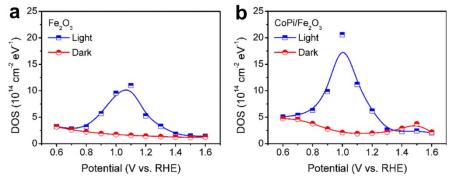
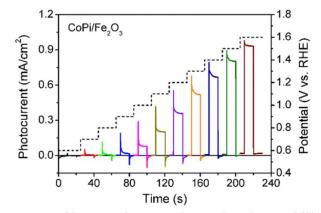
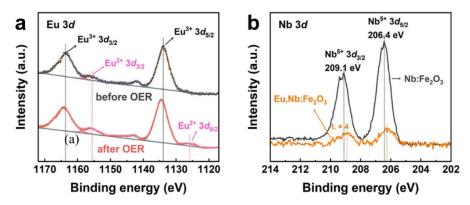
Supplementary Figures



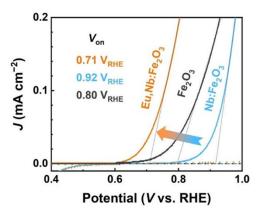
Supplementary Fig. 1 DOS of surface states in (a) bare Fe₂O₃ and (b) CoPi/Fe₂O₃ as a function of potential under illumination (456 nm LED: 5 mW cm⁻²) and in the dark. Reprinted with permission from ref.23. Copyright 2023 Royal Society of Chemistry.



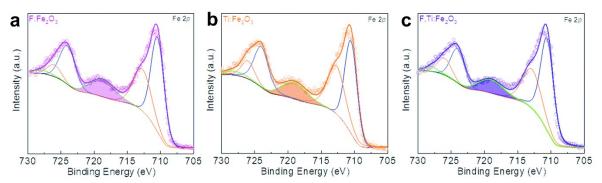
Supplementary Fig. 2 CoPi/Fe₂O₃ photoanodes under chopped illumination (AM 1.5 G: 100 mW cm⁻²). Reprinted with permission from ref.23. Copyright 2023 Royal Society of Chemistry.



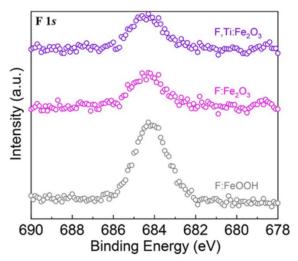
Supplementary Fig. 3 (a) Eu 3d XPS spectra of Eu,Nb: Fe₂O₃ before and after the OER, (b) Nb 3d XPS spectra of Nb: Fe₂O₃ and Eu,Nb: Fe₂O₃. Reprinted with permission from ref.25. Copyright 2024 Royal Society of Chemistry.



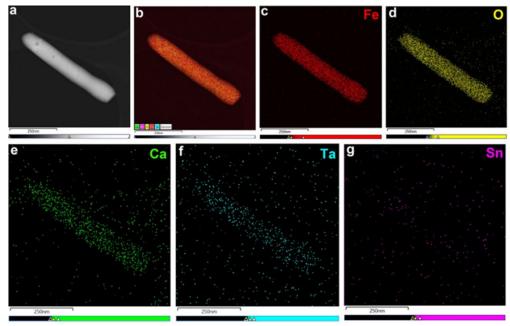
Supplementary Fig. 4 Extracted Von of Fe₂O₃, Nb : Fe₂O₃ and Eu,Nb : Fe₂O₃ Reprinted with permission from ref.25. Copyright 2024 Royal Society of Chemistry.



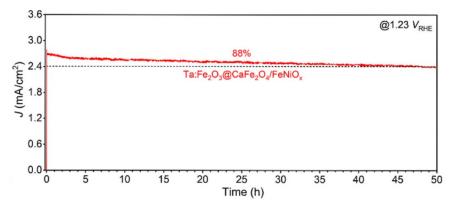
Supplementary Fig. 5 Fe 2p XPS spectra of (a) F:Fe₂O₃, (b) Ti:Fe₂O₃, and (c) F,Ti:Fe₂O₃. The coloured peaks denote the Fe³⁺ satellite peak. Reprinted with permission from ref.26. Copyright 2022 Royal Society of Chemistry.



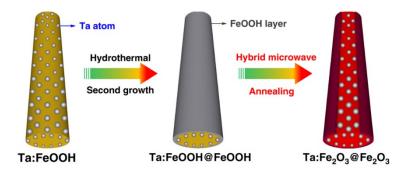
Supplementary Fig. 6 F *Is* XPS spectra of F:FeOOH, F:Fe₂O₃ and F,Ti:Fe₂O₃. Reprinted with permission from ref.26. Copyright 2022 Royal Society of Chemistry.



Supplementary Fig. 7 HAADF (a) and element mapping of $Ta:Fe_2O_3@CaFe_2O_4$ nanorod (b, all elements. c, Fe. d, O. e, Ca. f, Ta. g, Sn). Reprinted with permission from ref.27. Copyright 2023 American Chemical Society.



Supplementary Fig. 8 The long-term stability of $Ta:Fe_2O_3@CaFe_2O_4/FeNiO_x$ photoanode for 50 h at 1.23 V_{RHE} in comparison with $Ta:Fe_2O_3$ photoanode. Reprinted with permission from ref.27. Copyright 2023 American Chemical Society.



Supplementary Fig. 9 Schematic synthesis procedure of Ta:Fe₂O₃@Fe₂O₃ homojunction nanorods. Reprinted with permission from ref.28. Copyright 2020 Springer Nature.

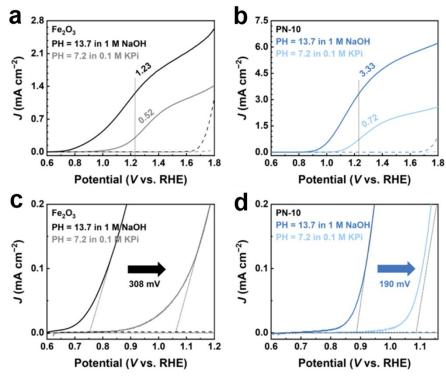


Figure S10. J–V curves of bare Fe₂O₃ (a, c) and PN-10 (b, d) in alkaline (pH 13.7) and near-neutral (pH 7.2) electrolytes. Reprinted with permission from ref.29. Copyright 2024 Springer Nature.