

Supplementary Information

***Operando* X-ray absorption spectroscopy of phosphomolybdic acid redox mediators for electrochemical conversion of biomass to green hydrogen**

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XAS Experimental conditions

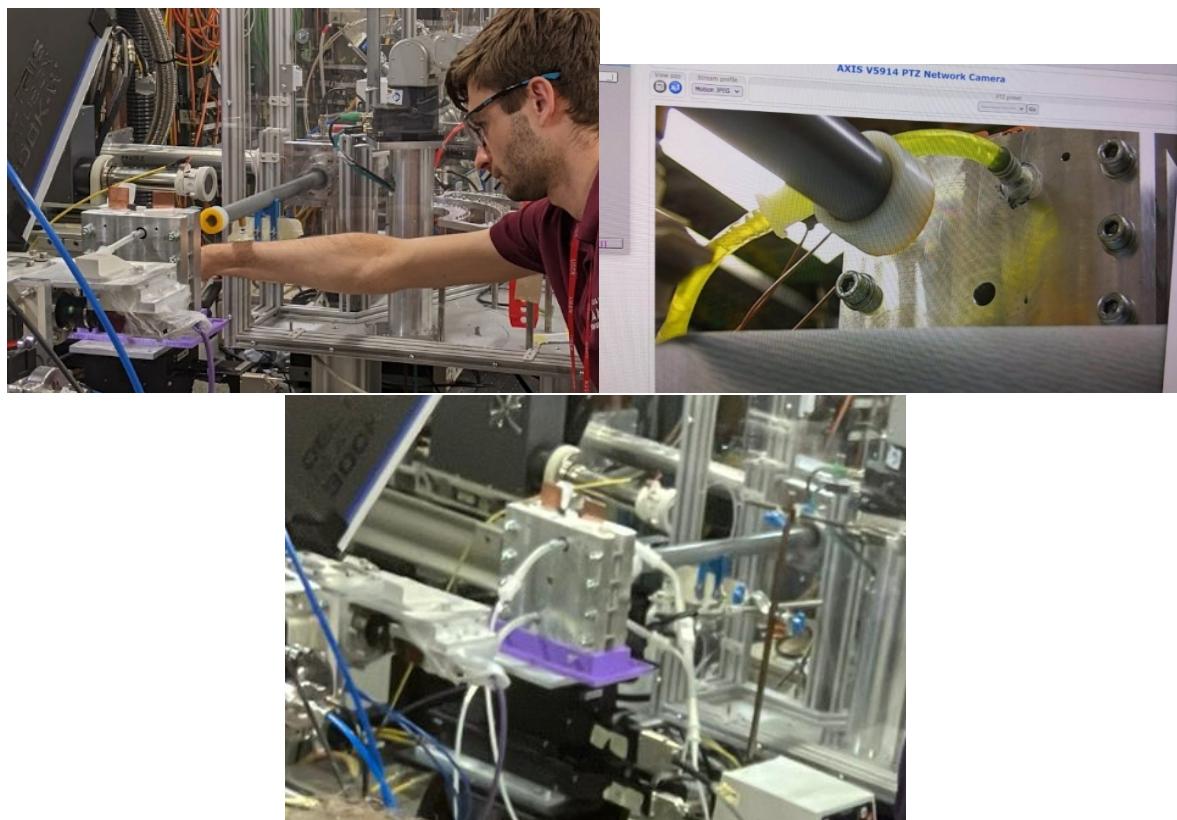


Figure S1. A selection of images of the assembled flow cell, situated in position on the sample stage of the B18 beamline at Diamond Light Source.

Table S1: Complete list of standard materials characterised as part of this study.

Standard Material:	Purity/Supplier:	Mass of Standard/mg	Mass of Cellulose/mg
MoO_3	Sigma-Aldrich	24.30	56.71
MoO_2	Sigma-Aldrich	21.66	58.34
$\text{H}_3[\text{PMo}_{12}\text{O}_{40}]\cdot x\text{H}_2\text{O}$	Thermo Scientific Chemicals	32.20	48.87
$\text{Na}_2\text{MoO}_4\cdot 2\text{H}_2\text{O}$	Sigma-Aldrich	40.29	40.16
Na_2MoO_4	Sigma-Aldrich	39.64	40.59
$(\text{NH}_4)_6\text{Mo}_7\text{O}_{24}\cdot 2\text{H}_2\text{O}$	Sigma-Aldrich	29.77	50.42

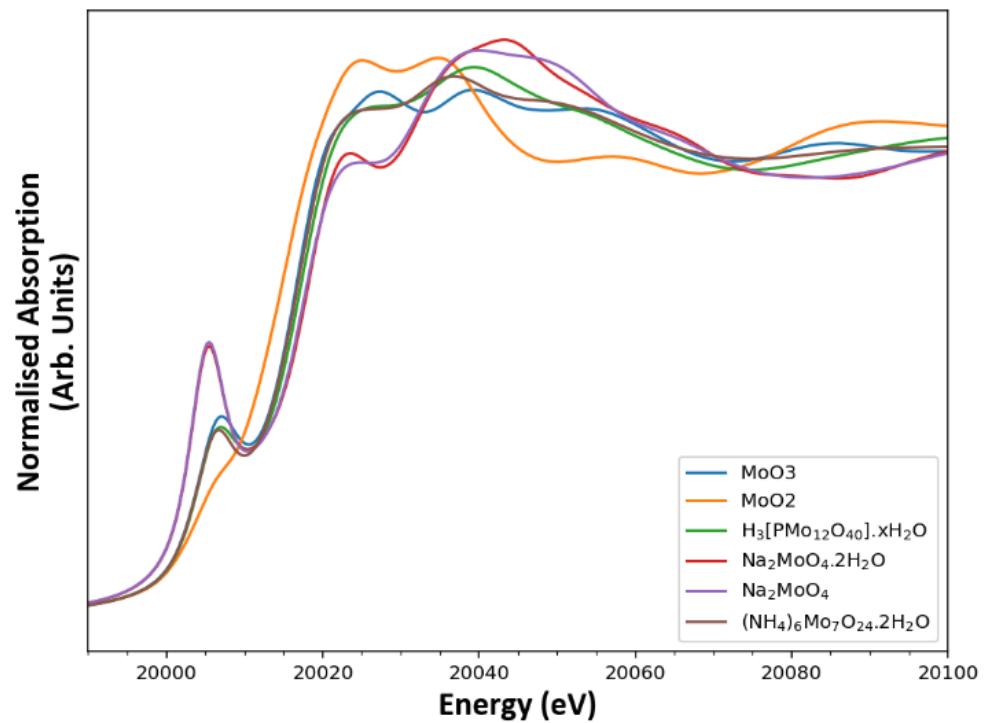


Figure S2: XANES spectra of Calibration Standards listed in Table S1

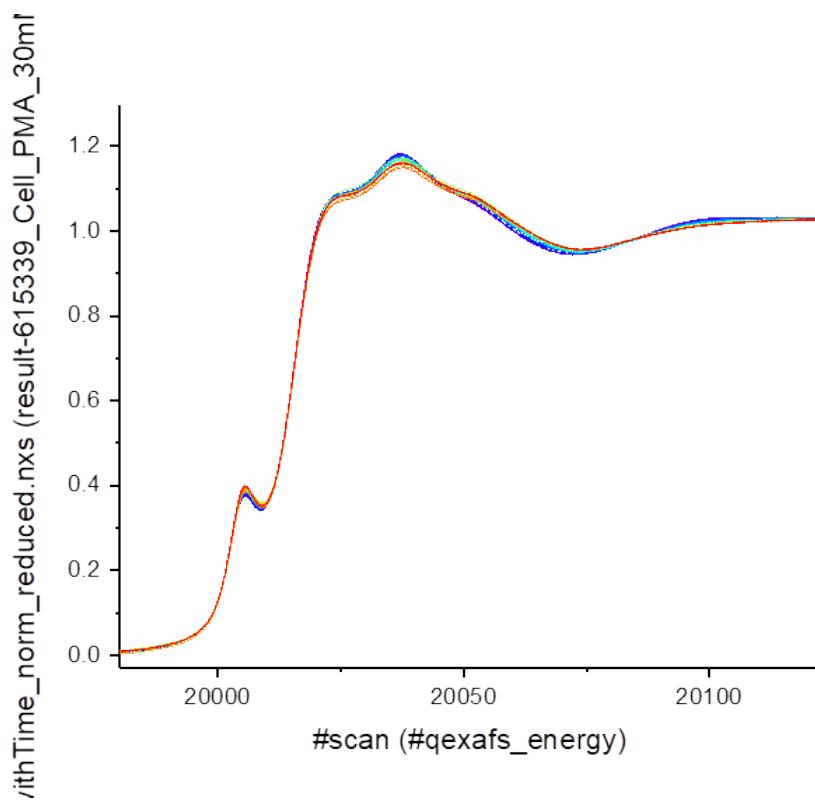


Figure S3: Plot of sequential quick XAS spectra obtained during an *operando* reoxidation of the Electrochemically-reduced sample, plotted over a colour gradient from red to blue. Each spectrum took 31.55 seconds.

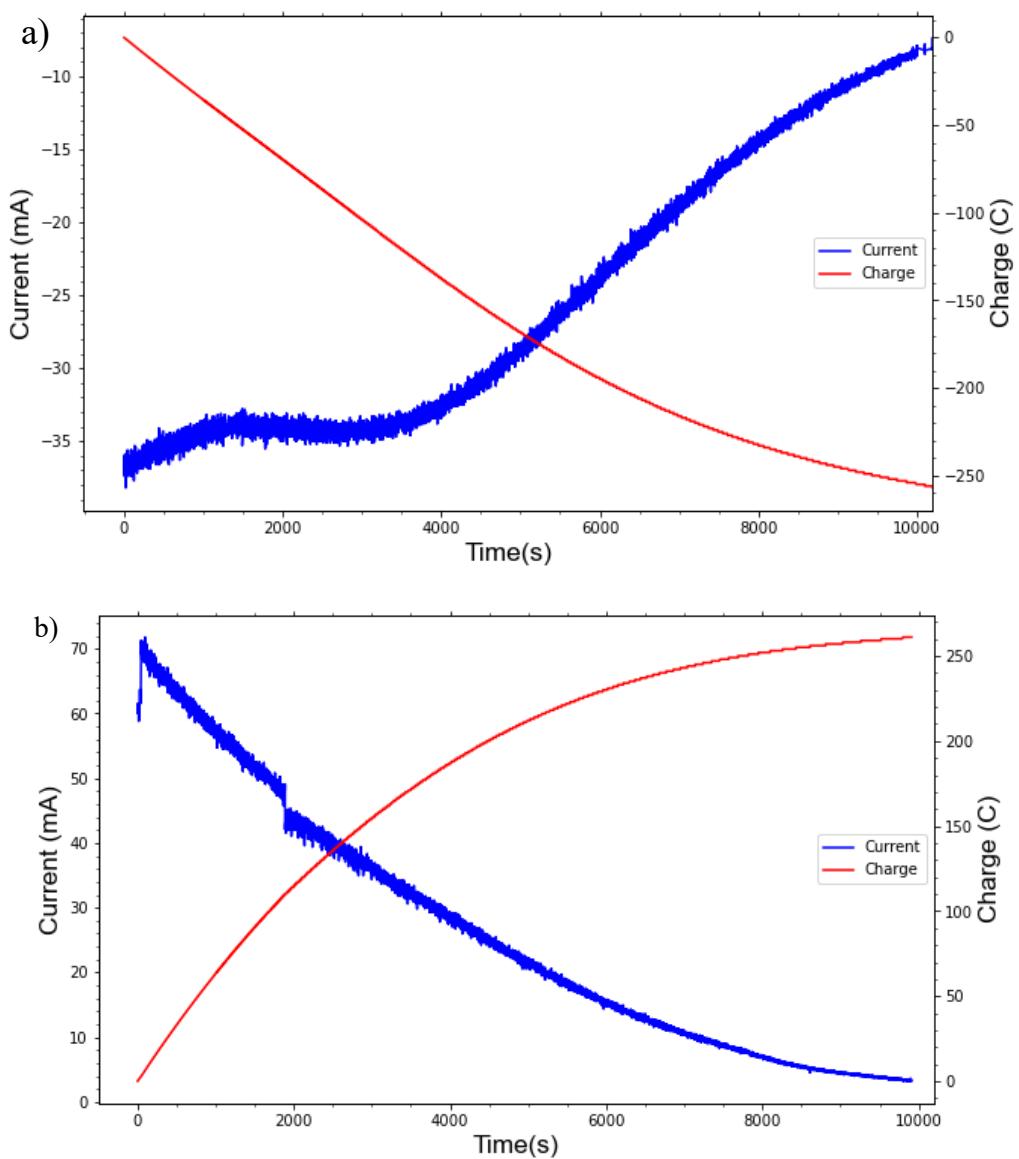


Figure S4. Plots of current and charge passage as a function of time for an H-cell operating with a PMA anolyte in a) reduction mode and b) oxidation mode.

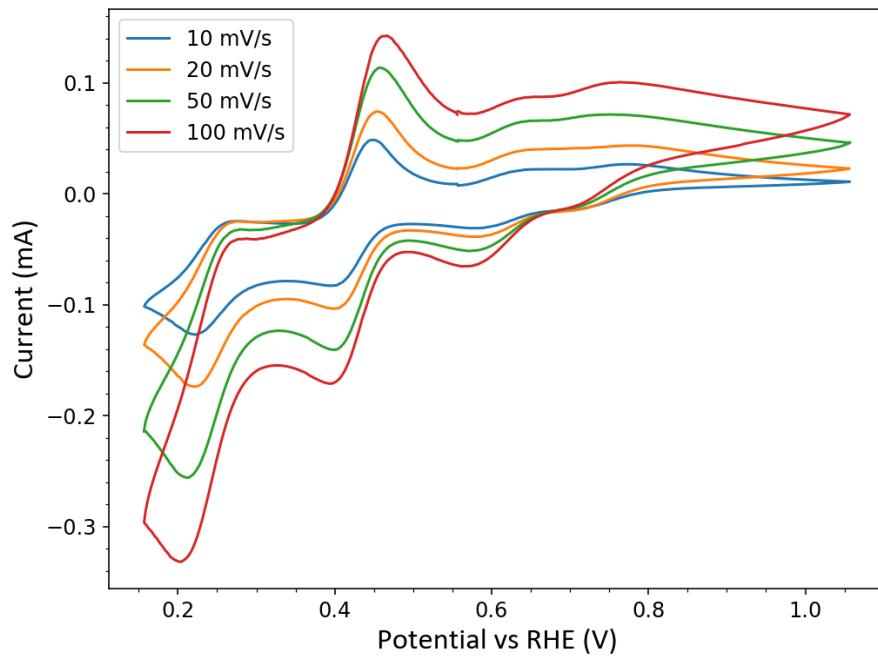


Figure S5. Plots of cyclic voltammetry of 0.01 mol dm⁻³ PMA at scan rates ranging from 10 to 100 mV/s.

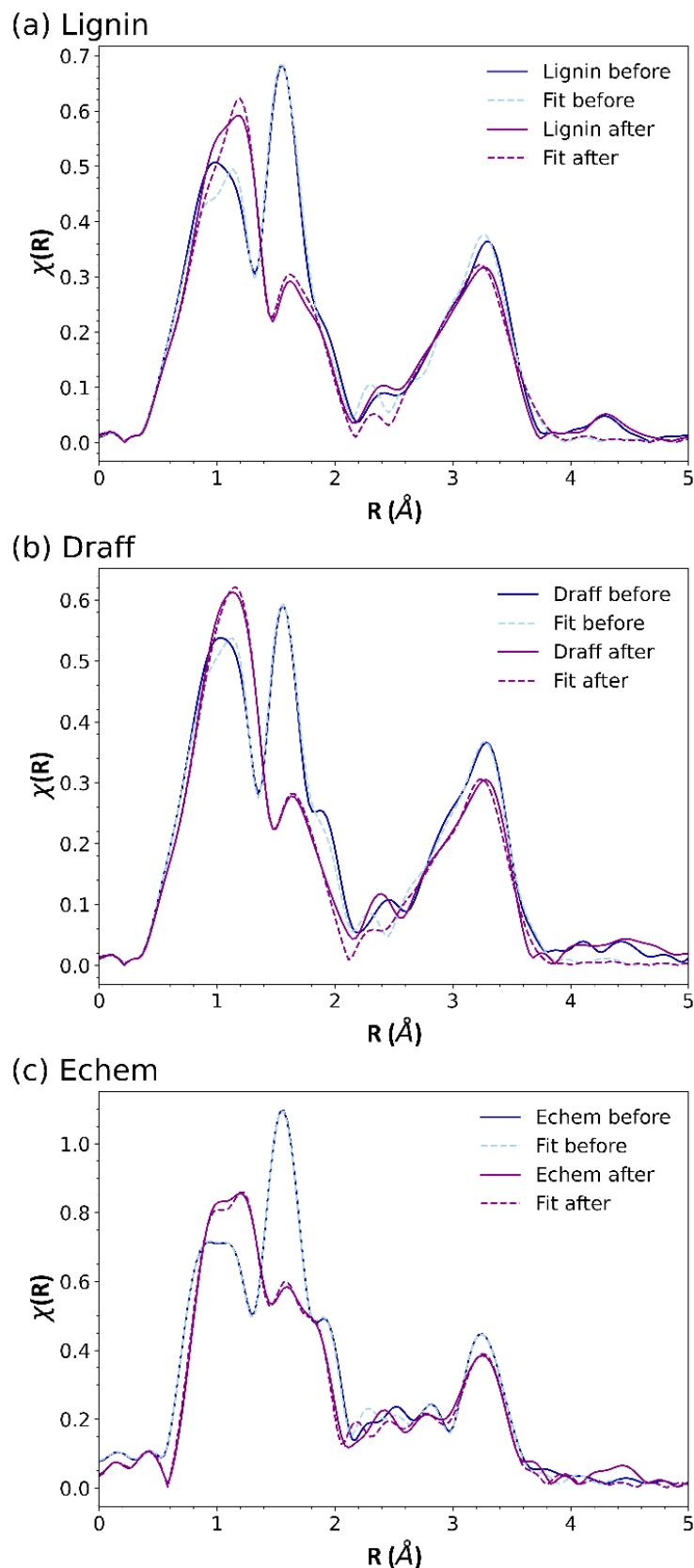


Figure S6. Fitting of EXAFS spectra (in R-space) of Lignin, Draft and Echem samples before and after re-oxidation of reduced PMA (starting photoelectron scattering paths obtained from geometrically relaxed DFT calculations and modelled using Artemis software)