

Supplementary information

**Unlocking the Sensing and Scavenging Potential of Sc_2CO_2 and
 $\text{Sc}_2\text{CO}_2/\text{TMD}$ Heterostructures for Phosgene Detection**

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Table S1: The atomic positions of each atom of the unit cell structures of Fig 1 of the original manuscript. Data is presented in fractional coordinates.

| Struct 1 | | | |
|-----------------|-------------|-------------|-------------|
| Atom | x | y | z |
| Sc ₁ | 0.666666687 | 0.333333343 | 0.622142017 |
| Sc ₂ | 0.333333343 | 0.666666687 | 0.377857983 |
| C | 0 | 0 | 0.5 |
| O ₁ | 0.666666687 | 0.333333343 | 0.212495565 |
| O ₂ | 0.333333343 | 0.666666687 | 0.787504435 |
| Struct 2 | | | |
| Atom | x | y | z |
| Sc ₁ | 0.666666806 | 0.333333433 | 0.622141957 |
| Sc ₂ | 0.333333433 | 0.666666865 | 0.377857953 |
| C | 0 | 0 | 0.5 |
| O ₁ | 0 | 0 | 0.21249561 |
| O ₂ | 0 | 0 | 0.787504435 |
| Struct 3 | | | |
| Atom | x | y | z |
| Sc ₁ | 0.666666687 | 0.333333343 | 0.622142017 |
| Sc ₂ | 0.333333343 | 0.666666687 | 0.377857983 |
| C | 0 | 0 | 0.5 |
| O ₁ | 0.666666687 | 0.333333343 | 0.212495551 |
| O ₂ | 0 | 0 | 0.787504435 |
| Struct 4 | | | |
| Atom | x | y | z |
| Sc ₁ | 0.666666687 | 0.333333343 | 0.622142017 |
| Sc ₂ | 0.333333343 | 0.666666687 | 0.377857983 |
| C | 0 | 0 | 0.5 |
| O ₁ | 0.666666687 | 0.333333343 | 0.787504435 |
| O ₂ | 0.333333343 | 0.666666687 | 0.212495551 |

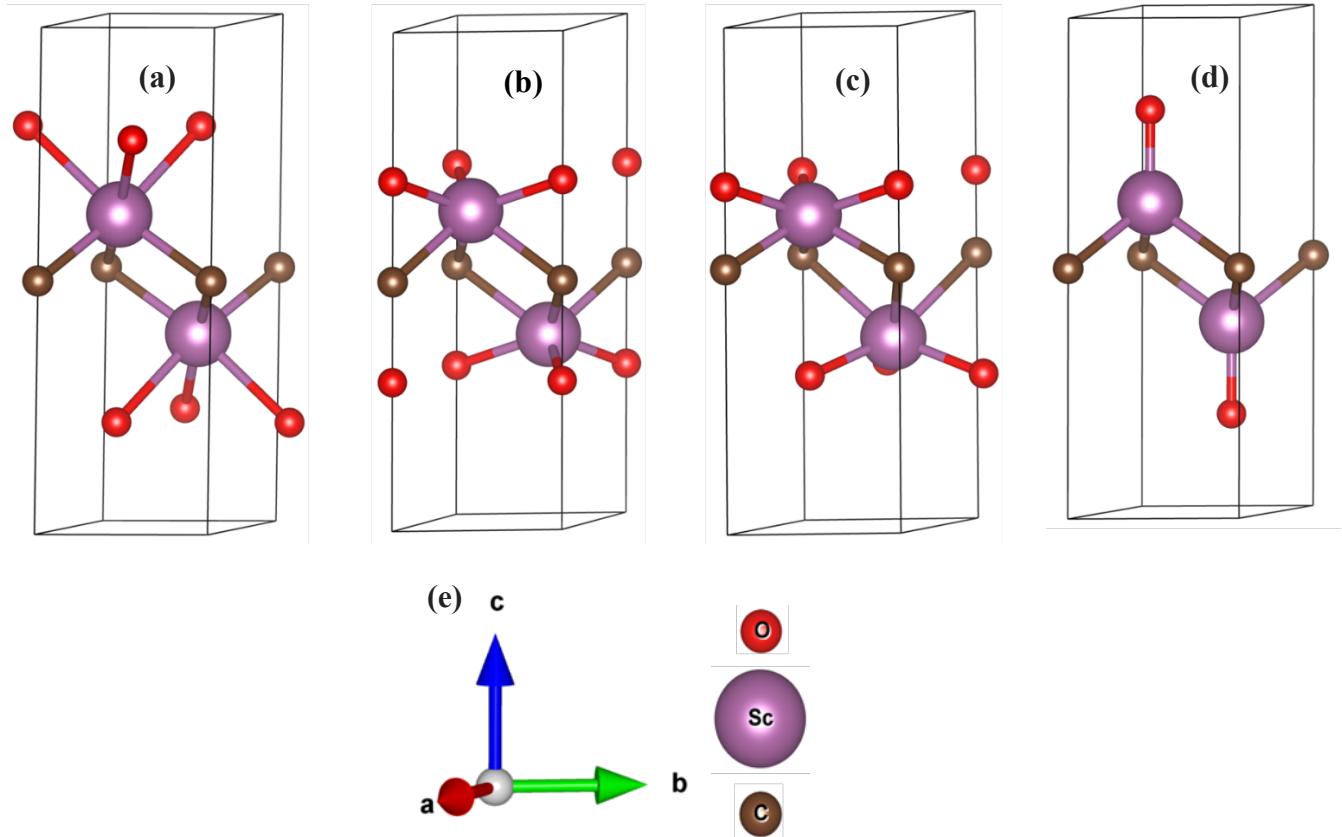


Figure S1. 3D representation of unit cell structures of different models of Sc_2CO_2 (a) Struct_1, (b) Struct_2, (c) Struct_3, and (d) Struct_4 (e) Lattice direction

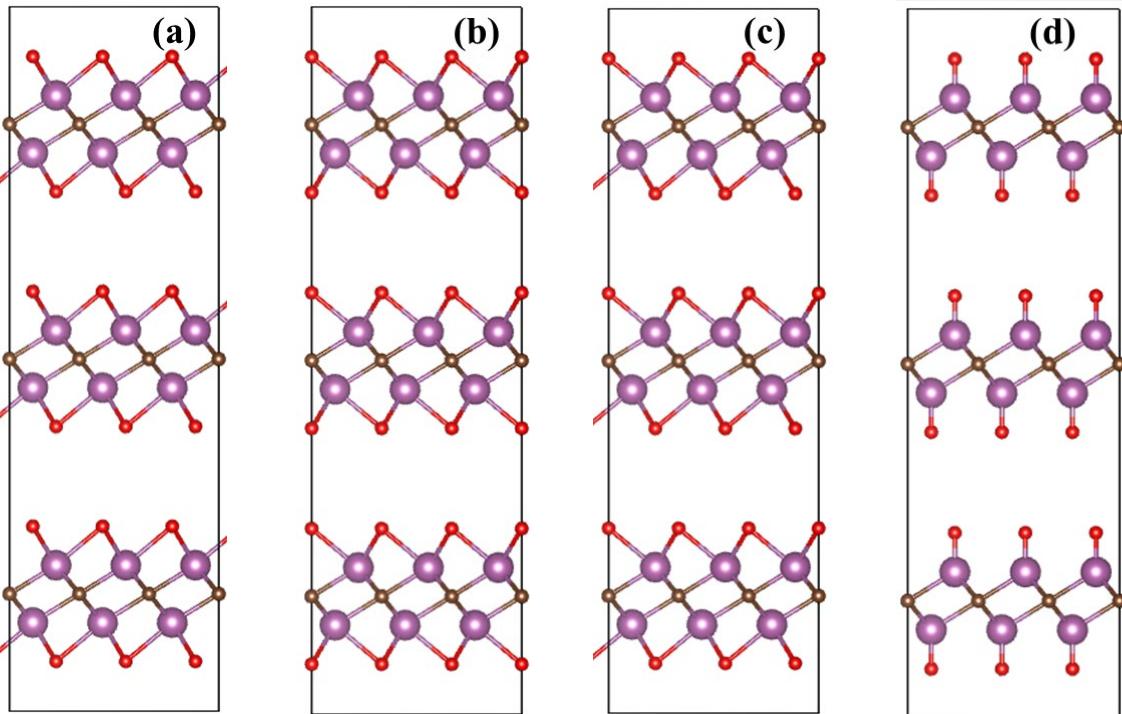


Figure S2. Supercell structures of different models of Sc_2CO_2 (a) Struct_1, (b) Struct_2, (c) Struct_3, and (d) Struct_4

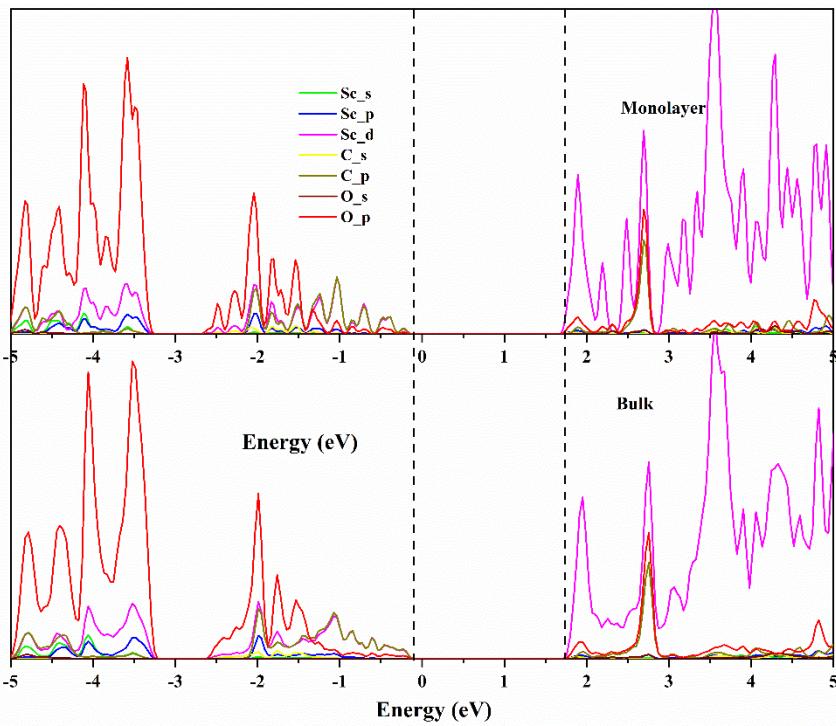


Figure S3. Partial DOS spectra of bulk and monolayer

Table S2: The calculated distance (d) of gas from MXene surface, adsorption energy (E_{ad}) without vdW correction, charge transfer (Δq), recovery time (τ) at 298 K and 373K, of CO, NO, CH₄, H₂S, COCl₂, O₂, and N₂ after interaction with Sc₂CO₂

| Model | d (Å) | E_{ad} (eV) | Δq (e) | τ (s)/298K | τ (s)/373K |
|---|-------|---------------|----------------|-----------------------|-----------------------|
| Sc ₂ CO ₂ | - | - | - | - | - |
| Sc ₂ CO ₂ + O ₂ | 1.6 | -1.04 | 0.067 | 3.73×10^5 | 109.8 |
| Sc ₂ CO ₂ + CO | 2.98 | -1.05 | -0.006 | 5.5×10^5 | 148 |
| Sc ₂ CO ₂ + N ₂ | 1.5 | -1.06 | 0.007 | 8.13×10^5 | 204 |
| Sc ₂ CO ₂ + NO | 2.49 | -1.16 | 0.034 | 5.13×10^7 | 5590 |
| Sc ₂ CO ₂ + CH ₄ | 2.53 | -1.32 | 0.049 | 2.68×10^{10} | 8.3×10^5 |
| Sc ₂ CO ₂ + H ₂ S | 2.14 | -1.75 | 0.175 | 5.44×10^{17} | 5.7×10^{11} |
| Sc ₂ CO ₂ + COCl ₂ | 3.00 | -1.97 | -0.005 | 2.95×10^{21} | 5.48×10^{14} |
| WSe ₂ _Sc ₂ CO ₂ | - | - | - | - | - |
| WSe ₂ _Sc ₂ CO ₂ +COCl ₂ | 1.6 | -1.905 | 1.867 | 2.33×10^{20} | 7.23×10^{13} |
| MoSe ₂ _Sc ₂ CO ₂ | - | - | - | - | - |
| MoSe ₂ _Sc ₂ CO ₂ +COCl ₂ | 1.8 | -1.915 | 1.880 | 3.44×10^{20} | 9.83×10^{13} |