

Supplementary information

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1 Raman vibrational modes:

According to group theory, the Raman spectrum of the magnetite spinel structure exhibits five Raman active modes ($A_{1g} + 3T_{2g} + E_g$)¹, the indexation of which has caused some controversy in the literature during the past two decades. During these years a large number of experimental studies, as well as some computational ones, were conducted in order to reach an accurate indexation²⁻⁵, but one of the most accepted interpretations nowadays is that presented by Shebanova *et al.* based on polarized Raman studies^{6,7}. The reported wavenumber values for these vibrational modes are $\nu_{T_{2g}(1)}=193\text{ cm}^{-1}$, $\nu_{E_g}=306\text{ cm}^{-1}$, $\nu_{T_{2g}(2)}=450\text{-}490\text{ cm}^{-1}$, $\nu_{T_{2g}(3)}=538\text{ cm}^{-1}$ and $\nu_{A_{1g}}=668\text{ cm}^{-1}$. Nevertheless, the typical room temperature Raman spectrum of magnetite nanoparticles is characterized by a high-intensity band corresponding to A_{1g} found at $\approx 660\text{-}670\text{ cm}^{-1}$, together with two low-intensity bands corresponding to E_g and $T_{2g(3)}$ located around ≈ 300 and $\approx 530\text{ cm}^{-1}$, respectively. The observation of the bands corresponding to $T_{2g(1)}$ and $T_{2g(2)}$ is not always reported.

In the case of $\gamma\text{-Fe}_2\text{O}_3$, even if group theory predicts the same number of Raman active vibrational modes as for Fe_3O_4 , a spectrum consisting of three vibrational bands is typically recorded. In this case, the main A_{1g} is located around $\approx 710\text{ cm}^{-1}$, blue-shifted compared to that of magnetite. Two lower intensity broad bands corresponding to the vibrational modes T_{2g} and E_g are usually located at $\approx 350\text{ cm}^{-1}$ and $\approx 500\text{ cm}^{-1}$, respectively^{2,4}.

2 Supplementary tables:

Table S1: Computed cell parameter applying the DICVOL04 cell refinement method. It must be taken into account that, given a cubic cell structure, $a=b=c$ and $\alpha=\beta=\gamma=90^\circ$. The lattice parameter values (a) for magnetite and maghemite have been taken from references [8-10]

Material	a (Å)
$\text{Fe}_3\text{O}_4^{8,9}$	8.396
$\gamma\text{-Fe}_2\text{O}_3^{10}$	8.33
40:120_S	8.363 ± 0.005
40:120_L	8.352 ± 0.004
40:115_S	8.346 ± 0.005
40:125_S	8.342 ± 0.005
40:115_S_OAm	8.331 ± 0.008

Table S2: position and mixing parameter (η) of the fitted Raman peaks.

	$\lambda=633$ nm				$\lambda=514$ nm			
	$\text{A}_{1g}\text{Fe}_3\text{O}_4$		$\text{A}_{1g}\gamma\text{-Fe}_2\text{O}_3$		$\text{A}_{1g}\text{Fe}_3\text{O}_4$		$\text{A}_{1g}\gamma\text{-Fe}_2\text{O}_3$	
	Position	η	Position	η	Position	η	Position	η
40:120_S	670	0.998	702	0.988	664	0.999	706	0.986
40:120_L	670	0.999	707	0.999	666	0.999	714	0.981
40:115_S	670	0.970	704	0.862	664	0.999	704	0.999
40:125_S	670	0.996	710	0.999	662	0.971	714	0.999
40:115_S_OAm	670	0.999	716	0.999	665	0.999	706	0.997

Table S3: Raman and infrared active modes corresponding to each space groups related to $\gamma\text{-Fe}_2\text{O}_3$. Built using the Bilbao Crystallographic Server [1]

Space group	Wyckoff positions	IR active modes	Raman active modes
Fd-3m	8a	-	T_{2g}
	16d	2T_{1u}	-
	32e	2T_{1u}	$\text{A}_{1g} + \text{E}_g + \text{T}_{2g}$
P4₁32, P4₃32	4b	1T_1	$\text{E} + \text{T}_2$
	8c	3T_1	$\text{A}_1 + 2\text{E} + 3\text{T}_2$
	12d	5T_1	$\text{A}_1 + 3\text{E} + 4\text{T}_2$
	24e	9T_1	$3\text{A}_1 + 6\text{E} + 9\text{T}_2$
P4₁2₁2, P4₃2₁2	4a	$\text{A}_2 + 2\text{E}$	$\text{A}_1 + 2\text{B}_1 + \text{B}_2 + 3\text{E}$
	8b	$3\text{A}_2 + 6\text{E}$	$3\text{A}_1 + 3\text{B}_1 + 3\text{B}_2 + 6\text{E}$

3 Supplementary figures:

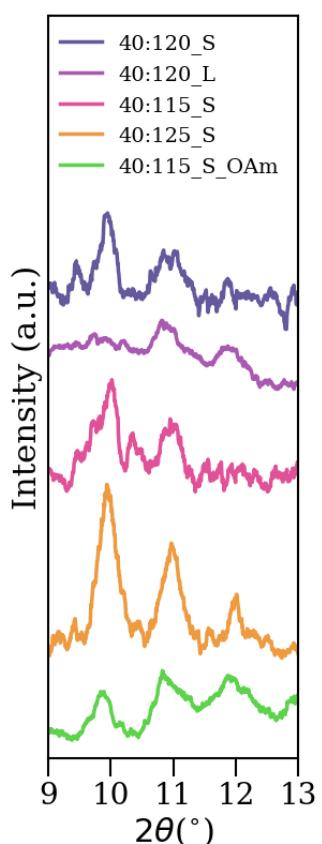


Figure S1: a vertically enlarged section of the diffractogram to enhance the visualization of peaks (210) and (211).

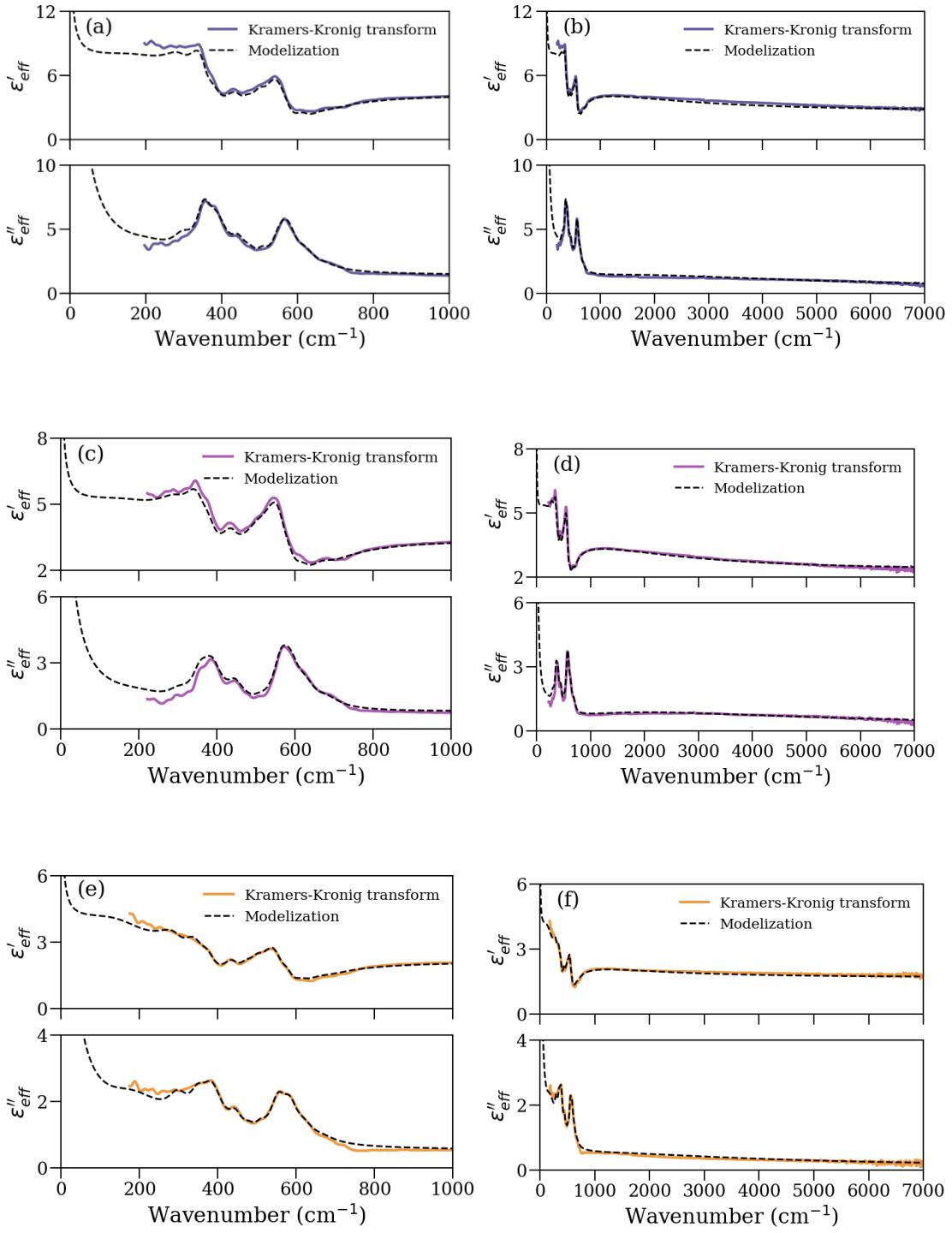


Figure S2: The real and imaginary parts of the effective permittivity of the samples computed using Kramers-Kronig analysis (solid lines) and obtained from the modelization (dashed lines) for samples 40:120_S (FIR (a), MIR (b)), 40:120_L (FIR (c), MIR (d)) and 40:125_S (FIR (e), MIR (f)).

4 References

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