

Supporting Information

Improvement of humidity stability of ethyl gallate by cocrystal strategy and study of dissolution behavior of the cocrystal

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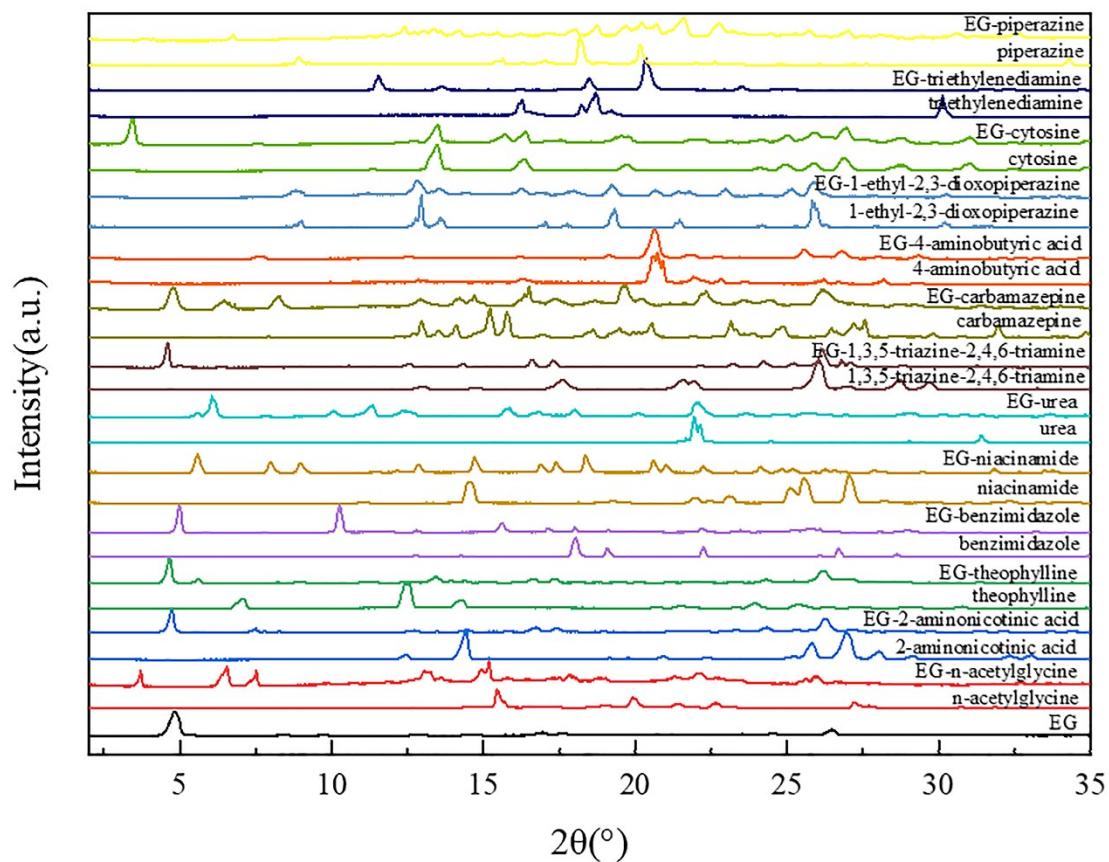


Fig. S1 PXRD patterns of LAG experiment product with new phases.

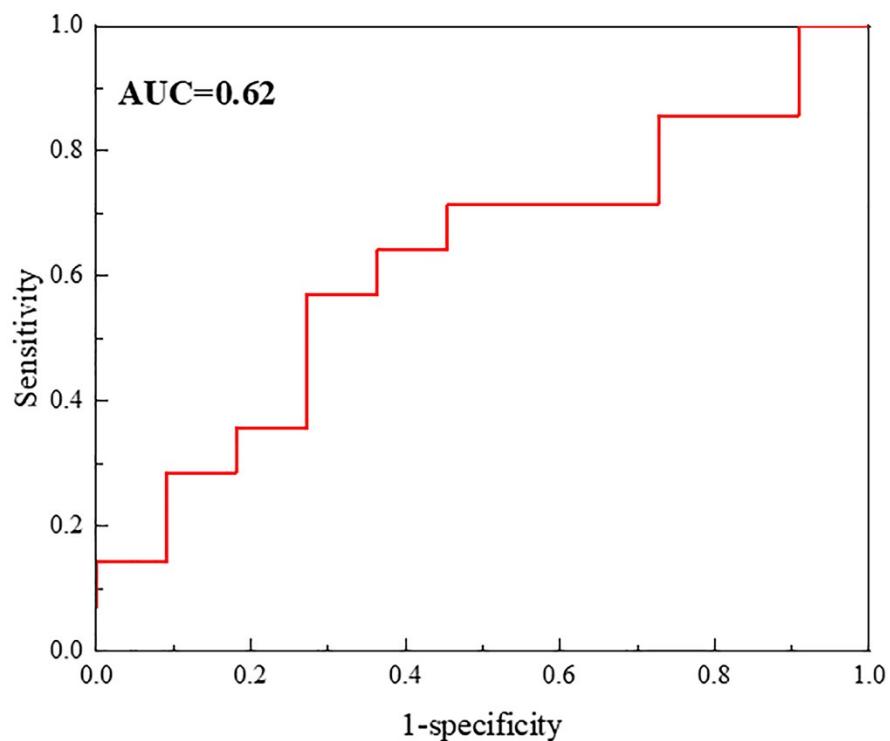


Fig. S2 ROC curve for COSMO-RS methodology.

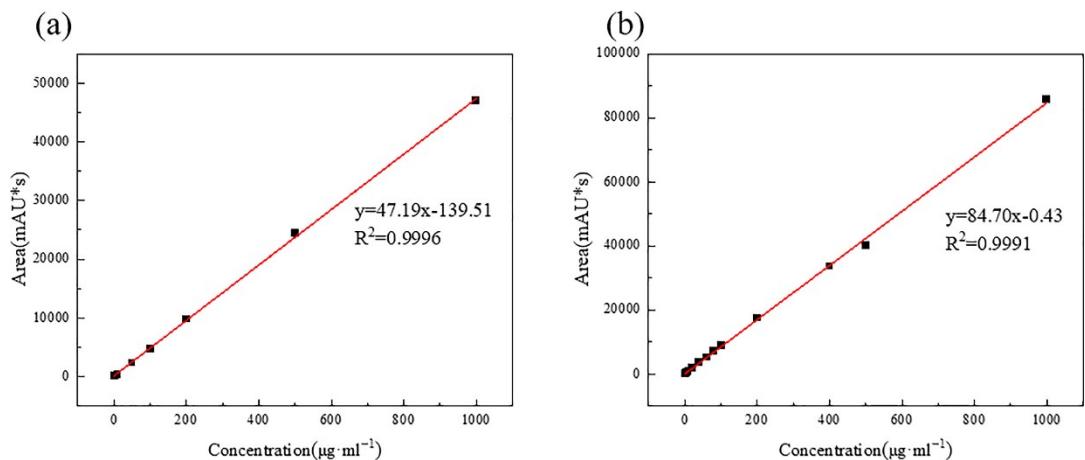


Fig. S3 The HPLC standard curve of pure EG and 2MIE: (a) EG; (b) 2MIE.

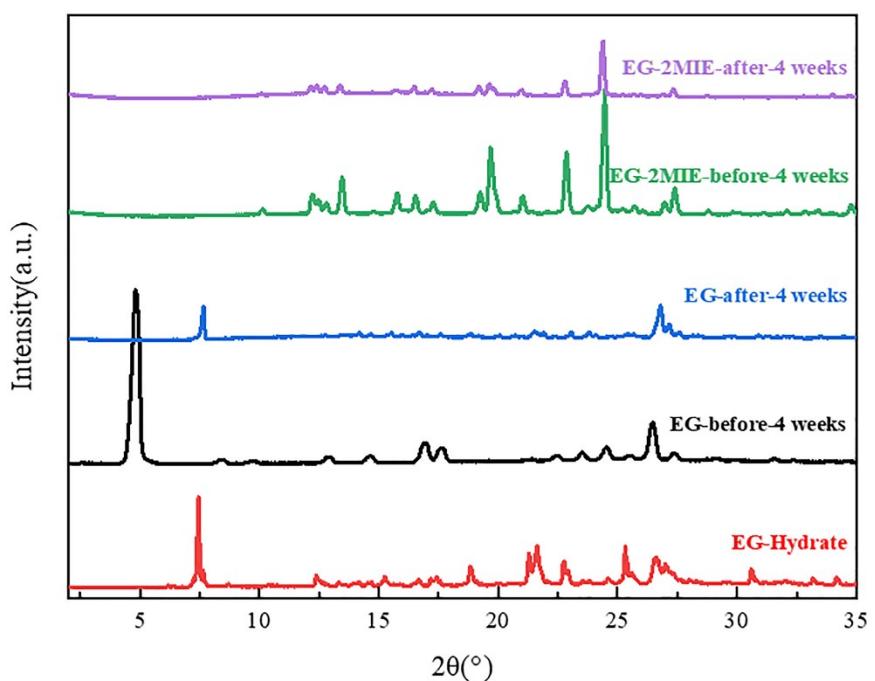


Fig. S4 PXRD patterns of EG-2MIE and EG under accelerated conditions ($40\text{ }^\circ\text{C}$, 75% RH) for 4 weeks.

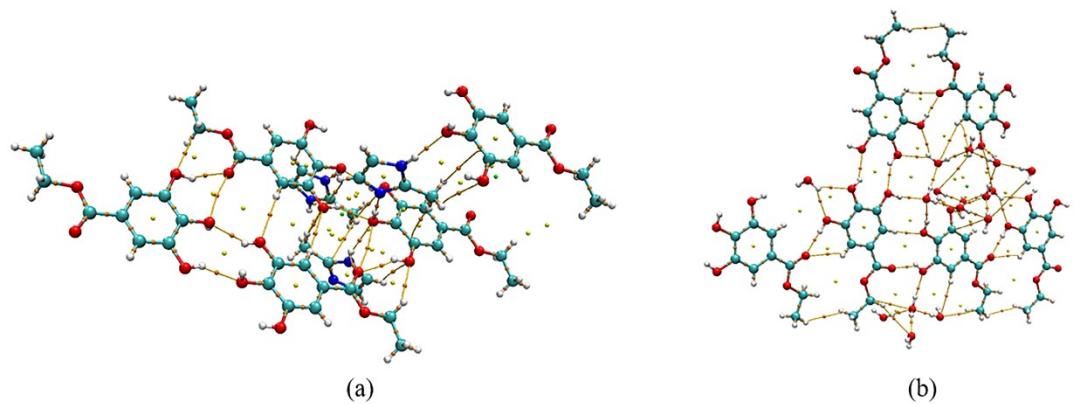


Fig. S5 The topological geometric structures of EG-2MIE and EG-Hydrate, with purple, orange, yellow, and green spheres corresponding to the $(3, -3)$, $(3, -1)$, $(3, +1)$, and $(3, +3)$ critical points of electron density (nuclear critical point, bond critical point, ring critical point, and cage critical point), respectively, the orange curve corresponds to the key path.

Table S1. The excess enthalpy (ΔH_{ex}) values and PXRD result of EG and the coformers.

API	Coformer	$\Delta H_{\text{ex}}(\text{kcal}\cdot\text{mol}^{-1})$	New form
EG	piperazine	-5.89	Yes
	triethylenediamine	-5.78	Yes
	nizatidine	-4.94	No
	cytosine	-4.60	Yes
	1-ethyl-2, 3-dioxopiperazine	-4.49	Yes
	palbociclib	-4.39	No
	4-aminobutyric acid	-3.83	Yes
	2-Imidazolidone	-3.50	No
	carbamazepine	-3.12	Yes
	1, 3, 5-triazine-2, 4, 6-triamine	-3.10	Yes
	2-methylimidazole	-3.00	Yes
	glutamine	-2.93	No
	urea	-2.87	Yes
	taurine	-2.61	No
	niacinamide	-2.58	Yes
	lansoprazole	-2.49	No
	2-aminopyrimidine	-2.39	No
	picolinamide	-2.04	No
	benzimidazole	-1.91	Yes
	theophylline	-1.80	Yes
	serine	-1.69	No
	mannitol	-1.37	No
	2-aminonicotinicacid	-1.26	Yes

API	Coformer	$\Delta H_{\text{ex}}(\text{kcal}\cdot\text{mol}^{-1})$	New form
	N-acetylglycine	-1.10	Yes
	oleanolic acid	-1.09	No

Table S2. Hydrogen bond parameters of EG-2MIE and EG-Hydrate.

D–H \cdots A	D–H (Å)	H \cdots A (Å)	D \cdots A (Å)	D–H \cdots A (°)	Symmetry
EG-2MIE					
O ₅ –H ₅ \cdots O ₉	0.66	2.18	2.6575(15)	131	
O ₁₀ –H ₁₀ \cdots O ₄	0.65	2.14	2.6457(14)	136	
O ₉ –H ₉ \cdots N ₄	0.82	1.91	2.6709(15)	154	
O ₄ –H ₄ \cdots N ₂	0.82	1.90	2.7062(18)	166	
O ₈ –H ₈ \cdots O ₇	0.84	2.01	2.8067(15)	158	1 + x, y, z
C ₁₀ –H _{10A} \cdots O ₈	0.97	2.59	3.2176(18)	123	-1 + x, y, z
O ₃ –H ₃ \cdots O ₂	0.88	1.91	2.7615(15)	163	-1 + x, y, z
N ₃ –H _{3A} \cdots O ₄	0.88	1.88	2.7431(16)	166	1 - x, 1 - y, 2 - z
N ₁ –H ₁ \cdots O ₉	0.88	1.93	2.7380(2)	152	1 - x, 1 - y, 1 - z
C ₁ –H _{1B} \cdots O ₅	0.96	2.51	3.4410(3)	163	1 - x, 1 - y, 1 - z
EG-Hydrate					
O ₃ –H ₃ \cdots O ₁₃	0.84	1.83	2.6647(11)	175	
O ₄ –H ₄ \cdots O ₁₂	0.84	1.95	2.7307(12)	153	
O ₅ –H ₅ \cdots O ₇	0.84	1.97	2.6486(12)	137	
O ₈ –H ₈ \cdots O ₁₂	0.84	1.86	2.6902(12)	170	
O ₉ –H _{9A} \cdots O ₈	0.84	2.08	2.7887(12)	142	1 - x, -y, 2 - z
O ₁₀ –H ₁₀ \cdots O ₂	0.84	1.81	2.6539(12)	178	-1 + x, y, 1 + z
O ₁₁ –H _{11C} \cdots O ₅	0.87	1.87	2.7358(11)	173	
O ₁₁ –H _{11D} \cdots O ₁₃	0.87	2.02	2.8643(12)	164	1 - x, 1 - y, 1 - z
O ₁₂ –H _{12A} \cdots O ₁₄	0.87	1.91	2.7805(12)	174	2 - x, -y, 1 - z
O ₁₂ –H _{12B} \cdots O ₁₅	0.87	1.97	2.8250(12)	168	1 - x, -y, 1 - z
O ₁₃ –H _{13A} \cdots O ₁₅	0.87	1.88	2.7475(12)	177	
O ₁₃ –H _{13B} \cdots O ₁₄	0.87	1.90	2.7687(12)	174	
O ₁₄ –H _{14A} \cdots O ₁₁	0.87	1.92	2.7853(12)	178	2 - x, 1 - y, 1 - z
O ₁₄ –H _{14B} \cdots O ₉	0.87	2.23	2.9704(12)	142	1 + x, y, -1 + z
O ₁₄ –H _{14B} \cdots O ₁₀	0.87	2.11	2.8528(12)	143	1 + x, y, -1 + z

D–H···A	D–H (Å)	H···A (Å)	D···A (Å)	D–H···A (°)	Symmetry
O ₁₅ –H _{15A} ···O ₁₁	0.87	1.89	2.7616(12)	178	x, -1 + y, z
O ₁₅ –H _{15B} ···O ₃	0.87	2.05	2.8892(12)	161	2 – x, –y, 1 – z
C ₁₄ –H ₁₄ ···O ₄	0.95	2.48	3.4226(14)	172	

Table S3. C–H···π weak interaction information of EG-2MIE and EG-Hydrate.

C–H···Cg _i *	H···Cg _i (Å)	C–H···Cg _i (°)	Symmetry
EG-2MIE			
C ₉ –H _{9A} ···Cg ₄	2.94	118	1 – x, 2 – y, 2 – z
EG-Hydrate			
C ₂ –H _{2A} ···Cg ₂	2.77	153	1 – x, 1 – y, 1 – z

*EG-2MIE: Cg₄=N₃–C₆–N₄–C₇–C₈; EG-Hydrate: Cg₂=C₁₃–C₁₄–C₁₅–C₁₆–C₁₇–C₁₈.

Table S4. π···π weak interaction information of EG-2MIE and EG-Hydrate.

Cg _i ···Cg _j	Cg _i ···Cg _j (Å)	Dihedral Angle between Cg _i and Cg _j (°)	Slippage (Å)	Symmetry
EG-2MIE				
Cg ₄ ···Cg ₄	3.68	0	0.528	1 – x, 1 – y, 2 – z
EG-Hydrate				
Cg ₁ ···Cg ₁	3.50	0	1.159	2 – x, 1 – y, 1 – z

*EG-2MIE: Cg₄=N₃–C₆–N₄–C₇–C₈; EG-Hydrate: Cg₁=C₄–C₅–C₆–C₇–C₈–C₉.

Table S5. Experimental solid-liquid equilibrium data (mass fraction) of EG-2MIE-EtOH system at 298.15 K, 303.15 K and 308.15 K.

T(K)	Mass fraction			
	EG	2MIE	Ethanol	Equilibrium solid phase
298.15	0.6232	0.0000	0.3768	EG
	0.5897	0.0068	0.4035	EG
	0.4793	0.0109	0.5098	EG + Cocrystal
	0.3161	0.0115	0.6724	Cocrystal
	0.2844	0.0149	0.7007	Cocrystal
	0.1687	0.0202	0.8111	Cocrystal
	0.1016	0.0294	0.8690	Cocrystal

T(K)	Mass fraction			Equilibrium solid phase
	EG	2MIE	Ethanol	
303.15	0.0736	0.0352	0.8912	Cocrystal
	0.0312	0.0566	0.9122	Cocrystal
	0.0118	0.0852	0.9030	Cocrystal
	0.0104	0.1249	0.8647	Cocrystal
	0.0108	0.4290	0.5602	2MIE + Cocrystal
	0.0056	0.4706	0.5238	2MIE
	0.0000	0.5272	0.4728	2MIE
	0.6396	0.0000	0.3604	EG
	0.5629	0.0076	0.4295	EG
	0.5168	0.0107	0.4725	EG + Cocrystal
	0.4256	0.0114	0.5630	Cocrystal
	0.3370	0.0124	0.6506	Cocrystal
	0.2183	0.0191	0.7626	Cocrystal
	0.1697	0.0235	0.8068	Cocrystal
	0.1046	0.0362	0.8592	Cocrystal
	0.0542	0.0546	0.8912	Cocrystal
308.15	0.0334	0.0834	0.8832	Cocrystal
	0.0153	0.1511	0.8336	Cocrystal
	0.0160	0.4688	0.5152	2MIE + Cocrystal
	0.0121	0.5016	0.4863	2MIE
	0.0000	0.6099	0.3901	2MIE
	0.7993	0.0000	0.2007	EG
	0.6722	0.0079	0.3199	EG
	0.5250	0.0143	0.4607	EG + Cocrystal
	0.4528	0.0108	0.5364	Cocrystal
	0.4073	0.0138	0.5789	Cocrystal
	0.3363	0.0143	0.6494	Cocrystal
	0.2622	0.0129	0.7249	Cocrystal
	0.1224	0.0234	0.8542	Cocrystal
	0.0686	0.0504	0.8810	Cocrystal
	0.0262	0.1028	0.8710	Cocrystal
	0.0125	0.2102	0.7773	Cocrystal

T(K)	Mass fraction			Equilibrium solid phase
	EG	2MIE	Ethanol	
0.0089	0.4850	0.5061		2MIE + Cocrystal
0.0049	0.5388	0.4563		2MIE
0.0000	0.6922	0.3078		2MIE