

Stereochemically (in)active lone pairs in Sb(III) complexes of a soft scorpionate ligand

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A. Colour changes

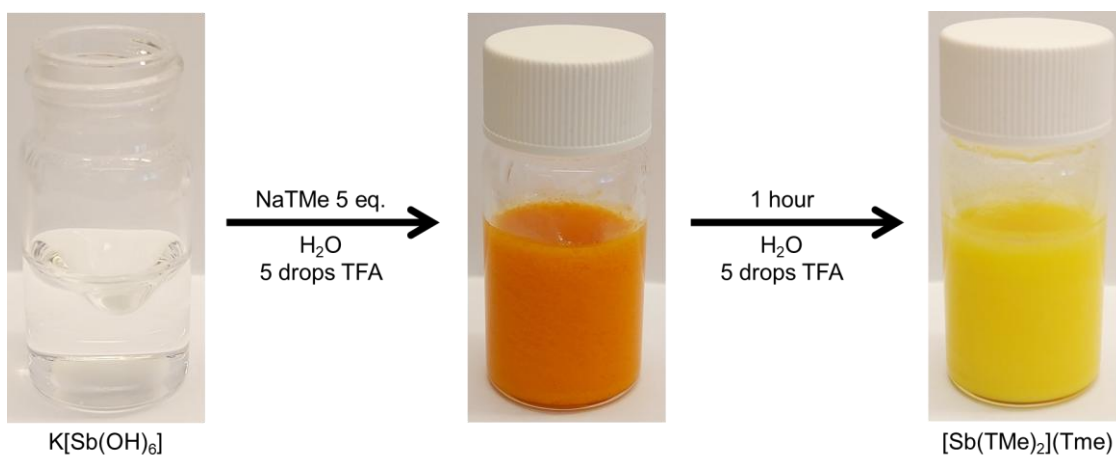


Figure S1. Colour changes during the synthesis of $[\text{Sb}(\text{TMe})_2](\text{TMe})$.

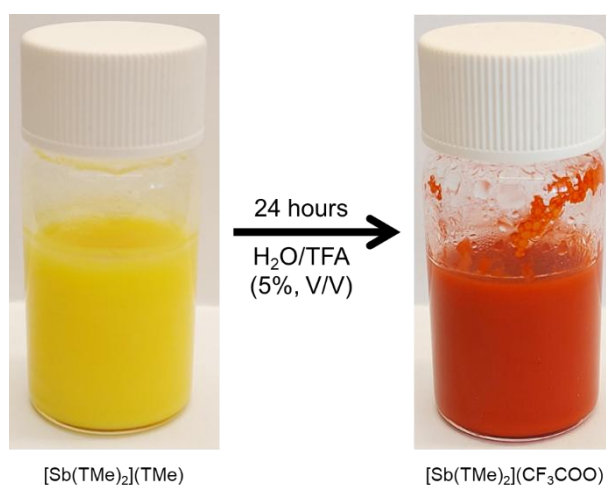


Figure S2. Colour changes during the synthesis of $[\text{Sb}(\text{TMe})_2](\text{CF}_3\text{COO})$.



Figure S3. Co-crystallisation of $[\text{Sb}(\text{TMe})_2](\text{PF}_6) \cdot \text{Me}_2\text{CO}$ (red) and $[\text{Sb}_2(\text{TMe})_3(\text{C}_4\text{H}_6\text{N}_2\text{S})](\text{PF}_6)_3 \cdot 1.7\text{Me}_2\text{CO}$ (yellow).

B. Crystallography

Table S1. Crystal data and structure refinements

Complex	[Sb ₄ Cl ₈ (TMe) ₄] ·4MeOH (1)	[Sb(TMe) ₂] ₂ (TMe) ·Me ₂ CO·3.6H ₂ O (2)	[Sb(TMe) ₂] ₂ (CF ₃ COO) ·1.6 CF ₃ COOH (3a)	[Sb(TMe) ₂] ₂ (PF ₆) ·Me ₂ CO (3b)	[Sb ₂ (TMe) ₃](C ₄ H ₈ N ₂ S) ·1.7Me ₂ CO (4)
Formula	C ₅₂ H ₈₀ N ₂₄ O ₄ S ₁₂ Cl ₈ Sb ₄ B ₄	C ₃₉ H _{60.3} B ₃ N ₁₈ O _{4.4} S ₉ Sb	C _{30.7} H _{36.6} B ₂ F _{7.3} N ₁₂ O _{5.7} S ₆ Sb	C ₂₇ H ₃₈ B ₂ N ₁₂ O ₆ PS ₆ Sb	C _{45.1} H _{64.2} B ₃ F ₁₈ N ₂₀ O _{1.7} P ₃ S ₁₀ Sb ₂
Formula weight	2303.96	1294.97	1148.85	1027.39	1945.20
Temperature/K	100.01(10)	100.00(12)	100.00(10)	100.00(10)	100.00(10)
Crystal system	monoclinic	triclinic	monoclinic	monoclinic	monoclinic
Space group	P2 ₁ /n	P1 ₁ [−]	I2/a	I2/a	C2/c
a/Å	10.86977(5)	10.5707(2)	10.36050(10)	45.083(3)	32.0046(9)
b/Å	26.79870(10)	14.9955(3)	17.2928(2)	9.2529(5)	14.8912(3)
c/Å	15.13528(6)	19.6610(3)	28.3074(3)	30.1304(17)	34.1458(12)
α/°	90	77.2890(10)	90	90	90
β/°	93.6973(4)	79.4900(10)	91.2010(10)	96.992(5)	114.476(4)
γ/°	90	72.973(2)	90	90	90
Volume/Å ³	4399.66(3)	2883.54(10)	5070.50(9)	12475.4(12)	14811.0(8)
Z	2	2	4	12	8
ρ _{calc} /g/cm ³	1.739	1.491	1.505	1.641	1.745
μ/mm ^{−1}	14.993	7.307	7.327	9.052	9.912
F(000)	2288.0	1332.0	2310.0	6216.0	7779.0
Crystal size/mm ³	0.22 × 0.14 × 0.06	0.172 × 0.064 × 0.018	0.2 × 0.12 × 0.04	0.19 × 0.04 × 0.03	0.111 × 0.095 × 0.011
Reflections collected	111568	69071	55956	15324	46676
Indep. refl [R _{int} , R _σ]	8952 [0.0396, 0.0157]	11622 [0.0357, 0.0236]	5151 [0.0378, 0.0156]	15324 [na*, 0.0799]	11519 [0.0583, 0.0587]
Data/restraints/param.	8952/0/510	11622/64/716	5151/6/276	15324/774/831	11519/1773/948
Goodness-of-fit on F ²	1.072	1.036	1.074	0.944	1.151
R1, wR2 [I>=2σ (I)]	0.0189, 0.0465	0.0394, 0.1057	0.0295, 0.0863	0.0439, 0.1019	0.0658, 0.1353
R1, wR2 [all data]	0.0197, 0.0470	0.0425, 0.1088	0.0319, 0.0886	0.0736, 0.1072	0.0789, 0.1419
Largest diff. peak/hole / e Å ^{−3}	0.75/−0.68	1.69/−1.37	1.11/−0.57	0.50/−0.81	1.60/−0.97
CCDC number	2388628	2388626	2388627	2388624	2388625

*Two-Component twin refined using HKLF5 data

i) $[\text{Sb}(\text{TMe})\text{Cl}_2]_4 \cdot 4\text{CH}_3\text{OH}$ (**1**):

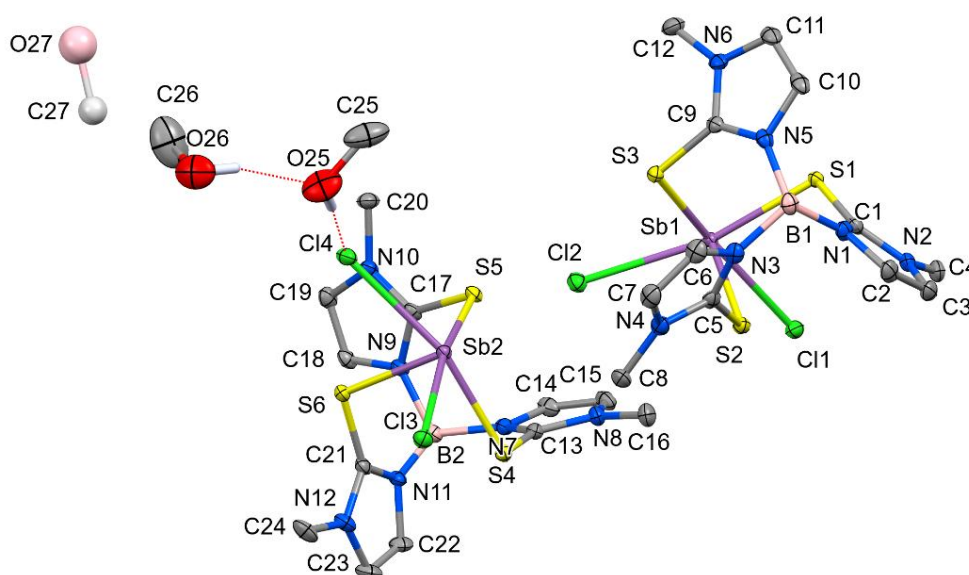


Figure S4. Asymmetric unit of **1** showing 50% probability ellipsoids with atomic labelling scheme. The minor component disordered MeOH molecule (pale colours) was refined with isotropic displacement parameters. Hydrogen atoms omitted for clarity except those involved in H-bonds (red dashed lines).

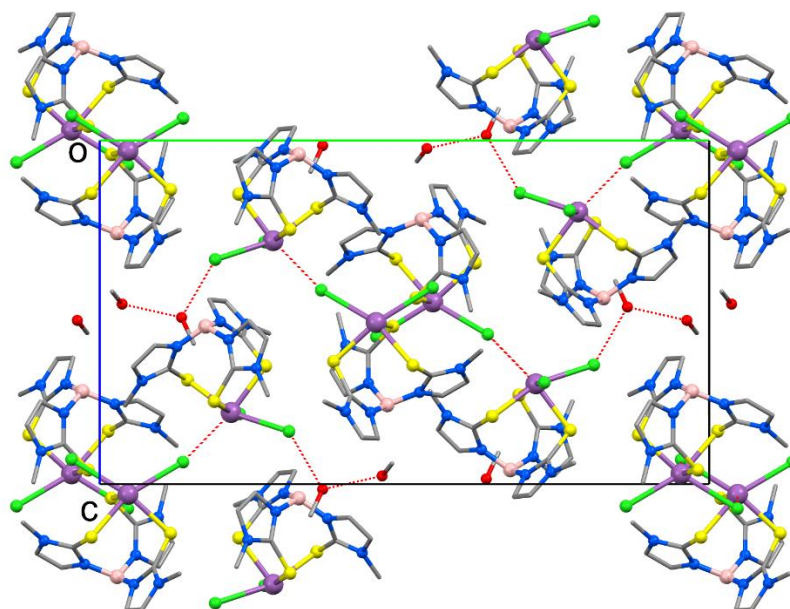


Figure S5. Unit cell packing of **1** viewed down the *a* axis. Hydrogen atoms and minor disorder omitted for clarity.

One of the two independent methanol solvate molecules is disordered (Figure S4), the major component is hydrogen bonded to both a coordinated chloro ligand (O26 $\cdots$ Cl4, 3.0573(19) Å) and to the second methanol (O26 $\cdots$ O25, 2.679(4) Å). There are no particularly striking interactions between neighbouring [SbCl₂(TMe)]₄·4MeOH assemblies (Figure S5), although there are many long C–H $\cdots$ (S, Cl) interactions (Table S2, Figure S5).

Table S2. Selected geometric parameters (Å, °) for [Sb(TMe)Cl₂]₄·4CH₃OH (I)

Sb1—S1	2.5761 (4)	Sb2—S4	2.5640 (5)
Sb1—S2	2.5155 (4)	Sb2—S5	2.6165 (4)
Sb1—S3	2.5590 (4)	Sb2—S6	2.5386 (4)
Sb1—Cl1	2.9268 (5)	Sb2—Cl3	2.7805 (5)
Sb1—Cl2	2.9419 (5)	Sb2—Cl4	2.8800 (5)
Sb1—Cl1 ⁱ	3.1582 (5)	Sb2—Cl2	3.1684 (5)
Cl1—Sb1—Cl1 ⁱ	101.116 (11)	Cl3—Sb2—Cl2	110.879 (13)
Cl1—Sb1—Cl2	114.463 (13)	Cl3—Sb2—Cl4	104.179 (14)
Cl2—Sb1—Cl1 ⁱ	113.524 (13)	Cl4—Sb2—Cl2	106.630 (13)
S1—Sb1—Cl1	82.823 (14)	S4—Sb2—Cl2	89.075 (13)
S1—Sb1—Cl1 ⁱ	65.616 (13)	S4—Sb2—Cl3	75.871 (14)
S1—Sb1—Cl2	161.855 (14)	S4—Sb2—Cl4	162.732 (14)
S2—Sb1—Cl1	72.309 (14)	S4—Sb2—S5	91.111 (14)
S2—Sb1—Cl1 ⁱ	156.939 (14)	S5—Sb2—Cl2	70.716 (13)
S2—Sb1—Cl2	88.938 (14)	S5—Sb2—Cl3	166.748 (15)
S2—Sb1—S1	91.434 (14)	S5—Sb2—Cl4	87.496 (14)
S2—Sb1—S3	92.107 (15)	S6—Sb2—Cl2	160.350 (14)
S3—Sb1—Cl1	162.748 (14)	S6—Sb2—Cl3	88.001 (14)
S3—Sb1—Cl1 ⁱ	90.373 (13)	S6—Sb2—Cl4	72.352 (14)
S3—Sb1—Cl2	71.449 (14)	S6—Sb2—S4	90.435 (15)
S3—Sb1—S1	90.410 (14)	S6—Sb2—S5	89.656 (14)
Sb1—Cl1—Sb1 ⁱ	78.884 (11)	Sb1—Cl2—Sb2	165.327 (18)

Table S3. Hydrogen-bond geometry ($\text{\AA}$, $^\circ$) for $[\text{Sb}(\text{TMe})\text{Cl}_2]_4 \cdot 4\text{CH}_3\text{OH}$ (**1**)

$D\cdots H\cdots A$	$D\cdots H$	$H\cdots A$	$D\cdots A$	$D\cdots H\cdots A$
O25—H25 $\cdots$ C14	0.84	2.22	3.0573 (19)	174
O26—H26 $\cdots$ O25	0.84	1.87	2.679 (4)	161
C2—H2 $\cdots$ S4 ⁱⁱ	0.95	2.91	3.515 (2)	123
C3—H3 $\cdots$ Cl3 ⁱⁱ	0.95	2.65	3.583 (2)	167
C4—H4A $\cdots$ Cl1	0.98	2.71	3.497 (2)	138
C7—H7 $\cdots$ Cl4 ⁱⁱⁱ	0.95	2.91	3.490 (2)	120
C7—H7 $\cdots$ S6 ⁱⁱⁱ	0.95	2.77	3.712 (2)	170
C8—H8A $\cdots$ Cl2	0.98	2.94	3.745 (2)	140
C8—H8C $\cdots$ S2 ⁱⁱ	0.98	2.91	3.706 (2)	139
C10—H10 $\cdots$ O26 ^{iv}	0.95	2.30	3.238 (3)	168
C12—H12A $\cdots$ Cl1 ⁱ	0.98	2.87	3.751 (2)	149
C12—H12B $\cdots$ Cl4 ^v	0.98	2.84	3.596 (2)	135
C12—H12C $\cdots$ Cl3 ^v	0.98	2.93	3.884 (2)	166
C14—H14 $\cdots$ O25 ^{vi}	0.95	2.45	3.282 (3)	146
C15—H15 $\cdots$ S1 ⁱ	0.95	2.97	3.543 (2)	120
C16—H16B $\cdots$ Cl2	0.98	2.85	3.622 (2)	136
C16—H16C $\cdots$ Cl1	0.98	2.70	3.580 (2)	150
C18—H18 $\cdots$ Cl3 ^{vi}	0.95	2.75	3.677 (2)	165
C19—H19 $\cdots$ Cl2 ^{vi}	0.95	2.83	3.723 (2)	158
C20—H20A $\cdots$ Cl3 ^{vii}	0.98	2.93	3.859 (2)	160
C20—H20C $\cdots$ S3 ^{vi}	0.98	2.84	3.732 (2)	151
C23—H23 $\cdots$ Cl4 ^{viii}	0.95	2.55	3.502 (2)	176
C24—H24B $\cdots$ Cl3	0.98	2.87	3.769 (2)	154

Symmetry codes: (i) $-x+1, -y+1, -z+1$; (ii) $-x+2, -y+1, -z+1$; (iii) $x+1/2, -y+1/2, z+1/2$; (iv) $-x+3/2, y+1/2, -z+3/2$; (v) $x-1/2, -y+1/2, z+1/2$; (vi) $x-1/2, -y+1/2, z-1/2$; (vii) $x-1, y, z$; (viii) $x+1/2, -y+1/2, z-1/2$.

ii) $[\text{Sb}(\text{TMe})_2](\text{TMe}) \cdot (\text{Me})_2\text{CO} \cdot 3.6\text{H}_2\text{O}$ (**2**)

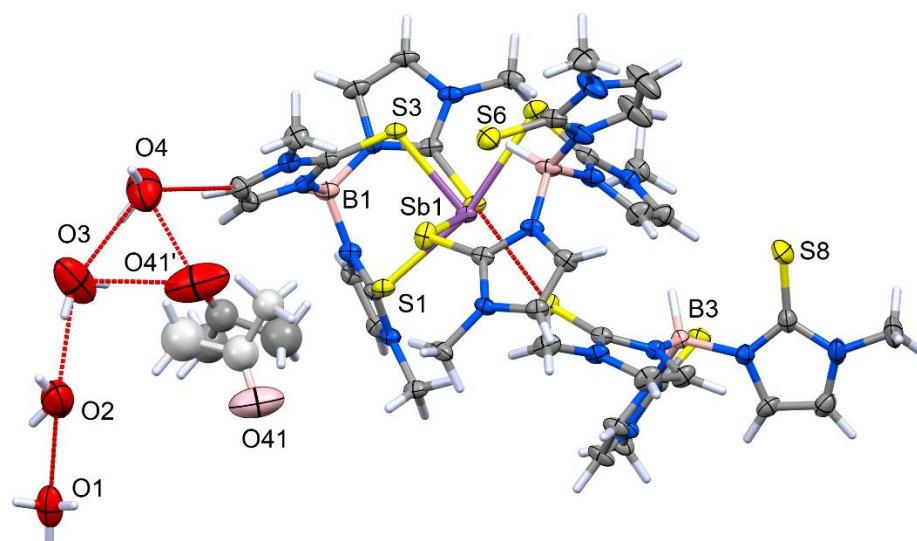


Figure S6. Asymmetric unit of $[\text{Sb}(\text{TMe})_2](\text{TMe}) \cdot (\text{Me})_2\text{CO} \cdot 3.6\text{H}_2\text{O}$ (**2**) showing 50% probability ellipsoids with hydrogen bonding scheme (red dashed lines). The carbon atoms of the disordered acetone molecule were refined with isotropic displacement parameters.

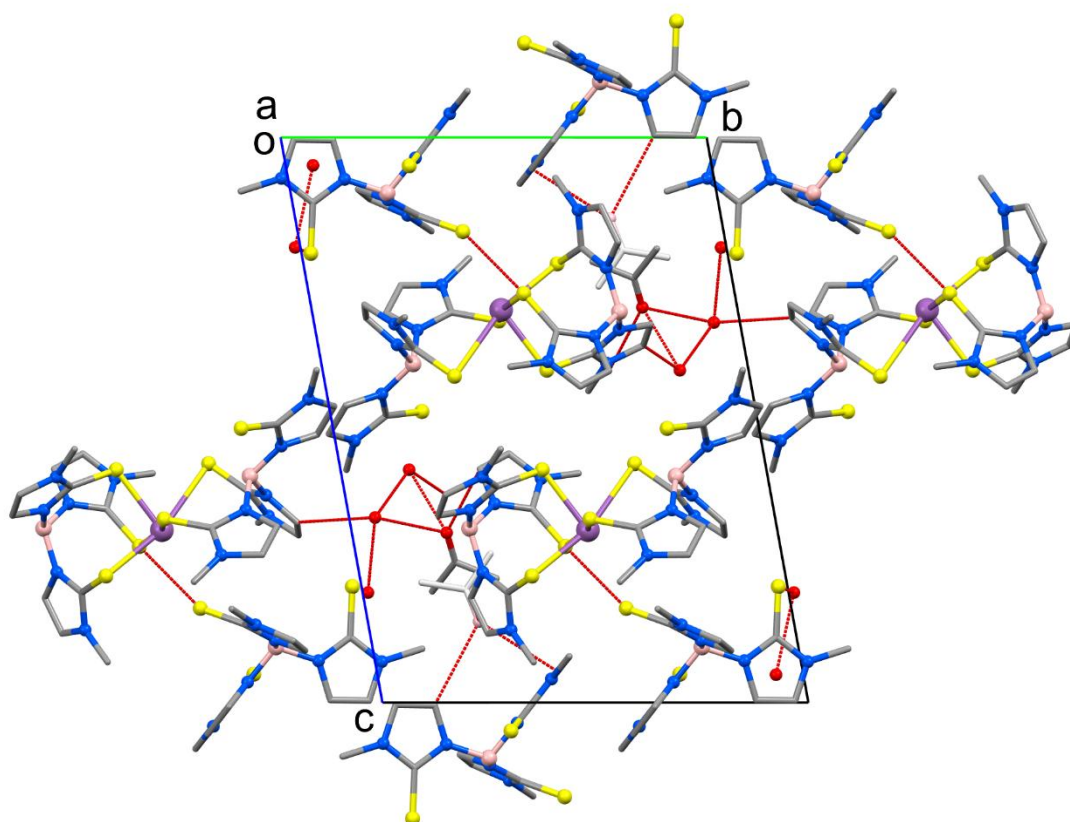


Figure S7. Unit cell packing of **2** viewed down the *a* axis. Hydrogen atoms omitted for clarity.

The solvate molecules are all involved in hydrogen bonding, there is are three H₂O···S bond there are also several C-H···O and C-H···S interactions (Table S5), but no notable specific interactions that could be said to stabilize the [Sb(κ^3 -TMe)(κ^2 -TMe)](TMe) structure.

Table S4. Selected geometric parameters ($\text{\AA}$, $^\circ$) for [Sb(TMe)₂](TMe)·Me₂CO·3.6H₂O (2)

Sb1—S1	2.6023 (7)	Sb1—S4	2.8727 (7)
Sb1—S2	2.6301 (7)	Sb1—S5	2.9245 (8)
Sb1—S3	2.5462 (7)	Sb1—S7	3.2579 (7)
S2—S7	3.1043 (10)		
S1—Sb1—S2	89.57 (2)	S3—Sb1—S1	91.54 (2)
S1—Sb1—S4	75.51 (2)	S3—Sb1—S2	91.20 (2)
S1—Sb1—S5	165.53 (2)	S3—Sb1—S4	83.28 (2)
S2—Sb1—S4	163.88 (2)	S3—Sb1—S5	74.09 (2)
S2—Sb1—S5	89.08 (2)	S4—Sb1—S5	103.83 (2)

Table S5. Hydrogen-bond geometry ($\text{\AA}$, $^\circ$) for $[\text{Sb}(\text{TMe})_2](\text{TMe})\cdot\text{Me}_2\text{CO}\cdot 3.6\text{H}_2\text{O}$ (2)

$D\cdots H\cdots A$	$D\cdots H$	$H\cdots A$	$D\cdots A$	$D\cdots H\cdots A$
$\text{O1}\cdots\text{H1B}\cdots\text{S9}^{\text{viii}}$	0.85	2.71	3.429 (3)	144
$\text{O1}\cdots\text{H1C}\cdots\text{O1}^{\text{ix}}$	0.82	2.49	2.917 (7)	113
$\text{O2}\cdots\text{H2B}\cdots\text{S8}^{\text{viii}}$	0.85	2.83	3.332 (5)	119
$\text{O2}\cdots\text{H2C}\cdots\text{O1}$	0.85	2.26	2.859 (7)	128
$\text{O3}\cdots\text{H3B}\cdots\text{O41}'$	0.86 (2)	2.15 (7)	2.938 (18)	152 (12)
$\text{O3}\cdots\text{H3C}\cdots\text{O2}$	0.84 (2)	1.97 (3)	2.800 (8)	168 (10)
$\text{O4}\cdots\text{H4D}\cdots\text{O3}$	0.85 (2)	2.06 (8)	2.812 (11)	148 (13)
$\text{O4}\cdots\text{H4E}\cdots\text{S5}$	0.843	0.843	3.554	130.41
$\text{C3}\cdots\text{H3}\cdots\text{S9}^{\text{i}}$	0.95	2.78	3.636 (3)	151
$\text{C4}\cdots\text{H4A}\cdots\text{S7}$	0.98	2.89	3.746 (3)	147
$\text{C7}\cdots\text{H7}\cdots\text{S5}^{\text{ii}}$	0.95	2.63	3.423 (3)	141
$\text{C8}\cdots\text{H8B}\cdots\text{S4}^{\text{iii}}$	0.98	2.79	3.713 (3)	157
$\text{C8}\cdots\text{H8C}\cdots\text{S5}$	0.98	2.98	3.842 (3)	147
$\text{C10}\cdots\text{H10}\cdots\text{S8}^{\text{iv}}$	0.95	2.83	3.762 (3)	166
$\text{C11}\cdots\text{H11}\cdots\text{O4}$	0.95	2.37	3.257 (9)	156
$\text{C12}\cdots\text{H12A}\cdots\text{S4}$	0.98	2.96	3.820 (4)	147
$\text{C27}\cdots\text{H27}\cdots\text{O1}^{\text{v}}$	0.95	2.59	3.513 (4)	163
$\text{C28}\cdots\text{H28B}\cdots\text{S9}^{\text{vi}}$	0.98	2.77	3.691 (3)	156
$\text{C30}\cdots\text{H30}\cdots\text{O41}^{\text{vii}}$	0.95	2.33	3.229 (9)	158
$\text{C14}\cdots\text{H14}\cdots\text{O3}^{\text{v}}$	0.95	2.62	3.513 (7)	157
$\text{C16}\cdots\text{H16B}\cdots\text{S2}^{\text{vi}}$	0.98	2.94	3.791 (3)	146
$\text{C16}\cdots\text{H16C}\cdots\text{S9}^{\text{vi}}$	0.98	2.81	3.749 (4)	162
$\text{C18}\cdots\text{H18}\cdots\text{O3}^{\text{v}}$	0.95	2.49	3.335 (6)	148
$\text{C19}\cdots\text{H19}\cdots\text{S8}$	0.95	2.92	3.743 (4)	146
$\text{C20}\cdots\text{H20C}\cdots\text{S8}$	0.98	2.96	3.890 (4)	159

Symmetry codes: (i) $-x+2, -y+1, -z+2$; (ii) $-x+2, -y+1, -z+1$; (iii) $x+1, y, z$; (iv) $x, y-1, z$; (v) $x, y+1, z$; (vi) $x-1, y, z$; (vii) $-x+1, -y+1, -z+2$; (viii) $x-1, y-1, z$; (ix) $-x, -y, -z+2$ (x) $-x+1, -y+1, -z+1$.

iii) $[\text{Sb}(\text{TMe})_2](\text{F}_3\text{C}_2\text{O}_2) \cdot \frac{1}{2}\text{Me}_2\text{CO} [+1.6\text{CF}_3\text{COOH}]$ (**3a**)

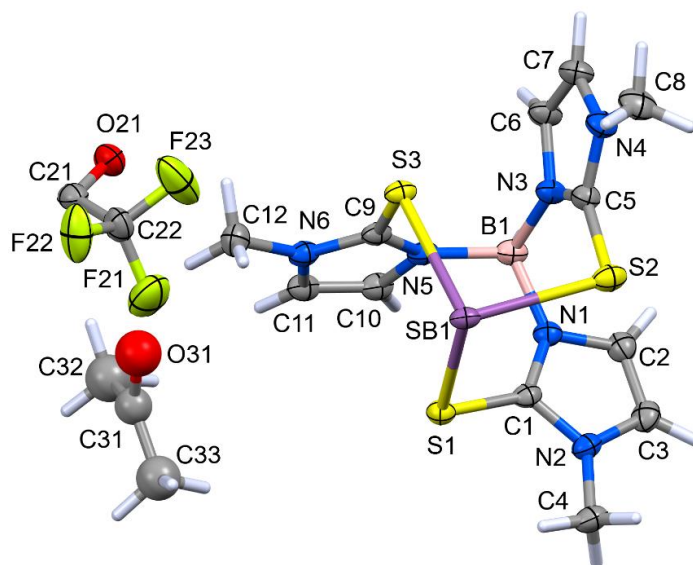


Figure S8. The asymmetric unit of **3a** showing 50% ADPs for non-hydrogen atoms except for the 25% occupancy acetone molecule. The trifluoroacetate anion is disordered around a 2-fold axis.

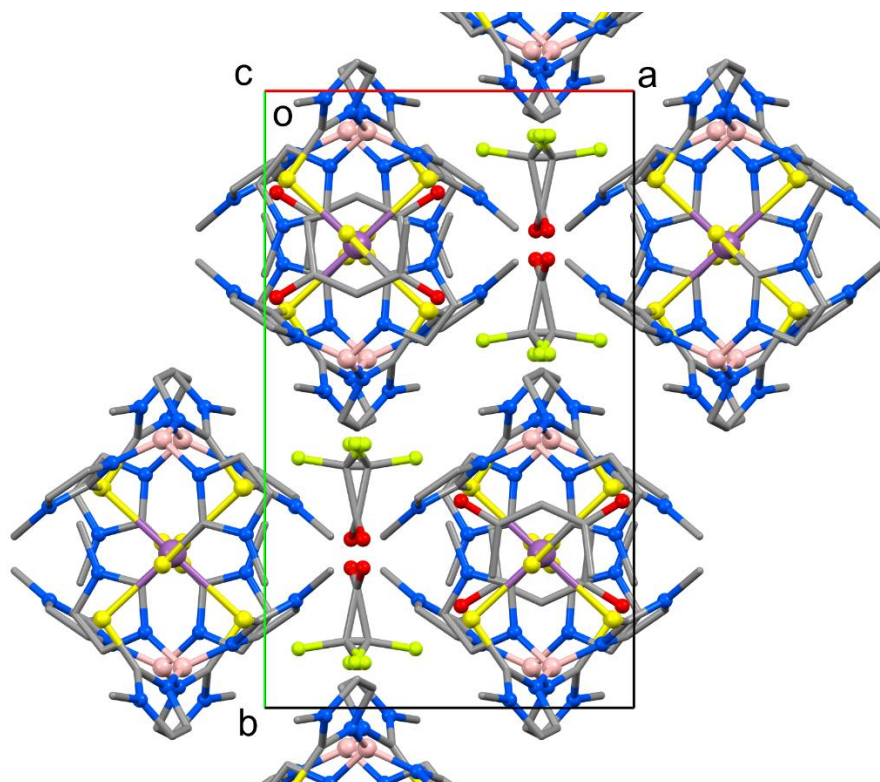


Figure S9. Packing diagram for **3a** viewed down the *c* axis.

Table S6. Selected geometric parameters ($\text{\AA}$, $^\circ$) for **3a**

Sb1—S1	2.7311 (5)	Sb1—S1 ⁱ	2.7310 (5)
Sb1—S2	2.7365 (6)	Sb1—S2 ⁱ	2.7365 (6)
Sb1—S3	2.7108 (5)	Sb1—S3 ⁱ	2.7108 (5)
S1 ⁱ —Sb1—S1	180.0	S3—Sb1—S1	94.238 (17)
S1 ⁱ —Sb1—S2 ⁱ	92.013 (17)	S3 ⁱ —Sb1—S1 ⁱ	94.239 (17)
S1 ⁱ —Sb1—S2	87.987 (17)	S3 ⁱ —Sb1—S2 ⁱ	91.973 (17)
S1—Sb1—S2	92.014 (17)	S3 ⁱ —Sb1—S2	88.027 (17)
S1—Sb1—S2 ⁱ	87.986 (17)	S3—Sb1—S2	91.972 (17)
S2—Sb1—S2 ⁱ	180.0	S3—Sb1—S2 ⁱ	88.027 (17)
S3—Sb1—S1 ⁱ	85.763 (18)	S3 ⁱ —Sb1—S3	180.0
S3 ⁱ —Sb1—S1	85.761 (18)		

Symmetry codes: (i) $-x+1/2, -y+1/2, -z+1/2$

iv) $[\text{Sb}(\text{TMe})_2](\text{PF}_6) \cdot \text{Me}_2\text{CO}$ (**3b**)

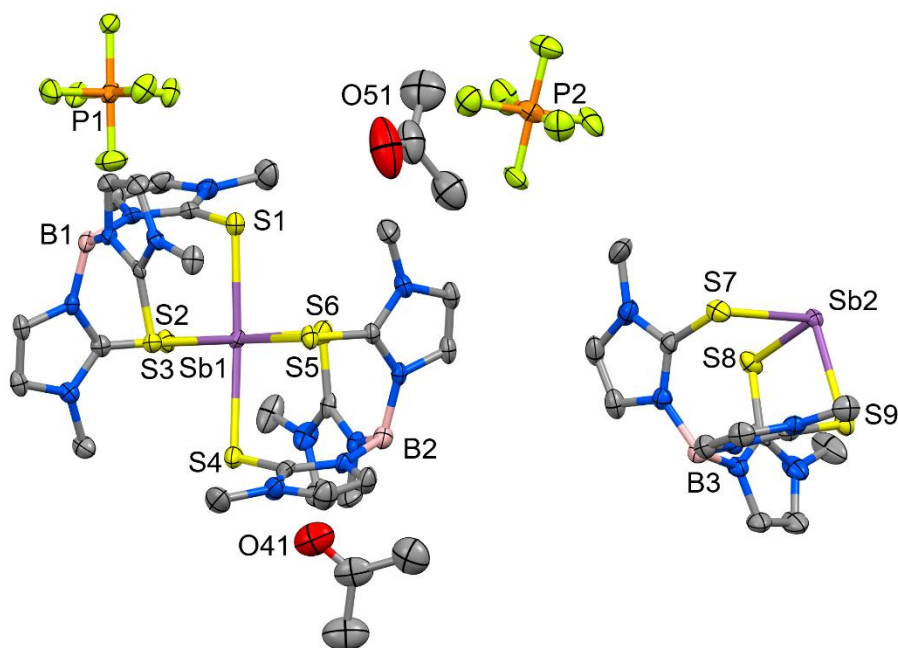


Figure S20. Asymmetric unit of (**3b**) showing 50% probability ellipsoids (hydrogen atoms omitted).

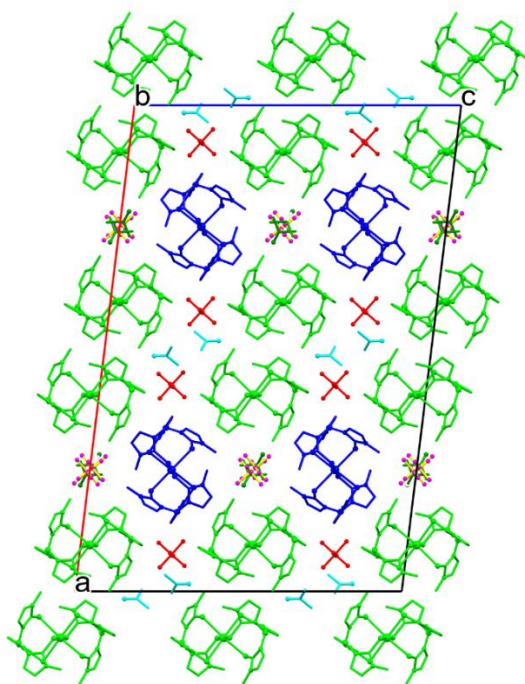


Figure S11. Unit cell packing of **3** viewed down the *b* axis. Atoms are coloured by symmetry equivalence, the strictly centrosymmetric cation is shown in blue. Hydrogen atoms omitted for clarity.

Table S7. Selected geometric parameters ($\text{\AA}$, $^\circ$) for **3b**

Sb1—S1	2.698 (2)	Sb2—S7	2.7307 (16)
Sb1—S2	2.714 (2)	Sb2—S8	2.7337 (19)
Sb1—S3	2.6825 (16)	Sb2—S9	2.730 (2)
Sb1—S4	2.757 (2)	Sb2—S7 ⁱ	2.7308 (16)
Sb1—S5	2.7338 (19)	Sb2—S8 ⁱ	2.7337 (19)
Sb1—S6	2.7692 (18)	Sb2—S9 ⁱ	2.730 (2)
S1—Sb1—S2	91.71 (6)	S7—Sb2—S7 ⁱ	180.00 (4)
S1—Sb1—S4	173.95 (6)	S7—Sb2—S8	90.41 (5)
S1—Sb1—S5	88.26 (6)	S7—Sb2—S8 ⁱ	89.59 (5)
S1—Sb1—S6	88.76 (6)	S7 ⁱ —Sb2—S8	89.59 (5)
S2—Sb1—S4	90.94 (6)	S7 ⁱ —Sb2—S8 ⁱ	90.41 (5)
S2—Sb1—S5	179.64 (5)	S8 ⁱ —Sb2—S8	180.0
S2—Sb1—S6	84.27 (5)	S9—Sb2—S7	94.43 (5)
S3—Sb1—S1	89.81 (5)	S9—Sb2—S7 ⁱ	85.57 (5)
S3—Sb1—S2	94.06 (5)	S9 ⁱ —Sb2—S7 ⁱ	94.43 (5)
S3—Sb1—S4	84.58 (5)	S9 ⁱ —Sb2—S7	85.57 (5)
S3—Sb1—S5	86.30 (5)	S9 ⁱ —Sb2—S8	86.84 (6)
S3—Sb1—S6	177.77 (6)	S9 ⁱ —Sb2—S8 ⁱ	93.16 (6)
S4—Sb1—S6	96.91 (6)	S9—Sb2—S8	93.16 (6)
S5—Sb1—S4	89.12 (6)	S9—Sb2—S8 ⁱ	86.84 (6)
S5—Sb1—S6	95.37 (5)	S9—Sb2—S9 ⁱ	180.0

v) $[\text{Sb}_2(\text{TMe})_3(\text{C}_4\text{H}_6\text{N}_2\text{S})](\text{PF}_6)_3 \cdot 1.7\text{Me}_2\text{CO}$ (**4**)

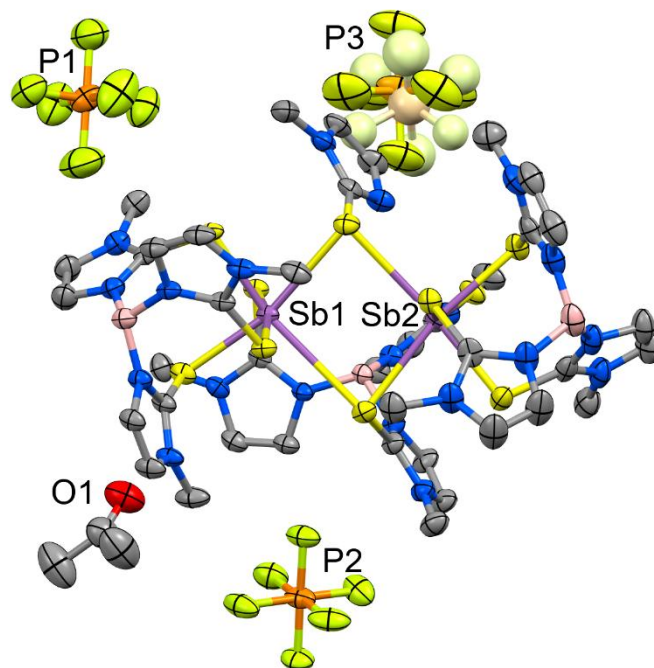


Figure S12. Asymmetric unit of $[\text{Sb}_2(\text{TMe})_3(\text{C}_4\text{H}_6\text{N}_2\text{S})](\text{PF}_6)_3 \cdot 1.7\text{Me}_2\text{CO}$ (**4**) showing 50% probability ellipsoids (except for the minor component of disorder); hydrogen atoms omitted.

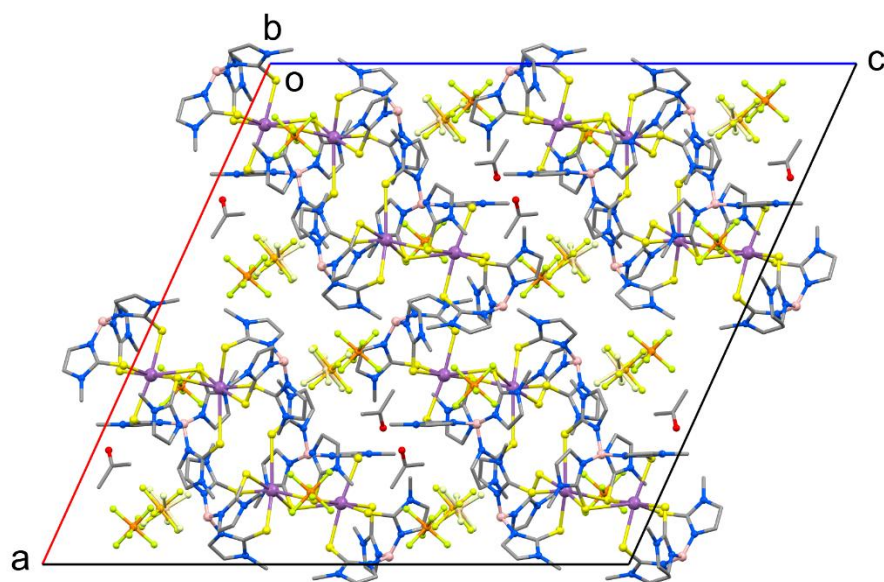


Figure S13. Unit cell packing of **4** viewed down the *b* axis. Hydrogen atoms omitted for clarity.

Table S8. Selected geometric parameters ($\text{\AA}$, $^\circ$) for **4**

Sb1—S1	2.672 (2)	Sb2—S4	2.519 (2)
Sb1—S2	2.507 (2)	Sb2—S5	2.630 (2)
Sb1—S3	2.570 (2)	Sb2—S6	2.515 (2)
Sb1—S7	2.794 (2)	Sb2—S8	2.889 (2)
Sb1—S8	3.385 (2)	Sb2—S9	3.265 (2)
Sb1—S10	3.004 (2)	Sb2—S10	3.191 (2)
S1—Sb1—S7	170.76 (7)	S4—Sb2—S5	89.47 (7)
S1—Sb1—S8	69.37 (6)	S4—Sb2—S8	90.30 (7)
S1—Sb1—S10	90.89 (6)	S4—Sb2—S9	155.93 (7)
S2—Sb1—S1	90.47 (7)	S4—Sb2—S10	73.21 (7)
S2—Sb1—S3	93.15 (7)	S5—Sb2—S8	169.09 (7)
S2—Sb1—S7	89.28 (7)	S5—Sb2—S9	67.65 (6)
S2—Sb1—S8	154.76 (7)	S5—Sb2—S10	92.64 (7)
S2—Sb1—S10	73.20 (7)	S6—Sb2—S4	94.73 (8)
S3—Sb1—S1	90.55 (7)	S6—Sb2—S5	92.69 (8)
S3—Sb1—S7	80.25 (6)	S6—Sb2—S8	76.46 (7)
S3—Sb1—S8	101.77 (7)	S6—Sb2—S9	93.78 (7)
S3—Sb1—S10	166.28 (7)	S6—Sb2—S10	166.76 (7)
S7—Sb1—S8	113.08 (6)	S8—Sb2—S9	113.60 (6)
S7—Sb1—S10	97.87 (6)	S8—Sb2—S10	97.73 (6)
S10—Sb1—S8	91.49 (6)	S10—Sb2—S9	99.45 (6)

C. NMR spectra

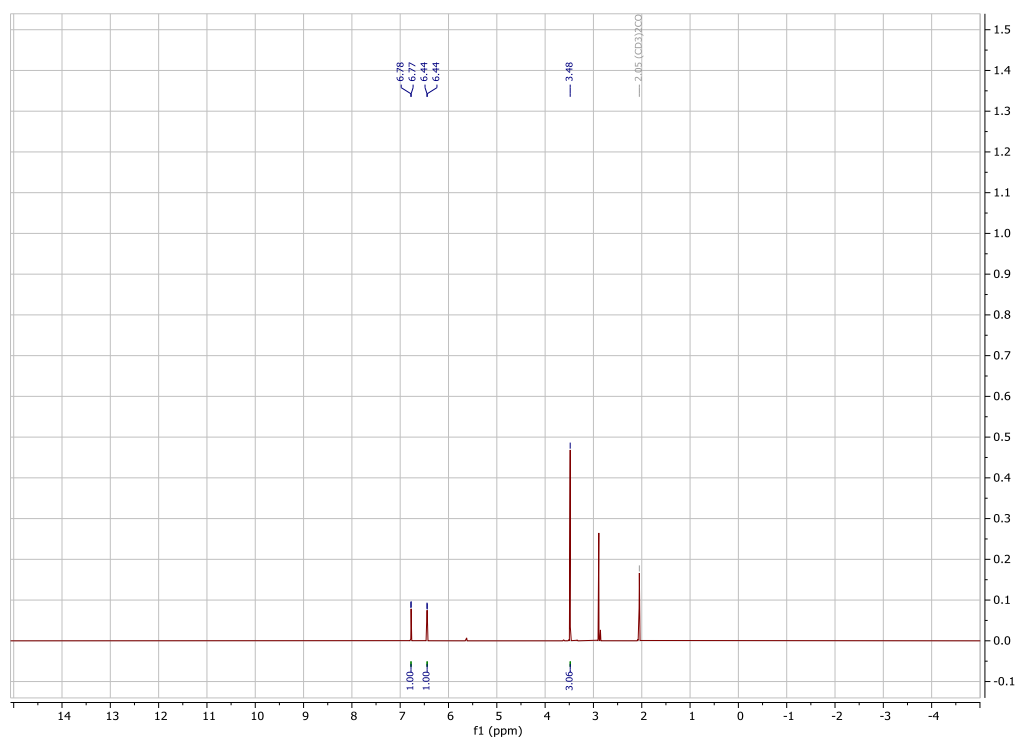


Figure S14. ^1H -NMR spectrum of NaTMe in d_6 -acetone.

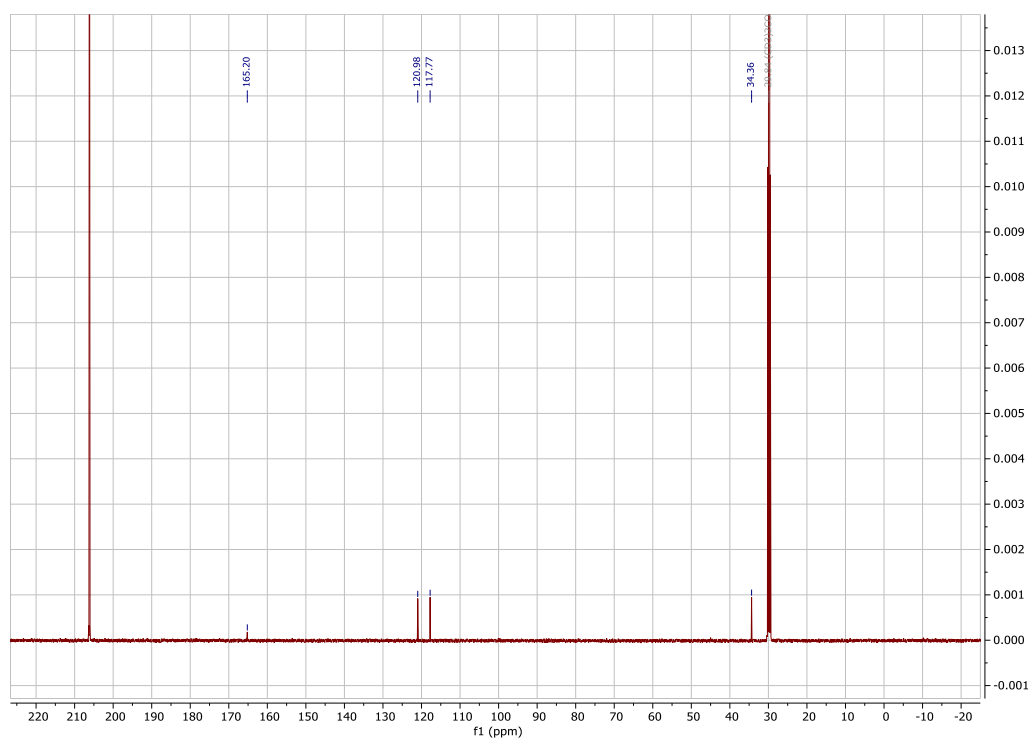


Figure S15. ^{13}C -NMR spectrum of NaTMe in d_6 -acetone.

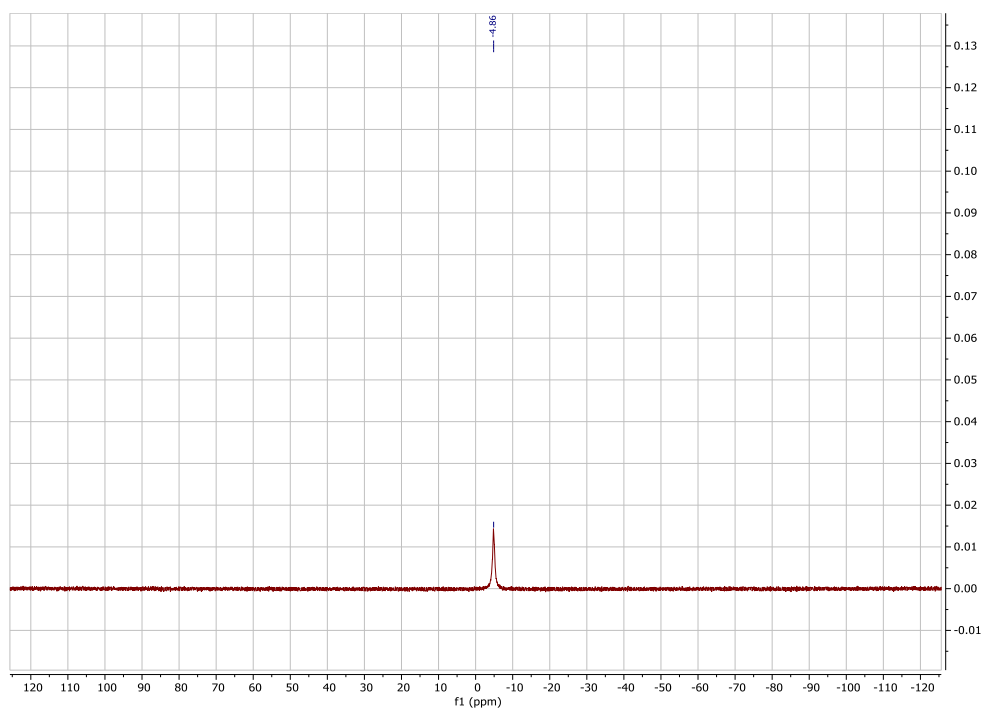


Figure S16. ^{11}B -NMR spectrum of NaTMe in d_6 -acetone.

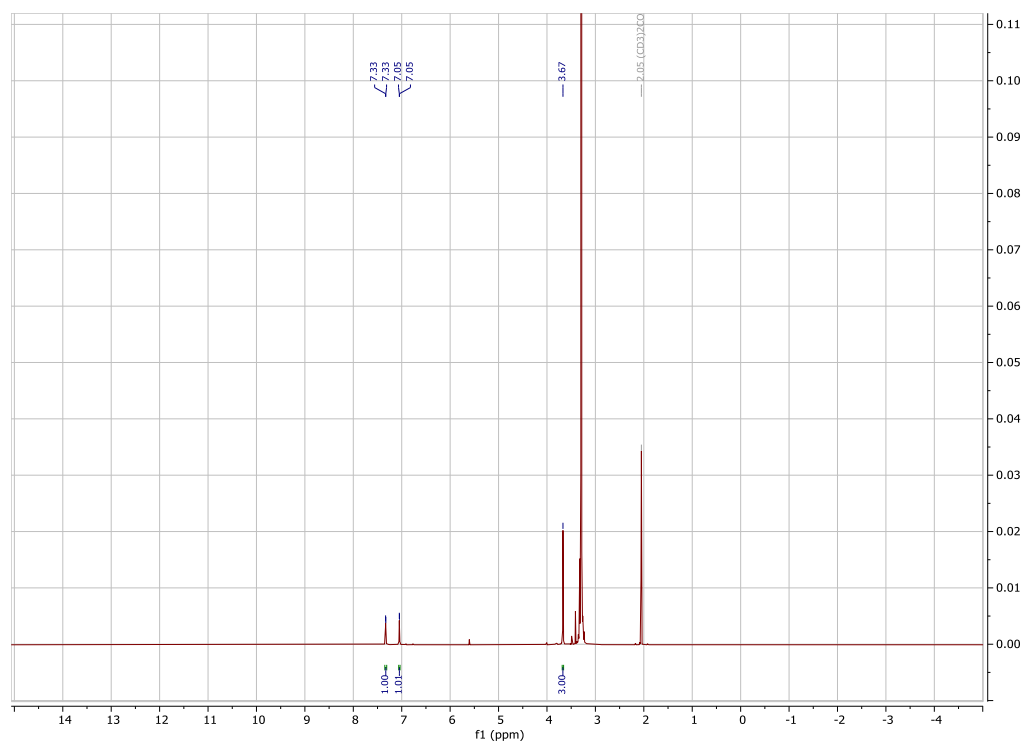


Figure S17. ¹H-NMR spectrum of [Sb(TMe)₂](CF₃COO) in d₆-acetone.

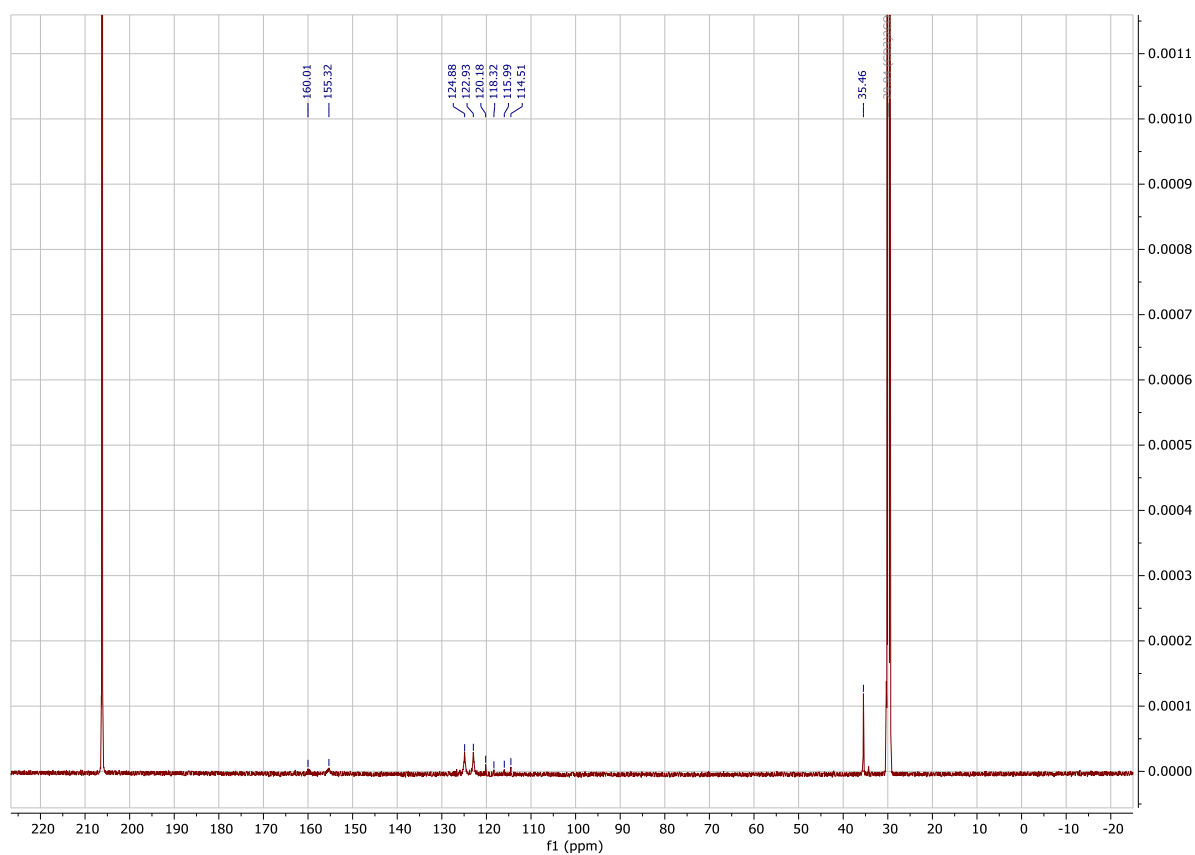


Figure S18. ¹³C-NMR spectrum of [Sb(TMe)₂](CF₃COO) in d₆-acetone.

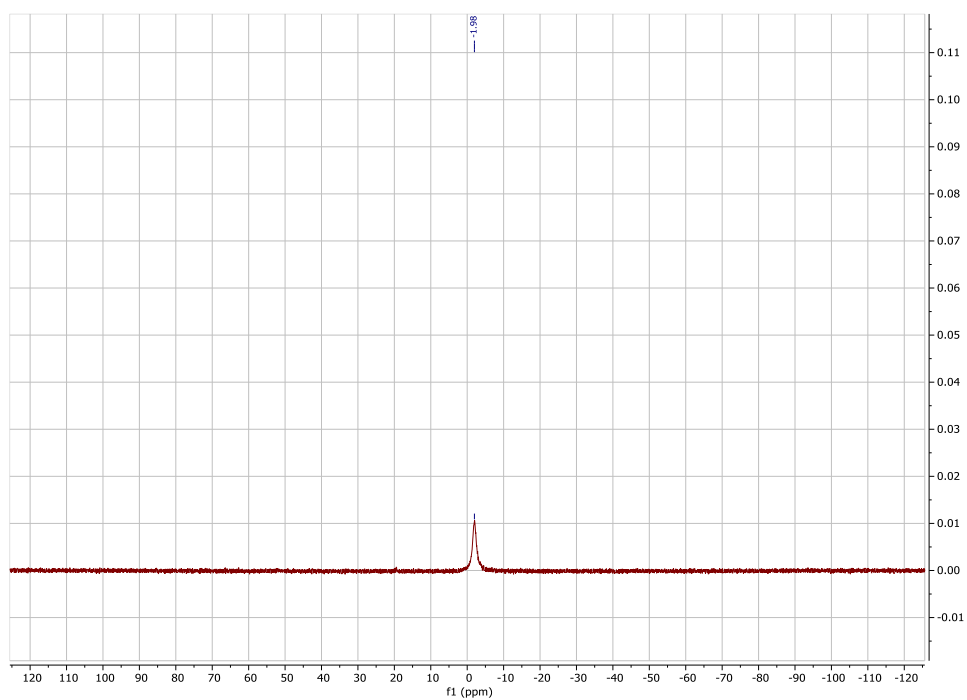


Figure S19. ^{11}B -NMR spectrum of $[\text{Sb}(\text{TMe})_2](\text{CF}_3\text{COO})$ in d_6 -acetone.

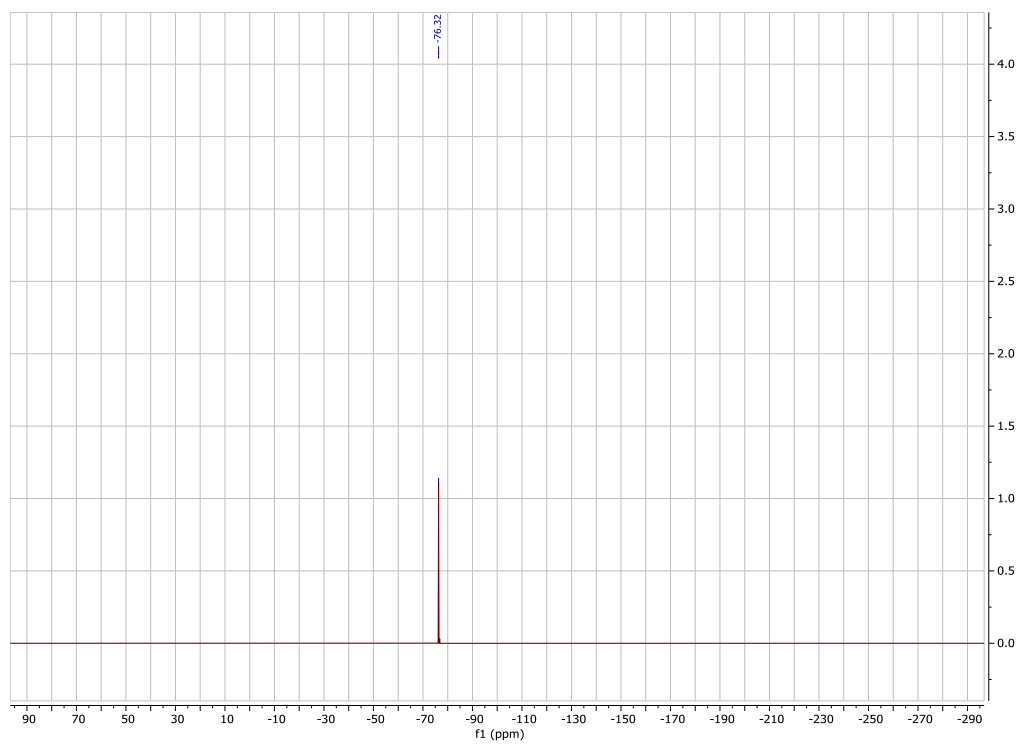


Figure S20. ^{19}F -NMR spectrum of $[\text{Sb}(\text{TMe})_2](\text{CF}_3\text{COO})$ in d_6 -acetone.

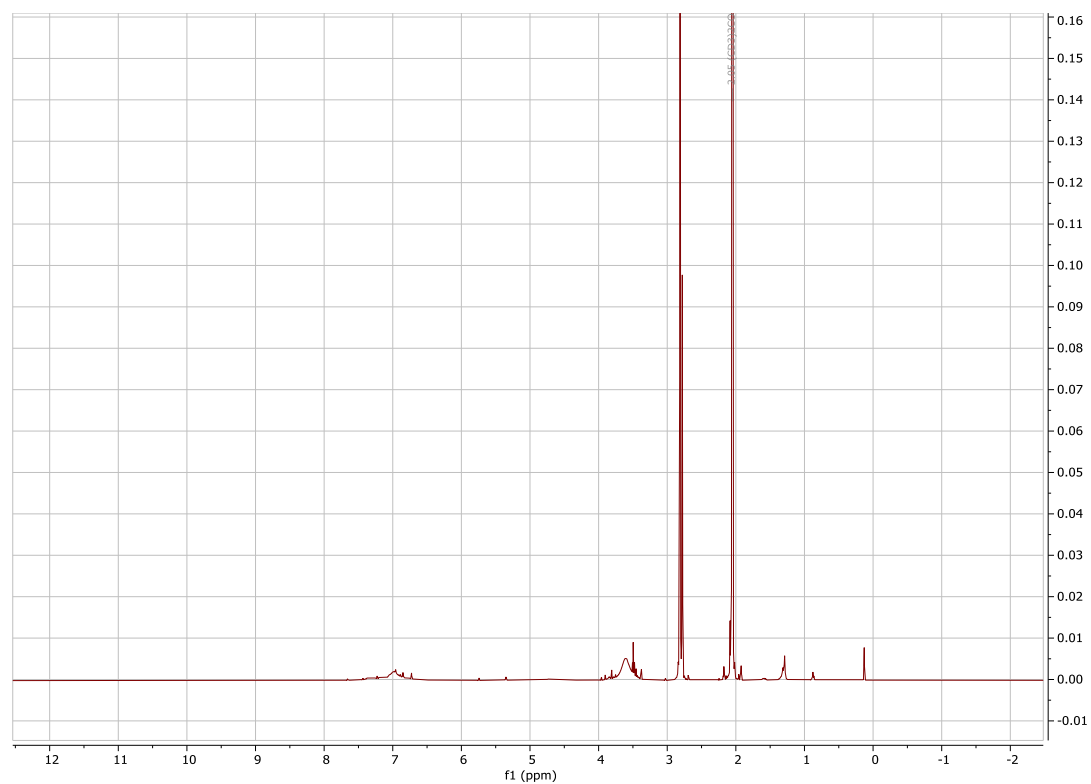


Figure S21. ^1H -NMR spectrum of $[\text{Sb}(\text{TMe})_2](\text{TMe})$ in d_6 -acetone at $25\text{ }^\circ\text{C}$.

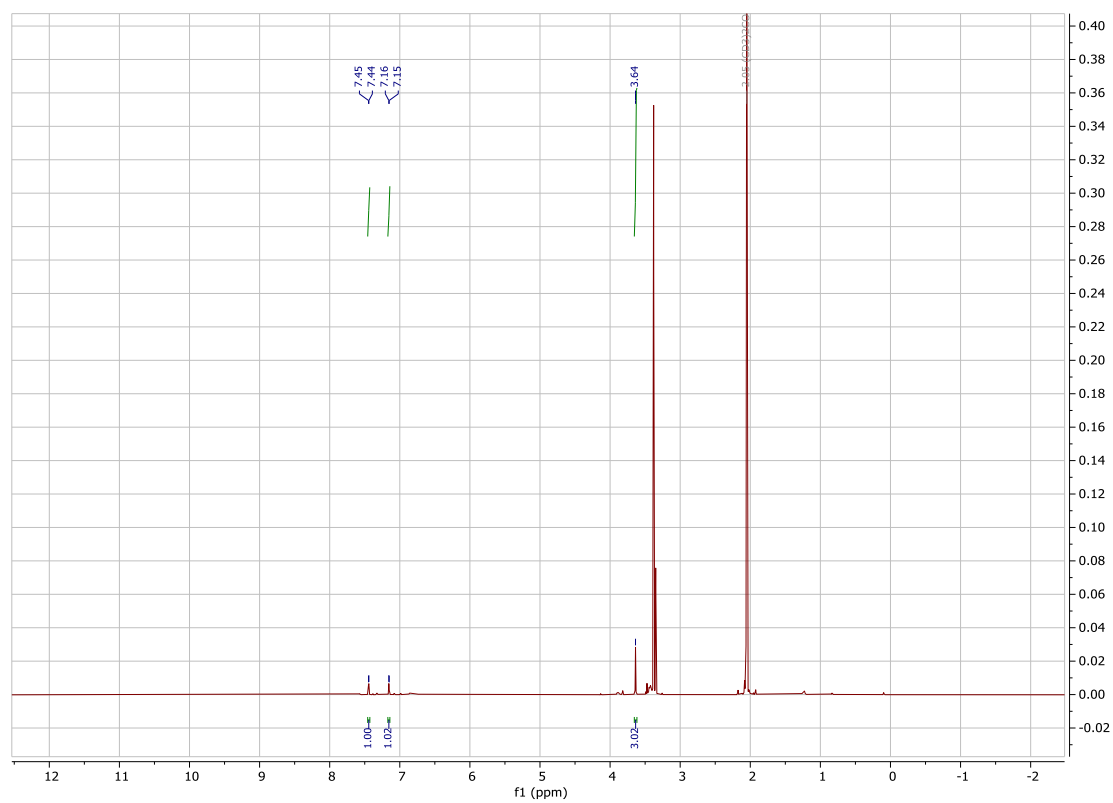


Figure S22. ^1H -NMR spectrum of $[\text{Sb}(\text{TMe})_2](\text{TMe})$ in d_6 -acetone at $-50\text{ }^\circ\text{C}$.

ESI mass spectra

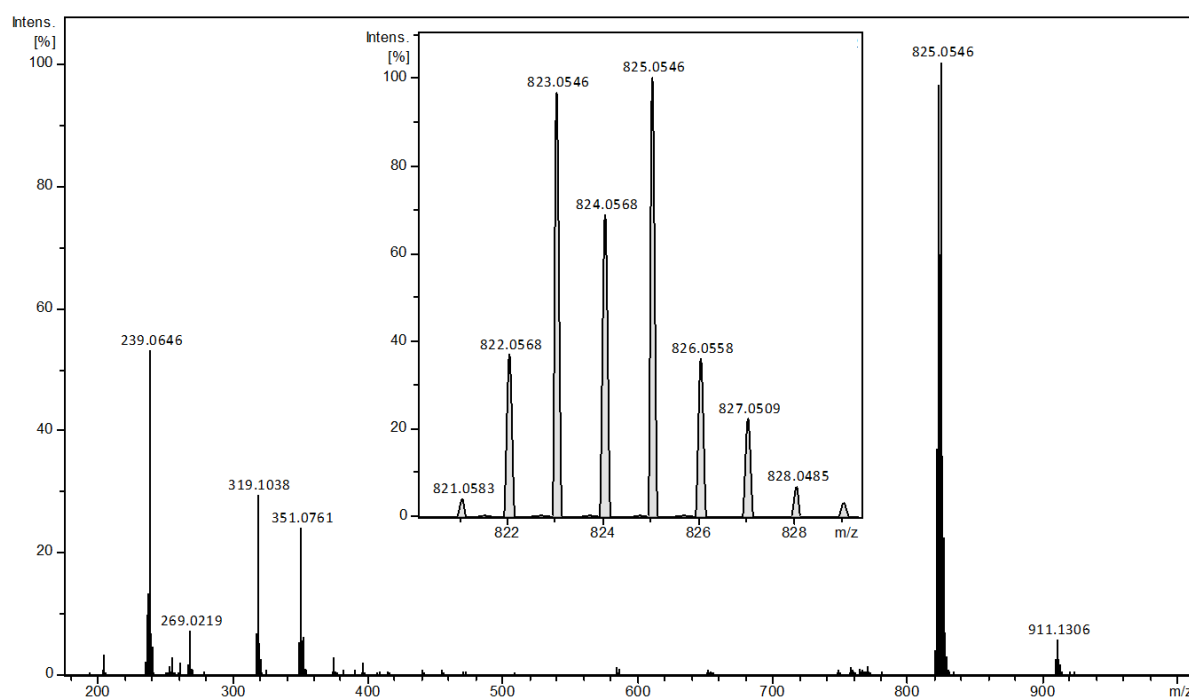


Figure S23. ESI mass spectra of $[Sb(TMe)_2](TMe)$ in acetonitrile recorded in positive mode showing the peaks corresponding to the ligand $[TMe]^+$ (m/z 351.07) and the complex $[Sb(TMe)_2]^+$ (m/z 825.05) (expanded in inset).

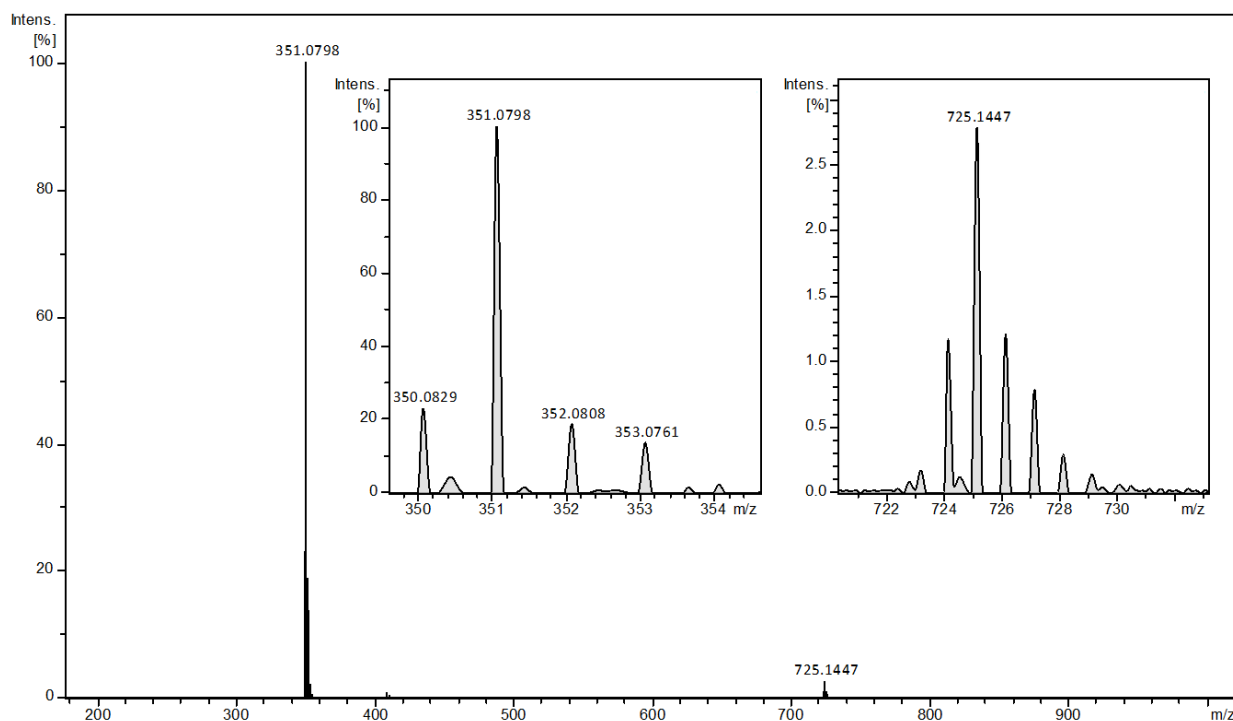


Figure S24. ESI mass spectra of $[Sb(TMe)_2](TMe)$ in acetonitrile recorded in negative mode showing the peaks corresponding to $[TMe]^-$ (m/z 351.07) (expanded in inset) and $[2(TMe)+Na]^-$ (m/z 725.14) (expanded in inset)

D. DFT

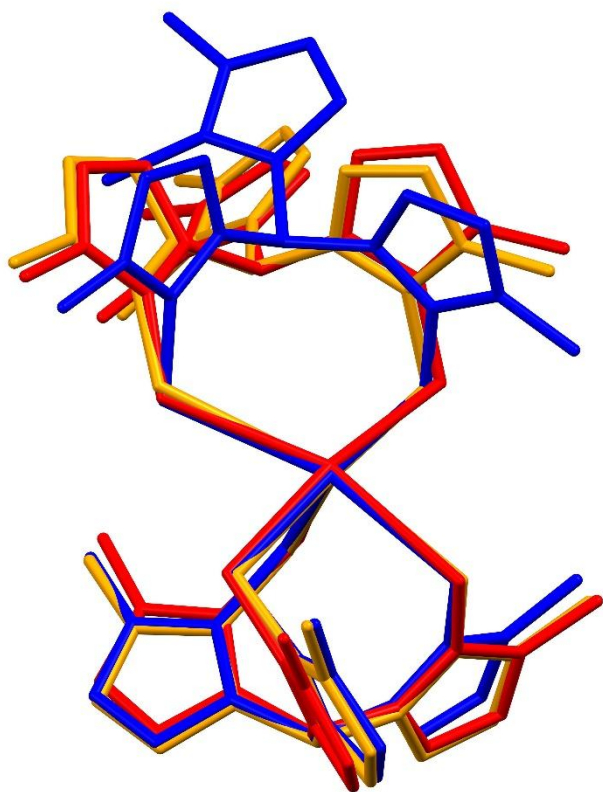


Figure S25. Overlaid structures of $[\text{Sb}(\kappa^3\text{-TMe})(\kappa^2\text{-TMe})]^+$. Crystal structure (blue), DFT gas phase optimised structure (red) and DFT solution phase optimised structure (orange).

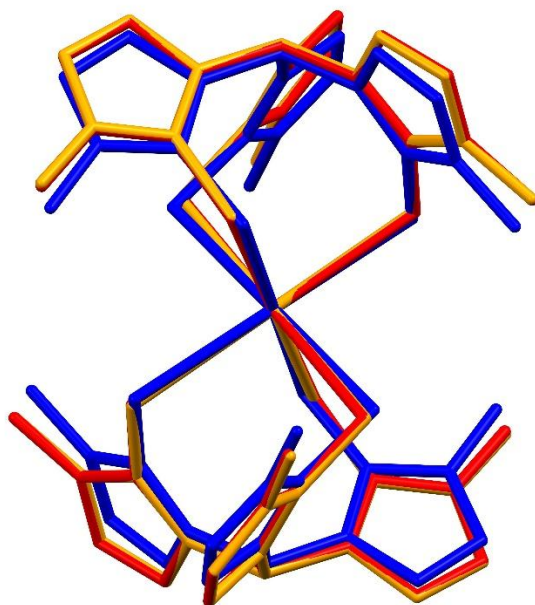


Figure S26. Overlaid structures of $[\text{Sb}(\kappa^3\text{-TMe})_2]^+$. Crystal structure (blue), DFT gas phase optimised structure (red) and DFT solution phase optimised structure (orange).

Table S9. Cartesian coordinates of the optimized structures of $[Sb(\kappa^3-TMe)(\kappa^2-TMe)]^+$

Gas phase			Solution phase (water)		
Gibbs free energy = -4269.02408396 Eh			Gibbs free energy = -4268.95181031 Eh		
Sb	11.78747874688100	10.51955379253116	Sb	11.87696880252036	10.44076715109836
S	10.85092044085619	8.38606919912946	S	10.97486790676986	8.39217615401922
S	14.29180426163680	9.34561834161537	S	14.31677965463584	9.3777633677384
S	11.44895466328085	9.40025798660145	S	11.33966195637927	9.14880208973831
N	12.78464969210794	6.50435313571971	N	12.95800562825202	6.53946461583247
N	12.65522340816899	6.94852229776698	N	12.91422872139281	7.16205704250254
N	13.78690666109309	7.23473372408439	N	13.81207136681754	7.14637604035734
N	15.33464632782961	8.63501699855756	N	15.23581005924952	8.57598397063202
N	11.23410671981837	6.75568511062712	N	11.31344703004784	6.55385415530566
N	9.38890101819625	7.62764359276877	N	9.38238716178279	7.29441135680995
C	12.14543083059738	7.26697187418858	C	12.3265386663565	7.3438727530579
C	13.70192918762884	5.70990043026850	C	13.95552992283636	5.85787537228271
H	14.33007426432124	5.01572019768665	H	14.59618165394434	5.15542716110928
C	13.62384707658245	5.97955684957779	C	13.92910012336933	6.2378127533597
H	14.15692729426607	5.56681913493653	H	14.52558271833663	5.93252673620638
C	12.22536232880352	7.51163330475520	C	12.53003097934748	7.80745138086948
H	12.1976763973663	8.59679652546906	H	12.51992198863568	8.88815437642403
H	12.93680694386045	7.2082072900602	H	13.26367657664565	7.53636499777620
H	11.23158522656570	7.15045369990942	H	11.54447613724319	7.47123539192692
C	14.44644028266385	8.37139167905442	C	14.43102770381361	8.32911522516178
C	14.28152233859782	6.79695265260020	C	14.24197734422184	6.65840735458212
H	13.91132977639474	5.90311285137374	H	13.88618912551517	5.72132536807325
C	15.23915985712498	7.65941774761340	C	15.12653887624180	7.53886002462010
H	15.85901258539131	7.66166656485872	H	15.68953974380016	7.51915399428055
C	16.26751767886066	9.75508508766413	C	16.12415217922563	9.72342674253455
H	16.68466812610922	9.82892732157952	H	16.51757426015541	9.69894232051198
H	17.07119987032554	9.59809371039162	H	16.94825793479694	9.67060532398291
H	15.74277943466755	10.67456829066298	H	15.57110625521192	10.64703171319160
C	10.68562756413566	7.88275332182519	C	10.68101853696411	7.62220927506756
C	10.25415563499483	5.78769462838417	C	10.38472175327608	5.54784308508270
H	10.45243963032766	4.79662303794482	H	10.65306901358567	4.59001799342039
C	9.11092570200624	6.32383495289029	C	9.18693687518456	6.00156850981734
H	8.13737873152529	5.89320786192009	H	8.22613566229158	5.52211202341820
C	8.43680096502309	8.55168999139966	C	8.34805756341350	8.14491603965591
H	8.65496373142422	9.57330484369277	H	8.33559171365779	9.10792496280839
H	7.43738205450081	8.28660485202383	H	7.39201293017479	7.64805573630421
H	8.49397531671523	8.47240904511818	H	8.53047135407538	8.28645988814480
B	12.67386524803784	6.42060712859031	B	12.78620400819687	6.32563963488497
H	12.6104636486075	5.2059044702242	H	13.02677493158117	5.17241521013636
S	8.98136736653836	10.81147309934298	S	8.87488736988837	10.97043296524991
S	12.95040552481306	12.36415712935308	S	12.75121892370273	12.26039875946123
S	7.35821857958782	11.96110386209694	S	7.59024704414510	12.11959874072423
N	8.54366917879692	13.30675721261024	N	8.33454955970436	13.30945604728357
N	7.15117913335038	12.74699157437093	N	6.91405747242058	12.83300828982575
N	10.89633062627861	14.14773042849078	N	10.76492552708889	13.97093896665613
N	12.89646182740950	14.83270329119769	N	12.78367471302599	14.49333545960572
C	8.25809967092919	12.34632192705977	C	8.05539068872259	12.41224156585234
C	7.58634237019838	14.29876264631902	C	7.35357187292523	14.28809115964542
H	7.57197247079865	15.14383279068992	H	7.36098431202522	15.10860135671316
C	6.72196763175789	13.95768209486194	C	6.47185536647037	13.99542384690943
H	5.84092017209897	14.45015448472424	H	5.57725450418986	14.49697169761368
C	6.47872672798353	11.98901785460530	C	6.23696505829399	12.15328263674880
H	7.21068082173488	11.61285962148255	H	6.91944871950375	12.00771290145540
H	5.93759934609306	11.14375871487371	H	5.86094154966223	11.18279693739096
H	5.77875096617498	12.64933389108701	H	5.40249267786159	12.77270707468541
C	12.20688485039024	13.80345085014086	C	12.06646607767520	13.60886719113495
C	10.77885078515915	15.40101109101387	C	10.67854239064568	15.09611817672817
H	9.82841436194930	15.88611655371247	H	9.74094454486633	15.57798524736982
C	12.01199442021679	15.83326053754918	C	11.92597225726012	15.42501266240085
H	12.33526417266574	16.74869737218943	H	12.27989928914410	16.22727003849591
C	14.33885465192018	14.88384221720950	C	14.22761496351298	14.47279487073921
H	14.84539215292266	15.05555051123592	H	14.74423833075349	14.57636100090668
H	14.68792415045117	13.94181430743674	H	14.53199807839624	13.54212302305315
H	14.56708587537199	15.69752912405866	H	14.48800652495275	15.30875064976475
C	8.29834796638653	13.32601403796021	C	8.43082353176007	13.59324982241000
N	9.24586240277530	13.92539875820776	N	9.24042124758702	10.04655951122908
C	9.76423189606692	15.01764577876824	C	9.71312119724790	15.30439809460090
H	10.53797715946490	15.62834604619873	H	10.37520023922144	15.86967833893068
C	9.14422945263644	15.13919498948989	C	9.20706848185818	15.62790988768674
H	9.26808314828021	15.86515558061449	H	9.33720422517027	16.50474866593361
N	8.24054707464263	14.09780167765709	N	8.41236268231698	14.56911459056218
C	7.3578053138376	13.82126695988730	C	7.67955697567200	15.0089007485665
H	7.5429356124415	12.83329408061659	H	8.04166014270375	13.67619291684067
H	7.54324848724630	14.57569837859769	H	7.82587449273885	15.43749567262389
H	6.31624180115021	13.88117169684475	H	6.61660007708234	14.35369197525959
B	9.70938287383776	13.28889926772075	B	9.55939158879828	13.27127273943586
H	10.07501694457463	12.16908241791277	H	9.84463648077727	12.14197818022765

Table S10. Cartesian coordinates of the optimized structures of $[Sb(\kappa^3-TMe)_2]^+$

Gas phase			Solution phase (water)				
Gibbs free energy = -4268.97092410 Eh			Gibbs free energy = -4269.02762266 Eh				
Sb	2.44223250227251	4.32344012048253	7.07549898305106	Sb	2.44065442821084	4.32295695847702	7.07524325320397
B	2.20490388074176	7.60481760364469	4.32038326912074	B	2.21193874861312	7.61369913267816	4.31962928305627
H	2.14064933336021	8.52683515396898	3.55068339409436	H	2.14797834860566	8.53736976741297	3.55348895872195
S	0.35424892717875	6.18018517510145	6.95985251736598	S	0.35779332700624	6.19607509522511	6.96004758104738
S	2.40678109850927	4.08280731030873	4.28876061739605	S	2.40218782871131	4.09077126146519	4.27636491415740
S	4.31297607766855	6.38998850451057	6.87391098914224	S	4.32470714979934	6.38643235458367	6.86444785094978
N	0.79191439120416	6.94387385514570	4.31112921804617	N	0.79848090615952	6.95571789495990	4.31171036616454
N	-1.11100882435031	5.93696414058389	4.69290865617666	N	-1.09410499330441	5.92534089506513	4.68294087337264
N	3.32058676135078	6.66716692167449	3.76331588402485	N	3.32142111706776	6.67414365546540	3.75508418960991
N	4.66673064394295	5.05646502686968	3.15154032043804	N	4.66328246014096	5.06406660980361	3.13131856887309
N	2.57212916351422	8.23062359788180	5.70206539497625	N	2.58637369569516	8.23444866749086	5.70066862279251
N	3.28782674972766	8.75959911076769	7.69940330368838	N	3.29935573683193	8.75700694808667	7.70100897260569
C	0.04214850035460	6.35648034159025	5.27719266566091	C	0.04848812137127	6.36248039461893	5.27206498005615
C	0.08735379267391	6.88271775579069	3.12261054910339	C	0.10616267664880	6.87660285951083	3.11628996484737
H	0.48545461461996	7.29415722347227	2.21307469382140	H	0.50720539918298	7.28426587848109	2.20543616211968
C	-1.08876432661595	6.25955455054137	3.35248623184504	C	-1.06495092597228	6.24100613996325	3.34110003872963
H	-1.90637341240278	6.02932069676092	2.69357498752585	H	-1.87746554892605	5.99478021693221	2.68081397264084
C	-2.19933958489638	5.25275410442842	5.37608121883467	C	-2.19656508215258	5.24942720196861	5.35398877470158
H	-2.98088537812258	5.04689390209109	4.64863136192151	H	-2.92495412009036	4.96625806452610	4.59881190683163
H	-2.59848247919094	5.87769592044444	6.17336026319299	H	-2.66748872667474	5.91095452254383	6.07961974155201
H	-1.84361144576390	4.31792947697561	5.80659679339982	H	-1.83680162223303	4.35761028346088	5.86331485061328
C	3.47471926829195	5.31913257886389	3.74901823177653	C	3.47469622094398	5.32723668286274	3.73317167578412
C	4.43145436275220	7.23365931180649	3.16550540461137	C	4.43531500036340	7.24277912802018	3.16334991833054
H	4.52823924775205	8.29861234715633	3.05782252497429	H	4.53811907173821	8.30888661698688	3.06528020226023
C	5.26771379691060	6.24285290535426	2.78701498907175	C	5.26724753721241	6.25150963239691	2.77488102354600
H	6.21935790449262	6.27381043843643	2.28792891906944	H	6.21847833377347	6.27834467387259	2.27356603561836
C	5.22158913418718	3.72944853232855	2.92741652468482	C	5.21310452950821	3.74131312585617	2.86499086866330
H	4.57018587433267	3.15047124249649	2.27440110334117	H	4.57514610001558	3.19372337536427	2.17294617381873
H	5.32861580954746	3.20319815503941	3.87456252962133	H	5.30457555807172	3.17675283609587	3.79073400636200
H	6.19689022498939	3.84352530901052	2.46024410984587	H	6.19734798978329	3.87049618581205	2.42218093208541
C	3.34605481809565	7.80634468805620	6.73240103473377	C	3.35864230742373	7.80915300407563	6.73023200109926
C	2.03561423083344	9.45855153753064	6.04429394502807	C	2.04244798915358	9.45811984079930	6.04925611703548
H	1.38949948266274	9.99474884427429	5.37342070970446	H	1.39093131711725	9.99496704098011	5.38287324864401
C	2.47285225514687	9.78918841561019	7.27874833421554	C	2.48056072847393	9.78550942572331	7.28480532221446
H	2.2888267540789	10.65840965270100	7.88395935822464	H	2.29742863075351	10.65032017262841	7.89743576809179
C	3.97471302767406	8.69988803326574	8.98161470520724	C	4.00152580210732	8.71828409162788	8.97719337730012
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H	2.74234600205459	0.11928419889223	10.59959775127157	H	2.73751215159604	0.10926096025350	10.59750719492945
S	4.52970233382958	2.46607585108644	7.19107961401672	S	4.52414977675089	2.45173493068841	7.18974448812733
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S	0.57130512981064	2.25742123845001	7.27628005818269	S	0.55725258151022	2.25739062792533	7.28710983967094
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C	4.84154477733881	2.28946477018868	8.87375453336194	C	4.83441877206631	2.28577140845628	8.87766144280770
C	4.79591791628584	1.76294052736589	11.02826059157480	C	4.77829777869932	1.77145145689422	11.03338873244627
H	4.39759271043729	1.35147512339083	11.93768674760369	H	4.37809418933199	1.36335018869287	11.94441477060459
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C	1.40894728235858	3.32729468145155	10.40173709610449	C	1.40849445686289	3.31842506291387	10.41751844336565
C	0.45170597967612	1.41281349683826	10.98461854149968	C	0.44941914909888	1.40244046625581	10.98837467721873
H	0.35468291572850	0.34785447154744	11.09203160986311	H	0.34737754340889	0.33629646874695	11.08684596202958
C	-0.38441396831239	2.40369860985358	11.36320284563172	C	-0.38300968588423	2.39328335865426	11.37688177367636
H	-1.33615092260975	2.37281476852825	11.86211561356560	H	-1.33400197639414	2.36597896134617	11.87862350496925
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C	2.84736668826790	-0.81209673977381	8.10594278963430	C	2.84298306439435	-0.81185251957563	8.10175822627953
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C	0.909217779705623	-0.05230508392073	5.16828994132090	C	0.88209973139377	-0.07432388857854	5.17446857951044
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