

Molecular dynamics simulations of discotic ionic liquid crystals: comparison of a full-charge and a scaled-charge force field

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Reference pdb structure of the cation of **2[BF₄]** salt with atom names as in Table S1

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ATOM	72	H21	LC	X	1	69.790	12.240	11.870	0.00	0.00
ATOM	73	H22	LC	X	1	69.680	11.830	13.570	0.00	0.00
ATOM	74	H23	LC	X	1	71.860	10.750	13.440	0.00	0.00
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ATOM	81	H2A	LC	X	1	62.990	11.410	11.260	0.00	0.00
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ATOM	85	H2E	LC	X	1	62.850	13.780	11.820	0.00	0.00
ATOM	86	H2F	LC	X	1	60.690	14.880	11.970	0.00	0.00
ATOM	87	H2G	LC	X	1	60.600	14.480	13.670	0.00	0.00
ATOM	88	H2H	LC	X	1	62.620	15.750	14.100	0.00	0.00
ATOM	89	H2I	LC	X	1	62.730	16.160	12.390	0.00	0.00
ATOM	90	H2J	LC	X	1	60.560	17.260	12.530	0.00	0.00
ATOM	91	H2K	LC	X	1	60.470	16.850	14.240	0.00	0.00
ATOM	92	H2M	LC	X	1	62.500	18.130	14.640	0.00	0.00
ATOM	93	H2N	LC	X	1	62.580	18.540	12.940	0.00	0.00
ATOM	94	H2O	LC	X	1	60.420	19.650	13.100	0.00	0.00
ATOM	95	H2P	LC	X	1	60.330	19.230	14.800	0.00	0.00
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ATOM	97	H2R	LC	X	1	62.460	20.920	13.520	0.00	0.00
ATOM	98	H2S	LC	X	1	58.030	10.890	12.610	0.00	0.00
ATOM	99	H2T	LC	X	1	58.140	11.280	10.920	0.00	0.00
ATOM	100	H2U	LC	X	1	55.980	10.250	10.540	0.00	0.00
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ATOM	102	H2W	LC	X	1	55.980	12.190	12.810	0.00	0.00
ATOM	103	H2X	LC	X	1	56.080	12.600	11.120	0.00	0.00
ATOM	104	H2Y	LC	X	1	53.930	11.550	10.730	0.00	0.00
ATOM	105	H2Z	LC	X	1	53.810	11.160	12.430	0.00	0.00
ATOM	106	H30	LC	X	1	53.920	13.510	12.990	0.00	0.00
ATOM	107	H31	LC	X	1	54.030	13.920	11.290	0.00	0.00
ATOM	108	H32	LC	X	1	51.860	12.870	10.900	0.00	0.00
ATOM	109	H33	LC	X	1	51.760	12.470	12.610	0.00	0.00
ATOM	110	H34	LC	X	1	51.860	14.820	13.180	0.00	0.00
ATOM	111	H35	LC	X	1	51.970	15.230	11.480	0.00	0.00
ATOM	112	H36	LC	X	1	49.810	14.190	11.090	0.00	0.00
ATOM	113	H37	LC	X	1	49.710	13.780	12.800	0.00	0.00
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ATOM	115	H39	LC	X	1	49.910	16.540	11.650	0.00	0.00
ATOM	116	H3A	LC	X	1	59.540	3.700	10.010	0.00	0.00
ATOM	117	H3B	LC	X	1	58.290	4.410	11.020	0.00	0.00
ATOM	118	H3C	LC	X	1	56.710	4.140	7.740	0.00	0.00
ATOM	119	H3D	LC	X	1	57.820	2.970	8.420	0.00	0.00
ATOM	120	H3E	LC	X	1	56.730	6.040	10.060	0.00	0.00
ATOM	121	H3F	LC	X	1	58.090	7.020	9.530	0.00	0.00
ATOM	122	H3G	LC	X	1	58.600	5.490	6.990	0.00	0.00
ATOM	123	H3H	LC	X	1	59.820	6.020	8.110	0.00	0.00
ATOM	124	H3I	LC	X	1	60.310	22.030	13.660	0.00	0.00
ATOM	125	H3J	LC	X	1	60.200	21.630	15.350	0.00	0.00
ATOM	126	H3K	LC	X	1	61.000	23.900	15.050	0.00	0.00
ATOM	127	H3M	LC	X	1	62.230	22.890	15.790	0.00	0.00
ATOM	128	H3N	LC	X	1	62.330	23.300	14.080	0.00	0.00
ATOM	129	H3O	LC	X	1	73.860	14.810	12.730	0.00	0.00
ATOM	130	H3P	LC	X	1	73.750	14.410	14.430	0.00	0.00
ATOM	131	H3Q	LC	X	1	76.030	15.010	13.820	0.00	0.00
ATOM	132	H3R	LC	X	1	75.940	13.340	14.310	0.00	0.00
ATOM	133	H3S	LC	X	1	76.040	13.730	12.610	0.00	0.00
ATOM	134	H3T	LC	X	1	47.750	15.510	11.270	0.00	0.00
ATOM	135	H3U	LC	X	1	47.640	15.100	12.970	0.00	0.00
ATOM	136	H3V	LC	X	1	46.400	17.130	12.480	0.00	0.00
ATOM	137	H3W	LC	X	1	47.750	17.460	13.540	0.00	0.00
ATOM	138	H3X	LC	X	1	47.860	17.870	11.840	0.00	0.00

END

Table S1. FF parameters obtained from the LigParGen web site^[1]

name	Mass / a.u.	ϵ / kJmol ⁻¹	σ / nm	q / e
C00	12.0110	3.55000E-01	2.92880E-01	0.3271
C01	12.0110	3.55000E-01	2.92880E-01	0.0099
C02	12.0110	3.55000E-01	2.92880E-01	0.3271
C03	12.0110	3.55000E-01	2.92880E-01	-0.3970
C04	12.0110	3.55000E-01	2.92880E-01	0.1075
C05	12.0110	3.55000E-01	2.92880E-01	-0.3970
O06	15.9990	2.90000E-01	5.85760E-01	-0.3624
C07	12.0110	3.50000E-01	2.76144E-01	0.2313
C08	12.0110	3.50000E-01	2.76144E-01	-0.1006
C09	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0A	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0B	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0C	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0D	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0E	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0F	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0G	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0H	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0I	12.0110	3.50000E-01	2.76144E-01	-0.2220
O0J	15.9990	2.90000E-01	5.85760E-01	-0.2994
C0K	12.0110	3.50000E-01	2.76144E-01	0.3207
C0M	12.0110	3.50000E-01	2.76144E-01	-0.2190
C0N	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0O	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0P	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0Q	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0R	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0S	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0T	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0U	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0V	12.0110	3.50000E-01	2.76144E-01	-0.1480
C0W	12.0110	3.50000E-01	2.76144E-01	-0.2220
C0X	12.0110	3.50000E-01	2.76144E-01	-0.0079
N0Y	14.0070	3.25000E-01	7.11280E-01	0.2035
C0Z	12.0110	3.50000E-01	2.76144E-01	-0.2118
H10	1.0080	2.50000E-01	1.25520E-01	0.1261
O11	15.9990	2.90000E-01	5.85760E-01	-0.2994
C12	12.0110	3.50000E-01	2.76144E-01	0.3207
C13	12.0110	3.50000E-01	2.76144E-01	-0.2190
C14	12.0110	3.50000E-01	2.76144E-01	-0.1480
C15	12.0110	3.50000E-01	2.76144E-01	-0.1480
C16	12.0110	3.50000E-01	2.76144E-01	-0.1480
C17	12.0110	3.50000E-01	2.76144E-01	-0.1480
C18	12.0110	3.50000E-01	2.76144E-01	-0.1480
C19	12.0110	3.50000E-01	2.76144E-01	-0.1480
C1A	12.0110	3.50000E-01	2.76144E-01	-0.1480
C1B	12.0110	3.50000E-01	2.76144E-01	-0.1480
C1C	12.0110	3.50000E-01	2.76144E-01	-0.1480
C1D	12.0110	3.50000E-01	2.76144E-01	-0.2220
C1E	12.0110	3.50000E-01	2.76144E-01	-0.2118
H1F	1.0080	2.50000E-01	1.25520E-01	0.1261
C1G	12.0110	3.50000E-01	2.76144E-01	-0.2118
H1H	1.0080	2.50000E-01	1.25520E-01	0.1261
H1I	1.0080	2.42000E-01	1.25520E-01	0.1443
H1J	1.0080	2.42000E-01	1.25520E-01	0.1443
H1K	1.0080	2.50000E-01	1.25520E-01	-0.0101
H1M	1.0080	2.50000E-01	1.25520E-01	-0.0101
H1N	1.0080	2.50000E-01	1.25520E-01	-0.0012
H1O	1.0080	2.50000E-01	1.25520E-01	-0.0012
H1P	1.0080	2.50000E-01	1.25520E-01	-0.0101

H3O	1.0080	2.50000E-01	1.10000E-01	0.0740
H3P	1.0080	2.50000E-01	1.10000E-01	0.0740
H3Q	1.0080	2.50000E-01	1.10000E-01	0.0740
H3R	1.0080	2.50000E-01	1.10000E-01	0.0740
H3S	1.0080	2.50000E-01	1.10000E-01	0.0740
H3T	1.0080	2.50000E-01	1.10000E-01	0.0740
H3U	1.0080	2.50000E-01	1.10000E-01	0.0740
H3V	1.0080	2.50000E-01	1.25520E-01	0.0740
H3W	1.0080	2.50000E-01	1.10000E-01	0.0740
H3X	1.0080	2.50000E-01	1.10000E-01	0.0740
B	10.810	0.355000	0.29288	1.177448
F	18.9984	0.31181	0.25522	-0.544362

^[1] a) Potential energy functions for atomic-level simulations of water and organic and biomolecular systems. Jorgensen, W. L.; Tirado-Rives, J. Proc. Nat. Acad. Sci. USA 2005, 102, 6665-6670

b) 1.14*CM1A-LBCC: Localized Bond-Charge Corrected CM1A Charges for Condensed-Phase Simulations. Dodda, L. S.; Vilseck, J. Z.; Tirado-Rives, J.; Jorgensen, W. L. J. Phys. Chem. B, 2017, 121 (15), pp 3864-3870

c) LigParGen web server: An automatic OPLS-AA parameter generator for organic ligands. Dodda, L. S.; Cabeza de Vaca, I.; Tirado-Rives, J.; Jorgensen, W. L. Nucleic Acids Research, Volume 45, Issue W1, 3 July 2017, Pages W331-W336

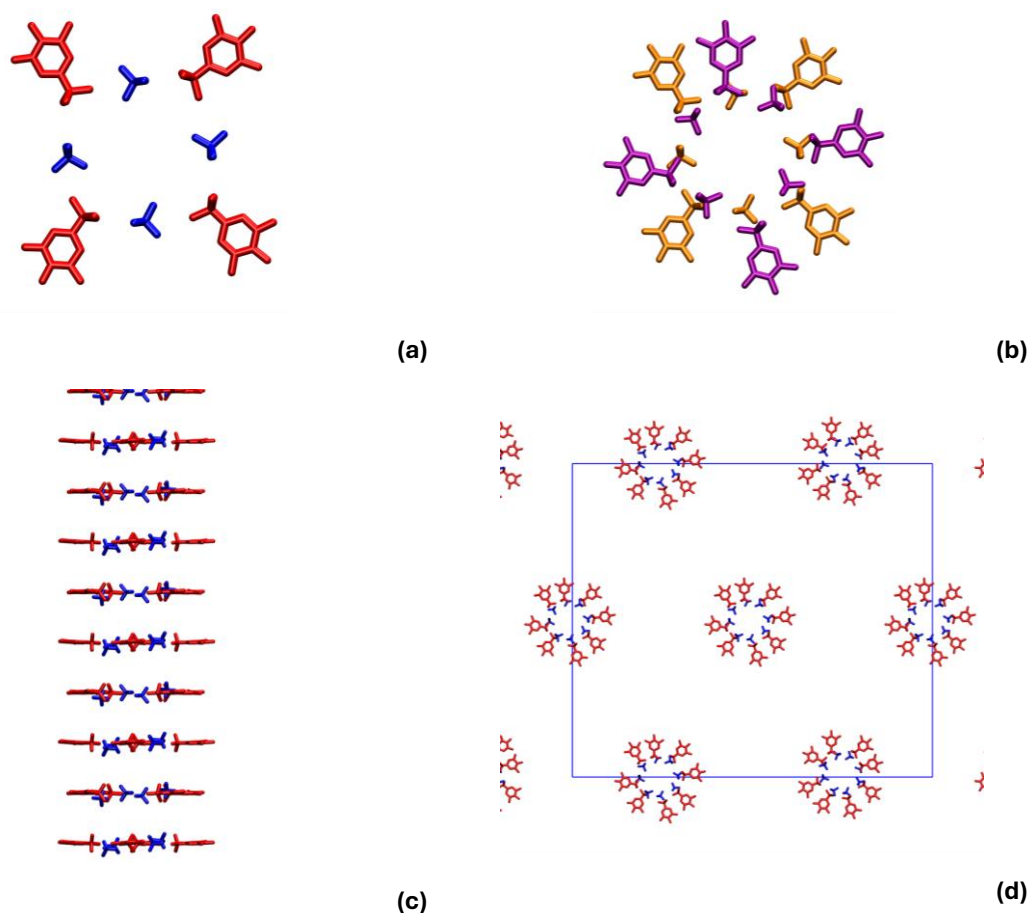


Figure S1. Discotic unit composed of 4 gallic acid derivatives (this example for compound **1** in Figure 1 of the main text) and 4 tetrafluoroborate anions, (a), respectively represented as red and blue sticks. Two discotic units stacked one on top of the other, (b), with the orange unit being below the purple one. Multiple stacked disks forming the column, (c), with the same color code of (a). Hexagonal arrangement of the columns seen from above with the simulation box indicated as blue line and with the closest periodic images represented, (d). In all these images the alkyl chains are not shown for the sake of clarity.

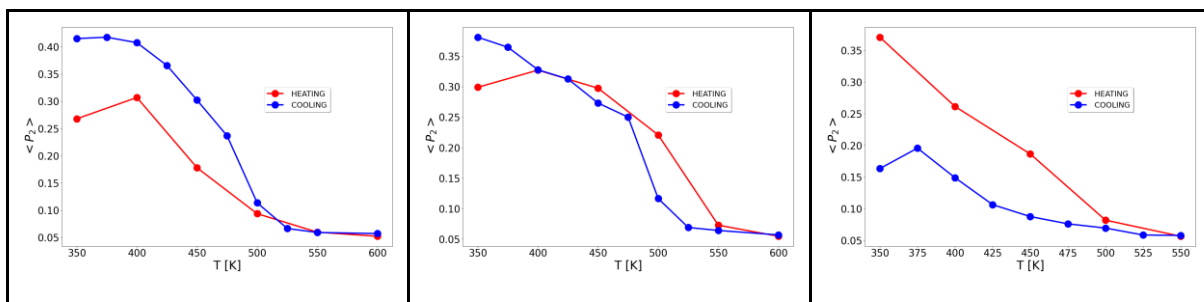


Figure S2. Orientational order parameters, $\langle P_2 \rangle$, in heating (red circles) and cooling (blue circles) run. The lines are a guide to the eye. From left to right: 1[BF₄], 2[BF₄] and 3[BF₄]. Small boxes made of 320 ion pairs.

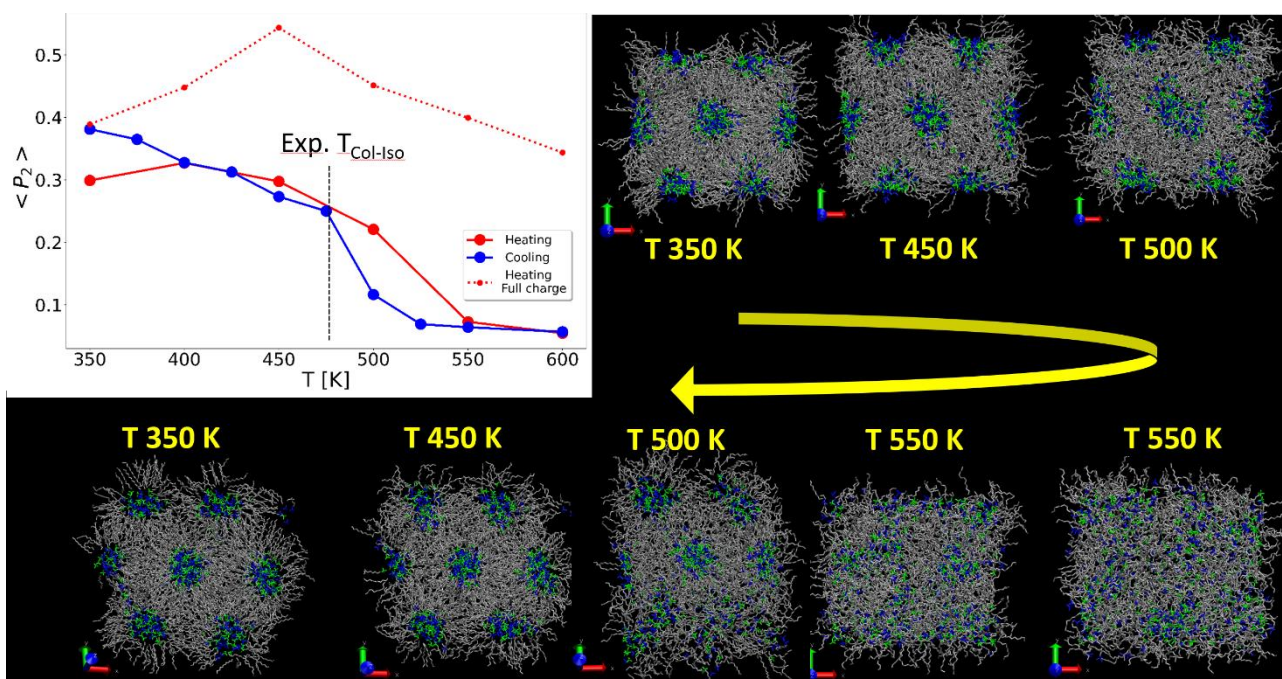


Figure S3. History of the thermal treatment of the small box (320 ion pairs) of 2[BF₄]. The system is simulated for 100 ns at each temperature first in heating, up to 600 K, then in cooling, starting each simulation from the last configuration of the previous temperature run. The last 50 ns of each run are used to calculate the orientational order parameter, $\langle P_2 \rangle$, of the vector normal to the phenyl rings.

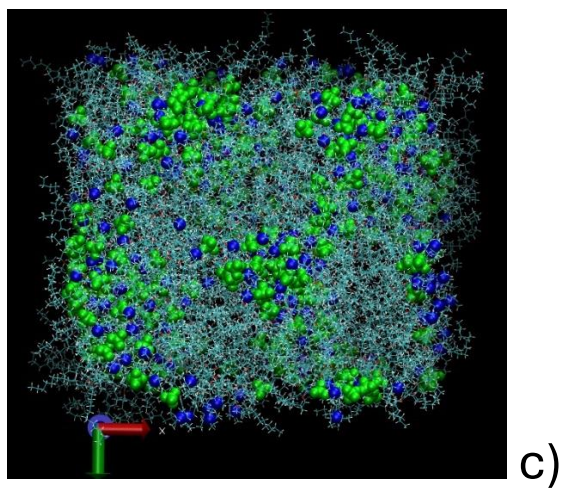
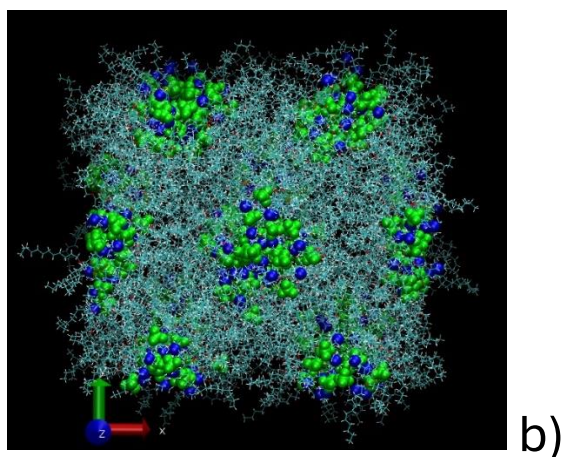
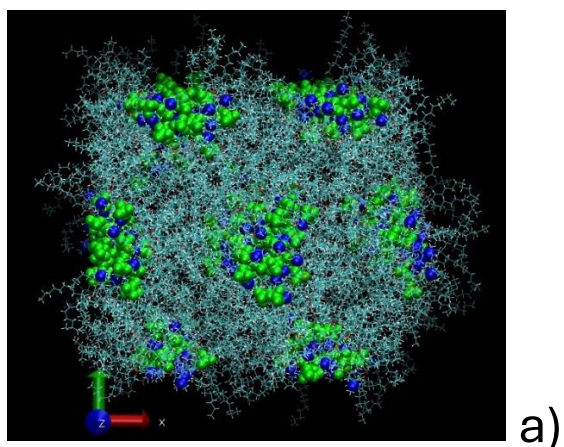


Figure S4. Snapshots of the final configurations at 600 K of the heating runs using the full-charge Force Field and small boxes of 320 ion pairs. a) **1[BF₄]**; b) **2[BF₄]**; c) **3[BF₄]**. The hexagonal packing of the ionic moieties (tetrafluoroborate in green and the nitrogen of the ammonium group in blue) is clearly visible, indicating the lack of melting even at 600 K, although **3[BF₄]** appears less ordered than the other systems.