

Stepwise transformation of a redox-active tetradentate ONNO ligand in the coordination sphere of tin(IV)

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Content:

General information about physical measurements	2
SC XRD studies	4
NMR studies	6
UV-vis spectroscopy investigations	11
Electrochemical investigations	12
Magnetochemical studies	12
DFT investigations	13
References	35

General information about physical measurements

NMR spectra were recorded on Bruker Avance Neo 300 MHz and Bruker 400 MHz spectrometers.

Elemental analysis was performed with an Elementar Vario EL Cube elemental analyzer.

IR spectra were recorded on an FSM-1201 Fourier transform spectrometer in Nujol mulls (range 4000–400 cm⁻¹) in a KBr cell.

UV-vis-NIR spectra for solutions and solid samples as nujol mulls of complexes **5** and **8** were recorded in the range of 220–1400 nm on a Shimadzu UV-2600i spectrophotometer equipped with an ISR-2600Plus integrating sphere.

EPR spectra were registered on a Bruker MiniScope MS-5000 Magnettech spectrometer. Parameters of the recorded EPR spectra were determined by simulation of the spectra using the EasySpin program package.¹

The magnetic susceptibility of complexes **5-7** was measured with a Quantum Design MPMSXL SQUID magnetometer (San Diego, California, USA) in the temperature range of 2–300 K with a magnetic field of up to 5 kOe. Diamagnetic corrections were made using the Pascal constants. The effective magnetic moment was calculated as $\mu_{\text{eff}}(T) = [(3k/N_A\mu_B^2)\chi T]^{1/2} \approx (8\chi T)^{1/2}$.

Cyclic voltammetry curves for **5** and **8** were recorded using a Smartstat PS-50 potentiostat. Measurements were performed in 2 mM solution (in CH₂Cl₂) of studied complexes. Supporting electrolyte was 0.2 M tetra-*n*-butyl ammonium perchlorate (TBAP). Experiments were performed in a standard three-electrode cell under argon atmosphere. A glassy carbon disc electrode (2 mm diameter) was used as a working electrode. The auxiliary electrode is a platinum wire. The reference electrode was Ag/AgCl/KCl(sat) electrode, and potentials were corrected to Fc⁺/Fc redox pair as a standard ($E_{1/2} = 0.51$ V vs Ag/AgCl/KCl(sat)). All measurements were performed at room temperature.

Crystals suitable for SC XRD study were obtained directly from the reaction mixture for **1** (from a solution in diethyl ether) and **5** (from a solution in acetonitrile), and by recrystallization from a diethyl ether/acetonitrile mixture (in a volume ratio of 1:10) for **4** and from a solution in pentane for **8**.

The SC XRD data for **1**, **4**, **5**, and **8** were collected with *Rigaku OD Xcalibur E* diffractometer (Mo_{Kα}-radiation, ω -scans technique, $\lambda = 0.71073$ Å, $T = 100.0(2)$ K) using

CrysAlis^{Pro} software package.² The structures were solved *via* intrinsic phasing algorithm and refined by full-matrix least squares against F^2 using *SHELX*.^{3,4} Implemented in *CrysAlis^{Pro}* scaling algorithms were used to perform absorption corrections. All non-hydrogen atoms in **1**, **4**, **5**, and **8** were found from Fourier syntheses of electron density and refined anisotropically. Hydrogen atoms of secondary amines in **1** and **4** are found from Fourier syntheses of electron density and refined isotropically and freely. All other hydrogen atoms in all structures were placed in calculated positions and refined isotropically in the “riding” model with $U(H)_{\text{iso}} = 1.2U_{\text{eq}}$ of their parent atoms ($U(H)_{\text{iso}} = 1.5U_{\text{eq}}$ for methyl groups).

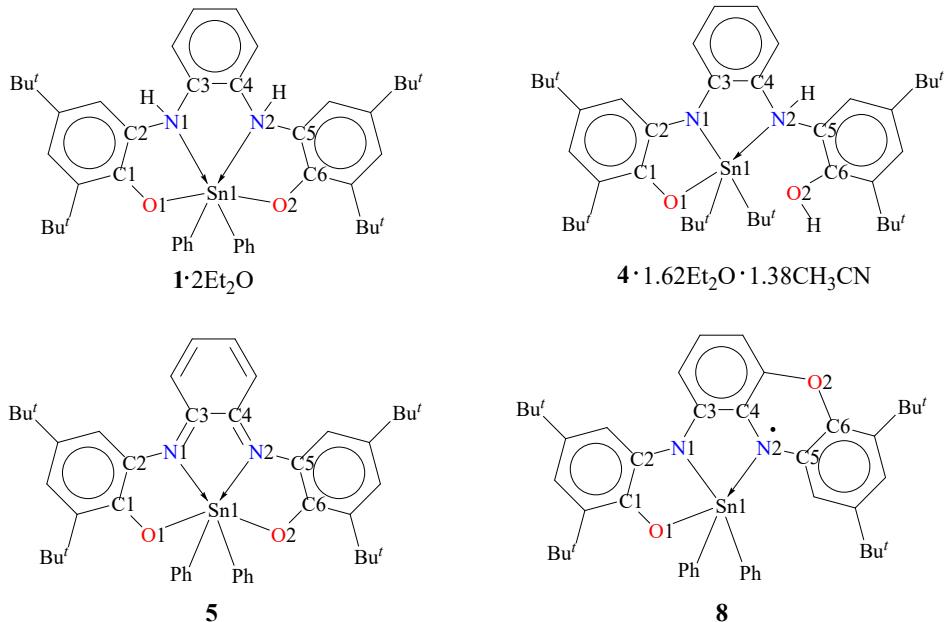
The crystallographic data and structures refinement details for all compounds are given in Table S1. CCDC 2390615 (**1**·2Et₂O), 2390616 (**4**·1.62Et₂O·1.38CH₃CN), 2390617 (**5**), and 2390618 (**8**) contain the supplementary crystallographic data for this paper. This data is provided free of charge by The Cambridge Crystallographic Data Centre: ccdc.cam.ac.uk/structures. The corresponding CIF are also available as ESI.

Density functional theory (DFT) calculations were performed using the Gaussian 09 software package.⁵ The stationary points on the potential energy surfaces were located by full geometry optimization with calculation of the force constant. The geometry optimization was carried out on the PBE0/Jorge DZP-DKH full electronic basis for all elements⁶ and was shown to give good results for tin(IV) complexes with redox-active ligands.⁷

SC XRD studies

Table S1. X-ray diffraction data collection and the structure refinement for complexes **1**·2Et₂O, **4**·1.62Et₂O·1.38CH₃CN, **5**, and **8**.

	1 ·2Et ₂ O	4 ·1.62Et ₂ O·1.38CH ₃ CN	5	8
Formula	C ₅₄ H ₇₆ N ₂ O ₄ Sn	C ₄₈ H ₇₇ N ₃ O ₃ Sn	C ₄₆ H ₅₄ N ₂ O ₂ Sn	C ₄₆ H ₅₃ N ₂ O ₂ Sn
Formula weight	935.85	862.81	785.60	784.59
Crystal size, mm	0.42×0.32×0.19	0.64×0.44×0.35	0.42×0.22×0.10	0.39×0.23×0.19
Crystal system	Orthorhombic	Triclinic	Triclinic	Orthorhombic
Space group	C222 ₁	P-1	P-1	Pbca
Unit cell dimentions				
<i>a</i> , Å	12.3983(2)	13.0130(2)	9.4959(3)	28.6053(3)
<i>b</i> , Å	15.2626(2)	13.5064(2)	12.4916(3)	9.73400(10)
<i>c</i> , Å	26.9230(4)	14.5664(2)	16.9054(5)	29.2307(2)
α , deg	90	72.5596(13)	80.714(2)	90
β , deg	90	84.7853(13)	85.784(2)	90
γ , deg	90	76.7192(13)	89.996(2)	90
<i>V</i> , Å ³	5094.66(13)	2376.49(7)	1973.53(10)	8139.11(13)
<i>Z</i>	4	2	2	8
<i>d</i> _{calc} , g/cm ³	1.220	1.206	1.322	1.281
μ , mm ⁻¹	0.545	0.578	0.687	0.666
<i>F</i> ₀₀₀	1984	920	820	3272
θ range, deg	2.25-29.57	1.84-30.03	1.89-26.02	2.32-30.03
HKL indices	-16 ≤ <i>h</i> ≤ 17 -19 ≤ <i>k</i> ≤ 21, -37 ≤ <i>l</i> ≤ 24	-18 ≤ <i>h</i> ≤ 18 -19 ≤ <i>k</i> ≤ 19, -20 ≤ <i>l</i> ≤ 20	-11 ≤ <i>h</i> ≤ 11 -15 ≤ <i>k</i> ≤ 15, -20 ≤ <i>l</i> ≤ 20	-40 ≤ <i>h</i> ≤ 40 -13 ≤ <i>k</i> ≤ 13, -41 ≤ <i>l</i> ≤ 41
Reflections collected	15290	54645	35932	340306
Independent reflections [$> 2\sigma(I)$]	6678 (<i>R</i> _{int} = 0.0281)	12178 (<i>R</i> _{int} = 0.0377)	6514 (<i>R</i> _{int} = 0.0635)	9636 (<i>R</i> _{int} = 0.0957)
Data/restraints/parameters	7146 / 24 / 331	13875 / 841 / 751	7781 / 96 / 536	11913 / 48 / 505
<i>R</i> ₁ ; w <i>R</i> ₂ [$> 2\sigma(I)$]	0.0282; 0.0558	0.0357; 0.0905	0.0354; 0.0682	0.0618; 0.1140
<i>R</i> ₁ ; w <i>R</i> ₂ (all data)	0.0328; 0.0575	0.0442; 0.0953	0.0509; 0.0735	0.0793; 0.1216
<i>S(F</i> ²)	1.042	1.070	1.075	1.019
$\Delta\rho_{\max}/\Delta\rho_{\min}$, e/Å ³	0.52 / -0.62	0.78 / -0.92	0.68 / -0.49	1.06 / -1.42



Scheme S1. Numbering of atoms in complexes **1·2Et₂O**, **4·1.62Et₂O·1.38CH₃CN**, **5**, and **8** to compare selected bond lengths.

Table S2. Comparison of selected bond lengths (Å) of complexes **1·2Et₂O**, **4**, **5**, and **8** according to the numbering given in Scheme S1.

Bond	1·2Et₂O	4·1.62Et₂O·1.38CH₃CN	5	8
Sn(1)-O(1)	2.091(2)	2.072(2)	2.261(2)	2.078(2)
Sn(1)-O(2)	2.091(2)	-	2.263(2)	-
Sn(1)-N(1)	2.351(2)	2.090(2)	2.288(2)	2.110(3)
Sn(1)-N(2)	2.351(2)	2.533(2)	2.196(2)	2.252(3)
O(1)-C(1)	1.349(3)	1.350(2)	1.285(3)	1.338(4)
N(1)-C(2)	1.471(3)	1.415(2)	1.368(3)	1.400(4)
O(2)-C(6)	1.349(3)	1.395(3)	1.304(3)	1.381(4)
N(2)-C(5)	1.471(3)	1.444(3)	1.379(3)	1.390(4)
N(1)-C(3)	1.456(3)	1.399(2)	1.352(3)	1.356(5)
N(2)-C(4)	1.456(3)	1.461(3)	1.347(3)	1.347(4)

NMR studies

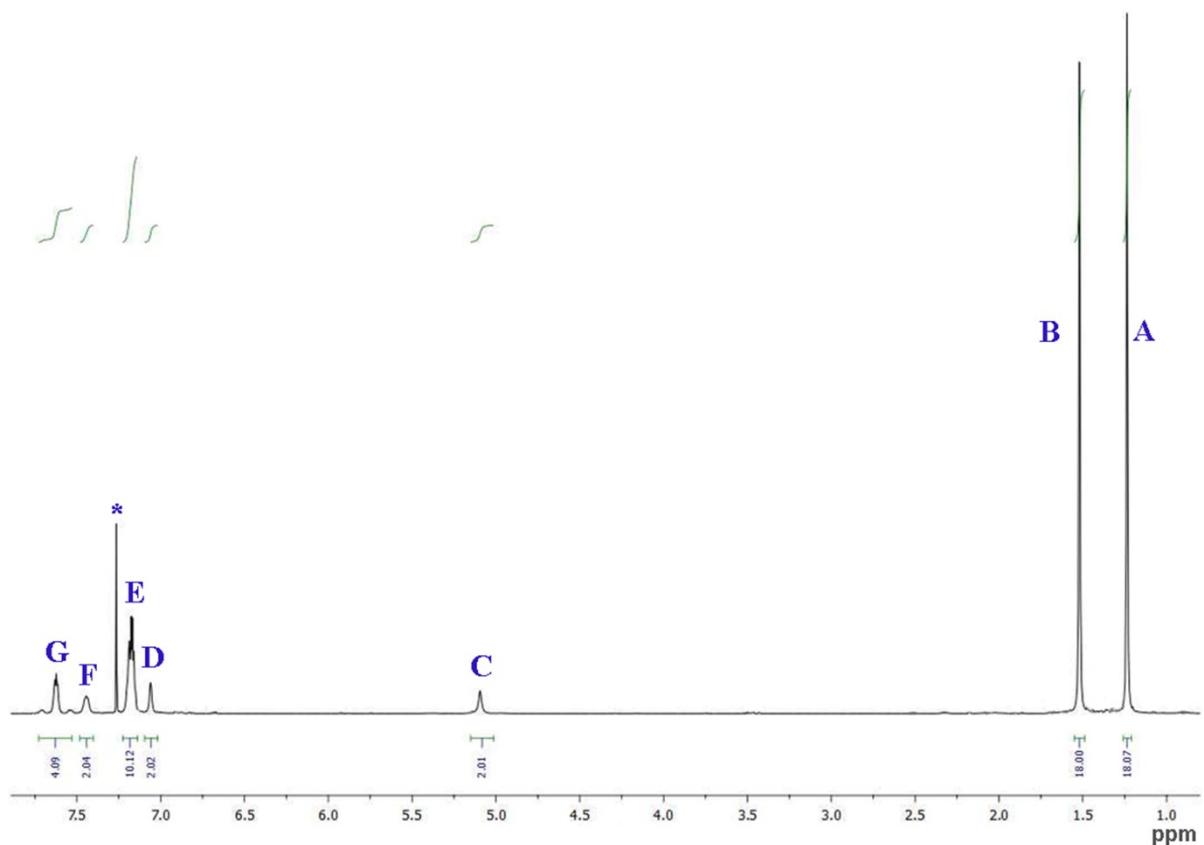


Figure S1. The ^1H NMR spectrum of complex **1** in CDCl_3 at 298 K; * - solvent residual signal; δ/ppm : A – 1.24 (s, 18H, Bu'), B – 1.52 (s, 18H, Bu'), C – 5.09 (br.s, 2H, NH), D – 7.06 (s, 2H, $\text{C-H}_{\text{O-Ph-N}}$), E – 7.17 (m, 6H + 2H + 2H, Sn-Ph + $\text{C-H}_{\text{N-Ph-N}}$ + $\text{C-H}_{\text{O-Ph-N}}$), F – 7.44 (m, 2H, $\text{C-H}_{\text{N-Ph-N}}$), G – 7.62 (m, 4H, $J_{\text{SnH}} = 32.8$ Hz, Sn-Ph).

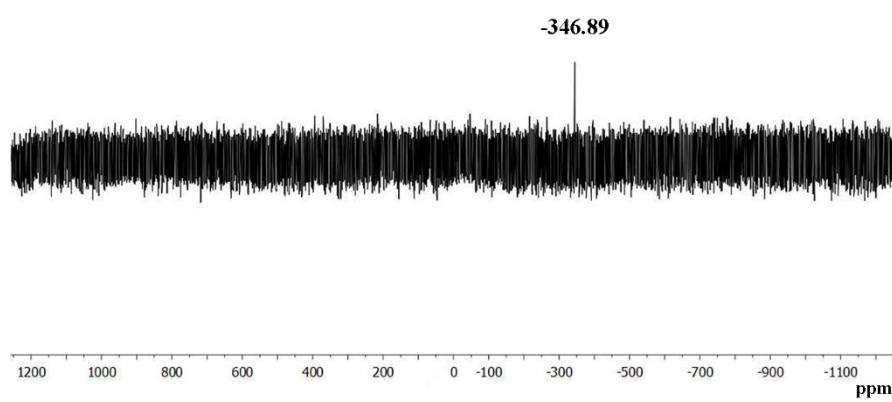


Figure S2. The ^{119}Sn NMR spectrum of complex **1** in CDCl_3 at 298 K; δ/ppm : -346.89.

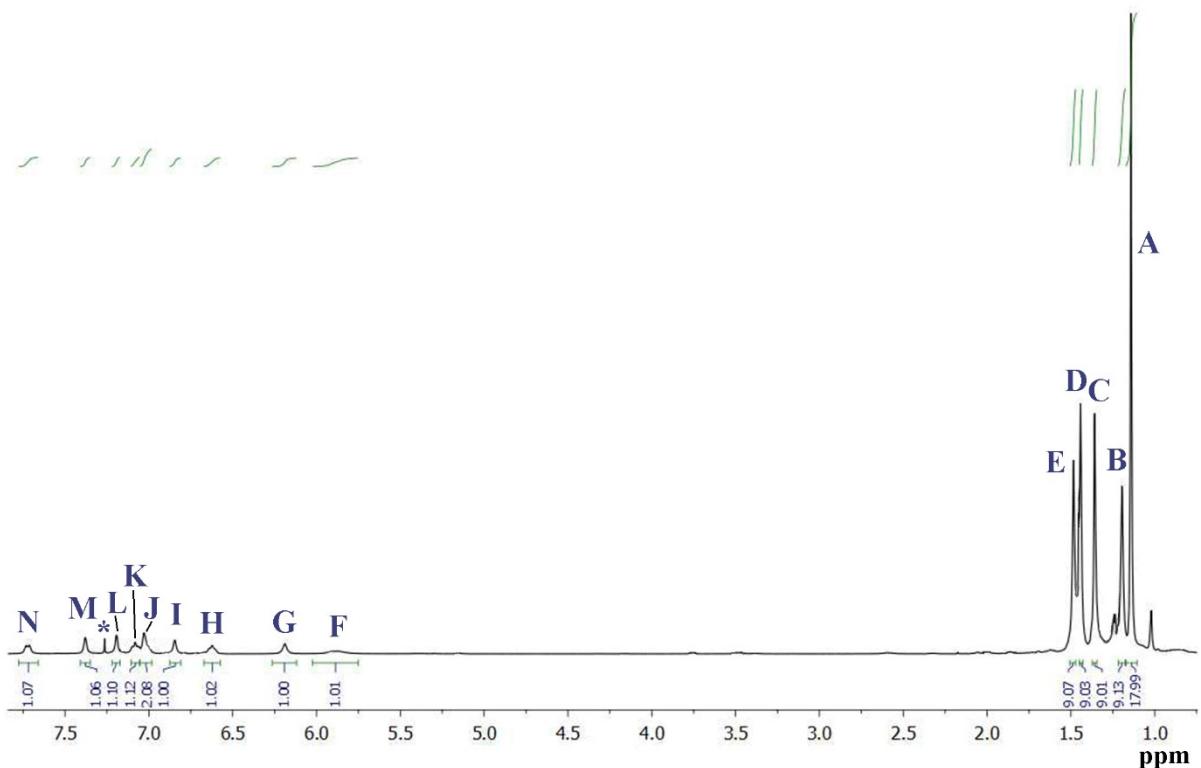


Figure S3. The ^1H NMR spectrum of complex **4** in CDCl_3 at 298 K; * - solvent residual signal; δ/ppm : A – 1.14 (s, 18H, $J_{\text{SnH}} = 46.98$ Hz, Sn-Bu^f), B – 1.20 (s, 9H, Bu^f), C - 1.36 (s, 9H, Bu^f), D - 1.44 (s, 9H, Bu^f), E - 1.48 (s, 9H, Bu^f), F - 5.86 (br.s, 1H, OH), G - 6.19 (br.s, 1H, NH), H - 6.62 (t, 1H, $J_{\text{HH}} = 6.9$ Hz, C- $\text{H}_{\text{N-Ph-N}}$), I - 6.85 (s, 1H, C- $\text{H}_{\text{O-Ph-N}}$), J - 7.03 (m, 1H+1H, C- $\text{H}_{\text{O-Ph-N}}$ + C- $\text{H}_{\text{N-Ph-N}}$), K - 7.08 (t, 1H, $J_{\text{HH}} = 7.6$ Hz, C- $\text{H}_{\text{N-Ph-N}}$), L - 7.19 (s, 1H, C- $\text{H}_{\text{O-Ph-N}}$), M - 7.38 (s, 1H, C- $\text{H}_{\text{O-Ph-N}}$), N - 7.72 (d, 1H, $J_{\text{HH}} = 7.7$ Hz, C- $\text{H}_{\text{N-Ph-N}}$).

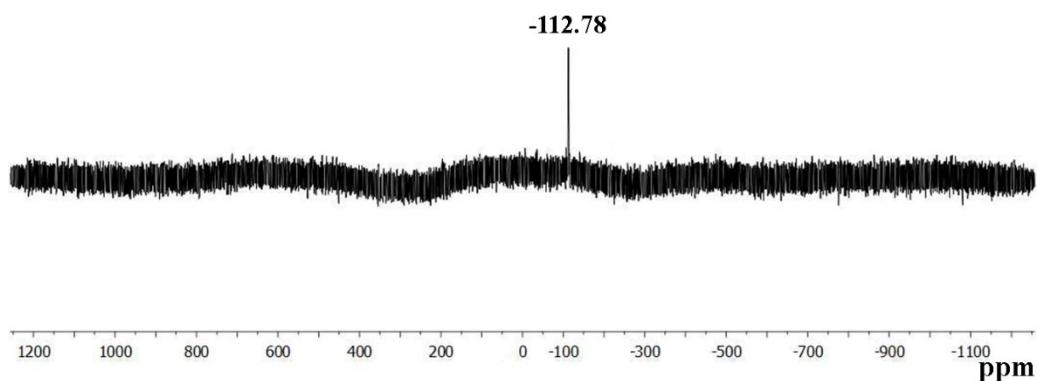


Figure S4. The ^{119}Sn NMR spectrum of complex **4** in CDCl_3 at 298 K; δ/ppm : -112.78.

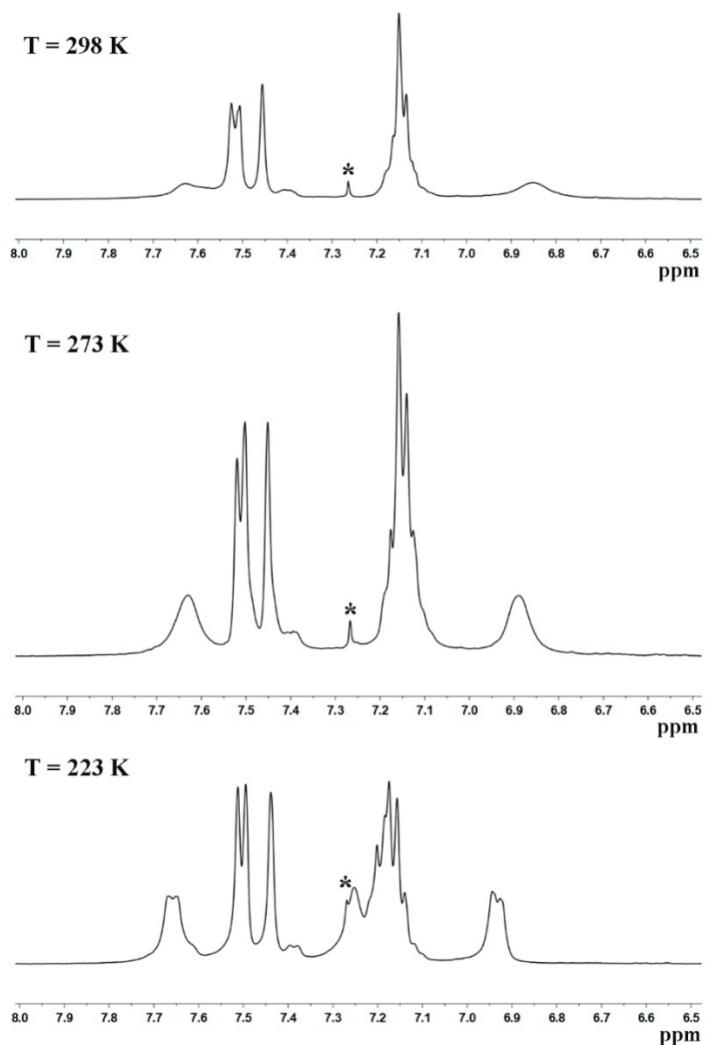


Figure S5. The weak-field region of the ^1H NMR spectrum of complex **5** at various temperatures in CDCl_3 , * - solvent residual signal.

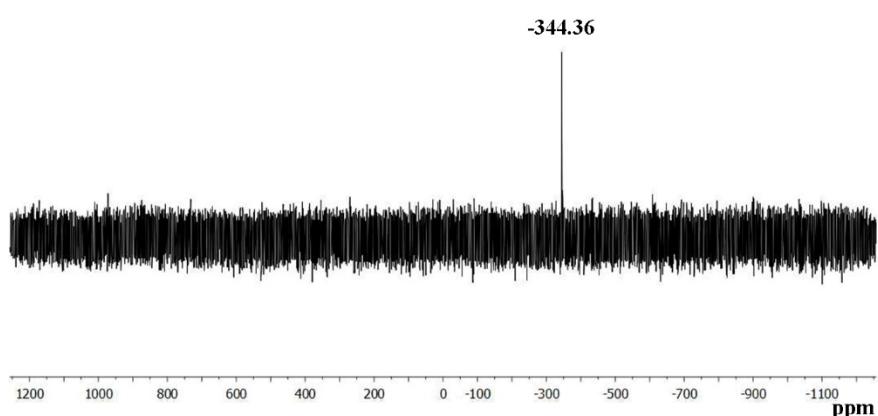
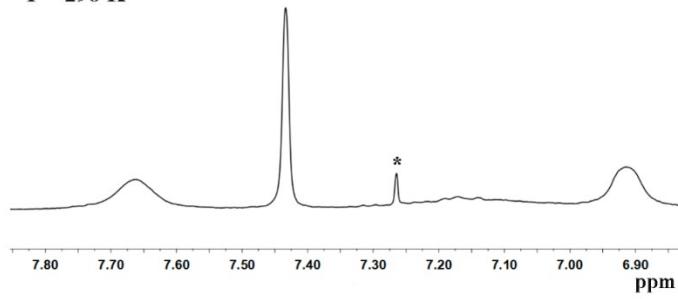
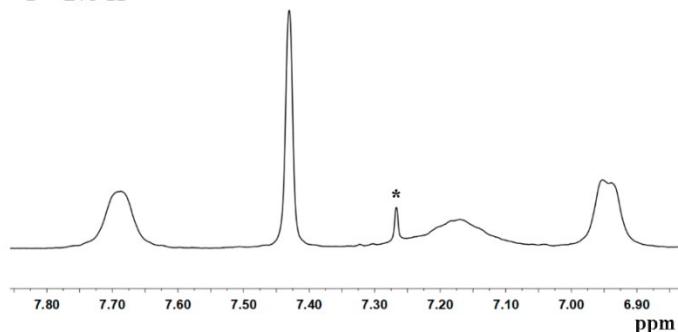


Figure S6. The ^{119}Sn NMR spectrum of complex **5** in CDCl_3 at 298 K ; δ/ppm : -344.36.

T = 298 K



T = 273 K



T = 223 K

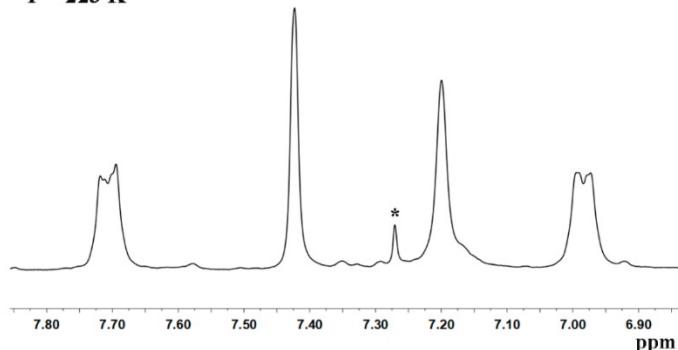


Figure S7. The weak-field region of the ¹H NMR spectrum of complex **6** at various temperatures in CDCl₃, * - solvent residual signal.

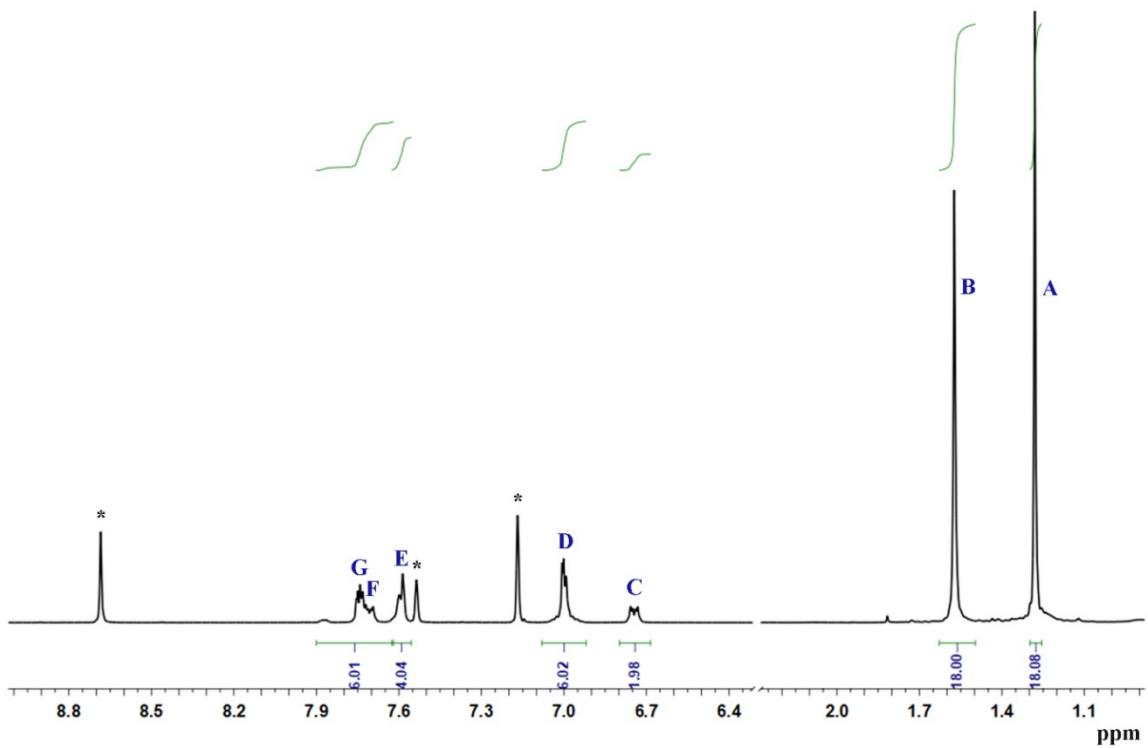


Figure S8. The ^1H NMR spectrum of complex **5** in Pyridine- d_5 at 298 K; * - solvent residual signal; δ/ppm : A – 1.28 (s, 18H, Bu'), B – 1.57 (s, 18H, Bu'), C – 6.74 (m, 2H, H_{phen}), D – 6.99 (m, 6H, $J_{\text{SnH}} = 13.5$ Hz, Sn-Ph), E – 7.59 (s + s, 2H + 2H, H_{arom}), F – 7.71 (m, 2H, H_{phen}), G – 7.74 (m, 4H, $J_{\text{SnH}} = 47.8$ Hz, Sn-Ph).

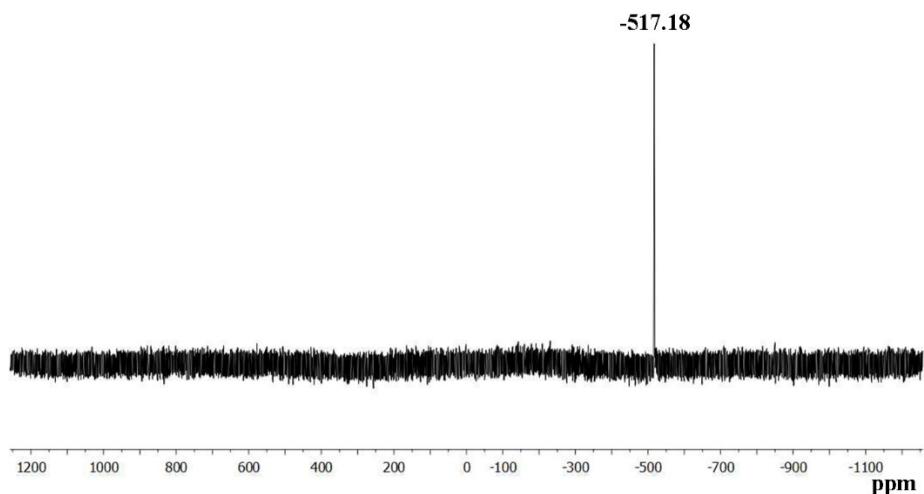


Figure S9. The ^{119}Sn NMR spectrum of complex **5** in Pyridine- d_5 at 298 K; δ/ppm : -517.18.

UV-vis spectroscopy investigations

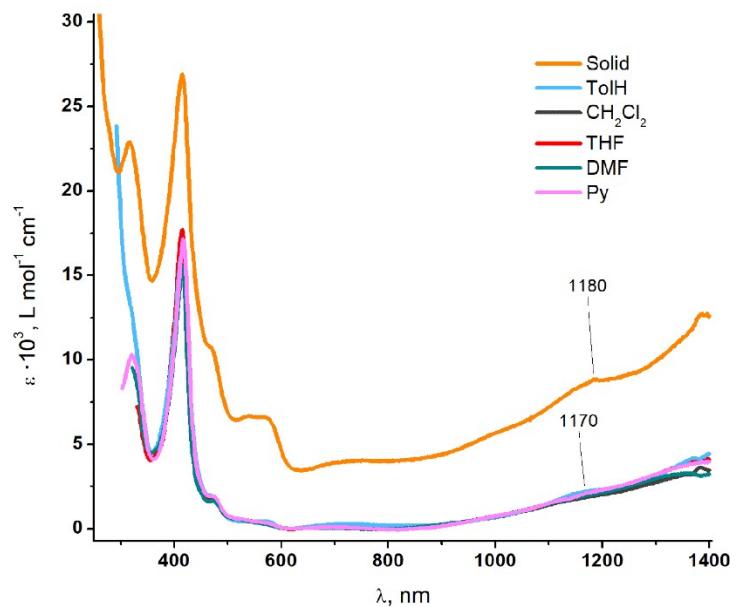


Figure S10. UV-vis spectra for complex 8 in different solvents and in solid.

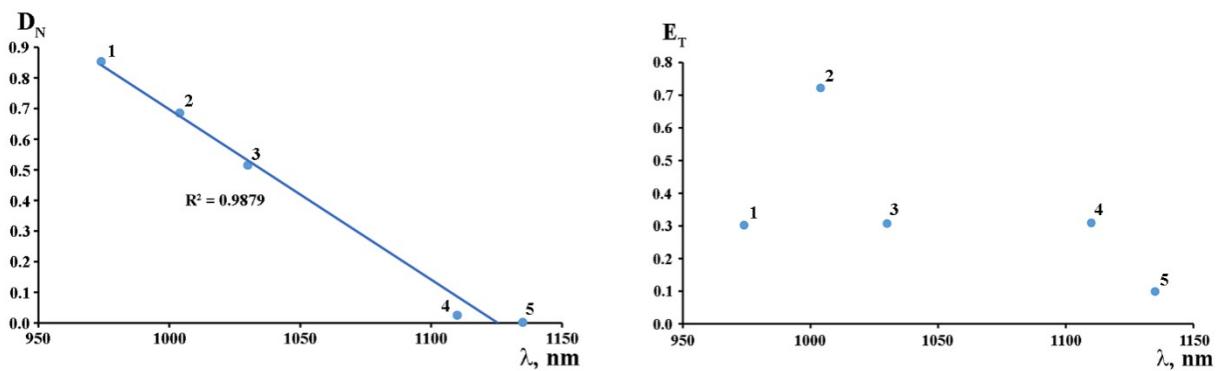


Figure S11. Dependences of the λ_{max} value for complex 5 on the solvent donor number D_N (on the left) and the empirical polarity parameter E_T (on the right). Solvents: 1 – Py, 2 – DMF, 3 – THF, 4 – CH₂Cl₂, 5 – Toluene.

Electrochemical investigations

Table S3. Redox potentials (V) of **5** and **8** in CH_2Cl_2 at 20 °C vs Fc^+/Fc , the energy of frontiers orbitals and HOMO-LUMO gap value (ΔE) calculated from the CV data

	$E_{1/2}(\text{Red-2})$	$E_{1/2}(\text{Red-1})$	$E_{1/2}(\text{Ox-1})$	$E_{1/2}(\text{Ox-2})$	$E(\text{HOMO})$	$E(\text{LUMO})$	ΔE
8		-0.81	-0.13		-3.99	-4.67	0.68
5	-1.37	-0.85	0.04	0.41	-3.95	-4.84	0.89

^afor reversible process $E = E_{1/2} = (E_{\text{red}} + E_{\text{ox}})/2$;

^c $E(\text{LUMO}) = -(4.8 + E(\text{ox1}))$, ^d $E(\text{HOMO}) = -(4.8 + E(\text{red1}))$ ⁸ [DOI: 10.1002/adma.201004554];

^e $\Delta E = E(\text{ox1}) - E(\text{red1})$.

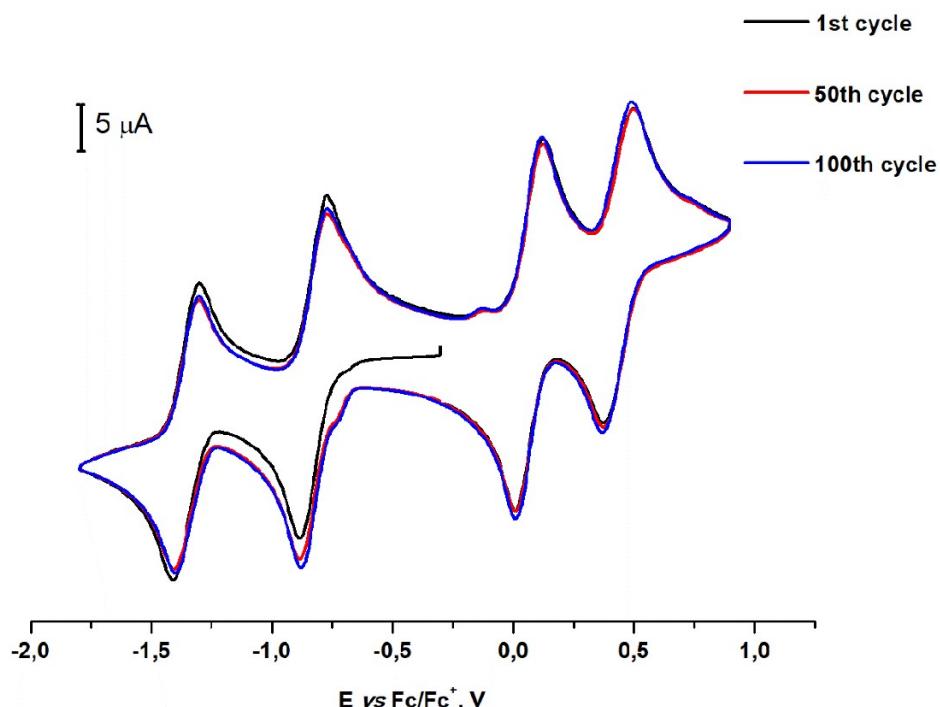


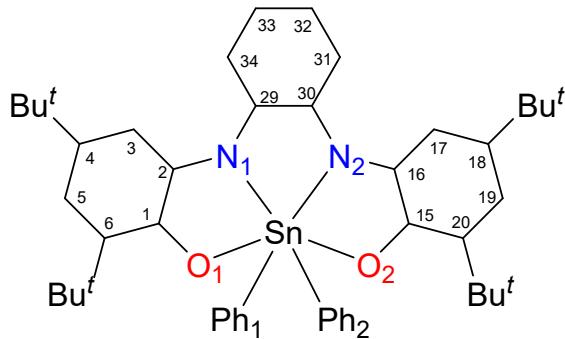
Figure S12. The 1st, 50th and 100th cycles of voltammogram of compound **5** in dichloromethane solution under argon atmosphere at 293 K.

Magnetochemical studies

Table S4. The best fit values for the analysis of the $\mu_{\text{eff}}(T)$ dependencies for the complexes **5-7** using the equation $\chi = C/(T-\theta) + \chi_0$ (where C is the Curie constant, θ – Weiss temperature, and χ_0 – temperature independent part of magnetic susceptibility), and the fraction of the high-spin paramagnetic state $\omega = C_{\text{exp}} / C_{\text{teor}} = C/0.75$.

Complex	$C, \text{K}\cdot\text{cm}^3/\text{mol}$	θ, K	$\chi_0, \text{cm}^3/\text{mol}$	$C_{\text{exp}} / C_{\text{teor}}, \%$
5	0.0065 ± 0.0003	-1.56 ± 0.04	$2.1 \cdot 10^{-5} \pm 0.2 \cdot 10^{-5}$	0.87
6	0.0189 ± 0.0015	-4.5 ± 0.6	$4.4 \cdot 10^{-5} \pm 0.6 \cdot 10^{-5}$	2.5
7	0.0224 ± 0.0007	-1.00 ± 0.01	$2.5 \cdot 10^{-4} \pm 0.7 \cdot 10^{-4}$	3.0

DFT investigations



Scheme S2. Numbering of atoms in complex **5** to compare the values of selected bond lengths and angles obtained from X-ray diffraction experiment and DFT calculations.

Table S5. The values of selected bond lengths (\AA) and angles ($^\circ$) for complex **5** obtained from X-ray diffraction experiment and DFT calculations.

Bond	X-ray	5(OSS)	5(CSS)	5(T)	5⁺	5⁻	5·Py(OSS)	5·Py(CSS)	5·Py(T)
C1-01*	1.295	1.277	1.284	1.274	1.346	1.300	1.285	1.291	1.280
C2-N1*	1.374	1.356	1.367	1.353	1.348	1.382	1.356	1.368	1.348
C1-C2*	1.442	1.462	1.452	1.466	1.481	1.442	1.456	1.447	1.464
C2-C3*	1.404	1.416	1.416	1.415	1.414	1.407	1.418	1.416	1.419
C3-C4*	1.372	1.379	1.377	1.380	1.378	1.390	1.378	1.379	1.377
C4-C5*	1.413	1.428	1.424	1.428	1.440	1.409	1.424	1.419	1.427
C5-C6*	1.373	1.377	1.381	1.377	1.369	1.392	1.381	1.383	1.379
C6-C1*	1.434	1.443	1.440	1.444	1.453	1.429	1.440	1.438	1.442
C29-N1*	1.355	1.372	1.338	1.386	1.360	1.352	1.356	1.327	1.381
C29-C30	1.455	1.440	1.465	1.430	1.446	1.437	1.451	1.474	1.433
C30-C31*	1.422	1.411	1.429	1.403	1.418	1.418	1.419	1.436	1.406
C31-C32*	1.358	1.386	1.369	1.392	1.376	1.383	1.380	1.365	1.392
C32-C33	1.416	1.402	1.422	1.396	1.413	1.407	1.409	1.427	1.396
Sn-Ph*	2.134	2.060	2.065	2.059	2.047	2.080	2.085	2.091	2.077
Sn-O*	2.272	2.226	2.273	2.355	2.346	2.282	2.298	2.262	2.340
Sn-N1,2*	2.243	2.184	2.219	2.169	2.212	2.175	2.282	2.319	2.244
Sn-N3	-	-	-	-	-	-	2.552	2.494	2.641
Angle									
O1-Sn-O2	147.4	143.7	143.2	144.2	149.9	141.2	148.7	149.2	148.6
N1-Sn-N2	71.3	74.1	73.4	74.2	73.6	74.1	71.2	70.4	71.8
Ph-Sn-Ph	146.6	148	154.2	145.5	149.5	150.4	171.1	174.6	167.2

* Values are an average of two fragments of the molecule.

Atomic coordinates in the optimized structures

5_CSS

Charge = 0, S = 0, E = -8.008172744586289E+03

Sn	-0.008854017	-0.807139146	0.394866957
O	2.158023250	-1.512552427	0.298082280
O	-2.155375808	-1.516594828	0.253779361
N	1.320413184	0.967181444	0.513974013
N	-1.332011786	0.966160968	0.527928154
C	3.020690436	-0.601614201	0.034744519
C	2.624330490	0.792133282	0.144255147
C	3.530319443	1.824832214	-0.198541526
C	4.820779433	1.533380277	-0.585310086
C	5.215450578	0.165284473	-0.612651515
C	4.380194510	-0.898062709	-0.337112128
C	5.825204271	2.606871832	-1.005433445
C	5.226626140	4.013045323	-0.944873850
C	6.288044601	2.342153501	-2.447286013
C	7.040482843	2.558936586	-0.064675257
C	4.850401230	-2.351139714	-0.428280131
C	6.327928161	-2.451301754	-0.815716913
C	4.028286723	-3.088739005	-1.498538141
C	4.672284344	-3.041173136	0.934809258
C	-3.027287958	-0.600443121	0.021563071
C	-2.642290300	0.789742327	0.176929118
C	-3.562086477	1.824457753	-0.117878643
C	-4.852339707	1.535833284	-0.509437946
C	-5.228127342	0.167158463	-0.602754461
C	-4.377845968	-0.896920707	-0.374416450
C	-5.870443740	2.616464382	-0.875618037
C	-5.297492367	4.026286904	-0.722336613
C	-6.311932297	2.431928374	-2.336563029
C	-7.095446631	2.492656173	0.045169978
C	-4.825275958	-2.348855404	-0.558077158
C	-3.985066468	-3.001396490	-1.668822024
C	-4.646215083	-3.126622767	0.756399319
C	-6.298943430	-2.447902076	-0.960789893
C	0.728597055	2.067004901	0.993141266
C	-0.736974942	2.060730325	1.013551738
C	-1.408467876	3.184878017	1.587252971
C	-0.708017024	4.246858782	2.092236776
C	0.714132111	4.255625979	2.066404719
C	1.407952712	3.199495332	1.540223491
C	0.009354940	-1.171963657	-1.635663318
C	-0.021174568	-0.100115775	-2.534397230
C	0.043741746	-2.474367860	-2.151747182
C	-0.016485382	-0.320032672	-3.914133517
C	0.048115468	-2.700145080	-3.527333709

C	0.017858520	-1.620705006	-4.412087123
C	0.005792567	-1.363679462	2.385518243
C	1.171179649	-1.199775038	3.148571907
C	-1.135397281	-1.873574773	3.020099165
C	1.190847104	-1.513014039	4.507282109
C	-1.117049722	-2.195677175	4.378138624
C	0.045376247	-2.014536844	5.125242553
H	3.169754862	2.853237958	-0.224228954
H	6.247181058	-0.056282913	-0.897943411
H	4.361363113	4.122629496	-1.621980503
H	5.983555165	4.754580090	-1.253850640
H	4.901796856	4.280105089	0.076364503
H	5.434903664	2.369448543	-3.147986595
H	6.773958541	1.356864427	-2.549658966
H	7.018455258	3.108731073	-2.764336217
H	7.543434861	1.577120338	-0.090647941
H	6.740140834	2.755188652	0.979914898
H	7.784493884	3.322158751	-0.357341325
H	6.532013252	-2.008656939	-1.806918607
H	6.618372016	-3.514882872	-0.866489433
H	6.988068048	-1.965168798	-0.075195463
H	4.181169570	-2.633658689	-2.493691005
H	2.953393716	-3.056183571	-1.266249623
H	4.345427480	-4.146229748	-1.555816540
H	5.016973671	-4.089651657	0.874358611
H	3.617393041	-3.039453421	1.248817471
H	5.271709285	-2.533500258	1.712396236
H	-3.214506371	2.857212174	-0.103764724
H	-6.255066063	-0.055470753	-0.904126376
H	-4.985594032	4.233972564	0.316585670
H	-6.065314373	4.772280788	-0.991344640
H	-4.429590825	4.193608146	-1.384037451
H	-5.451951533	2.522657014	-3.023514877
H	-7.056672915	3.199947721	-2.614709778
H	-6.772727304	1.443663487	-2.505245622
H	-6.810384721	2.629077718	1.103500863
H	-7.583193112	1.507160661	-0.046615821
H	-7.847802427	3.261209126	-0.209736048
H	-2.910785761	-2.955245953	-1.435174653
H	-4.150219591	-2.491527909	-2.634972626
H	-4.274979721	-4.061378896	-1.790196685
H	-5.236006823	-2.666484702	1.569884321
H	-3.589210833	-3.153901357	1.060376103
H	-4.997668830	-4.166983477	0.630623512
H	-6.971461990	-2.023210001	-0.194087158
H	-6.570058556	-3.510605352	-1.084107972
H	-6.505202222	-1.944145443	-1.921757340
H	-2.492613939	3.144614704	1.700487859

H	-1.245804886	5.072904087	2.569495307
H	1.259028597	5.088675881	2.522910935
H	2.496164665	3.172020778	1.612977452
H	-0.048903382	0.932149127	-2.164427597
H	0.069141319	-3.337275255	-1.475814418
H	-0.040007084	0.533283379	-4.601377889
H	0.075754606	-3.726448767	-3.910533847
H	0.021407482	-1.795724455	-5.494046525
H	2.085309385	-0.827211357	2.676336787
H	-2.051564763	-2.018160126	2.441625467
H	2.112167994	-1.371377033	5.083755448
H	-2.021098383	-2.595289402	4.851936501
H	0.060904336	-2.267654229	6.191671370

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O	-2.207494423	-1.526479144	0.176582087
N	1.307073441	0.895486198	0.632827864
N	-1.324255355	0.897209294	0.647017095
C	3.037492232	-0.584389815	-0.005738407
C	2.588840609	0.792388965	0.201561077
C	3.454275122	1.871576113	-0.100062745
C	4.738616077	1.642661946	-0.546040308
C	5.182610637	0.291718467	-0.676617588
C	4.396204873	-0.812876766	-0.436936287
C	5.693546824	2.773367314	-0.928289652
C	5.070905643	4.154467133	-0.718415716
C	6.073550118	2.635427252	-2.411890881
C	6.963181001	2.681457164	-0.065921769
C	4.907889158	-2.241569686	-0.626160627
C	6.375708527	-2.268784203	-1.059165012
C	4.077209461	-2.942742230	-1.714228015
C	4.791534426	-3.013563641	0.699371590
C	-3.045452423	-0.580682409	-0.017496208
C	-2.611523231	0.791265029	0.236424632
C	-3.494794557	1.869937362	-0.014540133
C	-4.780007734	1.643095289	-0.457304672
C	-5.202793595	0.293598672	-0.648056156
C	-4.397599578	-0.809069822	-0.462946683
C	-5.754654043	2.775989351	-0.778615968
C	-5.150336006	4.154869546	-0.508488133
C	-6.147280172	2.703504207	-2.263415345
C	-7.014246976	2.625092479	0.090024352
C	-4.886070095	-2.232126179	-0.740248136
C	-4.040598291	-2.848088872	-1.867821489
C	-4.763078454	-3.087667236	0.531643908

C	-6.352241527	-2.257280270	-1.179874550
C	0.714091068	1.988620717	1.212760259
C	-0.725587941	1.983539065	1.232264796
C	-1.398614374	3.018644891	1.914465733
C	-0.696532073	4.039116854	2.535828331
C	0.705675261	4.046435375	2.512795309
C	1.398136844	3.030683313	1.872737689
C	0.007248058	-1.126551898	-1.635255306
C	-0.003403101	-0.025376570	-2.498430946
C	0.023990920	-2.411708691	-2.193951501
C	0.003018441	-0.199847812	-3.884399259
C	0.031063547	-2.592680208	-3.576250531
C	0.020133003	-1.484233199	-4.424700602
C	0.013782785	-1.657640451	2.290130747
C	1.138839246	-1.471058273	3.106331508
C	-1.069697269	-2.379852797	2.807682824
C	1.175322340	-1.974510528	4.405973750
C	-1.033356075	-2.893111040	4.105276525
C	0.087990724	-2.689873262	4.907643127
H	3.067030186	2.888311725	-0.030426937
H	6.210587075	0.128128046	-1.009563403
H	4.168978740	4.299387943	-1.338231625
H	5.795904521	4.936368794	-1.003421290
H	4.796203616	4.327356802	0.337022165
H	5.179543542	2.706033949	-3.056395946
H	6.565263628	1.670876984	-2.624493561
H	6.772812766	3.439500985	-2.705200617
H	7.491380542	1.722731750	-0.204339531
H	6.721292069	2.780535803	1.007179883
H	7.666867201	3.490095506	-0.334604704
H	6.535076342	-1.765107264	-2.029107925
H	6.699716454	-3.317197862	-1.177106810
H	7.040232595	-1.801086465	-0.310825506
H	4.181705564	-2.423243824	-2.683748893
H	3.009717554	-2.965199089	-1.448696904
H	4.429531176	-3.982261960	-1.845955007
H	5.167534838	-4.044804081	0.568407434
H	3.747060062	-3.065097117	1.042661646
H	5.397615719	-2.531766166	1.488149270
H	-3.119950927	2.887896956	0.089918363
H	-6.228420719	0.128989173	-0.987415436
H	-4.867661362	4.280694984	0.551482456
H	-5.889743099	4.938419510	-0.748529638
H	-4.257019266	4.343860586	-1.128979896
H	-5.260745082	2.817585725	-2.912034180
H	-6.860399327	3.510667218	-2.511466376
H	-6.628108559	1.743545112	-2.517814624
H	-6.763230425	2.677159028	1.164318998

H	-7.530033447	1.666053173	-0.088028859
H	-7.732287479	3.434726701	-0.134012526
H	-2.972295070	-2.863768638	-1.604524068
H	-4.156440615	-2.271663001	-2.803162682
H	-4.370485928	-3.884769752	-2.064469650
H	-5.366822948	-2.661894423	1.353386452
H	-3.716698681	-3.155795187	0.864268254
H	-5.132140384	-4.110659951	0.334696025
H	-7.027199012	-1.848819946	-0.406679967
H	-6.657675280	-3.301562728	-1.364966146
H	-6.517134322	-1.696676110	-2.116954032
H	-2.485136522	2.974536660	2.008896265
H	-1.242396524	4.820245001	3.075256680
H	1.260696802	4.834005112	3.033277213
H	2.487869643	2.998982924	1.929886212
H	-0.016409968	0.994433871	-2.094215238
H	0.033846954	-3.296712141	-1.546506683
H	-0.005386938	0.675540770	-4.543700527
H	0.045399010	-3.605941809	-3.993389574
H	0.025388957	-1.623615312	-5.511863552
H	2.008080887	-0.927669281	2.721524925
H	-1.958138800	-2.535706181	2.189930548
H	2.062877356	-1.810958499	5.027749530
H	-1.891084079	-3.457564001	4.488810384
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5_T

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Sn	-0.009635348	-0.856011400	0.417668762
O	2.243874977	-1.539910451	0.194632559
O	-2.237144541	-1.534676391	0.147646065
N	1.302062053	0.848053824	0.699377443
N	-1.314736689	0.853207941	0.705143634
C	3.046180846	-0.576959012	-0.025893195
C	2.571059979	0.785807000	0.233631354
C	3.409074664	1.890110112	-0.049151971
C	4.688893588	1.699500197	-0.527746555
C	5.157905013	0.362806128	-0.714911309
C	4.400117837	-0.764631364	-0.493205712
C	5.616187090	2.861079126	-0.883672125
C	4.972852952	4.222082391	-0.612915659
C	5.972766352	2.784134574	-2.377614060
C	6.902052244	2.761241059	-0.046424830
C	4.933933547	-2.175749392	-0.741979598
C	6.392743431	-2.159247295	-1.205051597
C	4.093351569	-2.853408443	-1.837426109
C	4.858276451	-2.996195398	0.557127725
C	-3.051422362	-0.570477559	-0.037934258
C	-2.588026211	0.788150075	0.256581703

C	-3.441486093	1.891639344	0.014152921
C	-4.723321918	1.702576908	-0.456834067
C	-5.176143186	0.367424081	-0.686327027
C	-4.402349918	-0.757835093	-0.509529282
C	-5.668221565	2.864174236	-0.762809168
C	-5.036103116	4.222760574	-0.455744979
C	-6.046294526	2.833033224	-2.252934958
C	-6.940311071	2.722038058	0.089233407
C	-4.920292294	-2.163304609	-0.819562037
C	-4.072302972	-2.781066711	-1.944316309
C	-4.837246969	-3.042267120	0.439596063
C	-6.379586338	-2.144446203	-1.281786214
C	0.711174688	1.930239480	1.331387341
C	-0.718811224	1.928018941	1.343737598
C	-1.395090392	2.917780558	2.073592756
C	-0.692603747	3.909043779	2.752991953
C	0.702829002	3.912779204	2.738488496
C	1.396875471	2.923004247	2.047989626
C	0.006970113	-1.123794509	-1.623491770
C	0.008103761	-0.011647453	-2.472544971
C	0.011169226	-2.401888600	-2.198330311
C	0.013793379	-0.168452375	-3.860462260
C	0.017790760	-2.565541740	-3.582909152
C	0.018652817	-1.446287015	-4.416968402
C	0.011794275	-1.791382730	2.250642916
C	1.149787595	-1.685931733	3.063283775
C	-1.088998052	-2.508207882	2.738286424
C	1.182659266	-2.263805301	4.331772622
C	-1.056926894	-3.094706156	4.004388041
C	0.077998280	-2.972356153	4.804250058
H	3.006405502	2.896078830	0.071890711
H	6.181473119	0.233042492	-1.074534331
H	4.058745118	4.373845018	-1.212866306
H	5.679784194	5.026600114	-0.879681748
H	4.711942712	4.350872026	0.452171688
H	5.066448262	2.859936892	-3.004039159
H	6.480482077	1.838814326	-2.634377873
H	6.650289183	3.612977199	-2.652224465
H	7.446183936	1.819009243	-0.229267124
H	6.677093862	2.815653048	1.033500143
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H	6.521150989	-1.622952110	-2.161964144
H	6.735877200	-3.196469873	-1.361765481
H	7.062831215	-1.700171702	-0.456379367
H	4.160382811	-2.292291335	-2.786812618
H	3.033612394	-2.914376890	-1.547746990
H	4.467817160	-3.877305887	-2.020426169
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H	3.822144831	-3.077505149	0.919339942
H	5.473566836	-2.533527458	1.350184522
H	-3.047440833	2.897531961	0.158804268
H	-6.199766141	0.236806112	-1.045344347

H	-4.761887911	4.318923828	0.609446453
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H	-4.132036109	4.405011012	-1.062458183
H	-5.150490166	2.938857651	-2.890050564
H	-6.736044261	3.663459667	-2.489742417
H	-6.549036567	1.891814925	-2.533537645
H	-6.699850283	2.743716973	1.166946665
H	-7.476243427	1.779706492	-0.116376913
H	-7.636698806	3.553290583	-0.123456581
H	-3.010103605	-2.835052464	-1.662046497
H	-4.152901197	-2.182255007	-2.869218793
H	-4.429634826	-3.802884346	-2.168676546
H	-5.446758399	-2.618193114	1.257920071
H	-3.798502467	-3.137808648	0.789059842
H	-5.223494914	-4.053780631	0.218334497
H	-7.055622710	-1.729178764	-0.513227281
H	-6.709255086	-3.177654494	-1.486508286
H	-6.515118263	-1.566604697	-2.213079931
H	-2.483424279	2.875476113	2.152483108
H	-1.241928562	4.668058371	3.319794694
H	1.259631377	4.675359980	3.293181301
H	2.487131880	2.887537698	2.102652101
H	0.004931930	1.002947185	-2.054936585
H	0.011041675	-3.295050699	-1.561996865
H	0.014465210	0.715076761	-4.508897375
H	0.022251461	-3.573377924	-4.013220126
H	0.023401590	-1.571892988	-5.505830349
H	2.031267868	-1.148074696	2.699167737
H	-1.987219170	-2.601892713	2.122094459
H	2.080704056	-2.163460769	4.951875297
H	-1.928358657	-3.653065990	4.365241506
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N	1.292639507	1.350110352	0.420575983
N	-1.379639326	1.314236590	0.390966355
N	0.082527194	-3.034224398	-0.011696380
C	3.061283517	-0.178319275	0.132230761
C	2.619026510	1.199019511	0.125407575
C	3.510040874	2.237595543	-0.235798237
C	4.837908308	1.974612638	-0.501471988
C	5.282366777	0.632904748	-0.360982629
C	4.462585799	-0.439489020	-0.057724959
C	5.828396533	3.049723243	-0.952173443
C	5.176260061	4.430108286	-1.047020744
C	6.383590305	2.684383900	-2.338385367
C	6.988147583	3.134242639	0.053122444
C	5.011853071	-1.862337424	0.086506544

C	6.525179351	-1.919820419	-0.139225271
C	4.351250705	-2.804491478	-0.933266564
C	4.730142536	-2.366229081	1.512264740
C	-3.058275451	-0.268942195	-0.109300773
C	-2.707157715	1.116541341	0.120745960
C	-3.678612689	2.133580426	-0.049856775
C	-4.992138107	1.829229968	-0.341702158
C	-5.337264747	0.458208050	-0.456999733
C	-4.435541407	-0.587548923	-0.371275471
C	-6.065666911	2.895755831	-0.565573155
C	-5.510580033	4.313623678	-0.418037655
C	-6.646939914	2.752280093	-1.981443909
C	-7.190282233	2.709984940	0.465880482
C	-4.870690225	-2.039717622	-0.590676742
C	-4.216034223	-2.565429268	-1.878582848
C	-4.453107562	-2.912516255	0.604383666
C	-6.388423863	-2.169906851	-0.748070630
C	0.673130633	2.426756677	0.886553231
C	-0.799988236	2.385425812	0.920643506
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C	1.329670775	3.579261158	1.437600412
C	0.179330820	-0.448951779	-1.892979179
C	-0.757083439	0.251885540	-2.665384686
C	1.237780172	-1.078699241	-2.560176501
C	-0.641127284	0.323547482	-4.054728832
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C	0.417438198	-0.312890929	-4.702043552
C	-0.127496015	-0.841471250	2.253859753
C	0.944917618	-0.511335622	3.093653421
C	-1.292347697	-1.350474638	2.844070967
C	0.856501911	-0.673288285	4.477518466
C	-1.385367993	-1.518963023	4.226757688
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C	-0.468808465	-5.045287564	-1.186380101
C	0.776464783	-5.151139837	0.861759435
C	0.177625249	-5.805497975	-0.213444105
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H	3.266961050	-2.875120765	-0.761728562
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C	-3.451063736	-2.271147201	-0.226890073
C	-4.773884333	-2.038959750	-0.533563176
C	-5.256887215	-0.702335036	-0.432919113
C	-4.474301540	0.393509015	-0.126486064
C	-5.730647335	-3.142926692	-0.985599389
C	-5.048188302	-4.510717803	-1.040654080
C	-6.261404813	-2.812943772	-2.390199419
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C	-4.393265156	2.753103869	-1.021026737
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C	4.943754410	-1.878922294	-0.350323781
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C	5.414664434	-4.373795233	-0.379716686
C	6.536465061	-2.869148407	-2.008937435
C	7.144775212	-2.780540090	0.422760216
C	4.906740271	1.988407411	-0.675181480
C	4.236288911	2.501894586	-1.959901791
C	4.533591712	2.893258744	0.510752363
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C	-0.610367443	-4.516282393	2.212357011
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C	-1.164615263	1.173388837	-2.532042931
C	0.600248881	-0.388004592	-4.005944669
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C	-0.393618696	0.334899209	-4.664619015
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H	4.595151357	-4.586421239	-1.088347746
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5·Py_T

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C	3.356997252	2.320978182	-0.212931686
C	4.670088549	2.142022010	-0.587524089
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C	4.477656772	-0.313844568	-0.258470133
C	5.570991966	3.295111761	-1.029279293
C	4.842478706	4.639958990	-1.004091631
C	6.057617808	3.036726030	-2.464469007
C	6.782718594	3.386189553	-0.087537923
C	5.118649772	-1.703933115	-0.223603264
C	6.605212118	-1.657875517	-0.587379872
C	4.422824822	-2.648373844	-1.216852599
C	4.997563582	-2.274651424	1.199827485
C	-3.069147789	-0.259331810	-0.136653988
C	-2.656276280	1.104724914	0.200093583
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C	-4.873877957	1.945626166	-0.356453251
C	-5.277695984	0.599559194	-0.600506876
C	-4.437635991	-0.491383228	-0.524820852
C	-5.887221040	3.071584597	-0.562449401
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C	-6.367196870	3.064962045	-2.023054889
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C	-6.413273603	-1.949744144	-1.174156567
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C	-0.786802314	2.299576281	1.201619189
C	-1.471351395	3.265874623	1.958337893

C	-0.787807692	4.293943338	2.600782468
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C	0.917528430	-1.087822744	4.500545159
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H	6.771227457	-1.288710175	-1.615075624
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H	-6.058821157	5.227044937	-0.417077570
H	-4.439871230	4.684631478	-0.918126829
H	-5.522784342	3.226853114	-2.716292844
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H	-3.081590725	-2.268493827	-2.034772350

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H	-2.551379575	3.175479184	2.088440215
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H	-1.141584327	1.215544531	-4.471274990
H	1.781767590	-1.960178625	-4.423010722
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H	1.886997523	-0.318006389	2.737682702
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Charge = -1, S = 1/2, E = -8.008271959912147E+03

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N	1.305436050	0.965675696	0.500726780
N	-1.315918384	0.963000806	0.526127789
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5⁺

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