

Electronic Supplementary Information (ESI)

Hexaazatrinaphthylene Based Near-infrared Photothermal Organic-Small-Molecule for Efficient Photo-Thermo-Electric Conversion

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1. Experimental section

Materials. All reagents were obtained from commercial sources and used without further purification. **HTA** was synthesized following a previously reported procedure.¹ The phase purity of **HTA** was confirmed by infrared (IR) spectrum (Fig. S1).

Measurements. Diffuse reflectance spectrum was measured at room temperature over the 250–1500 nm range using a Shimadzu UV-3600i Plus spectrophotometer equipped with an integrating sphere. A BaSO₄ plate was employed as the reference (100% reflectance), and finely ground sample powders were coated directly onto the reference plate. Electron paramagnetic resonance (EPR) spectra were recorded on a CIQTEK EPR-200Plus spectrometer operating in the X-band with a 100 kHz magnetic field modulation. Fourier-transform infrared (IR) spectra were collected in the range of 4000 to 400 cm⁻¹ on a Nicolet iS50 spectrometer.

The NIR photothermal conversion measurement. NIR-I/II photothermal conversion measurements were conducted by pressed pellets of **HTA** with lasers of 808 nm and 1064 nm produced by Changchun New Industries Optoelectronics Tech. Co., Ltd, respectively. The temperature of samples was recorded by a HIKMICRO K20 infrared camera. The temperature detecting range of the infrared camera was set as auto. The sample pellet with a diameter of 5 mm and a weight of 15.9 mg was prepared using a tablet press.

The photo-thermo-electric conversion measurement. The photo-thermo-electric conversion of **HTA** was investigated by integrating its sample powder with a commercial thermoelectric generator. Approximately 50 mg of **HTA** powder was uniformly coated onto the surface of the thermoelectric chip using a thin layer of thermally conductive adhesive. Simulated solar irradiation at intensities of 1 Sun (1000 W/m²), 2 Suns (2000 W/m²) and 4 Suns (4000 W/m²)

were provided by a xenon lamp (CEL-PF300-T9, CEAULIGHT) equipped with an attenuator, and the light intensity was measured using a TES1333 solar power meter. Three thermoelectric devices—TEC1-12701 ($40 \times 40 \times 5.2 \text{ mm}^3$), TEC1-12703 ($40 \times 40 \times 4.4 \text{ mm}^3$), and TEC1-12706 ($40 \times 40 \times 3.8 \text{ mm}^3$)—served as thermoelectric generators. The open-circuit voltage and current were measured with a Keithley 2450 source meter. Surface temperature variations during illumination were monitored using a HIKMICRO K20 infrared thermal camera.

Calculation of optical absorption lines and spin density. The structural models of g-HTA, s-[HTA]^{••} and t-[HTA]^{••} were constructed in GaussView 5.0 and optimized using the B3LYP² function with the 6-311G(d,p) basis set in Gaussian 16,³ followed by time-dependent density functional theory (TD-DFT) calculations. The resulting wavefunction data were all analyzed using the Multiwfn program.⁴

2. Additional tables and graphics

Table S1. Reported photothermal materials and their performance in photo-thermo-electric conversion when integrating with thermoelectric generators under the irradiation of 1 Sun.

Items	Photo-thermo-electric conversion (under 1 Sun)		Reference
	Open circuit voltage (mV)	Maximum output power density (W/m ²)	
Organic-small-molecule photothermal materials			
HTA	211	0.69	This work
DDPA-PDN	83	/	<i>Adv. Funct. Mater.</i> 2021 , 2106247
GDPA-QCN	58	/	<i>Angew. Chem. Int. Ed.</i> 2022 , e202117087
Organic cocrystal photothermal materials			
TQC	427	1.31	<i>ACS Energy Lett.</i> 2023 , 8, 4179–4185.
Organic-inorganic hybrid photothermal materials			
{[BaMn(ONDI) ₂ (H ₂ O) ₃]·H ₂ O} _n }	313	0.873	<i>Chem. Eng. J.</i> 2024 , 491, 152054.
{[Ni ₄ Cl ₂ (OND I) ₂ (bpy) ₄]·2Cl·2H ₂ O·xDMF·y H ₂ O} _n }	250	0.53	<i>Chem. Eng. J.</i> 2024 , 499, 156059.
Inorganic photothermal materials			
NF@RGO-CNT	58-59	0.251	<i>Ind. Eng. Chem. Res.</i> 2022 , 61, 16565–16576.
Ni ₃ S ₂ /NF	59	0.175	<i>ACS Sustain. Chem. Eng.</i> 2020 , 8, 10833–10841.
CNT foam/PVA	96.35	0.4	<i>ACS Appl. Nano Mater.</i> 2021 , 4, 8906–8912.
PNPG/MoS ₂	110	0.23	<i>ACS Appl. Mater. Interfaces.</i>

			2022 , 14, 1034–1044.
3D porous CPP	19.76	0.5	<i>ACS Sustain. Chem. Eng.</i> 2021 , 9, 4571–4582.
PCC sponge	60	0.4	<i>Adv. Energy Mater.</i> 2019 , 1900250
CNTP	100	0.24	<i>Energy Convers. Manag.</i> 2021 , 241, 114306.
MnO/C-600	177	0.77	<i>Chem. Eng. J.</i> 2023 , 451, 138534.
MC10	168.3	0.6	<i>Energy Environ. Mater.</i> 2022 , 0, 1–9.
PC-x hydrogels	165.8	0.65	<i>Chem. Eng. J.</i> 2023 , 458, 141511.
PCC-800	201	0.8	<i>ACS Sustain. Chem. Eng.</i> 2022 , 10, 16427–16439.
MoS ₂ -x NSAs	98.2	0.749	<i>Energy Convers. Manag.</i> 2022 , 252, 115070.
MS _x -CPC	110	1.087	<i>Small</i> 2022 , 18, 2201949.
T-MSMD	89	0.721	<i>ACS Appl. Mater. Interfaces</i> 2021 , 13, 4305–4315.
Polymer photothermal materials			
CP@PVA	112.9	1.04	<i>Sci. China Mater.</i> 2022 , 65, 2491–2501.

Table S2. Reported NIR-II photothermal materials and their photothermal conversion efficiency (PCE, η) under the irradiation of a 1064 nm laser.

Items	η_{1064}	Reference
Organic-small-molecules		
BAF4	80 %	<i>Angew. Chem. Int. Ed.</i> , 2021 , 60, 22376–22384
CY-1234	76.01 %	<i>Small</i> , 2023 , 2307829
2	62.26 %	<i>Angew. Chem. Int. Ed.</i> , 2024 , e202400913
HTA	61.68 %	This work
Zn ₄ -H ₂ Pc/DP	58.3 %	<i>Chem. Sci.</i> , 2019 , 10, 8246–8252
DAF-OH ₂ CGBox-4 ⁴⁺	47.4 %	<i>Angew. Chem. Int. Ed.</i> , 2023 , 135, e202301267
DAF ₂ CGBox-4 ⁴⁺	37.6%	
TTF ₂ CGBox-4 ⁴⁺	39.9%	
CSM ₂	31.6 %	<i>Mater. Horiz.</i> , 2020 , 7, 1379–1386
COFs		
CNPs	50.6 %	<i>Chem. Commun.</i> , 2020 , 56, 7793–7796
Py-BPy-COF NPs	55.2 %	<i>J. Am. Chem. Soc.</i> , 2019 , 141, 14433–14442
HOFs		
TQC@PFC-1	32 %	<i>J. Mater. Chem. B</i> , 2023 , 11, 8649–8656
Complexes		
[Sr(BCA) ₂ (H ₂ O) ₂] _n	84.5 %	<i>Inorg. Chem. Front.</i> , 2024 , 11, 4867–4875
Cu-THQNPs	51.34%	<i>ACS Appl. Mater. Interfaces</i> , 2018 , 10, 25203–25212
Au@MOF	48.5%	<i>Nano Lett.</i> , 2019 , 19, 6772–6780
rPMo·cTMB	48.4 %	<i>Adv. Healthcare Mater.</i> , 2022 , 11, 2102352
pMOF-a	32.2 %	<i>Chem. Commun.</i> , 2022 , 58, 11095–11098
THPTS-Pb	15.2 %	<i>Inorg. Chem.</i> , 2024 , 63, 3327–3334
Polymers		

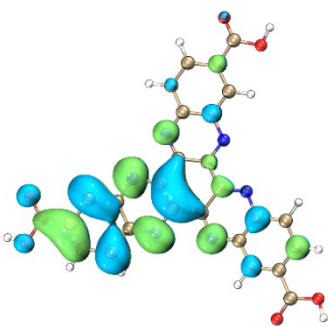
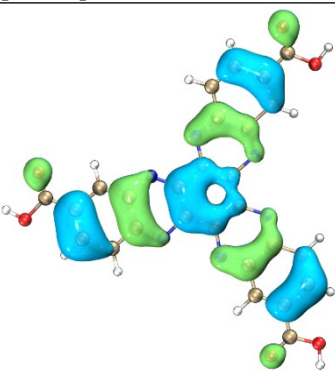
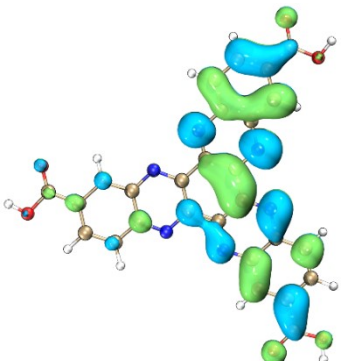
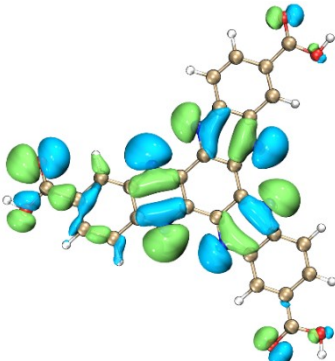
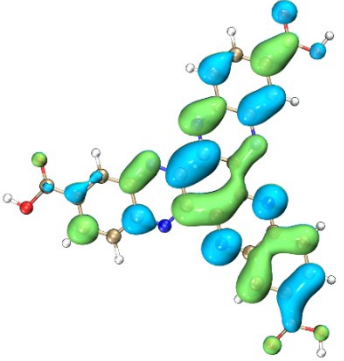
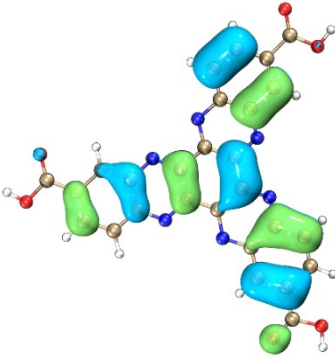
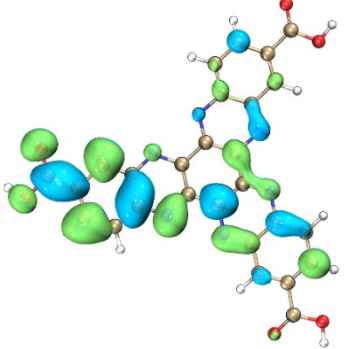
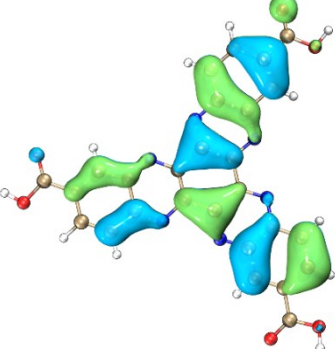
Y1	67.9 %	Chem. Sci., 2021 , 12, 5177–5184
Y2	69.5 %	
Y3	76.5%	
PBBTDTS	65.0 %	Chem. Commun., 2020 , 56, 1093–1096
SPNs3	60.0 %	Chem. Commun., 2019 , 55, 9487–9490
HPW@PANI	57.76 %	Int. J. Nanomed., 2022 , 17, 5565–5579
2MPT ²⁺ •CB	54.6 %	Angew. Chem., Int. Ed., 2019 , 58, 15526–15531
N1@2P	53.8 %	Small, 2023 , 19, 2300203
P ₃	46.0 %	ACS Nano, 2019 , 13, 7345–7354
NP ^{PSP} -Pt	43.2 %	Adv. Mater., 2023 , 35, 2300048
2NDTA	35 %	Adv. Funct. Mater., 2024 , 2401627
SPNs	21.2 %	ACS Appl. Mater. Interfaces, 2020 , 12, 33492–33499.
SP1	2.3 %	Angew. Chem.Int. Ed., 2023 , 62, e202301617
SP2	46.4 %	
SP3	44.9 %	
SP4	46.5 %	
SP5	42.4 %	
Inorganic materials		
N-Doping CDs	81.3 %	Carbon, 2020 , 162, 220–233
AuPBs	80.8 %	ACS Nano, 2018 , 12, 2643–2651
Au ₃ Cu nanocrystals	75.2 %	Nanoscale Horiz., 2018 , 3, 624–631
MoO ₂ NPs	55.6 %	Sci. China Mater., 2020 , 63, 1085–1098
CS–RuO ₂ NPs	52.5 %	Chem. Commun., 2020 , 56, 3019–3022
Pd Ncap	49.2 %	ACS Appl. Mater. Interfaces, 2023 , 15, 39081–39098
Sb-Doped SnO ₂	48.3 %	Nanoscale, 2018 , 10, 2542–2554
Ni ₉ S ₈	46.0 %	Nanoscale, 2019 , 11, 20161–20170
Nb ₂ C (MXene)	45.6 %	J. Am. Chem. Soc., 2017 , 139, 16235–16247

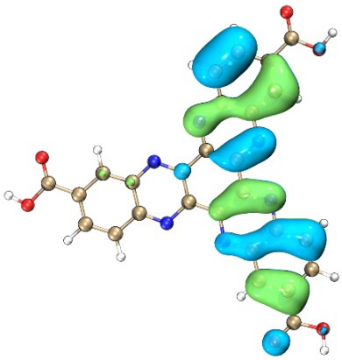
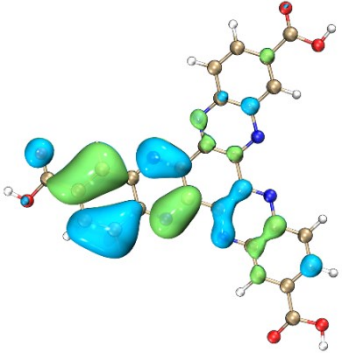
V ₂ C	45.0 %	<i>ACS Nano</i> , 2019 , 13, 1499–1510
TeO ₂ /(NH ₄) _x WO ₃ nanoribbons	43.6 %	<i>Nano Lett.</i> , 2019 , 19, 1179–1189
NIR-II-CD/BP hybrids	28.4 %	<i>ACS Appl. Mater. Interfaces</i> , 2019 , 11, 44949–44960
Si–Au	24.1 %	<i>J. Mater. Chem. B</i> , 2019 , 7, 4393–4401
EGaIn @SiO ₂ -RGD	22.43 %	<i>Nano Lett.</i> , 2019 , 19, 2128–2137
SnSe–PVP nanorods	20.3 %	<i>Mater. Horiz.</i> , 2018 , 5, 946–952
Au NSs	13.0 %	<i>J. Mater. Chem. B</i> , 2019 , 7, 2001–2008

Table S3 List of the triplet diradical states for t-[HTA]^{••} (the lowest singly unoccupied molecular orbital (SUMO) and the singly occupied molecular orbital (SOMO))

Excited State	Wavelength (nm)	Oscillator strength	Electronic transition (%)
4	1186.14 nm	0.0056	β -SOMO-1 \rightarrow β -SUMO (98.37 %)
6	1012.84 nm	0.0097	α -SOMO \rightarrow α -SUMO (81.837 %)
7	956.95 nm	0.0335	α -SOMO \rightarrow α -SUMO+1 (87.643 %)
8	827.20 nm	0.0735	β -SOMO-5 \rightarrow β -SUMO (77.695 %)
9	754.18 nm	0.0009	β -SOMO-6 \rightarrow β -SUMO (89.755 %)
10	718.86 nm	0.0003	β -SOMO-9 \rightarrow β -SUMO (85.643 %)
12	618.26 nm	0.0588	α -SOMO \rightarrow α -SUMO+2 (81.837 %)
15	547.81 nm	0.0748	β -SOMO-11 \rightarrow β -SUMO (68.022 %)

Table S4 List of the molecular orbitals for [HTA]^{••}

Energy level	[HTA] ^{••}	Energy level	[HTA] ^{••}
<i>α</i>-SOMO (-4.938 eV)		<i>β</i>-SOMO-11 (-8.459 eV)	
<i>α</i>-SUMO (-3.076 eV)		<i>β</i>-SOMO-9 (-8.209 eV)	
<i>α</i>-SUMO+1 (-3.005 eV)		<i>β</i>-SOMO-6 (-7.758 eV)	
<i>α</i>-SUMO+2 (-1.945 eV)		<i>β</i>-SOMO-5 (-7.503 eV)	

		β SOMO-1 (-6.999 eV)	
		β SUMO (-5.506 eV)	

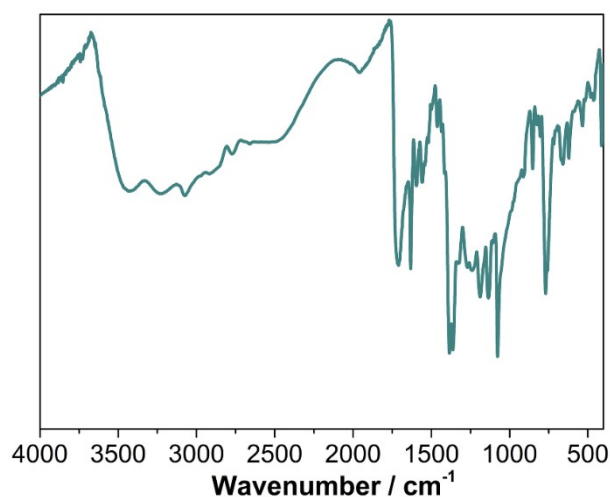


Fig. S1 IR spectrum of **HTA**.

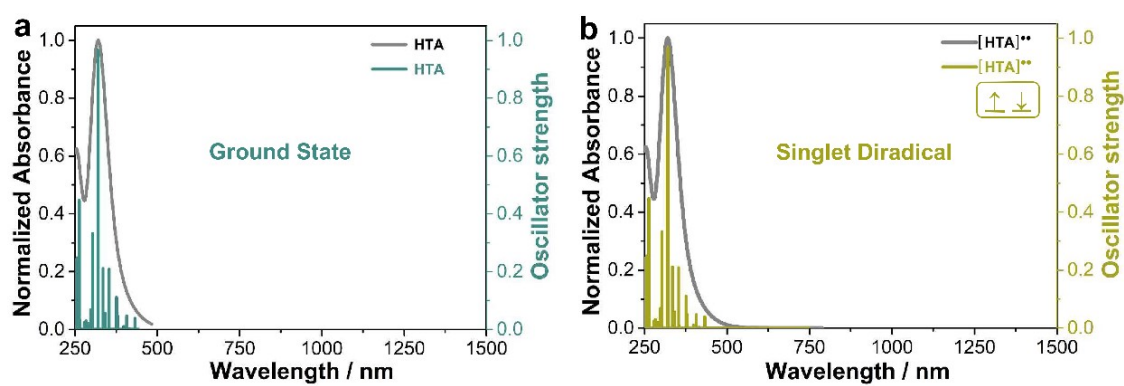


Fig. S2 Calculated optical absorption curves and lines of **HTA** based on ground state (a) and singlet diradical state (b).

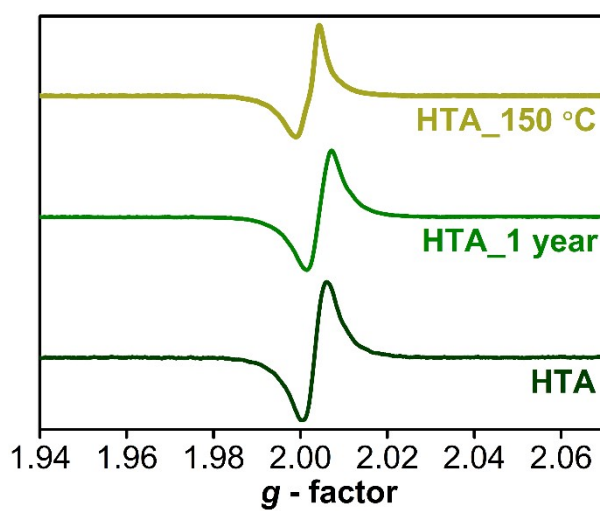


Fig. S3 Solid-state EPR spectra of **HTA** (dark green solid line), the sample **HTA** stored in air

for one year (green solid line) and the sample HTA after annealing at 150 °C for 12 hours (grass green solid line).

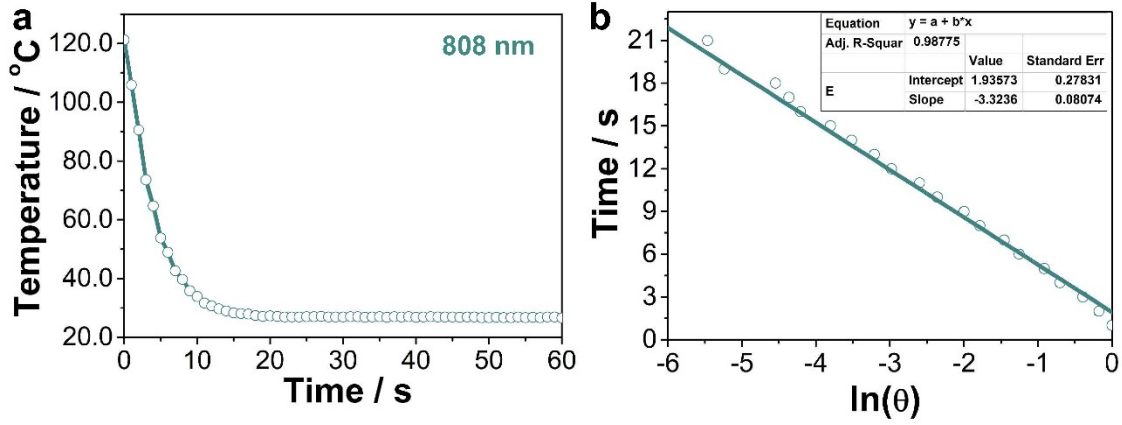


Fig. S4 Temperature decaying curve of compound HTA after removing the laser source of 808 nm (1.25 W/cm²) (a) and the corresponding time-lnθ linear curve (b). The photothermal conversion efficiency ($\eta_{808} = 71.09\%$) was also calculated based on reported method⁵: $\eta_{808} =$

$$\frac{hS\Delta T_{max}}{I(1 - 10^{-A_{808}})}$$
 where the I is the laser power (1.25 W/cm²), A_{808} is the absorbance of the samples at the wavelength of 808 nm (0.50, F(R)), and ΔT_{max} is the maximum temperature

change (94.6 K). hs can be calculated based on the formula of $\tau_s = \frac{\sum_i m_i C_{p,i}}{hs}$, where τ_s is the sample system time constant, m_i (0.0159 g) and $C_{p,i}$ (1.33 J·(g·°C)⁻¹) are the mass and heat capacity of system components. When the laser turns off, τ_s can be estimated according to the

formula: $t = -\tau_s \ln \theta$. The θ can be obtained according to the formula: $\theta = \frac{T - T_{surr}}{T_{max} - T_{surr}}$, where T is the temperature of sample, T_{max} is the maximum system temperature, and T_{surr} is the environment temperature.

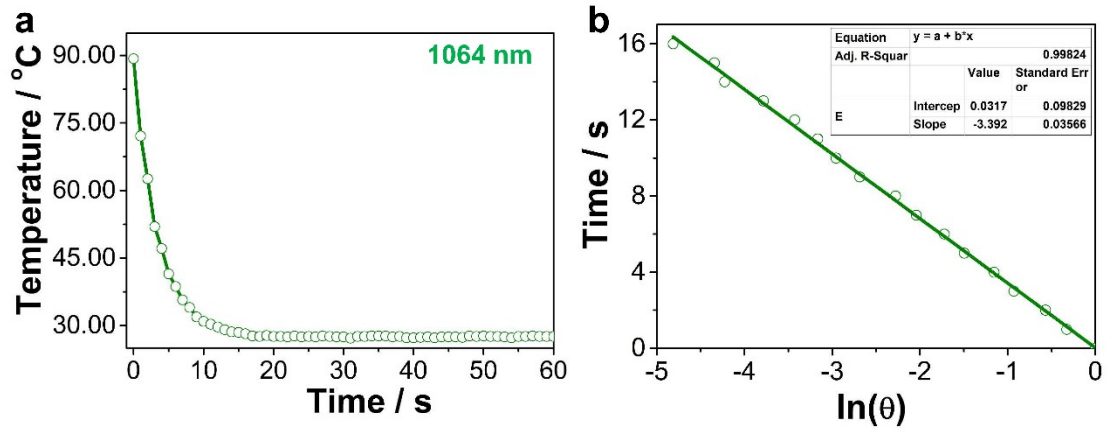


Fig. S5 Temperature decaying curve of compound **HTA** after removing the laser source of 1064 nm (1.25 W/cm²) (a) and the corresponding time-lnθ linear curve (b). The photothermal conversion efficiency ($\eta_{1064} = 61.68\%$) was also calculated based on reported method¹: $\eta_{1064} =$

$\frac{hS\Delta T_{max}}{I(1 - 10^{-A_{1064}})}$, where the I is the laser power (1.25 W/cm²), A_{1064} is the absorbance of the samples at the wavelength of 1064 nm (0.31, F(R)), and ΔT_{max} is the maximum temperature

$$\sum_i m_i C_{p,i}$$

change (63.7 K). hs can be calculated based on the formula of $\tau_s = \frac{\sum_i m_i C_{p,i}}{hs}$, where τ_s is the sample system time constant, m_i (0.0159 g) and $C_{p,i}$ (1.33 J·(g·°C)⁻¹) are the mass and heat capacity of system components. When the laser turns off, τ_s can be estimated according to the

formula: $t = -\tau_s \ln \theta$. The θ can be obtained according to the formula: $\theta = \frac{T - T_{surr}}{T_{max} - T_{surr}}$, where T is the temperature of sample, T_{max} is the maximum system temperature, and T_{surr} is the environment temperature.

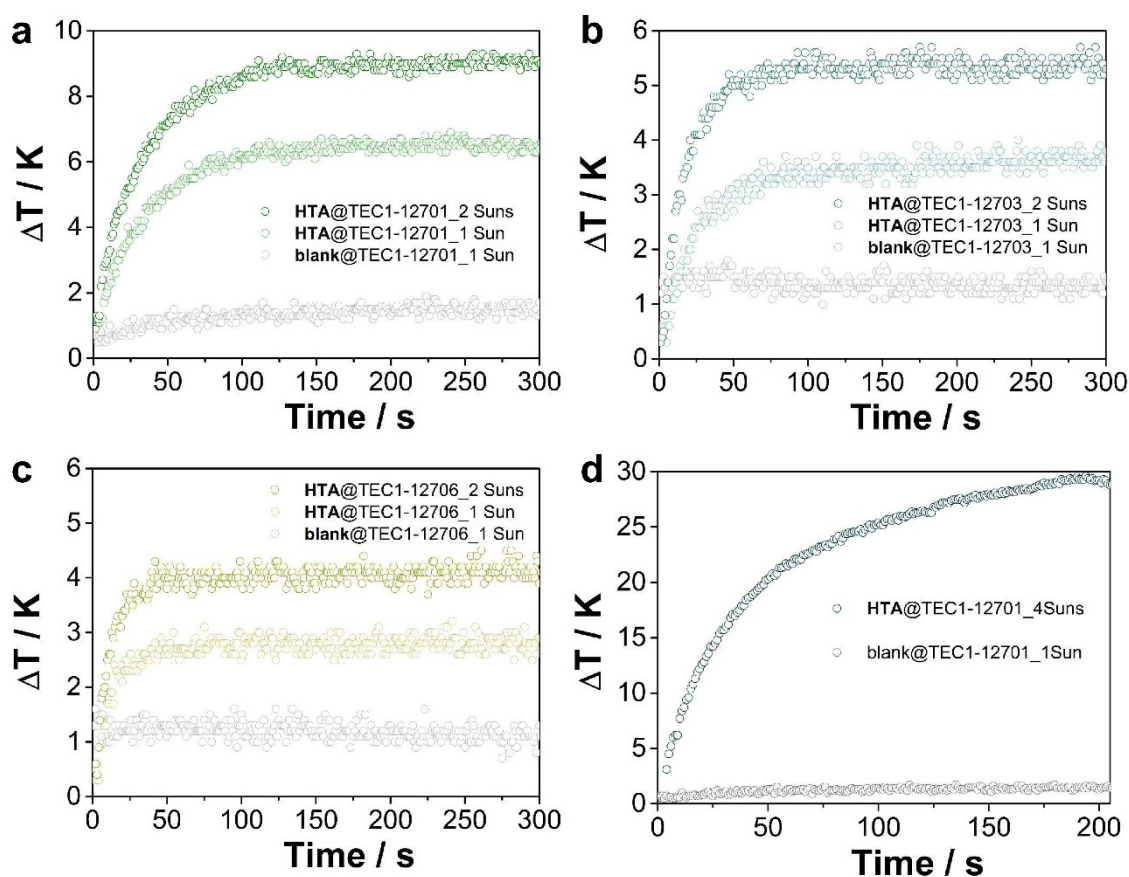


Fig. S6 (a-c) The temperature difference (ΔT) of different photo-thermo-electric conversion devices under the irradiation of 1 Sun and 2 Suns. (d) The temperature difference (ΔT) of HTA@TEC1-12701 under the irradiation of 4 Suns.

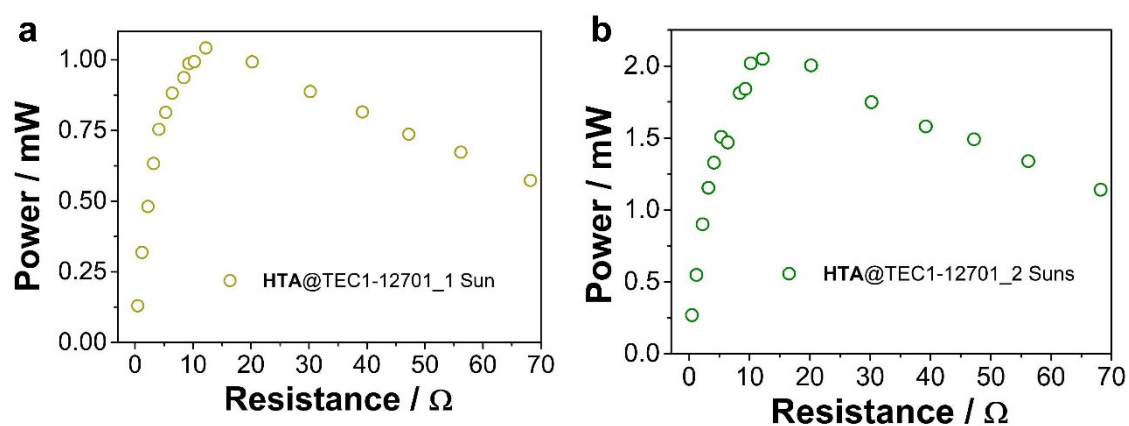


Fig. S7 Output power of HTA@TEC1-12701 under irradiation of 1 Sun (a) and 2 Suns (b) when loading different external resistances, respectively.

Movie S1.

A device HTA@TEC1-12701 drives a small fan under 4 Suns.

3. Reference

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