

Electronic Supplementary Information (ESI)

A Ni-added polyoxometalate: synthesis, structure and catalytic performance

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Fig. S1. (a) The asymmetric structural unit of **1**; (b) Ball and stick model diagram of **1a**.

Fig. S2. (a) The packing mode of **1a** along the *b*-axis; (b) The packing mode of **1a** along the *c*-axis.

Fig. S3. The FT-IR spectrum of **1**.

Fig. S4. The TG curve of **1**.

Fig. S5. Experimental and simulated PXRD patterns of **1**.

Fig. S6. UV-Vis diffuse reflectance spectrum of **1** and the corresponding α/S -Energy curve (inset).

Fig. S7. (a) The Knoevenagel condensation of benzaldehyde with malononitrile detected by GC. Reaction conditions: 1 mmol benzaldehyde, 0.5 μ mol catalyst, 1.2 mmol malononitrile, 3 mL CH₃OH, room temperature, 20 min. (b) Internal standard curve after fitting.

Table S1. BVS calculations of some oxygen atoms in **1**.

Table S2 Comparison of some POM-based catalysts for the Knoevenagel condensation reaction of benzaldehyde with malononitrile.

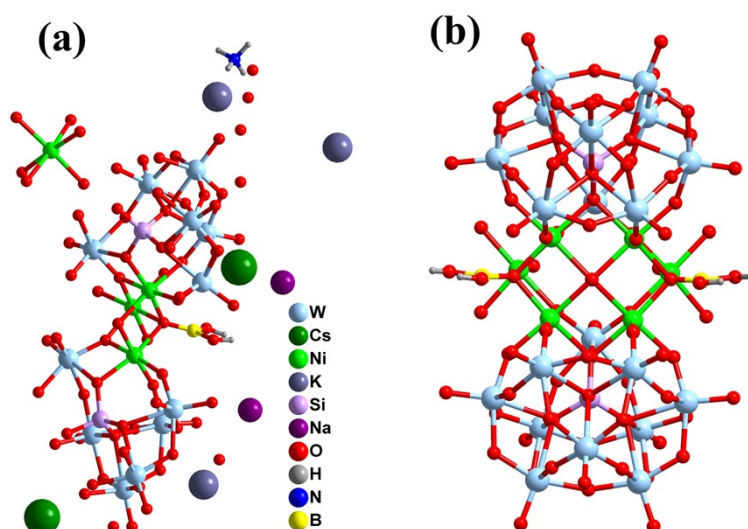


Fig. S1. (a) The asymmetric structural unit of **1**; (b) Ball and stick model diagram of **1a**.

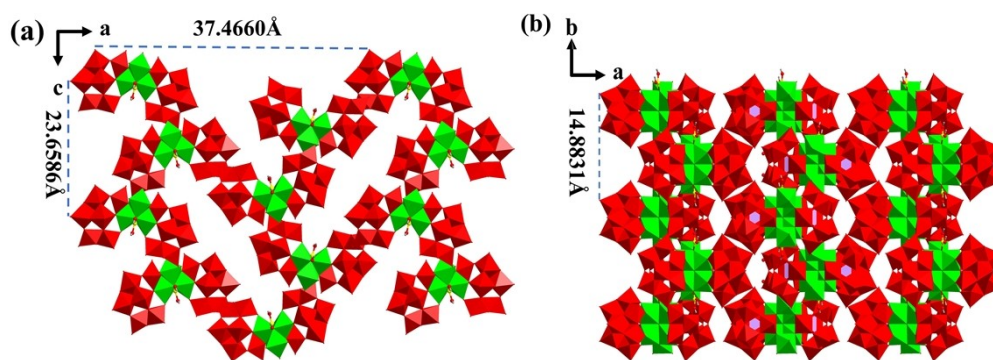


Fig. S2. (a) The packing mode of **1a** along the *b*-axis. (b) The packing mode of **1a** along the *c*-axis. Color labels for polyhedra: WO₆, red; SiO₄, purple; NiO₆, green.

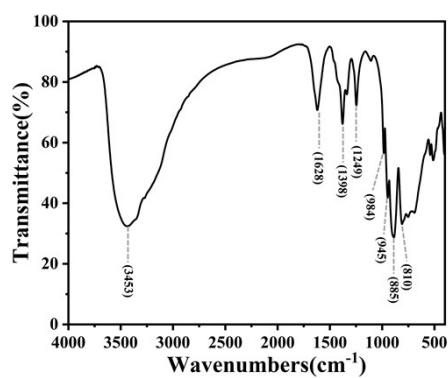


Fig. S3. The FT-IR spectrum of **1**.

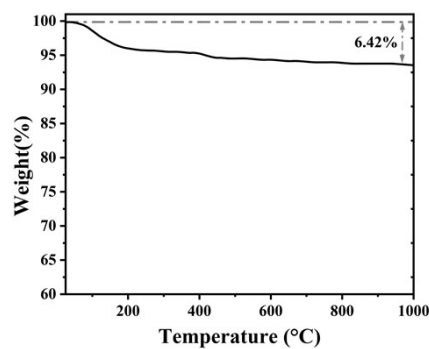


Fig. S4. TG curve of **1**.

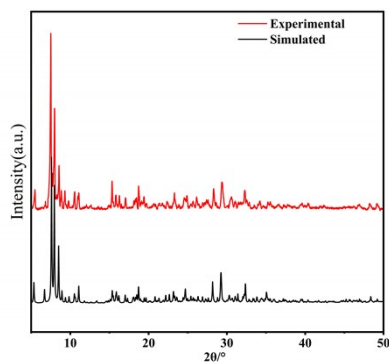


Fig. S5. Experimental and simulated PXRD patterns of **1**.

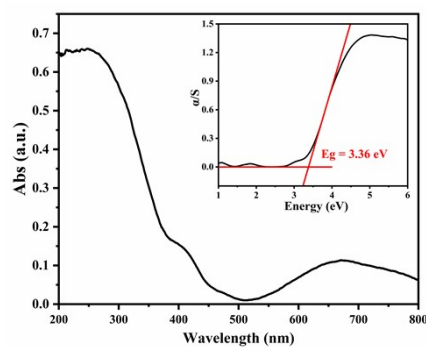


Fig. S6. UV-Vis diffuse reflectance spectrum of **1** and the corresponding α /S-Energy curve (inset).

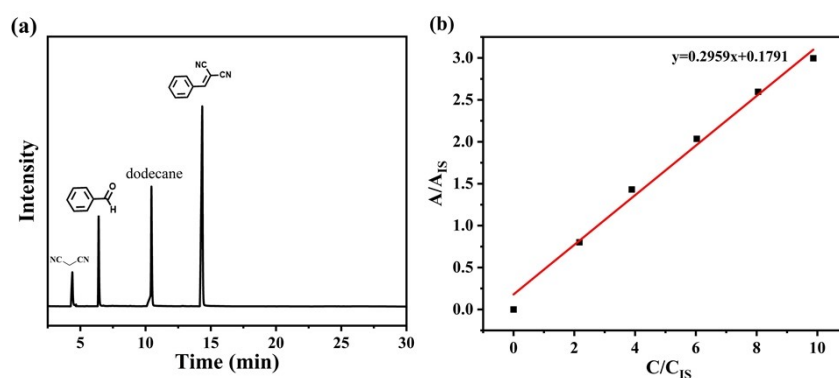


Fig. S7. (a) The Knoevenagel condensation of benzaldehyde with malononitrile detected by GC. Reaction conditions: 1 mmol benzaldehyde, 0.5 μ mol catalyst, 1.2 mmol malononitrile, 3 mL CH₃OH, room temperature, 20 min. (b) Internal standard curve after fitting.

Table S1. BVS calculations of some atoms in **1**.

Atom	BVS	Atom	BVS
O3	0.328	O34	0.311
O13	2.020	O43	0.322
O20	2.117	O47	1.000
O22	2.044	O48	0.825
O30	1.064	B1	2.864

Table S2. Comparison of some POM-based catalysts for the Knoevenagel condensation reaction of benzaldehyde with malononitrile.

Compounds	Time (min)	Temp. (°C)	Conversion (%)	Ref.
Na ₈ H[A-PW ₉ O ₃₄]·7H ₂ O	360	25	92	1
Na ₈ H[B-PW ₉ O ₃₄]·19H ₂ O	360	25	89	1
[γ -SiW ₁₀ O ₃₄ (H ₂ O) ₂] ⁴⁻	150	32	90	2
[H(γ -SiW ₁₀ O ₃₂) ₂ (μ -O) ₄] ⁷⁻	150	32	96	2
Na ₁₆ [SiNb ₁₂ O ₄₀]·xH ₂ O	120	25	100	3
Mg _{0.73} Al _{0.22} (OH) ₂ [PW ₁₂ O ₄₀] _{0.04} ·0.98H ₂ O	360	60	93	4
K ₇ HNb ₆ O ₁₉	45	RT	99	5

$[\text{H}_2\text{N}(\text{CH}_3)_2]_2\text{Na}_{18}\text{Cs}_2\text{H}_{13}[(\text{Cs}_7(\text{H}_2\text{O})_6)@\{(\text{PO}_4)@(\text{Ni}_4(\text{OH})_3(\text{WO}_4))_3@(\text{B}-\alpha\text{-PW}_9\text{O}_{34})_3\}_2]\cdot 30\text{H}_2\text{O}$	60	30	99	6
$[\text{H}_2\text{N}(\text{CH}_3)_2]_2\text{Na}_3\text{H}_5[\{\text{Ti}(\text{C}_2\text{O}_4)(\text{H}_2\text{O})\}\{\text{Ti}(\text{C}_2\text{O}_4)\}(\text{B}-\beta\text{-SbW}_9\text{O}_{33})_2]\cdot 22\text{H}_2\text{O}$	90	30	99	7
$\text{K}_2\text{Na}_{14}\text{H}_{10}[\{\text{Ni}_6(\text{OH})_3(\text{H}_2\text{O})_6(\text{GeW}_9\text{O}_{34})\}\{\text{Ni}_8(\mu_6\text{-O})(\text{OH})_3(\text{H}_2\text{O})\text{BO}(\text{OH})_2\text{B}_2\text{O}_3(\text{OH})_2(\text{GeW}_9\text{O}_{34})_2\}_2]\cdot 46\text{H}_2\text{O}$	90	30	100	8
$\text{K}_4\text{Na}_4\text{H}_{12}[(\{\text{Ni}_8(\mu_6\text{-O})\})@(\{\text{B}_4\text{O}_8(\text{OH})_3\}_2)@(\text{B}-\alpha\text{-GeW}_9\text{O}_{34})_2]\cdot 16\text{H}_2\text{O}$	90	30	100	9
1	50	RT	100	This paper

References

- 1 S. Zhao, Y. Chen and Y. F. Song, *Appl. Catal., A.*, 2014, **475**, 140–146.
- 2 A. Yoshida, S. Hikichi and N. Mizuno, *J. Organomet. Chem.*, 2007, **692**, 455–459.
- 3 W. Ge, X. Wang, L. Zhang, L. Du, Y. Zhou and J. Wang, *Catal. Sci. Technol.*, 2016, **6**, 460–467.
- 4 Y. Jia, Y. Fang, Y. Zhang, H. N. Miras and Y. F. Song, *Chem. – Eur. J.*, 2015, **21**, 14862–14870.
- 5 Q. Xu, Y. Niu, G. Wang, Y. Li, Y. Zhao, V. Singh, J. Niu and J. Wang, *Mol. Catal.*, 2018, **453**, 93–99.
- 6 C. Lian, H. L. Li and G. Y. Yang, *Inorg. Chem.*, 2022, **61**, 11335–11341.
- 7 H. L. Li, C. Lian and G. Y. Yang, *Tungsten*, 2024, **6**, 465–471.
- 8 C. Lian, H. L. Li and G. Y. Yang, *Sci. China. Chem.*, 2023, **66**, 1394–1399.
- 9 C. Lian and G. Y. Yang, *Inorg. Chem.*, 2023, **62**, 21409–21415.