

A novel sandwich Ni-added polyoxometalate with nonlinear optical properties and photothermal conversion performance

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1. Experimental section

FT-IR spectrum were measured by using a Nicolet iS10 FT-IR spectrometer in the range of 400–4000 cm^{-1} with KBr pallets. Powder X-ray diffraction (PXRD) patterns were recorded on a Bruker D8 Advance XRD diffractometer with Cu K α radiation ($\lambda = 1.54056 \text{ \AA}$). Thermogravimetric analyses were conducted in under N_2 flowing on a Shimadzu DTG-60H with the heating rate of $10 \text{ }^\circ\text{C min}^{-1}$ from 40 to 1000 $^\circ\text{C}$. UV-Vis absorption spectra were obtain using a SP-1901 UV-Vis spectrophotometer. Compound **1** was irradiated using a near-infrared (808 nm) VCL-808nmM1-7W laser. A Fotric 326C+#L25 thermal imaging camera was employed to monitor the sample, collecting real-time temperature data during the process.

2. Supporting Figures

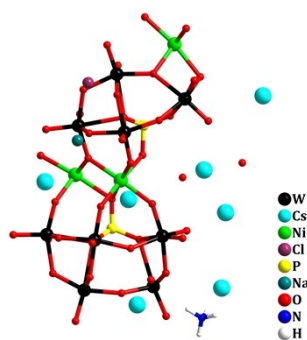


Figure S1 Asymmetric unit of **1**.

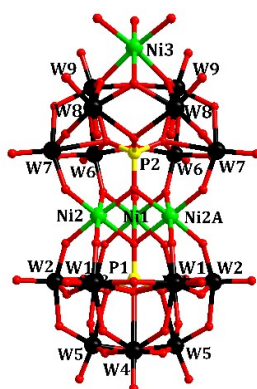


Figure S2 Ball-and-stick view of **1**.

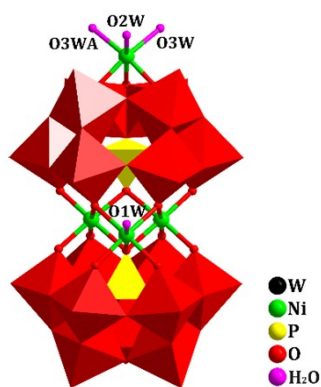


Figure S3 Polyhedral view of **1**.

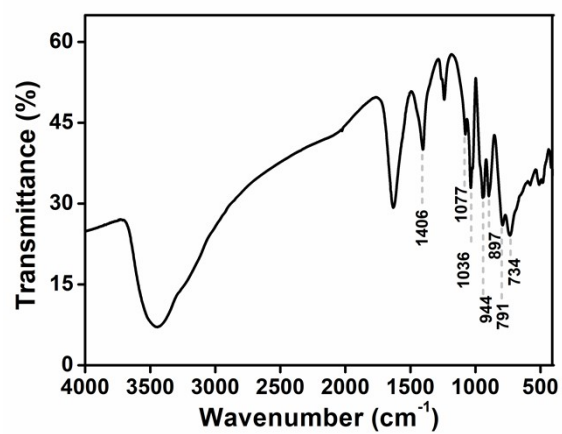


Figure S4 The FT-IR of 1.

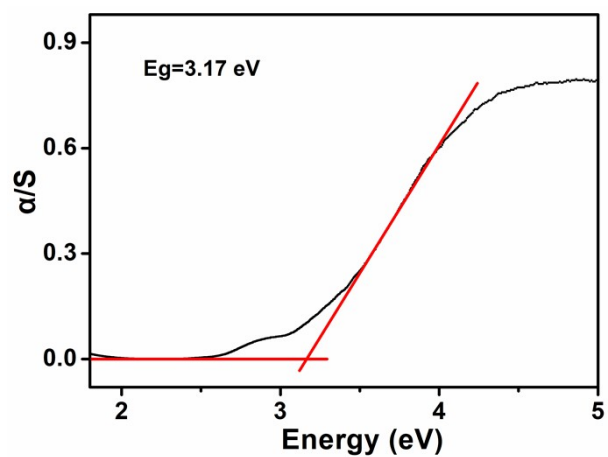


Figure S5 Relationship between Kubelka -Munk function and Energy (eV).

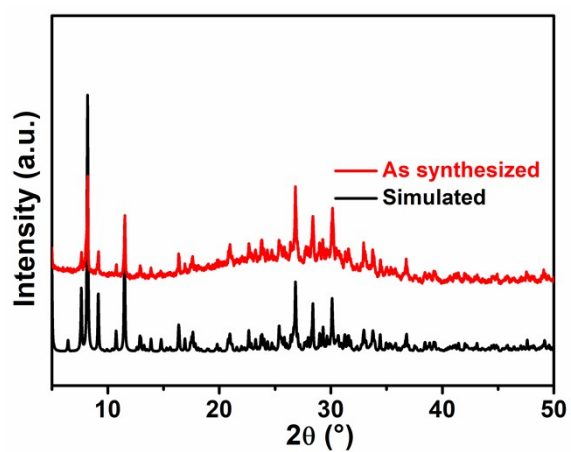


Figure S6 Synthesized and simulated PXRD pattern of 1.

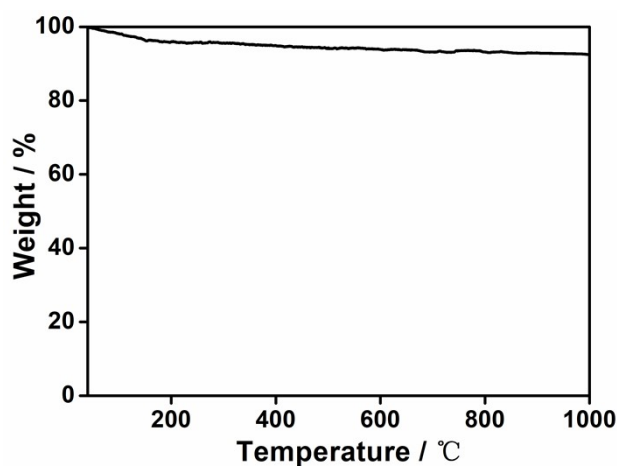


Figure S7 Thermogravimetric analysis curve of 1.

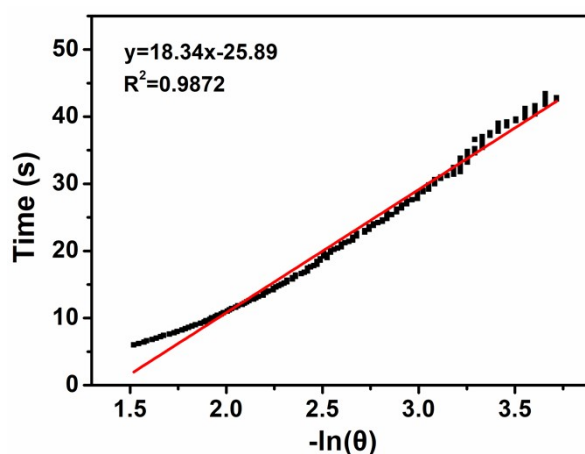


Figure S8 The corresponding time $-\ln\theta$ linear curve of 1.

Table S1 Temperature increase (ΔT) corresponding to the power density for various photothermal materials.

Photothermal materials	Light source	Power density	ΔT (°C)	conversion efficiency	Ref.
$(\text{NH}_4)\text{Cs}_{7.5}\text{Na}_{0.5}\text{H}_2[\text{Ni}(\text{H}_2\text{O})_3\text{Ni}_3(\text{H}_2\text{O})(\text{PW}_8\text{O}_{31})(\text{PW}_9\text{O}_{34})]\text{Cl}\cdot 9\text{H}_2\text{O}$	808 nm laser	1 W cm^{-2}	70 °C	20.5%	This work
$[\text{Ag}_{29.78}\text{Cu}_{1.22}\text{Br}_2(\text{FUR})_{20}(\text{TPP})_{10}]^{3+}$	808 nm	1 W cm^{-2}	88 °C	-	1
DTC cocrystals	808 nm	0.7 W cm^{-2}	42.3 °C	18.8%	2
$\{[\text{La}_3(\text{bcbp})_3(\text{NO}_3)_6\text{O}][\text{La}(\text{NO}_3)_6]_{1/3}\}_n$	808 nm	1.5 W cm^{-2}	88 °C	77%	3
PT-B-COF	808 nm	1.8 W cm^{-2}	43	31.2%	4
PT-N-COF	808 nm	1.8 W cm^{-2}	94	66.4%	

3. Calculation for the photothermal conversion efficiency

The conversion efficiency was evaluated using the approach detailed in previous studies^{5,6}. A comprehensive explanation is provided below:

The photothermal conversion efficiency (η) is obtained according to Equation (1):

$$\eta = \frac{hS(T_{max} - T_{sur})}{I(1 - 10^{-A})} \quad \text{* MERGEFORMAT (1)}$$

Here, T_{max} represents the maximum temperature (97.1 °C), T_{surrr} represents the initial temperature (27.1 °C), I denotes the incident laser power (1 W cm⁻²), and A refers to the absorbance of the sample at 808 nm (0.09).

According to the total energy balance for this system:

$$\sum_i m_i C_{pi} \frac{dT}{dt} = Q_s - Q_{loss} \quad \text{MERGEFORMAT (2)}$$

Here, m_i (0.0126 g) and $C_{p,i}$ (0.8 J⁻¹°C⁻¹) represent the mass and specific heat capacity of the system components (compound **1** and quartz glass), respectively. Q_s represents the thermal energy generated by **1** under NIR laser irradiation, while Q_{loss} represents the thermal energy lost to the surrounding environment.

$$\Delta T = \int_0^t dT \quad \text{MERGEFORMAT (3)}$$

$$Q_{loss}(T) = hS\Delta T \quad \text{MERGEFORMAT (4)}$$

At the maximum temperature, the system reaches thermal equilibrium, expressed as:

$$Q_s = Q_{loss} = hS\Delta T_{max} \quad \text{MERGEFORMAT (5)}$$

To determine hS , a dimensionless driving temperature θ is introduced, defined as follows:

$$\theta = \frac{T - T_{surrr}}{T_{max} - T_{surrr}} = \frac{\Delta T}{\Delta T_{max}} \quad \text{MERGEFORMAT (6)}$$

$$d\theta = \frac{dT}{T_{max} - T_{surrr}} = \frac{dT}{\Delta T_{max}} \quad \text{MERGEFORMAT (7)}$$

The sample system time constant τ_s :

$$\tau_s = \frac{\sum_i m_i C_{p,i}}{hS} \quad \text{MERGEFORMAT (8)}$$

By combining equations (2), (7), and (8), the following can be derived:

$$\frac{d\theta}{dt} = \frac{1}{\Delta T_{max}} \cdot \frac{Q_s - Q_{loss}}{\sum_i m_i C_{p,i}} = \frac{Q_s - Q_{loss}}{hS\tau_s\Delta T_{max}} \quad \text{MERGEFORMAT (9)}$$

$$\frac{Q_{loss}}{hS\tau_s\Delta T_{max}} = \frac{hS\Delta T}{hS\tau_s\Delta T_{max}} = \frac{\theta}{\tau_s} \quad \text{MERGEFORMAT (10)}$$

When the laser is off, $Q_s = 0$, therefore:

$$\frac{d\theta}{dt} = \frac{1}{\tau_s} \cdot \frac{Q_s}{hS\Delta T_{max}} - \frac{\theta}{\tau_s} \quad \text{MERGEFORMAT (11)}$$

$$\frac{d\theta}{dt} = -\frac{\theta}{\tau_s} \quad \text{MERGEFORMAT (12)}$$

$$t = -\tau_s \ln \theta \quad (13)$$

So hS could be calculated from the slope of cooling time (t) vs $-\ln \theta$. Therefore, τ_s is 18.34 s (Fig. S8) and the photothermal conversion efficiency η is 20.5%.

References

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