

H/F Replacement in secondary alcohols of sydnones as examples of isostructural OH...O=C hydrogen bonded dimer structures

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1. Relevant secondary alcohol dimers from the literature

Table 1S. Secondary alcohols identified through CCDC search and dimeric –like structures identified

Index	CCD Ref.	DIMER	Fragment	ANG1	ANG2	DIST1	DIST2
				O-H...O >150°	O-H...O >150°	H...O <vdW	H...O <vdW
1	ADOHUY	YES	1	175.498	173.071	1.937	1.933
	ADUNAP	NO	1	173.506	168.773	1.923	1.917
2	ADUNAP	NO	2	173.506	151.576	1.923	2.031
	ADUNAP	NO	3	151.576	168.773	2.031	1.917
3	AKUFIX	NO	1	160.313	178.159	1.993	1.951
4	ARAFPY10	NO	1	158.146	154.245	1.947	1.991
5	AZELAV	NO	1	177.711	154.421	1.969	1.934
6	BARJOX	NO	1	163.534	154.156	2.006	2.024
7	BUQDIC	NO	1	160.662	160.103	2.009	2.09
8	BUVFAA	NO	1	169.063	158.963	2.033	1.913
9	CAQRUH	NO	1	164.587	173.054	2.012	1.948
10	CEHDAX	NO	1	152.591	172.274	2.036	2.008
11	COBQEP	NO	1	158.045	167.558	2.036	1.982
	COKKIY		1	170.713	159.461	2.088	2.02
12	COKKIY		2	170.62	159.461	2.051	2.02
	COKKIY	YES	3	170.62	170.713	2.051	2.088
13	CUQJII	NO	1	173.469	169.002	1.948	1.987
	DUNFEY	NO	1	171.677	155.669	1.922	2.095
	DUNFEY	NO	2	154.725	171.677	2.04	1.922
14	DUNFEY	NO	3	154.725	155.669	2.04	2.095
	EBOPOE	NO	1	153.469	156.929	1.972	1.931
	EBOPOE	NO	2	155.407	153.469	2.241	1.972
	EBOPOE	NO	3	155.407	156.929	2.241	1.931
15	EDEMOR	YES	1	153.18	162.555	2.028	1.975
16	EGALAB	YES	1	172.271	178.141	1.942	1.923
17	ELEJUB01	YES	1	164.113	174.283	1.897	1.864
18	ELEKAI01	YES	1	179.295	177.757	1.83	1.83
19	EZEPEF	YES	1	152.996	150.364	2.105	2.179
	FELNIU	NO	1	168.236	168.524	1.927	1.89
20	FELNIU	NO	2	168.236	164.13	1.927	1.99
	FELNIU	NO	3	164.13	168.524	1.99	1.89
21	FIBMIQ	NO	1	174.141	176.153	1.977	1.936
22	FIDSES	NO	1	154.691	177.031	1.999	2.104
23	FIQDOZ	YES	1	150.797	150.797	2.244	2.244
24	FLOCOS	NO	1	154.383	154.383	2.21	2.21
25	FUHYAL	YES	1	168.845	178.438	2.022	1.992
26	GEGRAP	NO	1	166.971	158.007	2.001	2.108
27	GEJPUH	NO	1	166.812	161.227	2.049	2.049
28	GOGQEZ	NO	1	164.571	163.096	2	2.064
29	HBMHPO10	YES	1	153.947	156.697	2.316	2.182
30	HIJBK	NO	1	170.71	174.596	1.973	1.969
31	HIKWEG	YES	1	166.685	160.122	1.916	1.982
	HIWREM	NO	1	169.477	169.258	1.947	1.929
	HIWREM	NO	2	169.477	167.815	1.947	1.963
	HIWREM	NO	3	169.258	167.815	1.929	1.963

32	HIWREM	NO	4	163.33	169.258	2.004	1.929
	HIWREM	NO	5	163.33	167.815	2.004	1.963
	HIWREM	NO	6	163.33	169.477	2.004	1.947
	IBAQEL	NO	1	151.872	156.975	2.32	2.3
	IBAQEL	NO	2	151.872	155.538	2.32	2.269
	IBAQEL	NO	3	155.538	156.975	2.269	2.3
33	ILAROF	NO	1	162.603	158.672	1.973	1.977
34	IQONIO	NO	1	160.758	168.136	2.029	1.839
35	IRAGEP	NO	1	170.403	168.492	1.939	1.995
36	ISAFEQ	YES	1	171.085	161.617	1.953	2.08
37	JEVPAF	NO	1	169.297	170.141	1.952	1.946
38	JONZUJ	NO	1	154.333	154.582	2.121	2.157
39	JUXKAR	NO	1	150.414	162.309	2.288	2.027
40	KEBTUK	NO	1	173.742	166.247	1.969	2.253
41	KEXSIS	YES	1	178.354	170.954	2.035	2.23
42	KIHNOF	YES	1	176.458	164.305	1.871	1.927
43	KIWPAL	YES	1	159.895	158.1	1.993	2.012
	KIWPAL		2	152.897	159.895	2.022	1.993
	KIWPAL		3	152.897	158.1	2.022	2.012
	KIWPAL		4	152.897	150.345	2.022	2.07
	KIWPAL		5	150.345	159.895	2.07	1.993
	KIWPAL		6	150.345	158.1	2.07	2.012
44	KUPYOM	YES	1	172.655	170.026	2.069	2.048
45	LAGQES	NO	1	170.92	160.577	1.869	1.858
	LAGQES	NO	2	170.92	174.909	1.869	1.859
	LAGQES	NO	3	170.92	174.707	1.869	1.969
	LAGQES	NO	4	174.909	160.577	1.859	1.858
	LAGQES	NO	5	174.707	160.577	1.969	1.858
	LAGQES	NO	6	174.707	174.909	1.969	1.859
	LAPDEO	NO	1	173.678	169.447	1.906	1.911
	LAPDEO	NO	2	173.678	170.579	1.906	1.946
	LAPDEO	NO	3	169.447	170.579	1.911	1.946
46	LEDGEK	NO	1	172.205	155.546	1.957	2.107
47	LETNOR	NO	1	172.561	172.246	2.167	2.168
48	LIGVUT	NO	1	156.498	159.741	2.04	2.012
49	LISSUE	YES	1	168.183	167.225	2.033	2.056
50	LONVUI	NO	1	163.136	164.581	2.012	2.009
	LONVUI	NO	2	171.122	164.581	1.943	2.009
	LONVUI	NO	3	160.109	164.581	2.077	2.009
	LONVUI	NO	4	171.122	163.136	1.943	2.012
	LONVUI	NO	5	160.109	163.136	2.077	2.012
	LONVUI	NO	6	171.122	160.109	1.943	2.077
51	LUQHUD	NO	1	153.35	163.566	1.996	1.901
52	LUYHAR	NO	1	175.718	171.487	1.881	2.018
53	LUZJUL	YES	1	151.001	151.001	2.193	2.193
54	MEHNPQ	NO	1	173.418	173.488	1.965	1.956
55	MEQCEU	NO	1	163.359	164.122	1.95	2.19
56	MIDMUK	NO	1	172.647	168.294	2.188	2.097
57	MOJPUW	YES	1	153.171	157.485	2.116	2.211
58	MOYPUO	NO	1	158.19	168.659	1.963	2.017
59	NANFOB	NO	1	158.33	174.119	2.081	1.989
60	NOQKIO	NO	1	179.118	168.245	1.877	1.857

61	OGIXIO	NO	1	167.028	159.771	1.967	1.987
62	OHUKIO	NO	1	174.923	174.773	1.975	1.934
63	OMAFEP	YES	1	172.359	172.085	1.989	1.964
64	PAVQEI	NO	1	176.158	178.751	1.846	1.913
65	PESPOT	NO	1	168.206	175.445	2.148	2.088
66	QETZUL	NO	1	152.467	156.298	1.943	2.002
	QETZUL	NO	2	152.467	164.434	1.943	1.975
	QETZUL	NO	3	152.467	164.576	1.943	1.989
	QETZUL	NO	4	156.298	164.434	2.002	1.975
	QETZUL	NO	5	164.576	156.298	1.989	2.002
	QETZUL	NO	6	164.576	164.434	1.989	1.975
67	QOMBUS	NO	1	156.679	161.65	2.097	1.874
68	QULBII	NO	1	165.907	168.249	2.042	2.005
69	QUWZEQ	NO	1	150.846	159.578	2.194	2.154
70	RAGDEM	NO	1	172.31	165.237	2.086	2.031
71	REPDEX	YES	1	169.627	169.627	1.889	1.889
72	REXHEI	NO	1	172.477	169.539	2.002	2.071
73	RUVSOS	NO	1	175.306	175.847	1.898	2.108
74	RUVSOS01	NO	1	167.71	176.547	2.101	1.885
75	RUVVEL	NO	1	179.731	173.373	2.131	1.975
76	SAKJOG	NO	1	166.899	171.21	1.924	1.869
77	SAKKAT	NO	1	163.959	172.888	1.916	1.927
78	SAZRUK	NO	1	173.774	177.79	1.954	1.953
79	SEHLEA	NO	1	167.812	164.378	2.021	2.078
80	SELTEM	YES	1	173.587	175.25	1.97	1.862
81	SEWFAD	NO	1	162.661	162.661	2.038	2.038
82	SIMDUR	NO	1	162.888	159.881	2.208	2.395
83	SIMKAE	NO	1	159.653	166.154	1.996	2.019
84	SUQBIR	NO	1	172.554	178.027	1.959	2.066
85	SUQBUD	NO	1	174.527	176.254	1.974	2.252
86	TANFUJ	YES	1	152.333	150.407	2.031	2.153
87	TESTON10	NO	1	167.368	163.324	1.993	1.805
88	TUDSOD	NO	1	156.036	160.699	2.037	2.015
	TUDSOD	NO	2	159.683	160.699	2.059	2.015
	TUDSOD	NO	3	159.683	156.036	2.059	2.037
	UMAHAT	NO	1	158.59	157.93	1.954	1.959
	UMAHAT	NO	2	158.59	151.965	1.954	1.964
	UMAHAT	NO	3	151.965	157.93	1.964	1.959
	UMAHAT	NO	4	155.613	158.59	1.974	1.954
	UMAHAT	NO	5	155.613	157.93	1.974	1.959
	UMAHAT	NO	6	155.613	151.965	1.974	1.964
	UMAHAT	NO	7	152.498	158.59	1.975	1.954
	UMAHAT	NO	8	152.498	157.93	1.975	1.959
	UMAHAT	NO	9	152.498	151.965	1.975	1.964
	UMAHAT	NO	10	152.498	155.613	1.975	1.974
89	UTILUG	NO	1	156.55	169.581	2.108	1.869
90	UVEMOA	NO	1	167.727	167.064	2.067	2.037
91	UYOQAB	NO	1	151.568	158.035	2.038	2.042
92	UZIPIF	NO	1	155.614	160.608	2.084	2.002
	UZIPIF	NO	2	155.614	150.688	2.084	2.166
	UZIPIF	NO	3	150.688	160.608	2.166	2.002
93	VAHCOZ	NO	1	167.597	164.638	1.972	1.986

94	VAWMUF	YES	1	158.743	159.337	2.049	2.109
95	VEMNUA	NO	1	174.254	159.785	1.985	1.974
96	VIQQUK	NO	1	164.847	173.271	2.016	2.037
97	VUYYUJ	YES	1	165.327	166.589	1.908	1.902
98	WAKWIQ	NO	1	175.416	172.22	1.94	1.955
99	WAQYUI	NO	1	156.392	164.152	1.957	1.933
100	WEGLON	NO	1	166.39	167.065	1.942	1.942
101	WIMLEL	NO	1	171.104	163.447	2.022	2.043
102	WIPVOG	YES	1	166.45	173.316	2.142	1.99
103	XIDZES	NO	1	175.446	172.468	2.02	1.983
104	XIFSIP	NO	1	155.345	159.686	1.975	2.029
105	XODQAL	NO	1	151.498	154.53	2.112	2.067
106	YOFWEW	NO	1	155.35	172.702	1.808	1.646
107	YOWPAE	NO	1	179.44	177.965	1.955	2.148
108	ZEPZEB	NO	1	164.303	157.868	1.845	1.969
109	ZIBTEN	NO	1	150.853	157.537	2.058	2.013
110	ZIBTUB	NO	1	171.072	168.115	2.06	2.014
111	ZICZEQ	NO	1	168.909	164.796	1.705	1.831
112	ZOYMOP03	NO	1	170.911	170.671	2.068	1.912
113	ZUDKAM	NO	1	158.352	157.084	2.044	2.047
	ZUDKAM		2	158.352	157.802	2.044	2.052
	ZUDKAM		3	157.433	158.352	2.045	2.044
	ZUDKAM		4	157.433	157.084	2.045	2.047
	ZUDKAM		5	157.433	157.802	2.045	2.052
	ZUDKAM		6	157.084	157.802	2.047	2.052
114	ZUKRUS	NO	1	165.281	168.839	1.94	1.817
115	JOCVAC	YES	1	154.769	163.307	2.053	1.936
116	KUHRIR	NO					
117	TIMFAA	NO					
118	UFABUC	NO					
119	UHUDAD	NO					
120	UZIPEB	NO					
121	XEJAO	NO					
122	XEZJES	NO					
123	YIQMUJ	YES					
124	YOVOX	NO					

Identified separately from the search

2. IR spectroscopy revealing the O-H...O=C interaction

The ATR-IR spectra for compounds **Alc-1**, **Alc-2**, **Keto-1** and **Keto-2** show that the CO exocyclic group in the sydnone presents a redshift in the frequency from 1752-1771 cm^{-1} (Figure 1S) in ketones down to 1722 cm^{-1} for **Alc-2** and 1719 cm^{-1} **Alc-1**.

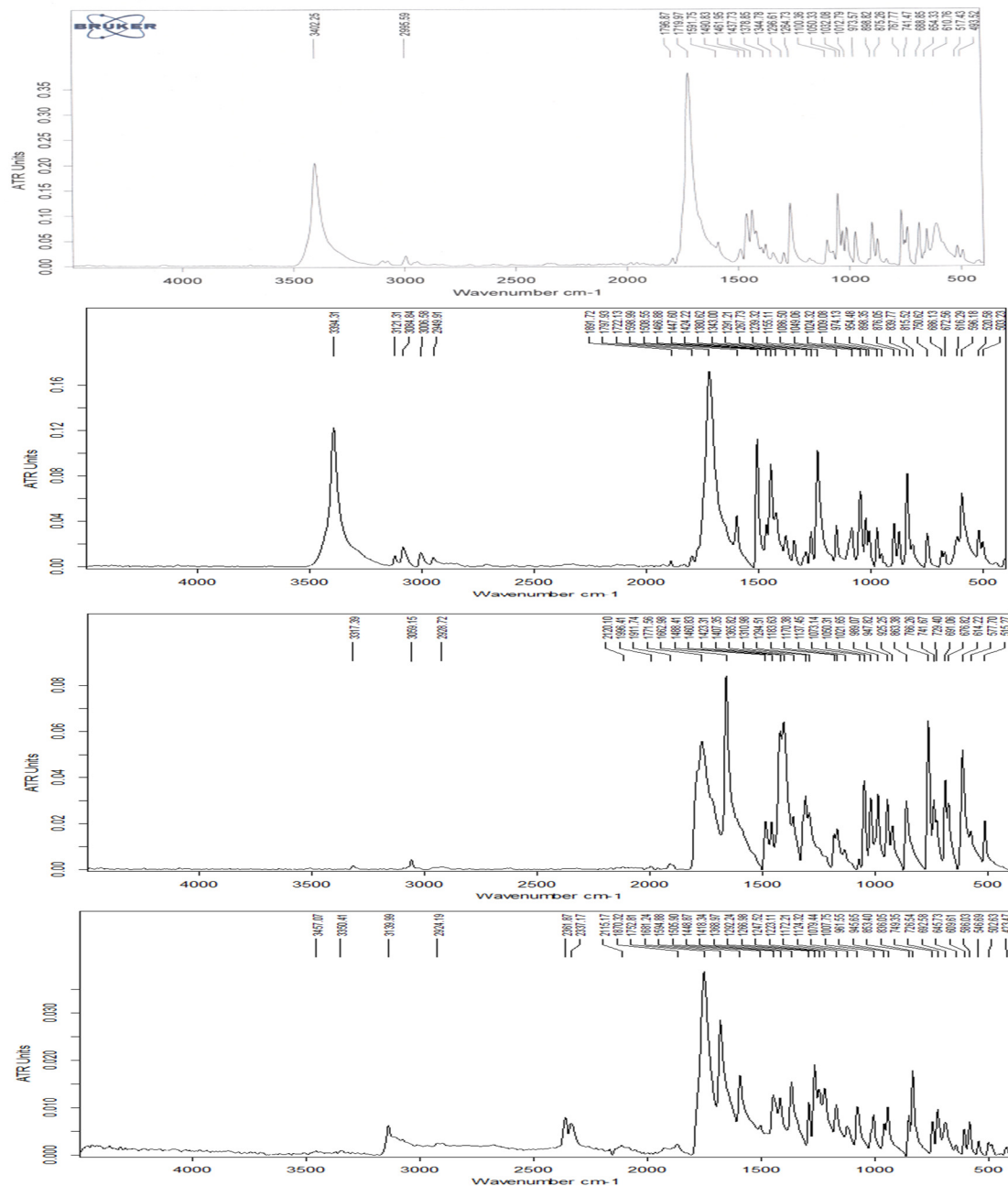
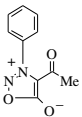
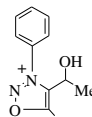
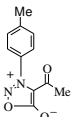
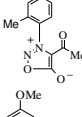
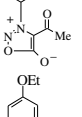
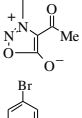
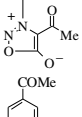
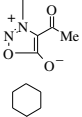
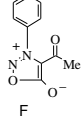
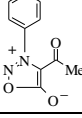


Figure 1S. ATR-IR Spectra of **Alc-1**, **Alc-2**, **Keto-1** and **Keto-2** (from top)

Table 2S also confirms such a trend with literature IR data. However, to gain more insight we performed DFT simulated spectra for **Alc-1** and **Keto-1** as models. For the purpose of the calculation we have employed the monomer of **Keto-1** and for **Alc-1** we have envisaged three scenarios: monomer extracted from the .cif file (optimized or used as is), the O-H...O=C H bonded dimer (optimized or used as is) and the most stable configuration presenting an intramolecular H-bond between the O-H and C=O groups of the sydnone.

Table 2S. Literature data-mining on the IR spectra of certain ketones and analogous alcohols

No.	Ketone	Other	CO	Alcohol	OH	CO
1		3060 1426 1053	1780 1662 1793 1665 1771*		3402*	1719*
2		2933 1509 1316 1050	1783 1686		3436	1770
3		N/A	1780		3418	1725
4		N/A	1780 1658		3412	1734
5		N/A	1792 1660		3442	1725
6		N/A	1790 1655		3406	1731
7		- N/A	1762 1664		3394	1728
8		N/A	1791 1734 1668		3340	1707
9		1752*			3394*	1722*

*Current study ATR-IR-Figure 1S

J. Chin. Chem. Soc. **1995**, 42, 987-989

J. Chin. Chem. Soc. **1992**, 39, 107-110

J. Org. Chem. **1983**, 48, 1382-1384

Bull. Chem. Soc. Jpn. **1972**, 45, 2944-2945

Synth. Commun. **1996**, 26, 2757-2764

Figure 2S presents the stacked simulated spectra. The band at 1791 cm^{-1} for the Keto-1 corresponds to the $1752\text{--}1771\text{ cm}^{-1}$ range for ATR IR spectra (**Keto-1** and **Keto-2**). All the alcohol configurations, in either monomer or dimeric forms, present the CO sydnone vibration at a lower frequency confirming that IR vibration is correlated with the H-bond. For the optimized monomer and the intramolecular H-bonded monomer the CO bond vibrates at a higher frequency due to more relaxed configuration obtained by optimization. For the dimer form (either optimized or not) and monomer (non-optimized) the CO band ranges from 1692.5 to 1703.7 cm^{-1} . All three structures adopt the configurations constrained by the crystalline packing. These bands are analogous to the 1722 cm^{-1} bands from the ATR IR spectra. The conclusion is that historically the H-bonds between OH and sydnone CO could be predicted by IR spectroscopy and are confirmed by X-ray diffraction analysis.

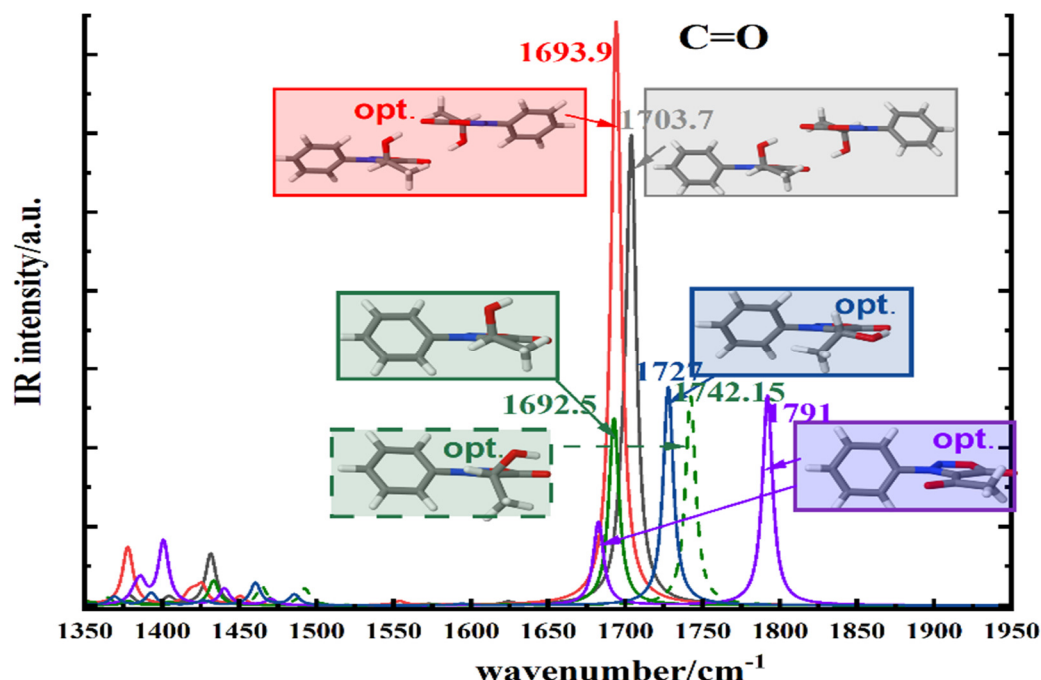


Figure 2S. Calculated IR spectra of **Alc-1** and **Keto-1**: **Alc1**- dimeric form as extracted from the X-Ray spectra (solid red) , **Alc1**-dimeric form as extracted from the X-ray spectra – optimized (solid dark gray), **Alc-1** monomer as extracted from X-ray spectra (solid green), **Alc-1**-monomer as extracted from X-Ray spectra – optimized (dash green), **Alc1**-monomer – proposed structure to form intramolecular H bond - optimized (solid blue), **Keto-1** monomer as extracted from the X-Ray optimized (solid purple).

For the calculation of the IR spectra, the CAM-B3LYP [1] functional was used. This is one of the functionals that was proved reliable for the simulation of the IR spectra [2,3]. A scaling factor of 0.95 was applied to the calculated spectra [4]. Besides the conformations provided by the X-ray spectra, we have investigated a monomer conformation that provides an intramolecular hydrogen bonding.

[1] Yanai, T.; Tew, D. P.; Handy, N. C. A new hybrid exchange–correlation functional using the Coulomb-attenuating method (CAM-B3LYP) *Chem. Phys. Lett.* **2004**, 393, 51-57. <https://doi.org/10.1016/j.cplett.2004.06.011>

[2] Srivastava, R.; Al-Omary, F.A.M.; El-Emam, A.A.; Pathak, S.K.; Karabacak, M.; Narayan, V.; Chand, S.; Prasad, O.; Sinha, L. A combined experimental and theoretical DFT (B3LYP, CAM-B3LYP and M06-2X) study on electronic structure, hydrogen bonding, solvent effects and spectral features of methyl 1H-indol-5-carboxylate *J. Mol. Struct.* **2017**, 1137, 725-741. <https://doi.org/10.1016/j.molstruc.2017.02.084>

[3] Komjati, B.; Urai, A.; Hosztafi, S.; Kokosi, J.; Kovats, B.; Nagy, J.; Horvath, P. Systematic study on the TD-DFT calculated electronic circular dichroism spectra of chiral aromatic nitro compounds: A comparison of B3LYP and CAM-B3LYP *Spectrochim. Acta A* **2016**, 155, 95-102. <https://doi.org/10.1016/j.saa.2015.11.002>

[4] Jarota, A.; Drwal, D.; Pieta, J.; Pastorczak, E. Wide-range IR spectra of diarylethene derivatives and their simulation using the density functional theory *Sci. Rep.* **2022**, 12, 16834. <https://doi.org/10.1038/s41598-022-20264-x>

3. Condensed sydnones with tertiary alcohol structure generating O-H...O=C dimers

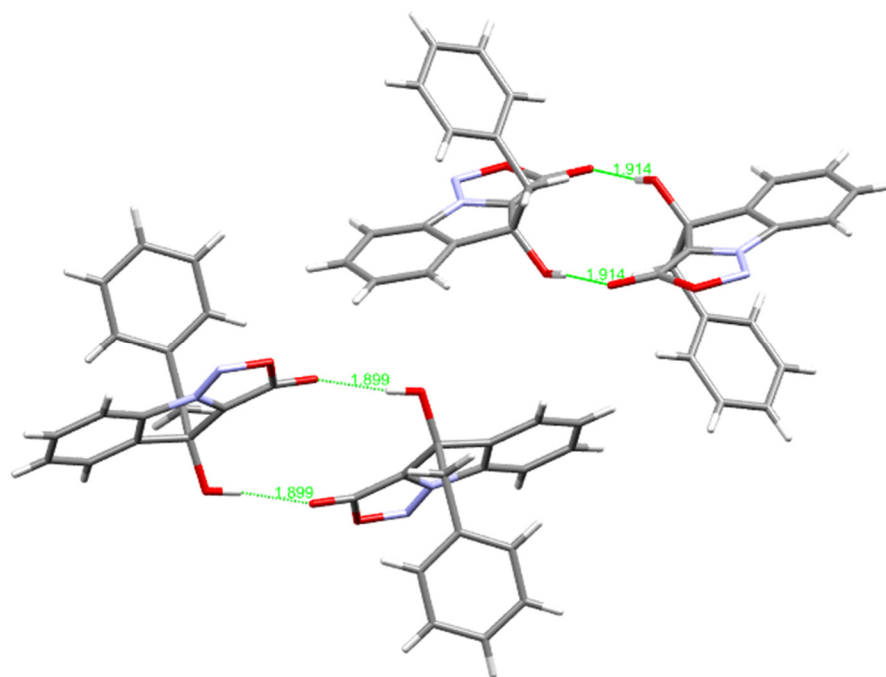


Figure 3S. OH...O=C dimers of *ABELID* [5] (two independent molecules)

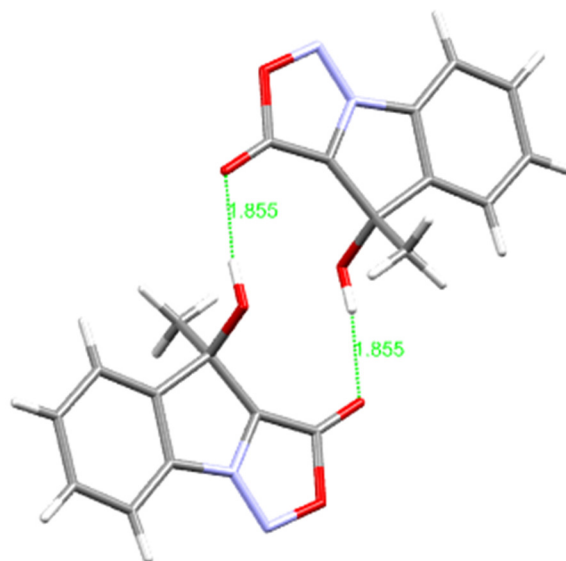


Figure 4S. OH...O=C dimers of *TAPJUP* [6]

[5] Riddle, G.B.; Grossie, D.A.; Turnbull, K. The sydnone compound 4-hydroxy-4-benzyl-sydno-[3,4-a]-indole *Acta Cryst. E* **2004**, *60*, o1568-o1570. <https://doi.org/10.1107/S1600536804020082>

[6] Grossie, D.A.; Turnbull, K. 5-Hydroxy-5-methylsydno[3,4-a]indole *Acta Cryst. C* **1992**, *48*, 377-379. <https://doi.org/10.1107/S0108270191009125>

4. Primary interactions in 4-acetylsydnone deposited in the CSD

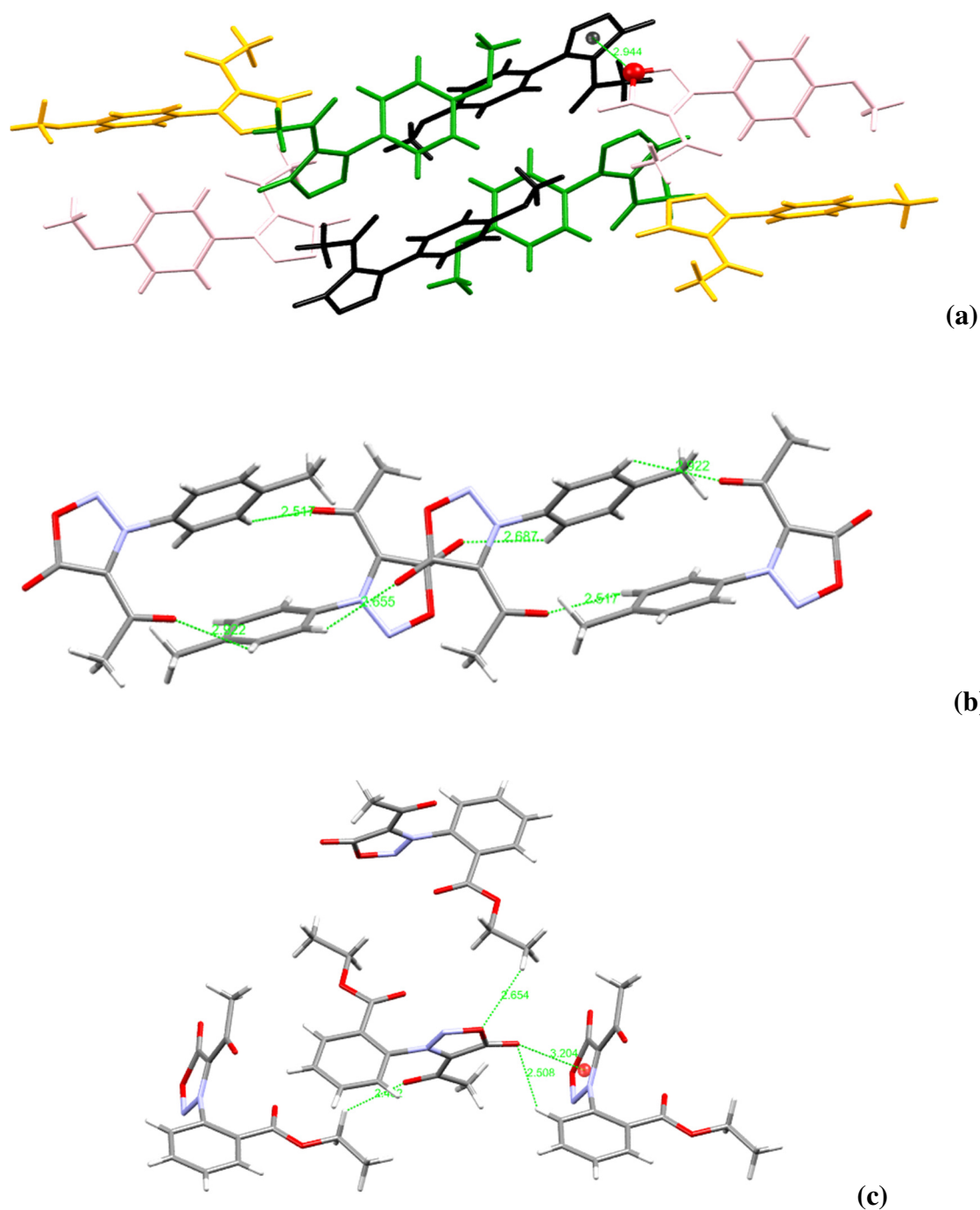


Figure 5S. Known CCDC 4-acetylsydnone: a. *ERUCAW*-four independent molecules-highlight $O\cdots\pi$ contact; b. *FIXPIL* – Examples of H-bonds involving the $C=O$ bonds; c. *MIHTUU* – Contacts involving the O atoms in the sydnone and acetyl moiety.

5. X-Ray-Diffraction supplementary data for Alc-1,2 and Keto-1,2

Table 3S. Bond lengths in crystals of the investigated compounds

Alc-1				Alc-2			
Atom-Atom	Length/Å	Atom-Atom	Length/Å	Atom-Atom	Length/Å	Atom-Atom	Length/Å
O1-N2	1.378(2)	C1-C6	1.366(3)	F1-C4	1.347(3)	C1-C6	1.370(3)
O1-C8	1.402(3)	C2-C3	1.373(4)	O1-N2	1.377(3)	C2-C3	1.373(4)
O2-C8	1.223(3)	C3-C4	1.359(4)	O1-C8	1.398(3)	C3-C4	1.360(4)
O3-C9	1.419(3)	C4-C5	1.374(3)	O2-C8	1.228(3)	C4-C5	1.363(4)
N1-N2	1.311(3)	C5-C6	1.371(3)	O3-C9	1.424(3)	C5-C6	1.377(4)
N1-C1	1.448(3)	C7-C8	1.396(3)	N1-N2	1.313(3)	C7-C8	1.402(3)
N1-C7	1.346(3)	C7-C9	1.496(3)	N1-C1	1.445(3)	C7-C9	1.502(4)
C1-C2	1.369(3)	C9-C10	1.511(3)	N1-C7	1.341(3)	C9-C10	1.507(4)
				C1-C2	1.366(3)	-	
Keto-1				Keto-2			
Atom-Atom	Length/Å	Atom-Atom	Length/Å	Atom-Atom	Length/Å	Atom-Atom	Length/Å
O1-N2	1.351(10)	C1-C6	1.367(12)	F1-C4	1.362(2)	C1-C6	1.373(3)
O1-C8	1.432(11)	C2-C3	1.367(13)	O1-N2	1.370(2)	C2-C3	1.381(3)
O2-C8	1.173(10)	C3-C4	1.369(17)	O1-C8	1.419(3)	C3-C4	1.358(3)
O3-C9	1.220(9)	C4-C5	1.377(16)	O2-C8	1.206(2)	C4-C5	1.372(3)
N1-N2	1.307(9)	C5-C6	1.365(14)	O3-C9	1.215(2)	C5-C6	1.383(3)
N1-C1	1.448(10)	C7-C8	1.445(11)	N1-N2	1.307(2)	C7-C8	1.418(3)
N1-C7	1.345(9)	C7-C9	1.458(10)	N1-C1	1.447(2)	C7-C9	1.459(3)
C1-C2	1.379(11)	C9-C10	1.495(12)	N1-C7	1.358(2)	C9-C10	1.500(3)
				C1-C2	1.379(3)	-	

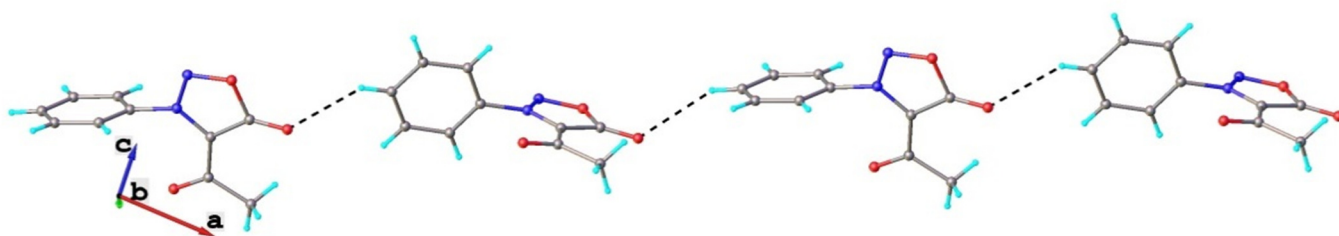


Figure 6S. View of 1D supramolecular chain showing the role of C-H...O hydrogen bonds in the crystal structure of keto-1. H-bond parameters (Å, °): C5-H...O2 [C5-H 0.930, H...O2 2.63, C5...O2(−0.5 + x, 1.5 − y, z − 0.5) 3.43(1), ∠C5HO2 145.2].

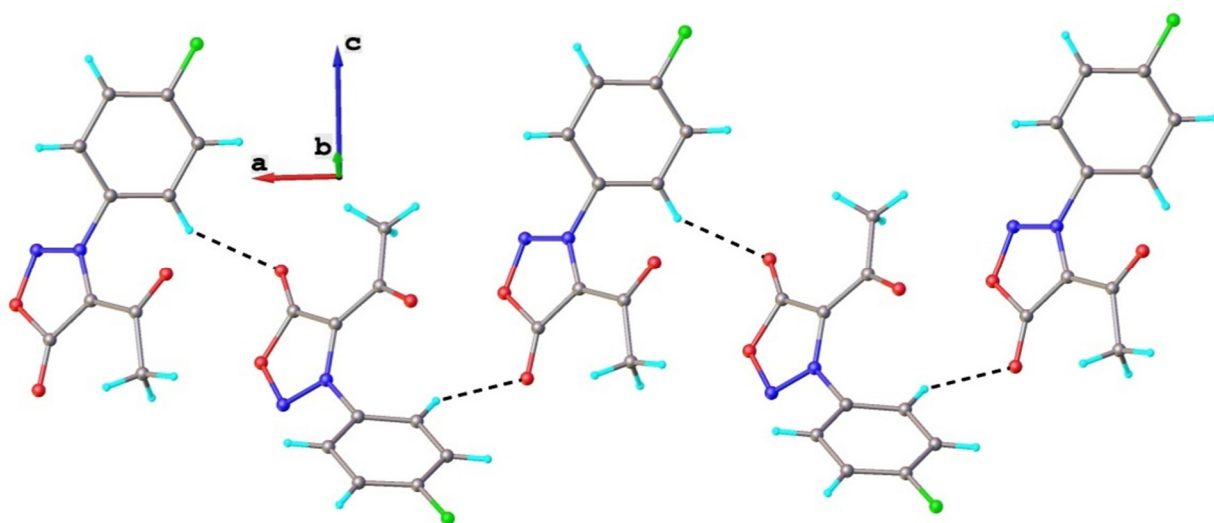


Figure 7S. View of 1D supramolecular chain showing the role of C-H...O hydrogen bonds in the crystal structure of keto-2. H-bond parameters (Å, °): C2-H...O2 [C2-H 0.952, H...O2 2.37, C2...O2(−0.5 + x, y, 0.5 − z) 3.191(2), ∠C2HO2 144.8].

6. Hirshfeld surface analysis of keto-1 and keto-2

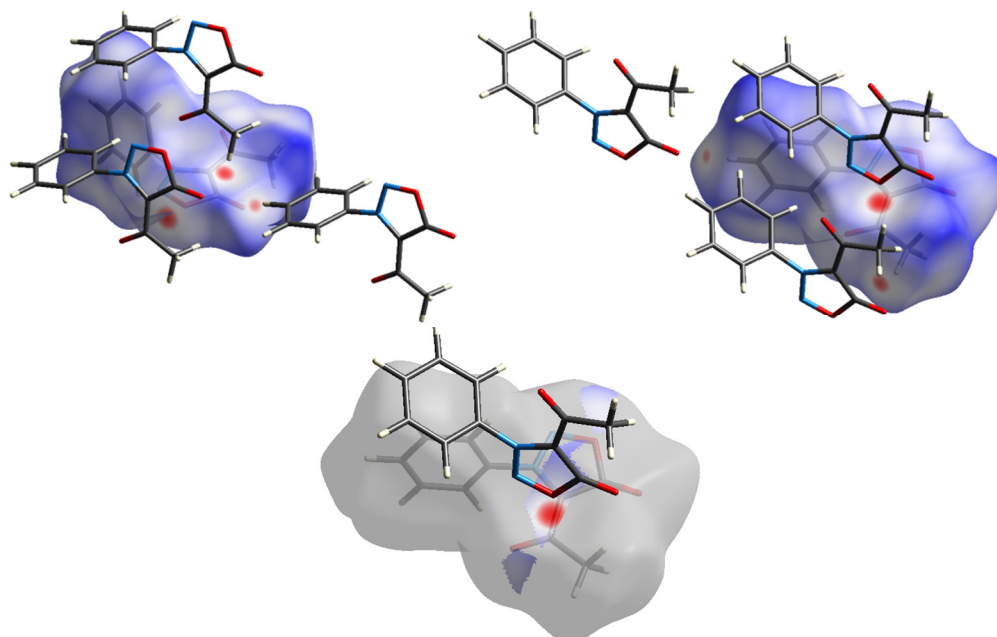


Figure 8S. −O...C=O, CO...π, CH...π and C=O...H(*para*) contacts in **Keto-1**

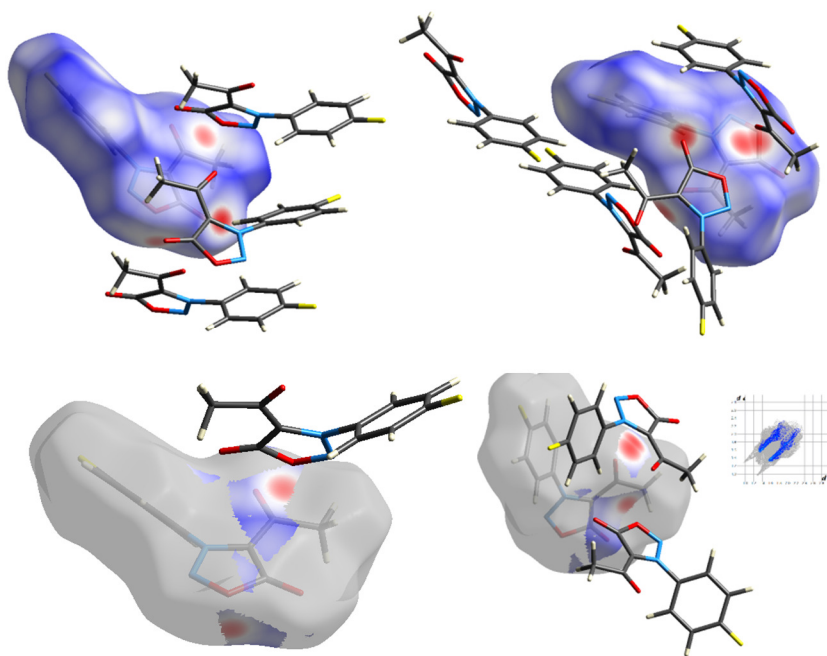


Figure 9S. $-\text{O}\cdots\text{C}=\text{O}$, $\text{CO}\cdots\pi$, and $\text{C}=\text{O}\cdots\text{H}(\text{ortho})$ contacts in **Keto-2**

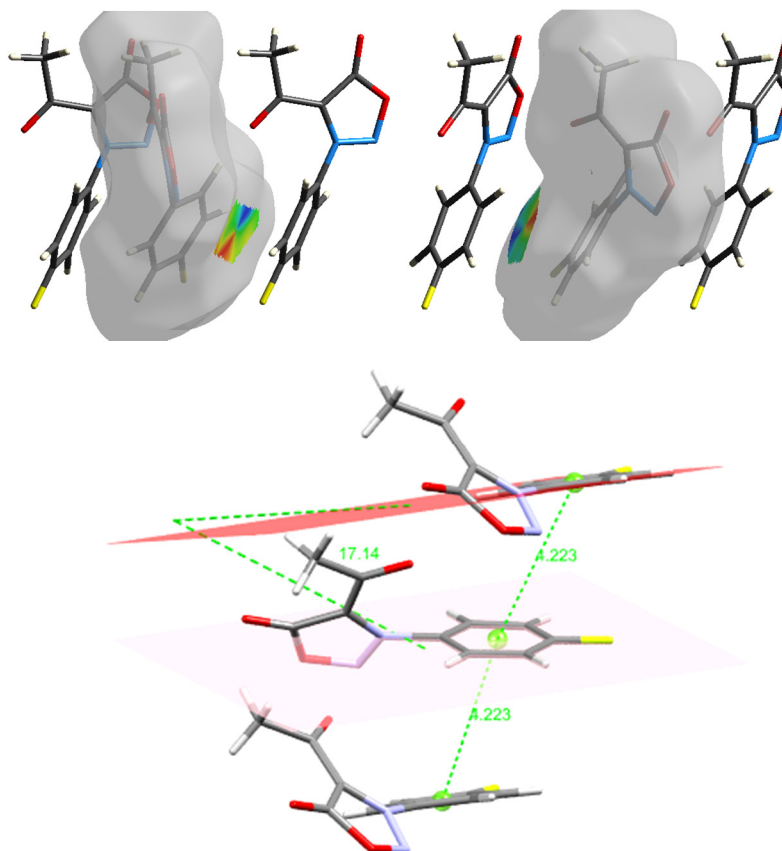


Figure 10S. HF surface Shape index mode presents complementarity surfaces indicating a $\pi\cdots\pi$ interaction. However, $\text{C}_g\cdots\text{C}_g$ distances of 4.22 Å and interplanar angles for benzene ring planes of 17° are not favorable

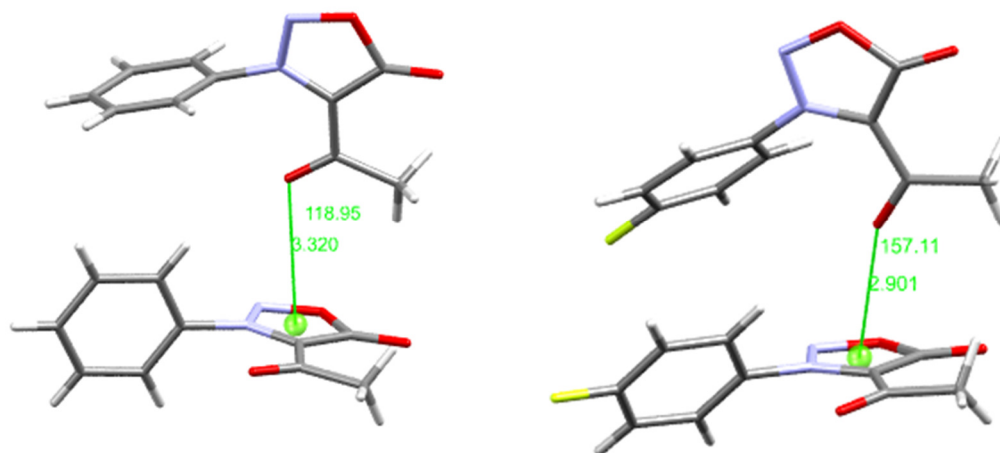
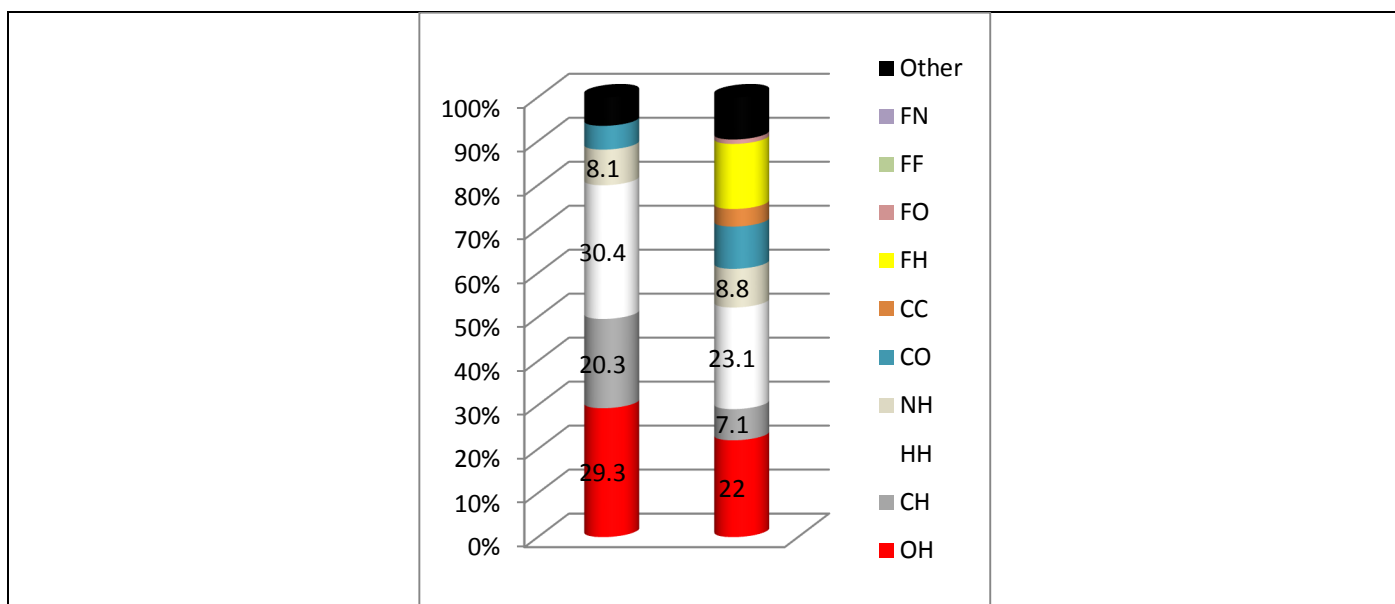


Figure 11S. Measurements of possible $C=O \cdots \pi$ interactions to justify increased $O \cdots C$ contacts in **Keto-2** (more favorable angle and vdW contact distance $< 3.22 \text{ \AA}$)

Table 4S. Fingerprint plots for ketones **Keto-1** and **Keto-2** and percentage interactions

Keto-1						
	All	C...H 20.3%	H...H 30.4%	N...H 8.1%	O...C 5.4%	O...H 20.9%
	O...N 3.9%					
Keto-2						
	All	C...H 7.1%	H...H 23.1%	N...H 8.8%	O...C 9.6%	O...H 22.0%
	O...N 3.1%	C...C 4.0%	F...H 14.8%			



7. Crystal features and analysis of Alc-3

Table 5S. Complete crystal data refinement for **Alc-3**

Empirical formula	C ₁₀ H ₁₀ N ₂ O ₃
Formula weight	206.20
Temperature/K	288(19)
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	41.5257(16)
b/Å	12.1880(5)
c/Å	7.6472(4)
α/°	90
β/°	90.053(4)
γ/°	90
Volume/Å ³	3870.4(3)
Z	16
ρ _{calc} /g/cm ³	1.415
μ/mm ⁻¹	0.894
F(000)	1728.0
Crystal size/mm ³	0.1 × 0.1 × 0.05
Radiation	Cu Kα (λ = 1.54184)
2θ range for data collection/°	6.386 to 133.192
Index ranges	-42 ≤ h ≤ 49, -14 ≤ k ≤ 14, -9 ≤ l ≤ 9

9. DFT – supplementary data

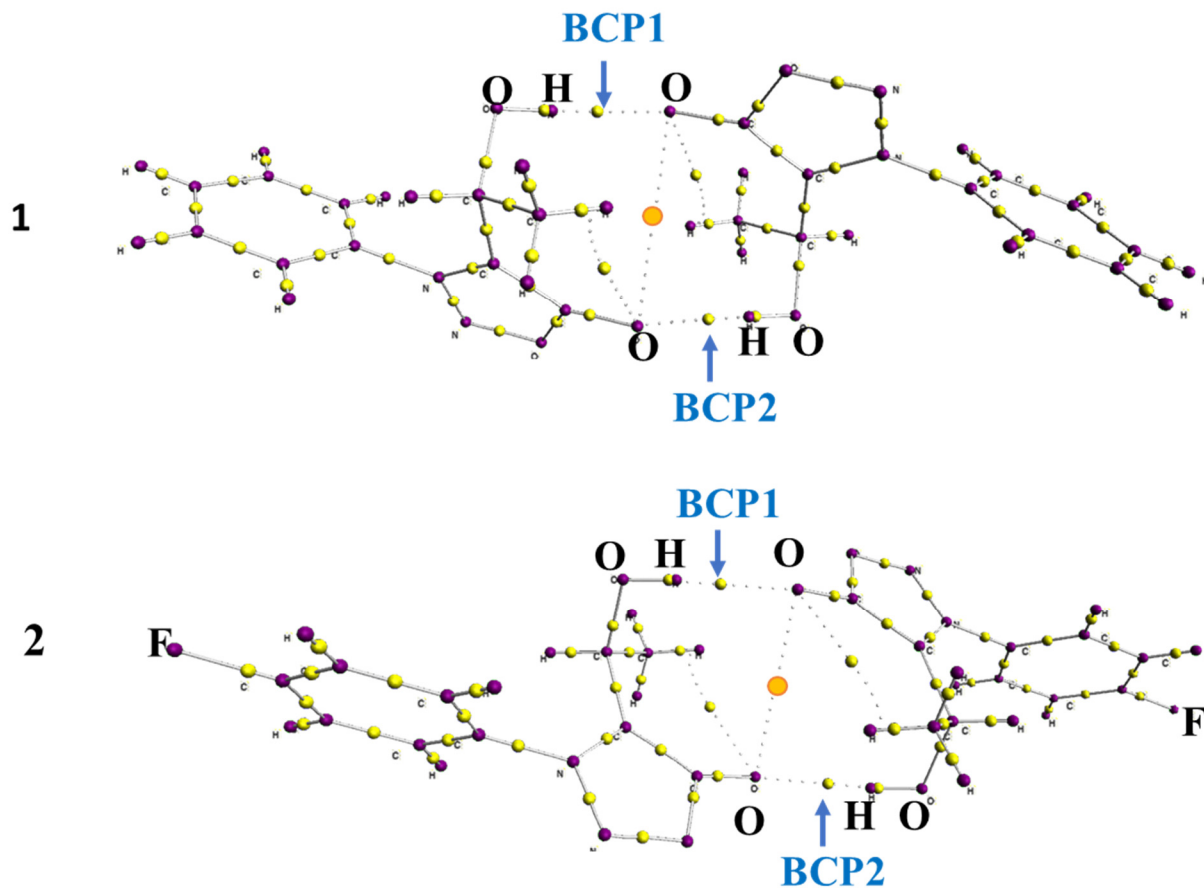


Figure 13S - Molecular graphs of HB dimeric complexes ($\text{-O-H}\cdots\text{O-}$), showing positions of the two attractors (purple spheres) linking the bond paths (BPs) and the corresponding bond critical points (BCP) (yellow spheres) for **Alc-1** and **Alc-2**, obtained with M06-D3/6-311++G**. The orange mid-point represents the ring critical point. Graphs were performed with the aid of the QTAIM extension available in Avogadro

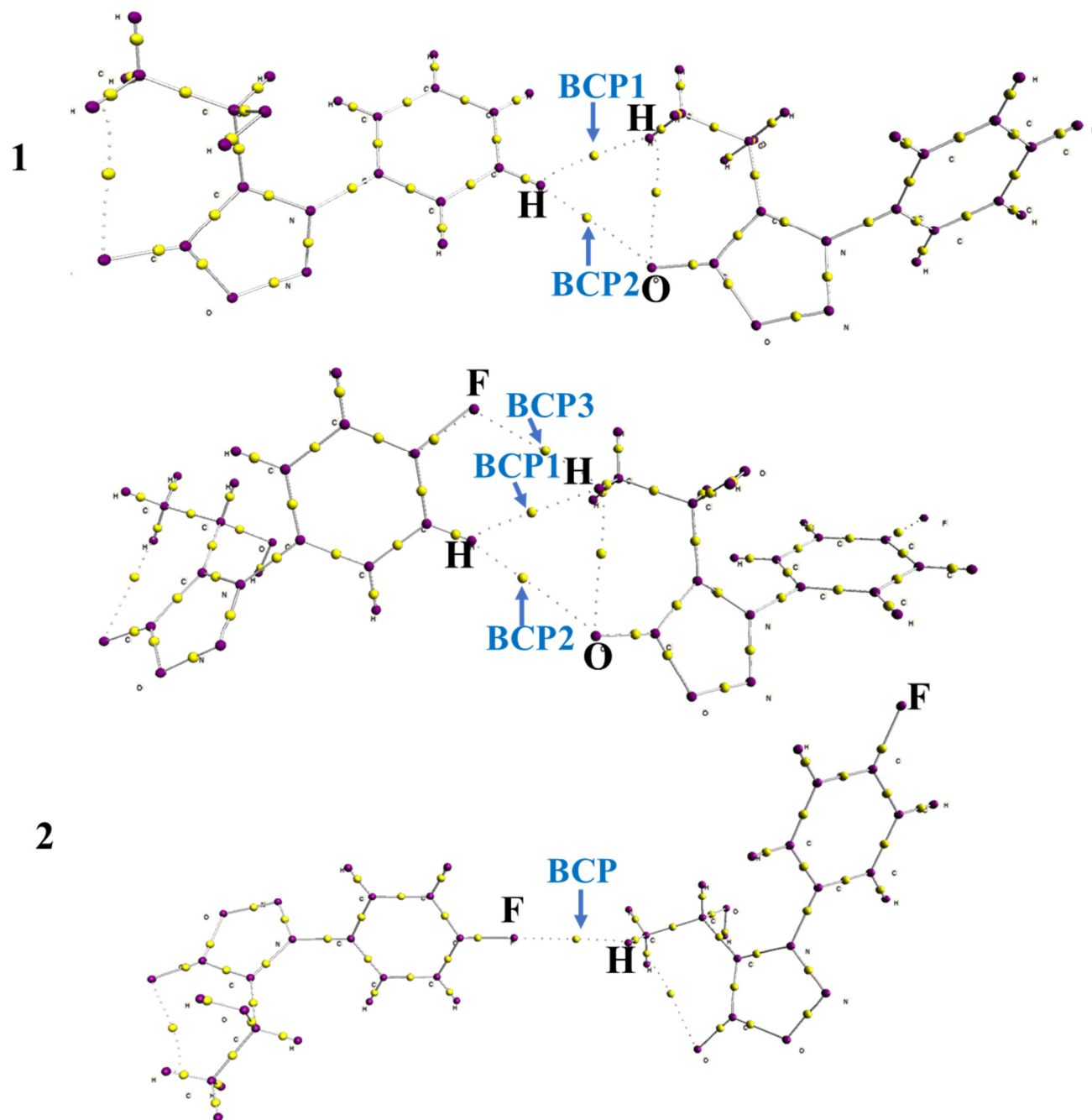


Figure 14S - Molecular graphs of HB dimeric complexes ($C=O \cdots H-C/C-H \cdots H-C/C-F \cdots H-C$), showing positions of the two attractors (purple spheres) linking the bond paths (BPs) and the corresponding bond critical points (BCP) (yellow spheres) for **Alc-1** and **Alc-2**, obtained with M06-D3/6-311++G**. Graphs were performed with the aid of the QTAIM extension available in Avogadro