

## Supporting Information

# Designing of *in-situ* nanostructured MWCNT- Phloroglucinol modified webs for electrochemical-based dual screening of stress biomarkers

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**Statement of the problem addressed/Originality of the approach:** This work accounts for the selective sensing of stress biomarker *i.e.*, glutathione (GT) and hydrogen peroxide ( $H_2O_2$ ) both in pH 7 buffer media. This work involves an eco-friendly approach for a Point-of-care testing of specific analytes in the presence of interferences.

**Contribution of the work to create new knowledge in the field:** This work incorporated the fabrication of a novel *in-situ* electrochemical platform involving transformation of a phyto-constituent *i.e.*, Phloroglucinol (benzene 1,3,5-triol) using Cyclic Voltammetric Technique in aqueous media on a GCE platform. This concept enables a sustainable approach using a glassy carbon modified multiwalled carbon nanotube-chitosan polymer based electrode (GCE/MWCNT-Chit@PG-Redox) based system.

**Relevance of the work to advance research and impact to the field of food science and technology:** This work is quite relevant for the development of a sustainable platform, which is eco-friendly and food-science-derived phloroglucinol-based sensor. Phloroglucinol, found in various plants, is an essential phenolic compound and a vital tannin.

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**SF1:** Scanning electron microscopy images

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**SF5:** Real sample analysis

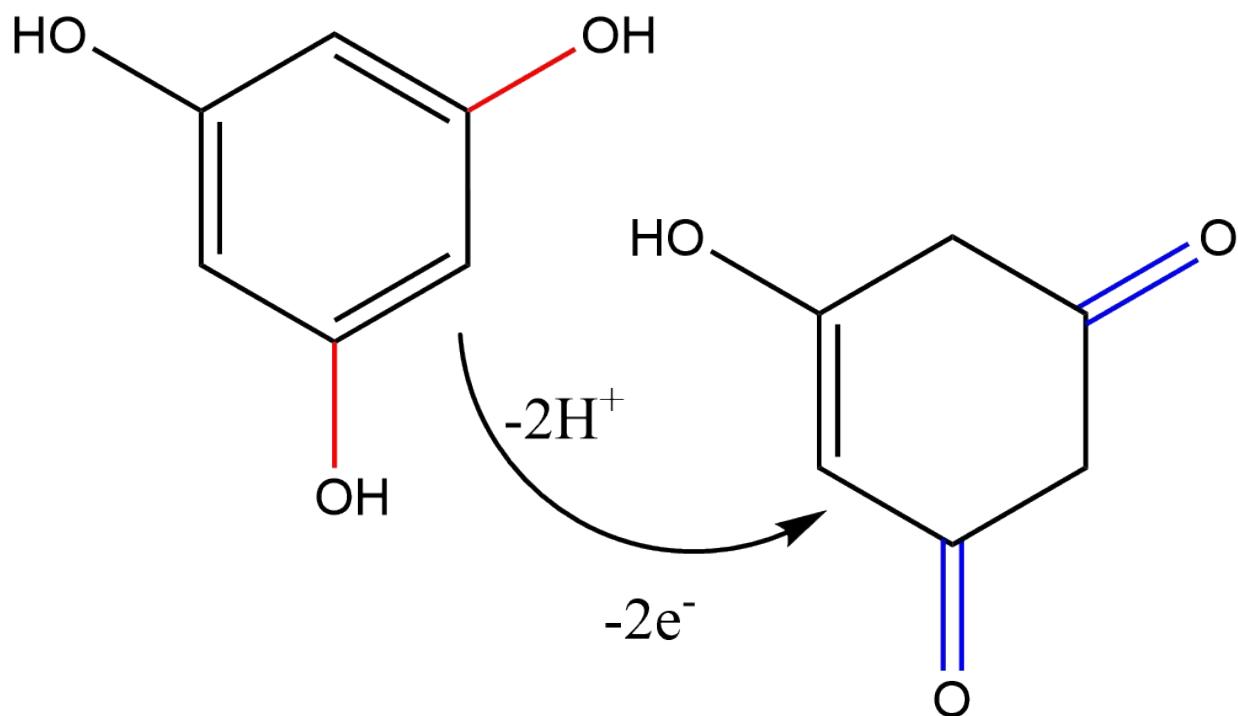
**SF6:** Differential pulse voltammetric (DPV) signals of GCE/MWCNT-Chit@PG-Redox platform using H<sub>2</sub>O<sub>2</sub> concentration range in buffer media, its calibration plot and its interference effect.

**SF7:** Real sample analysis

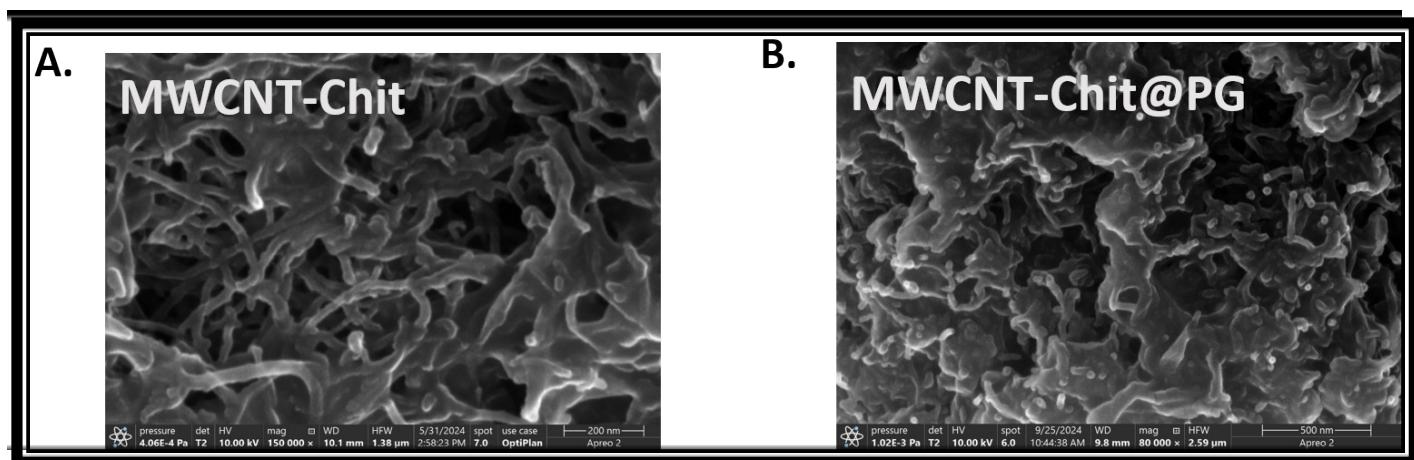
**ST1:** Tables for real sample analysis (GT)

**ST1:** Tables for real sample analysis (H<sub>2</sub>O<sub>2</sub>)

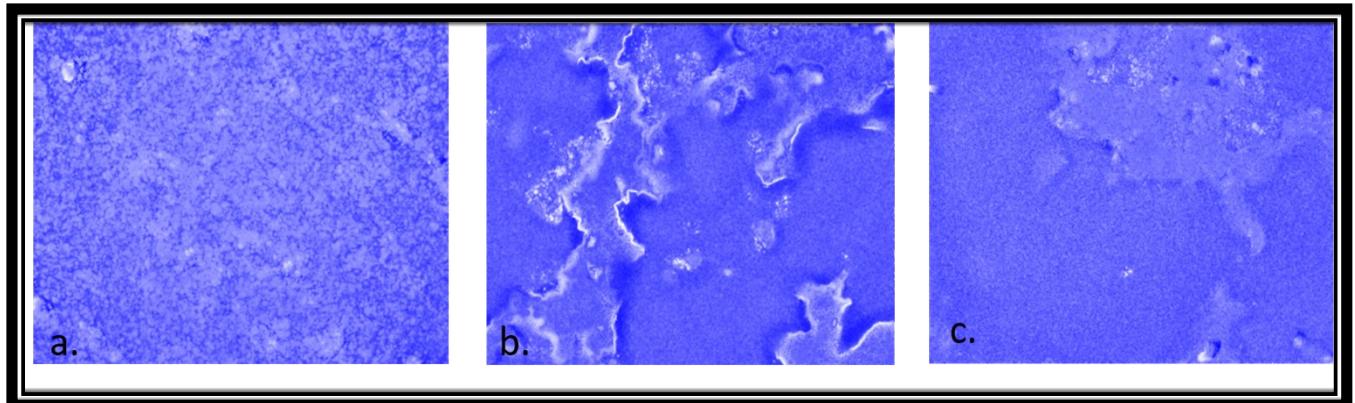
## List of Supplementary Schemes, Figures and Tables



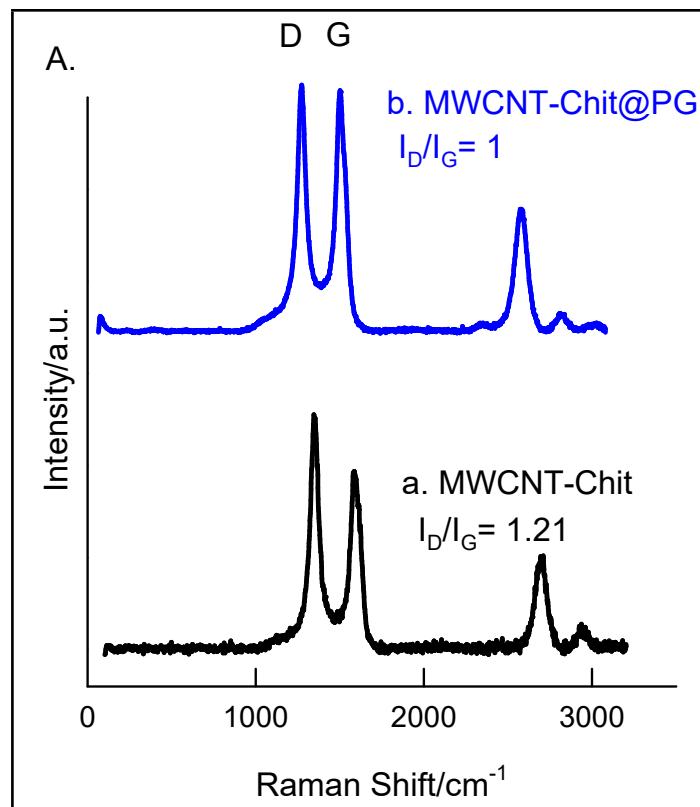
**Supplementary Scheme. S1** Schematic representation for the organic transformation of phloroglucinol under electrochemical treatment in pH 7 aqueous buffer media.



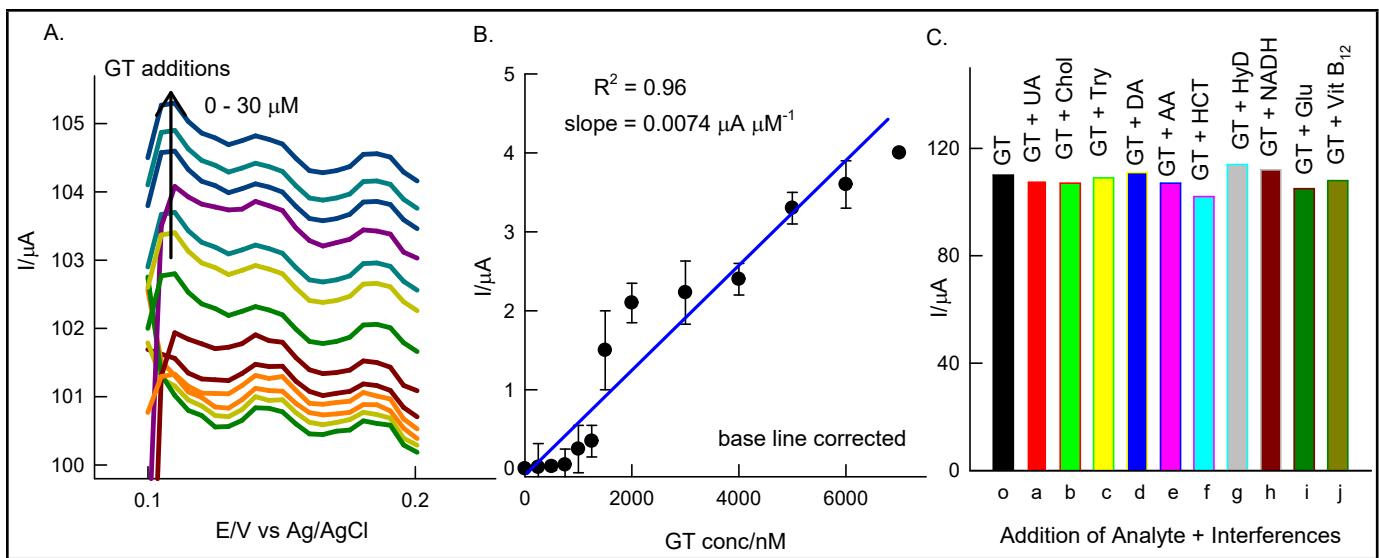
**Supplementary Figure. SF1** Scanning electron microscopic images for the MWCNT-Chit (a.) and MWCNT-Chit@PG-redox (b.).



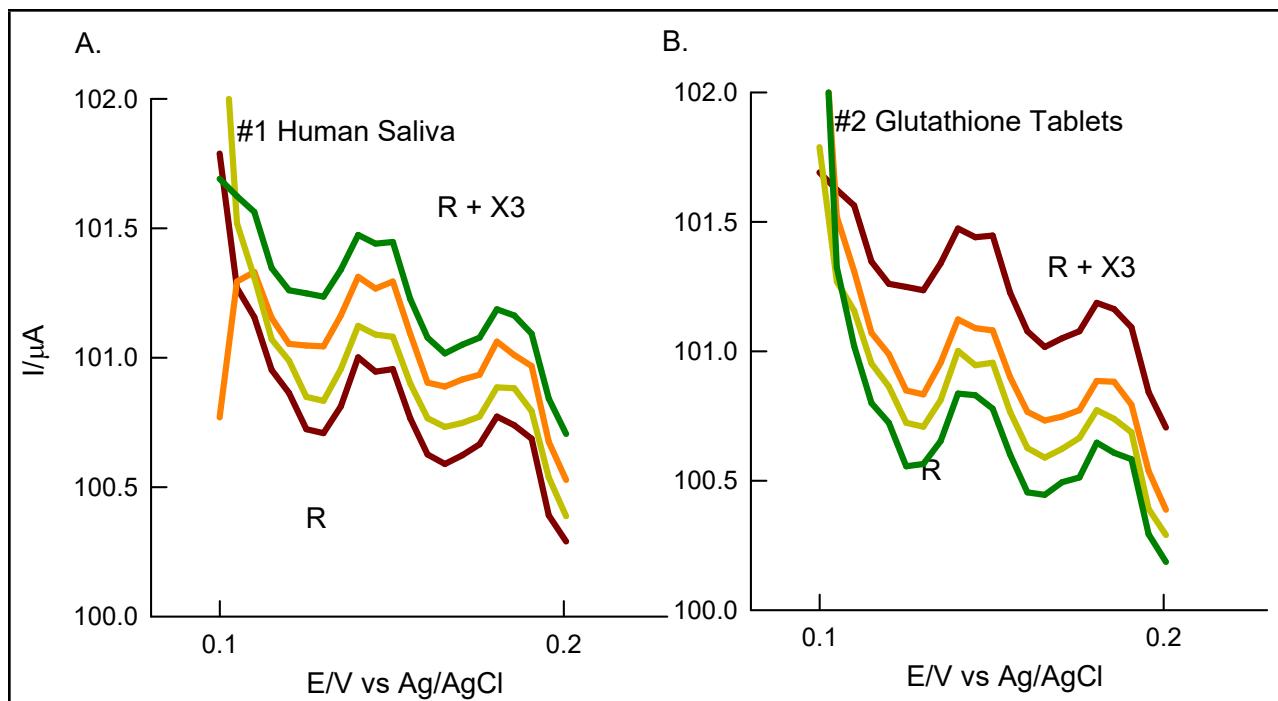
**Supplementary Figure. SF2** Optical Profilometry 2D images of screen printed carbon electrode (SPCE) (a), MWCNT-Chit (b) and MWCNT-Chit@PG (c) prepared in a SPCE surface.



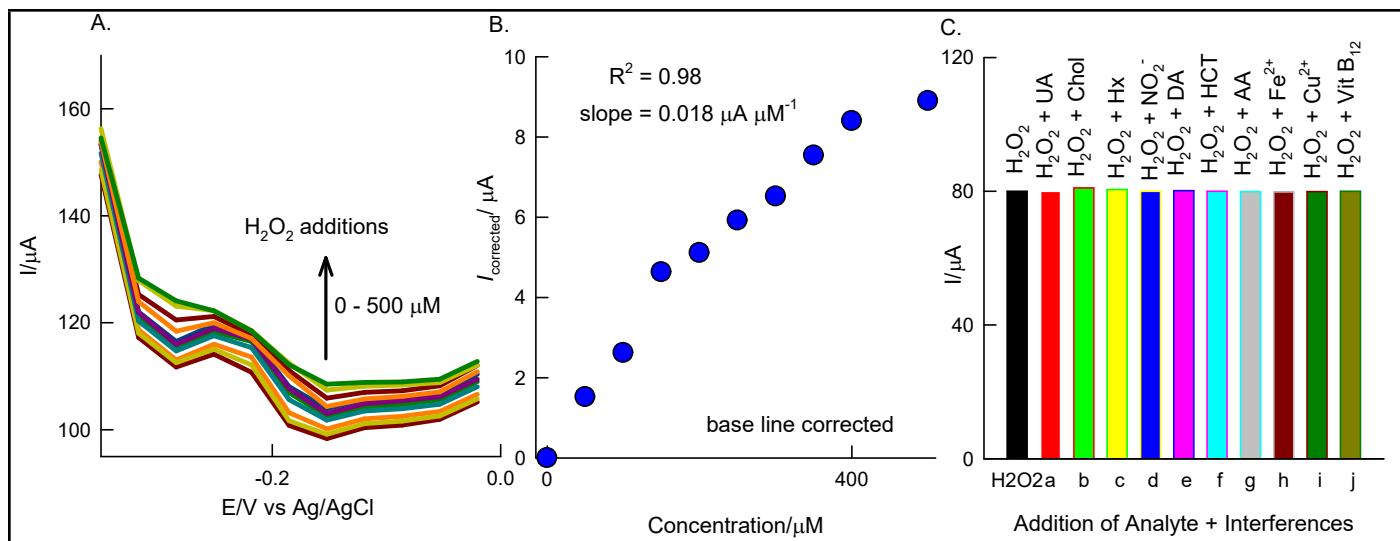
**Supplementary Figure. SF3** Raman spectroscopic response of SPCE/MWCNT-Chit (curve a) and SPCE/MWCNT-Chit@PG (curve b). Note: SPCE= Screen printed carbon electrodes.



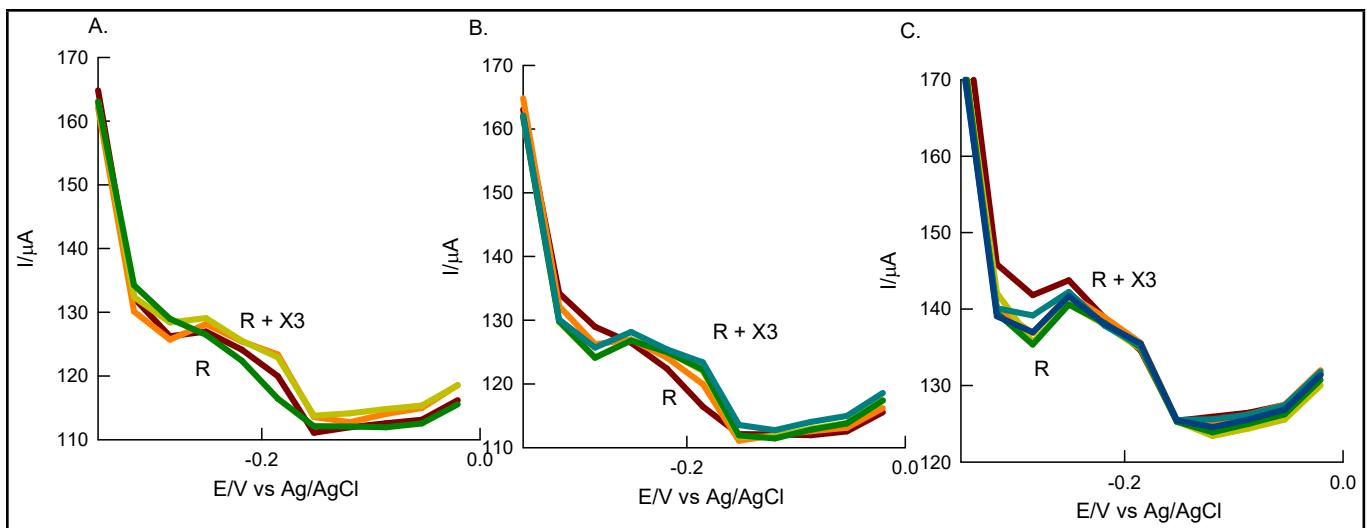
**Supplementary Figure. SF4** (A) Differential pulse voltammetric (DPV) signals of GCE/MWCNT-Chit@PG-Redox platform using GT concentration range in buffer media Note: Optimal DPV conditions are: initial potential= 0 V; final potential= 0.2 V; increment potential= 0.004 V; amplitude= 0.05 V; pulse width= 0.06 s; pulse period= 0.5 s. (B) Respective calibration graph for DPV concentration effect with their error bars after triplicate analysis. (C) Effect of interfering chemicals along with glutathione (GT) (hydrodynamic conditions) in buffer media.



**Supplementary Figure. SF5** (A) Differential pulse voltammetric (DPV) signals of GCE/MWCNT-Chit@PG-Redox platform using real samples (#1 Human Saliva, #2 Glutathione Tablets using standard addition approach with successive additions of GT dissolved in buffer media after 1:1 dilution). Note: Optimal DPV conditions are: initial potential= 0 V; final potential= 0.2 V; increment potential= 0.004 V; amplitude= 0.05 V; pulse width= 0.06 s; pulse period= 0.5 s.



**Supplementary Figure. SF6** (A) Differential pulse voltammetric signals GCE/MWCNT-Chit@PG-Redox platform on successive additions of hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) dissolved in pH 7 PBS. (B) Plots of  $i_{\text{pa}}$  vs. analyte additions (base line corrected). (C) Effect of interfering chemicals along with hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) (hydrodynamic conditions) in buffer media. Note: Optimal DPV conditions are: initial potential= 0.4 V; final potential= 0 V; increment potential= 0.004 V; amplitude= 0.05 V; pulse width= 0.06 s; pulse period= 0.5 s.



**Supplementary Figure. SF7** Differential pulse voltammetric (DPV) signals of GCE/MWCNT-Chit@PG-Redox platform using real samples (#1 Orange juice (packaged), #2 Orange juice (real), #3 Milk sample) using standard addition approach with successive additions of hydrogen peroxide ( $\text{H}_2\text{O}_2$ ) dissolved in buffer media after 1:1 dilution. DPV parameters as SF4.

**Supplementary Table ST1** Detection of glutathione (GT) added and found in various real samples (#1 Human Saliva, #2 Glutathione Tablets using standard addition approach with successive additions of GT dissolved in buffer media after 1:1 dilution with buffer

S. No.	Sample	Analyte	Spiked Value (nM)	Found Value (nM)	Recovery Percentage (%)
1.	Human Saliva	R	-		
		R +X1	150	170	120
		R+X2	300	300	100
		R+X3	500	680	134
2.	Glutathione Tablet	R	-		
		R +X1	150	180	120
		R+X2	300	300	100
		R+X3	700	690	98.7

**Supplementary Table ST2** Detection of hydrogen peroxide ( $H_2O_2$ ) added and found in various real samples (#1 Orange juice (packaged), #2 Orange juice (real), #3 Milk samples) after 1:1 dilution with buffer

S. No.	Sample	Analyte	Spiked Value ( $\mu M$ )	Found Value ( $\mu M$ )	Recovery Percentage (%)
1.	Orange juice (packaged) #1	R	-		
		R +X1	50	45	90
		R+X2	100	88.5	88.5
		R+X3	250	245	98
2.	Orange juice (real) #2	R	-		
		R +X1	100	110	110
		R+X2	150	170	113
		R+X3	200	205	102.5
3.	Milk samples #3	R	-		
		R +X1	50	43.25	86.5
		R+X2	100	91	91
		R+X3	150	110	73.5