

Supplementary information for

**Ionic Liquid-Mediated CO₂/N₂ Separation in MoSe₂ Nanochannels: A
Molecular Dynamics Perspective**

Xun Liu^{1#}, Shuang Wang^{1#}, Pan He¹, Zhirui Huang¹, Libo Li^{2}, Daohui Zhao^{1*}*

¹ Hubei Collaborative Innovation Center for Advanced Organic Chemical Materials, Ministry of Education Key Laboratory for the Synthesis and Application of Organic Functional Molecules, School of Chemistry and Chemical Engineering, Hubei University, Wuhan 430062, China

² State Key Laboratory of Pulp and Paper Engineering, School of Chemistry & Chemical Engineering, Guangdong Provincial Key Lab of Green Chemical Product Technology, South China University of Technology, Guangzhou, 510610, China

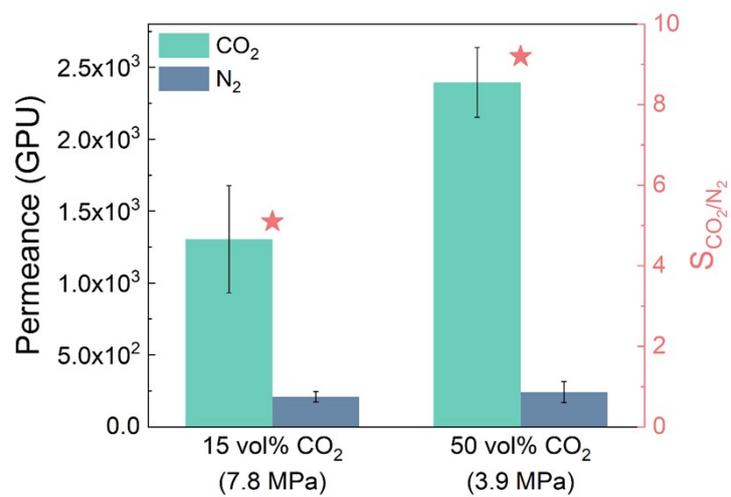


Figure S1 Permeance and selectivity of gases under different feed ratios and pressures.

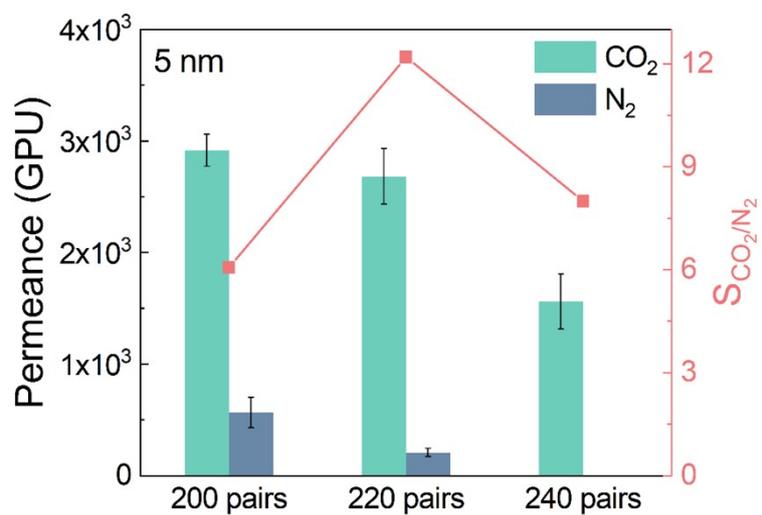


Figure S2 Gas permeance and separation selectivity at an interlayer spacing of 5 nm.

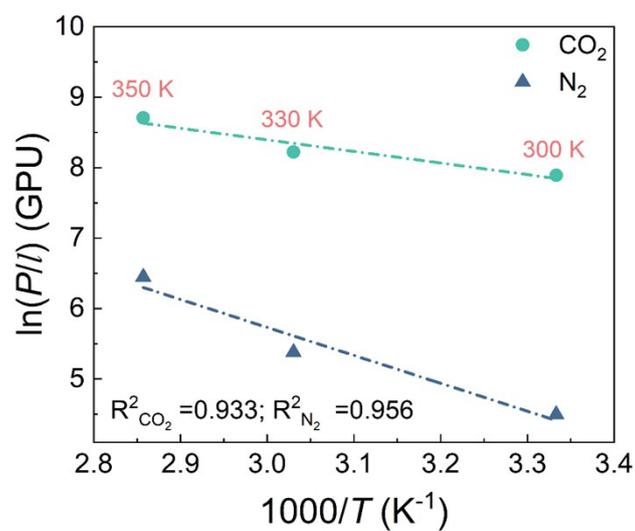


Figure S3 Arrhenius plots for gas permeance at an interlayer spacing of 4 nm.

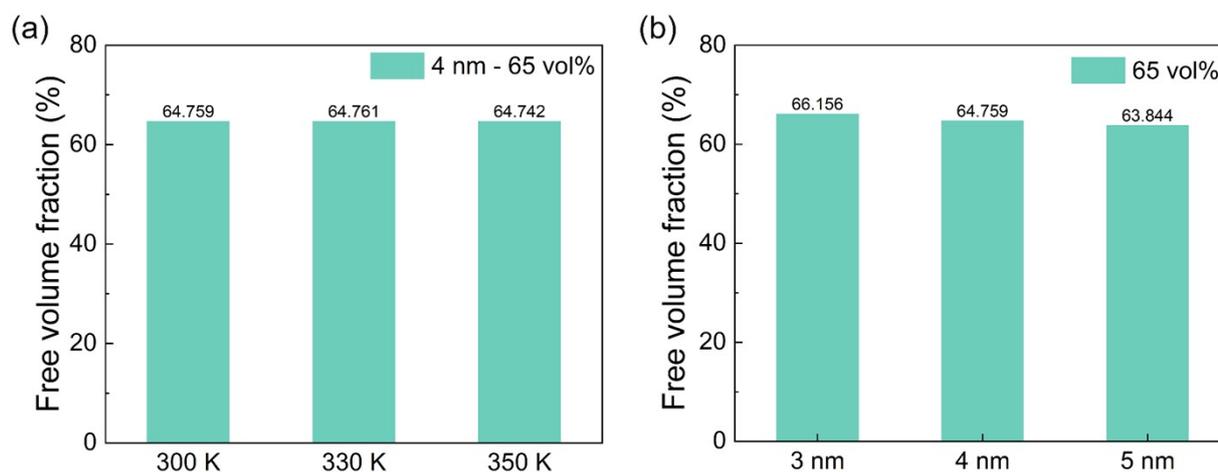


Figure S4 Free volume fraction within the nanochannel as a function of (a) temperature and (b) interlayer spacing.

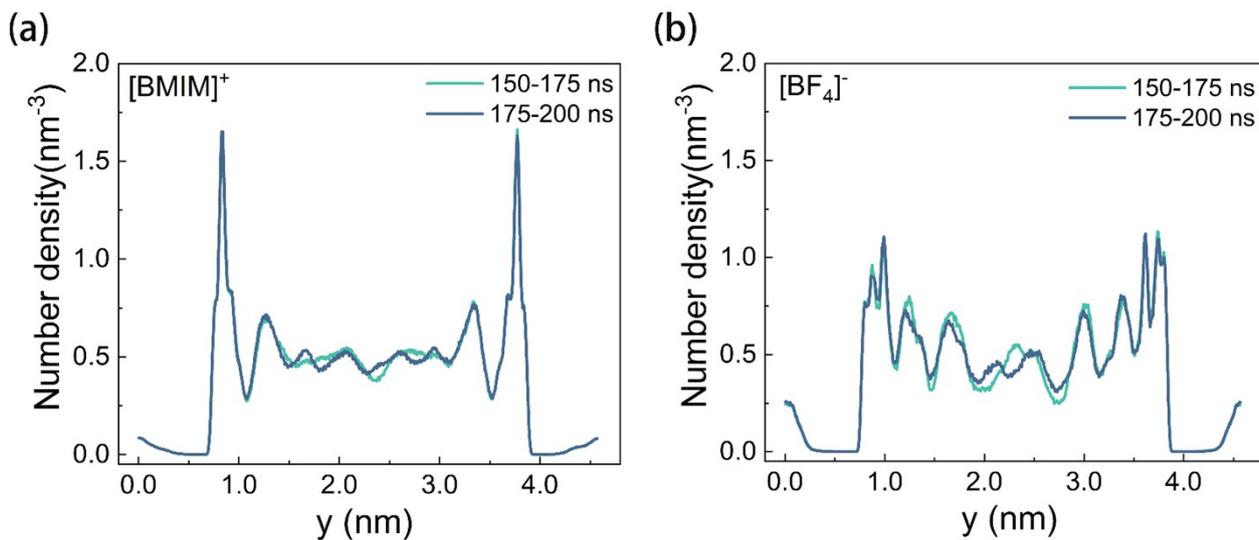


Figure S5 Number density distributions of (a) [BMIM]⁺ and (b) [BF₄]⁻ over two consecutive 25 ns intervals from the 150-200 ns simulation trajectory of the IL system.

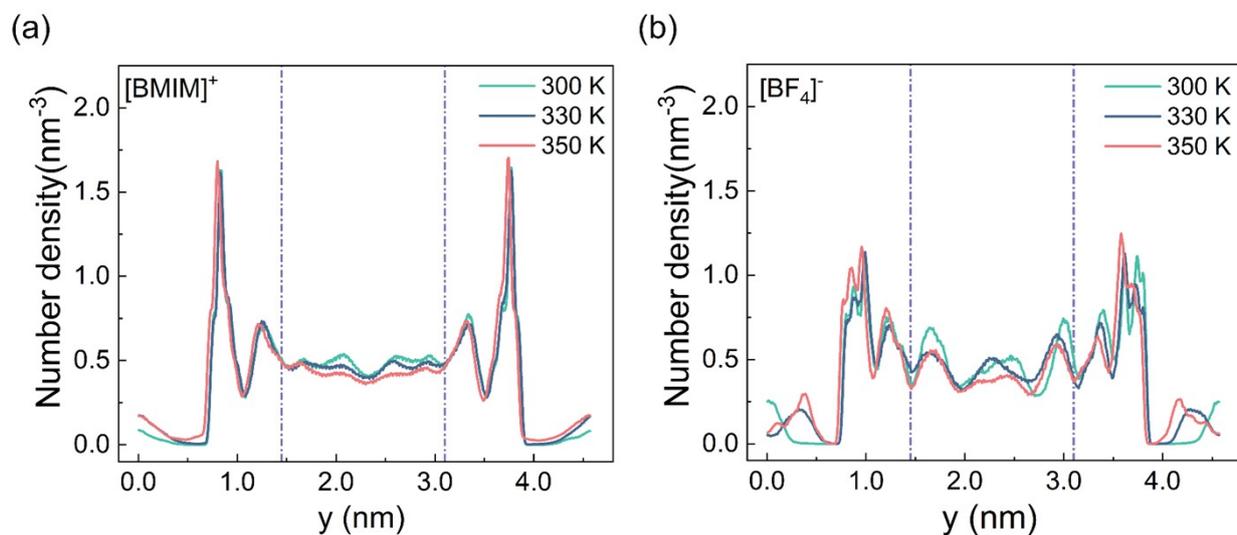


Figure S6 Number density distributions of (a) cations and (b) anions along the y -axis at different temperatures under a fixed interlayer spacing of 4 nm.

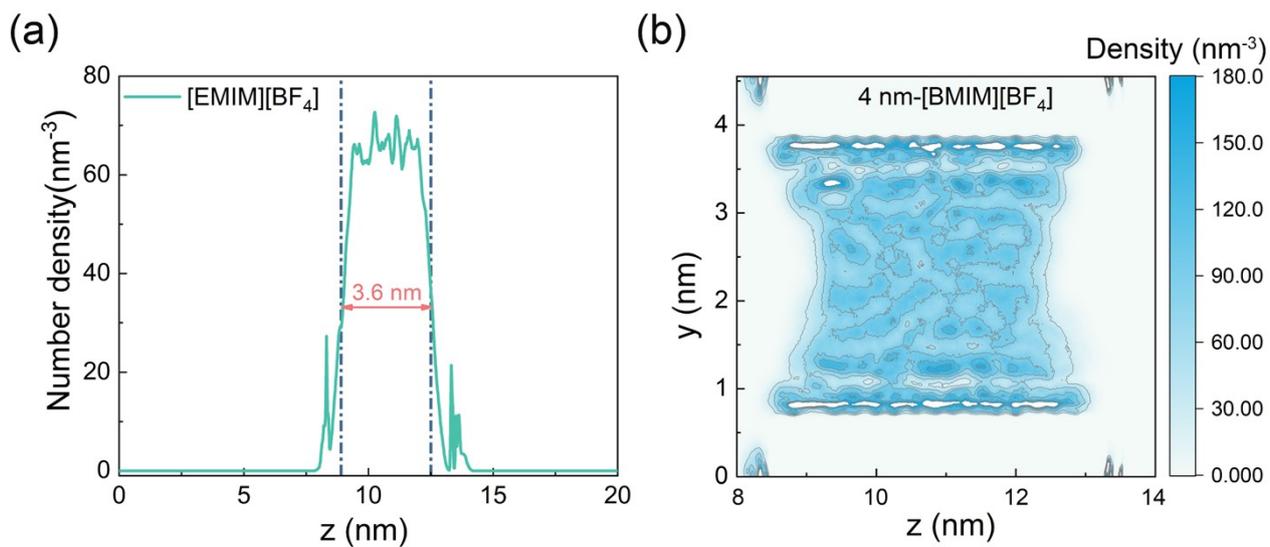


Figure S7 (a) Number density distribution of ILs along the z -axis and (b) the corresponding 2D density distribution at an interlayer spacing of 4 nm.

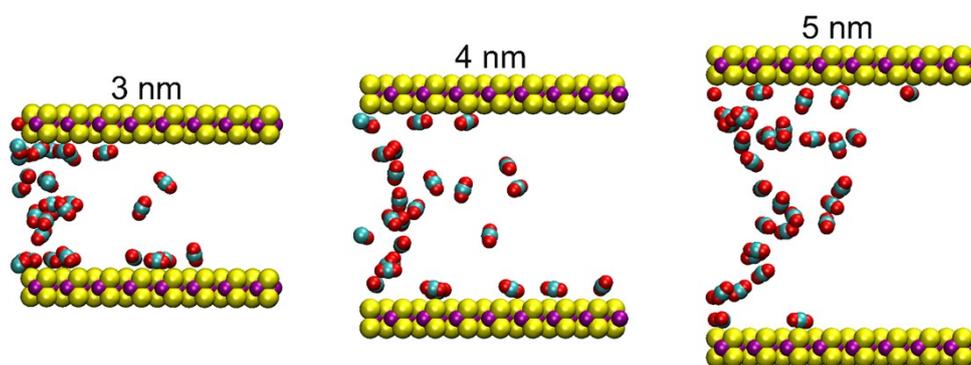


Figure S8 Snapshot illustrating the distribution of CO₂ within the nanochannels.

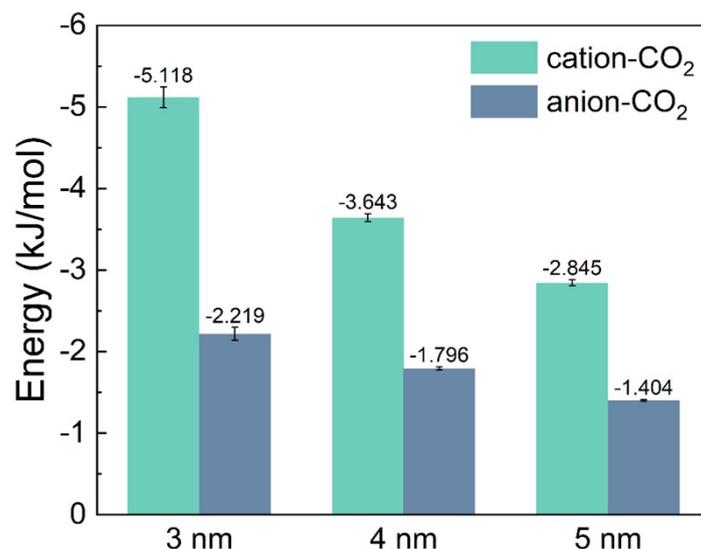


Figure S9 Interaction energy between CO₂ and cations/anions at different interlayer spacings.

Table S1 The permeability, diffusion coefficients, solubility coefficients, and corresponding selectivity of CO₂ and N₂ at three different layer spacings.

Interlayer spacing	Gas	Permeability (Barrer)	Diffusion coefficient (cm ² /s)	Solubility coefficient (cm ³ (STP [*])/(cm ³ ·cmHg))
3 nm	CO ₂	6.91	0.9168 × 10 ⁻⁵	7.54 × 10 ⁻⁵
	N ₂	0.215	13.17 × 10 ⁻⁵	1.63 × 10 ⁻⁷
	selectivity	32.14	0.0696	462.58
4 nm	CO ₂	9.61	4.446 × 10 ⁻⁵	2.16 × 10 ⁻⁵
	N ₂	0.322	26.66 × 10 ⁻⁵	1.21 × 10 ⁻⁷
	selectivity	29.84	0.167	178.51
5 nm	CO ₂	9.65	6.156 × 10 ⁻⁵	1.57 × 10 ⁻⁵
	N ₂	0.752	37.86 × 10 ⁻⁵	1.99 × 10 ⁻⁷
	selectivity	12.83	0.163	78.89

*STP stands for standard temperature and pressure (0 °C, 1.0 atm).