

Supporting Information

Oxidative Decarbonylation Coupling of Anilines with Isocyanates for the Synthesis of Unsymmetric Azobenzenes

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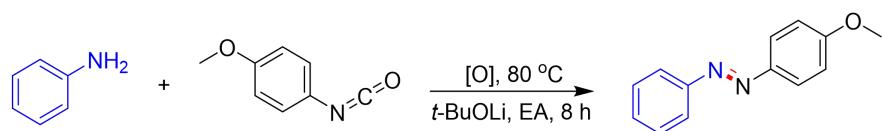
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A. General information

Melting points were measured on an Electrothermal SGW-X4 microscopy digital melting point apparatus and are uncorrected; ^1H -NMR, ^{13}C -NMR and ^{19}F NMR spectra were obtained on Bruker-500 or Bruker-400 and referenced to 7.28 ppm and 77.16 ppm for chloroform solvent with TMS as internal standard (0 ppm). Chemical shifts were reported in parts per million (ppm, δ) downfield from tetramethylsilane. Proton coupling patterns are described as singlet (s), doublet (d), triplet (t), multiplet (m); Column chromatography was performed on silica gel (200-300 mesh) using ethyl acetate/petroleum ether; Unless otherwise stated, all the reagents, including anilines **1** and isocyanates **2** were purchased from commercial sources (Energy Chemical, Bide Medicine, J&K Chemie, TCI, Fluka, Acros, SCRC, Macklin), used without further purification. Mass spectroscopy data of the products were collected on an HRMS-QTOF instrument.

B. The process of optimizing reaction conditions

Table S1. Screening of the oxidant ^a



Entry	Oxidant	Yield ^b (%)
1	PhIO	trace
2	PIFA	trace
3	PIDA	61
4	IBX	trace
5	PhI[Ph(CO)O] ₂	66
6	PhI[4-OMe-Ph(CO)O] ₂	75

7	PhI[4-NO ₂ -Ph(CO)O] ₂	24
8	PhI[4- <i>t</i> -Bu-Ph(CO)O] ₂	56
9	H ₂ O ₂	n.d.
10	Nb ₂ O ₅ (10 %)+ H ₂ O ₂	trace
11	Nb ₂ O ₅ (1 eq.)+ H ₂ O ₂	n.d.
12	TBHP	n.d.
13	DTBP	n.d.
14	<i>m</i> -CPBA	n.d.
15	K ₂ FeO ₄	n.d.

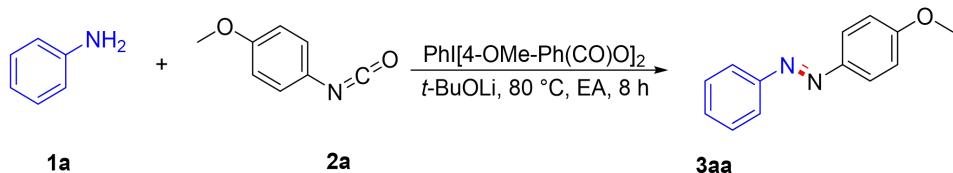
^a Reaction conditions: **1a** (0.3 mmol), **2a** (0.15 mmol), oxidant (2 eq.), *t*-BuOLi (1 eq.), and EA (1 mL) at 80 °C under air for 8 h. ^b Isolated yield. n.d. = not detected.

Table S2. Screening of the temperature ^a

1a	2a	3aa
Entry	Temperature	Yield ^b (%)
1	rt	trace
2	70	60
3	80	75
5	90	65
6	100	trace

^a Reaction conditions: **1a** (0.3 mmol), **2a** (0.15 mmol), PhI[4-OMe-Ph(CO)O]₂ (2 eq.), *t*-BuOLi (1 eq.), and EA (1 mL) at 80 °C under air for 8 h. ^b Isolated yield.

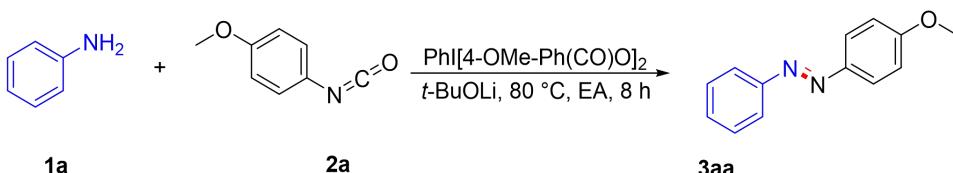
Table S3. Screening of aniline additions ^a



Entry	Aniline additions	Yield ^b (%)
1	1.0 eq.	34
2	1.5 eq.	57
3	2.0 eq.	75

^a Reaction conditions: **1a**, **2a** (0.15 mmol), PhI[4-OMe-Ph(CO)O]₂ (2 eq.), *t*-BuOLi (1 eq.), and EA (1 mL) at 80 °C under air for 8 h. ^b Isolated yield.

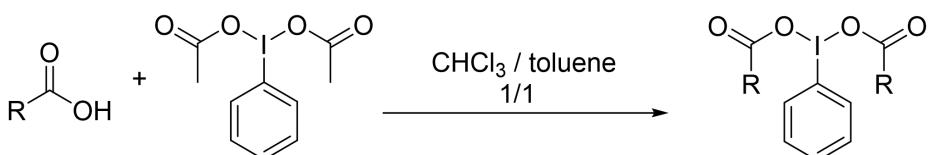
Table S4. Screening of solvent additions ^a



Entry	Solvent additions	Yield ^b (%)
1	1.0 mL	75
1	1.5 mL	70
2	2 mL	68
3	2.5 mL	60

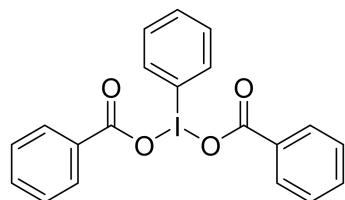
^a Reaction conditions: **1a** (0.3 mmol), **2a** (0.15 mmol), PhI[4-OMe-Ph(CO)O]₂ (2 eq.), *t*-BuOLi (1 eq.), and EA at 80 °C under air for 8 h. ^b Isolated yield.

C. Preparation of hypervalent iodine (III) reagents^[1]

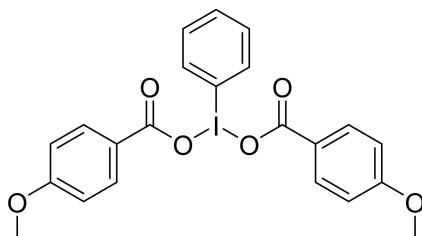


A mixture of (diacetoxyiodo) benzene (10 mmol), carboxylic acid (20 mmol) and CHCl_3 /toluene (1:1, v/v, 100 mL) were added to a pressure-resistant tube and stirred for 30 min at 40 °C using a heating mantle. After the reaction was finished, the solvent was removed in vacuo to give crude product, which was used directly in the following reaction without any further purification.

Characterization of hypervalent iodine (III) reagents



Phenyl- λ^3 -iodanediyl dibenzoate^[1]: white solid (3.8 g, 86% yield); ^1H NMR (500 MHz, CDCl_3) δ 8.29 – 8.25 (m, 2H), 7.96 (d, $J = 7.1$ Hz, 4H), 7.67 – 7.63 (m, 1H), 7.57 (dd, $J = 8.3, 7.0$ Hz, 2H), 7.54 – 7.50 (m, 2H), 7.39 (t, $J = 7.7$ Hz, 4H); ^{13}C NMR (125 MHz, CDCl_3) δ 171.4, 134.9, 132.6, 131.7, 131.0, 130.2, 130.1, 128.2, 122.3.



Phenyl- λ^3 -iodanediyl bis(4-methoxybenzoate e)^[1]: white solid (4.0 g, 80% yield); ^1H NMR (500 MHz, CDCl_3) δ 8.24 (d, $J = 7.9$ Hz, 2 H), 7.90 (d, $J = 8.5$ Hz, 4H), 7.62 (t, $J = 7.2$ Hz, 1H), 7.55 (t, $J = 7.8$ Hz, 2H), 6.86 (d, $J = 8.7$ Hz, 4H), 3.84 (s, 6H); ^{13}C NMR (125 MHz, CDCl_3) δ 171.2, 163.1, 134.9, 132.1, 131.6, 130.9, 122.6, 122.3, 113.5, 55.4.

D. Typical procedure for the preparation of azobenzene 3

Typical procedure for preparation of 3aa-3as

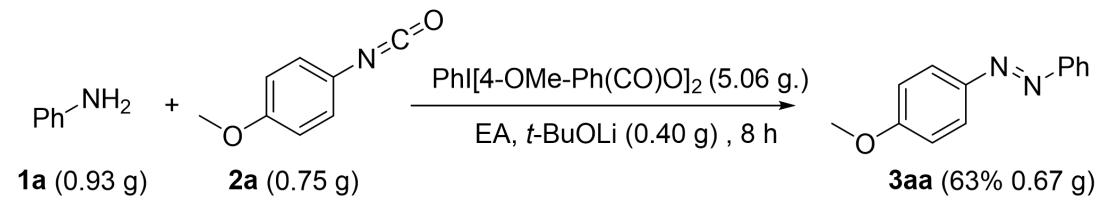
Aniline compounds **1** (0.3 mmol), 4-methoxyphenyl isocyanate **2a** (0.15 mmol, 23.4 mg, 1eq), *t*-BuOLi (0.15 mmol, 12.0 mg), PhI[4-OMe-Ph(CO)O]₂ (0.3 mmol, 151.8

mg) and EA (1.0 mL) were added to a 10 mL pressure-resistant tube. The reaction tube was sealed, and it was stirred for 8 h at 80 °C by heating the mantle. After the reaction was completed and cooling to room temperature, the reaction mixture was concentrated in vacuo. The crude mixture was purified by column chromatography using silica gel with eluent (petroleum ether: ethyl acetate = 50: 1) to afford the **3aa-3as**.

Typical procedure for preparation of **3ba-3bk**

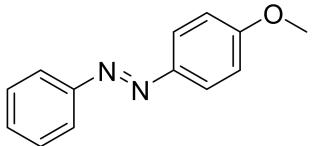
Aniline **1a** (0.3 mmol, 27.9 mg, 2eq), isocyanates **2** (0.15 mmol), *t*-BuOLi (0.15 mmol, 12.0 mg), PhI[4-OMe-Ph(CO)O]₂ (0.3 mmol, 151.8 mg) and EA (1.0 mL) were added to a 10 mL pressure-resistant tube under air. The reaction tube was sealed, and it was stirred for 8 h at 80 °C by heating the mantle. After the reaction was completed and cooling to room temperature, the reaction mixture was concentrated in vacuo. The crude mixture was purified by column chromatography using silica gel with eluent (petroleum ether: ethyl acetate = 100: 1) to afford the **3ba-3bk**.

Gram reaction of **3aa**

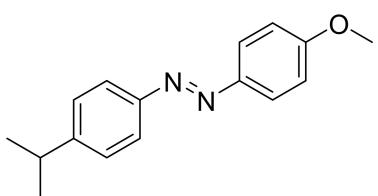


Aniline **1a** (10 mmol, 0.93 g), 4-methoxyphenyl isocyanate **2a** (5 mmol, 0.75 g), *t*-BuOLi (5 mmol, 0.40 g), PhI[4-OMe-Ph(CO)O]₂ (10 mmol, 5.06 g) and ethyl acetate (10.0 mL) were added to a 25 mL pressure-resistant tube under air. The reaction tube was sealed, and it was stirred for 8 h at 80 °C by heating the mantle. After the reaction was completed and cooling to room temperature, the reaction mixture was concentrated in vacuo. The crude mixture was purified by column chromatography using silica gel with eluent (petroleum ether: ethyl acetate = 50: 1) to afford the **3aa** (0.67 g, 63% yield).

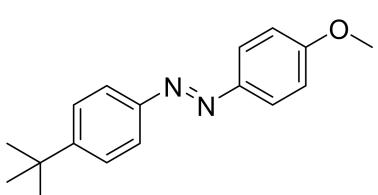
E. Spectroscopic data of azobenzene 3



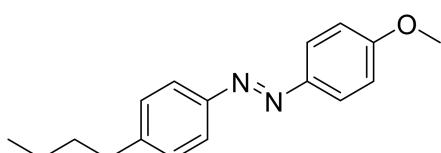
1-(4-Methoxyphenyl)-2-phenyldiazene (3aa)^[2]: Orange solid (23.9 mg, 75% yield); m.p. 52-54 °C; $R_f = 0.49$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.96 (d, $J = 9.0$ Hz, 2H), 7.91 (d, $J = 7.3$ Hz, 2H), 7.53 (t, $J = 7.5$ Hz, 2H), 7.47 (t, $J = 7.3$ Hz, 1H), 7.05 (d, $J = 9.0$ Hz, 2H), 3.92 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 162.1, 152.8, 147.0, 130.4, 129.1, 124.8, 122.6, 114.2, 55.6; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{12}\text{N}_2\text{O} [\text{M}]^+$ 212.0950; found 212.0945.



1-(4-Isopropylphenyl)-2-(4-methoxyphenyl)diazene (3ab): yellow solid (21.3 mg, 56% yield); m.p. 78-80 °C; $R_f = 0.49$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.93 (d, $J = 9.0$ Hz, 2H), 7.84 (d, $J = 8.4$ Hz, 2H), 7.38 (d, $J = 8.4$ Hz, 2H), 7.04 (d, $J = 8.9$ Hz, 2H), 3.92 (s, 3H), 1.33 (s, 3H), 1.31 (s, 3H), 1.28 (s, 1H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.8, 151.7, 151.1, 147.1, 127.1, 124.6, 122.6, 114.2, 55.6, 34.1, 23.9; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{16}\text{H}_{18}\text{N}_2\text{O} [\text{M}+\text{H}]^+$ 255.1491; found 255.1486.

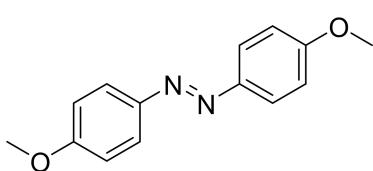


1-(4-(tert-Butyl)phenyl)-2-(4-methoxyphenyl)diazene (3ac): Orange solid (24.1 mg, 60% yield); m.p. 78-80 °C; $R_f = 0.50$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.93 (d, $J = 8.9$ Hz, 2H), 7.84 (d, $J = 8.6$ Hz, 2H), 7.54 (d, $J = 8.6$ Hz, 2H), 7.04 (d, $J = 9.0$ Hz, 2H), 3.92 (s, 3H), 1.40 (s, 9H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.8, 153.9, 150.7, 147.2, 126.0, 124.6, 122.3, 114.2, 55.6, 35.0 31.3; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{17}\text{H}_{20}\text{N}_2\text{O} [\text{M}+\text{H}]^+$ 269.1648; found 269.1642.

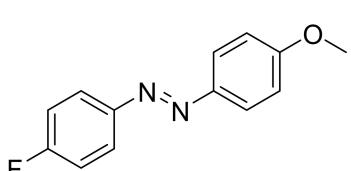


1-(4-Butylphenyl)-2-(4-methoxyphenyl)diazene (3ad): Orange solid (23.3 mg, 58% yield); m.p. 78-80 °C; $R_f = 0.45$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.93 (d, $J = 9.0$ Hz, 2H), 7.82

(d, $J = 8.2$ Hz, 2H), 7.33 (d, $J = 8.0$ Hz, 2H), 7.03 (d, $J = 9.0$ Hz, 2H), 3.92 (s, 3H), 2.70 (t, $J = 7.7$ Hz, 2H), 1.67 (t, $J = 7.0$ Hz, 2H), 1.43 – 1.38 (m, 2H), 0.97 (t, $J = 7.3$ Hz, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.8, 151.0, 147.1, 145.9, 129.1, 124.6, 122.5, 114.2, 55.6, 35.6, 33.5, 22.4, 14.0; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{17}\text{H}_{20}\text{N}_2\text{O} [\text{M}+\text{H}]^+$ 269.1648; found 269.1641.



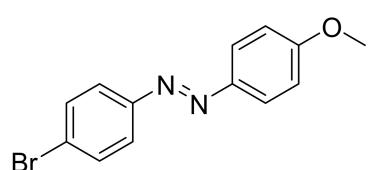
1,2-Bis(4-methoxyphenyl)diazene (3ae)^[3]: Orange solid (22.9 mg, 63% yield); m.p. 160 – 162 °C; $R_f = 0.45$ (petroleum ether: ethyl acetate = 25: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.92 – 7.89 (m, 4H), 7.05 – 7.01 (m, 4H), 3.91 (s, 6H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.6, 147.1, 124.4, 114.2, 55.6; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{14}\text{H}_{14}\text{N}_2\text{O}_2 [\text{M}+\text{H}]^+$ 243.1134; found 243.1138.



1-(4-Fluorophenyl)-2-(4-methoxyphenyl)diazene (3af)^[4]: Orange solid (24.8 mg, 72% yield); m.p. 96–98 °C; $R_f = 0.48$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.94 – 7.90 (m, 4H), 7.22 – 7.19 (m, 2H), 7.04 (d, $J = 9.0$ Hz, 2H), 3.92 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 164.0 (d, $J = 250.0$ Hz), 162.1, 149.3 (d, $J = 2.5$ Hz), 146.8, 124.7, 124.5 (d, $J = 8.8$ Hz), 116.0 (d, $J = 22.5$ Hz), 114.3, 55.6; ^{19}F NMR (471 MHz, Chloroform-*d*) δ -110.41 – -110.51 (m). HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{11}\text{FN}_2\text{O} [\text{M}+\text{H}]^+$ 231.0929; found 231.0928.

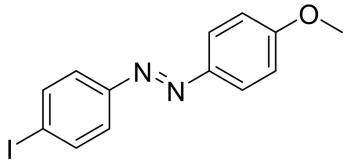


1-(4-Chlorophenyl)-2-(4-methoxyphenyl)diazene (3ag)^[4]: Orange solid (22.5 mg, 61% yield); m.p. 127–129 °C; $R_f = 0.49$ (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.94 (d, $J = 8.9$ Hz, 2H), 7.85 (d, $J = 8.6$ Hz, 2H), 7.49 (d, $J = 8.7$ Hz, 2H), 7.04 (d, $J = 9.0$ Hz, 2H), 3.92 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 162.3, 151.1, 146.9, 136.2, 129.3, 124.9, 123.8, 114.3, 55.7; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{11}\text{ClN}_2\text{O} [\text{M}+\text{H}]^+$ 247.0633; found 247.0632.



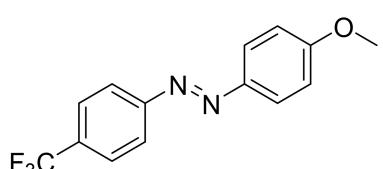
1-(4-Bromophenyl)-2-(4-methoxyphenyl)diazene

(3ah)^[4]: Orange solid (28.7 mg, 66% yield); m.p. 166–168 °C; R_f = 0.50 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (400 MHz, Chloroform-*d*) δ 7.94 (d, J = 8.9 Hz, 2H), 7.78 (d, J = 8.6 Hz, 2H), 7.65 (d, J = 8.7 Hz, 2H), 7.04 (d, J = 9.0 Hz, 2H), 3.92 (s, 3H); ^{13}C NMR (100 MHz, Chloroform-*d*) δ 162.4, 151.5, 146.9, 132.3, 124.9, 124.6, 124.1, 114.3, 55.6; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{11}\text{BrN}_2\text{O}$ [M+H]⁺ 291.0128; found 291.0127.



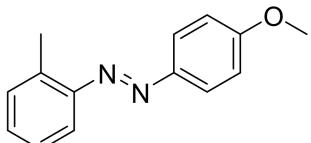
1-(4-Iodophenyl)-2-(4-methoxyphenyl)diazene (3ai):

Orange solid (36.8 mg, 65% yield); m.p. 178–180 °C; R_f = 0.50 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.94 (d, J = 9.0 Hz, 2H), 7.86 (d, J = 8.5 Hz, 2H), 7.64 (d, J = 8.5 Hz, 2H), 7.04 (d, J = 9.0 Hz, 2H), 3.92 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 162.4, 152.1, 146.9, 138.3, 125.0, 124.3, 114.3, 96.7, 55.6; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{11}\text{IN}_2\text{O}$ [M+H]⁺ 338.9988; found 338.9994.



1-(4-Methoxyphenyl)-2-(4-(trifluoromethyl)phenyl)

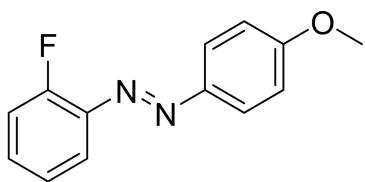
diazene (3aj)^[4]: Orange solid (25.2 mg, 60% yield); m.p. 133–135 °C; R_f = 0.45 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, CDCl_3) δ 7.98 (d, J = 8.9 Hz, 4H), 7.78 (d, J = 8.3 Hz, 2H), 7.06 (d, J = 8.9 Hz, 2H), 3.94 (s, 3H); ^{13}C NMR (125 MHz, CDCl_3) δ 162.8, 154.7, 146.9, 131.4 (q, J = 32.5 Hz), 126.2 (q, J = 3.8 Hz), 125.2, 124.2 (q, J = 272.3 Hz) 122.7, 114.4, 55.7; ^{19}F NMR (471 MHz, Chloroform-d) δ -62.43 HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{14}\text{H}_{11}\text{F}_3\text{N}_2\text{O}$ [M+H]⁺ 281.0897; found 281.0894.



1-(4-Methoxyphenyl)-2-(o-tolyl)diazene (3ak): Orange

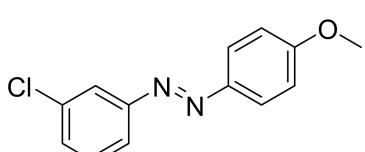
oil (22.0 mg, 65% yield); m.p. 53–55 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.95 (d, J = 8.9 Hz, 2H), 7.64 – 7.61 (m, 1H), 7.35 (d, J = 5.1 Hz, 2H), 7.29 (s, 1H), 7.04 (d, J = 9.0 Hz, 2H), 3.92 (s, 3H), 2.73 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.9, 150.8, 147.5, 137.6,

131.2, 130.3, 126.4, 124.8, 115.4, 114.2, 55.6, 17.5; HRMS (ESI Q-TOF) m/z: calcd for C₁₄H₁₄N₂O [M+H]⁺ 227.1178; found 227.1171.



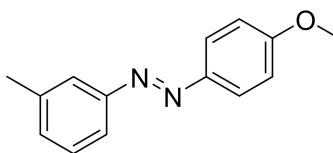
1-(2-Fluorophenyl)-2-(4-methoxyphenyl)diazene

(3al): Orange solid (23.5 mg, 68% yield); m.p. 98-100 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.98 (d, *J* = 9.0 Hz, 2H), 7.76 (t, *J* = 7.8 Hz, 1H), 7.47 – 7.40 (m, 1H), 7.28 – 7.20 (m, 2H), 7.04 (d, *J* = 9.0 Hz, 2H), 3.93 (s, 3H); ¹³C NMR (125 MHz, Chloroform-*d*) δ 162.5, 159.9(d, *J* = 262.5 Hz), 147.3, 140.8(d, *J* = 6.3 Hz), 131.7 (d, *J* = 8.8 Hz), 125.2, 124.3(d, *J* = 3.8 Hz), 117.8, 117.0 (d, *J* = 18.8 Hz), 114.3, 55.6; ¹⁹F NMR (471 MHz, Chloroform-*d*) δ -125.03 – -125.13 (m); HRMS (ESI Q-TOF) m/z: calcd for C₁₃H₁₁FN₂O [M+H]⁺ 231.0929; found 231.0927.



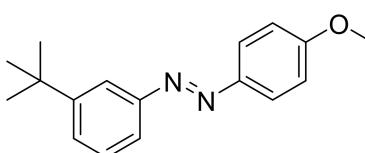
1-(3-Chlorophenyl)-2-(4-methoxyphenyl)diazene

(3am): Orange solid (25.8 mg, 70% yield); m.p. 128-130 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.95 (d, *J* = 9.0 Hz, 2H), 7.89 (t, *J* = 1.9 Hz, 1H), 7.81 (d, *J* = 7.6 Hz, 1H), 7.48 – 7.42 (m, 2H), 7.05 (d, *J* = 9.0 Hz, 2H), 3.93 (s, 3H); ¹³C NMR (125 MHz, Chloroform-*d*) δ 162.5, 153.6, 146.8, 135.1, 130.08, 130.1, 125.1, 122.1, 121.5, 114.3, 55.7; HRMS (ESI Q-TOF) m/z: calcd for C₁₃H₁₁ClN₂O [M+H]⁺ 247.0632; found 247.0629.



1-(4-Methoxyphenyl)-2-(m-tolyl)diazene (3an): Orange

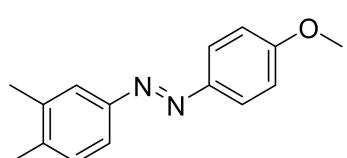
solid (20.7 mg, 61% yield); m.p. 78-80 °C R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.94 (d, *J* = 9.0 Hz, 2H), 7.72 (d, *J* = 1.8 Hz, 2H), 7.42 (dd, *J* = 8.4, 7.5 Hz, 1H), 7.29 (s, 1H), 7.04 (d, *J* = 9.0 Hz, 2H), 3.92 (s, 3H), 2.48 (s, 3H); ¹³C NMR (125 MHz, Chloroform-*d*) δ 162.0, 152.9, 147.1, 138.9, 131.2, 128.9, 124.7, 122.6, 120.2, 114.2, 55.6, 21.4; HRMS (ESI Q-TOF) m/z: calcd for C₁₄H₁₄N₂O [M+H]⁺ 227.1178; found 227.1174.



1-(3-(tert-Butyl)phenyl)-2-(4-methoxyphenyl)diazene

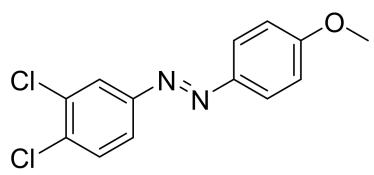
e (3ao): Orange solid (21.3 mg, 53% yield); m.p.

68-70 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.95 (d, J = 8.9 Hz, 3H), 7.71 (d, J = 7.6 Hz, 1H), 7.51 (d, J = 7.0 Hz, 1H), 7.45 (t, J = 7.7 Hz, 1H), 7.05 (d, J = 8.9 Hz, 2H), 3.92 (s, 3H), 1.43 (s, 9H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.9, 152.8, 152.3, 147.1, 128.7, 127.5, 124.7, 120.7, 118.8, 114.2, 55.6, 35.0, 31.4; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{17}\text{H}_{20}\text{N}_2\text{O}$ [M+H]⁺ 269.1648; found 269.1643.



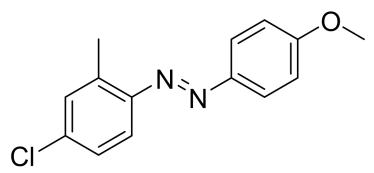
1-(3,4-Dimethylphenyl)-2-(4-methoxyphenyl)diazene

(3ap): Orange solid (18.4 mg, 51% yield); m.p. 78-80 °C; R_f = 0.49 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (400 MHz, Chloroform-*d*) δ 7.92 (d, J = 9.0 Hz, 2H), 7.71 – 7.63 (m, 2H), 7.30 – 7.27 (m, 1H), 7.05 – 7.01 (m, 2H), 3.91 (s, 3H), 2.38 (s, 3H), 2.36 (s, 3H); ^{13}C NMR (100 MHz, Chloroform-*d*) δ 161.8, 151.2, 147.1, 1396, 137.4, 130.2, 124.5, 123.2, 120.6, 114.2, 55.6, 19.9, 19.9; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}$ [M+H]⁺ 241.1335; found 241.1331.



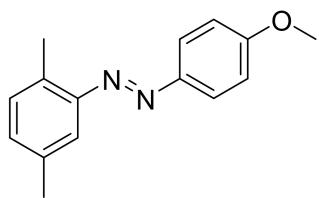
1-(3,4-Dichlorophenyl)-2-(4-methoxyphenyl)diazene

(3aq): Orange solid (22.3 mg, 58% yield); m.p. 178-180 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 8.00 (s, 1H), 7.94 (d, J = 9.0 Hz, 2H), 7.77 (d, J = 8.5 Hz, 1H), 7.60 (d, J = 8.5 Hz, 1H), 7.04 (d, J = 9.0 Hz, 2H), 3.93 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 162.7, 151.7, 146.7, 134.1, 133.4, 130.8, 125.2, 123.6, 122.6, 114.4, 55.7; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{10}\text{Cl}_2\text{N}_2\text{O}$ [M+H]⁺ 281.0242; found 281.0234.

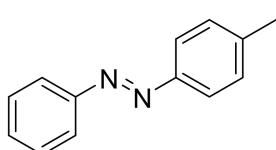


1-(4-Chloro-2-methylphenyl)-2-(4-methoxyphenyl)diazene

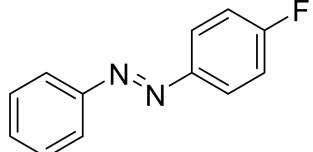
(3ar): Orange solid (19.5 mg, 50% yield); m.p. 188-190 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.93 (d, J = 8.9 Hz, 2H), 7.60 (d, J = 8.6 Hz, 1H), 7.34 (s, 1H), 7.26 – 7.23 (m, 1H), 7.04 (d, J = 8.9 Hz, 2H), 3.92 (s, 3H), 2.70 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 162.1, 149.2, 147.3, 139.3, 135.6, 130.9, 126.7, 124.9, 116.8, 114.3, 55.6, 17.4; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{14}\text{H}_{13}\text{ClN}_2\text{O}$ [M+H]⁺ 261.0789; found 261.0786.



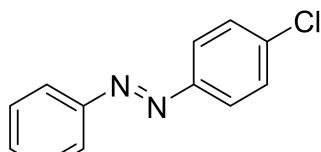
1-(2,5-Dimethylphenyl)-2-(4-methoxyphenyl)diazene (3as): Orange solid (19.8 mg, 55% yield); m.p. 80.0 – 82.0 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 50: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.96 – 7.92 (m, 2H), 7.44 (d, J = 1.9 Hz, 1H), 7.24 (d, J = 7.7 Hz, 1H), 7.18 (dd, J = 7.7, 1.9 Hz, 1H), 7.04 (d, J = 9.0 Hz, 2H), 3.92 (s, 3H), 2.68 (s, 3H), 2.39 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 161.8, 150.6, 147.5, 136.1, 134.6, 131.2, 131.0, 124.7, 115.7, 114.2, 55.6, 21.1, 17.1; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{15}\text{H}_{16}\text{N}_2\text{O}$ [M+H]⁺ 241.1335; found 241.1325.



1-Phenyl-2-(p-tolyl)diazene (3ba)^[2]: Orange solid (18.5 mg, 63% yield); m.p. 65-67 °C; R_f = 0.52 (petroleum ether: ethyl acetate = 100: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.95 – 7.91 (m, 2H), 7.86 (d, J = 8.2 Hz, 2H), 7.54 (t, J = 7.4 Hz, 2H), 7.49 (d, J = 7.0 Hz, 1H), 7.34 (d, J = 8.0 Hz, 2H), 2.47 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 152.7, 150.8, 141.6, 130.7, 129.8, 129.1, 122.9, 122.7, 21.5; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{13}\text{H}_{12}\text{N}_2$ [M]⁺ 196.1000; found 196.0997.

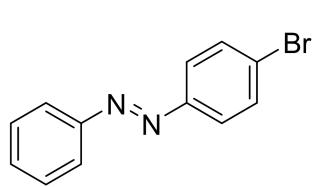


1-(4-Fluorophenyl)-2-phenyldiazene (3bb)^[2]: Orange solid (19.2 mg, 64% yield); m.p. 76-78 °C; R_f = 0.50 (petroleum ether: ethyl acetate = 100: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 8.03 – 7.94 (m, 4H), 7.59 – 7.50 (m, 3H), 7.24 (t, J = 8.6 Hz, 2H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 164.4 (d, J = 250.0 Hz), 152.5, 149.2 (d, J = 2.5 Hz), 131.1, 129.2, 124.9 (d, J = 8.8 Hz), 122.9, 116.1 (d, J = 22.5 Hz); ^{19}F NMR (471 MHz, Chloroform-*d*) δ -109.28. HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{12}\text{H}_9\text{FN}_2$ [M]⁺ 200.0750; found 200.0754.



1-(4-Chlorophenyl)-2-phenyldiazene (3bc)^[2]: Orange solid (19.4 mg, 60% yield); m.p. 81-83 °C; R_f = 0.50 (petroleum ether: ethyl acetate = 100: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.96 (d, J = 7.3 Hz, 2H), 7.92 (d, J = 8.7 Hz, 2H), 7.58 – 7.51 (m, 5H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 152.5,

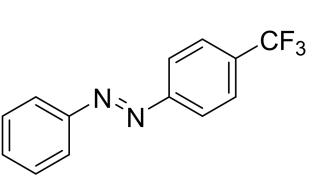
151.0, 137.0, 131.3, 129.4, 129.2, 124.2, 123.0; HRMS (ESI Q-TOF) m/z: calcd for C₁₂H₉ClN₂ [M]⁺ 216.0454; found 216.0452.



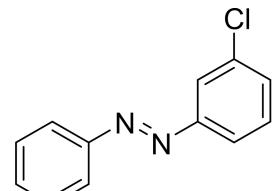
1-(4-Bromophenyl)-2-phenyldiazene (3bd)^[2]: Orange solid (22.6 mg, 58% yield); m.p. 90-92 °C; R_f = 0.50 (petroleum ether: ethyl acetate = 100: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.94 (d, *J* = 6.9 Hz, 2H), 7.83 (d, *J* = 8.6 Hz, 2H), 7.68 (d, *J* = 8.7 Hz, 2H), 7.53 (dd, *J* = 10.6, 7.2 Hz, 3H); ¹³C NMR (125 MHz, Chloroform-*d*) δ 152.5, 151.4, 132.4, 131.4, 129.2, 125.4, 124.4, 123.0; HRMS (ESI Q-TOF) m/z: calcd for C₁₂H₉BrN₂ [M]⁺ 259.9949; found 259.9945.



1-(4-Iodophenyl)-2-phenyldiazene (3be)^[2]: Orange solid (25.3 mg, 55% yield); m.p. 91-93 °C; R_f = 0.52 (petroleum ether: ethyl acetate = 100: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.94 (d, *J* = 6.8 Hz, 2H), 7.89 (d, *J* = 8.7 Hz, 2H), 7.68 (d, *J* = 8.7 Hz, 2H), 7.57-7.51 (m, 3H); ¹³C NMR (125 MHz, Chloroform-*d*) δ 152.5, 152.0, 138.4, 131.4, 129.2, 124.5, 123.0; HRMS (ESI Q-TOF) m/z: calcd for C₁₂H₉IN₂ [M]⁺ 307.9810; found 307.9805.

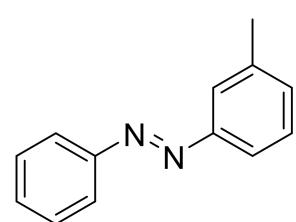


1-Phenyl-2-(4-(trifluoromethyl)phenyl)diazene (3bf)^[2]: Orange solid 19.9 mg, 53% yield); m.p. 81-83 °C; R_f = 0.52 (petroleum ether: ethyl acetate = 100: 1); ¹H NMR (400 MHz, Chloroform-*d*) δ 8.03 (d, *J* = 8.2 Hz, 2H), 7.99 (dd, *J* = 7.8, 1.9 Hz, 2H), 7.81 (d, *J* = 8.3 Hz, 2H), 7.57 (d, *J* = 6.6 Hz, 3H); ¹³C NMR (100 MHz, Chloroform-*d*) δ 154.4, 152.4, 132.2 (q, *J* = 40.0 Hz), 131.8, 129.2, 126.3 (q, *J* = 5 Hz), 124.0 (q, *J* = 337.5 Hz), 123.2, 123.0; ¹⁹F NMR (377 MHz, Chloroform-*d*) δ -62.53. HRMS (ESI Q-TOF) m/z: calcd for C₁₃H₉F₃N₂ [M]⁺ 250.0718; found 250.0715.

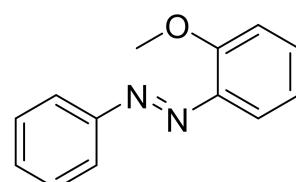


1-(3,4-Dimethylphenyl)-2-phenyldiazene (3bg): Orange solid (18.8 mg, 58% yield); m.p. 82-84 °C; R_f = 0.48 (petroleum ether: ethyl acetate = 100: 1); ¹H NMR (500 MHz, Chloroform-*d*) δ 7.98 – 7.92 (m, 3H), 7.87 (d, *J* = 6.1 Hz, 1H), 7.55 (dd, *J* = 10.1, 7.3 Hz, 3H), 7.49 (d, *J* = 6.1 Hz, 2H); ¹³C NMR (125 MHz, CDCl₃)

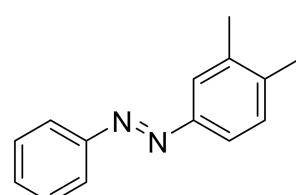
δ 153.5, 152.4, 135.2, 131.6, 130.7, 130.2, 129.2, 123.1, 122.4, 121.8; HRMS (ESI Q-TOF) m/z: calcd for $C_{12}H_9ClN_2 [M+H]^+$ 217.0527; found 217.0523.



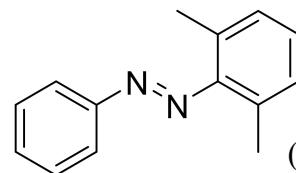
1-Phenyl-2-(m-tolyl)diazene (3bh)^[2]: Orange solid (17.1 mg, 58% yield); m.p. 65-67 °C; $R_f = 0.48$ (petroleum ether: ethyl acetate = 100: 1); 1H NMR (500 MHz, Chloroform-*d*) δ 7.98 – 7.94 (m, 2H), 7.77 (s, 2H), 7.56 (t, $J = 7.4$ Hz, 2H), 7.50 (t, $J = 7.2$ Hz, 1H), 7.47 – 7.42 (m, 1H), 7.33 (d, $J = 7.4$ Hz, 1H), 2.50 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 152.8, 152.7, 139.0, 131.8, 130.9, 129.1, 128.9, 123.0, 122.8, 120.5, 21.4; HRMS (ESI Q-TOF) m/z: calcd for $C_{13}H_{12}N_2 [M]^+$ 196.1000; found 196.0995.



1-(3-Chlorophenyl)-2-phenyldiazene (3bi)^[3]: Orange solid (22.3 mg, 70% yield); m.p. 56-58 °C; $R_f = 0.48$ (petroleum ether: ethyl acetate = 50: 1); 1H NMR (500 MHz, Chloroform-*d*) δ 7.99 – 7.94 (m, 2H), 7.72 (d, $J = 8.0$ Hz, 1H), 7.54 (t, $J = 7.5$ Hz, 2H), 7.51 – 7.45 (m, 2H), 7.15 – 7.11 (m, 1H), 7.08 – 7.03 (m, 1H), 4.05 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 157.0, 153.2, 142.3, 132.5, 130.8, 129.1, 123.0, 120.8, 117.0, 112.8, 56.4; HRMS (ESI Q-TOF) m/z: calcd for $C_{13}H_{12}N_2O [M+H]^+$ 213.1028; found 213.1022.



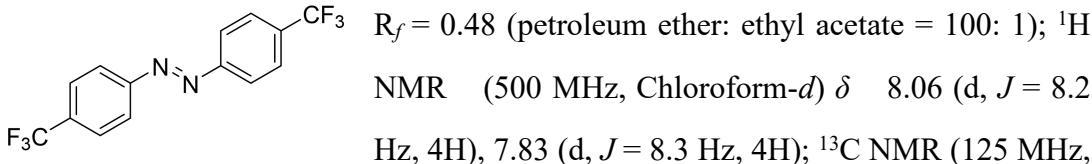
1-(3,4-Dimethylphenyl)-2-phenyldiazene (3bj)^[3]: Orange solid (17.6 mg, 56% yield); mp 78-80 °C; $R_f = 0.48$ (petroleum ether: ethyl acetate = 100: 1); 1H NMR (500 MHz, Chloroform-*d*) δ 7.92 (d, $J = 8.2$ Hz, 2H), 7.74 (s, 1H), 7.72 – 7.69 (m, 1H), 7.53 (t, $J = 7.5$ Hz, 2H), 7.48 (t, $J = 7.2$ Hz, 1H), 7.30 (d, $J = 8.0$ Hz, 1H), 2.39 (s, 3H), 2.37 (s, 3H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 152.8, 151.2, 140.3, 137.5, 130.7, 130.4, 129.1, 123.6, 122.8, 121.0, 20.0; HRMS (ESI Q-TOF) m/z: calcd for $C_{14}H_{14}N_2 [M+H]^+$ 211.1229; found 211.1231.



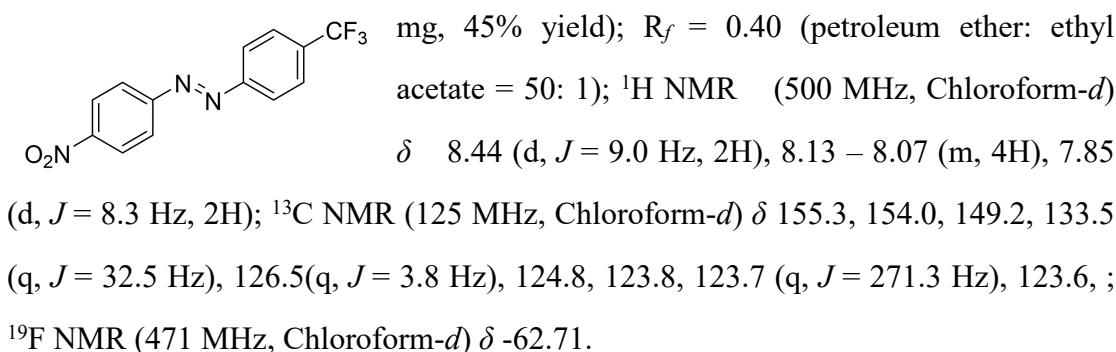
1-(2,6-Dimethylphenyl)-2-phenyldiazene (3bk): Orange solid (22.3 mg, 54% yield); mp 76-78 °C; $R_f = 0.48$ (petroleum ether: ethyl acetate = 100: 1); 1H NMR (500 MHz, Chloroform-*d*) δ 7.97 – 7.92 (m, 2H), 7.60 – 7.53 (m, 3H), 7.17 (t, $J = 4.6$ Hz, 3H),

2.39 (s, 6H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 152.8, 151.5, 131.1, 130.6, 129.1, 128.1, 122.5, 18.8; HRMS (ESI Q-TOF) m/z: calcd for $\text{C}_{14}\text{H}_{14}\text{N}_2$ [M+H]⁺ 211.1229; found 211.1226.

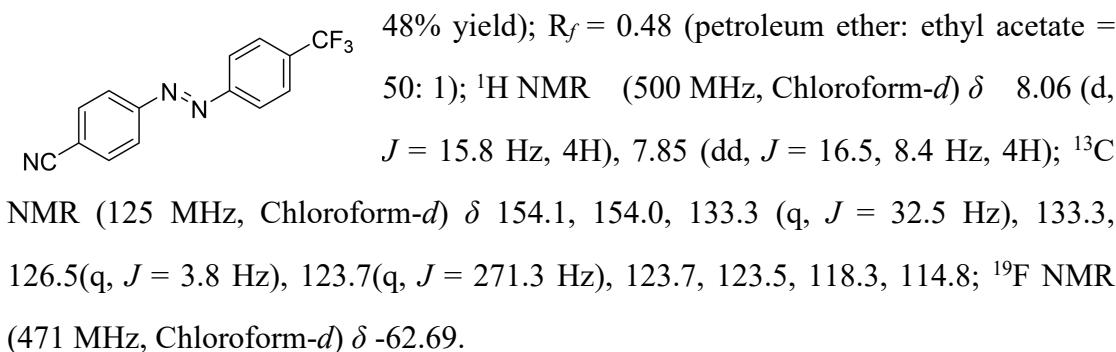
1,2-Bis(4-(trifluoromethyl)phenyl)diazene (3ca): Orange solid (25.3 mg, 53% yield);



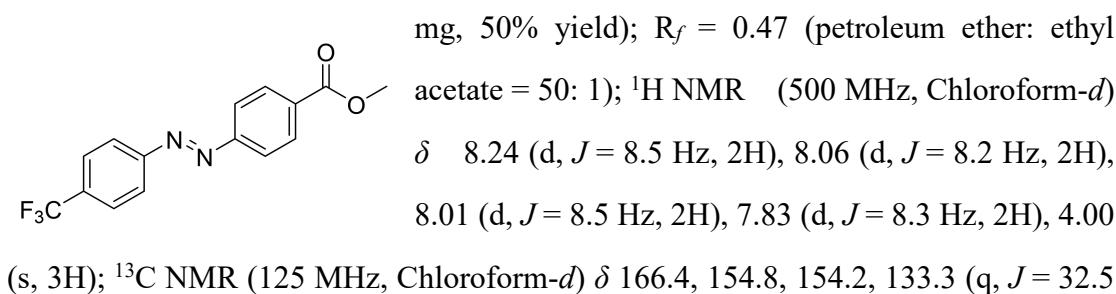
1-(4-Nitrophenyl)-2-(4-(trifluoromethyl)phenyl)diazene (3cb): Orange solid (19.9



4-((4-(Trifluoromethyl)phenyl)diazenyl)benzonitrile (3cc): Orange solid (19.8 mg,



Methyl -4-((4-(trifluoromethyl)phenyl)diazenyl)benzoate (3cd): Orange solid (23.1



Hz), 133.3, 130.7, 126.4(q, J = 3.8 Hz), 123.8(q, J = 271.3 Hz), 123.3, 122.7, 52.5; ^{19}F NMR (471 MHz, Chloroform-*d*) δ -62.62.

F. Potential application of this method

Typical procedure for preparation of 4aa: **3ae** (0.2 mmol), vinylene carbonate (0.4 mmol), $[\text{RhCp}^*\text{Cl}_2]_2$ (5 mol%), AgSbF_6 (30 mol%), NaOAc (30 mol%), DCE (1 mL) were stirred at 80 °C for 14 h. After the reaction was completed and cooled to room temperature, the reaction mixture was concentrated in vacuo. The crude residue was purified by column chromatography using silica gel and eluent (petroleum ether: ethyl acetate = 5:1) to obtain **4aa** in 65% yield.

(5-Methoxy-2-(4-methoxyphenyl)-2*H*-indazol-3-yl)methanol (4aa): orange liquid (37.1 mg, 65% yield); R_f = 0.45 (petroleum ether: ethyl acetate = 5: 1); ^1H NMR (500 MHz, Chloroform-*d*) δ 7.64 (d, J = 9.2 Hz, 1H), 7.58 (d, J = 8.9 Hz, 2H), 7.07 – 7.03 (m, 1H), 7.00 (d, J = 9.0 Hz, 2H), 6.90 (d, J = 2.3 Hz, 1H), 4.92 (s, 2H), 3.88 (s, 3H), 3.85 (s, 3H), 2.43 (s, 1H); ^{13}C NMR (125 MHz, Chloroform-*d*) δ 159.7, 155.4, 145.1, 132.9, 132.6, 126.6, 121.8, 121.7, 119.1, 114.3, 95.6, 55.6, 55.4, 54.2.

G. Control experiments

To investigate the reaction mechanism, we reacted phenylhydroxylamine **6a** (1 mmol, 109 mg) with 4-methoxyphenyl isocyanate **2a** (0.5 mmol, 74.5 mg), *t*-BuOLi (0.5 mmol, 40 mg), $\text{PhI}[4\text{-OMe-Ph}(\text{CO})\text{O}]_2$ (0.1 mmol, 53.8 mg), and EA (4 mL) for 0.5 hours, as illustrated in Fig. 1. The reaction intermediates were identified using high-resolution mass spectrometry (HRMS), and their molecular weights were determined accordingly.

HRMS (ESI) of 4-methoxy-N-((phenylamino)oxy)carbonyl)aniline **A**: calcd for C₁₄H₁₄N₂O₃ [M+H]⁺ 259.1077; found 259.1072.

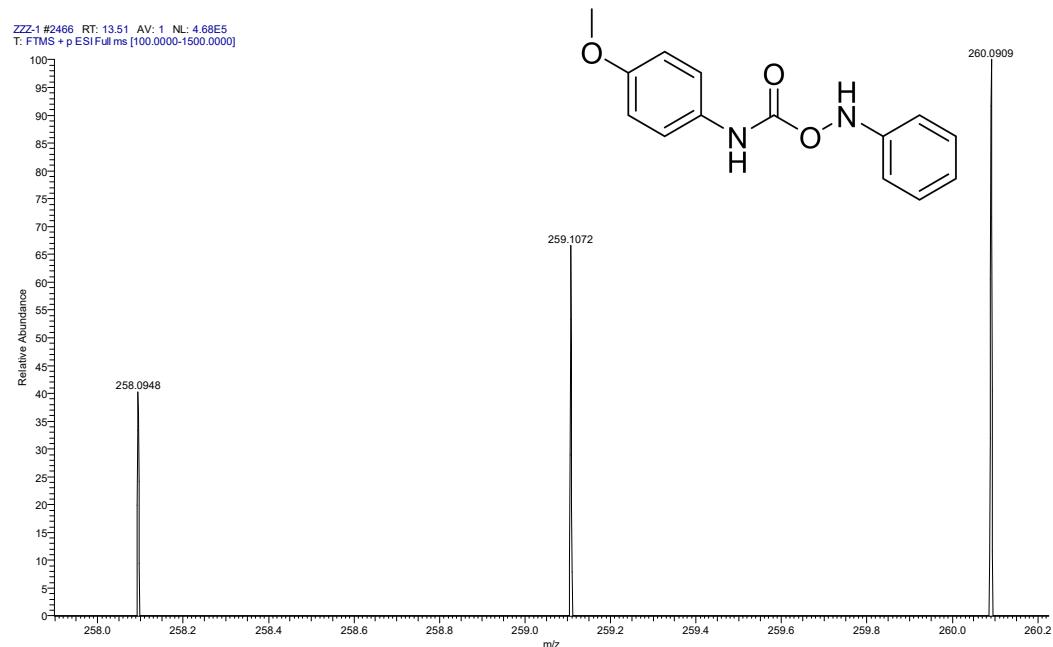


Figure 1. HRMS of Intermediates **A**

Experiment of competition experiment between **2a** and **2f**.

To a solution of aniline **1a** (0.3 mmol, 27.9 mg), 4-methoxyphenyl isocyanate **2a** (0.15 mmol, 22.4 mg), 4-trifluoromethylphenyl isocyanate **2f** (0.15 mmol, 28.1 mg), *t*-BuOLi (0.15 mmol, 0.12 g), and PhI[4-OMe-Ph(CO)O]₂ (0.3 mmol, 161.4 mg) in 2 mL of ethyl acetate, the reaction mixture was stirred at 80 °C for 1 hours. After completion, the mixture was cooled to room temperature. The ratio of peak areas of **3aa** and **3bg** was obtained as 1:1.2 by GC-MS.

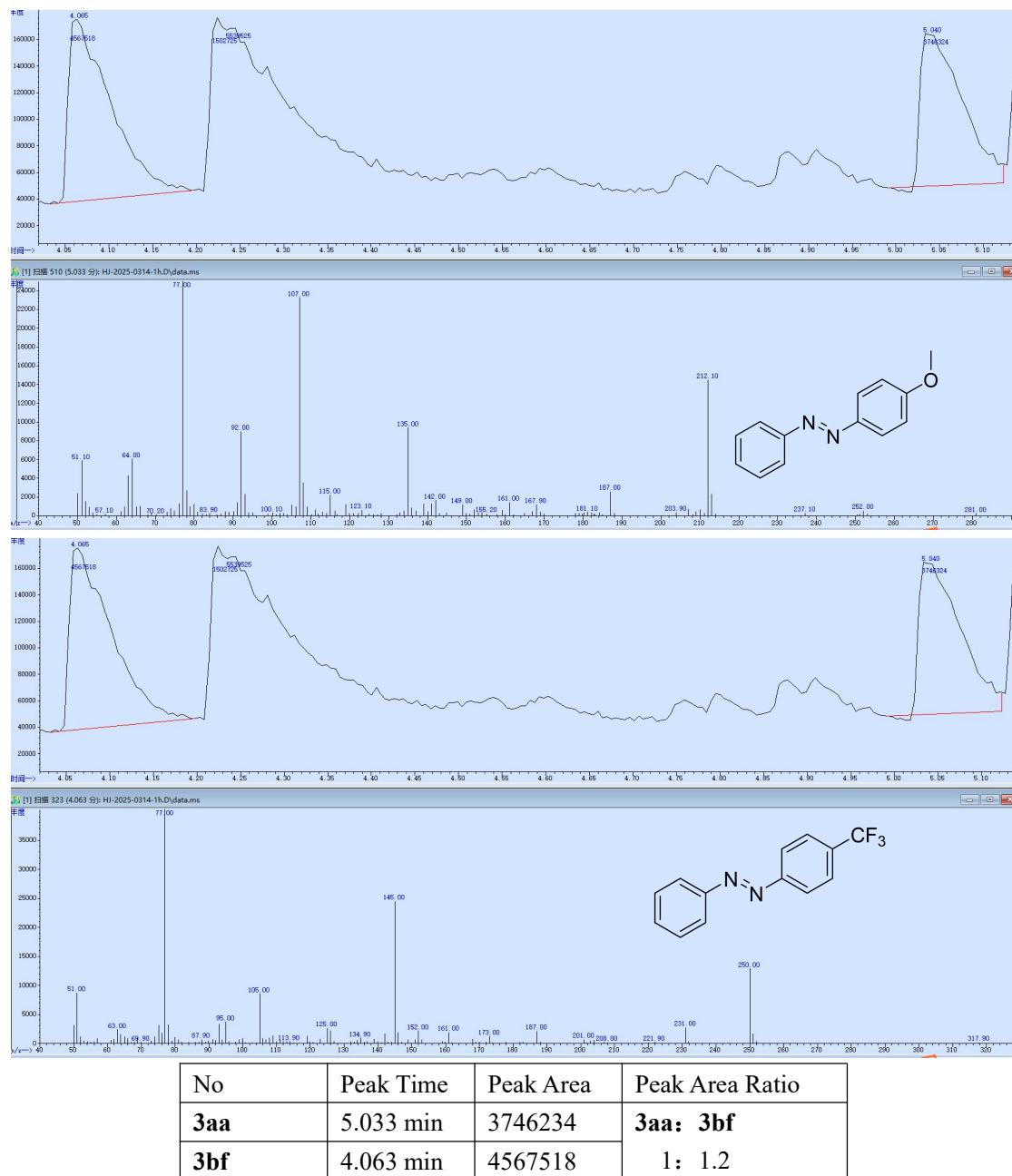


Table 1. **3aa** to **3bf** yield ratio

CO₂ verification experiment

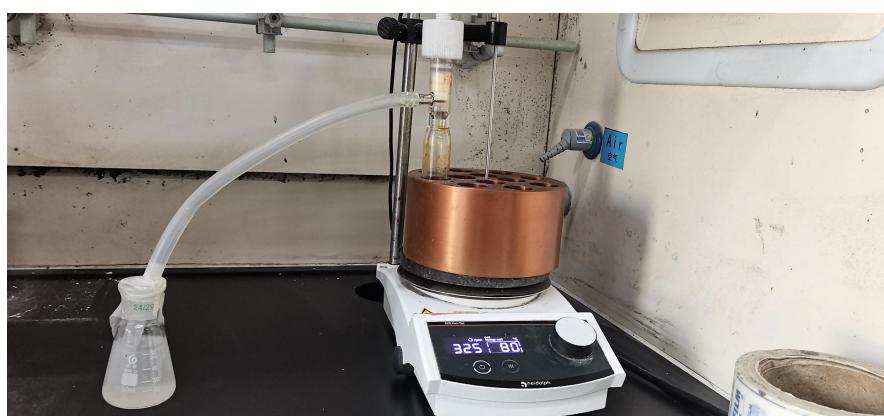
To a solution of aniline **1a** (4 mmol, 0.37 g), 4-methoxyphenyl isocyanate **2a** (2 mmol, 0.30 g), *t*-BuOLi (2 mmol, 0.16 g), PhI[4-OMe-Ph(CO)O]₂ (4 mmol, 2.02 g) in EA (10 mL). The reaction mixture was reacted at 80 °C for 8 h and cooled to room temperature. The branch port of the closed tube was introduced into a pre-prepared clarified saturated clear Ca(OH)₂ solution using arubber hose. Slowly twist the cork

and stop twisting the cork when there are stable and continuous bubbles, and the clarified $\text{Ca}(\text{OH})_2$ solution gradually becomes turbid after a continuous passage of the reacted gas mixture, which proves that CO_2 is produced during the reaction.

The diagram below shows the experimental setup:



Upon introduction of the gas mixture



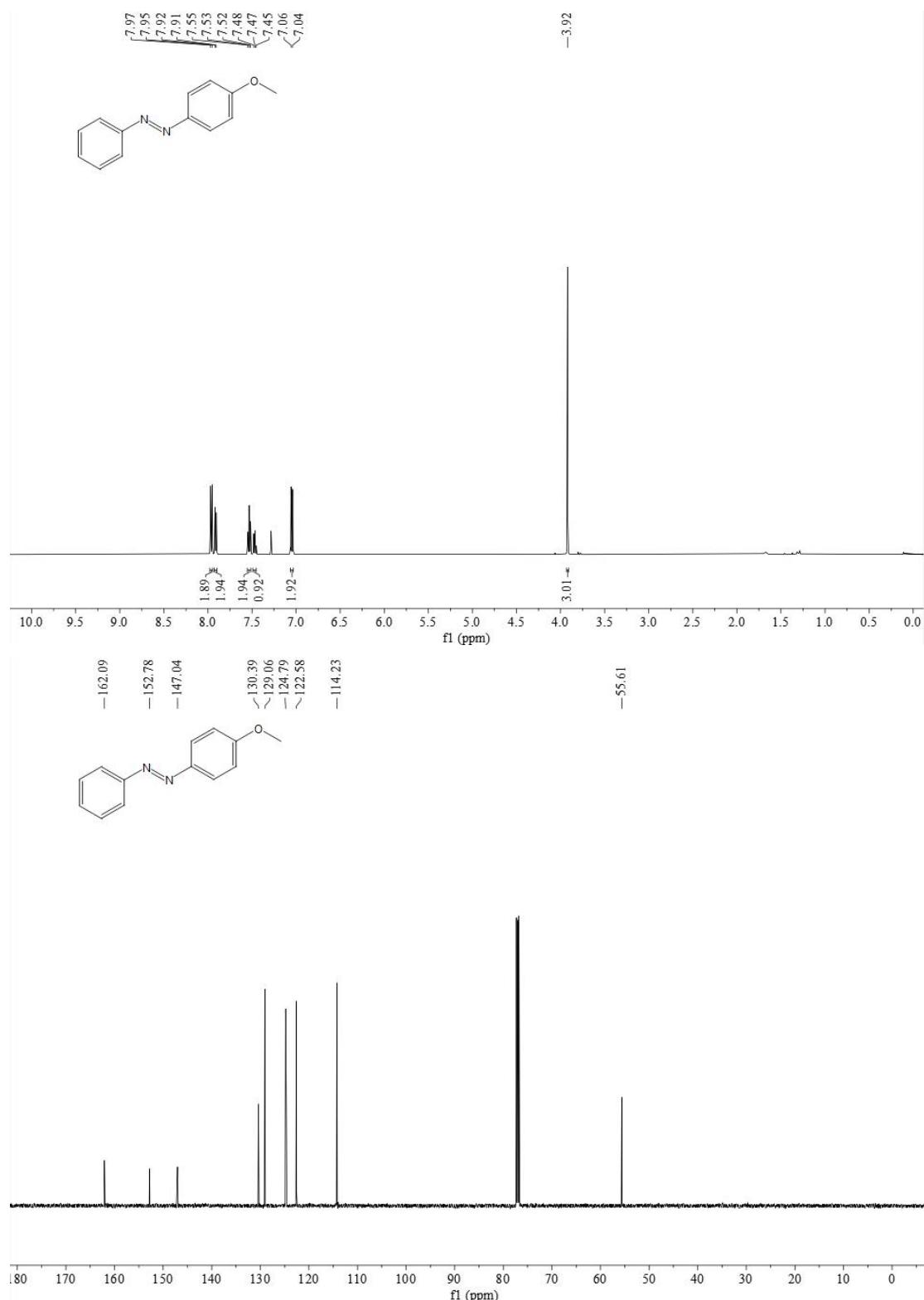
References

- [1] Yota Sakakibara, Kei Murakami, and Kenichiro Itami, C–H Acyloxylation of Polycyclic Aromatic Hydrocarbons. *Org. Lett.* 2022, **24**, 602–607.
- [2] Haiping Lv, Ronibala Devi Laishram, Jiayan Li, Photocatalyzed Oxidative Dehydrogenation of Hydrazobenzenes to Azobenzenes. *Green Chem.* 2019, **21**, 4055–4061.

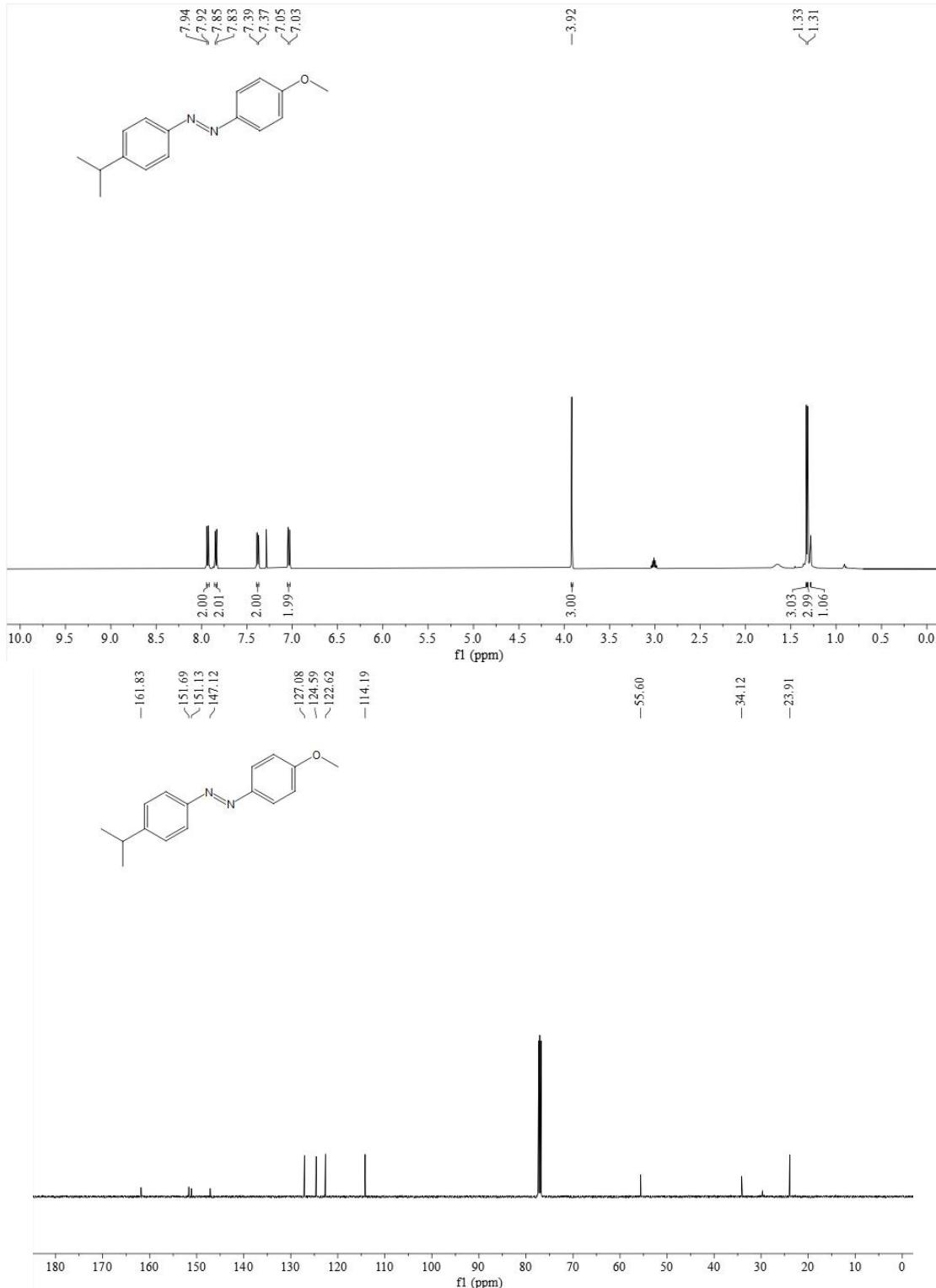
- [3] Van Hieu Tran, Hee-Kwon Kim, Visible-Light-Driven SAQS-Catalyzed Aerobic Oxidative Dehydrogenation of Alkyl 2-Phenylhydrazinecarboxylates. *RSC Adv.* 2022, **12**, 30304-30309.
- [4] L. Finck, M. Oestreich, Synthesis of Non-Symmetric Azoarenes by Palladium-Catalyzed Cross-Coupling of Silicon-Masked Diazenyl Anions and (Hetero)Aryl Halides. *Angew. Chem. Int. Ed.* 2022, **61**, e202210907.

H. Copies of ^1H , ^{13}C and ^{19}F NMR spectra

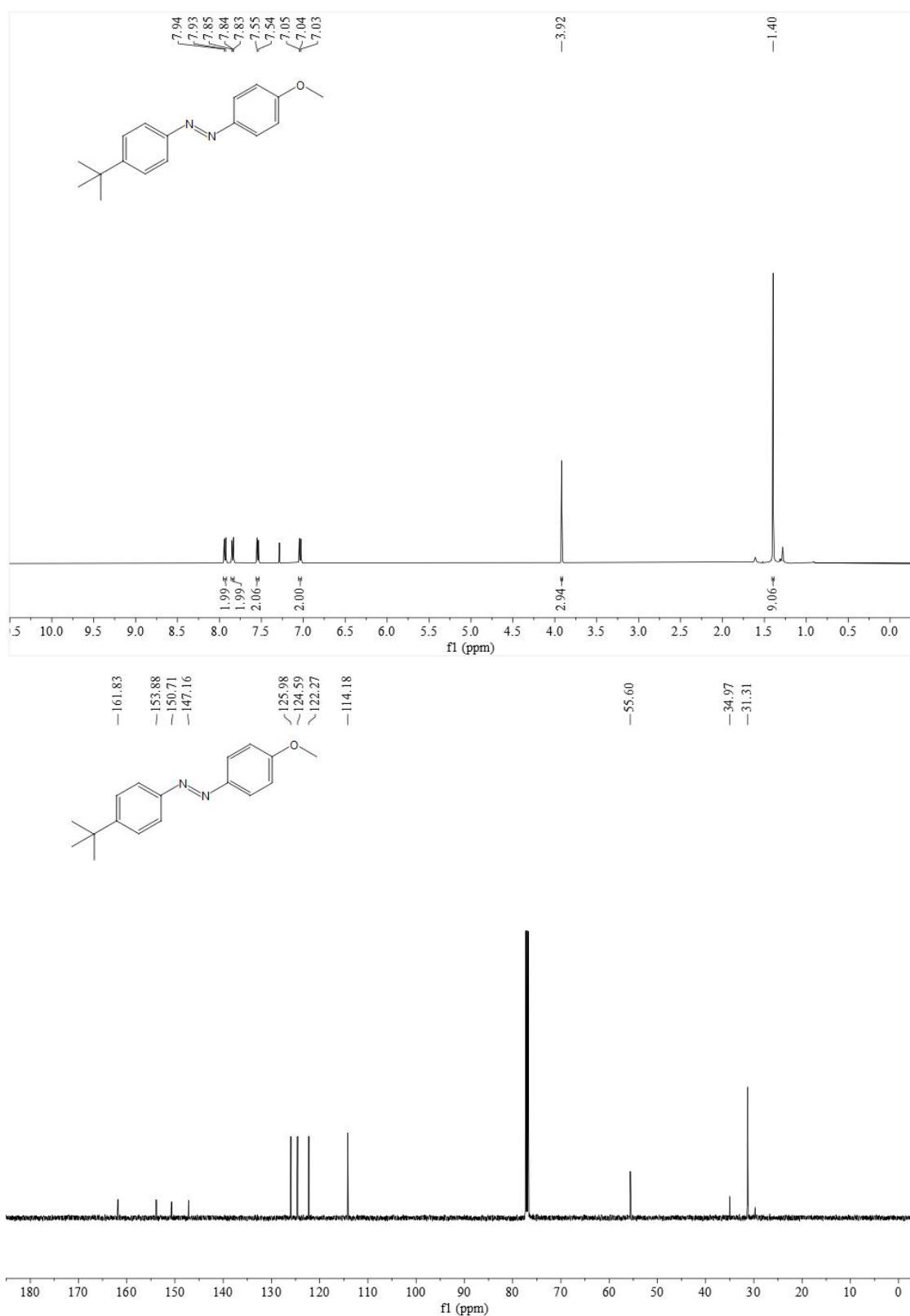
^1H NMR and ^{13}C NMR of 1-(4-Methoxyphenyl)-2-phenyldiazene 3aa (500 MHz, CDCl_3)



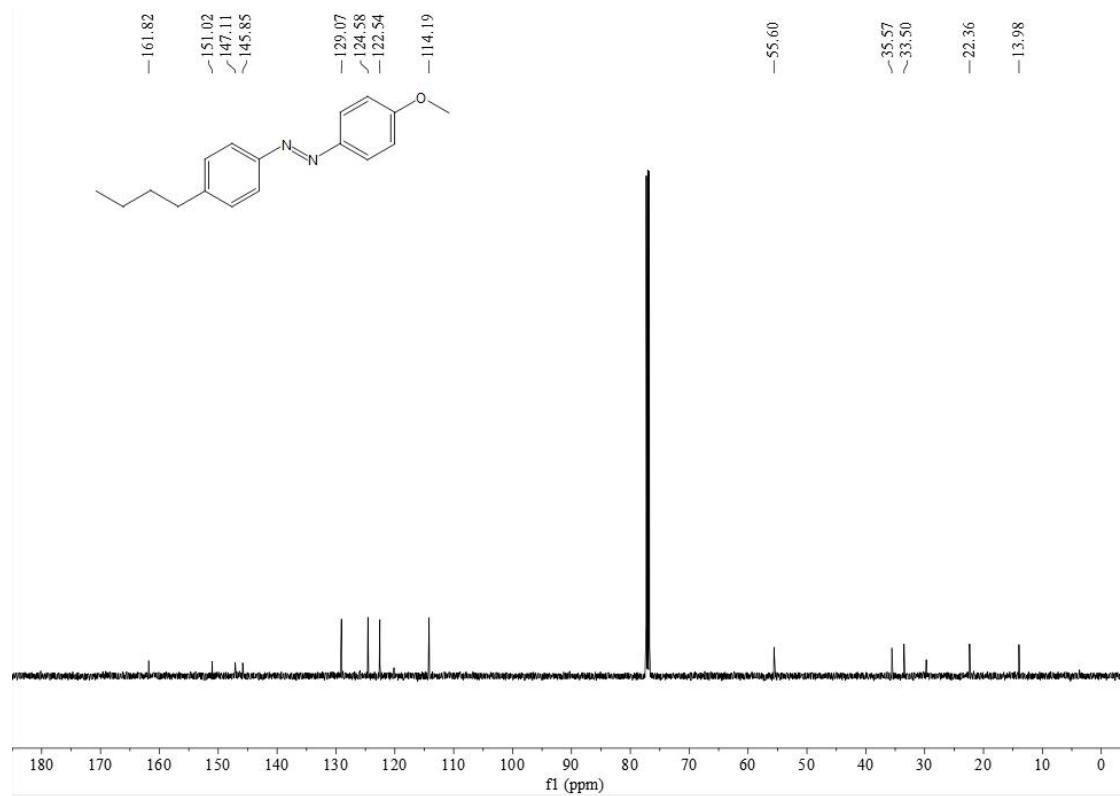
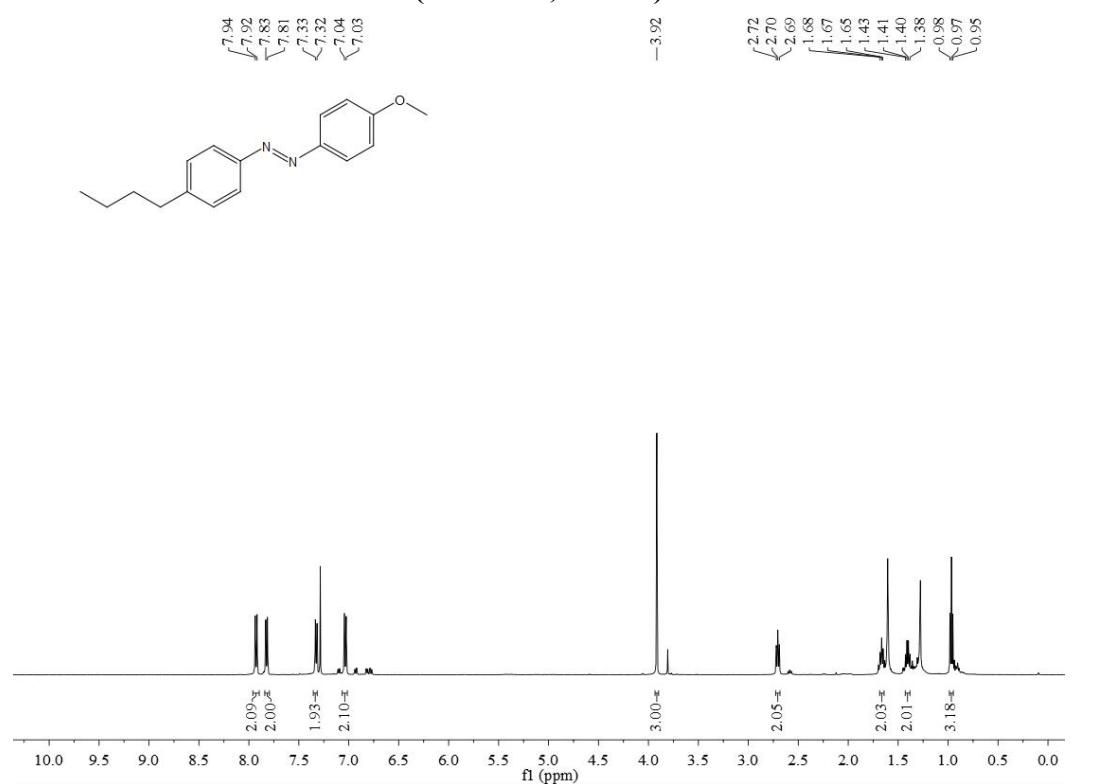
**¹H NMR and ¹³C NMR of 1-(4-Isopropylphenyl)-2-(4-methoxyphenyl) diazene
3ab (500 MHz, CDCl₃)**



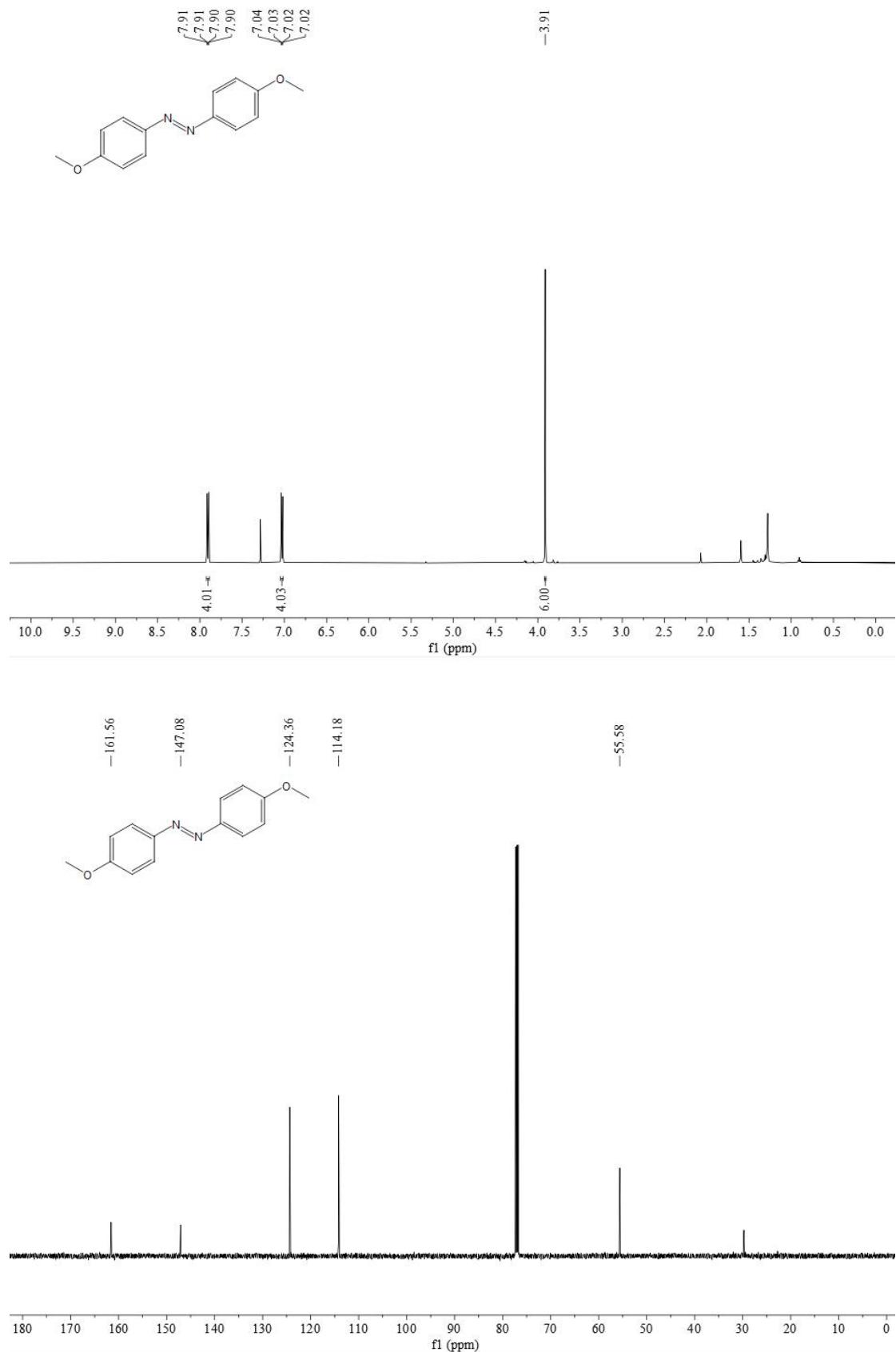
**¹H NMR and ¹³C NMR of 1-(4-(*tert*-Butyl)phenyl)-2-(4-methoxyphenyl)diazene
3ac (500 MHz, CDCl₃)**



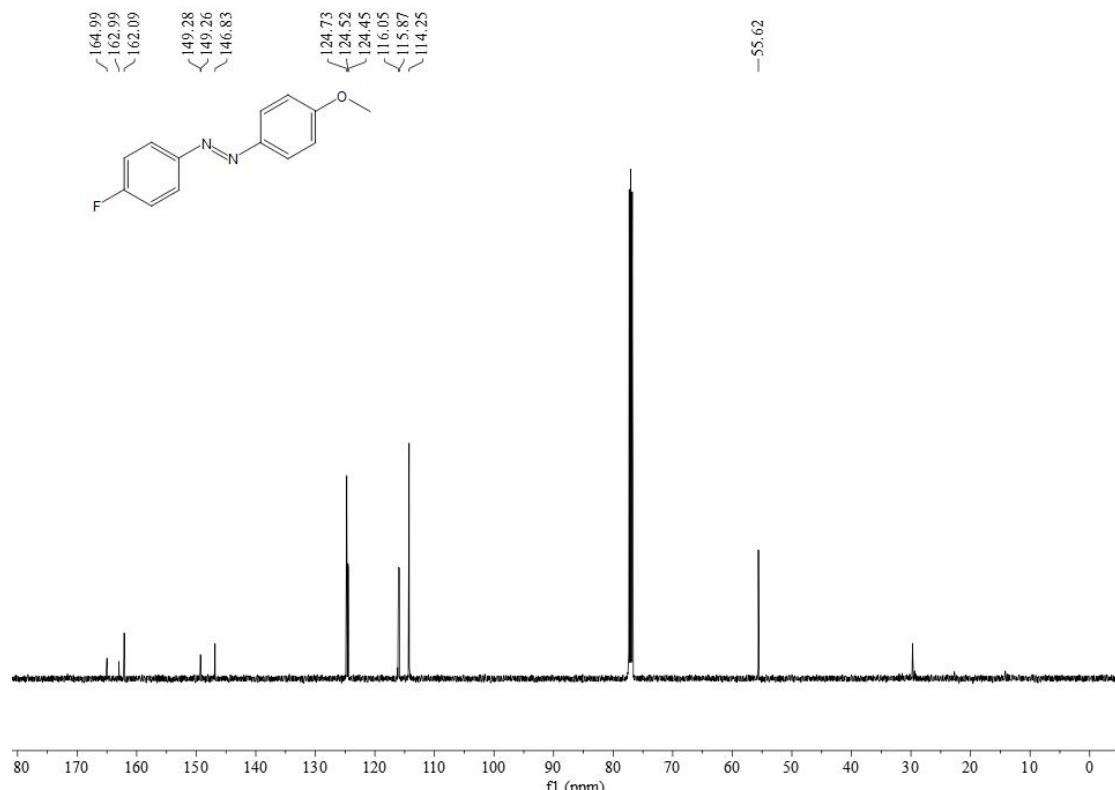
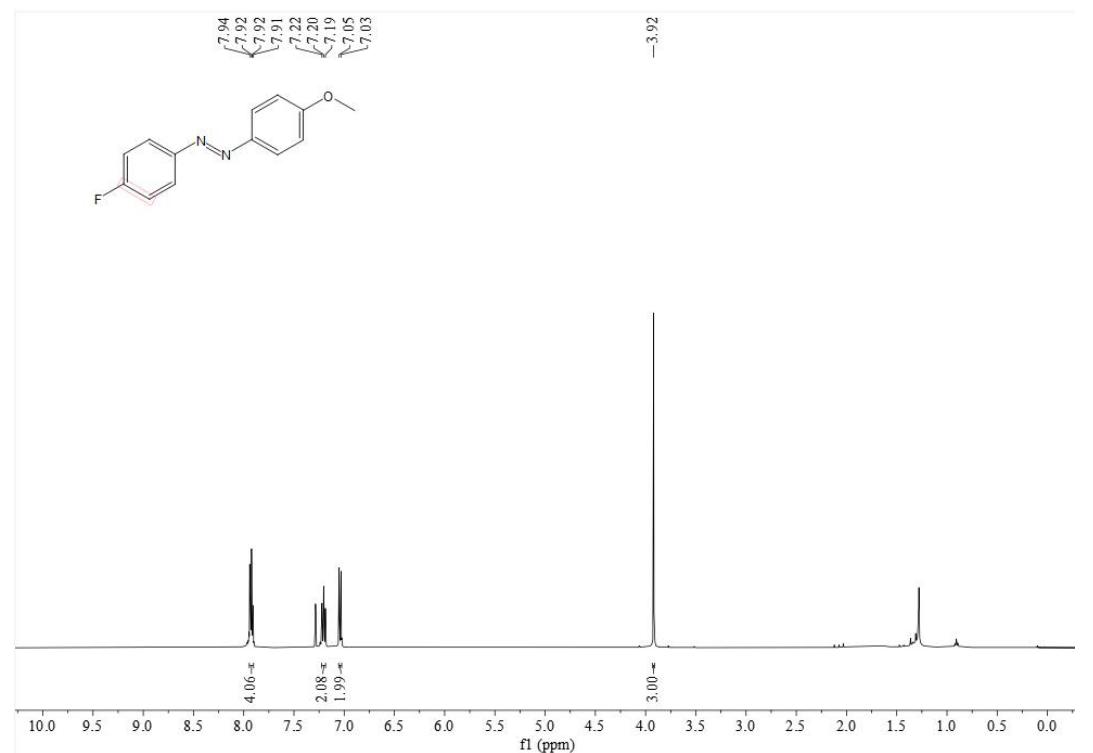
**¹H NMR and ¹³C NMR of 1-(4-Butylphenyl)-2-(4-methoxyphenyl)diazene 3ad
(500 MHz, CDCl₃)**

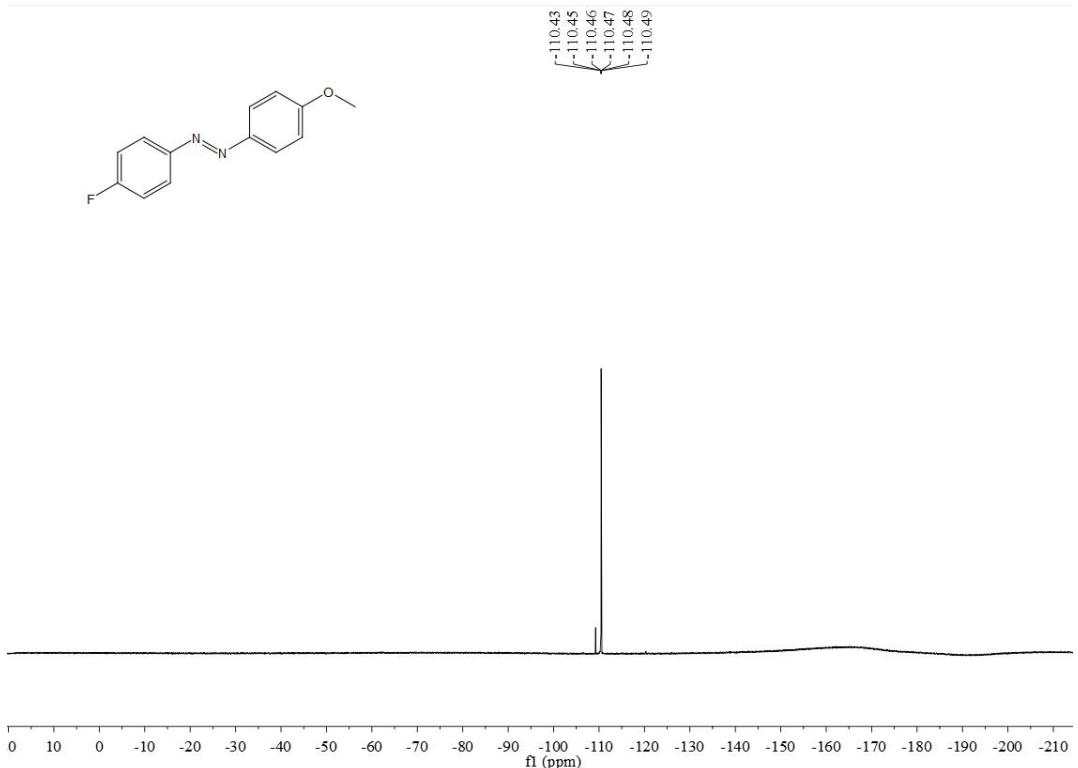


¹H NMR and ¹³C NMR of 1,2-Bis(4-methoxyphenyl)diazene 3ae (500 MHz, CDCl₃)

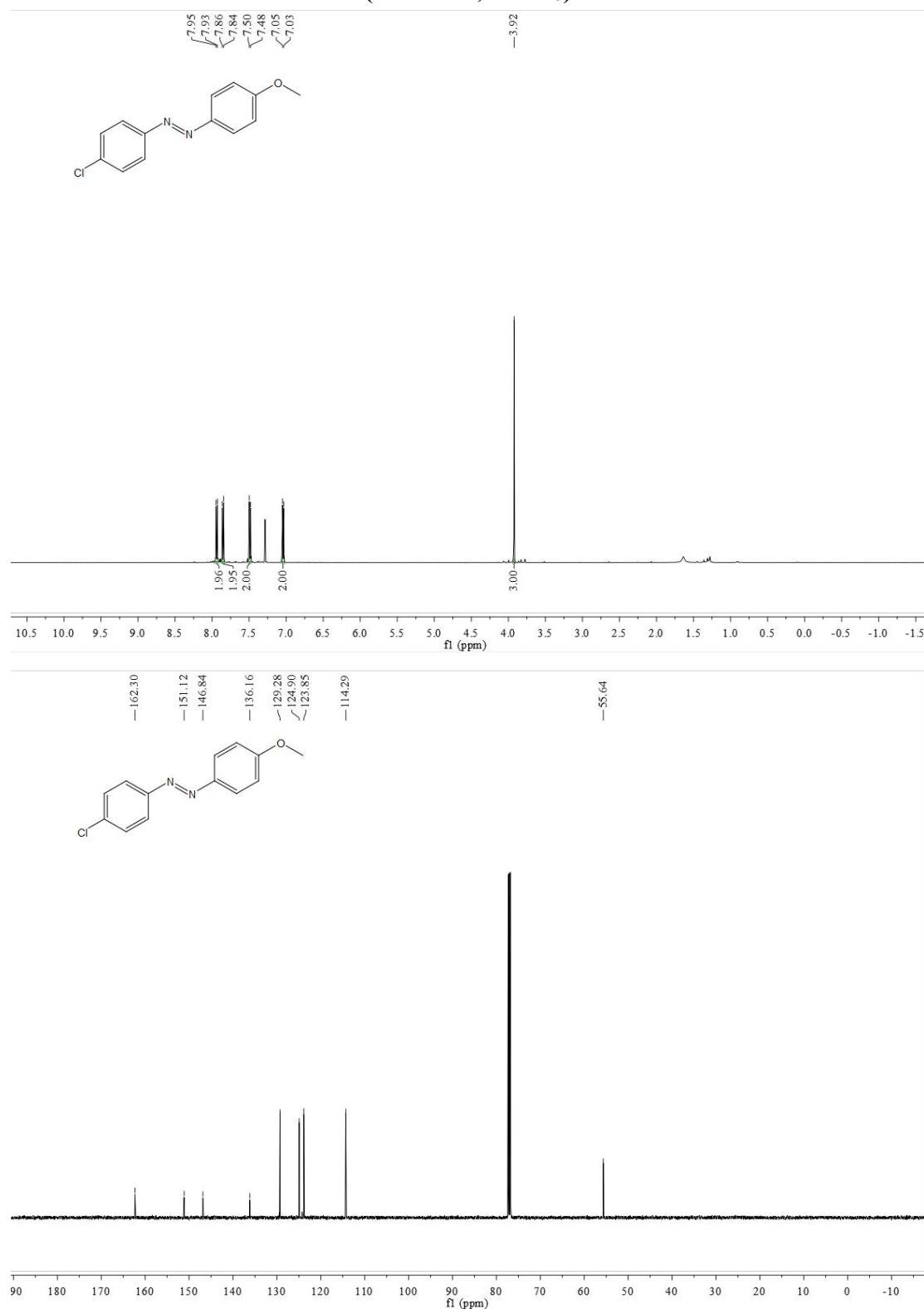


¹H NMR and ¹³C NMR and ¹⁹F NMR of 1-(4-Fluorophenyl)-2-(4-methoxyphenyl) diazene 3af (500 MHz, CDCl₃)

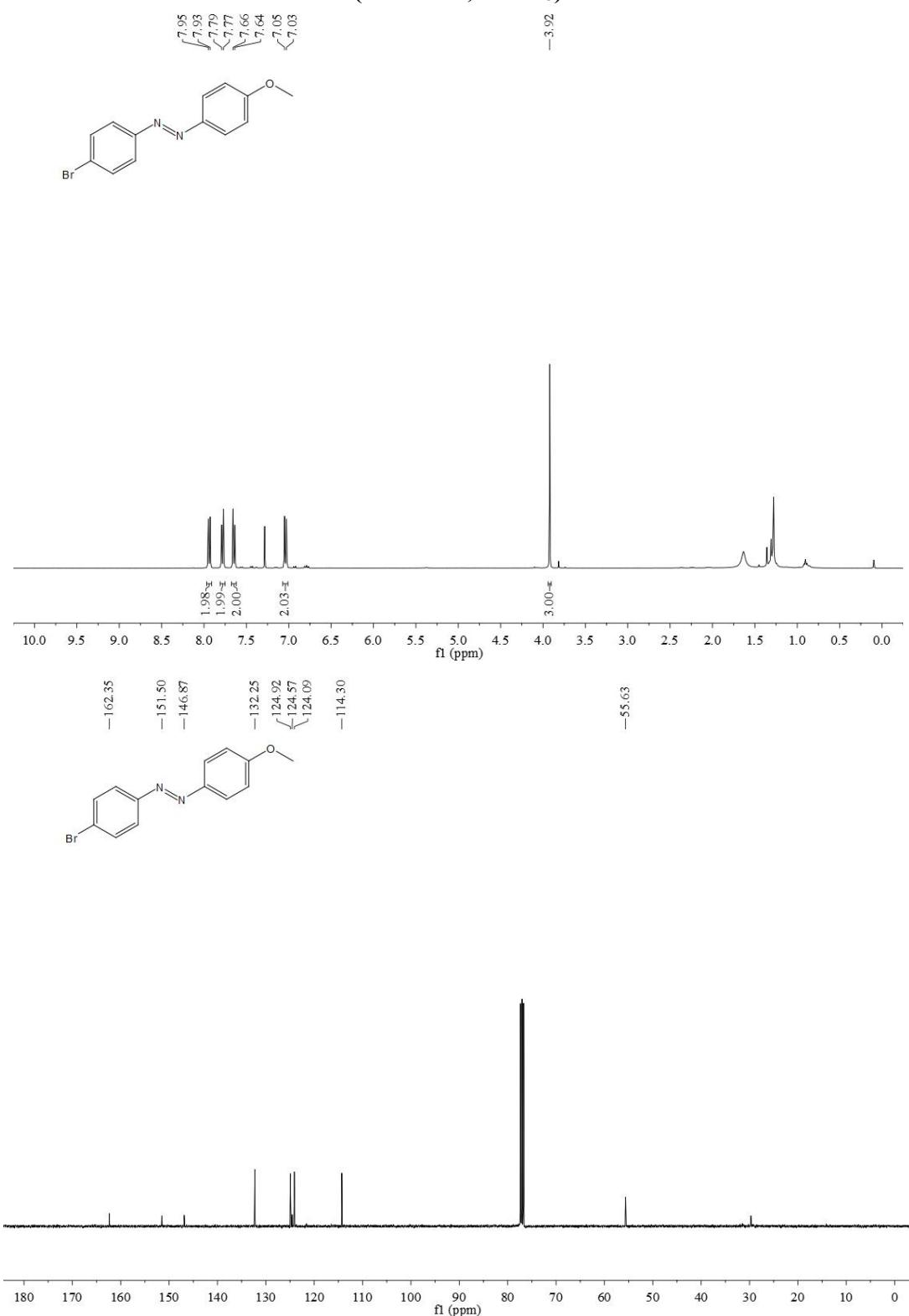




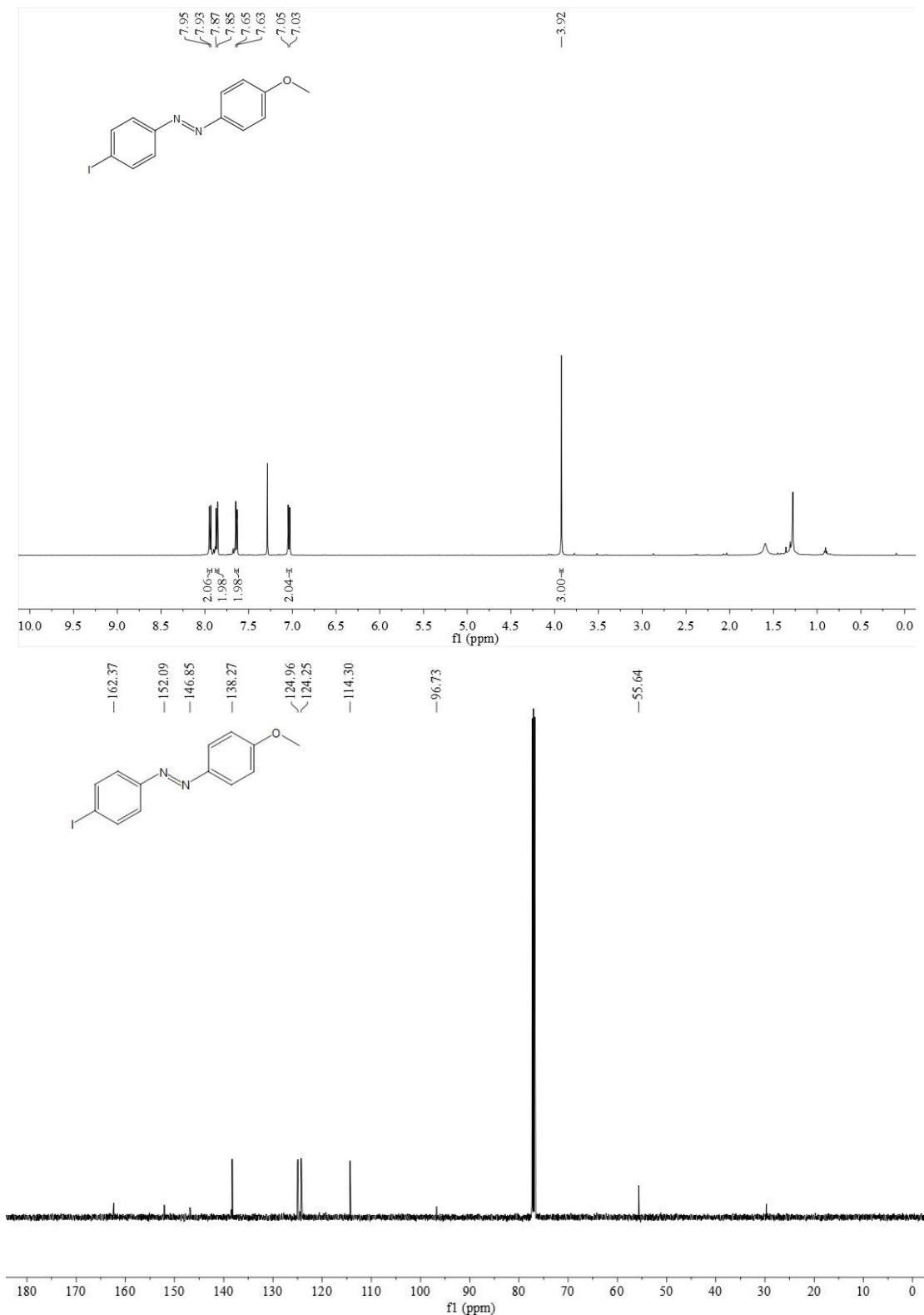
**¹H NMR and ¹³C NMR of 1-(4-Chlorophenyl)-2-(4-methoxyphenyl)diazene 3ag
(500 MHz, CDCl₃)**



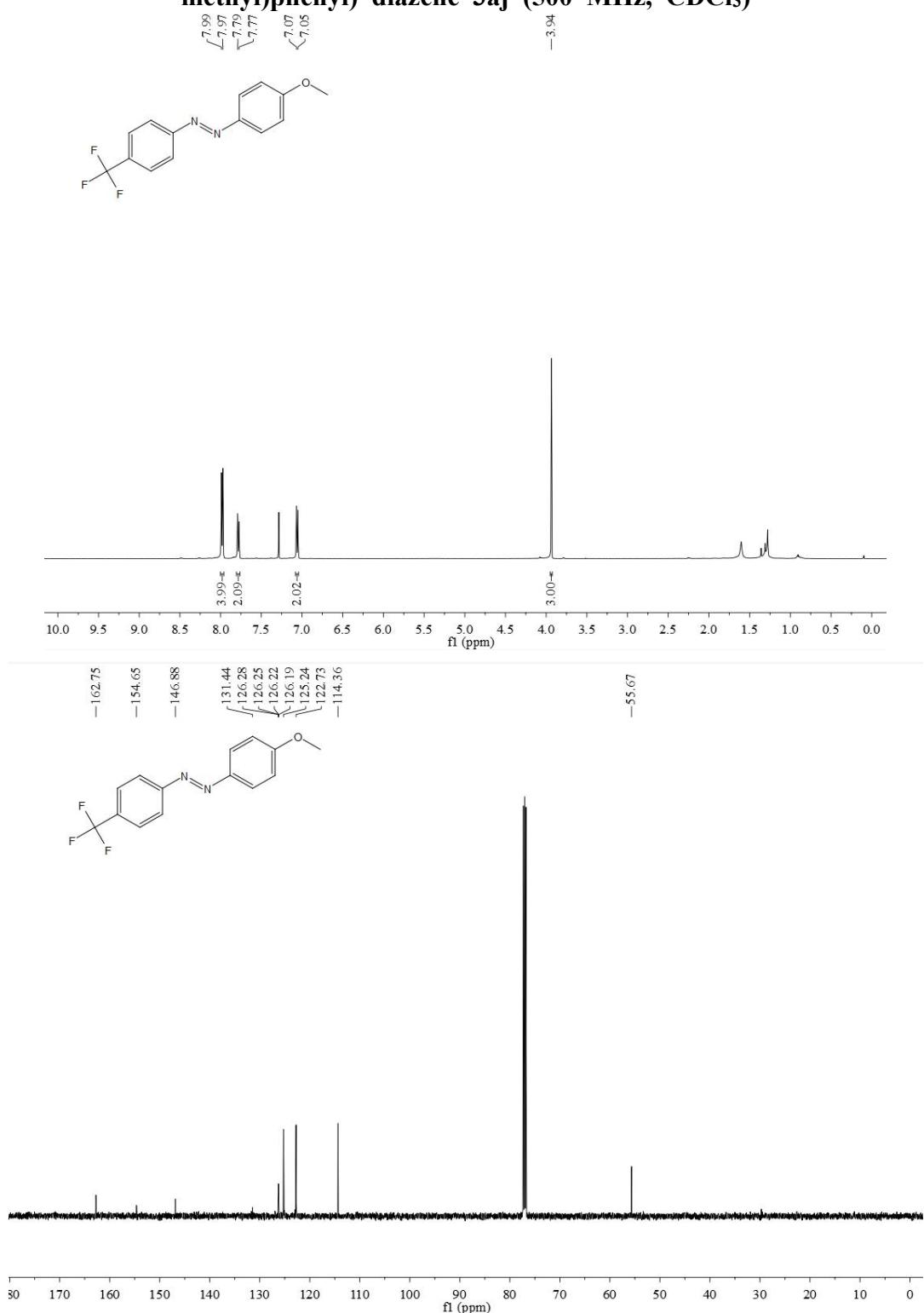
**¹H NMR and ¹³C NMR of 1-(4-Bromophenyl)-2-(4-methoxyphenyl)diazene 3ah
(400 MHz, CDCl₃)**

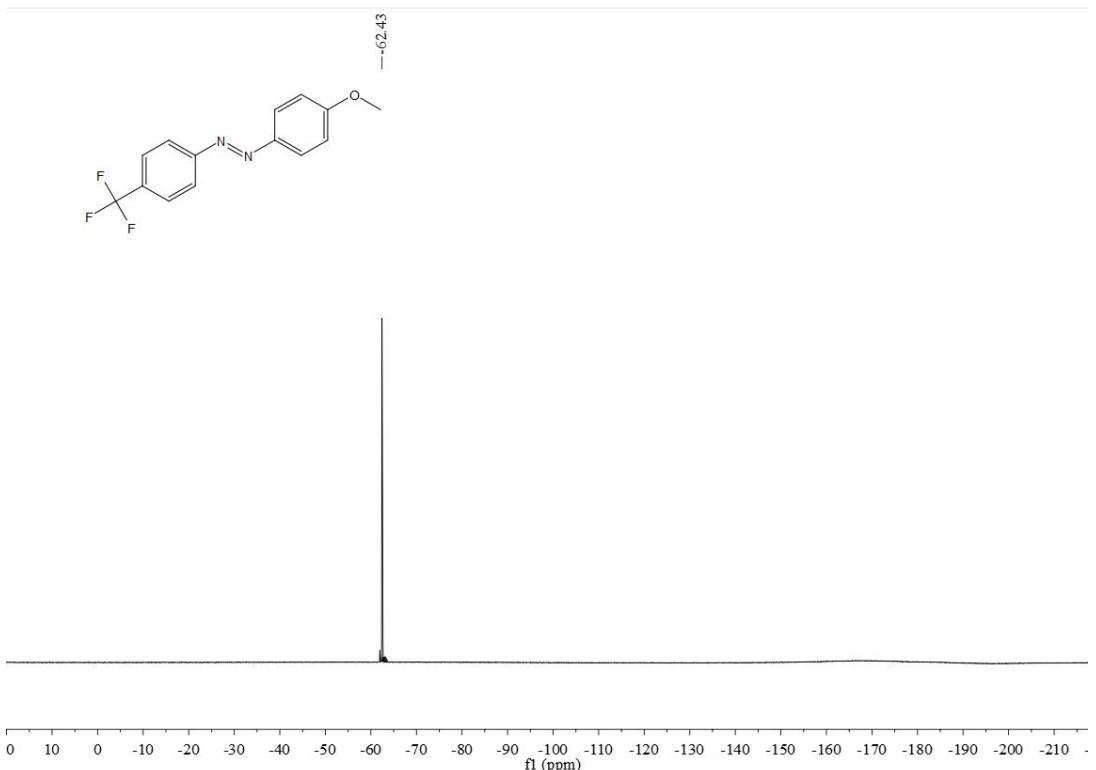


¹H NMR and ¹³C NMR of 1-(4-Iodophenyl)-2-(4-methoxyphenyl)diazene 3ai (500 MHz, CDCl₃)

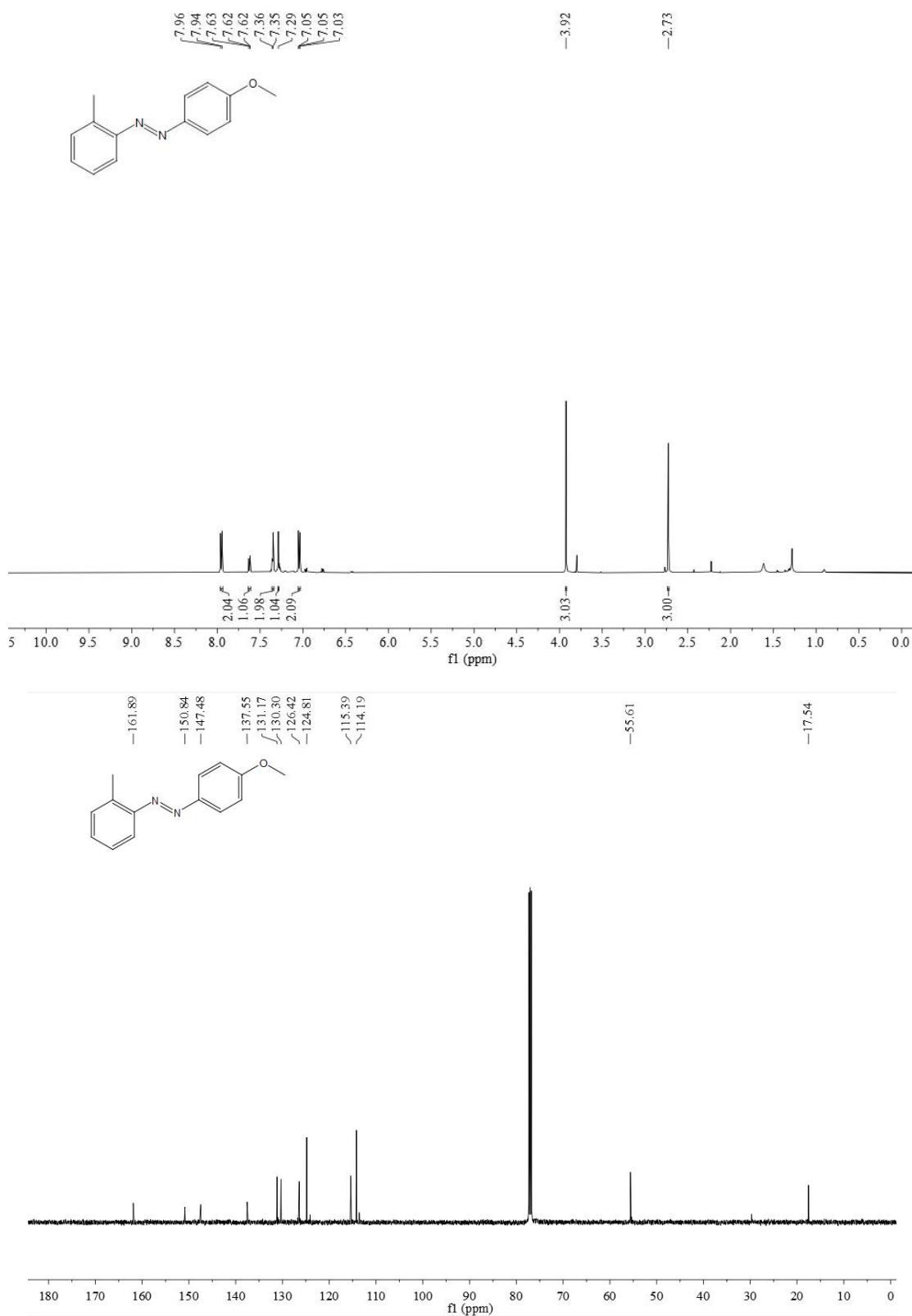


1H NMR and 13C NMR and 19F NMR 1-(4-Methoxyphenyl)-2-(4-(trifluoromethyl)phenyl) diazene 3aj (500 MHz, CDCl₃)

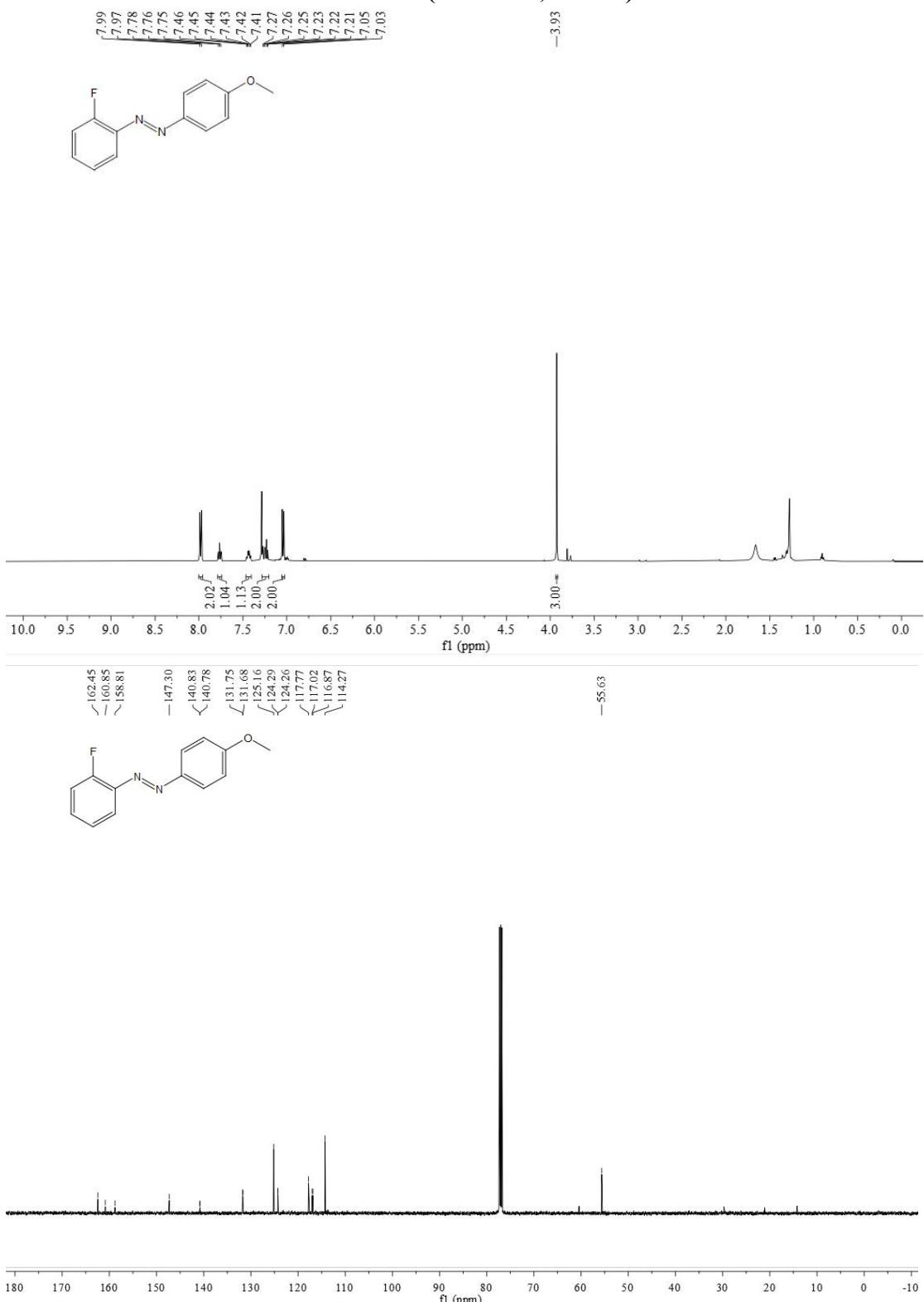


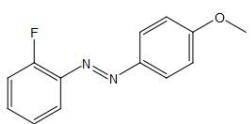


¹H NMR and ¹³C NMR of 1-(4-Methoxyphenyl)-2-(o-tolyl)diazene 3ak (500 MHz, CDCl₃)

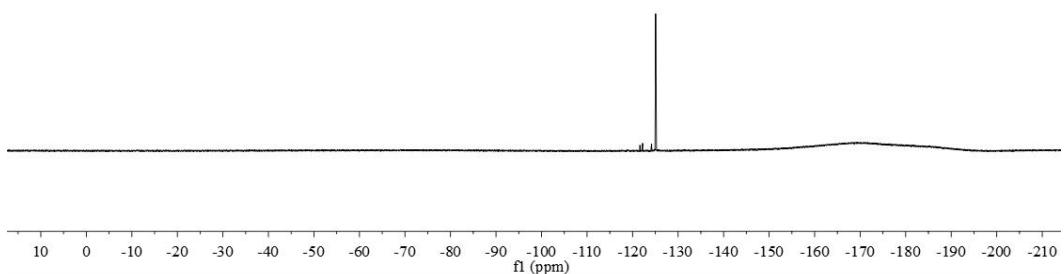


¹H NMR and ¹³C NMR and ¹⁹F NMR of 1-(2-Fluorophenyl)-2-(4-methoxyphenyl) diazene 3al (500 MHz, CDCl₃)

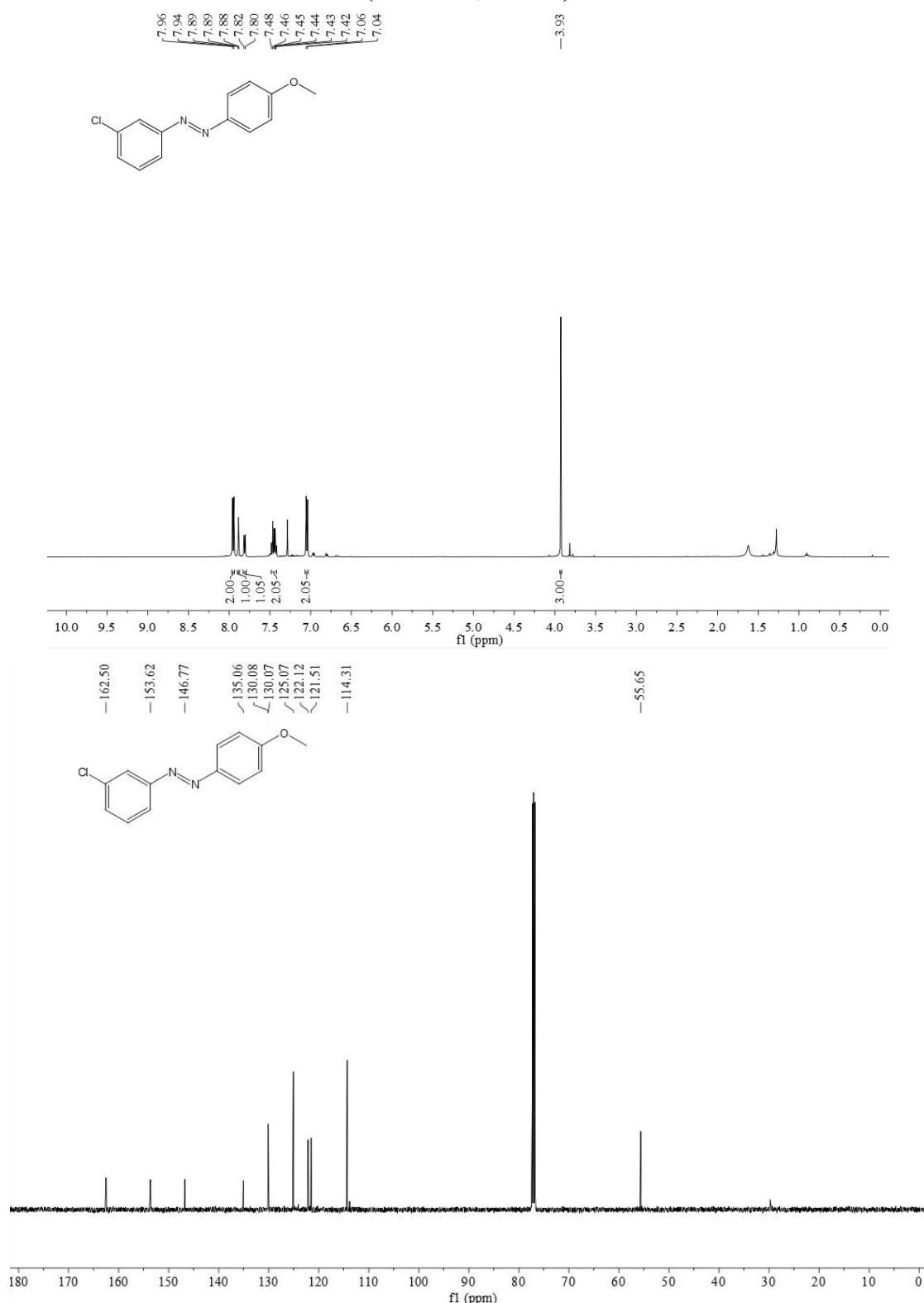




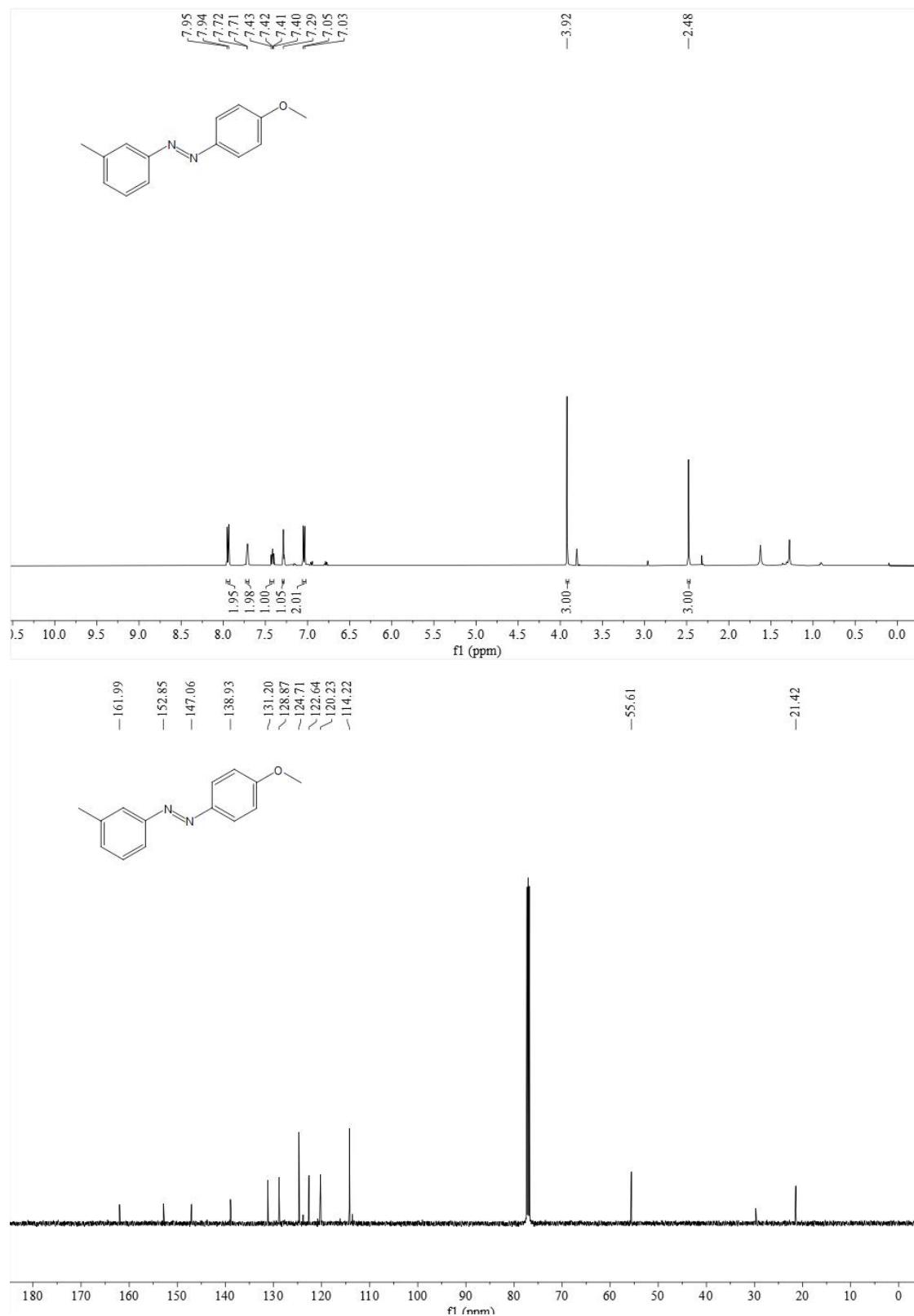
-125.06
-125.08
-125.09
-125.09
-125.10



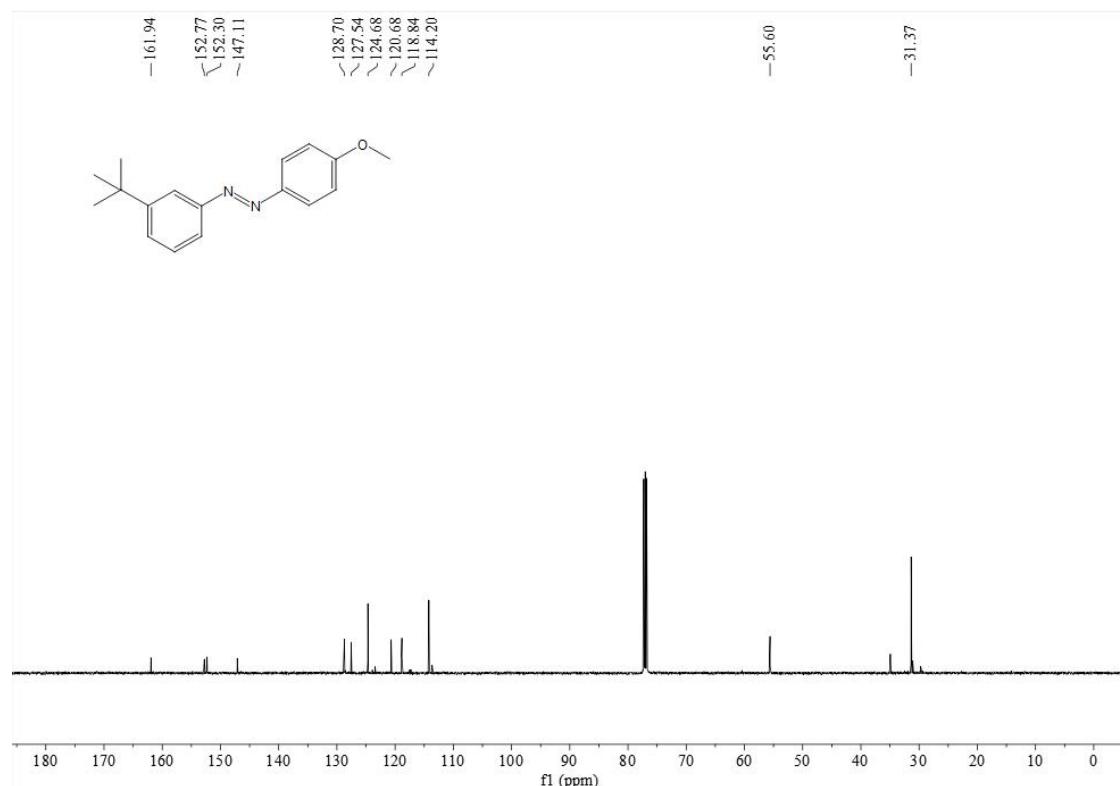
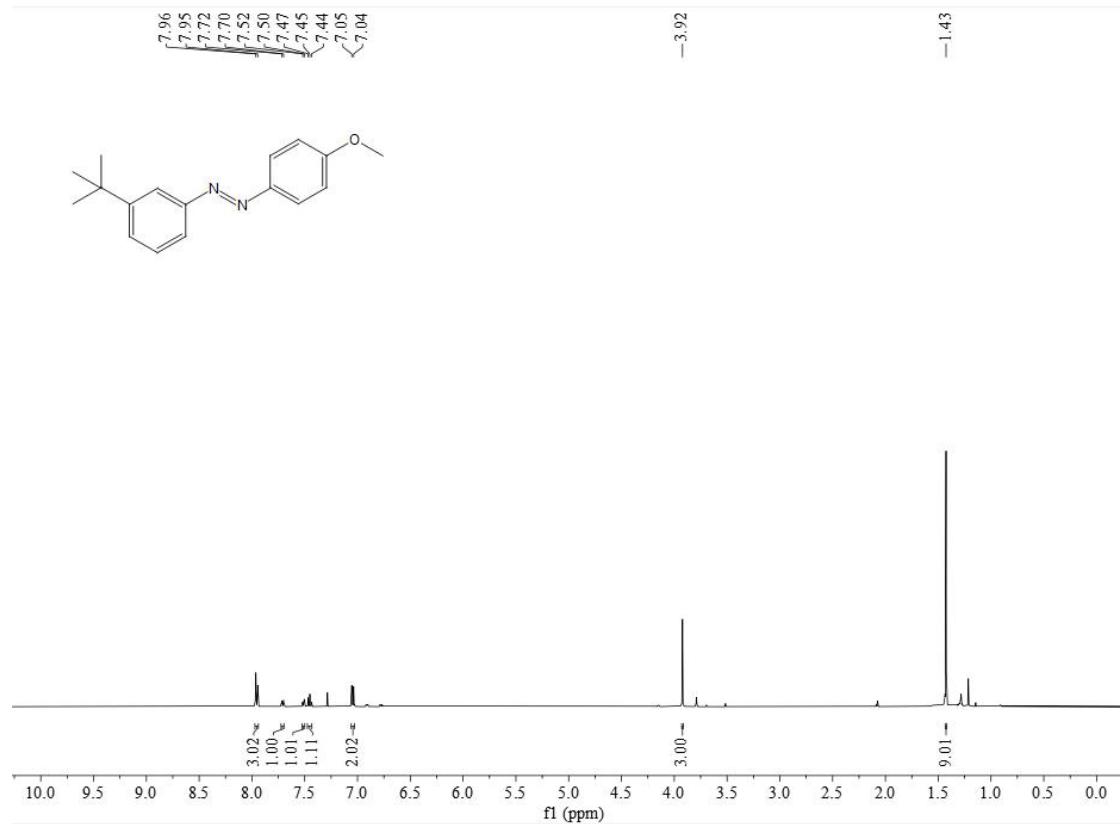
**¹H NMR and ¹³C NMR of 1-(3-Chlorophenyl)-2-(4-methoxyphenyl)diazene 3am
(500 MHz, CDCl₃)**



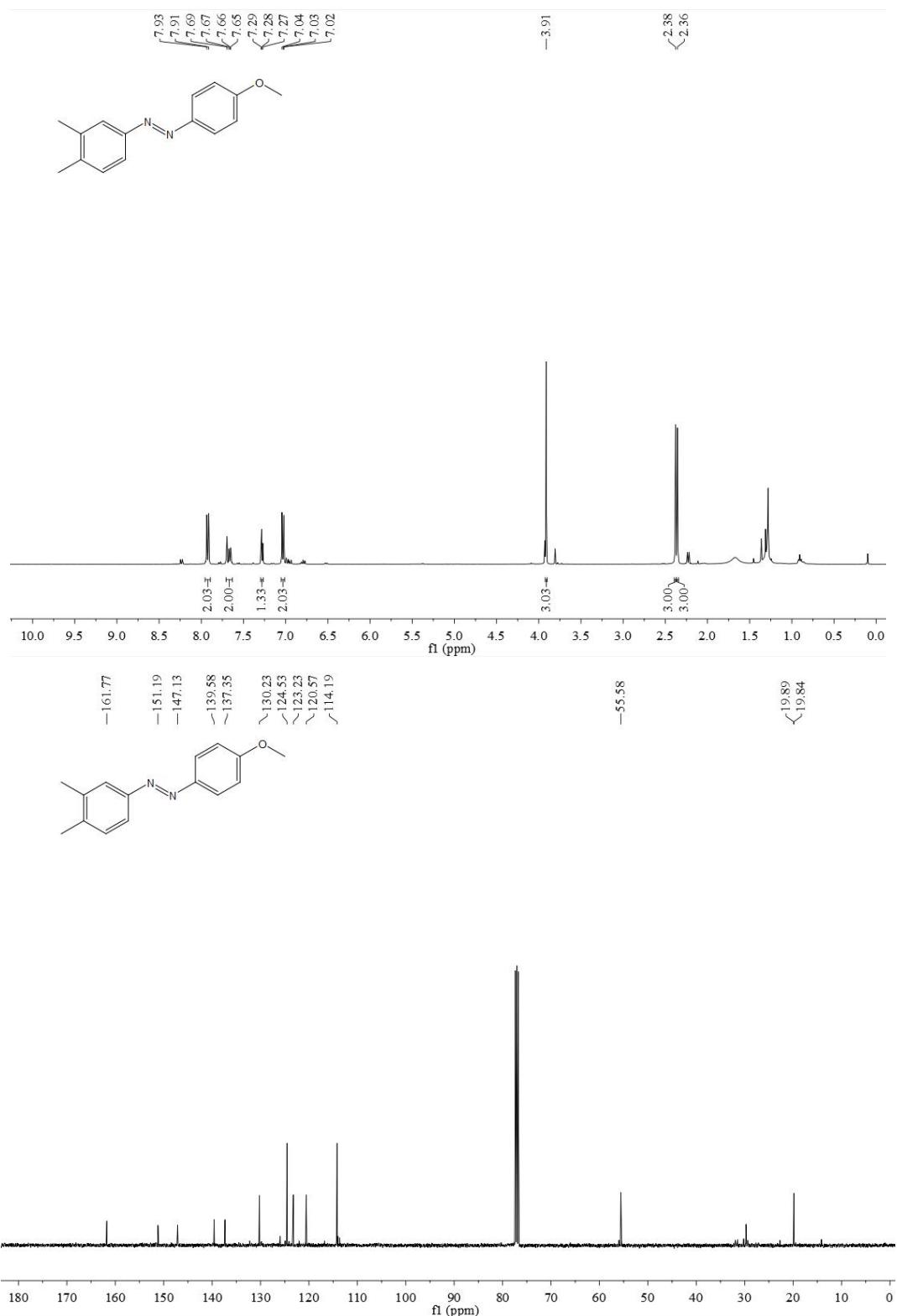
¹H NMR and ¹³C NMR of 1-(4-Methoxyphenyl)-2-(m-tolyl)diazene 3an (500 MHz, CDCl₃)



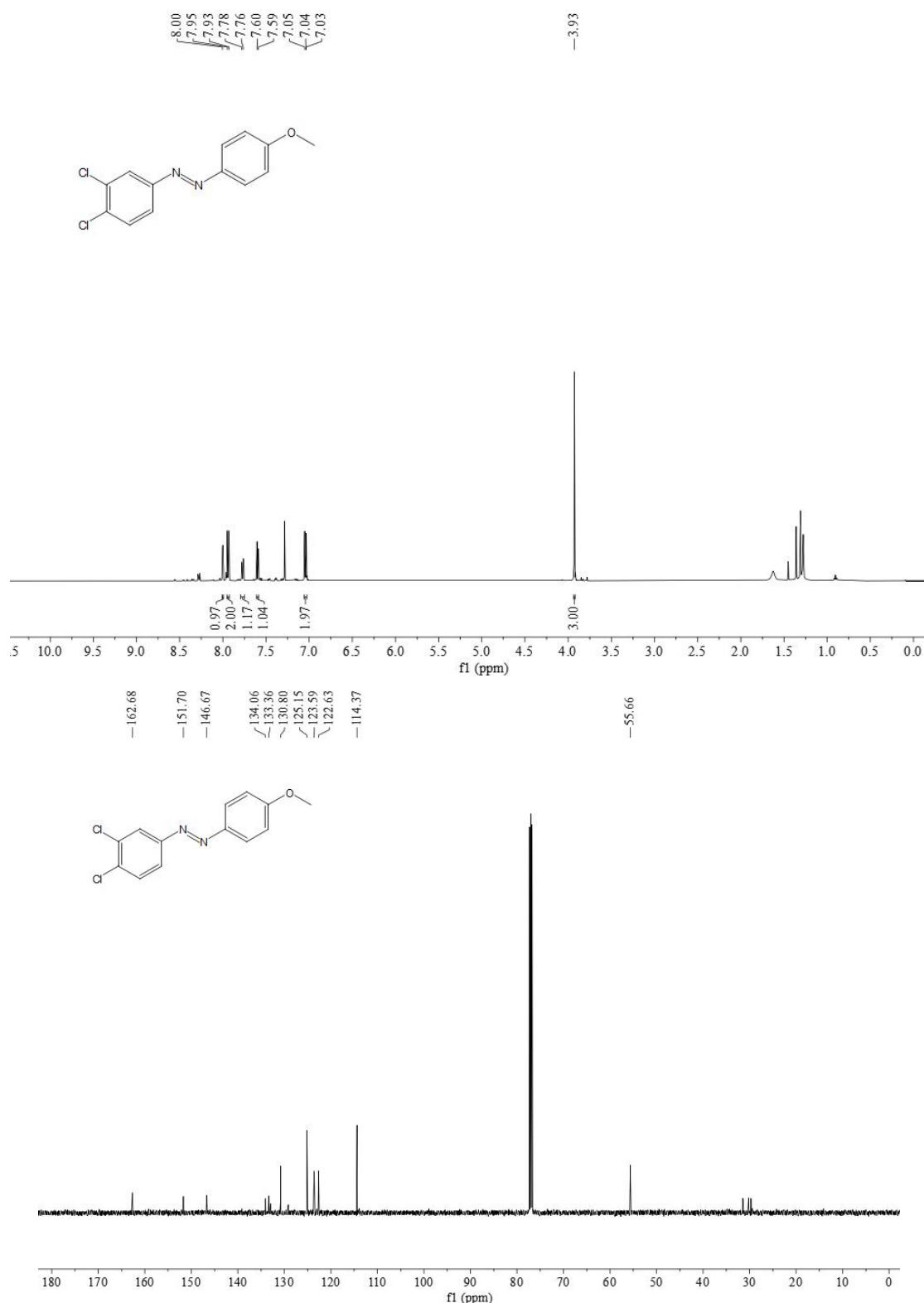
**¹H NMR and ¹³C NMR of 1-(3-(*tert*-Butyl)phenyl)-2-(4-methoxyphenyl)diazene
3ao (500 MHz, CDCl₃)**



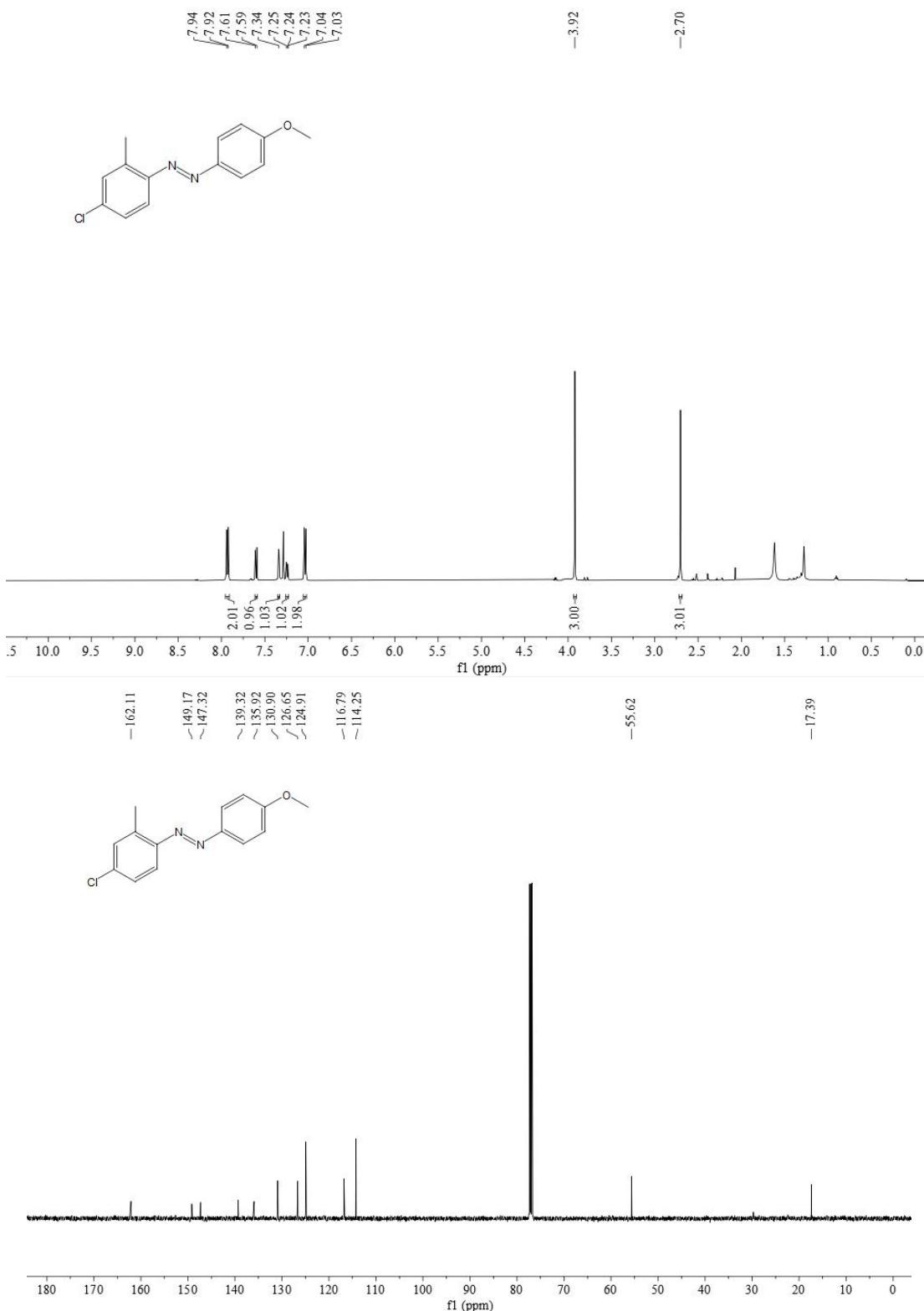
**¹H NMR and ¹³C NMR of 1-(3,4-Dimethylphenyl)-2-(4-methoxyphenyl)diazene
3ap (500 MHz, CDCl₃)**



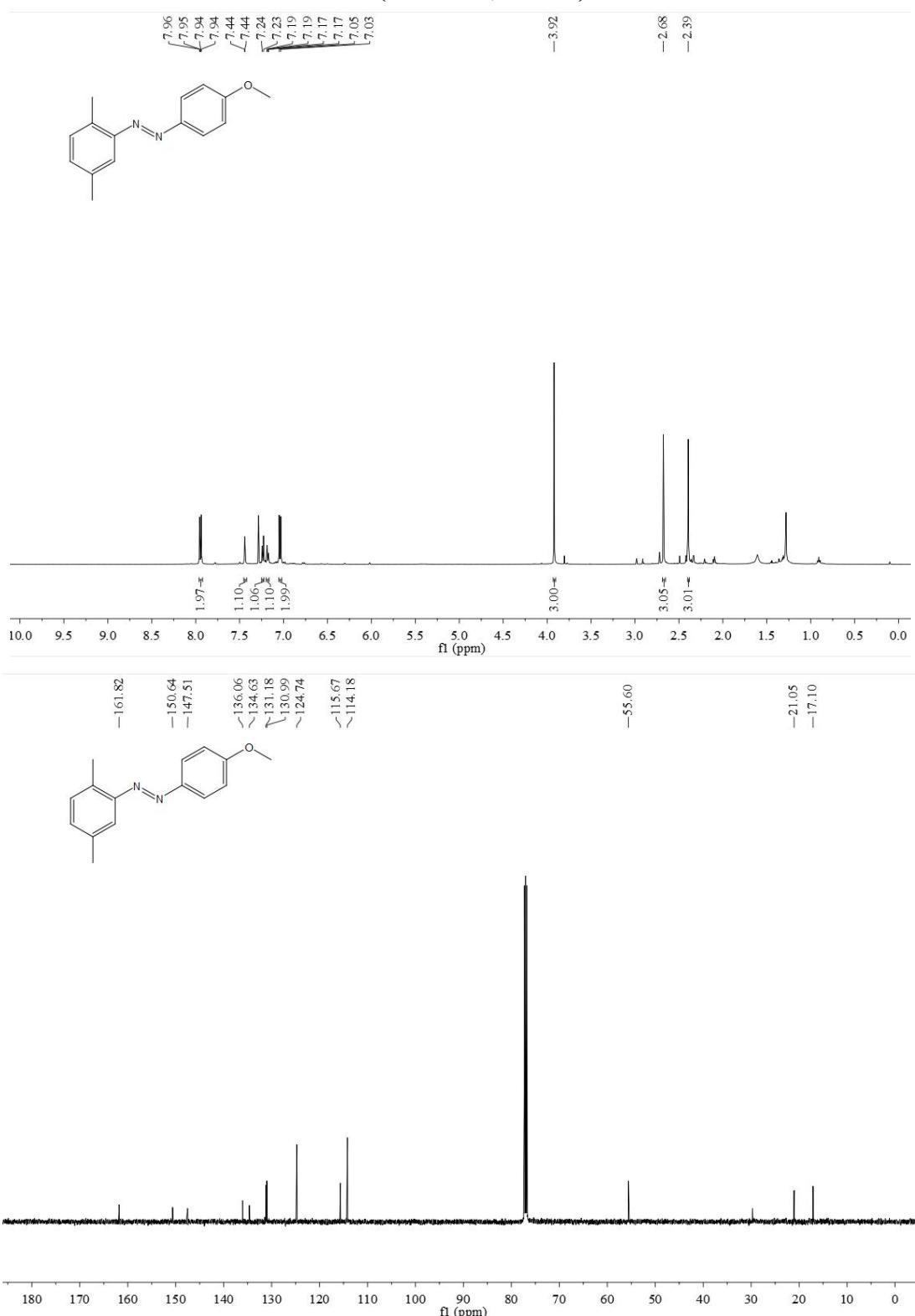
**¹H NMR and ¹³C NMR of 1-(3,4-Dichlorophenyl)-2-(4-methoxyphenyl)diazene
3aq (500 MHz, CDCl₃)**



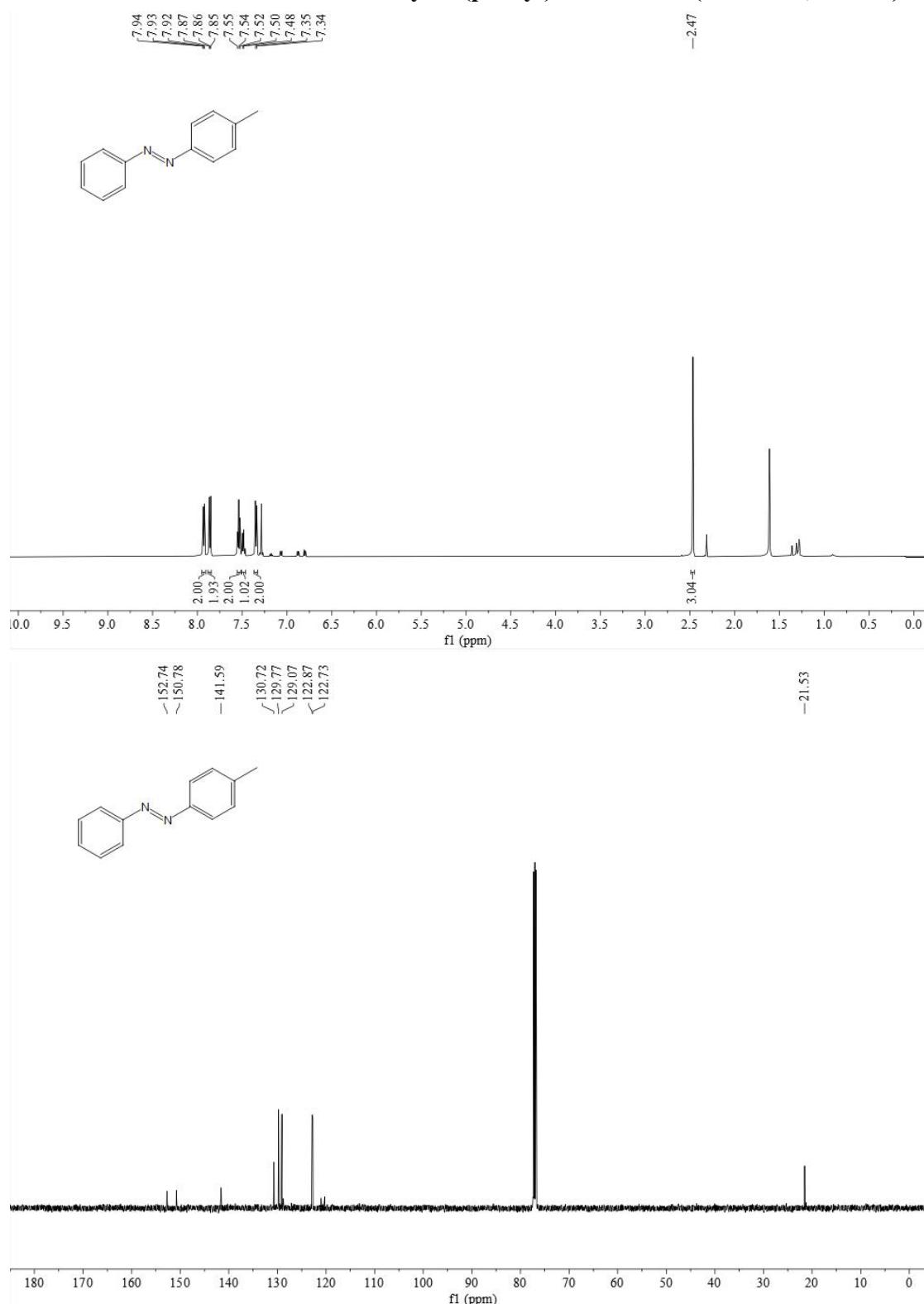
¹H NMR and ¹³C NMR of 1-(4-Chloro-2-methylphenyl)-2-(4-methoxyphenyl) diazene 3ar (500 MHz, CDCl₃)



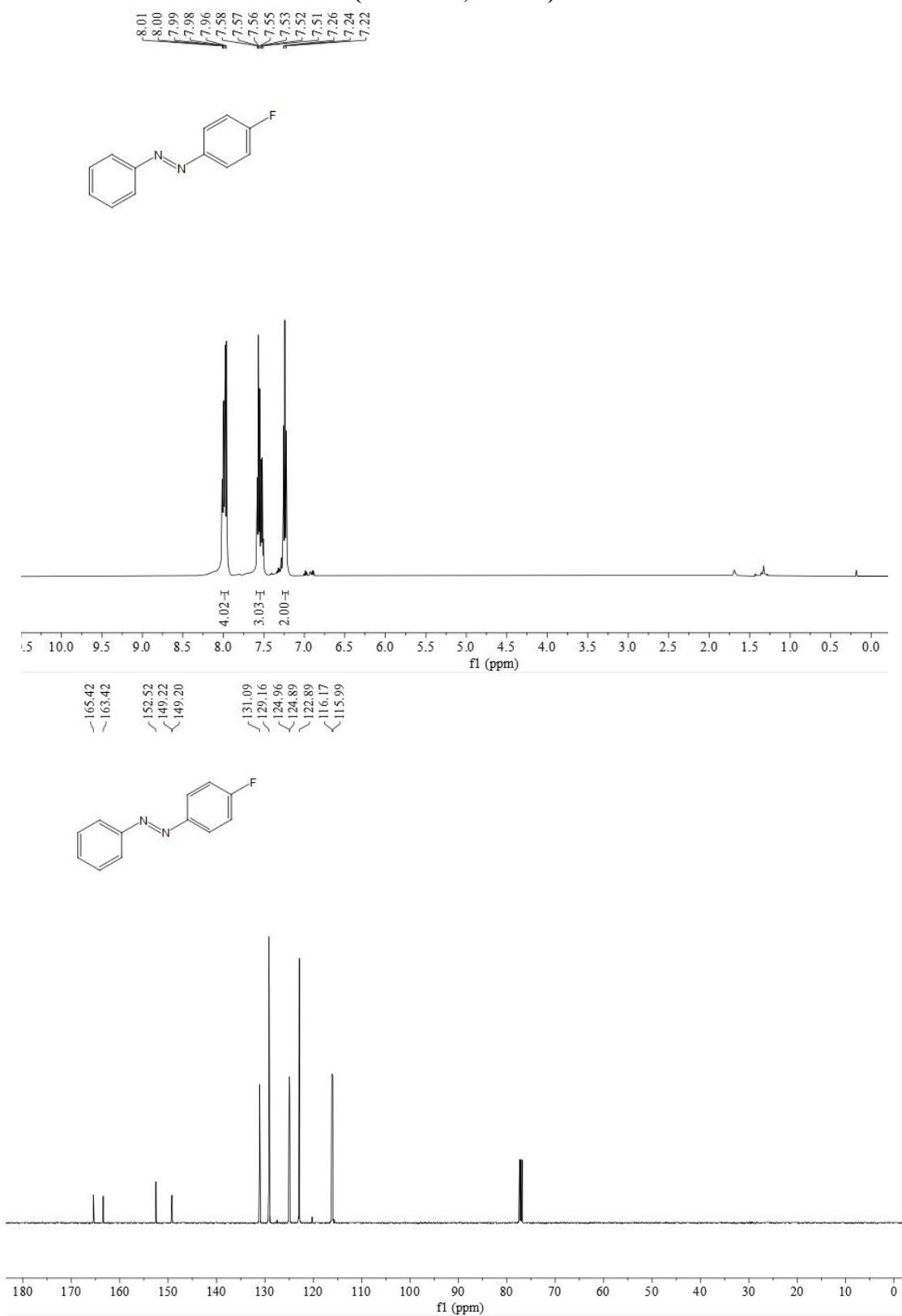
**¹H NMR and ¹³C NMR of 1-(2,5-Dimethylphenyl)-2-(4-methoxyphenyl)diazene
3as (500 MHz, CDCl₃)**

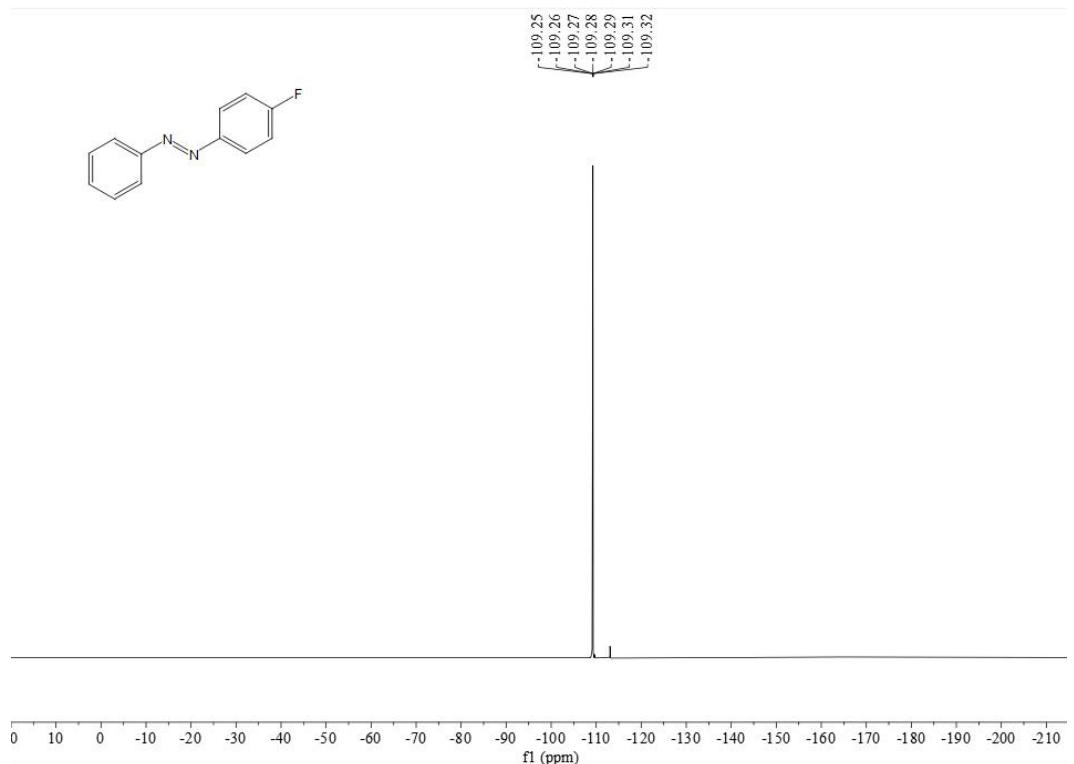


¹H NMR and ¹³C NMR of 1-Phenyl-2-(p-tolyl)diazene 3ba (500 MHz, CDCl₃)

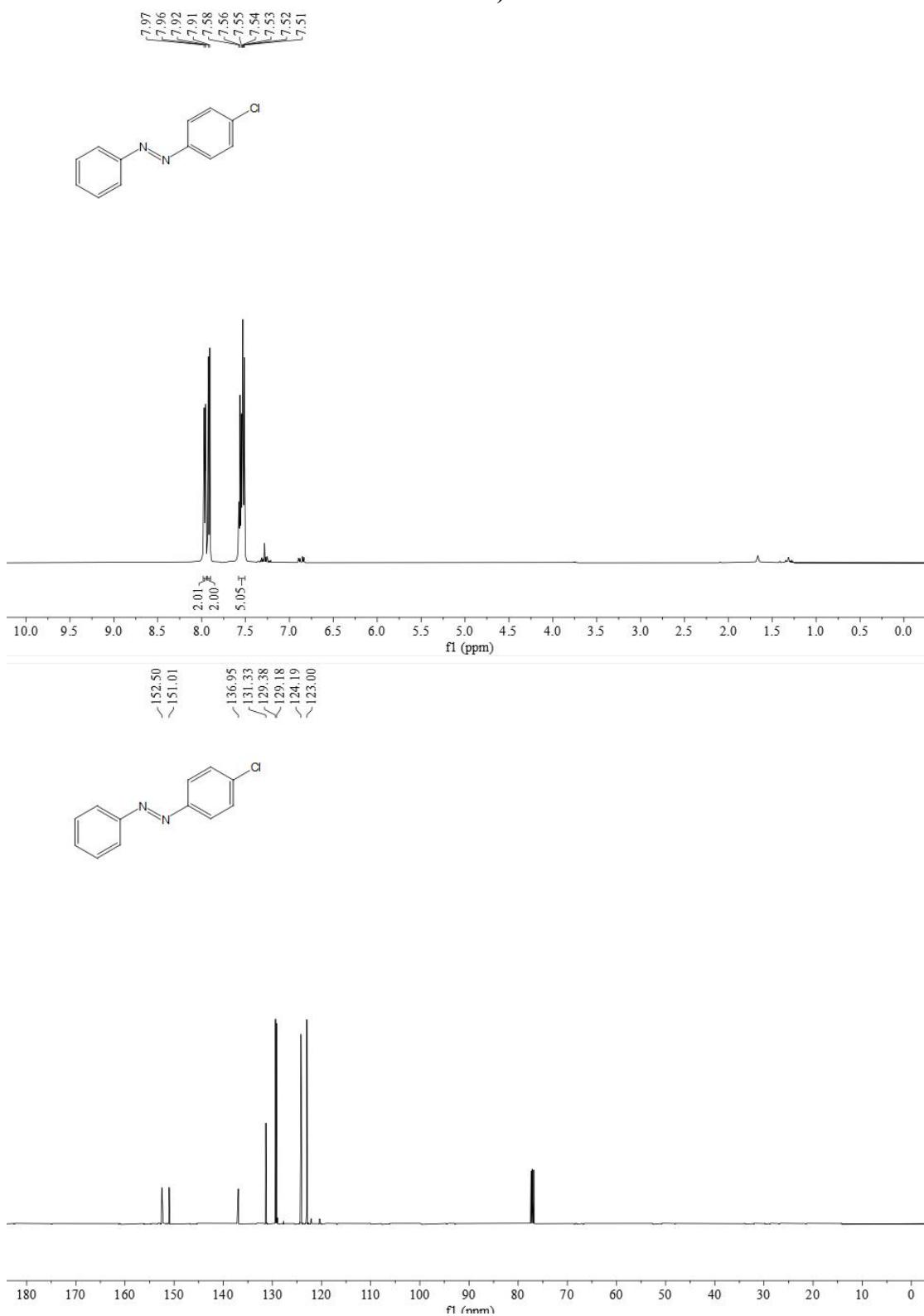


**¹H NMR, ¹³C NMR and ¹⁹F NMR of 1-(4-Fluorophenyl)-2-phenyldiazene 3bb
(500 MHz, CDCl₃)**

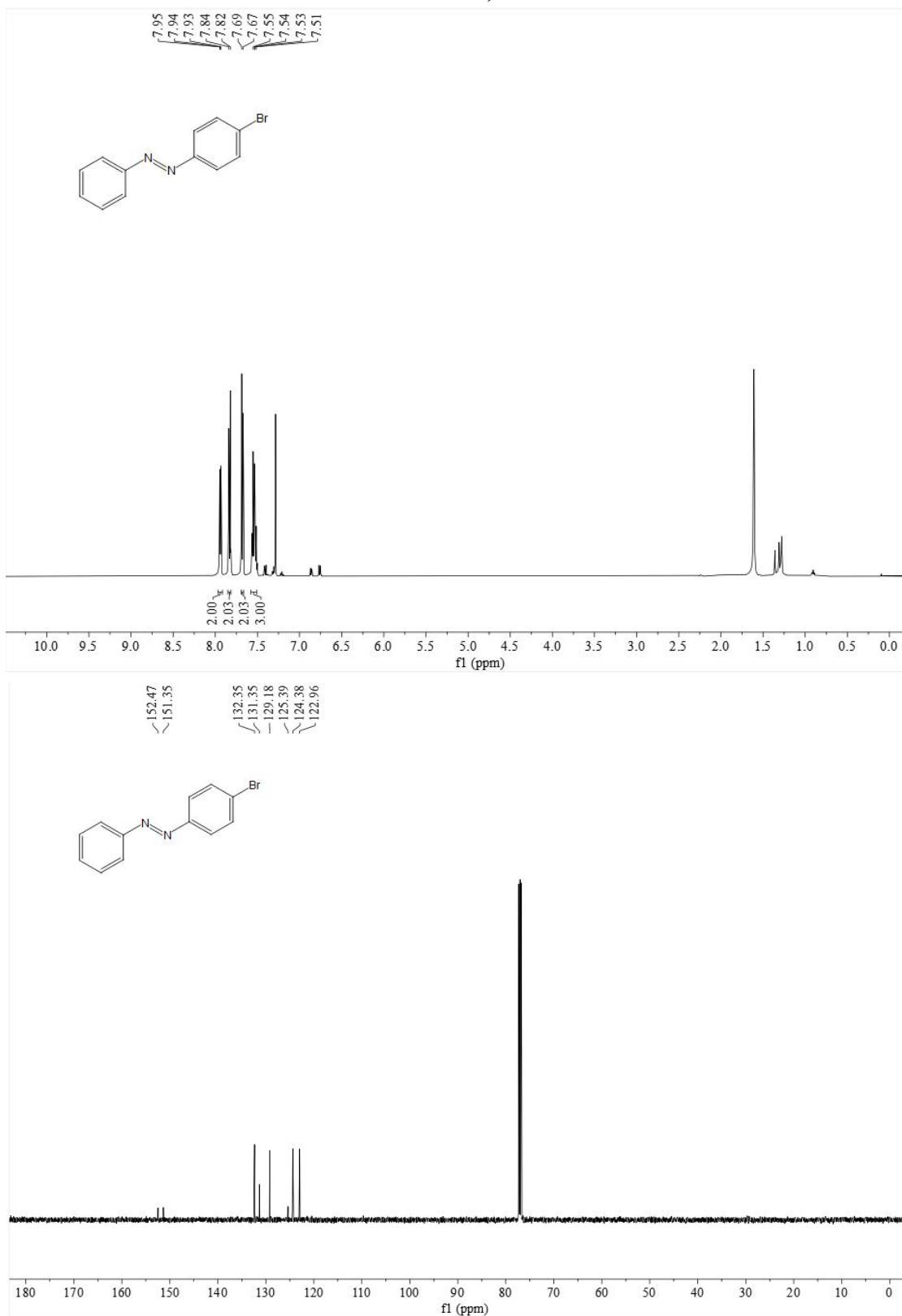




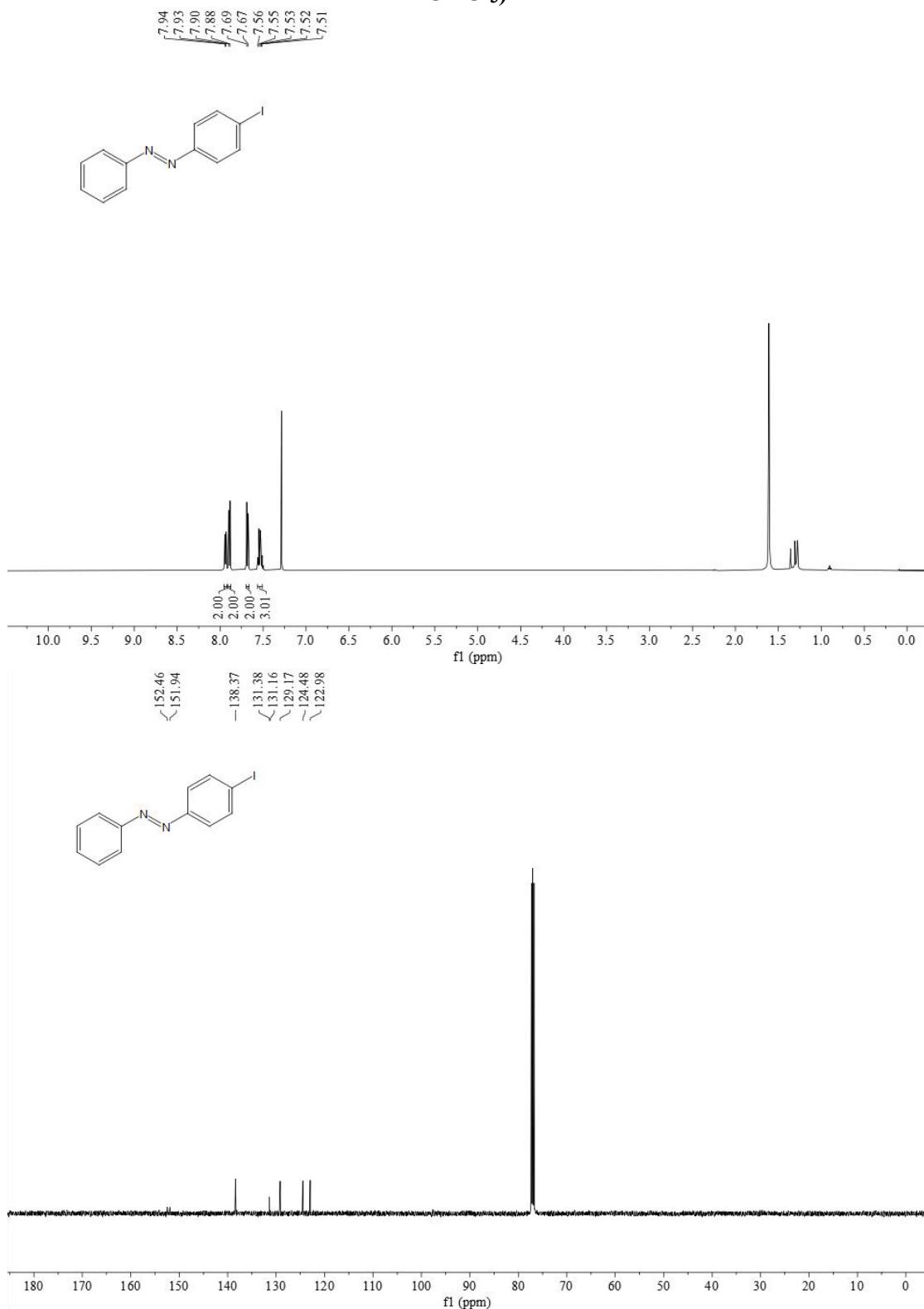
¹H NMR and ¹³C NMR of 1-(4-Chlorophenyl)-2-phenyldiazene 3bc (500 MHz, CDCl₃)



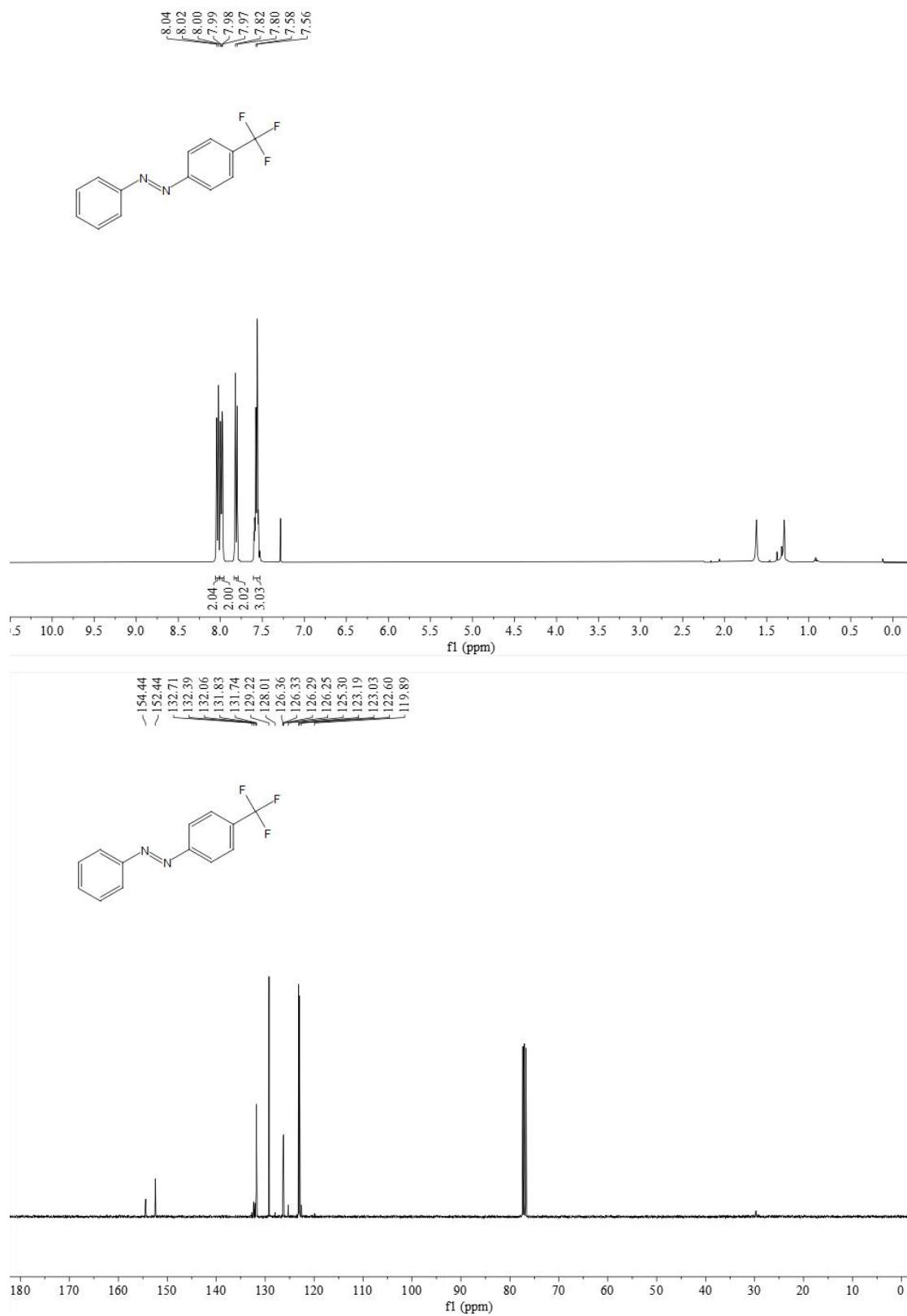
¹H NMR and ¹³C NMR of 1-(4-Bromophenyl)-2-phenyldiazene 3bd (500 MHz, CDCl₃)

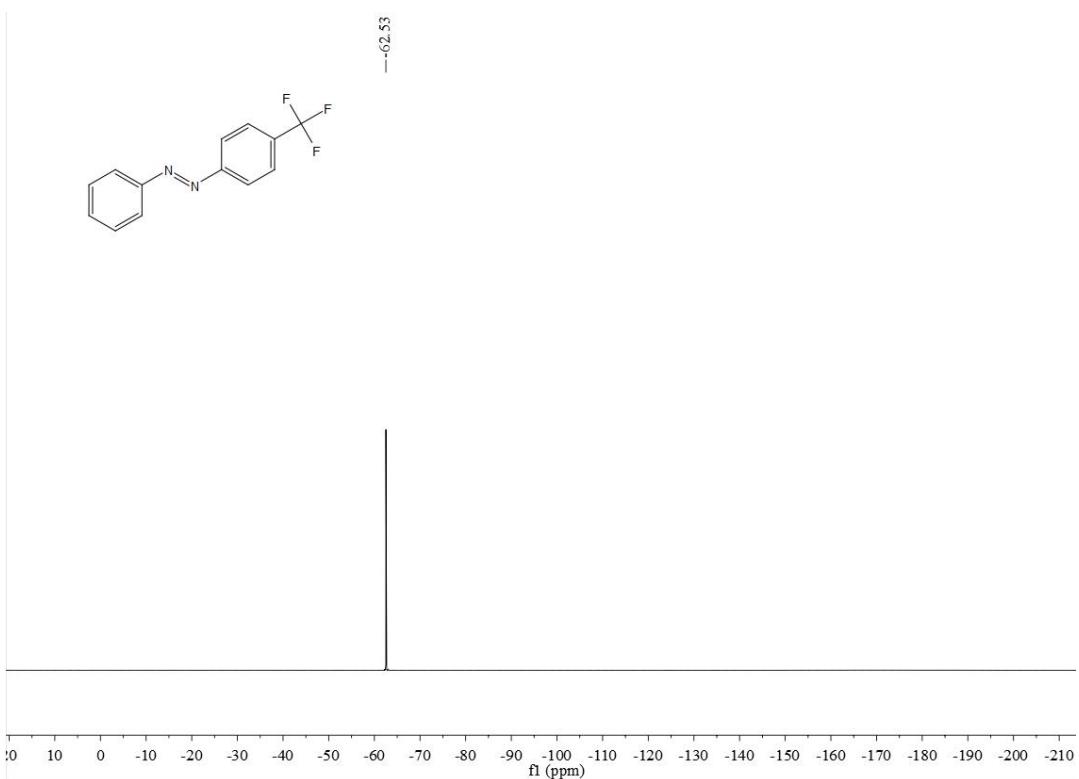


^1H NMR and ^{13}C NMR of 1-(4-Iodophenyl)-2-phenyldiazene 3be (500 MHz, CDCl_3)

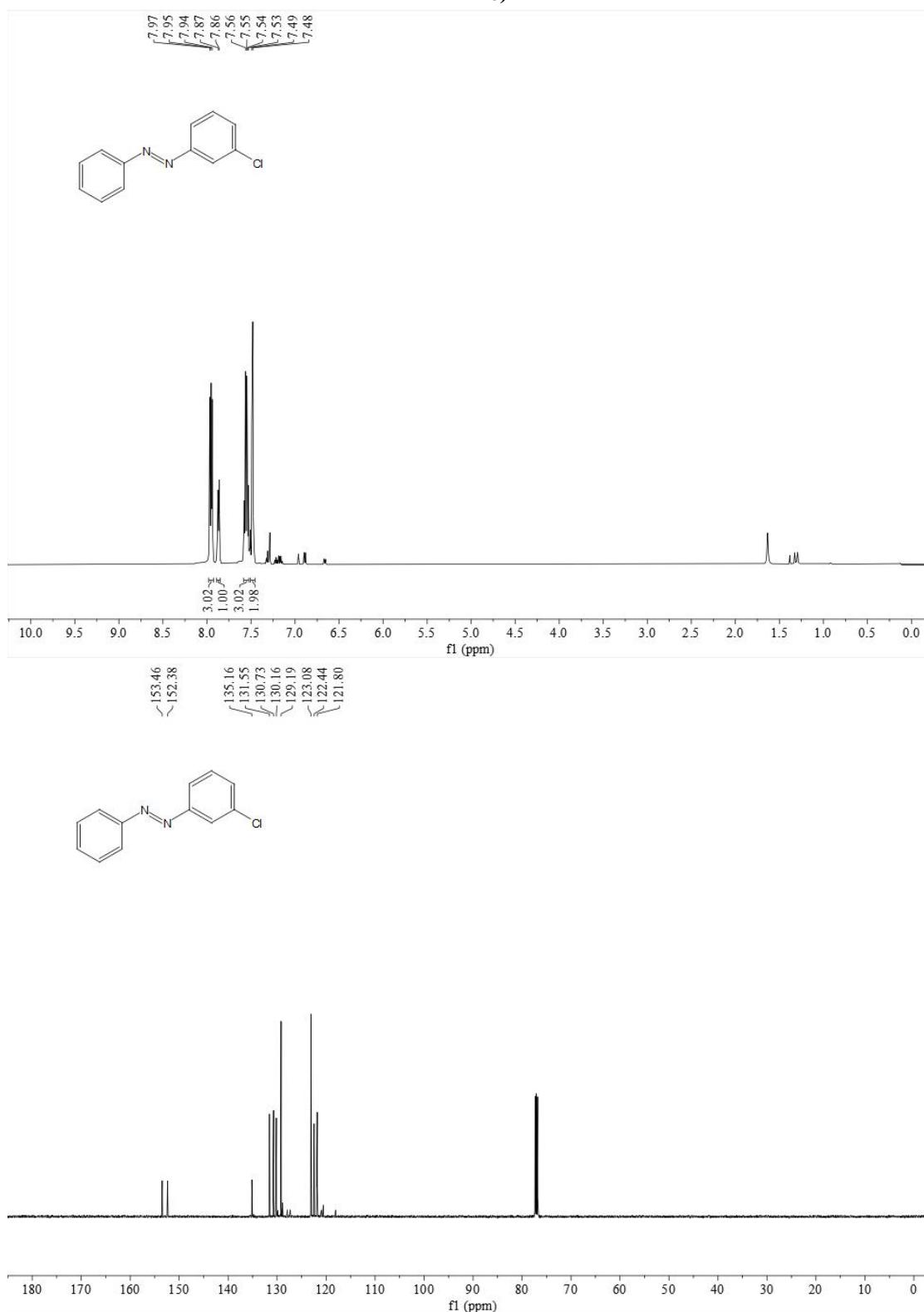


¹H NMR, ¹³C NMR and ¹⁹F NMR of 1-Phenyl-2-(4-(trifluoromethyl)phenyl)diazene 3bf (500 MHz, CDCl₃)

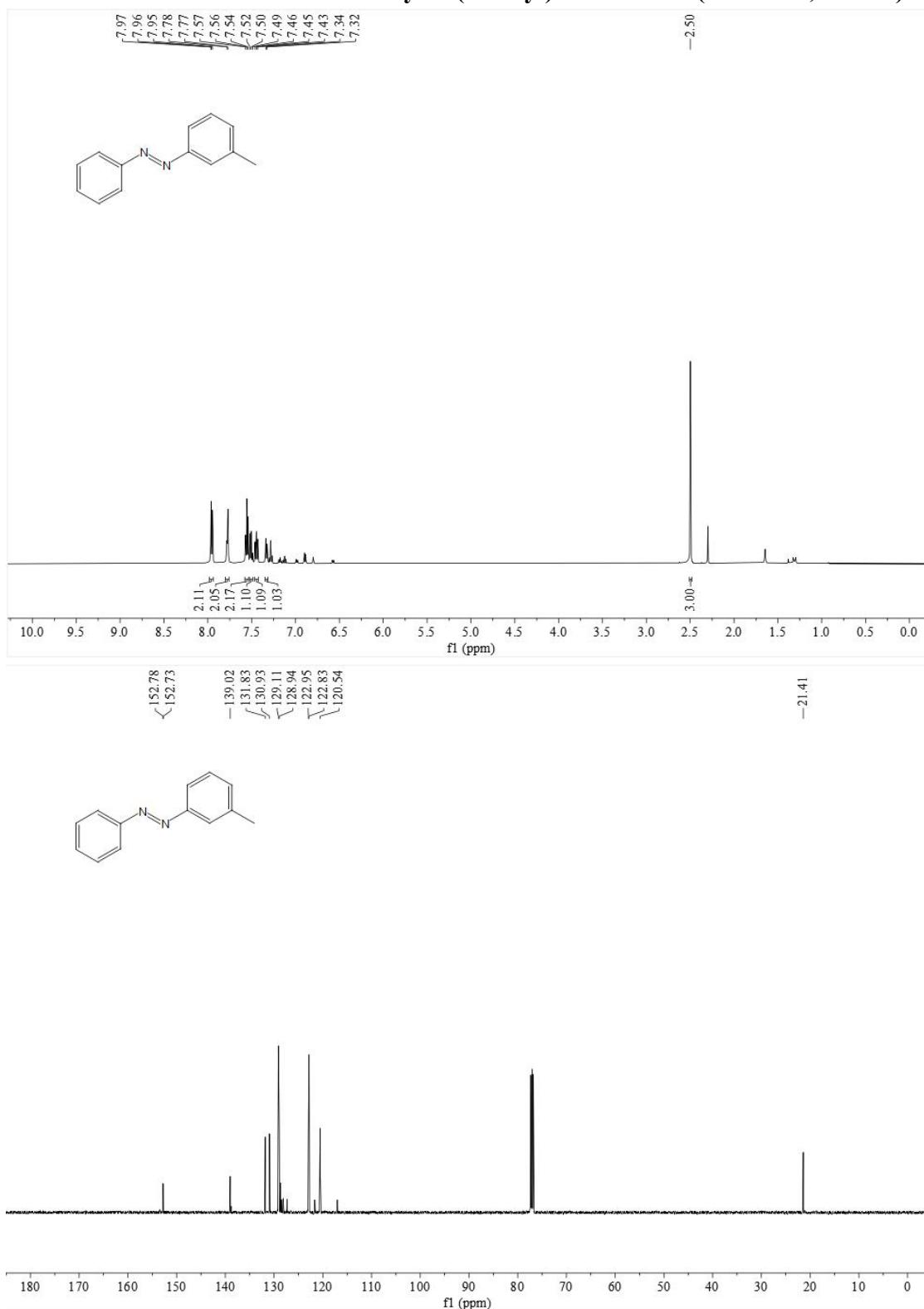




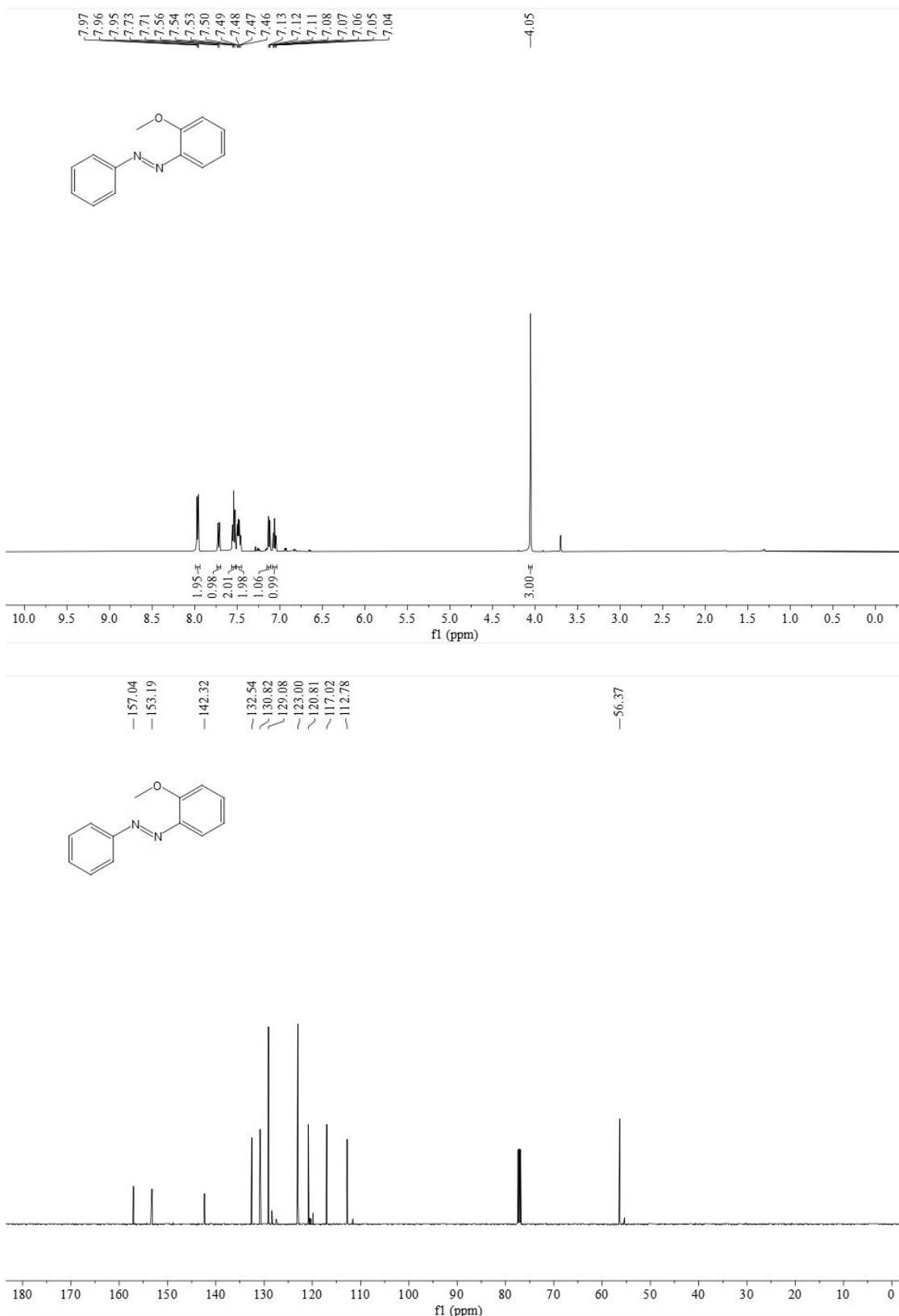
^1H NMR and ^{13}C NMR of 1-(3-Chlorophenyl)-2-phenyldiazene 3bg (500 MHz, CDCl_3)



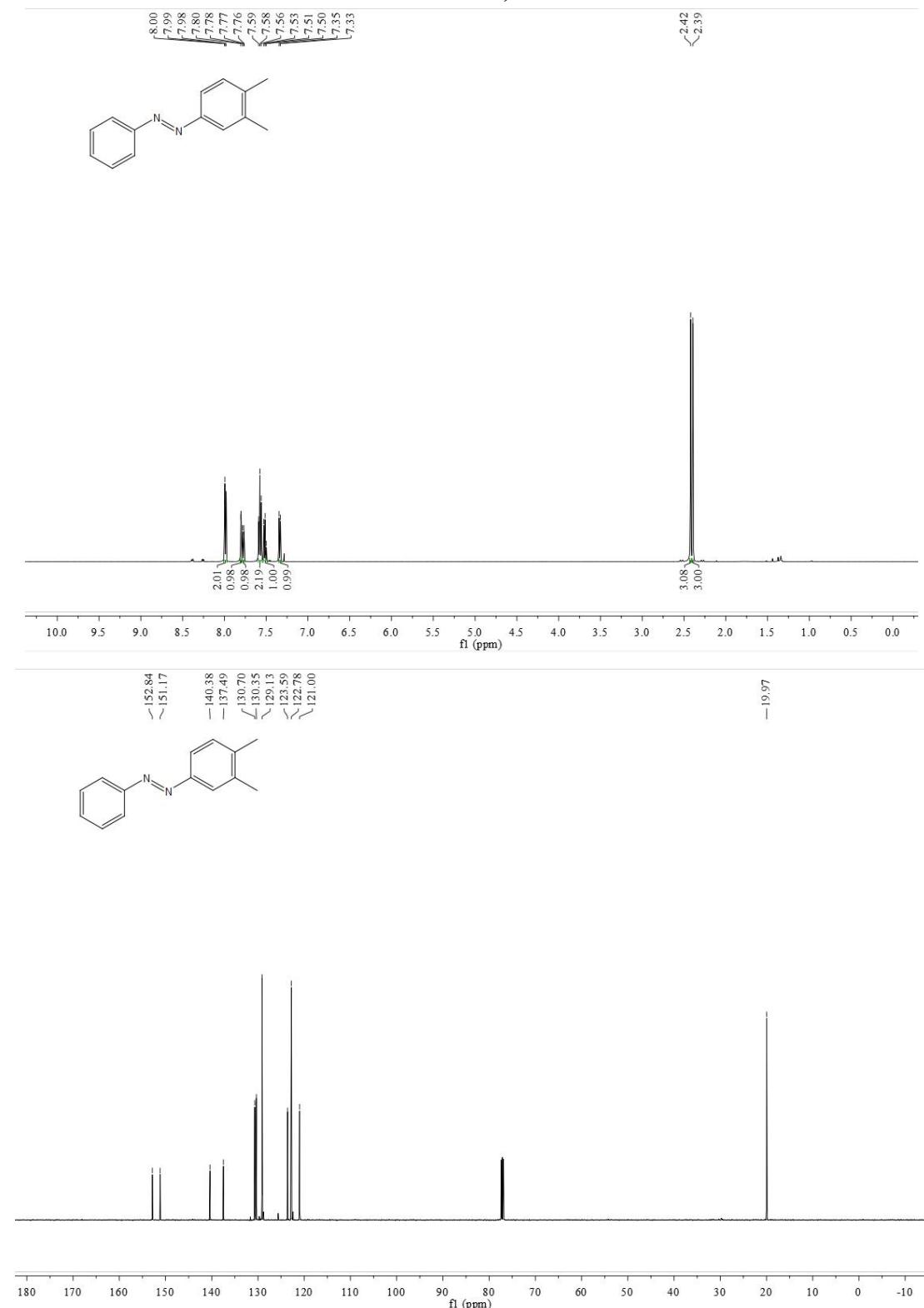
¹H NMR and ¹³C NMR of 1-Phenyl-2-(m-tolyl)diazene 3bh (500 MHz, CDCl₃)



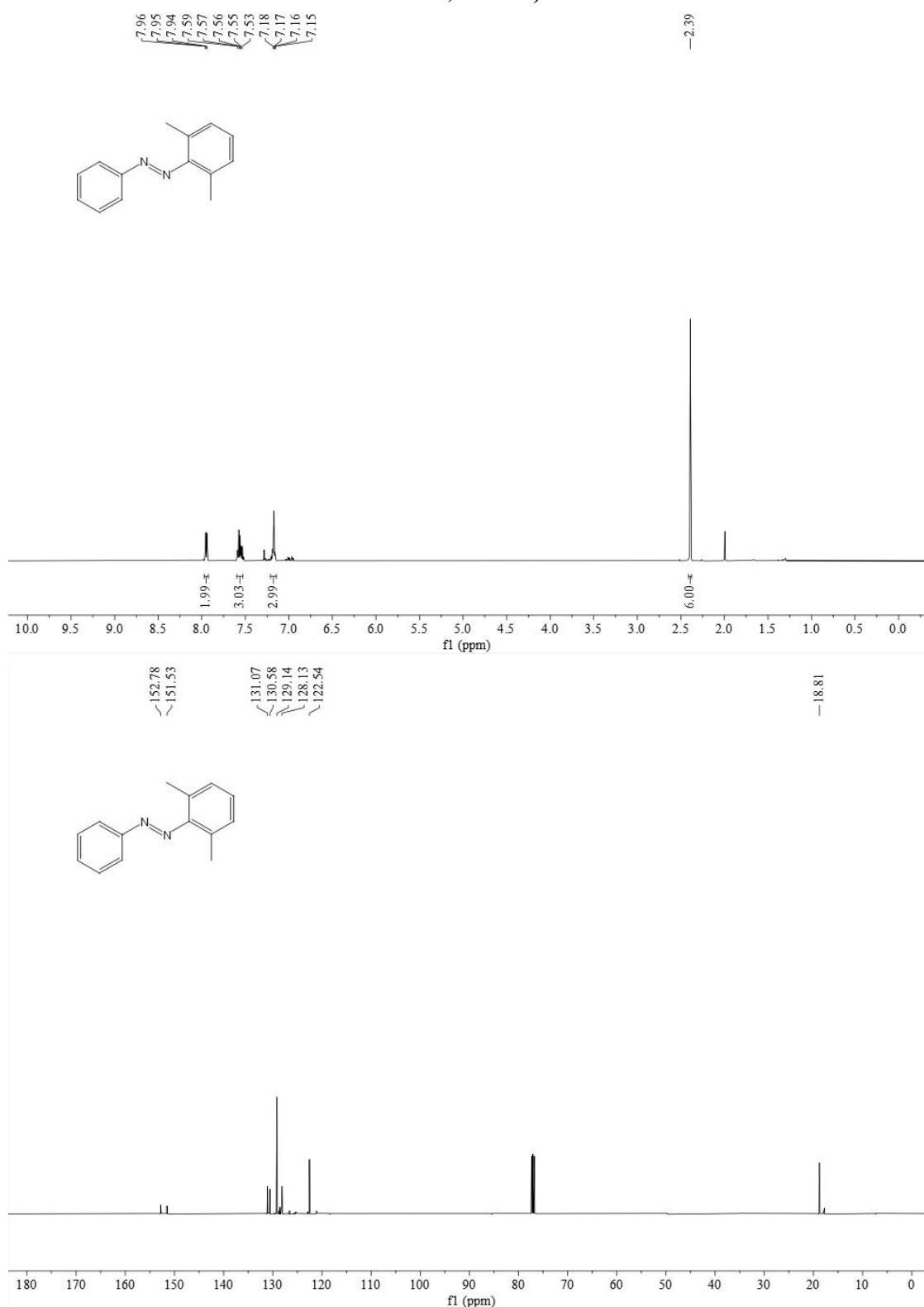
¹H NMR and ¹³C NMR of 1-(2-Methoxyphenyl)-2-phenyldiazene 3bi (500 MHz, CDCl₃)



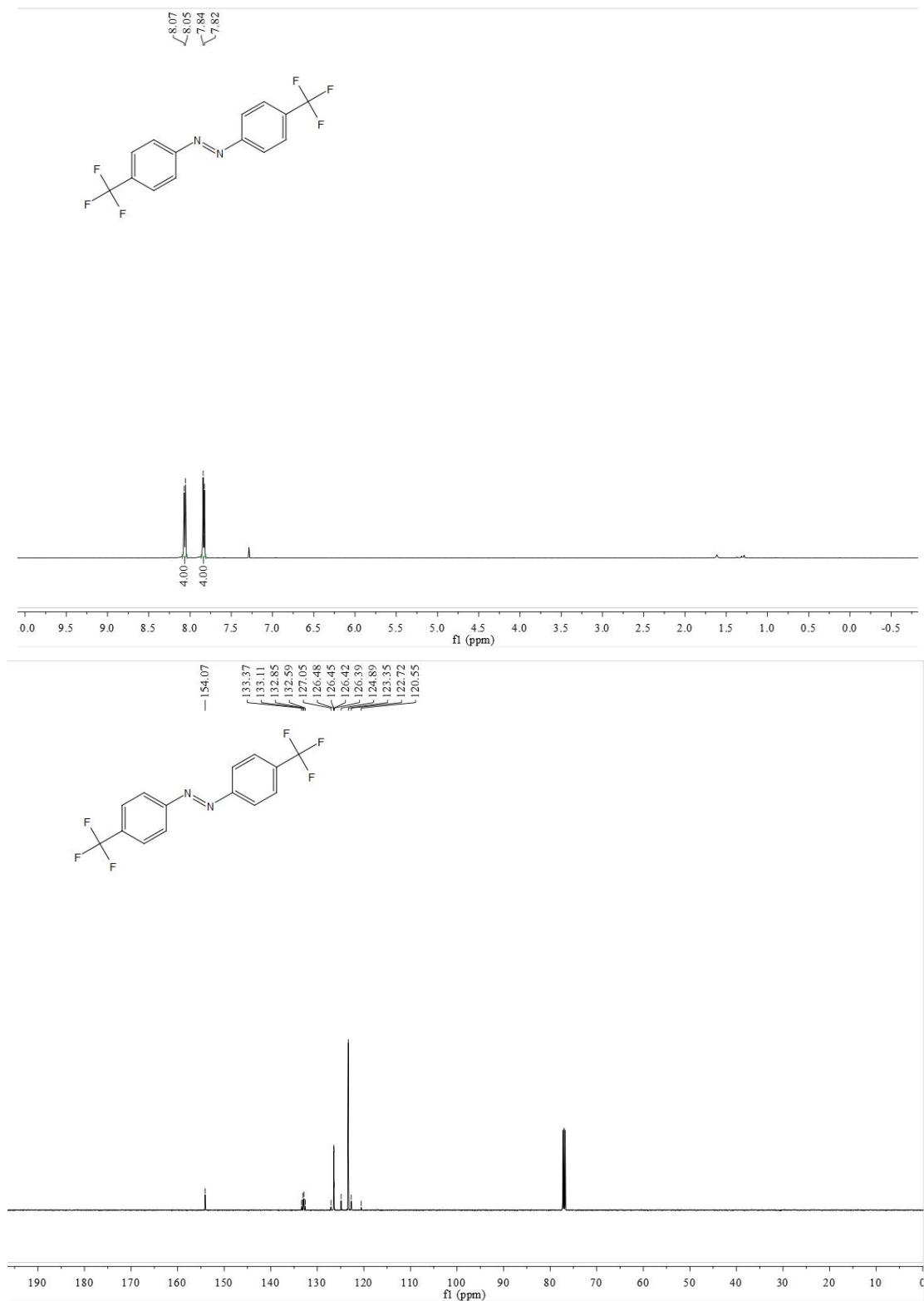
¹H NMR and ¹³C NMR of 1-(3,4-Dimethylphenyl)-2-phenyldiazene 3bj (500 MHz, CDCl₃)



¹H NMR and ¹³C NMR of 1-(2,6-Dimethylphenyl)-2-phenyldiazene 3bk (500 MHz, CDCl₃)

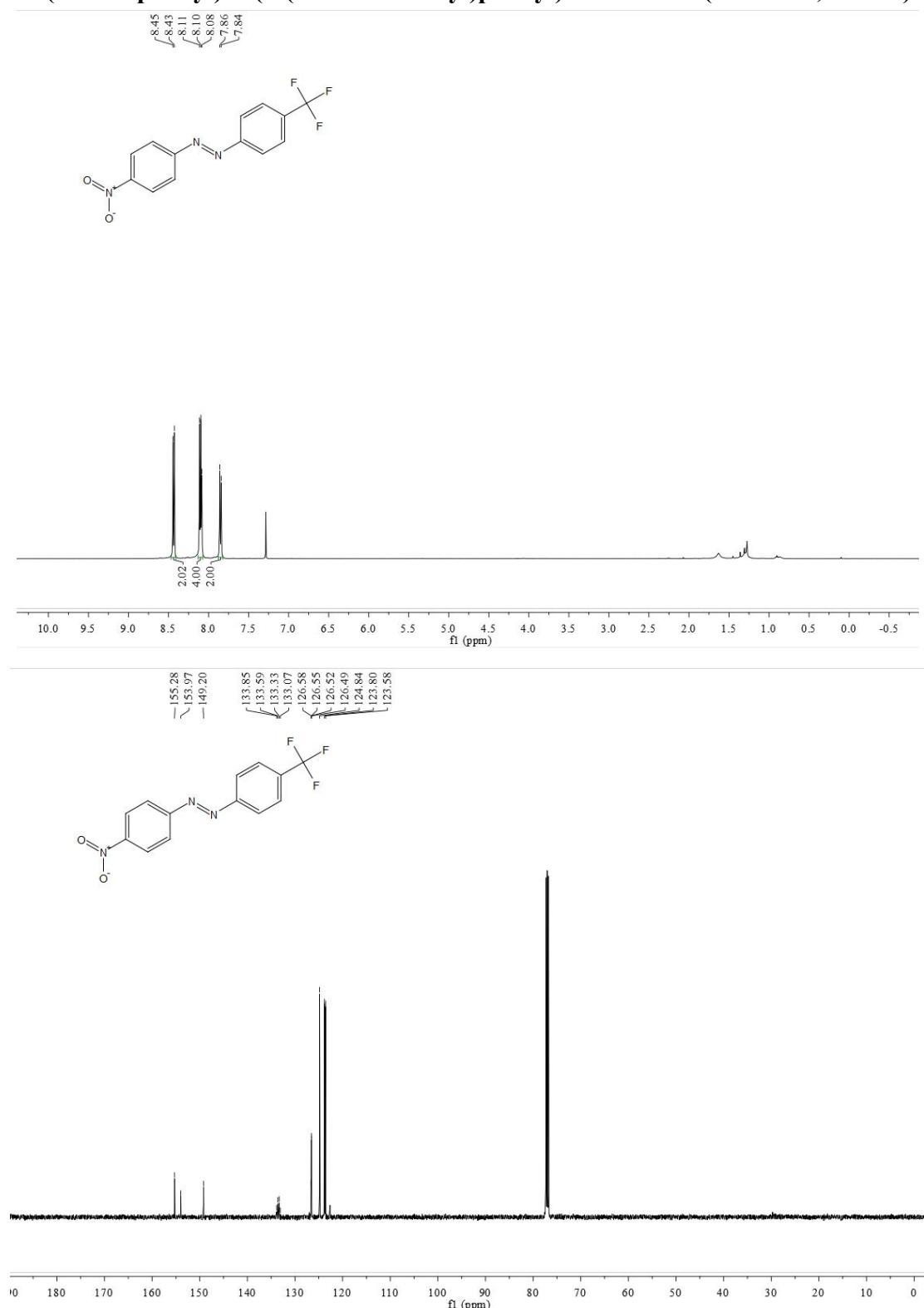


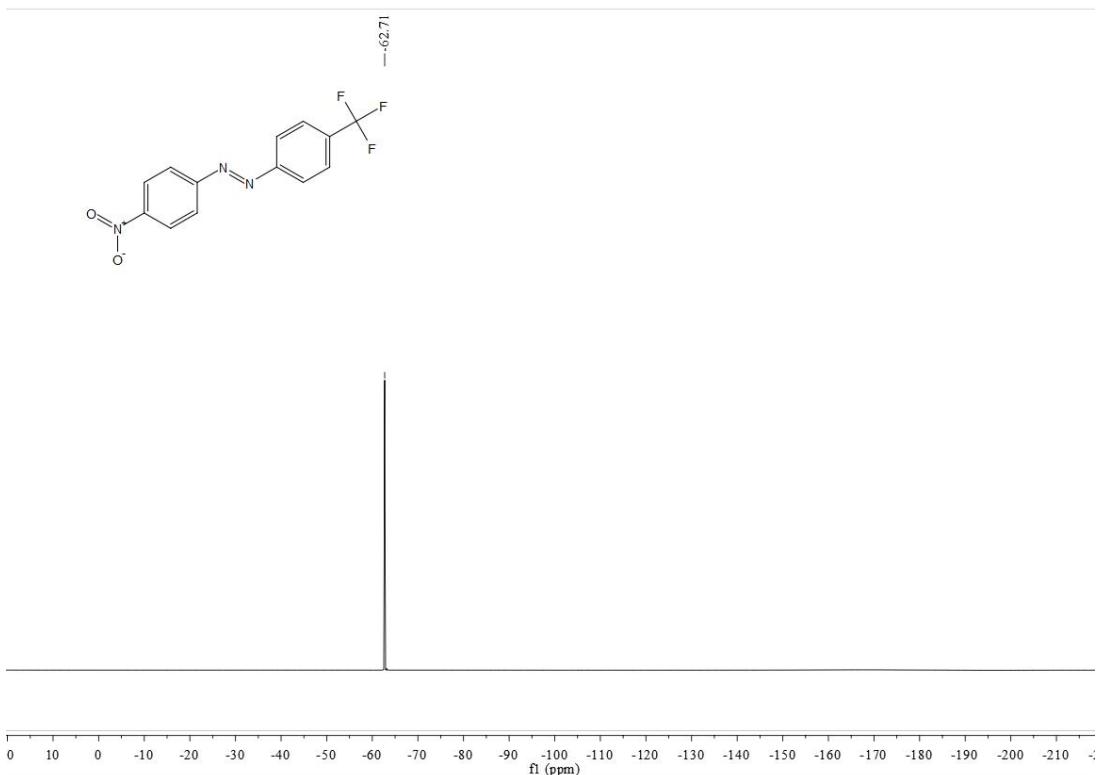
**¹H NMR, ¹³C NMR and ¹⁹F NMR of 1,2-Bis(4-(trifluoromethyl)phenyl)diazene
3ca (500 MHz, CDCl₃)**



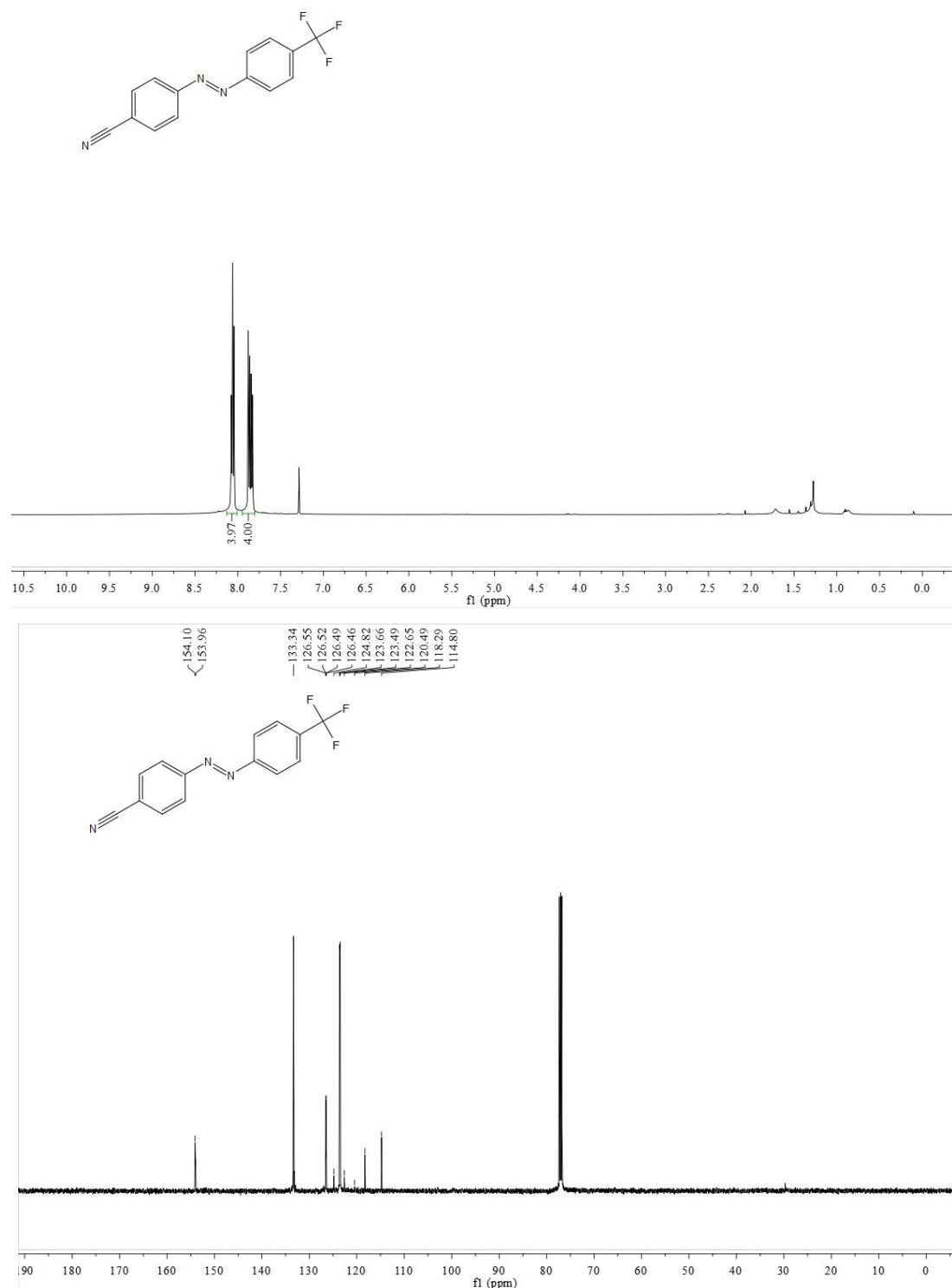


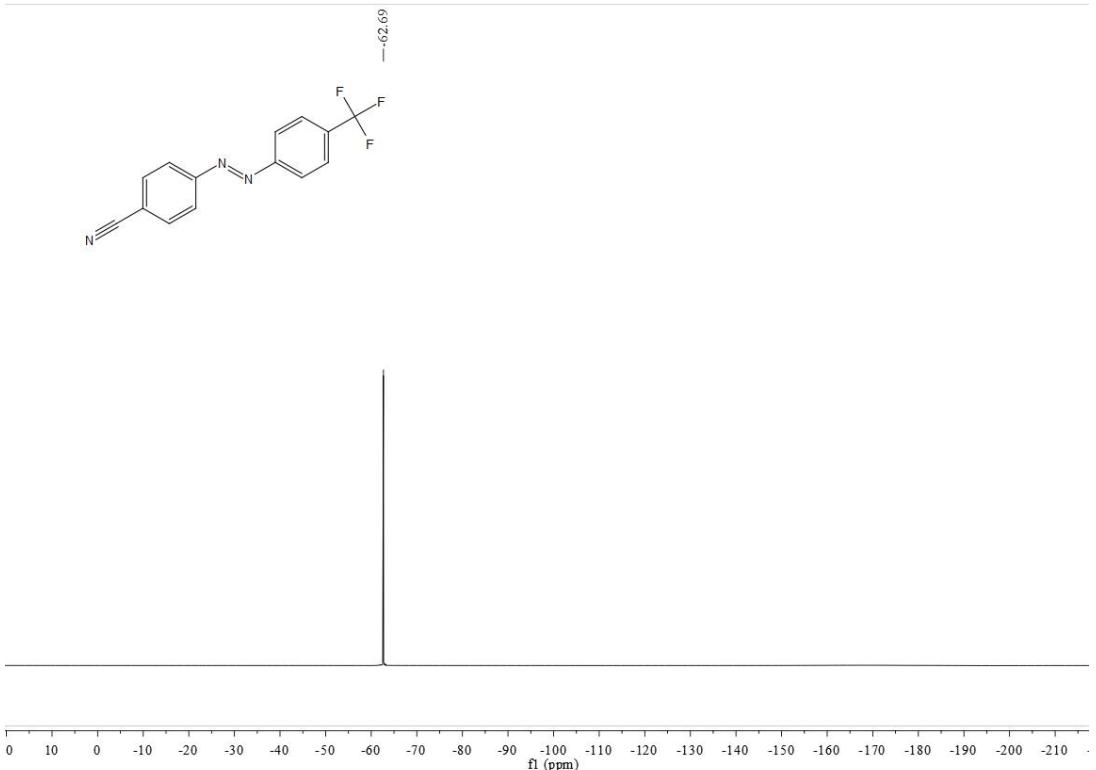
**¹H NMR, ¹³C NMR and ¹⁹F NMR of
1-(4-Nitrophenyl)-2-(4-(trifluoromethyl)phenyl)diazene 3cb (500 MHz, CDCl₃)**



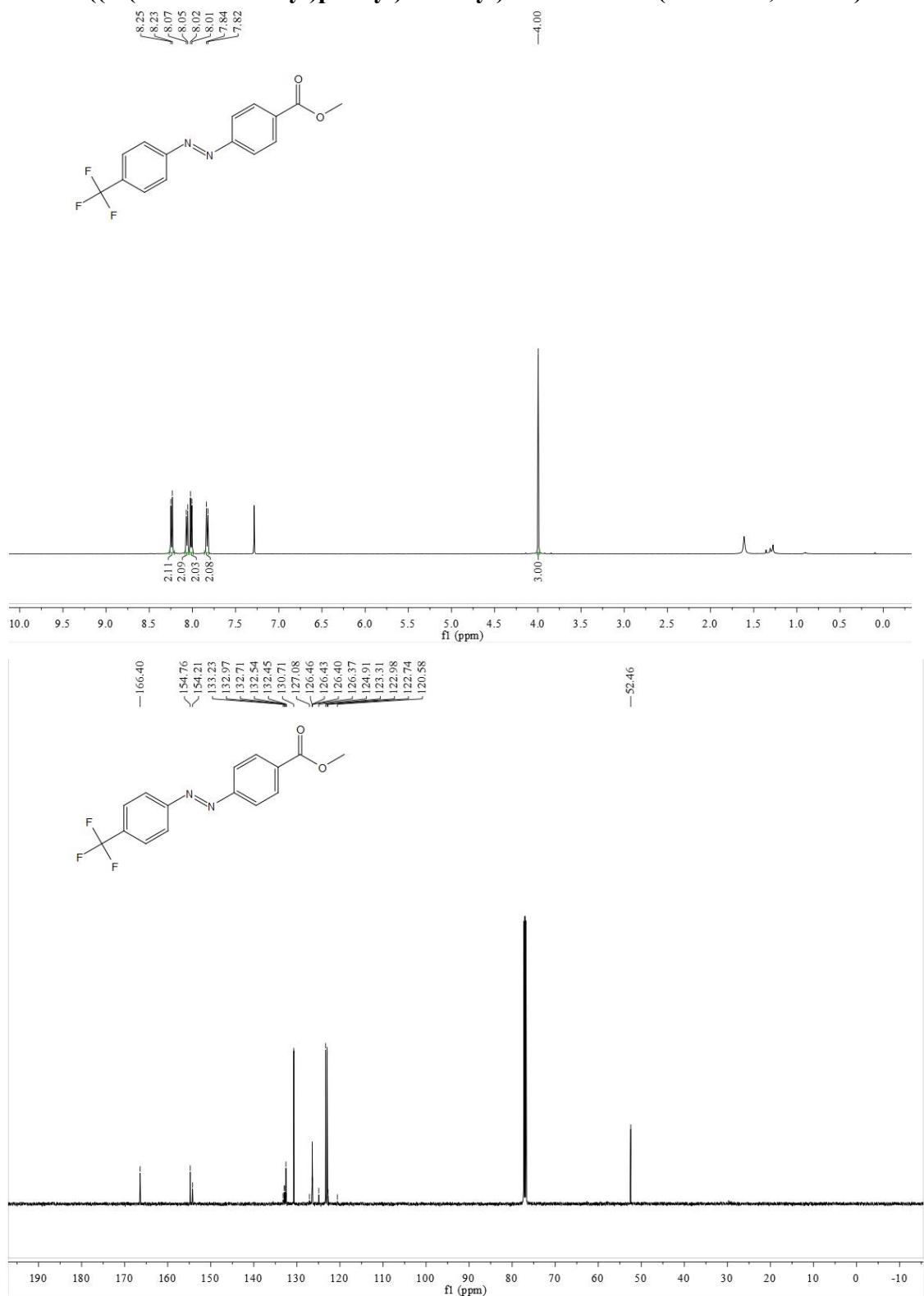


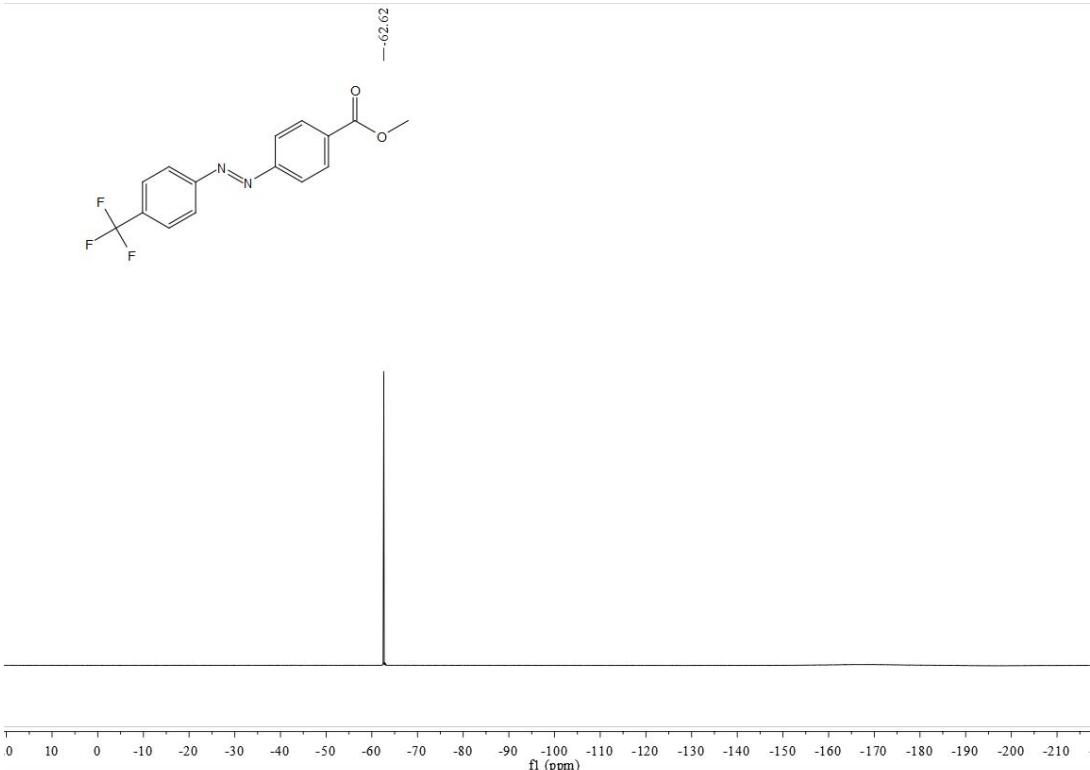
**¹H NMR, ¹³C NMR and ¹⁹F NMR of
4-((4-(Trifluoromethyl)phenyl)diazenyl)benzonitrile 3cc (500 MHz, CDCl₃)**





¹H NMR, ¹³C NMR and ¹⁹F NMR of Methyl -4-((4-(trifluoromethyl)phenyl)diazenyl)benzoate 3cd (500 MHz, CDCl₃)





¹H NMR and ¹³C NMR of (5-Methoxy-2-(4-methoxyphenyl)-2H-indazol-3-yl) methanol (4aa) (500 MHz, CDCl₃)

