

Supporting Information

Destabilization of ionic compounds under compression: a case of copper halides

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Figure

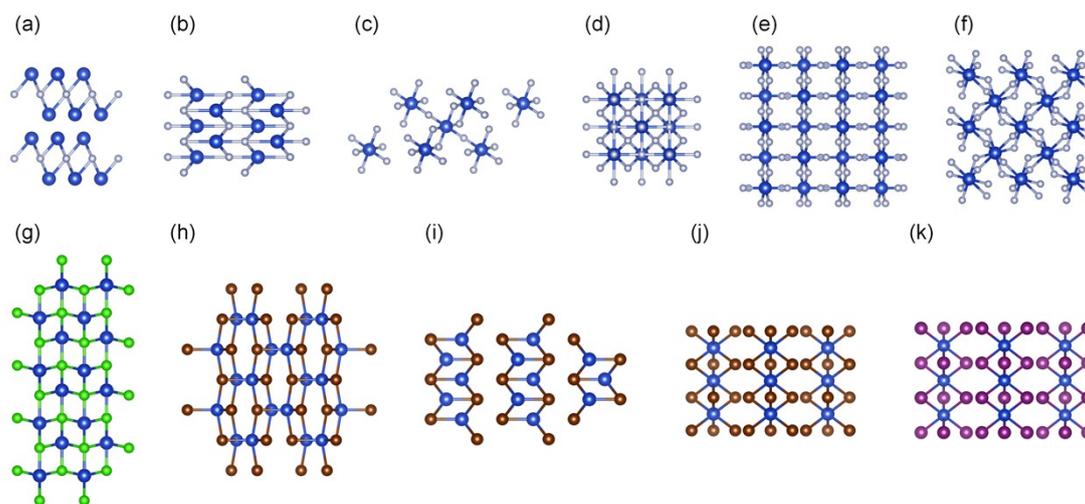


Figure S1. The crystal structures of the stable Cu-X compounds at high pressure. (a) $P\bar{3}m1$ phase for Cu_2F , (b) $Cmmm$ phase for CuF , (c) $R\bar{3}c$ phase for CuF_3 , (d) $Fm\bar{3}m$ phase for CuF_3 , (e) $Cccm$ phase for CuF_4 , (f) $P2_1/c$ phase for CuF_4 , (g) $Cmcm$ phase for CuCl , (h) $Cmcm$ phase for CuBr , (i) $P4/nmm$ phase for CuBr , (j) $P6_3/mcm$ phase for CuBr_3 , (k) $P6_3/mcm$ phase for CuI_3 .

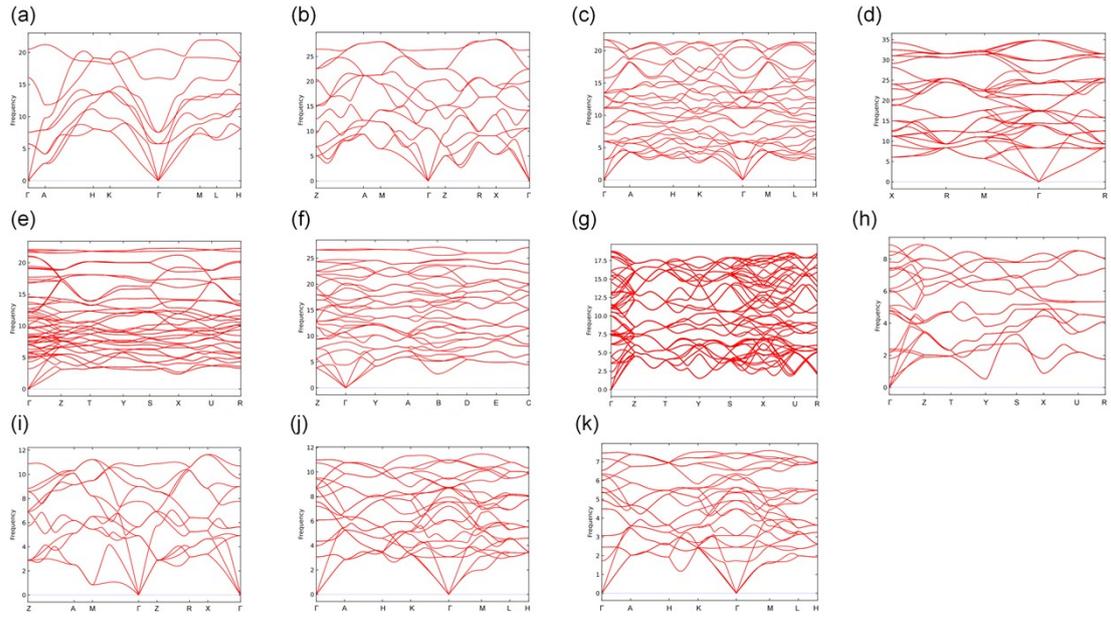


Figure S2. The phonon spectrum at the corresponding pressure. (a) $P\bar{3}m1$ phase for Cu_2F at 200 GPa, (b) $Cmmm$ phase for CuF at 300 GPa, (c) $R\bar{3}c$ phase for CuF_3 at 50 GPa, (d) $Fm\bar{3}m$ phase for CuF_3 at 400 GPa, (e) $Cccm$ phase for CuF_4 at 50 GPa, (f) $P2_1/c$ phase for CuF_4 at 150 GPa, (g) Cmc phase for CuCl at 200 GPa, (h) Cmc phase for CuBr at 40 GPa, (i) $P4/nmm$ phase for CuBr 100 GPa, (j) $P6_3/mcm$ phase for CuBr_3 at 80 GPa, (k) $P6_3/mcm$ phase for CuI_3 at 40 GPa.

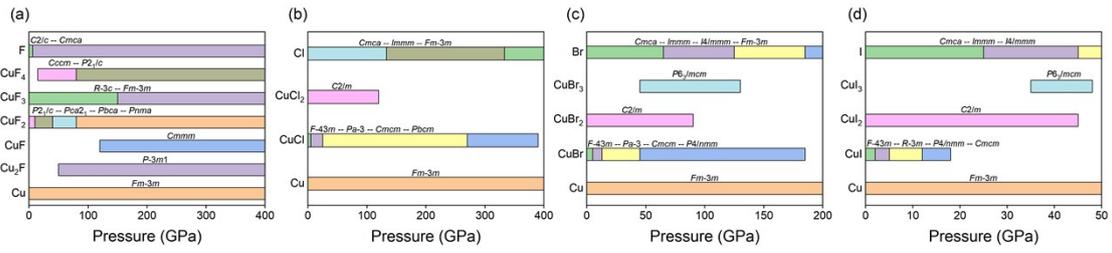


Figure S3. The pressure composition phase diagram of the (a) Cu-F, (b) Cu-Cl, (c) Cu-Br, and (d) Cu-I systems.

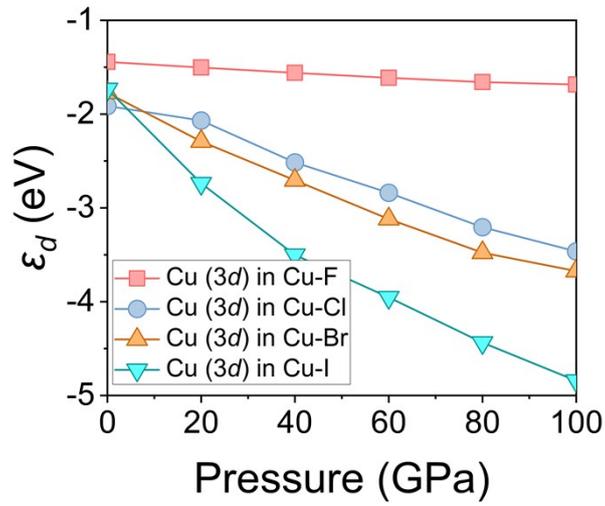


Figure S4. Pressure-driven shift of Cu-3d band centers (ϵ_d) in Cu-X systems. Evolution of ϵ_d with increasing pressure for Cu-F, Cu-Cl, Cu-Br, and Cu-I systems, illustrating the progressive downshift in *d*-orbital band centers under compression.

Table

Table S1. The parameters of the predicted structures for Cu-X compounds.

Space group	Pressure (GPa)	Lattice parameter (Å)		Atom	site	Wyckoff position		
Cu ₂ F (<i>P</i> -3 <i>m</i> 1)	200	<i>a</i> = 2.214	α = 90.000	Cu1	2d	0.333	0.667	0.814
		<i>b</i> = 2.214	β = 90.000	F1	1b	0.000	0.000	0.500
		<i>c</i> = 4.765	γ = 120.000					
CuF (<i>Cmmm</i>)	300	<i>a</i> = 2.269	α = 90.000	Cu1	2c	0.000	0.500	0.342
		<i>b</i> = 2.269	β = 90.000	F1	2c	0.000	0.500	0.870
		<i>c</i> = 4.436	γ = 90.000					
CuF ₃ (<i>R</i> -3 <i>c</i>)	50	<i>a</i> = 4.301	α = 90.000	Cu1	6b	0.000	0.000	0.000
		<i>b</i> = 4.301	β = 90.000	F1	18e	-0.333	-0.359	0.583
		<i>c</i> = 12.325	γ = 120.000					
CuF ₃ (<i>Fm</i> -3 <i>m</i>)	400	<i>a</i> = 4.229	α = 90.000	Cu1	4a	0.000	0.000	0.000
		<i>b</i> = 4.229	β = 90.000	F1	4b	0.000	0.000	0.500
		<i>c</i> = 4.229	γ = 90.000	F2	8c	0.750	0.250	0.250
CuF ₄ (<i>Cccm</i>)	50	<i>a</i> = 5.448	α = 90.000	Cu1	4e	0.750	0.750	1.000
		<i>b</i> = 5.991	β = 90.000	F1	8l	0.805	0.965	0.500
		<i>c</i> = 4.776	γ = 90.000	F2	8h	0.500	0.197	0.750
CuF ₄ (<i>P</i> 2 ₁ / <i>c</i>)	150	<i>a</i> = 3.691	α = 90.000	Cu1	2c	0.000	0.500	1.000
		<i>b</i> = 4.056	β = 96.785	F1	4e	0.793	0.632	0.594
		<i>c</i> = 4.106	γ = 90.000	F2	4e	0.646	0.339	1.178
CuCl (<i>Cmcm</i>)	150	<i>a</i> = 2.699	α = 90.000	Cu1	4c	0.000	0.905	1.250
		<i>b</i> = 8.085	β = 90.000	Cl1	4c	-0.500	0.141	1.250
		<i>c</i> = 3.360	γ = 90.000					
CuBr (<i>Cmcm</i>)	40	<i>a</i> = 4.789	α = 90.000	Cu1	4c	-0.500	0.111	0.750
		<i>b</i> = 5.493	β = 90.000	Br1	4c	-0.500	0.685	0.750
		<i>c</i> = 4.278	γ = 90.000					
CuBr ₃ (<i>P</i> 6 ₃ / <i>mcm</i>)	80	<i>a</i> = 4.985	α = 90.000	Cu1	2b	0.000	0.000	0.500
		<i>b</i> = 4.985	β = 90.000	Br1	6g	0.000	0.382	0.750
		<i>c</i> = 5.035	γ = 120.000					

CuBr (<i>P4/nmm</i>)	100	$a = 2.949$	$\alpha = 90.000$	Cu1 Br1	2c 2c	0.000 0.000	0.500 0.500	0.108 0.676
		$b = 2.949$	$\beta = 90.000$					
		$c = 5.248$	$\gamma = 90.000$					
CuI ₃ (<i>P6₃/mcm</i>)	40	$a = 5.641$	$\alpha = 90.000$	Cu1 I1	2b 6g	0.000 0.372	0.000 0.000	1.000 0.750
		$b = 5.641$	$\beta = 90.000$					
		$c = 5.626$	$\gamma = 120.000$					