Enantiopure and quasiracemic crystals of 7-substituted tryptophan derivatives: modulation of the molecular arrangement for functionalized crystals

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1. Crystal structure analysis

1-1. Photographic images of crystals



Fig. S1 Photographic images of (a) L-2, (b) L-3- α , and (c) L-3- β .

Table S1 Crystal data of L-1					
Crystal	L- 1-a	L -1-β	L-1-α'		
CCDC	997182 ¹	1937607 ²	2266484 ³		
Empirical formula	$C_{11}H_{12}N_2O_2$	$C_{11}H_{12}N_2O_2$	$C_{11}H_{12}N_2O_2$		
Formula weight	204.23	204.23	204.23		
Temperature [K]	123(2)	294	123(1)		
Wavelength [Å]	0.71073	0.71073	0.56083		
Crystal system	Triclinic	Monoclinic	Triclinic		
Space group	<i>P</i> 1	$P2_{1}$	<i>P</i> 1		
<i>a</i> [Å]	11.430(3)	9.60851(28)	18.315(2)		
<i>b</i> [Å]	11.464(4)	5.20198(14)	5.7707(6)		
<i>c</i> [Å]	35.606(9)	19.7511(6)	9.9414(12)		
α [°]	84.421(4)	90.0	89.015(9)		
β[°]	87.694(4)	93.9514(33)	104.269(10)		
γ[°]	60.102(2)	90.0	95.528(9)		
V[Å ³]	4025.6(19)	984.88(6)	1013.5(2)		
Ζ	16	4	4		
Ζ'	16	2	4		
$D_{ m calcd} [{ m g \ cm^{-3}}]$	1.348	1.377	1.338		
$\mu \ [\mathrm{mm}^{-1}]$	0.095	0.097	0.059		
<i>F</i> (000)	1728	432.0	432		
Crystal size [mm ⁻¹]	$0.62 \times 0.28 \times 0.14$		$0.295 \times 0.246 \times 0.047$		
Radiation	ΜοΚα	CuKa ₁	AgKα		
Index ranges	-16<=h<=16		-22<=h<=22		
	-16<=k<=16		-6<=k<=7		
	-51<= <i>l</i> <=51		-12<= <i>l</i> <=12		
Reflections collected	66471		21786		
Independent reflections	24736		6702		
	$R_{\rm int} = 0.0434$		$R_{\rm int} = 0.1545$		
Data / restraints / parameters	24736 / 3 / 2178		6702 / 3 / 546		
Goodness-of-fit on F^2	1.132		0.632		
Final R indices	$R_1 = 0.0847$		$R_1 = 0.0456$		
[<i>I</i> >2sigma(<i>I</i>)]	$wR_2 = 0.2481$		$wR_2 = 0.0763$		
R indices (all data)	$R_1 = 0.0966$		$R_1 = 0.1282$		
	$wR_2 = 0.2548$		$wR_2 = 0.0918$		
$R_{ m wp}$		0.0082			
R _p		0.0061			
Largest diff. peak and hole $[eÅ^{-3}]$	0.532 and -0.469		0.172 and -0.201		

<u>1-2. Crystal structure of L-tryptophan (1)</u>



Fig. S2 Single-crystal X-ray structures of L-1- α (CCDC: 997182) (C = gray, N = blue, O = red; cyan shading = hydrophilic region, pink shading = hydrophobic region). Packing structures viewed along the (a) *b*-axis and (b) *a*-axis.



Fig. S3 Single-crystal X-ray structures of L-1- β (CCDC: 1937607). The crystal consisted of two crystallographically independent molecules (C = gray or white, N = blue, O = red; cyan shading = hydrophilic region, pink shading = hydrophobic region). Packing structures viewed along the (a) *b*-axis and (b) *a*-axis.



Fig. S4 Single-crystal X-ray structures of L-1- α' (CCDC: 2266484) (C = gray, N = blue, O = red; cyan shading = hydrophilic region, pink shading = hydrophobic region). Packing structures viewed along the (a) *b*-axis and (b) *c*-axis.

	Table S2 Crystal data of L-2 and L-3					
Crystal	L-2	L- 3- a	L- 3-β			
CCDC	2419033	2419168	2419169			
Empirical formula	$C_{12}H_{11}N_3O_2$	$C_{10}H_{11}N_3O_2$	$C_{10}H_{11}N_3O_2$			
Formula weight	229.24	205.22	205.22			
Temperature [K]	296.15	223	223			
Wavelength [Å]	0.0251	1.54184	1.54184			
Crystal system	Monoclinic	Tetragonal	Monoclinic			
Space group	<i>P</i> 2 ₁	<i>I</i> 4	$P2_1$			
<i>a</i> [Å]	11.81(12)	18.69560(8)	9.02963(16)			
<i>b</i> [Å]	4.94(8)	18.69560(8)	5.82382(10)			
<i>c</i> [Å]	17.74(13)	5.95062(4)	9.72260(17)			
α [°]	90	90	90.0000			
β[°]	95.86(17)	90	112.326(2)			
γ [°]	90	90	90.0000			
<i>V</i> [Å ³]	1030(21)	2079.89(2)	472.955(16)			
Ζ	4	8	2			
Ζ'	2	1	1			
$D_{\text{calcd}} [ext{g cm}^{-3}]$	1.479	1.311	1.441			
$\mu [\mathrm{mm}^{-1}]$	0.000	0.782	0.861			
<i>F</i> (000)	186	864	216.0			
Crystal size [mm ⁻¹]	$0.001 \times 0.001 \times 0.001$	$0.6 \times 0.1 \times 0.05$	$0.400\times0.200\times0.015$			
Radiation	TEM	CuKa	CuKa			
Index ranges	-15<=h<=14	-16<=h<=22	-10<=h<=10			
	-6<=k<=6	-17<=k<=22	6<=k<=6			
	-22<= <i>l</i> <=21	6<= <i>l</i> <=7	-11<= <i>l</i> <=11			
Reflections collected	6125	6721	4931			
Independent reflections	3407	1974	1695			
	$R_{\rm int} = 0.1247$	$R_{\rm int} = 0.0206$	$R_{\rm int} = 0.1355$			
Data / restraints / parameters	3407 / 259 / 301	1974 / 1 / 137	1695 / 1 / 152			
Goodness-of-fit on F^2	1.185	1.089	1.031			
Final R indices	$R_1 = 0.1854$	$R_1 = 0.0238$	$R_1 = 0.0717$			
[<i>I</i> >2sigma(<i>I</i>)]	$wR_2 = 0.4603$	$wR_2 = 0.0618$	$wR_2 = 0.1823$			
R indices (all data)	$R_1 = 0.2544$	$R_1 = 0.0240$	$R_1 = 0.0726$			
	$wR_2 = 0.5176$	$wR_2 = 0.0619$	$wR_2 = 0.1837$			
Largest diff. peak and hole [eÅ ⁻³]	0.138 and -0.120	0.119 and -0.168	0.30 and -0.44			

1-3. Crystal structure of 7-substituted tryptophan L-2 and L-3

D–H···A	<i>D</i> –H [Å]	H…A [Å]	$D \cdots A [Å]$	D–H···A [°]
a (N2B–H4B…O1B)	0.86	1.87	2.69(5)	158
b (N1B–H1B…O1A)	0.89	1.80	2.68(5)	171
c (N1B–H2B…O2A)	0.89	1.85	2.72(4)	165
d (N1B−H3B…O2B)	0.89	1.95	2.74(4)	148
e (N1A–H1A…O1B)	0.89	2.01	2.69(4)	133
f (N1A–H1A····O2B)	0.89	2.33	3.00(5)	132
g (N1A–H2A…N3B)	0.89	2.07	2.94(4)	167
h (N1A–H3A…O2A)	0.89	1.89	2.78(4)	176
i (N2A–H4A…N3A)	0.86	1.98	2.78(5)	153

 Table S3. Hydrogen bonds parameters for L-2

Table S4. Graph-set analysis for $L-2^a$

Type of		1.		1		c	_	1.	
H-bond	а	b	с	d	e	I	g	h	1
а	$C_1^1(8)$								
h	$D_3^3(15)$	$n^{1}(2)$							
U	a>b	$D_1(2)$							
0	$D_3^3(15)$	$C_{2}^{2}(6)$	ר <u>י</u> 1(2)						
C	<c>a>c</c>	>b <c< td=""><td>$D_1(2)$</td><td></td><td></td><td></td><td></td><td></td><td></td></c<>	$D_1(2)$						
đ	$C_2^2(11)$	$D_3^3(10)$	$D_3^3(10)$	$C^{1}(5)$					
u	>a>d	d>b	<c>d>c</c>	$c_1(3)$					
e	$D_3^2(11)$	$R_2^2(10)$	$C_2^2(10)$	$D_3^3(10)$	$D^{1}(2)$				
C	>e>a <e< td=""><td>>b>e</td><td>>c>e</td><td>>e>d<e< td=""><td>$D_1(2)$</td><td></td><td></td><td></td><td></td></e<></td></e<>	>b>e	>c>e	>e>d <e< td=""><td>$D_1(2)$</td><td></td><td></td><td></td><td></td></e<>	$D_1(2)$				
f	$D_3^3(13)$	$C_2^2(10)$	$C_2^2(10)$	$D_{3}^{3}(8)$	$C_1^2(4)$	$D^{1}(2)$			
1	>f>a <f< td=""><td>>b>f'</td><td>>c>f</td><td>>f>d<f< td=""><td>>e<f< td=""><td>$D_1(2)$</td><td></td><td></td><td></td></f<></td></f<></td></f<>	>b>f'	>c>f	>f>d <f< td=""><td>>e<f< td=""><td>$D_1(2)$</td><td></td><td></td><td></td></f<></td></f<>	>e <f< td=""><td>$D_1(2)$</td><td></td><td></td><td></td></f<>	$D_1(2)$			
σ	$D_3^3(18)$	$C_2^2(15)$	$C_2^2(15)$	$D_3^3(22)$	$C_{2}^{2}(13)$	$C_{2}^{2}(13)$	$D^{1}(2)$		
8	>g>a <g< td=""><td>>b>g'</td><td>>c>g</td><td>>g>d<g< td=""><td>>e<g< td=""><td>>f < g</td><td>$D_1(2)$</td><td></td><td></td></g<></td></g<></td></g<>	>b>g'	>c>g	>g>d <g< td=""><td>>e<g< td=""><td>>f < g</td><td>$D_1(2)$</td><td></td><td></td></g<></td></g<>	>e <g< td=""><td>>f < g</td><td>$D_1(2)$</td><td></td><td></td></g<>	>f < g	$D_1(2)$		
h		$D_3^3(10)$		$D_{3}^{2}(8)$	$D_3^3(10)$	$D_3^3(10)$	$D_3^3(10)$	$C^{1}(5)$	
11	-	>b>h <b< td=""><td>-</td><td>$>_c>_g<_c$</td><td><e>h>e</e></td><td><f>h>f</f></td><td>< g > h > g</td><td>$c_1(3)$</td><td></td></b<>	-	$>_c>_g<_c$	<e>h>e</e>	<f>h>f</f>	< g > h > g	$c_1(3)$	
i		$D_3^3(20)$		$D_3^3(10)$	$D_3^3(20)$	$D_3^3(20)$	$D_3^3(20)$	$C_2^2(18)$	$C^{1}(6)$
1	-	>b>i <b< td=""><td>-</td><td><c>g'>c</c></td><td><e>i>e</e></td><td><f>i>f</f></td><td><g>j>g</g></td><td>>h<i< td=""><td>$c_1(0)$</td></i<></td></b<>	-	<c>g'>c</c>	<e>i>e</e>	<f>i>f</f>	<g>j>g</g>	>h <i< td=""><td>$c_1(0)$</td></i<>	$c_1(0)$

^a Analyzed by Mercury.⁴

	J	0 1		
D–H··· A	<i>D</i> –H [Å]	H…A [Å]	$D \cdots A$ [Å]	D–H···A [°]
a (N2–H4…O1)	0.870	2.024	2.862(2)	161.15
b (N1–H2…O2)	0.900	2.068	2.947(2)	165.53
c (N1–H1…N3)	0.900	2.182	3.020(2)	154.68
d (N1–H3…O2)	0.900	1.879	2.777(2)	174.35

Table S5. Hydrogen bonds parameters for L-3- α

Table S6. Graph-set analysis for L-**3**- α^{a}

Type of	2	h	2	A
H-bond	a	0	C	u
а	$C_{1}^{1}(8)$			
b	C ₂ ² (13) ≥a≥b	$C_{1}^{1}(5)$		
с	$R_2^2(16) > a < c$	$C_2^2(13) > b > c$	$C_{1}^{1}(8)$	
d	C ₂ ² (13) ≥a≥d	$C_2^2(10) > b > d$	$C_2^2(13) > c > d$	$C_{1}^{1}(5)$

^{*a*} Analyzed by Mercury.⁴

Table S7. Hydrogen bonds parameters for L-3- β

D–H···A	<i>D</i> –H [Å]	H…A [Å]	$D^{\cdots}A\left[\text{\AA} ight]$	D–H···A [°]
a (N2–H4…O1)	0.95(5)	1.86(6)	2.793(4)	168
b (N1–H1…N3)	0.94(6)	2.31(6)	3.129(5)	146
c (N1–H2…O2)	0.97(7)	1.81(7)	2.775(5)	174
d (N1–H3…O1)	1.00(8)	1.78(8)	2.762(5)	164

Table S8. Graph-set analysis for L-**3**- β^{a}

Type of H-bond	a	b	с	d
а	$C_{1}^{1}(8)$			
b	$R_2^2(16) > a < b$	$C_{1}^{1}(8)$		
с	$C_2^2(13) > a > c$	$C_2^2(13) > b > c$	$C_{1}^{1}(5)$	
d	$C_2^2(13) > a > d$	$C_2^2(13) > b > d$	$C_2^2(10) > c > d$	$C_{1}^{1}(5)$

^{*a*} Analyzed by Mercury.⁴



Fig. S5 Electron microscope image of the crystal L-2.



Fig. S6 Molecular arrangement of L-2 viewed along the (a) *a*-axis and (b) *c*-axis [C = gray (molecule A) and white (molecule B), N = blue, O = red].



Fig. S7 (a) Packing structure of L-2 viewed along the *b*-axis, and intermolecular hydrogen bonds formed along the *b*-axis between (b) molecules A and B (CN \cdots H₃N⁺), (c) molecules B and B (COO⁻ \cdots HN), and (d) molecules A and A (CN \cdots HN) [C = gray (molecule A), C = white (molecule B), N = blue, O = red].



Fig. S8 Packing structures of L-**3**- α (C = gray, N = blue, O = red). (a) Front view and (b) back view of the *ab*-plane. (c) Viewed along the *a*-axis. (d) Hydrogen bonds between columns.



Fig. S9 ¹H NMR spectrum (500 MHz, in D₂O) of L-3- α that consists of L-3 and THF.



Fig. S10 Column structures of (a) L-**3**- α along the *c*-axis and (b) L-**3**- β along the *b*-axis [C = light blue (L-**3**- α) and gray (L-**3**- β), N = blue, O = red]. Red dotted lines indicate hydrogen bonds formed along the column extension direction, while blue dotted lines indicate hydrogen bonds connecting adjacent columns.



Fig. S11 Comparison of molecular conformation for dimeric L-3 in α -crystal and β -crystal [C = light blue (L-3- α) and gray (L-3- β), N = blue, O = red].

Table S9 Crystal data for DL-1 and L-2/D-1					
Crystal	DL- 1 ⁵	L-2/D-1			
CCDC	997182	2419170			
Empirical formula	$C_{11}H_{12}N_2O_2$	$C_{12}H_{11}N_3O_2\ /\ C_{11}H_{12}N_2O_2$			
Formula weight	204.23	433.46			
Temperature [K]	143.15	223(2)			
Wavelength [Å]	0.71075	1.54184			
Crystal system	Monoclinic	Monoclinic			
Space group	$P2_1/c$	$P2_1$			
<i>a</i> [Å]	18.981(4)	9.43660(10)			
<i>b</i> [Å]	5.7815(12)	5.76640(10)			
<i>c</i> [Å]	9.3478(19)	19.6310(3)			
α [°]	90	90			
β [°]	101.695(7)	101.2100(10)			
γ [°]	90	90			
V[Å ³]	1004.5(4)	1047.84(3)			
Ζ	4	2			
Ζ'	1	1			
$D_{ m calcd} [{ m g \ cm^{-3}}]$	1.350	1.374			
$\mu \ [\mathrm{mm}^{-1}]$	0.095	0.795			
<i>F</i> (000)	432	456			
Crystal size [mm ⁻¹]	$0.23 \times 0.1 \times 0.04$	0.5 imes 0.2 imes 0.2			
Radiation	ΜοΚα	CuKα			
Index ranges	-15<=h<=24	-11<=h<=11			
	_7<=k<=7	-6<= <i>k</i> <=6			
	-12<= <i>l</i> <=9	-24<= <i>l</i> <=24			
Reflections collected	7205	16079			
Independent reflections	2271	4013			
	$R_{\rm int} = 0.0741$	$R_{\rm int} = 0.0374$			
Data / restraints / parameters	2271 / 0 / 137	4013 / 1 / 313			
Goodness-of-fit on F^2	1.068	1.060			
<pre>Final R indices [I>2sigma(I)]</pre>	$R_1 = 0.0636$	$R_1 = 0.0349$			
	$wR_2 = 0.1646$	$wR_2 = 0.0887$			
R indices (all data)	$R_1 = 0.0871$	$R_1 = 0.0404$			
	$wR_2 = 0.1804$	$wR_2 = 0.0922$			
Largest diff. peak and hole $[eÅ^{-3}]$	0.385 and -0.255	0.137 and -0.157			

1-4. Crystal structure of racemic crystal DL-1 and quasiracemic crystal L-2/D-1

<i>D</i> –Н··· <i>A</i>	<i>D</i> –H [Å]	H…A [Å]	$D \cdots A$ [Å]	D–H···A [°]
a (N1L–H2L…O1L)	0.95(3)	1.90(3)	2.837(3)	168(3)
b (N1L–H1L…O1D)	0.99(3)	1.84(3)	2.821(3)	169(3)
c (N1L-H3L····O2L)	0.99(4)	1.75(4)	2.728(3)	172(3)
d (N1D–H2D····O1D)	0.94(3)	1.93(3)	2.844(3)	165(3)
e (N1D–H3D…O2D)	1.12(5)	1.59(5)	2.707(3)	172(4)
f (N1D–H1D····O1L)	0.92(3)	1.93(3)	2.837(3)	173(3)

Table S10. Hydrogen bonds parameters for L-2/D-1

Table S11. Graph-set analysis for $L-2/D-1^a$

Type of	0	h	2	d	2	f
H-bond	a	b	C	u	e	1
а	$C_{1}^{1}(5)$					
h	$D_3^3(10)$	$n^{1}(2)$				
U	a>b	$D_1(2)$				
2	$C_2^2(10)$	$D_3^3(10)$	$c^{1}(r)$			
C	>a>c	c>b	$c_1(5)$			
Ŀ		$D_{3}^{3}(8)$				
a	-	>b>d <b< td=""><td>-</td><td>$L_{1}(5)$</td><td></td><td></td></b<>	-	$L_{1}(5)$		
		$D_3^3(10)$		$C_{2}^{2}(10)$		
e	-	- >b>e <b< td=""><td>-</td><td>>d>e</td><td>$L_1(5)$</td><td></td></b<>	-	>d>e	$L_1(5)$	
£	$D_{3}^{2}(8)$	$R_2^2(10)$	$D_3^3(10)$	$D_3^3(10)$	$D_3^3(10)$	م ¹ (2)
I	>f>a <f< td=""><td>>b>f</td><td>>f>c<f< td=""><td><f>d>f</f></td><td><f>e>f</f></td><td>$D_{1}(2)$</td></f<></td></f<>	>b>f	>f>c <f< td=""><td><f>d>f</f></td><td><f>e>f</f></td><td>$D_{1}(2)$</td></f<>	<f>d>f</f>	<f>e>f</f>	$D_{1}(2)$

^{*a*} Analyzed by Mercury.⁴



Fig. S12 Single-crystal X-ray structures of racemic crystal DL-1 [C = light blue (L-form) and purple (D-form), N = blue, O = red]. Packing structures viewed along the (a) *b*-axis and (b) *c*-axis. Short contacts between adjacent molecules are indicated as dotted lines.



Fig. S13 Single-crystal X-ray structures of quasiracemic crystal L-2/D-1 [C = light blue (L-2) and purple (D-1), N = blue, O = red]. Packing structures viewed along the (a) *b*-axis and (b) *a*-axis. Short contacts between adjacent molecules are indicated as dotted lines.



Fig. S14 Void spaces of adjacent molecules in (a) racemic crystal DL-1 and (b) quasiracemic crystal L-2/D-1 visualized in yellow using a 0.7 Å probe [C = light blue (L-form) and purple (D-form), N = blue, O = red].



Fig. S15. Graph-set analysis for hydrogen bonds formed by each atom of (a) crystal L-2, (b) α - and β - crystal L-3, and (c) quasiracemic crystal L-2/D-1.

2. Hirshfeld surface analysis

The interactions with neighboring molecules in the crystal structures of L-3- α and L-3- β were confirmed through Hirshfeld surface analysis using the CrystalExplorer package V.21.310⁶ (Fig. S16).



Fig. S16 (a) Hirshfeld surfaces of L-3 within the α -crystal and β -crystal mapped with normalized contact distance (d_{norm}). Red, white, and blue areas on the surfaces indicate that the intermolecular contact distances are shorter, equal to, and longer than their van der Waals radii, respectively. (b) Individual atomic contact percentage contributions to the Hirshfeld surface in L-3- α , L-3- β , and L-1.

3. Bravais-Friedel-Donnay-Harker (BFDH) morphology

Bravais–Friedel–Donnay–Harker (BFDH) morphologies of L-3- α , L-3- β , DL-1, and L-2/D-1 were obtained using Mercury (Figs S17 and S18).⁴ All the crystals grew in the same direction as the continuous hydrogen bonds were formed.



Fig. S17 BFDH morphologies of (a) L-3- α and (b) L-3- β .



Fig. S18 BFDH morphologies of (a) DL-1 and (b) L-2/D-1.

4. Theoretical calculations

The HOMO and LUMO of L-2 and D-1 calculated by time-dependent density functional theory (TD-DFT) are shown in Fig. S19.



Fig. S19 Calculated fluorescence wavelengths and molecular orbitals of L-2 and D-1.

5. References

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