

Supporting Information

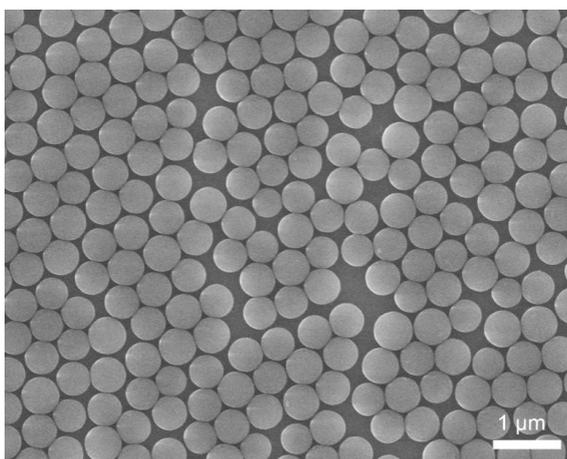


Fig. S1. SEM image of SiO₂ spheres.

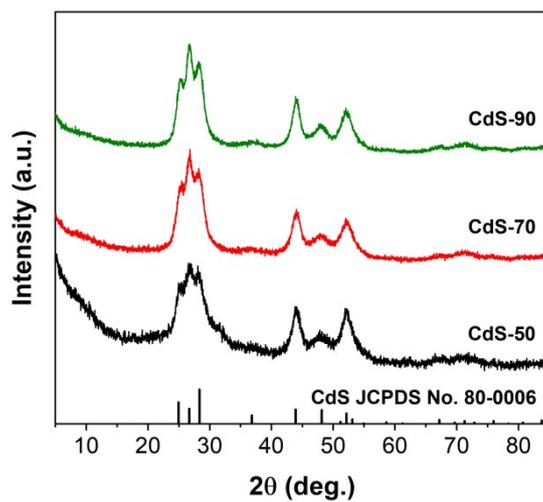


Fig. S2. XRD patterns of CdS hollow spheres prepared with different reaction temperatures.

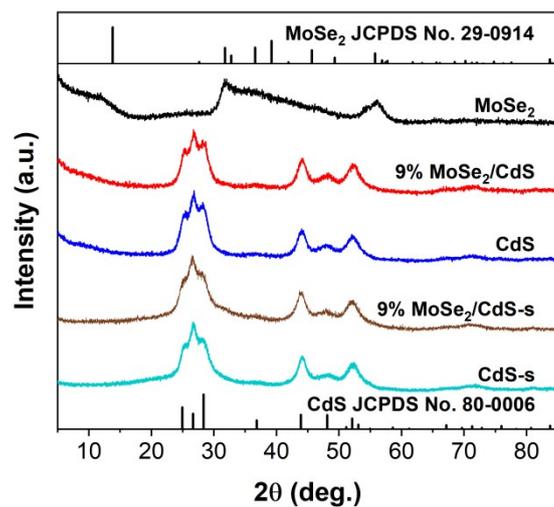


Fig. S3. XRD patterns of different samples.

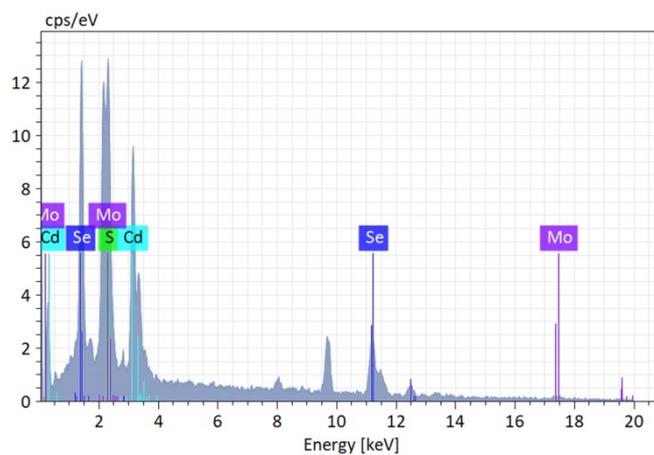


Fig. S4. EDX spectrum of 9% MoSe₂/CdS composite.

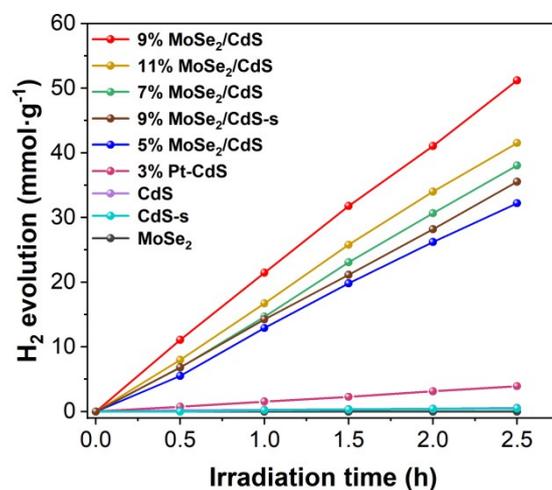


Fig. S5. PHE activities of different samples.

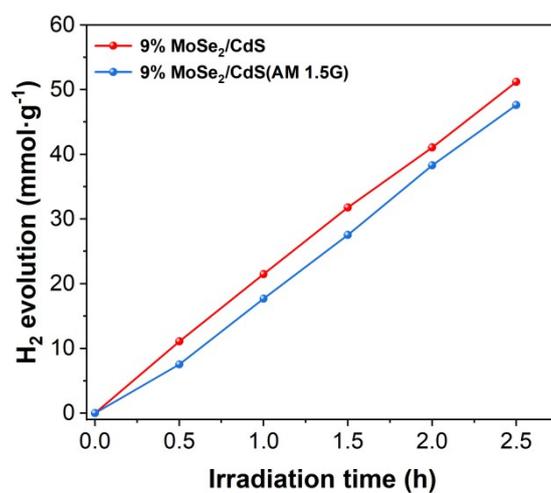


Fig. S6. PHE activities of 9% MoSe₂/CdS measured with 400-nm cut-off filter and AM 1.5G filter.

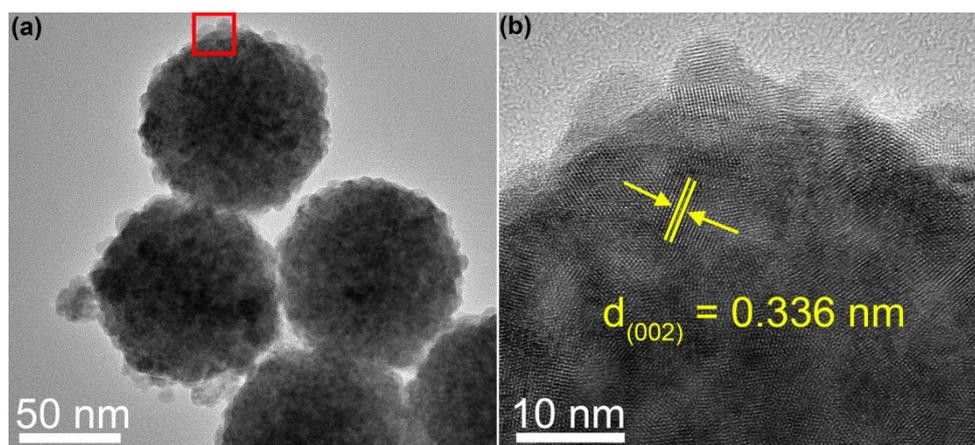


Fig. S7. (a) TEM and (b) HRTEM graphs of CdS-s. (b) is amplified from the marked area in (a).

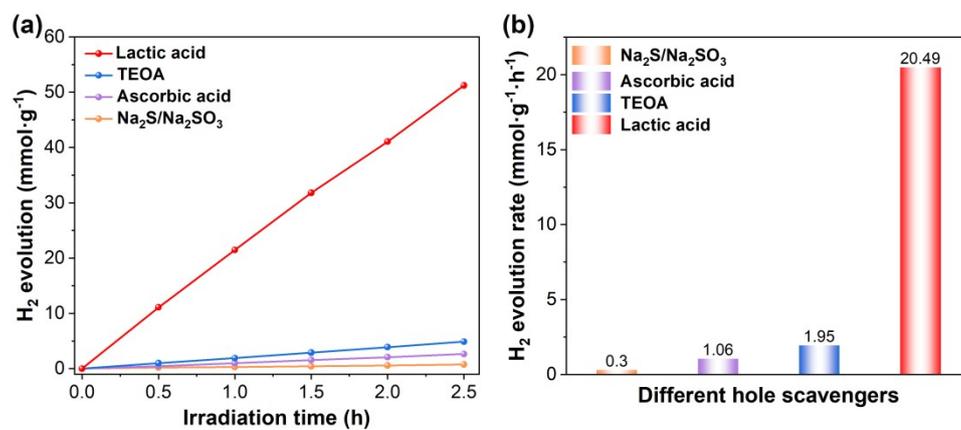


Fig. S8. Average PHE rates of 9% MoSe₂/CdS measured using different hole scavengers.

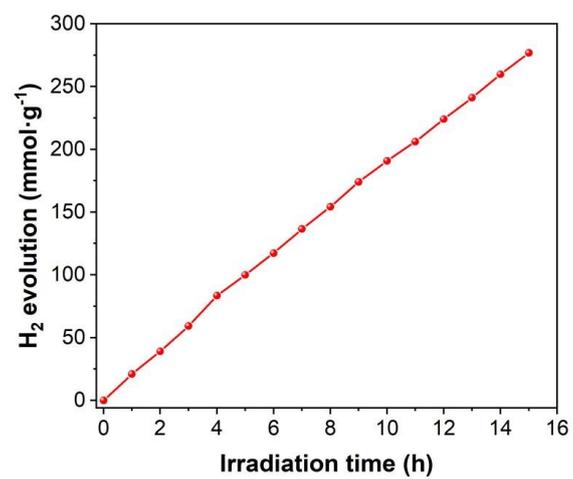


Fig. S9. Long-term PHE stability of 9% MoSe₂/CdS.

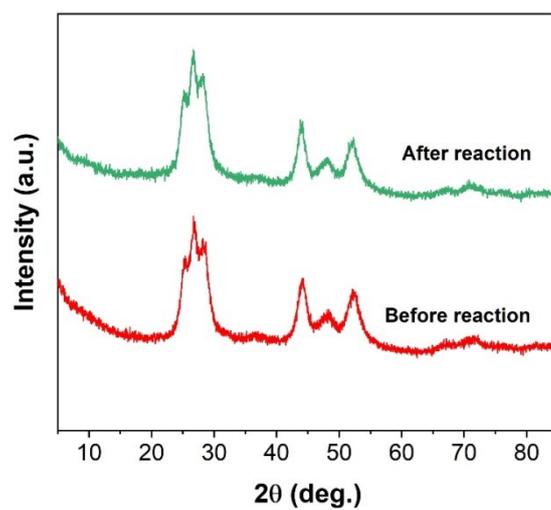


Fig. S10. XRD patterns of 9% MoSe₂/CdS before and after catalytic test.

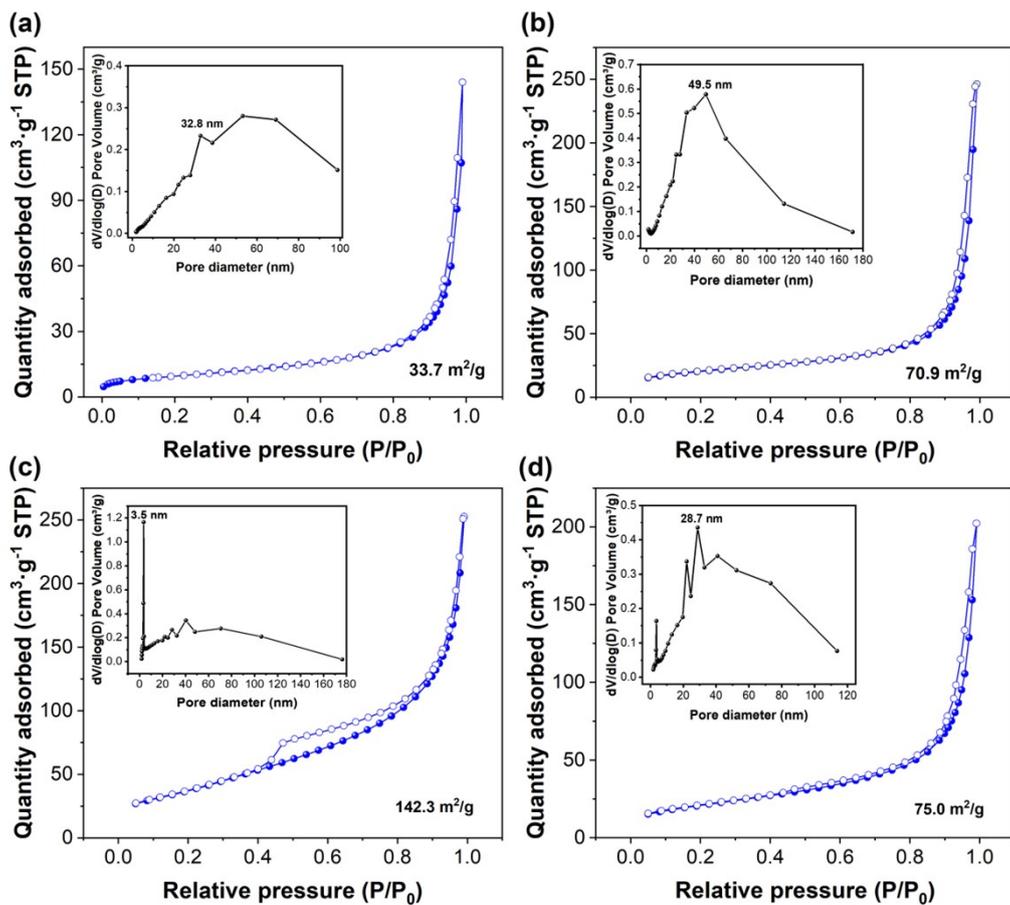


Fig. S11. N_2 adsorption-desorption isotherms and pore-size distribution of (a) CdS-s, (b) CdS, (c) MoSe_2 , and (d) 9% MoSe_2/CdS .

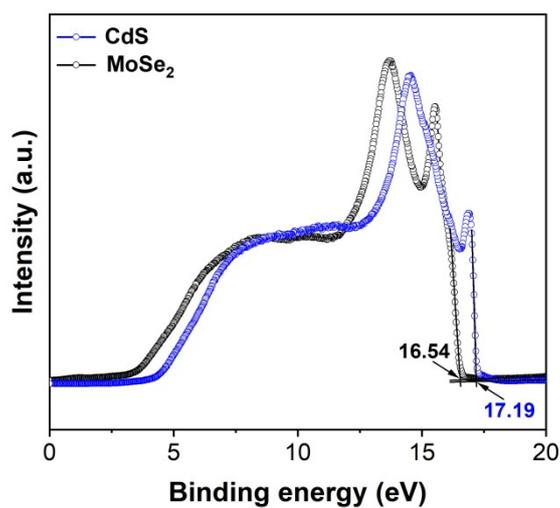


Fig. S12. UPS spectra of CdS and MoSe_2 .

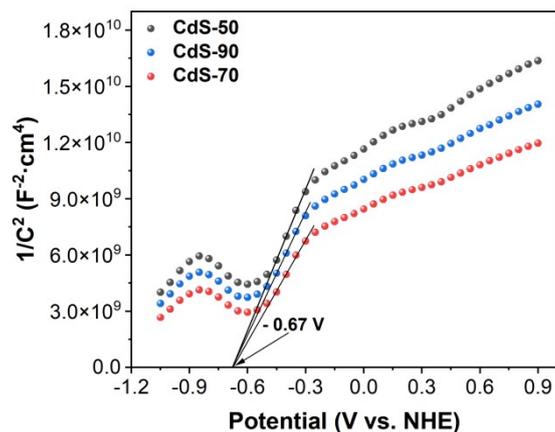


Fig. S13. Mott–Schottky curves of CdS hollow spheres with differing synthesis temperatures.

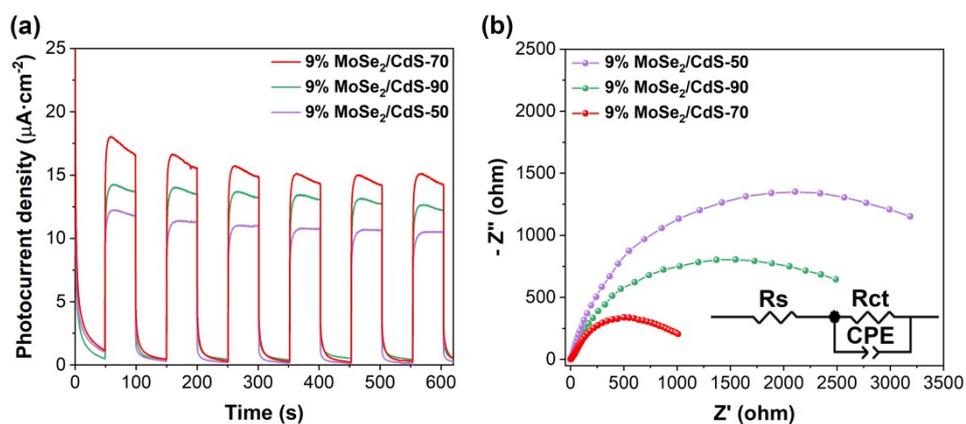


Fig. S14. (a) Photocurrent densities and (b) EIS Nyquist curves of 9% MoSe₂/CdS-50, 9% MoSe₂/CdS-70, and 9% MoSe₂/CdS-90.

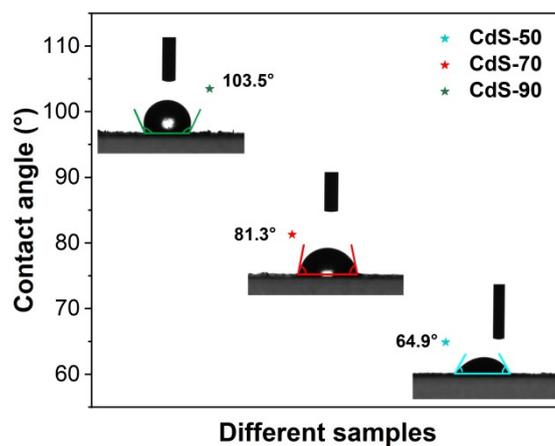


Fig. S15. Static water contact angles of CdS hollow spheres with different formation temperatures.

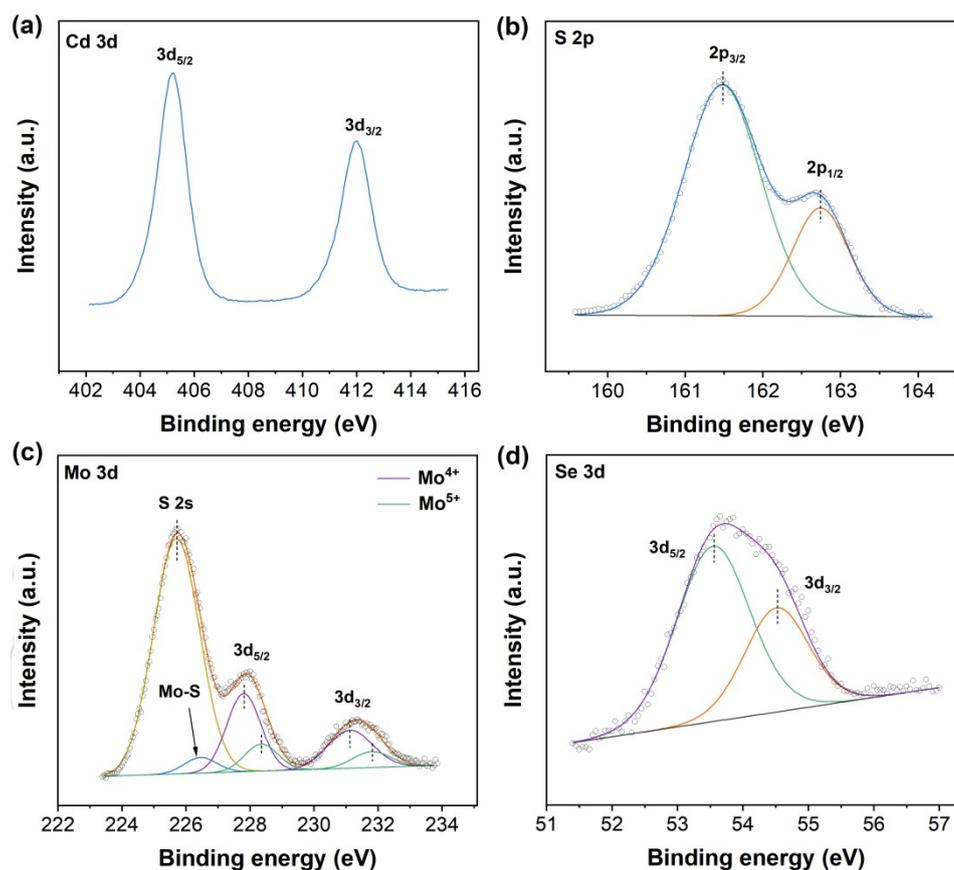


Fig. S16. XPS spectra of (a) Cd, (b) S, (c) Mo, and (d) Se elements of 9% MoSe₂/CdS after catalytic test.

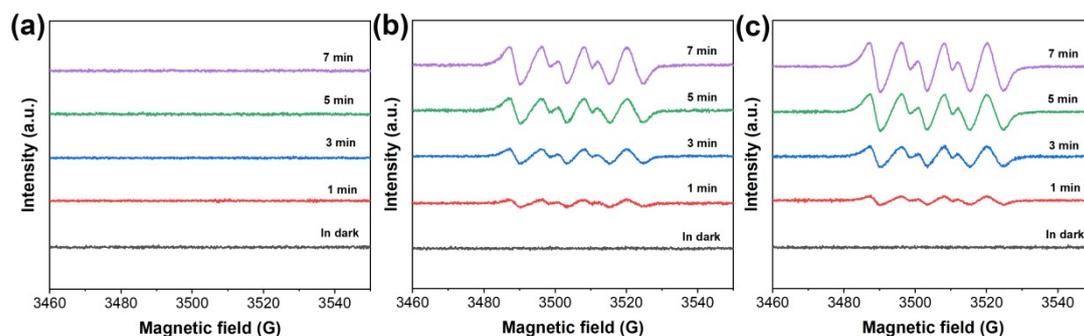


Fig. S17. DMPO·O₂⁻ EPR signals of (a) MoSe₂, (b) CdS, and (c) 9% MoSe₂/CdS detected at certain irradiation periods.

Table S1. PHE activity comparison on our MoSe₂/CdS heterojunction as well as CdS- and MoSe₂-based photocatalysts documented in literatures.

Photocatalyst	Hole scavenger (aqueous solution)	Light source (Xe lamp)	Maximum rate (mmol·g ⁻¹ ·h ⁻¹)	AQY (420 nm)	Reference
MoSe ₂ /CdS	Lactic acid	λ > 400 nm	20.49	10.36% (475 nm)	This work

CdS/Ni-based MOF	Lactic acid	AM 1.5G filter	4.5	6.36%	1
				11.83% (365 nm)	
Cu_{2-x}S/CdS	Na ₂ S/Na ₂ SO ₃	$\lambda \geq 420$ nm	5.75	0.8%	2
Pt, Au/CdS	TEOA	$\lambda > 380$ nm	6.71	3.72% (380 nm)	3
NiPS₃/CdS	Lactic acid	AM 1.5G filter	18.24	12.11% (475 nm)	4
ZnIn₂S₄/CdS	Na ₂ S/Na ₂ SO ₃	$\lambda > 400$ nm	5.68	4.8%	5
H-CeO₂/H-CdS	Lactic acid	$\lambda \geq 420$ nm	8.36	56.5%	6
CdS/Pt/NaTaO₃	Lactic acid	AM 1.5G filter	19.84	7.9% (350 nm)	7
Au/ZnWO₄/CdS	Na ₂ S/Na ₂ SO ₃	420-800 nm	5.48	2.62% (450 nm)	8
				8.8%	
0D/2D CdS/MoO_{3-x}	Lactic acid	350-W Xe lamp	7.44	14.3% (450 nm)	9
Cd/CdS	Na ₂ S/Na ₂ SO ₃	$\lambda > 420$ nm	10.6	12.1%	10
CdS/rGO	Na ₂ S/Na ₂ SO ₃	420-760 nm	2.17	-	11
CdS@TiO₂/Mxene	Lactic acid	-	16.2	-	12
		3-W 420-nm LED lamps			
PtP₂@C/CdS	Lactic acid		9.76	41.67%	13
CdS-MoS₂-CoO_x	TEOA	300-W Xe lamp	7.4	7.6%	14
CdS/WC	Lactic acid	$\lambda \geq 420$ nm	9.18	14.3%	15
CdS/graphene QDs/Pt single atoms	Lactic acid	$\lambda \geq 420$ nm	13.49	35.5%	16
1D CdS/2D MXene	Lactic acid	$\lambda > 420$ nm	8.87	27.8% (450 nm)	17
MA_{1-x}FA_xPbI₃/MoSe₂	perovskite-saturated	AM 1.5G filter	12.53	10.29%	18
Cd_{0.7}Zn_{0.3}S/MoSe₂	Na ₂ S/Na ₂ SO ₃	$\lambda \geq 420$ nm	0.67	1.96% (450 nm)	19
MoSe₂/N-doped C/ZnIn₂S₄	TEOA	$\lambda > 400$ nm	10.31	-	20
CdS/MoSe₂/UiO-66-NH₂	TEOA	500-W Xe lamp	0.426	-	21
				67.04%	
MoSe₂/Mn_{0.5}Cd_{0.5}S	Na ₂ S/Na ₂ SO ₃	$\lambda > 420$ nm	5.6	71.16% (400 nm)	22
MoSe₂/g-C₃N₄	TEOA	$\lambda > 400$ nm	4.0	-	23
Ta₃N₅/MoSe₂	Na ₂ S/Na ₂ SO ₃	300-W Xe lamp	0.81	-	24
TiO₂/MoSe₂	Lactic acid	300-W Xe lamp	0.69	-	25
MoSe₂/CdS	Lactic acid	$\lambda \geq 420$ nm	4.7	15.6% (450 nm)	26
CdS-Cu_{1.81}S	Na ₂ S/Na ₂ SO ₃	300-W Xe lamp	2.71	11.32%	27
Ag₂S/CdS/Cd₂SO₄(OH)₂	Na ₂ S/Na ₂ SO ₃	300-W Xe lamp	9.69	7.60%	28
				33.92%	
CdS-Pd	TEOA	300-W Xe lamp	0.95	27.49% (500 nm)	29
Pt single sites-ZnIn₂S₄	TEOA	$\lambda > 420$ nm	17.5	50.4%	30
CoP@ZnIn₂S₄	TEOA	$\lambda > 420$ nm	7.4	7.6%	31
Mo-ZnIn₂S₄@NiTiO₃	TEOA	$\lambda > 420$ nm	14.06	44.1%	32
ZnIn₂S₄/P-doped C₀/C	TEOA	$\lambda > 420$ nm	2.1	-	33
FeS₂@ZnIn₂S₄	TEOA	AM 1.5G filter	5.05	13.4%	34
Zn-doped Ni₂P/g-C₃N₄	TEOA	AM 1.5G filter	1.077	8.9%	35
F/C co-doped g-C₃N₄	TEOA	$\lambda > 420$ nm	3.87	15.4%	36
B-doped g-C₃N₄/Ni-	TEOA	60-W 420-nm	2.19	1.8%	37

MOF-74		LED lamp			
Cu-coordinated g-C₃N₄	TEOA	$\lambda > 420$ nm	6.53	-	38
NiS/g-C₃N₄	TEOA	420-nm LED lamps	0.244	-	39

Table S2. Fitting parameters for EIS Nyquist plots of MoSe₂, CdS, and 9% MoSe₂/CdS. (Rs: series resistance, Rct: charge transfer resistance, CPE: constant phase angle element)

Sample	Rs (Ω)	Rct (Ω)	CPE-T (F)	CPT-P
CdS	20.56	10125	1.64×10^{-4}	0.808
MoSe ₂	29.22	7149	1.33×10^{-4}	0.825
9% MoSe ₂ /CdS-70	28.42	889	1.73×10^{-4}	0.809
9% MoSe ₂ /CdS-90	22.82	3007	1.13×10^{-4}	0.877
9% MoSe ₂ /CdS-50	22.07	4342	1.04×10^{-4}	0.868

Table S3. Average diameters (calculated by Scherrer's formula), specific surface areas, and static water contact angles of CdS with differing preparation temperatures.

Sample	Average diameter (nm)	Specific surface area (m ² /g)	Water contact angle (°)
CdS-50	4.7	74.1	64.9
CdS-70	4.8	70.9	81.3
CdS-90	5.1	68.4	103.5

Table S4. XPS peak fitting data for MoSe₂, CdS, and 9% MoSe₂/CdS with and without light irradiation.

Species	MoSe ₂	CdS	MoSe ₂ /CdS	light-irradiated MoSe ₂ /CdS
Mo-S	-	-	226.4 eV	226.7 eV
Mo⁴⁺	228.13-231.23 eV	-	227.80-231.13 eV	227.93-231.18 eV
Mo⁵⁺	229.1-232.3 eV	-	228.4-231.9 eV	228.68-232.14 eV
Se²⁻	53.85-54.7 eV	-	53.6-54.5 eV	53.8-54.7 eV
Cd²⁺	-	404.7-411.4 eV	405.3-412.0 eV	405.0-411.7 eV
S²⁻	-	161.2-162.4 eV	161.6-162.8 eV	161.4-162.6 eV

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