SUPPLEMENTARY MATERIAL

Exploring quantum tunneling in heavy atom reactions using a rigorous theoretical approach to the dynamics. Formation of NO + O from the N + O_2 atmospheric reaction

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Zero point energy issue and N+O₂ reaction

There are two cases in which the zero point energy (ZPE) effect can be relevant in QCT. The first and most important one is related to polyatomic molecules, in which more than one vibrational degree of freedom is present. As a consequence, it is very likely that the initial vibrational energy of one vibrational mode leaks into the others, even before the interaction with the target takes place (ref. S1). The second case is in the presence of an endothermic reaction, when the product can be formed classically with a final distribution systematically lower than the ZPE. In this condition, the product lowest vibrational bin is covered by a strongly uneven distribution, which gives origin to inaccuracies (refs. S2 and S3). This condition is present when the barrier to reaction is lower than the product ZPE, and the total energy is just sufficient for reaction by overcoming the barrier (of course excluding tunneling in classical dynamics). This issue can be seen as part of a more general problem of final analysis by binning in QCT, when the final distribution does not span the whole width of a bin (refs. S2 and S3). On the contrary, when total energy is higher than final ZPE, the final vibrational action tends to be smoothly distributed at least on the lowest vibrational bin, with good accuracy.

In this work, we consider the products without a final state analysis. Using a standard histogram binning (HB) and then summing final products is exactly equivalent to avoid any final state analysis. If a Gaussian binning is used, however, the result can differ, because it depends on the final distribution in each bin. In the present conditions (exothermicity much higher than product ZPE, presence of a deep well in the strong coupling region), the strong interaction in the A+BC collision tends to give origin to wide product distributions, which will not produce significantly different results when using different binning schemes. A result using GB is presented in Figure S1, where the reaction probability at J=0 is shown, as calculated using histogram binning and Gaussian binning from the same set of trajectories used in this study. The two results differ only for larger statistical fluctuations associated with GB, as expected, because many trajectories are discarded using this procedure. No significant difference is observed concerning reaction threshold.



Figure S1. Comparison of histogram binning (HB) and Gaussian binning (GB) applied to the QCT reaction probability calculation for J=0, including the set of trajectories used in this work. The two results differ only for slightly worse statistics associated with GB.



Figure S2. QM (symbols) and QCT (lines) reaction probabilities multiplied by the (2J+1) term as a function of *J*, for a selection of E_{col} values. This figure complements Figure 1.



Figure S3. QM (black) and QCT (red) reaction probabilities as a function of E_{col} , for a selection of *J* values.



Figure S4. QM reaction probability as a function of E_{col} , for a selection of *J* values in the 0-90 interval and considering *p*=+ and K_0 =0.





$$V_{jK}^{J}(R) = \min_{r,\gamma} V(R,r,\gamma) + \frac{\left[J(J+1) + j(j+1) - 2K^{2}\right]}{2\mu_{R}R^{2}}$$

where the different symbols have the usual meaning.^{S4} In this expression the potential energy (PES) is minimized with respect to *r* and γ .



Figure S6. QM reaction probability as a function of E_{col} , for a selection of *J* values in the 0-30 interval and considering $K_0=0$ (a) and $K_0=1$ (b).



Figure S7. QM reaction probability as a function of E_{col} , for *J*=0 and considering j_0 values in the 1-17 interval.



Figure S8. Comparison of the QM and QCT cross sections obtained here with the QM approximate ones (*J*-shifting approximation) reported by Defazio et al. (time dependent QM (RWP))^{S5} and Sultanov and Balakrishna (time independent QM).^{S6}



Figure S9. QM (black) and QCT (red) rate constants for N + $O_2(v_0=0, j_0=1) \rightarrow$ NO + O and experimental (blue) thermal rate constant for N + $O_2 \rightarrow$ NO + O including error bars.^{S7}

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