Electronic Supporting Information for:

Mechanochemical Syntheses and ³⁵Cl NMR Crystallography

of Ionic Cocrystals of Phenothiazine Drugs

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Scheme S1. Molecular structures of (A) oxalic acid, (B) adipic acid, (C) succinic acid, (D) fumaric acid, (E) malonic acid, (F) glutaric acid, (G) terephthalic acid, and (H) malic acid.

	Ptz	Cpz	Pmz	Ptz ₂ O	Ptz ₂ S	Ptz ₂ F	Ptz ₂ A
No. Scans	16384	24576	2048	29888	15360	8192	131072
Experimental Time (hr)	4.55	10.24	2.28	16.60	4.27	4.55	18.20
Recycle delay (s)	1.0	1.5	4.0	2.0	1.0	2.0	0.5
No. echoes	60	n/a	30	20	125	125	n/a
Echo Length (µs)	600	600	350	500	200	200	100
Dwell Time (µs)	2	2	1.67	2	1	1	1
Spectral Width (kHz)	250	250	300	250	500	500	500
Time domain size	21452	1024	6960	10240	28118	28118	2236
Pulse width $[\pi/2]$ (µs)	5	2.5	1.665	5	1.665	1.665	3.5
¹ H decoupling field (kHz)	50	50	50	50	50	50	50

Table S1. Static ³⁵Cl SSNMR acquisition parameters for the QCPMG or Hahnecho experiments at 18.8 T

experiments at 18.8 T						
	Ptz ₂ G	PtzMi	PtzMo	Cpz ₂ S	Cpz ₂ F	CpzT
No. Scans	8192	12288	8192	40960	51200	8192
Experimental Time (hr)	1.14	6.83	1.00	8.53	7.11	2.28
Recycle delay (s)	0.5	2.0	1.0	0.75	0.5	1.0
No. echoes	30	30	125	20	50	25
Echo Length (µs)	500	400	200	400	400	300
Dwell Time (µs)	0.7	0.677	1	2	2	1.677
Spectral Width (kHz)	714	750	500	250	250	300
Time domain size	30000	19000	28118	5300	16000	5030
Pulse width $[\pi/2]$ (µs)	1.66	5	1.665	5	5	3.33
¹ H decoupling field (kHz)	50	50	50	50	50	50

Table S2. Static ³⁵Cl SSNMR acquisition parameters for the QCPMG experiments at 18.8 T

experiments at 7.1 1							
	Ptz	Cpz	Pmz	Ptz ₂ O	Ptz ₂ S	Ptz ₂ F	Ptz ₂ A
No. Scans	2160	16384	10240	2160	10952	58400	10800
Experimental Time (hr)	3.0	18.2	11.4	3	1.5	16.2	6
Recycle delay (s)	5.0	4.0	4.0	5.0	0.5	1.0	2.0
No. echoes	50	20	10	50	40	10	8
Echo Length (μs)	400	400	250	400	200	150	200
Dwell Time (µs)	0.5	0.5	0.5	1	0.5	0.5	0.5
Spectral Width (kHz)	1000	1000	1000	500	1000	1000	1000
Time domain size	51768	21048	7258	25884	25884	5558	5360
WURST pulse length (µs)	50	50	50	50	50	50	50
³⁵ Cl WURST pulse rf	15.9	15.9	15.9	15.9	15.9	15.9	15.9
(kHz)							
Sweep width of WURST	500	500	500	500	500	500	500
pulse (kHz)							
¹ H decoupling field (kHz)	33.7	36.9	33.7	33.7	36.9	42.6	36.9

Table S3. Static ³⁵Cl SSNMR acquisition parameters for the WURST-CPMG experiments at 9.4 T

experiments at 9.4 1						
	Ptz ₂ G	PtzMi	PtzMo	Cpz ₂ S	Cpz ₂ F	CpzT
No. Scans	25000	5400	20480	10240	40960	8192
Experimental Time (hr)	6.9	3.0	2.8	5.7	5.7	2.28
Recycle delay (s)	1.0	2.0	0.5	2.0	2.0	1.0
No. echoes	50	50	50	20	25	40
Echo Length (μs)	150	400	400	200	400	500
Dwell Time (µs)	0.5	0.5	0.5	0.67	0.5	0.5
Spectral Width (kHz)	1000	1000	1000	750	1000	1000
Time domain size	26518	51678	51768	3200	26128	49628
WURST pulse length (µs)	50	50	50	50	50	50
³⁵ Cl WURST pulse rf	13.0	13.0	14.5	14.5	17.2	14.5
(kHz)						
Sweep width of WURST	500	500	500	500	500	500
pulse (kHz)						
¹ H decoupling field (kHz)	50	33.7	26.1	51.8	39.9	30.2

Table S4. Static ³⁵Cl SSNMR acquisition parameters for the WURST-CPMG experiments at 9.4 T

Table S5. MAS ³⁵ Cl SSN	MR acqu	uisition p	arameter	s for the	Hahn-ecl	no experir	nents at 18	3.8 T	
	Cpz	Ptz ₂ F	Ptz ₂ S	Ptz ₂ O	Ptz ₂ A	PtzMi	Cpz ₂ S	Cpz ₂ F	CpzT
No. Scans	8192	49152	65536	3600	14400	3600	16384	16384	1800
Experimental Time (hr)	2.3	13.7	9.1	1	4	0.5	4.6	4.6	0.5
Recycle delay (s)	1.0	1.0	0.5	1	1	0.5	1.0	1.0	1.0
Dwell Time (µs)	2.5	0.5	0.5	1.7	0.5	1.7	2.5	2.5	1.7
Spectral Width (kHz)	200	1000	1000	300	1000	300	200	200	300
Time domain size	2048	1024	1024	1024	1024	1024	2048	2048	1024
Pulse width $[\pi/2]$ (µs)	3	1.7	1.7	2.5	1.7	2.5	3	3	500
¹ H decoupling field	50	30	30	30	30	30	50	50	30
(kHz)									
Spinning Speed (khz)	16	55	55	18	55	18	16	16	15

for spectra acquired at 14	.I T						
	Ptz	Cpz	Pmz	Ptz ₂ O	Ptz ₂ S	Ptz ₂ F	Ptz ₂ A
No. Scans	2048	4096	500	2048	1024	2048	4096
Experimental Time (hr)	2.28	3.41	0.97	2.28	2.56	3.41	3.41
Recycle delay (s)	4.0	3.0	7.0	4.0	9.0	6.0	3.0
Contact Time (ms)	3.0	1.0	3.0	1.0	3.0	3.0	1.0
¹ H Hartmann-Hahn	50	50	50	50	50	50	50
matching field (kHz)							
Dwell Time (µs)	5	5	5	5	5	5	11
Spectral Width (kHz)	100	100	100	100	100	100	45
Time domain size	4096	4096	4096	4096	4096	4096	2048
¹ H decoupling field (kHz)	50	50	50	50	50	50	50

Table S6. ¹H \rightarrow ¹³C{¹H} CP/MAS SSNMR (v_{rot} = 10 kHz) acquisition parameters for spectra acquired at 14.1 T

for spectra acquired at 14	.I T					
	Ptz ₂ G	PtzMi	PtzMo	Cpz ₂ S	Cpz ₂ F	CpzT
No. Scans	10240	2048	2048	4096	1024	2048
Experimental Time (hr)	11.38	1.71	2.28	4.55	2.84	1.71
Recycle delay (s)	4.0	3.0	4.0	4.0	10.0	3.0
Contact Time (ms)	2.0	1.0	1.0	2.5	1.0	1.0
¹ H Hartmann-Hahn	50	50	50	50	50	50
matching field (kHz)						
Dwell Time (µs)	5	5	5	5	5	5
Spectral Width (kHz)	100	100	100	100	100	100
Time domain size	8192	4096	4096	4096	4096	4096
¹ H decoupling field (kHz)	50	50	50	50	50	50

Table S7. ¹H \rightarrow ¹³C{¹H} CP/MAS SSNMR ($v_{rot} = 10 \text{ kHz}$) acquisition parameters for spectra acquired at 14.1 T

Table S8. Mechanoch	emical syntheses that fai	led to produce a Pmz coo	crystal.
Target Cocrystal	Milling Time	Solvent/Amount	Jar Material
	(minutes)/Milling	added (µL)	
	Frequency (Hz)		
Pmz ₂ Adipic	30/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Adipic	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Adipic	270/30	Ethanol/15	Teflon
Pmz ₂ Oxalic	90/30	Neat	Stainless Steel
Pmz ₂ Oxalic	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Malonic	90/30	Neat	Stainless Steel
Pmz ₂ Malonic	90/30	Acetonitrile/20	Stainless Steel
Pmz ₂ Malic	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Glutaric	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Fumaric	30/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Fumaric	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Fumaric	180/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Fumaric	180/35	Acetonitrile/10	Stainless Steel
Pmz ₂ Fumaric	180/35	Acetonitrile/10	Teflon
Pmz ₂ Fumaric	297/30	Ethanol/15	Teflon
Pmz ₂ Succinic	30/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Succinic	90/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Succinic	180/30	Acetonitrile/10	Stainless Steel
Pmz ₂ Succinic	180/35	Acetonitrile/10	Stainless Steel
Pmz ₂ Succinic	180/35	Acetonitrile/10	Teflon
Pmz ₂ Succinic	297/30	Ethanol/15	Teflon
Pmz ₂ Terepthalic	90/30	Acetonitrile/15	Stainless Steel
Pmz ₂ Terepthalic	180/35	Acetonitrile/10	Stainless Steel
Pmz ₂ Terepthalic	180/35	Acetonitrile/10	Teflon



Fig. S1. PXRD patterns of promethazine HCl, chlorpromazine HCl, and promazine HCl.



Fig. S2 Comparisons of calculated and experimental PXRD patterns of promethazine HCl cocrystals.



Fig. S3. Comparisons of calculated and experimental PXRD patterns of chlorpromazine HCl cocrystals.



Fig. S4. PXRD patterns of promethazine HCl and the novel Ptz cocrystals.



Fig. S5. PXRD patterns of chlorpromazine HCl and the novel Cpz cocrystals.

Table S9. Summary of results	from SCXRD analysis of novel (Cpz cocrystals.
	Cpz ₂ F	Cpz ₂ S
Chemical formula	C ₁₇ H ₂₀ ClN ₂ S 0.5(C ₄ H ₄ O ₄)	C ₁₇ H ₂₀ ClN ₂ S 0.5(C ₄ H ₆ O ₄)
Formula weight	413.34	414.35
Temperature/K	170.00	170.00
Crystal system	monoclinic	monoclinic
Space group	C2/c	C2/c
a/Å	26.9241(10)	26.8901(13)
b/Å	7.5422(3)	7.5402(4)
c/Å	21.1043(8)	21.1999(9)
α/°	90	90
β/°	106.9090(10)	106.365(2)
$\gamma/^{\circ}$	90	90
Volume/Å ³	4100.3(3)	4124.3(3)
Ζ	8	8
$ ho_{calc}/g \ cm^{-3}$	1.339	1.335
µ/mm ⁻¹	0.434	0.432
F(000)	1728.0	1736.0
Crystal size/mm ³	$0.326 \times 0.136 \times 0.121$	$0.345 \times 0.293 \times 0.243$
Radiation	Mo $K\alpha$ ($\lambda = 0.71073$)	Mo $K\alpha$ ($\lambda = 0.71073$)
2\Overlap range for data collection/°	5.806 to 50.852	5.824 to 55.126
	$-32 \le h \le 32$	$-34 \le h \le 34$
Index ranges	$-9 \le k \le 9$	$-9 \le k \le 9$
	$-25 \le l \le 25$	$-27 \le l \le 26$
Reflections collected	36003	75505
In daman dant nefte stiens	$3774 [R_{int} = 0.0460, R_{sigma} =$	4747 [$R_{int} = 0.0286, R_{sigma} =$
Independent reflections	0.0217]	0.0117]
Data/restraints/parameters	3774/0/246	4747/0/246
Goodness-of-fit on F ²	1.052	1.074
Final <i>R</i> indexes $[I \ge 2\sigma(I)]$	$R_1 = 0.0303, wR_2 = 0.0667$	$R_1 = 0.0305, wR_2 = 0.0737$
Final <i>R</i> indexes [all data]	$R_1 = 0.0405, wR_2 = 0.0715$	$R_1 = 0.0352, wR_2 = 0.0765$
Largest diff. peak/hole / e Å ⁻³	0.25/-0.24	0.34/-0.26



Fig. S6. ORTEP diagrams of (A) Cpz₂S and (B) Cpz₂F along the crystallographic *c*-axis.



Fig. S7. ¹H \rightarrow ¹³C{¹H} CP/MAS spectra (v_{rot} = 10 kHz) of **Cpz** acquired at 14.1 T compared to those of novel cocrystals. Asterisks denote isotropic chemical shifts, all other peaks are spinning sidebands.



Fig. S8. ${}^{1}\text{H} \rightarrow {}^{13}\text{C} \{{}^{1}\text{H}\}$ CP/MAS spectra ($v_{rot} = 10 \text{ kHz}$) of **Ptz** acquired at 14.1 T compared to those of novel cocrystals. Asterisks denote isotropic chemical shifts, all other peaks are spinning sidebands.



Fig. S9. ¹H \rightarrow ¹³C{¹H} CP/MAS spectra ($v_{rot} = 10 \text{ kHz}$) of **Ptz** acquired at 14.1 T compared to those of reported cocrystals. Asterisks denote isotropic chemical shifts, all other peaks are spinning sidebands.

reak Ass	signments fo	or Ptz and al	I peaks due 1	to Ptz in co	ocrystals ((ppm)
		cı⊖	10 H ⊕N 8	10		
	4	~ ⁵ ~6	7 _N6	9 5 4		
	Ĭ		Ĭ	Ī		
	3、	2 1	_s1≈	$>_2 - 3$		
Carbon	3、 1-4 ª	2 5 "	<u>S</u>	<u>2</u> 7-8 "	9	10 ^b
Carbon Ptz	1-4 ^{<i>a</i>} 133-122	$\frac{2}{\frac{5^{a}}{119}}$	1 6 ^b 149-143	2 7-8 <i>a</i> 62-59	9 14	10 ^{<i>b</i>} 48-42
Carbon Ptz Ptz ₂ O	1-4 ^{<i>a</i>} 133-122 131-123	2 5 a 119 119	1 <u>6 ^b</u> 149-143 149-144	2 7-8 ^{<i>a</i>} 62-59 62-59	9 14 15	10 ^{<i>b</i>} 48-42 46-42
Carbon Ptz Ptz ₂ O Ptz ₂ S	1-4 <i>a</i> 133-122 131-123 129-123	2 5 " 119 119 119 119	1≈ <u>6 ^b</u> 149-143 149-144 147-144	2 7-8 " 62-59 62-59 59-52	9 14 15 10	10 ^b 48-42 46-42 42-37
Carbon Ptz Ptz ₂ O Ptz ₂ S Ptz ₂ F	1-4 <i>a</i> 133-122 131-123 129-123 129-123	2 5 " 119 119 119 119 120	1 = 1 = 1 = 1 = 1 = 1 = 1 = 1 = 1 = 1 =	2 7-8 " 62-59 62-59 59-52 59-52	9 14 15 10 8	10 ^b 48-42 46-42 42-37 49-42
Carbon Ptz Ptz ₂ O Ptz ₂ S Ptz ₂ F Ptz ₂ A	1-4 <i>a</i> 133-122 131-123 129-123 129-123 133-124	2 5 " 119 119 119 120 119-115	1 = 6 ^b 149-143 149-144 147-144 147-143 151-144	2 7-8 " 62-59 62-59 59-52 59-52 59-50	9 14 15 10 8 9	10 ^{<i>b</i>} 48-42 46-42 42-37 49-42 42
Carbon Ptz Ptz ₂ O Ptz ₂ S Ptz ₂ F Ptz ₂ A Ptz ₂ G	1-4 <i>a</i> 133-122 131-123 129-123 129-123 133-124 131-122	2 5 " 119 119 119 120 119-115 118	1 S 6 ^b 149-143 149-144 147-144 147-143 151-144 147-140	2 7-8 <i>a</i> 62-59 62-59 59-52 59-52 59-52 59-50 61-58	9 14 15 10 8 9 10-8	10 ^b 48-42 46-42 42-37 49-42 42 42
Carbon Ptz Ptz ₂ O Ptz ₂ S Ptz ₂ F Ptz ₂ A Ptz ₂ G PtzMi	1-4 <i>a</i> 133-122 131-123 129-123 129-123 133-124 131-122 136-124	2 5 " 119 119 119 120 119-115 118 119	1≈ 6 ^b 149-143 149-144 147-144 147-143 151-144 147-140 147-144	7-8 " 62-59 62-59 59-52 59-52 59-50 61-58 60	9 14 15 10 8 9 10-8 11	10 ^b 48-42 46-42 42-37 49-42 42 42 42 43

Table S10. ¹H \rightarrow ¹³C{¹H} CP/MAS SSNMR ($v_{rot} = 10 \text{ kHz}, B_0 = 14.1 \text{ T}$)

^a Carbons 1-4 and 7-8 are too similar to accurately distinguish from another using CP/MAS SSNMR or high level DFT calculations. ^b Has multiple peaks due to unique environments in the crystal lattice

Table S11. ¹H \rightarrow ¹³C{¹H} CP/MAS SSNMR (v_{rot} = 10 kHz, B_0 = 14.1 T) Peak Assignments for Cpz and all peaks due to Cpz in cocrystals (ppm)



^a Carbons 1-4, 9-12, 6/7, 5/8, 13/15 are too similar to accurately distinguish from another using CP/MAS SSNMR or high level DFT calculations.

^b Has multiple peaks due to unique environments in the crystal lattice

Table S12. ¹H \rightarrow ¹³C{¹H} CP/MAS SSNMR ($v_{rot} = 10 \text{ kHz}, B_0 = 14.1 \text{ T}$) Peak Assignments for coformers and all peaks due to coformers in cocrystals (ppm)

O OF	ł		0	
но О	HO	2 3 1	H0 ⁻¹ _2-'	
OH O				
	2. OH	,	O II	O II
HO 2		0 0	HO_1_2_3_	2ОН
	0			
O 3=	=3 OH	HO 2 OI	H ((рн
1-2	2-1		н0_1_2_;	B OH
НÓ 3-	-3 O		110 2	ii O
Carbon	1	2	3	4
Oxalic Acid	164	-	-	-
Ptz_2O	165	-	-	-
Malonic Acid	175	41	-	-
PtzMo	168	36	-	-
Succinic Acid	180	29	-	-
Ptz_2S	174	30	-	-
Cpz_2S	173	41	-	-
Fumaric Acid	173	137	-	-
Ptz_2F	167	124	-	-
Cpz_2F	165	133	-	-
Malic Acid	181	41	69	180
PtzMi	176	40	67	176
Glutaric Acid	181	34	19	-
Ptz_2G	174-175	36	23	-
Adipic Acid	183	35	24	-
Ptz_2A	176	36	23	-
Terephthalic Acid	160	132	120	-
CpzT	173	*	*	-

*These carbons overlap with peaks in the same range as Cpz and we cannot accurately attribute individual peaks to Terephthalic acid



Fig. S10. ³⁵Cl EFG tensor orientations for the geometry-optimized structures of (A) **Ptz**, (B) **Pmz**, and (C) **Cpz**.



Fig. S11. ³⁵Cl EFG tensor orientations for the geometry-optimized structures of (A) Ptz₂F, (B) Ptz₂S, (C) Ptz₂A, and (D) Ptz₂O.



Fig. S12. ³⁵Cl EFG tensor orientations for the geometry-optimized structures of (A) **Cpz₂F** and (B) **Cpz₂S**.



Scheme S2. Bonding motifs of malonic acid within various crystal structures. (A) a dimer as found in bulk phase malonic acid (CSD Reference Code: MALNAC02), (B) a dimer as found in a picolinamide-malonic acid cocrystal (CSD Reference Code: HOGGUH), (C) a single carbonyl-acid interaction as found in a quinine-malonic acid salt (CSD Reference Code: VAMFOI), and (D) a double carbonyl-acid interaction as found in a theophylline-malonic acid cocrystal (CSD Reference Code: XEJXAM). See Refs. 110-113.