

Preparation of 2D Antimonene Nanosheets via Topochemical Reduction Conversion

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1. Experimental Section

1.1 materials

SbI₃ (99.0%) and NaBH₄ (99.99%) were purchased from Shanghai Aladdin Biochemical Technology Co., Ltd. Fluorophlogopite mica and silicon wafers were obtained from Nanjing MKNANO Technology Co., Ltd.

1.2 Characterization of the materials

TEM analysis was conducted on a JEOL JEM-F200 instrument operating at an accelerating voltage of 200 kV for microstructure characterization.

GI-XRD measurements were performed on a Rigaku SmartLab SE diffractometer using Cu K α radiation ($\lambda = 0.154$ nm) for phase composition analysis.

XPS characterization was carried out on a Thermo Scientific K-Alpha spectrometer equipped with a monochromatic Al K α excitation source (1486.69 eV) for chemical state analysis.

OM observations were made using a Nikon Eclipse LV1000D microscope for morphological examination.

Raman spectroscopy was performed on a WITec Alpha 300R system to probe the intrinsic vibrational modes of the crystal.

2. Supplementary Figures

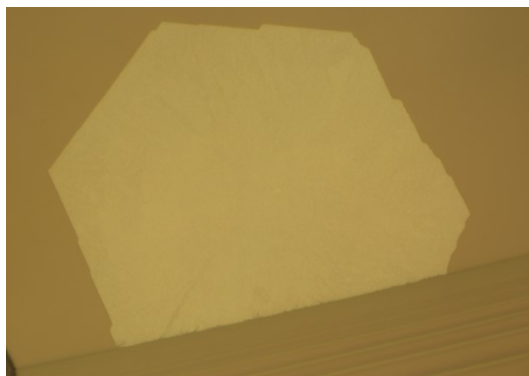


Figure. S1 OM image of antimonene after reduction

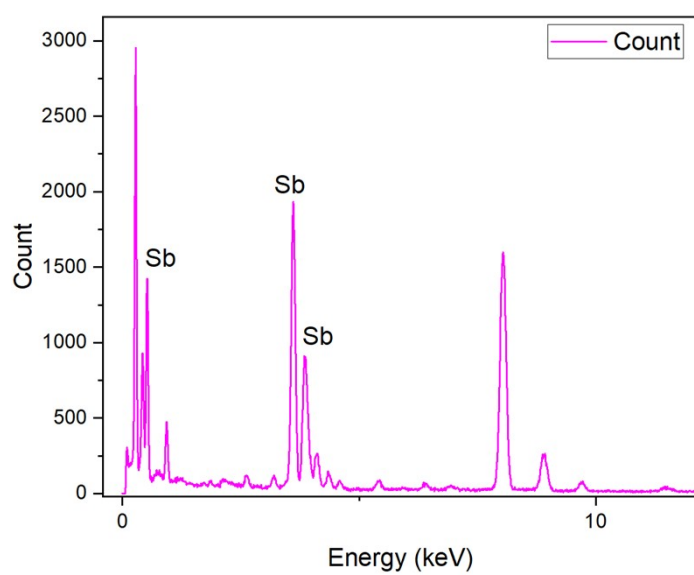


Figure. S2 EDS spectrum of antimonene nanosheets.



Figure. S3 FFT of antimonene nanosheets.

Table S1 Quantitative analysis

Element	Mass (%)	Atom (%)
Sb	100	100
I	0	0

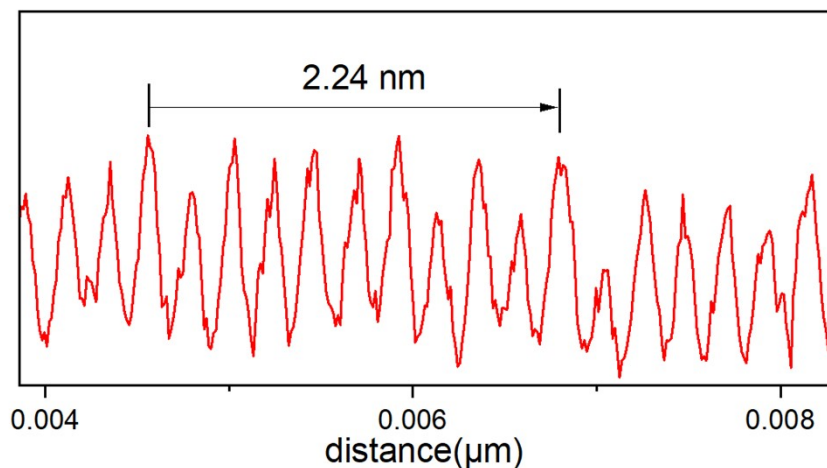


Figure. S4 The TEM lattice spacing of antimonene

Table S2 Comparison of different synthesis methods for antimonene

Methods	Synthesis temperature (°C)	Lateral size and shape (μm)	Synthesis atmosphere	Reference
CVD	660-750	<10, hexagon	Ar	1, 2
Liquid-phase exfoliation	25	<5, irregular	Air	3, 4
Wet-chemical synthesis	300	<3, hexagon	Ar	5, 6
Topochemical				
Reduction	110	~50, hexagon	Air	This work
Conversion				

References:

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