

Supporting Information

Construction of an Advanced MoS₂/MoO₃/NiFe-LDH/NF Heterostructure Catalyst toward Boosting Efficient Alkaline Oxygen Evolution Reaction

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1. Supplementary Figures

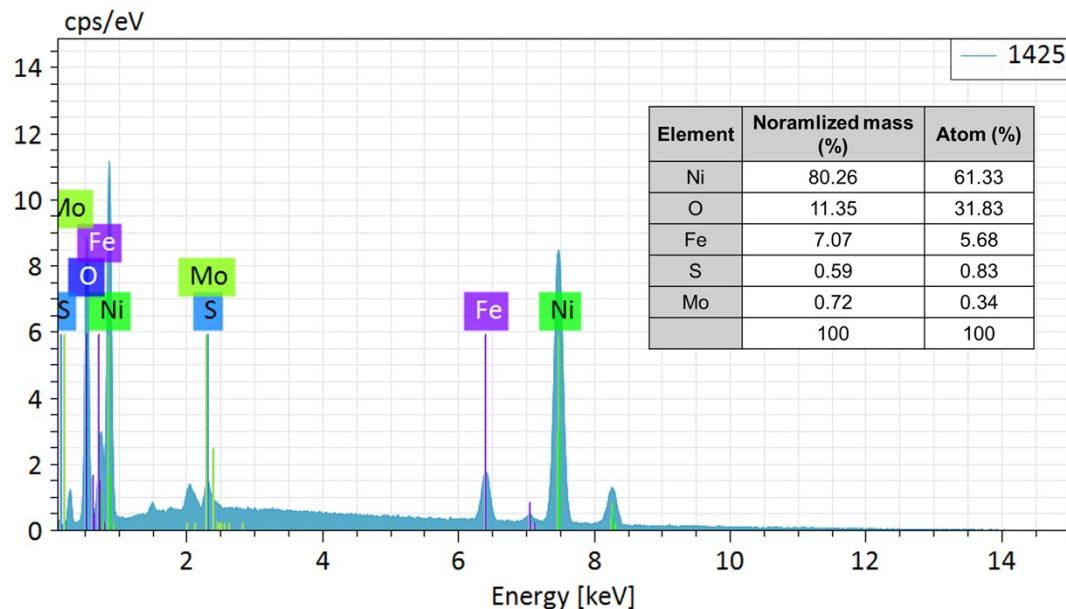


Figure S1 EDS test results of MoS₂/MoO₃/NiFe-LDH/NF.

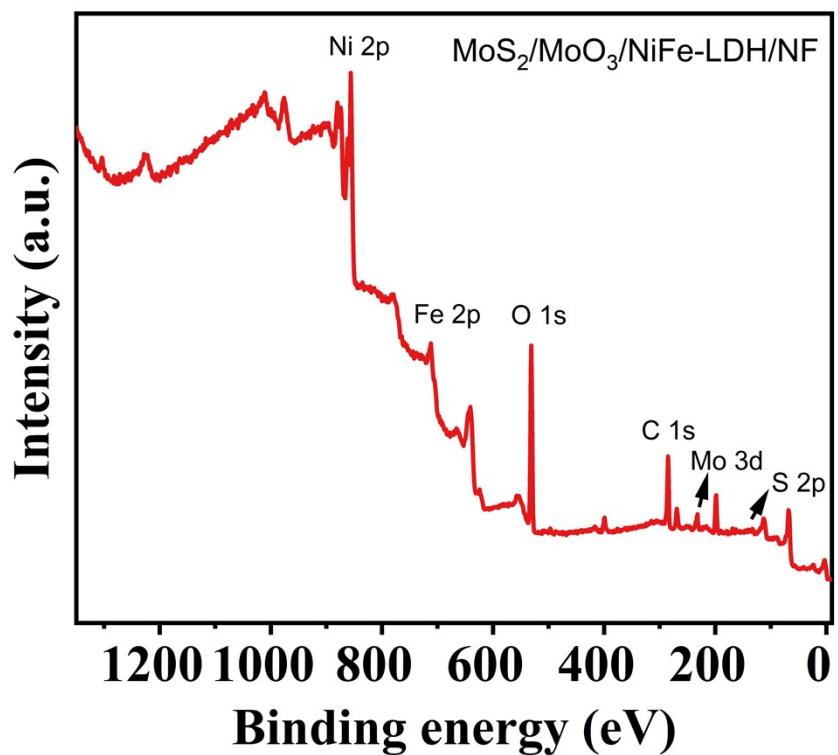


Figure S2 High-resolution XPS full spectrum of MoS₂/MoO₃/NiFe-LDH/NF.

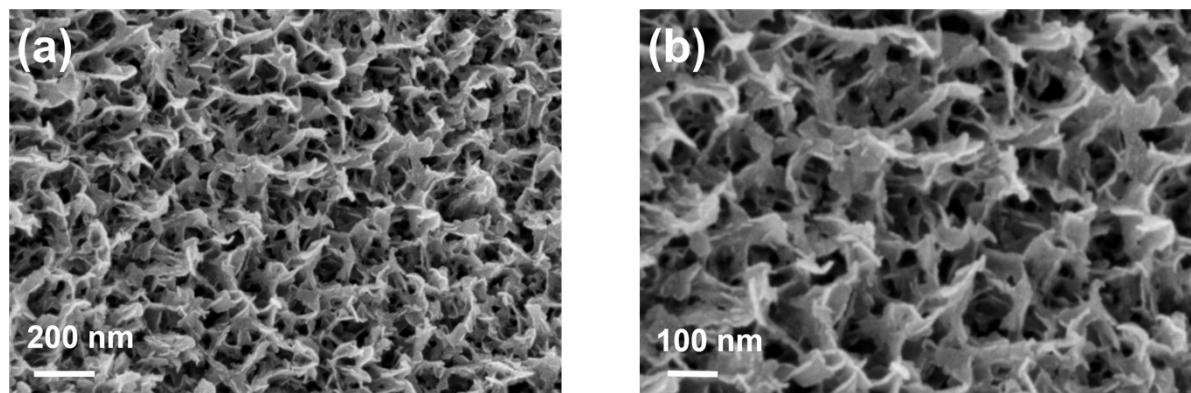


Figure S3 SEM images of MoS₂/MoO₃/NiFe-LDH/NF after durability test.

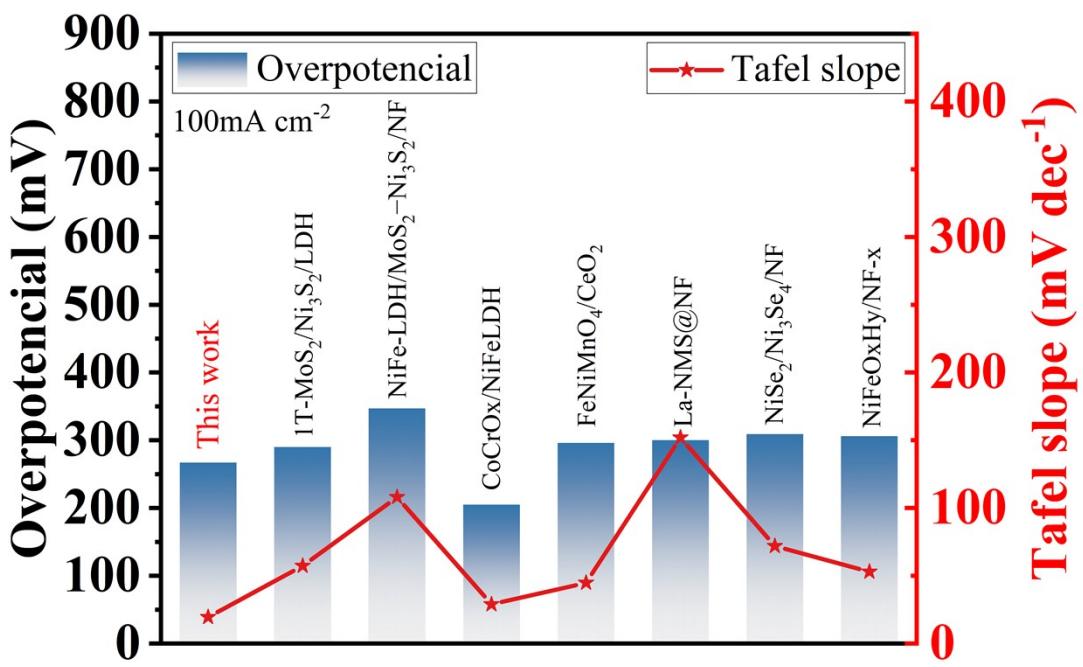


Figure S4 The comparison of OER performance for some representative non-noble. electrocatalysts.

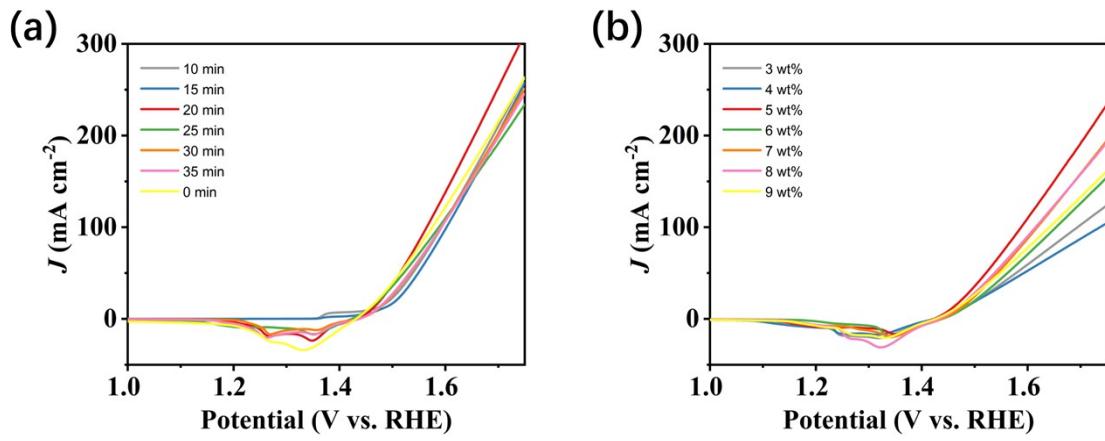


Figure S5 Control experiments using different time and different volumes of H_2O_2 .

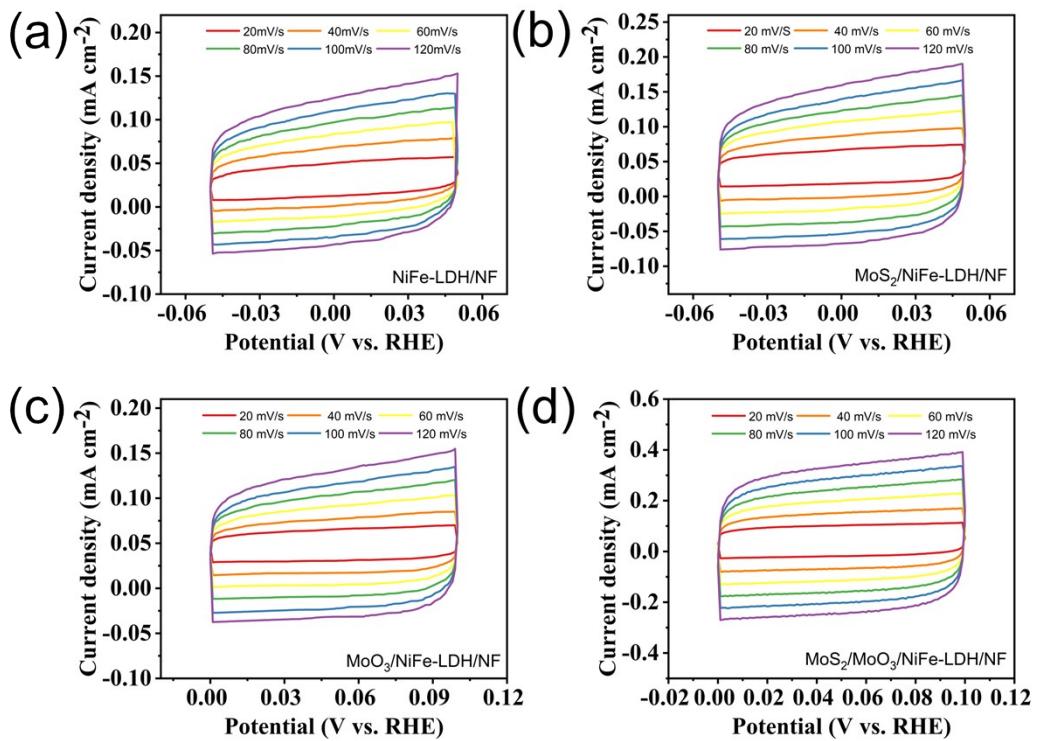


Figure S6 CV curves of (a) NiFe-LDH/NF, (b) MoS₂/ NiFe-LDH/NF, (c) MoO₃/NiFe-LDH/NF and (d) MoS₂/MoO₃/NiFe-LDH/NF for the OER reaction at different scan rates.

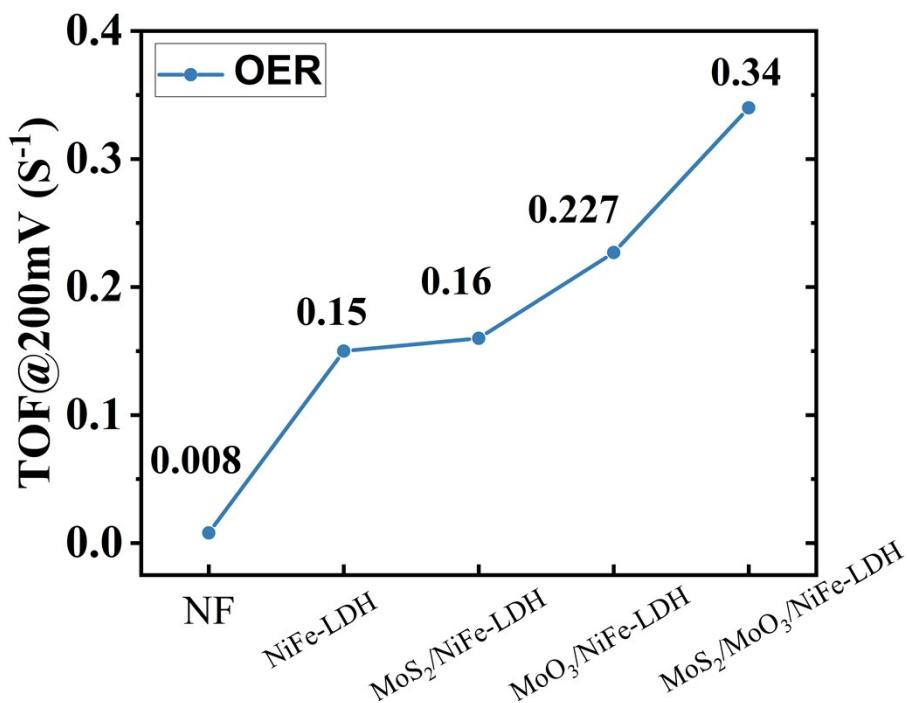


Figure S7 The corresponding turnover frequency at an overpotential of 200 mV.

2. Supplementary Tables

Table S1 Comparison of electrocatalytic performance for OER to other reported catalysts in 1.0 M KOH.

Catalysts	J (mA cm ⁻²)	η mV (vs.RHE)	Tafel slope (mV dec ⁻¹)	Electrolyte	reference
MoS₂/MoO₃/NiFe-LDH/NF	50	255	19.4	1.0M KOH	This work
1T-MoS ₂ /Ni ₃ S ₂ /LDH	100	267			¹
NiFe-LDH/MoS ₂ -Ni ₃ S ₂ /NF	100	290	57.3	1.0M KOH	²
CoCrOx/NiFeLDH	100	347	108	1.0M KOH	³
FeNiMnO ₄ /CeO ₂	100	205	44.8	1.0M KOH	⁴
La-NMS@NF	100	300	152	1.0M KOH	⁵
NiSe ₂ /Ni ₃ Se ₄ /NF	100	309	71.9	1.0 M KOH	⁶
NiFeOxHy/NF-x	100	306	53	1.0 M KOH	⁷

3. Notes and references

- W. Liu, J. Dong, B. An, H. Su, Z. Teng, N. Li, Y. Gao and L. Ge, *Journal of Colloid and Interface Science*, 2024, **673**, 228-238.
- S. Wang, X. Ning, Y. Cao, R. Chen, Z. Lu, J. Hu, J. Xie and A. Hao, *Inorganic Chemistry*, 2023, **62**, 6428-6438.
- L. Wan, D. Lin, J. Liu, Z. Xu, Q. Xu, Y. Zhen, M. Pang and B. Wang, *ACS Nano*, 2024, **18**, 22901-22916.
- H. Wu, Z. Wang, Y. Shi, Z. Li, F. Ding, Y. Ren, F. Li, H. Bian, C. Wang, Y. Yang, J. Gu, S. Tang, Y. Ma, Y. Deng and X. Meng, *Inorganic Chemistry Frontiers*, 2024, **11**, 3786-3798.
- W. Li, Z. Sun, R. Ge, J. Li, Y. Li, J. M. Cairney, R. Zheng, Y. Li, S. Li, Q. Li and B. Liu, 2023, **4**, 2300175.
- L. Tan, J. Yu, H. Wang, H. Gao, X. Liu, L. Wang, X. She and T. Zhan, *Applied Catalysis B: Environmental*, 2022, **303**, 120915.
- J. Deng, Z. Wang, H. Yang, R. Jian, Y. Zhang, P. Xia, W. Liu, O. Fontaine, Y. Zhu, L. Li and S. Chen, *Chemical Engineering Journal*, 2024, **482**, 148887.