

Supporting Information

Crucial Role of Oxygen Vacancies for the Efficient Hydrodeoxygenation of Lignin-based Phenolic Model Compounds with Ni/Ti_{1-x}Zr_xO₂

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Table S1. The data obtained from the O₂-TPD profile (Figure 5) of Ni-supported mono and mixed oxide catalysts

Catalysts	O₂ (mmol/g)		
	Peak I (<250 °C)	Peak II (250–400 °C)	Peak III (>400 °C)
Ni/ZrO ₂	0.037	0.059	0.197
Ni/Ti _{0.05} Zr _{0.95}	0.054	0.092	0.139
Ni/Ti _{0.25} Zr _{0.75}	0.092	0.225	0.149
Ni/Ti _{0.50} Zr _{0.50}	0.088	0.340	0.186
Ni/Ti _{0.25} Zr _{0.75}	0.112	0.326	0.218
Ni/Ti _{0.95} Zr _{0.05}	0.109	0.221	0.305
Ni/TiO ₂	0.037	0.097	0.158

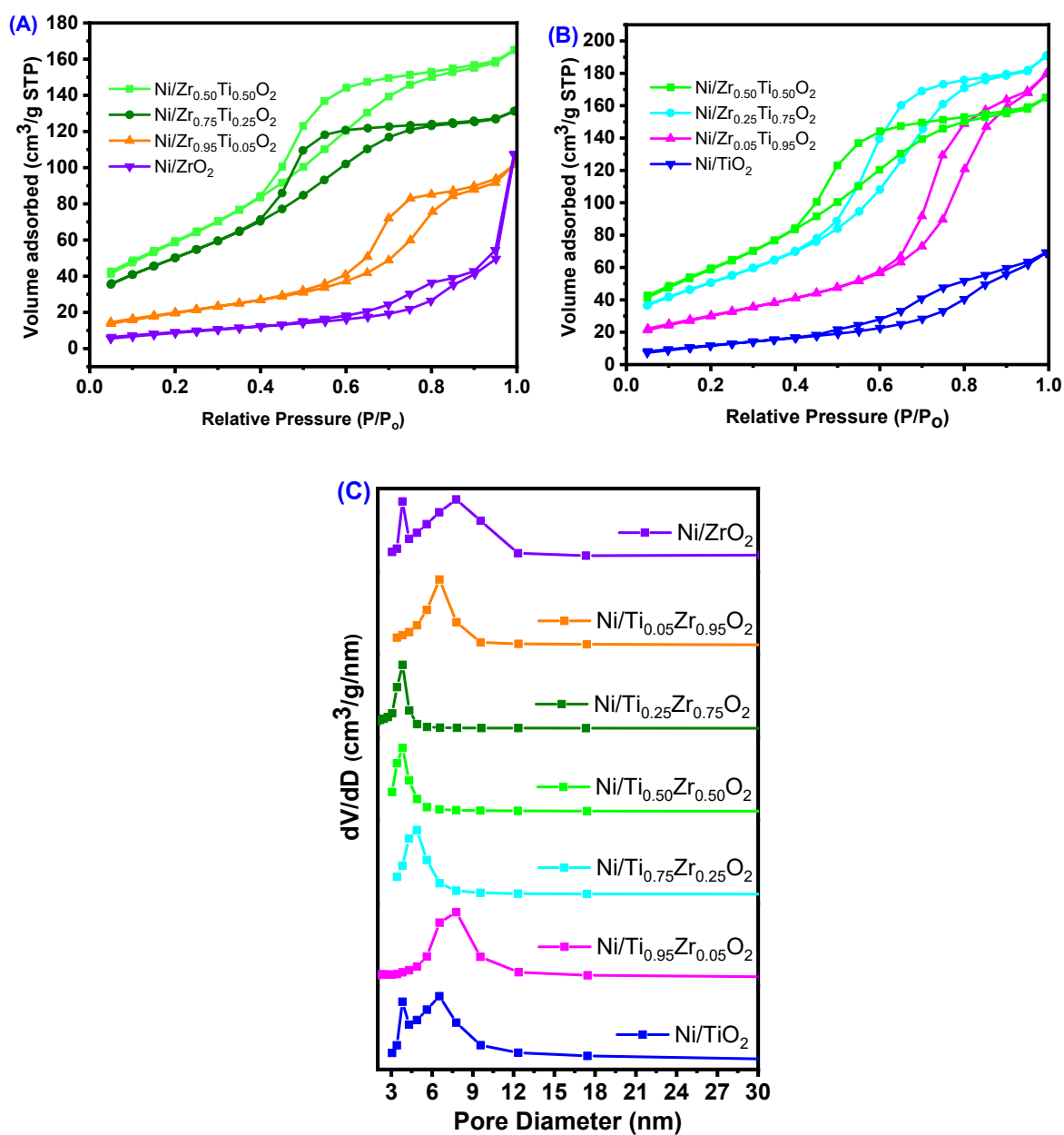


Figure S1: (A and B) Physisorption isotherm and (C) BJH pore size distribution of various Ni-supported mono and mixed oxide catalysts.

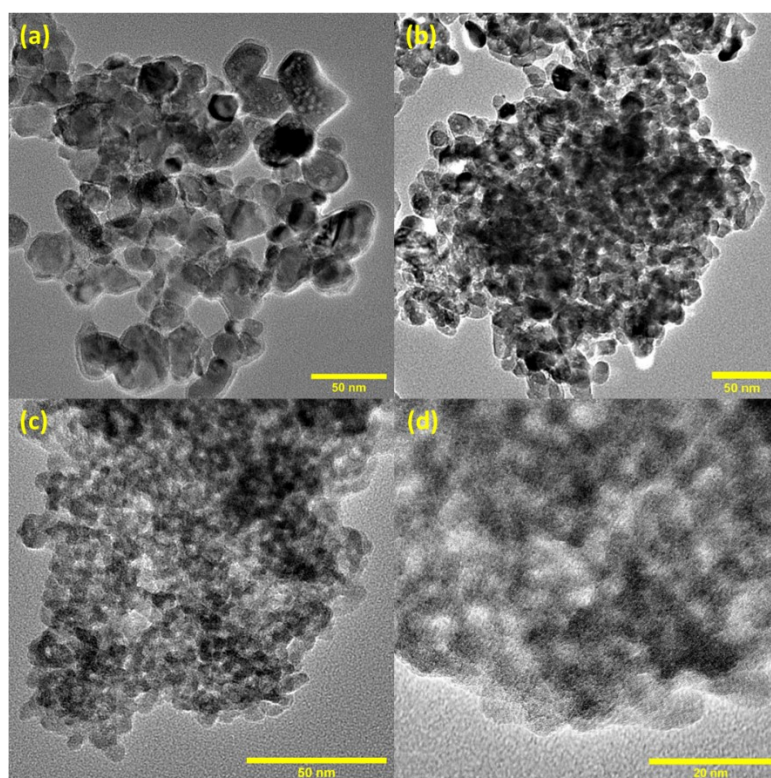
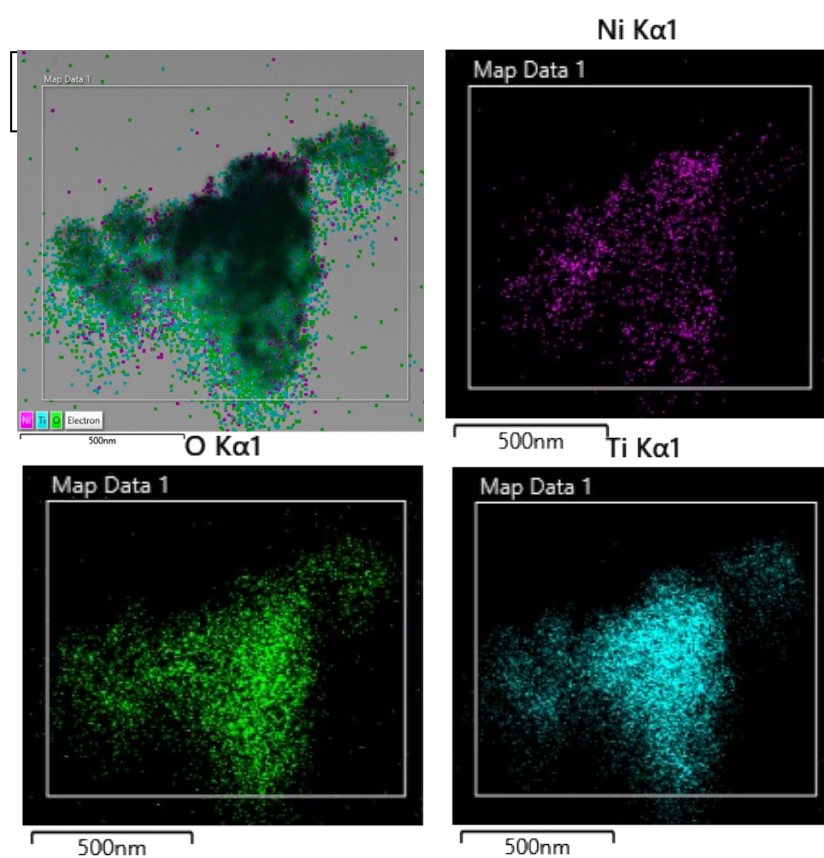


Figure S2. HRTEM images of (a) Ni/TiO₂, (b) Ni/ZrO₂ and (c-d) Ni/Ti_{0.50}Zr_{0.50}O₂ catalysts.



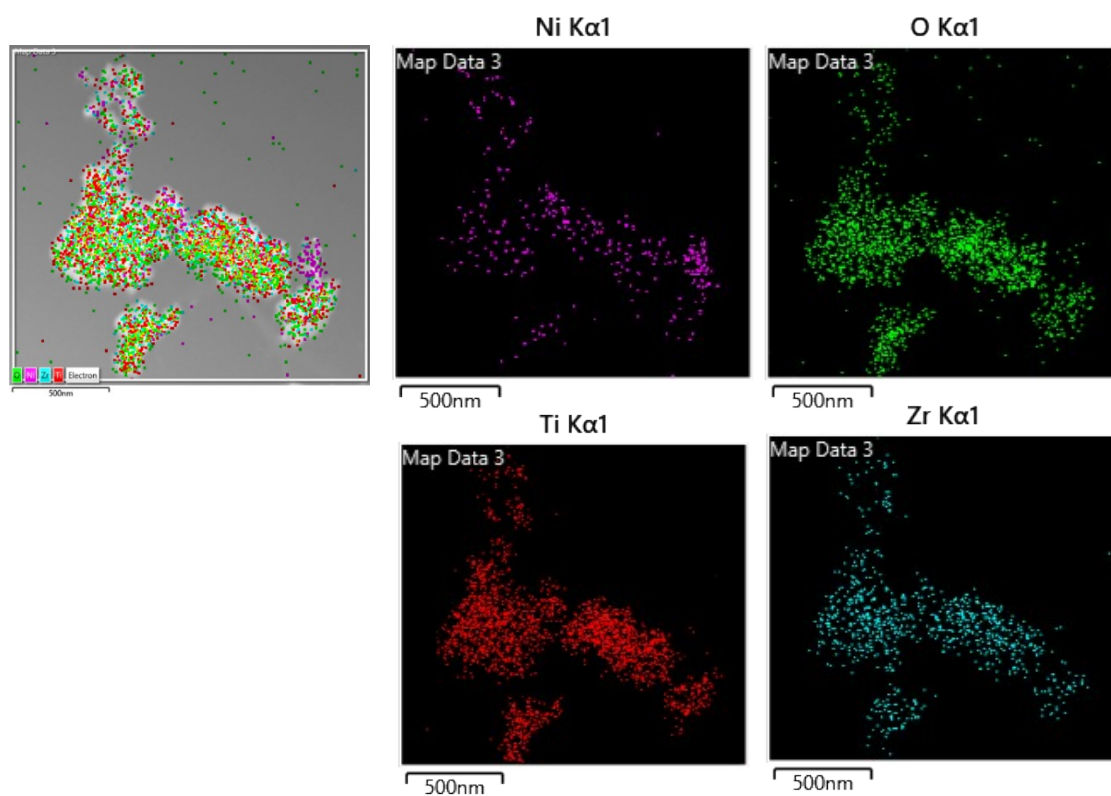
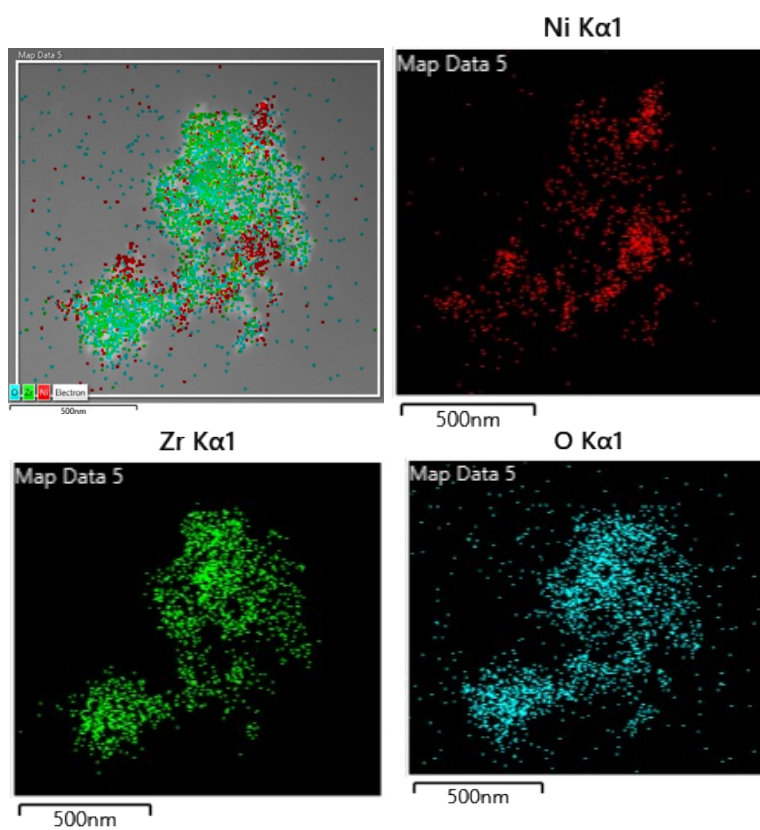


Figure S3. EDS elemental mapping images of a) Ni/TiO₂, b) Ni/ZrO₂, and c) Ni/Ti_{0.50}Zr_{0.50}O₂.

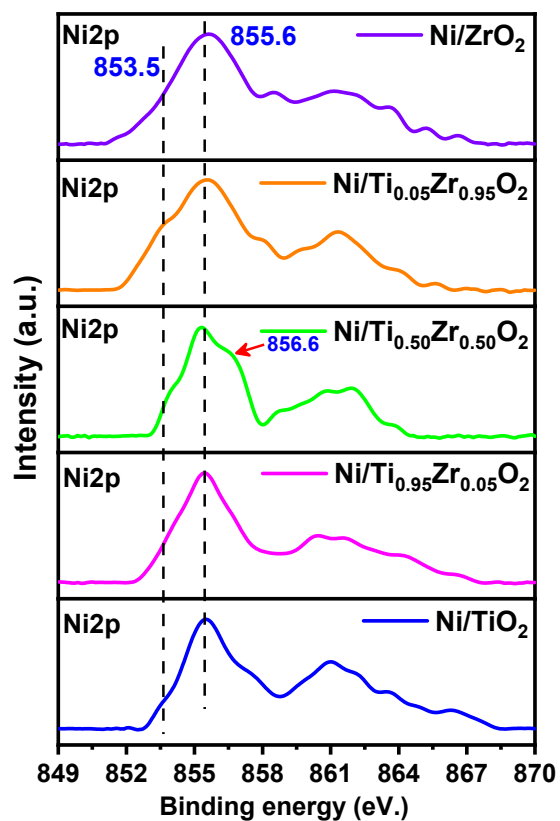


Figure S4. XPS spectra of Ni2p of various Ni-supported mono and mixed metal oxide catalysts.

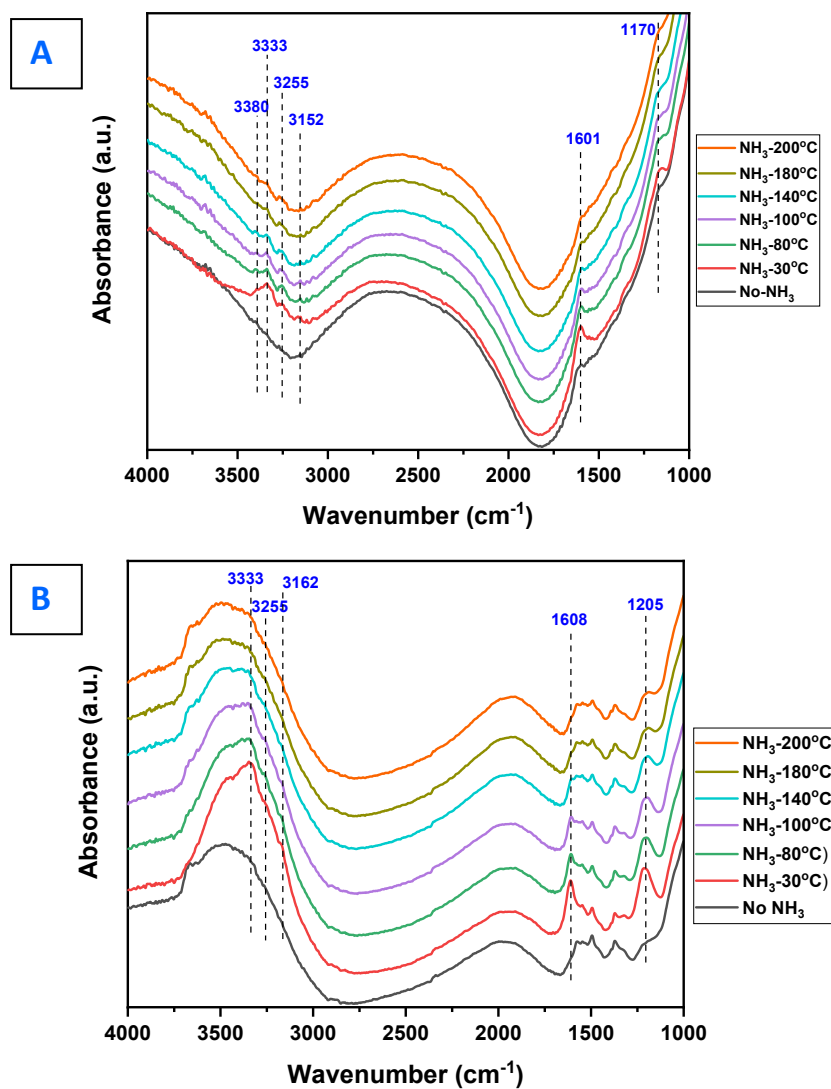


Figure S5. In-situ DRIFT spectra of NH₃ desorbed at various temperature (A) Ni/TiO₂ and (B) Ni/ZrO₂ catalyst.

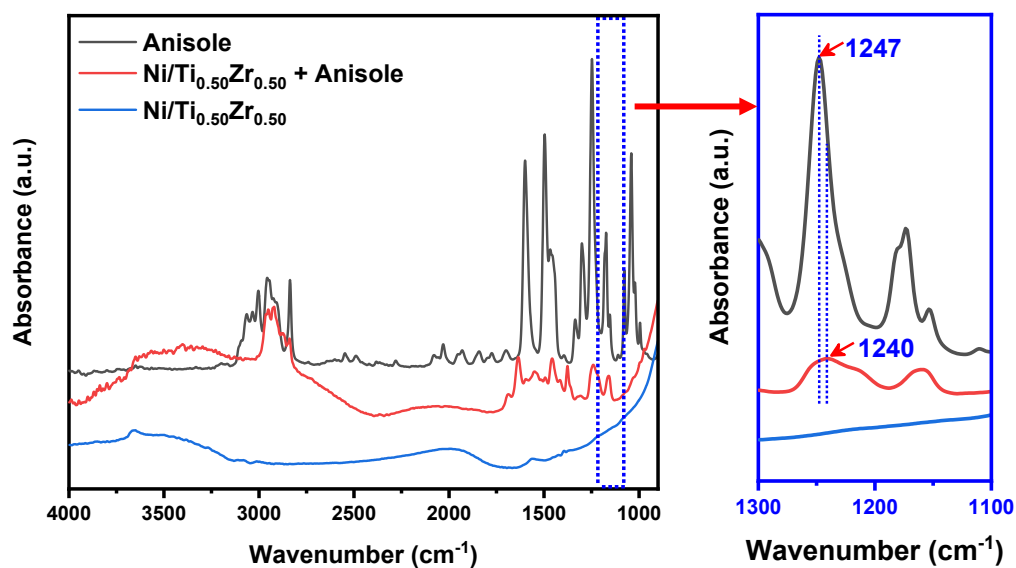


Figure S6. Catalytic activity of Ni/Ti_{0.50}Zr_{0.50}O₂ and previous studies related to Ni-based catalysts for HDO of anisole

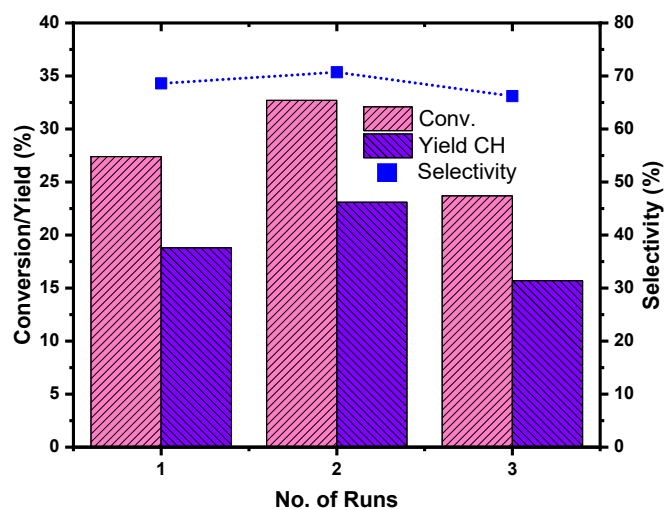


Figure S7. The recyclability of the Ni/Ti_{0.50}Zr_{0.50}O₂ catalyst for anisole HDO. (Reaction conditions: $m_{\text{cat:anisole}} = 1:2$, 20 g decane, 230 °C, 15 min, 30 bar H₂, 450rpm.)

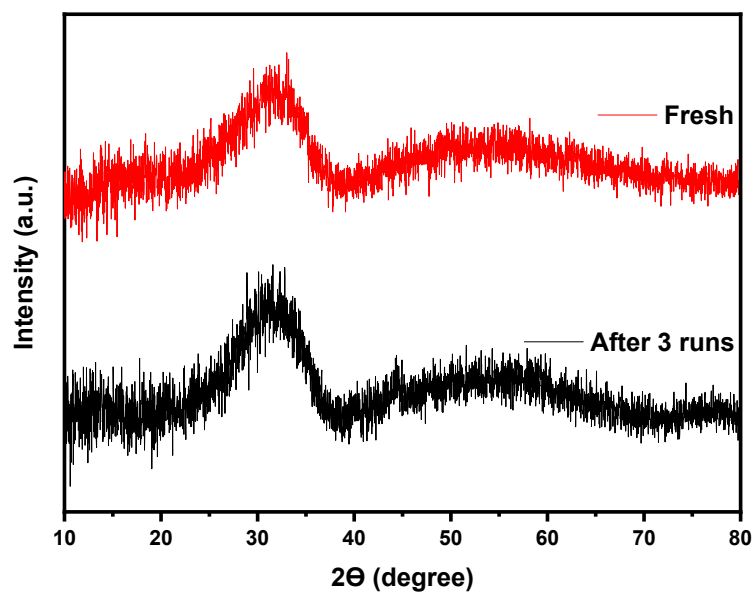


Figure S8. XRD pattern of fresh and 3rd use of Ni/ Ti_{0.50}Zr_{0.50}O₂ catalysts

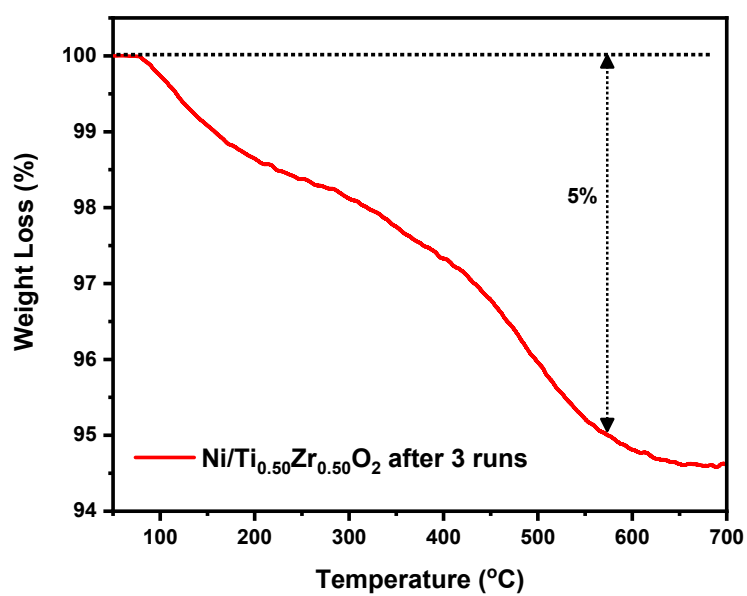


Figure S9. TGA analysis of the spent catalyst

Table S2. Catalytic activity of **3wt%Ni/Ti_{0.50}Zr_{0.50}O₂** and previous studies related to Ni-based catalysts for HDO of anisole

Entry	Catalysts	Reaction conditions	Conv. (%)	Selec. DeO (%)	Reference
1	3wt%Ni/Ti_{0.50}Zr_{0.50}O₂	230 °C, 30 bar H₂, 3 h	>99	94%	This work
2	30wt% Ni/Al ₂ O ₃ -ZrO ₂	230 °C, 10 bar H ₂ , 3 h	100	77.6	[1]
3	5wt% Ni/TiO ₂ -ZrO ₂	300 °C, 40bar H ₂ , 4h	60	~87	[2]
4	10wt% Ni/SiO ₂	300 °C, 50 bar H ₂ , 16 h	>99	80.4	[3]
5	Ni ₂ P/Ti _{0.98} Ce _{0.02} O ₂	300 °C, 7 bar H ₂ , 0.2 h	29.2	79.2	[4]
6	3wt% Ni/Nb ₂ O ₅	240 °C, 20bar H ₂ , 1 h	90	91.0	[5]
7	90wt% Ni/Nb ₂ O ₅	240 °C, 30bar H ₂ , 4h	>99	100	[6]
8	90wt%Ni /SiO ₂	280 °C, 60bar H ₂ , 0.75h	100	40	[7]
9	10wt%Ni/SBA-15	280°C, 35bar H ₂ , 6h	100	100	[8]
10	5wt%Ni/HSZ	200°C, 68bar H ₂ , 2.3 h	98	84	[9]
11	4.6% hie-Ni/ZSM-5	200 °C, 60 bar H ₂ , 2h decalin	100	88.1	[10]
12	4.96% Ni/Al-MCM-41(90)	280 °C, 48 bar H ₂ , 6h	96.5	97.6	[11]
13	60wt% Ni 5wt%Cu 30wt%Si	320 °C, 60bar H ₂ , 2.5h	~85	100	[12]
14	Ni ₁ Mo ₃ N/C	260 °C, 4 h, 10 bar H ₂ ,	99.9	97.8	[13]
15	Ni/C-H ₃ [P-(Mo ₃ O ₁₀) ₄]*13H ₂ O	240°C, 10 bar H ₂ , 3 h	100	100	[14]
16	Ni/C	230 °C, 10bar H ₂ , 2 h, water	15	57.4 (benzene)	[15]
17	Ni@C-500	230°C, 10 bar N ₂ , 2 h	15.4	57.4 (benzene)	[16]
18	NiPt/S	280 °C, 73 bar H ₂ , 6 h	98.5	72	[17]

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