Supplementary Information (SI) for CrystEngComm.
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Harmonic $NaNbO_3$ and $KNbO_3$ nanoparticles via solvothermal synthesis in glycols

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Electronic Supplementary Informations (ESI)

Table 1 Results of the Rietveld analyses of all samples.

$NaNbO_3$:		Proportions			
Sample	size $NaNbO_3$ [nm]	$NaNbO_3$ [%]	Nb_2O_5 [%]	$Na_7(H_3O)(Nb_6O_{19})(H_2O)_{14}$ [%]	$Na_2C_2O_4$ [%]
NN-h-1	96	97	3		
NN-h-2	34	69	31		
NN-h-3	97	21	7	72	
NN-h-4				100	
NN-t-1	86	100			
NN-t-2	162	100			
NN-t-3				100	
NN-t-4				100	
NN-Ox-1	56	90			10
NN-Ox-2	81	90			10
NN-Ox-3	195	67		19	14
NN-Ox-4	94	68		15	17
NN-tb-1	9	72		15	13
NN-tb-2	6	85.2		0.2	14.6
NN-tb-3	16	55		0.35	44.65
NN-tb-4	6	99		1	

$KNbO_3$:		
Sample	size $KNbO_3$ [nm] (orthorhombic)	size $KNbO_3$ [nm] (monoclinic)
KN-h-1	16	17
KN-h-2	58	71
KN-h-3	49	
KN-h-4	62	
KN-t-1	24	28
KN-t-2	48	
KN-t-3	56	50
KN-t-4	45	
KN-Ox-1		
KN-Ox-2	36	
KN-Ox-3	101	
KN-Ox-4	145	108
KN-tb-1		
KN-tb-2	5	10
KN-tb-3	7	7
KN-tb-4	4	4

glycol	boiling point [° C]	dipole moment* [10^{-30} C · m]
EG	197	7.94
DEG	244	8.97
TrEG	291	9.97
TEG	327	10.84

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* T. Uchida, Y. Kurita, N. Koizumi and M. Kubo, Journal of Polymer Science, 1956, **21**, 313–322

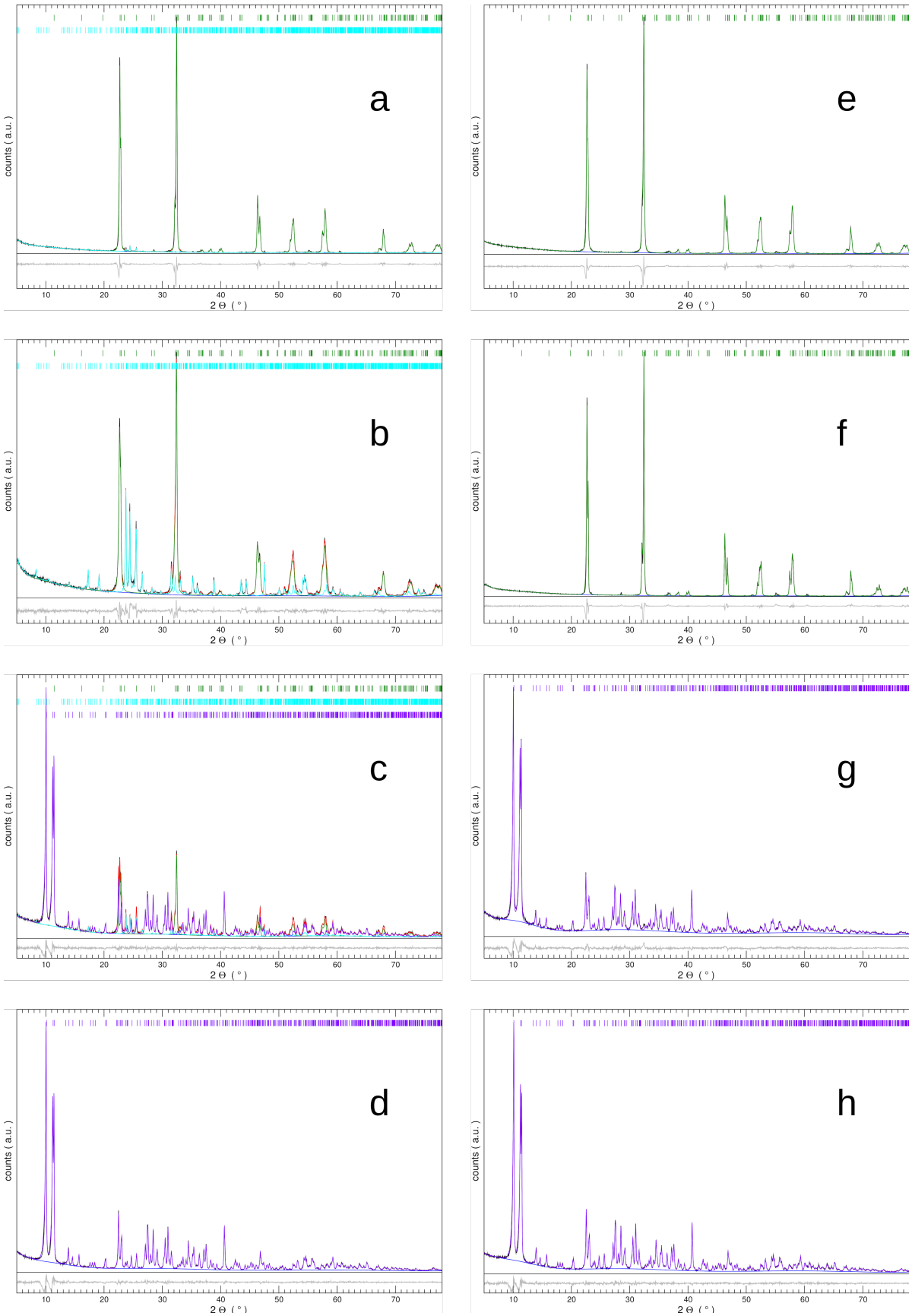

Fig. S 1 Diffractograms of the samples synthesized with the niobium oxides (h-form and t-form) and sodium hydroxide: a NN-h-1, b NN-h-2, c NN-h-3, d NN-h-4, e NN-t-1, f NN-t-2 , g NN-t-3 and h NN-t-4. Colors are: Measurement | , fit | , background | , residual | , $NaNbO_3$ | , Nb_2O_5 | and $Na_7(H_3O)(Nb_6O_{19})(H_2O)_{14}$ | .

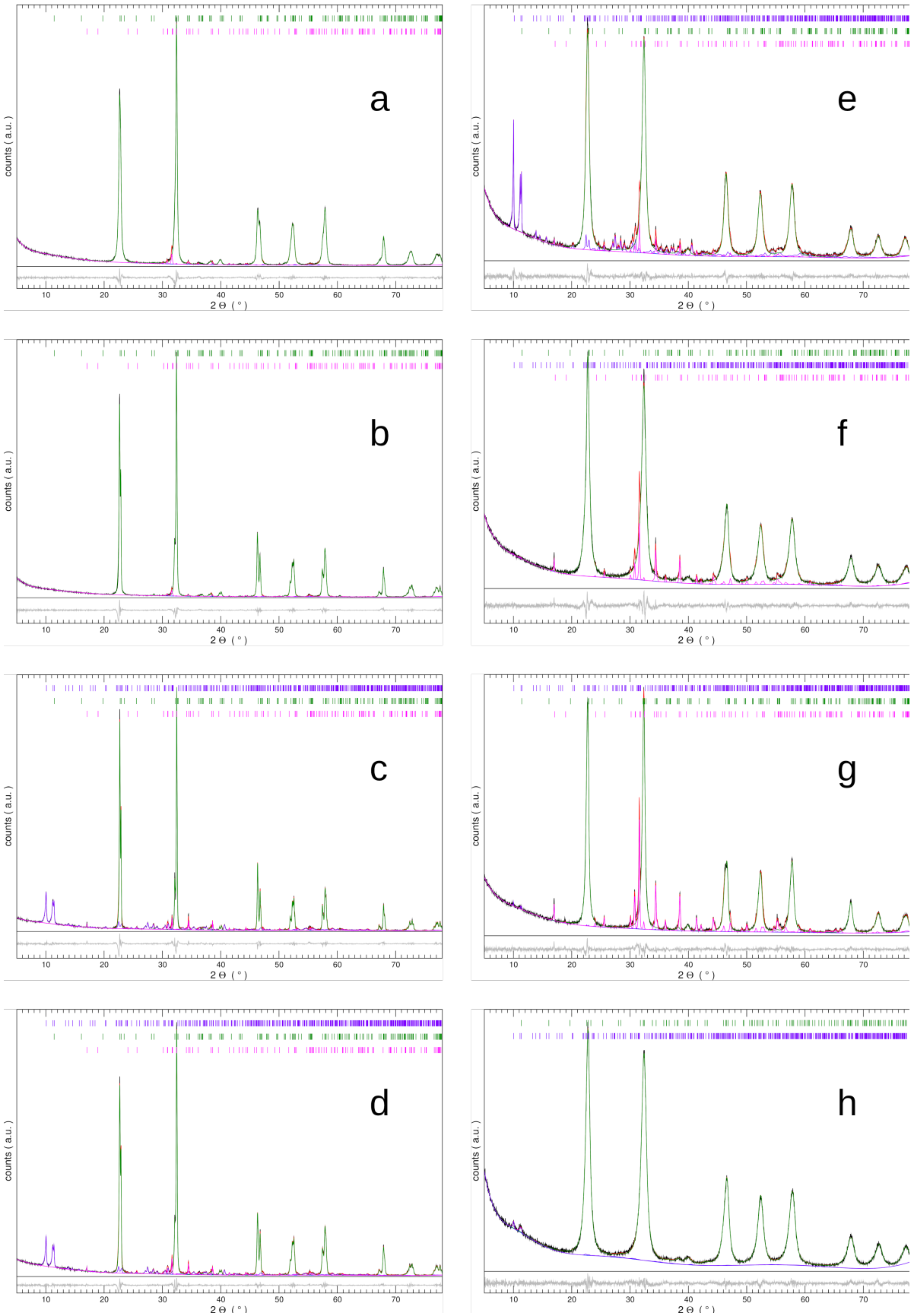

Fig. S 2 Diffractograms of the samples synthesized with niobium oxalate and sodium hydroxide: a NN-Ox-1, b NN-Ox-2, c NN-Ox-3, d NN-Ox-4 and with niobium oxalate and sodium tert-butoxide: e NN-tB-1, f NN-tB-2 , g NN-tB-3 and h NN-tB-4. Colors are: Measurement | , fit | , background | , residual | , $NaNbO_3$ | , $Na_7(H_3O)(Nb_6O_{19})(H_2O)_{14}$ | and $Na_2C_2O_4$ | .

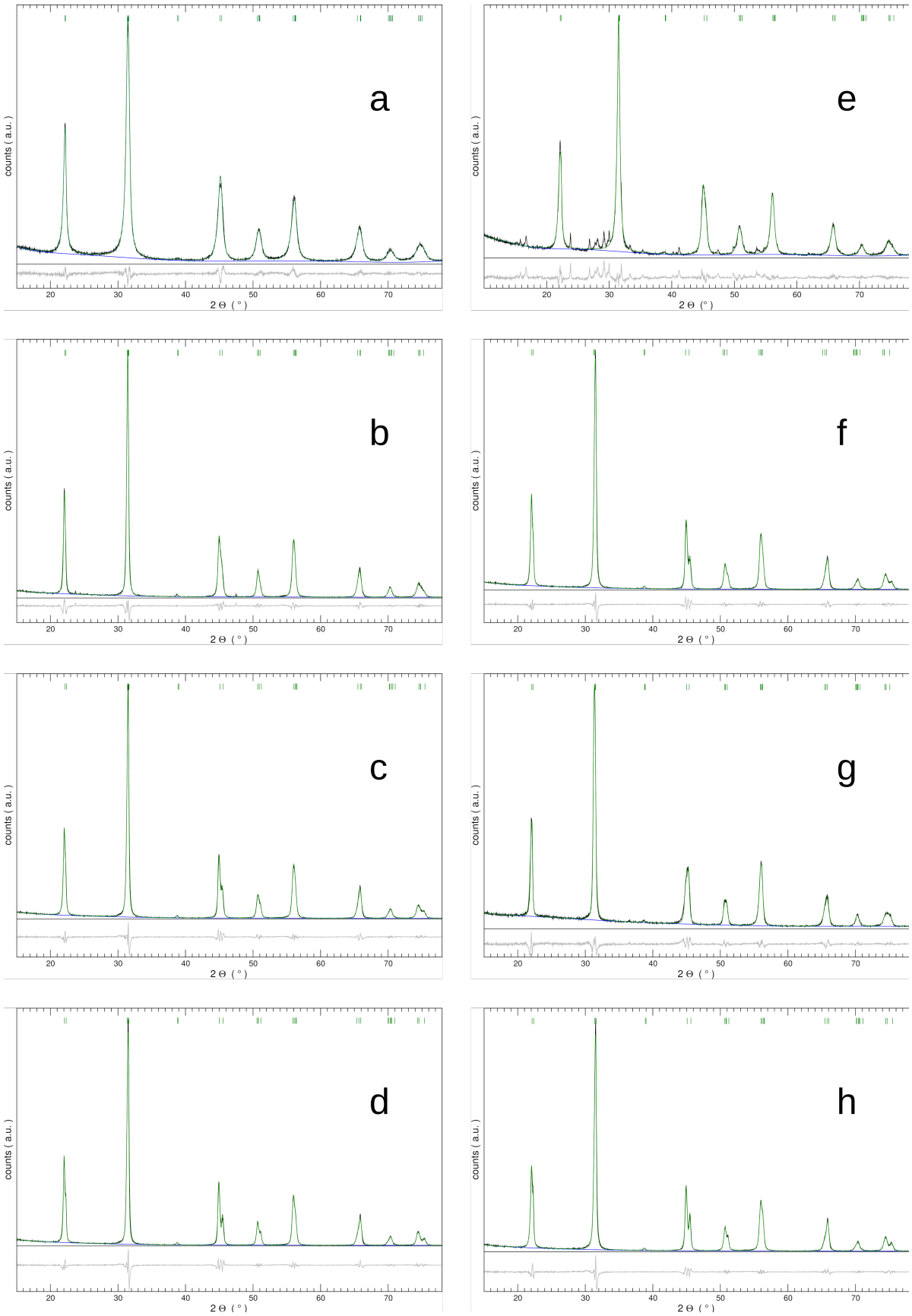

Fig. S 3 Diffractograms of the samples synthesized with the niobium oxides (h-form and t-form) and potassium hydroxide: a KN-h-1, b KN-h-2, c KN-h-3, d KN-h-4, e KN-t-1, f KN-t-2 , g KN-t-3 and h KN-t-4. Colors are: Measurement | , fit | , background | , residual | and $KNbO_3$ | .

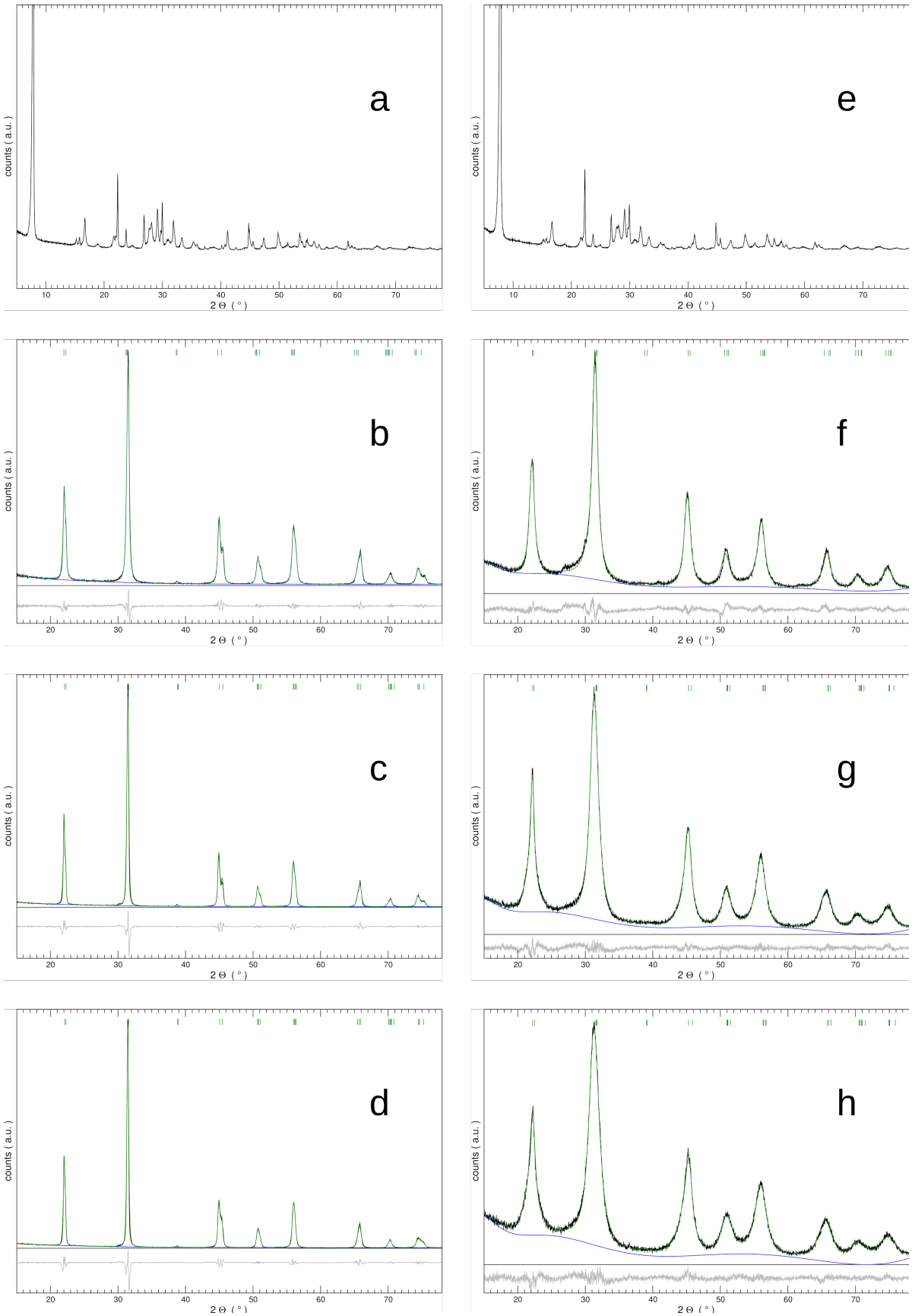

Fig. S 4 Diffractograms of the samples synthesized with niobium oxalate and potassium hydroxide: a KN-Ox-1, b KN-Ox-2, c KN-Ox-3, d KN-OX-4 and niobium oxalate and potassium tert-butoxide: e KN-tB-1, f KN-tB-2 , g KN-tB-3 and h KN-tB-4. Colors are: Measurement | , fit | , background | , residual | .