

Supporting Information

Atomically Dispersed Cu-Pd Dual Sites on Nitrogen-Doped Carbon for Efficient Electroreduction of Nitrite to Ammonia

Supporting Tables

Tab. S1 Surface valence composition of Pd and Cu species in Cu-Pd-N-C catalysts with different Pd loadings.

Sample	Pd ²⁺ (%)	Pd ⁴⁺ (%)	Pd ²⁺ /Pd ⁴⁺	Cu ⁺ (%)	Cu ²⁺ (%)	Cu ⁺ /Cu ²⁺
Cu-Pd ₂₀ -N-C	56.38	43.62	1.29	26.00	74.00	0.35
Cu-Pd ₄₀ -N-C	58.91	41.09	1.43	23.72	76.28	0.31

Tab. S2 Reaction Conditions and Performance Metrics of Different Catalysts

Catalyst	Potential (vs. RHE)	Electrolyte	Concentration (NO ₂ ⁻)	NH ₃ Yield (mg h ⁻¹ cm ⁻²)	FE (%)
V-TiO ₂	-0.7 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	9.2	93.2
Ni ₁ Ru	-0.6 V	0.5 M Na ₂ SO ₄ + 0.1 M NO ₂ ⁻	0.1 M	8.3	95.9
Ni@MDC	-0.8 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	6.3	65.4
Ag nanoarray	-0.7 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	5.8	85.0
Ni@JBC-800	-0.5 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	4.1	83.4
TiO ₂ x nanobelt array	-0.7 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	7.9	92.7
Cu ₃ P nanowires	-0.5 V	0.1 M PBS + 0.1 M NO ₂ ⁻	0.1 M	1.6	91.2
Cu-Pd ₂₀ -N-C	-0.6 V	0.1 M NaOH + 0.1 M NO ₂ ⁻	0.1 M	9.7	95.1

Supporting Figures

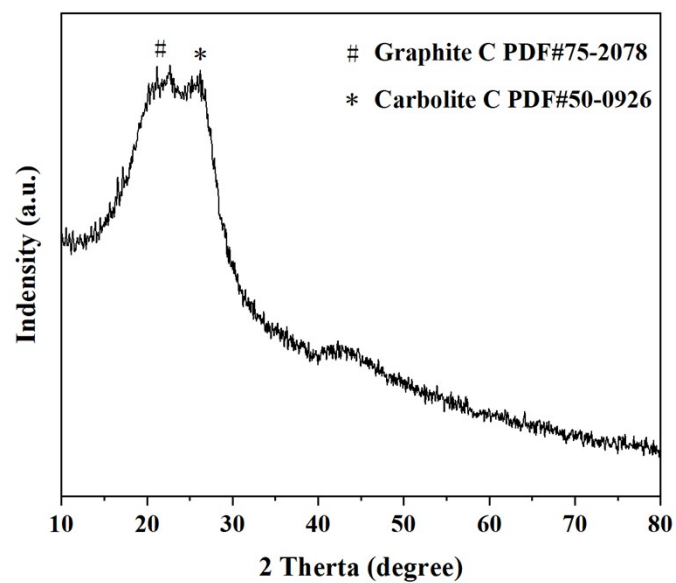


Fig. S1 XRD pattern of Cu-Pd-N-C.

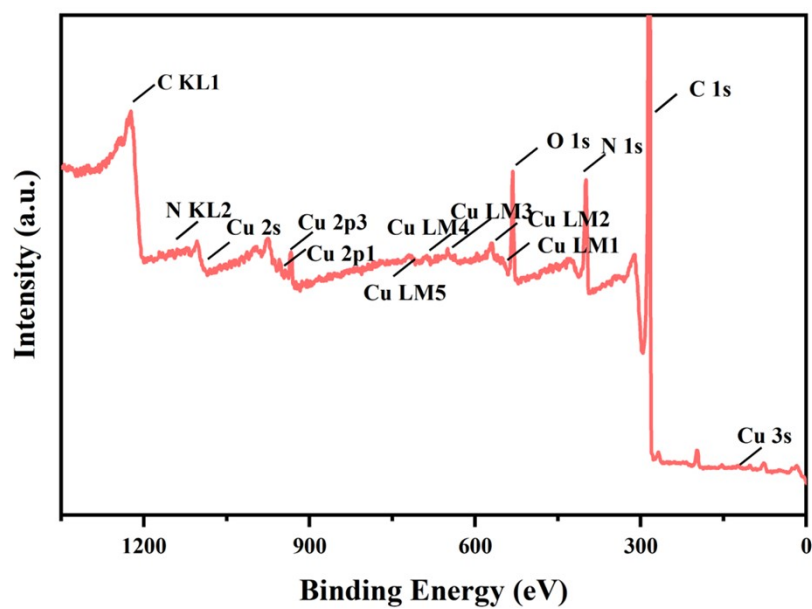


Fig. S2 XPS survey spectrum of Cu-Pd-N-C.

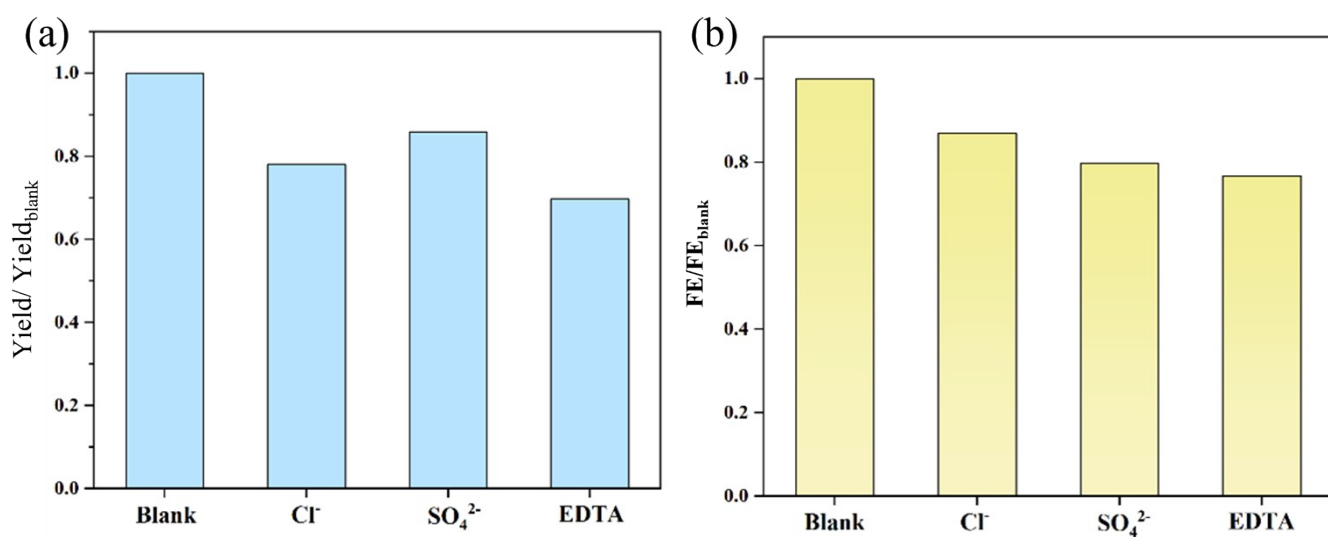


Fig. S3 Normalized (a) NH₃ yield and (b) Faradaic efficiency of Cu-Pd₂₀-N-C in simulated wastewater.

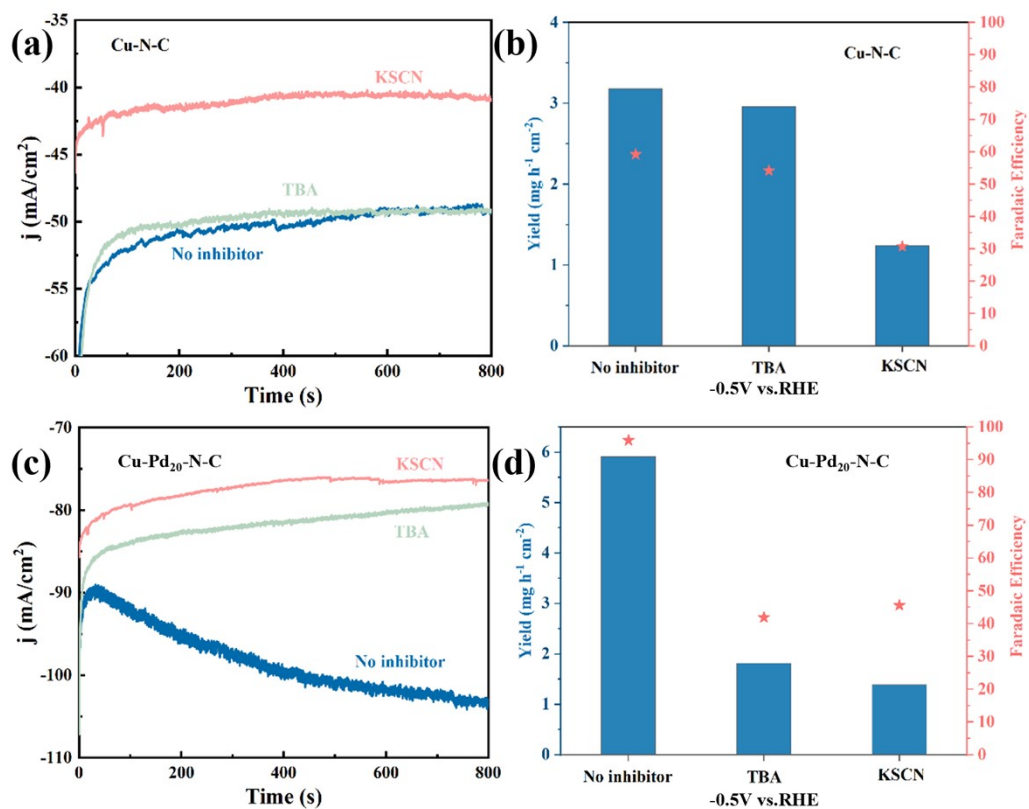


Fig. S4 (a) I-t curves of Cu-N-C with KSCN, TBA; (b) NH₄⁺ yield and FE of Cu-Pd-N-C with KSCN, TBA; (c) I-t curves of Cu-Pd₂₀-N-C with KSCN, TBA; (d) NH₄⁺ yield and FE of Cu-Pd₂₀-N-C with KSCN, TBA.