

Supporting Information

Computational high-throughput screening of high-performance transition metal C₈N₈ single-atom electrocatalysts for oxygen reduction reaction.

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complexes.

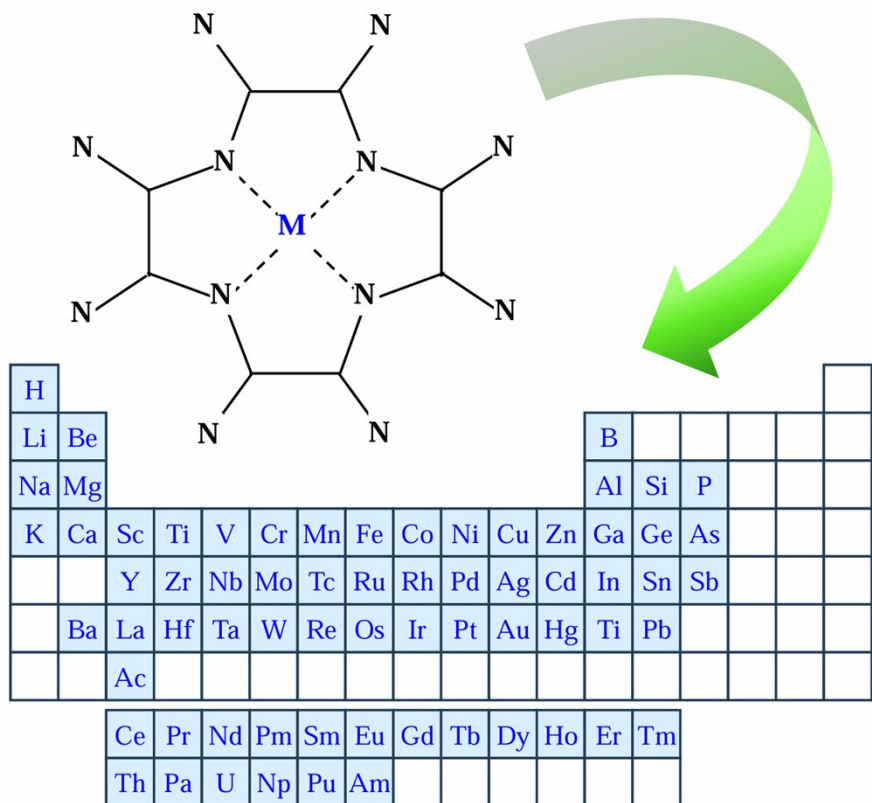


Figure. S2. Linear relationship between formation energy and d-band center.

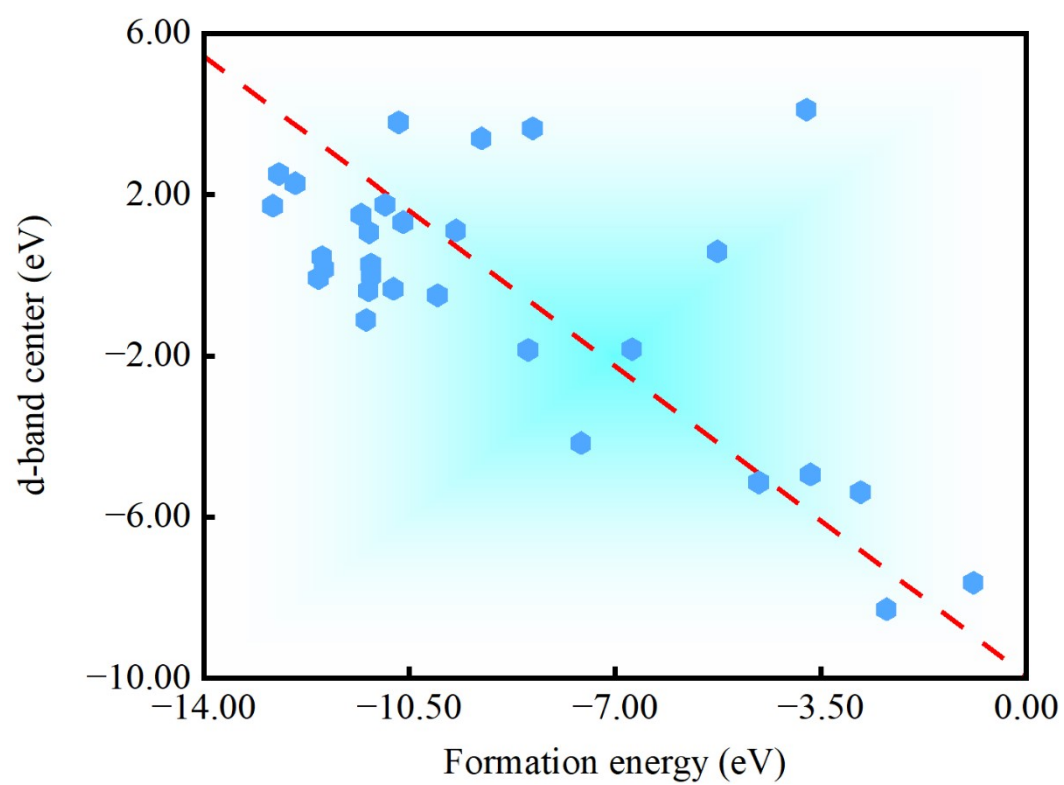


Figure. S3. The structure diagram of Fe-C₈N₈ monolayer.

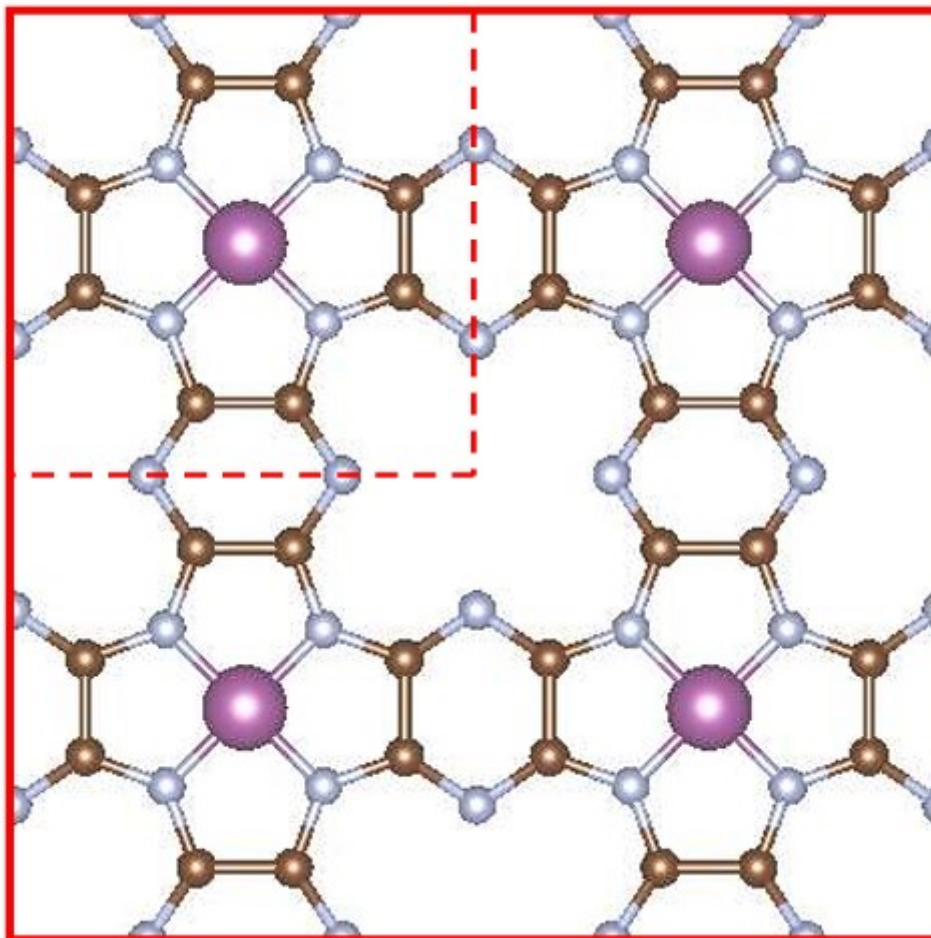
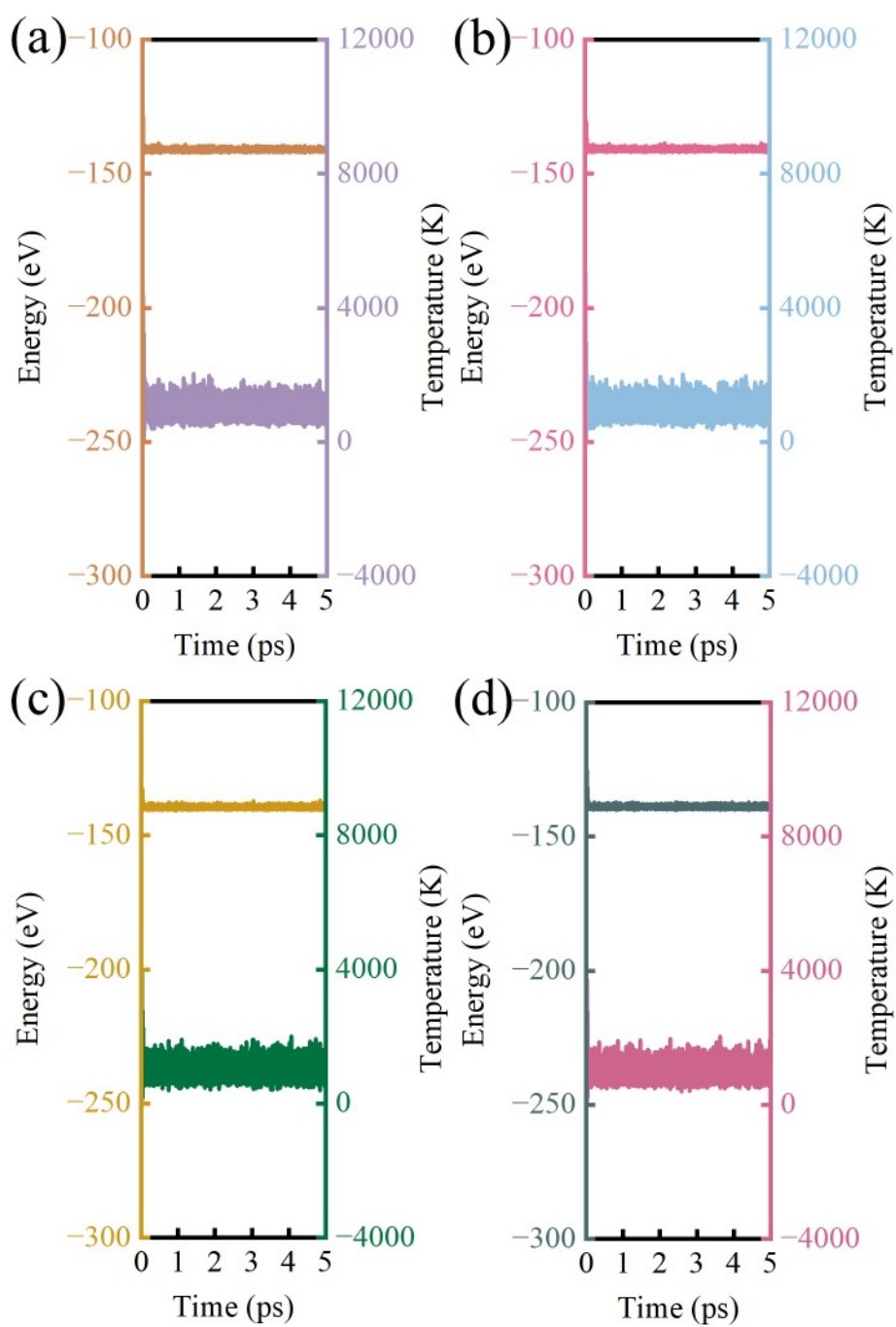


Figure. S4. AIMD energy change of Fe-C₈N₈, Mn-C₈N₈, Co-C₈N₈, and Ru-C₈N₈ at 1000K.



The POSCAR of Fe-C₈N₈.

POSCAR

1.0

7.0798144341	0.0000000000	0.0000000000
0.0000000000	7.0798144341	0.0000000000
0.0000000000	0.0000000000	15.0000000000

C	N	Fe
---	---	----

8	8	1
---	---	---

Direct

0.156021993	0.395461122	0.500000000
0.843977992	0.604538848	0.500000000
0.604538848	0.156021993	0.500000000
0.395461122	0.843977992	0.500000000
0.843977992	0.395461122	0.500000000
0.156021993	0.604538848	0.500000000
0.395461122	0.156021993	0.500000000
0.604538848	0.843977992	0.500000000
0.290396987	0.000000000	0.500000000
0.709602983	0.000000000	0.500000000
0.000000000	0.290396987	0.500000000
0.000000000	0.709602983	0.500000000
0.330963654	0.330963654	0.500000000

0.669036375	0.669036375	0.500000000
0.669036375	0.330963654	0.500000000
0.330963654	0.669036375	0.500000000
0.500000000	0.500000000	0.500000000

The POSCAR of Mn-C₈N₈.

POSCAR

1.0

7.0798144341	0.0000000000	0.0000000000
0.0000000000	7.0798144341	0.0000000000
0.0000000000	0.0000000000	15.000000000

C	N	Mn
---	---	----

8	8	1
---	---	---

Direct

0.156021993	0.395461122	0.500000000
0.843977992	0.604538848	0.500000000
0.604538848	0.156021993	0.500000000
0.395461122	0.843977992	0.500000000
0.843977992	0.395461122	0.500000000
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0.290396987	0.000000000	0.500000000
0.709602983	0.000000000	0.500000000
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0.669036375	0.669036375	0.500000000
0.669036375	0.330963654	0.500000000
0.330963654	0.669036375	0.500000000
0.500000000	0.500000000	0.500000000

The POSCAR of Co-C₈N₈.

POSCAR

1.0

7.0798144341	0.0000000000	0.0000000000
0.0000000000	7.0798144341	0.0000000000
0.0000000000	0.0000000000	15.0000000000

C	N	Co
---	---	----

8	8	1
---	---	---

Direct

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0.843977992	0.604538848	0.500000000
0.604538848	0.156021993	0.500000000

0.395461122	0.843977992	0.500000000
0.843977992	0.395461122	0.500000000
0.156021993	0.604538848	0.500000000
0.395461122	0.156021993	0.500000000
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0.290396987	0.000000000	0.500000000
0.709602983	0.000000000	0.500000000
0.000000000	0.290396987	0.500000000
0.000000000	0.709602983	0.500000000
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0.669036375	0.669036375	0.500000000
0.669036375	0.330963654	0.500000000
0.330963654	0.669036375	0.500000000
0.500000000	0.500000000	0.500000000

The POSCAR of Ru-C₈N₈.

POSCAR

1.0

7.0798144341	0.0000000000	0.0000000000
0.0000000000	7.0798144341	0.0000000000
0.0000000000	0.0000000000	15.0000000000

C N Ru

8 8 1

Direct

0.156021993	0.395461122	0.500000000
0.843977992	0.604538848	0.500000000
0.604538848	0.156021993	0.500000000
0.395461122	0.843977992	0.500000000
0.843977992	0.395461122	0.500000000
0.156021993	0.604538848	0.500000000
0.395461122	0.156021993	0.500000000
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0.709602983	0.000000000	0.500000000
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0.000000000	0.709602983	0.500000000
0.330963654	0.330963654	0.500000000
0.669036375	0.669036375	0.500000000
0.669036375	0.330963654	0.500000000
0.330963654	0.669036375	0.500000000
0.500000000	0.500000000	0.500000000

(1) The adsorption energies E_{ads} are calculated by the equation.

$$E_{ads} = E_{total} - E_{catalysts} - E_{adsorbates} \quad \#(1)$$

Where E_{total} , $E_{catalysts}$ and $E_{adsorbates}$ represent the energies of the total system after the adsorption, the catalyst, and the adsorbed species, respectively.

(1) The energy of the d band center $E_{d-band\ center}$ can be calculated by the

$$E_{d-band\ center} = \frac{\int E \cdot n(E) dE}{\int n(E) dE} \quad \#(2)$$

following formula:

Where $n(E)$ is the density of states at energy E , and the integration is performed over the entire energy range of the d orbital.

Figure. S5. Based on the Gibbs free energies of adsorbed oxygen-containing intermediates, the scaling relation of ΔG_{OOH^*} vs ΔG_{OH^*} .

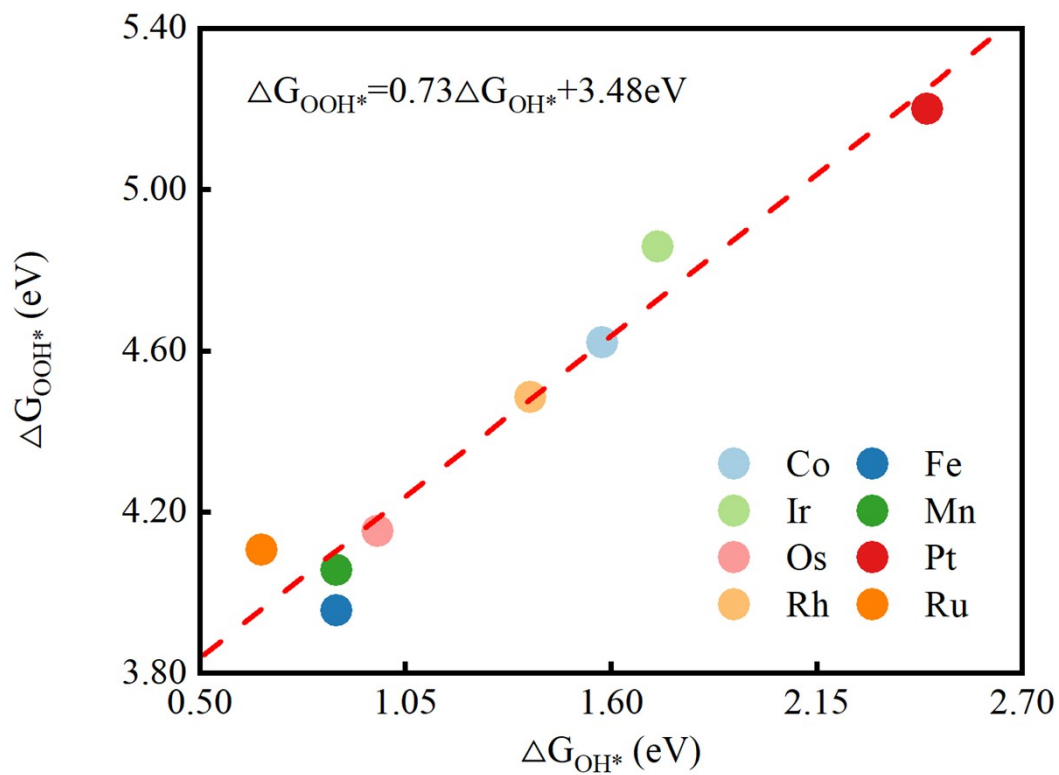


Figure. S6. Schematic of the molecular orbital energy levels of the O₂ molecule.

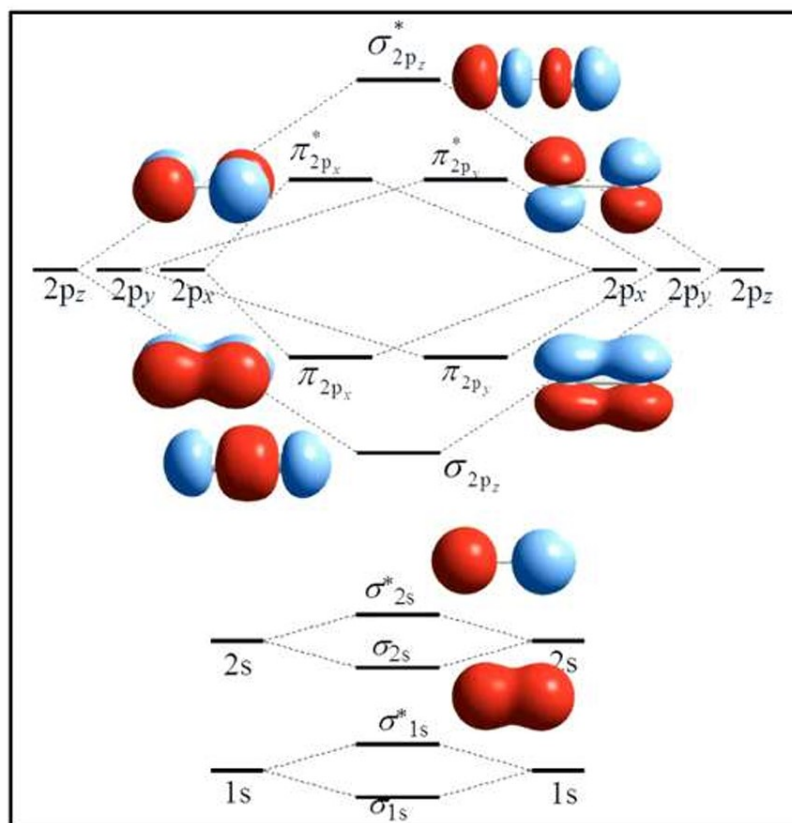


Figure. S7. The major interactions and energy levels of the scalar relativistic Kohn–Sham β -spin of isolated FeO_2 with correlation to the orbitals from Fe and O_2 .

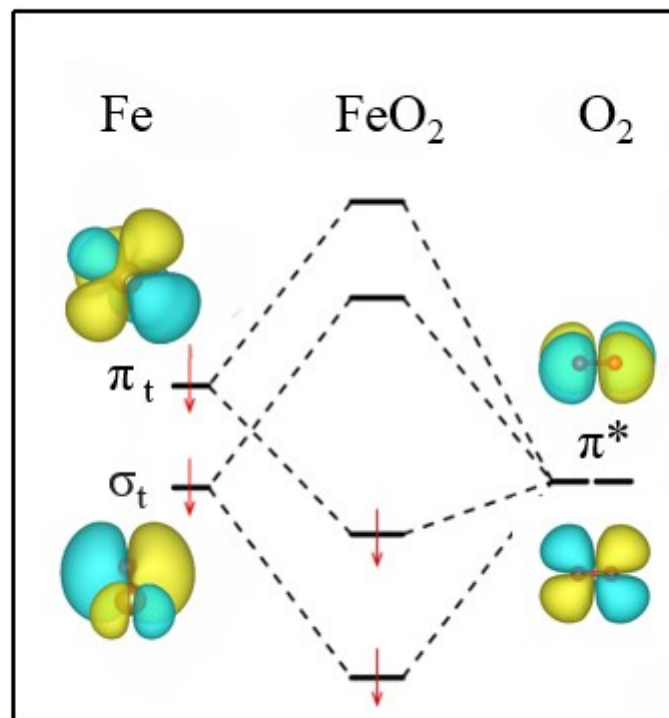


Figure. S8. Schematic interaction of the 3d orbitals of Fe, the 2p orbitals of the O₂ molecule and the projected electronic density of states (PDOS).

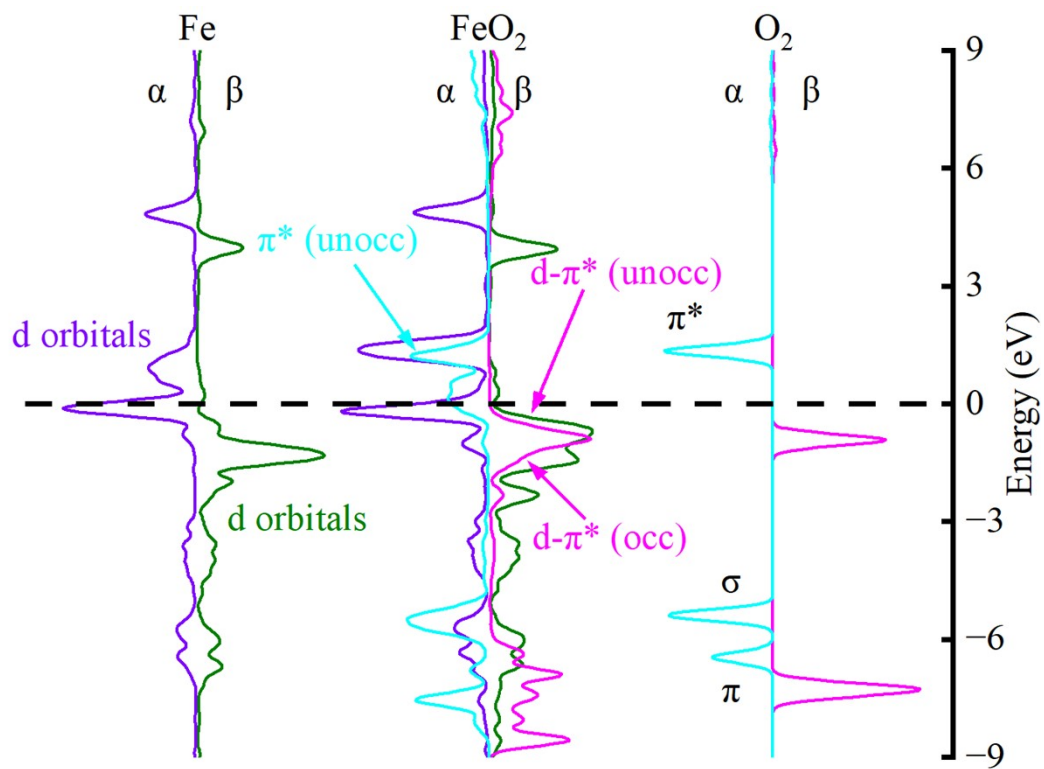


Figure. S9. ENCUT convergence test.

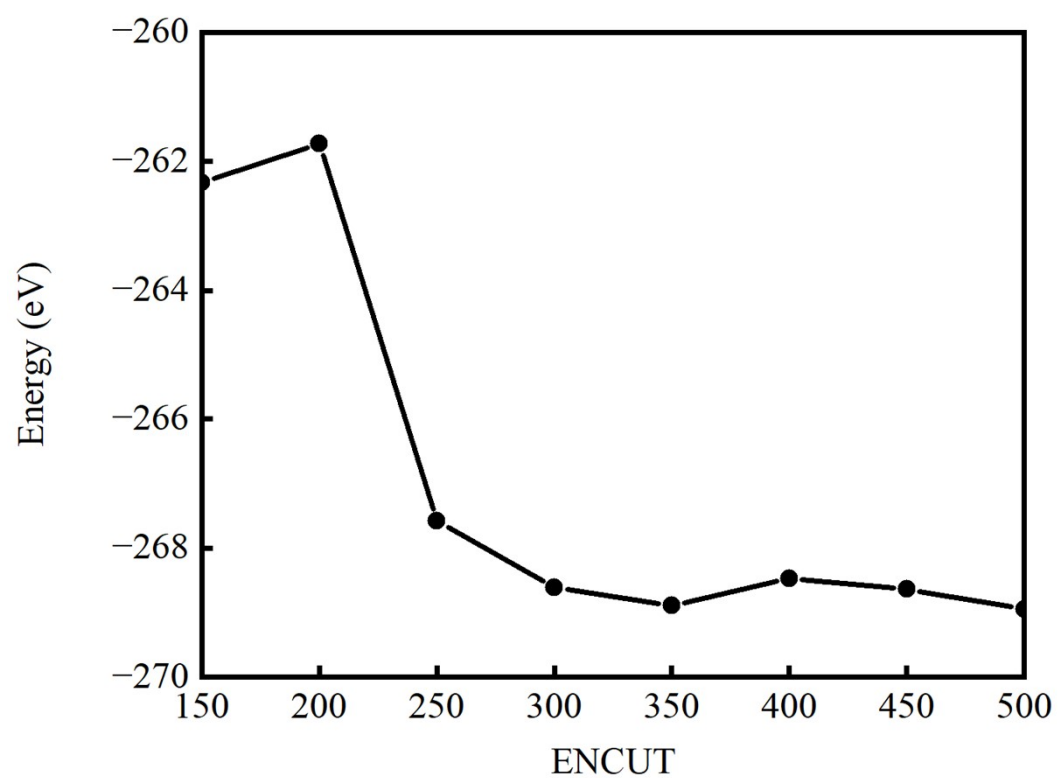


Figure. S10. KPOINTS convergence test.

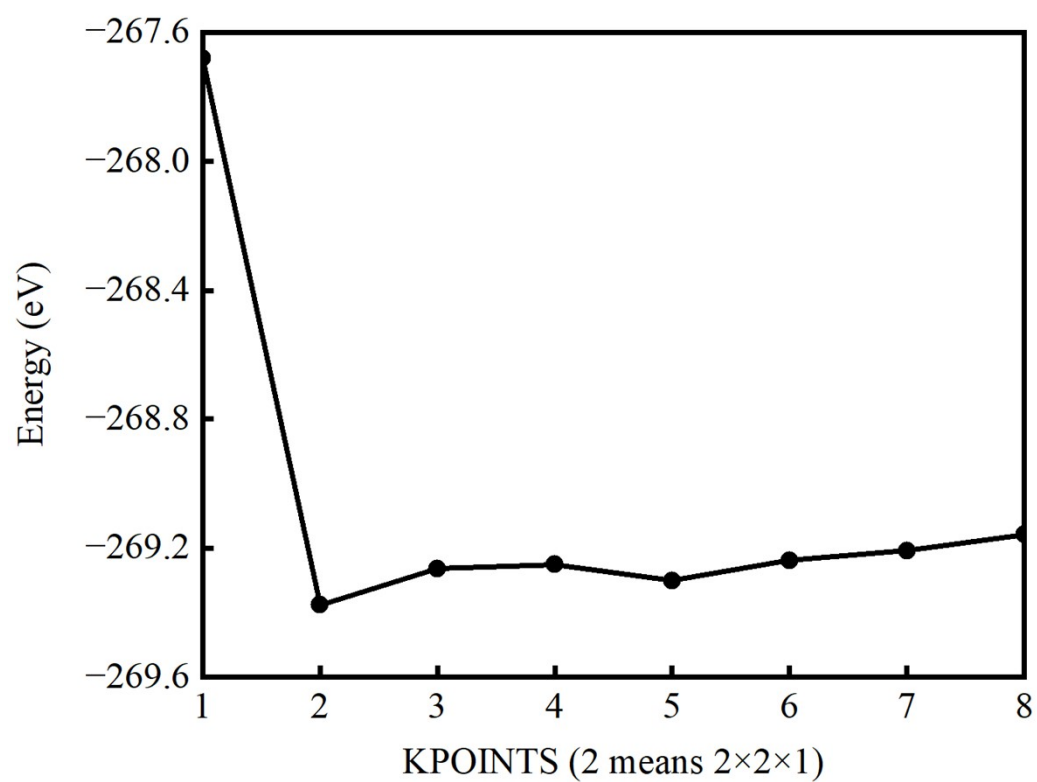


Table S1. Formation energies of TM-C₈N₈ monolayers.

structure	E _a	E _b	E _{atom}	E _f
Sc	-143.24	-132.82	-2.02	-8.40
Ti	-147.93	-132.82	-2.39	-12.72
V	-149.22	-132.82	-3.58	-12.82
Cr	-149.45	-132.82	-5.45	-11.18
Mn	-149.16	-132.82	-5.15	-11.19
Fe	-148.05	-132.82	-3.24	-11.99
Co	-146.71	-132.82	-1.84	-12.05
Ru	-144.34	-132.82	-0.29	-11.23
Cu	-140.64	-132.82	-0.24	-7.58
Zn	-137.38	-132.82	-0.01	-4.55
Y	-138.64	-132.82	-2.08	-3.74
Zr	-144.34	-132.82	-2.25	-9.27
Nb	-146.90	-132.82	-3.18	-10.91
Mo	-147.17	-132.82	-4.65	-9.70
Tc	-147.25	-132.82	-3.28	-11.15
Ru	-145.97	-132.82	-1.21	-11.95
Rh	-144.09	-132.82	-1.26	-10.02
Pd	-141.01	-132.82	-1.48	-6.71
Ag	-135.84	-132.82	-0.20	-2.82
Cd	-133.73	-132.82	-0.01	-0.90

Hf	-146.65	-132.82	-3.15	-10.68
Ta	-148.76	-132.82	-3.50	-12.44
W	-148.71	-132.82	-4.57	-11.31
Re	-148.03	-132.82	-4.61	-10.60
Os	-146.86	-132.82	-2.89	-11.14
Ir	-144.82	-132.82	-1.23	-10.77
Pt	-141.82	-132.82	-0.53	-8.47
Au	-136.68	-132.82	-0.19	-3.67
Hg	-135.21	-132.82	-0.01	-2.38
Rf	-139.20	-132.82	-1.12	-5.26
Db	-139.20	-132.82	-1.12	-5.26
Sg	-139.20	-132.82	-1.12	-5.26
Bh	-139.20	-132.82	-1.12	-5.26
Hs	-139.20	-132.82	-1.12	-5.26
Mt	-139.20	-132.82	-1.12	-5.26
Ds	-139.20	-132.82	-1.12	-5.26
Rg	-139.20	-132.82	-1.12	-5.26
Cn	-139.20	-132.82	-1.12	-5.26

Table S2. Adsorption energies of single O molecule on TM-C₈N₈ monolayers

structure	E _{total}	E _{catalysts}	E _{adsorbates}	E _{ads}
Sc	-147.77	-143.24	-1.54	-2.62
Ti	-154.89	-147.93	-1.54	-4.98
V	-156.58	-149.22	-1.54	-5.38
Cr	-156.40	-149.45	-1.54	-4.96
Mn	-155.37	-149.16	-1.54	-4.22
Fe	-153.03	-148.05	-1.54	-3.00
Co	-150.54	-146.71	-1.54	-1.88
Ni	-146.83	-144.34	-1.54	-0.57
Cu	-142.88	-140.64	-1.54	-0.32
Zn	-139.67	-137.38	-1.54	-0.36
Y	-143.27	-138.64	-1.54	-2.74
Zr	-151.40	-144.34	-1.54	-5.10
Nb	-155.04	-146.90	-1.54	-6.16
Mo	-155.09	-147.17	-1.54	-5.94
Tc	-154.65	-147.25	-1.54	-5.41
Ru	-151.92	-145.97	-1.54	-3.97
Rh	-148.50	-144.09	-1.54	-2.45
Pd	-143.69	-141.01	-1.54	-0.75
Ag	-137.94	-135.84	-1.54	-0.26
Cd	-138.01	-133.73	-1.54	-2.33

Hf	-153.94	-146.65	-1.54	-5.32
Ta	-156.93	-148.76	-1.54	-6.19
W	-156.48	-148.71	-1.54	-5.79
Re	-155.22	-148.03	-1.54	-5.19
Os	-152.82	-146.86	-1.54	-3.99
Ir	-149.32	-144.82	-1.54	-2.54
Pt	-144.77	-141.82	-1.54	-1.02
Au	-138.79	-136.68	-1.54	-0.22
Hg	-137.63	-135.21	-1.54	-0.53
Rf	-140.49	-139.20	-1.54	0.63
Db	-140.49	-139.20	-1.54	0.63
Sg	-140.49	-139.20	-1.54	0.63
Bh	-140.49	-139.20	-1.54	0.63
Hs	-140.49	-139.20	-1.54	0.63
Mt	-140.49	-139.20	-1.54	0.63
Ds	-140.49	-139.20	-1.54	0.63
Rg	-140.49	-139.20	-1.54	0.63
Cn	-140.49	-139.20	-1.54	0.63

Table S3. Comparison of the Calculated ORR Electrocatalytic Performance of TM-C₈N₈ Materials with Experimental Data of TM-PC Materials.

structure	ΔG_{O^*}	ΔG_{OH^*}	ΔG_{OOH^*}	η_{ORR}
Fe- C ₈ N ₈ (theory)	2.16	0.87	3.96	0.26
Fe-PC_(experiment)	1.65	1.00	3.91	0.54
Mn- C ₈ N ₈ (theory)	1.76	0.87	4.06	0.36
Mn-PC_(experiment)	1.65	1.03	4.05	0.56
Co- C ₈ N ₈ (theory)	3.46	1.58	4.62	0.93
Co-PC_(experiment)	3.13	1.40	4.27	0.58
Ru- C ₈ N ₈ (theory)	1.96	0.67	4.11	0.42
Ru-PC_(experiment)	4.49	2.38	5.11	1.42