

Insulator–Metal Transition Induced by Band-Filling Modulation in Molecular Mott Insulators

Mayu Kaneko,^a Masahiro Masuda,^a Ryota Teruya^a, Tetsu Sato,^b Tomoyuki Akutagawa^b and Masaki Matsuda^{*a}

^a Department of Chemistry, Kumamoto University, 2-39-1 Kurokami, Chuo-ku, Kumamoto 860-8555, Japan. E-mail: masaki@kumamoto-u.ac.jp

^b Institute of Multidisciplinary Research for Advanced Materials, Tohoku University, 2-1-1 Katahira, Aoba-ku, Sendai 980-8579, Japan.

Contents

Fig. S1 TG analysis curves of $x\text{-Li(Pc)Br}_x$ and $x\text{-Li(tbp)I}_x$	2
Table S1 Fractional coordinates, thermal parameters and occupancies for $x\text{-Li(Pc)Br}_{0.75}$ and $x\text{-Li(tbp)I}_{0.75}$	3
Table S2 Crystal data and crystal refinement for reproduced $x\text{-Li(Pc)}$ by chemical reduction of $x\text{-Li(Pc)Br}_x$	4
Fig. S2 Crystal structure of reproduced $x\text{-Li(Pc)}$ by chemical reduction of $x\text{-Li(Pc)Br}_x$	5
Fig. S3 Crystal structure of $x\text{-Li(Pc)Br}_x$ employing the split-atom model for Br atoms.	6

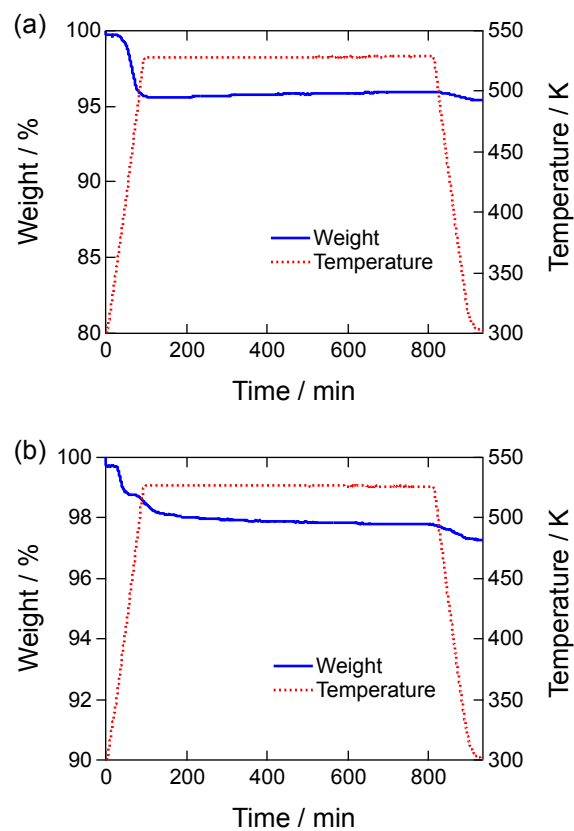


Fig. S1 TG analysis curves of $x\text{-Li(Pc)Br}_x$ (a) and $x\text{-Li(tbp)I}_x$ (b).

Table S1.1 Fractional coordinates, thermal parameters, and occupancy for x-Li(Pc)Br_{0.75}.

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{11}/\text{\AA}^2$	$U_{22}/\text{\AA}^2$	$U_{33}/\text{\AA}^2$	$U_{12}/\text{\AA}^2$	$U_{13}/\text{\AA}^2$	$U_{23}/\text{\AA}^2$	Occ.
Br	0	0	0.25	0.0524	0.0524	0.513	0.0000	0.0000	0.0000	0.75
Li	0.5	0.5	0.5	0.0270	0.0270	0.0790	0.0000	0.0000	0.0000	1
N1	0.3734(2)	0.4395(2)	0.5	0.0264	0.0258	0.0450	0.0004	0.0000	0.0000	1
N2	0.2711(2)	0.5810(2)	0.5	0.0300	0.0301	0.0340	0.0015	0.0000	0.0000	1
C1	0.3556(3)	0.3431(3)	0.5	0.0300	0.0320	0.0310	−0.0023	0.0000	0.0000	1
C2	0.2506(3)	0.3253(3)	0.5	0.0310	0.0350	0.0360	−0.0054	0.0000	0.0000	1
C3	0.1956(3)	0.2420(3)	0.5	0.0400	0.0340	0.0520	−0.0050	0.0000	0.0000	1
C4	0.0963(3)	0.2526(4)	0.5	0.0380	0.0500	0.0670	−0.0160	0.0000	0.0000	1
C5	0.0541(3)	0.3429(4)	0.5	0.0270	0.0590	0.0810	−0.0080	0.0000	0.0000	1
C6	0.1090(3)	0.4268(4)	0.5	0.0280	0.0450	0.0590	−0.0013	0.0000	0.0000	1
C7	0.2078(3)	0.4156(3)	0.5	0.0290	0.0320	0.0330	−0.0042	0.0000	0.0000	1
C8	0.2876(3)	0.4855(3)	0.5	0.0270	0.0350	0.0340	0.0007	0.0000	0.0000	1
H3	0.224143	0.181389	0.5	0.0510 ^{a)}						1
H4	0.057389	0.198004	0.5	0.0620 ^{a)}						1
H5	−0.012729	0.347608	0.5	0.0670 ^{a)}						1
H6	0.080466	0.487401	0.5	0.0530 ^{a)}						1

a) thermal parameters for hydrogen atoms are U_{eq} .**Table S1.2** Fractional coordinates, thermal parameters, and occupancy for x-Li(tbp)I_{0.75}.

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{11}/\text{\AA}^2$	$U_{22}/\text{\AA}^2$	$U_{33}/\text{\AA}^2$	$U_{12}/\text{\AA}^2$	$U_{13}/\text{\AA}^2$	$U_{23}/\text{\AA}^2$	Occ.
I	0	0	0.25	0.0551	0.0551	0.1425	0.0000	0.0000	0.0000	0.75
Li	0.5	0.5	0.5	0.0450	0.0450	0.0460	0.0000	0.0000	0.0000	1
N	0.3699(3)	0.4391(2)	0.5	0.0340	0.0292	0.0320	0.0000	0.0000	0.0000	1
C1	0.3509(3)	0.3437(3)	0.5	0.0330	0.0370	0.0290	−0.0053	0.0000	0.0000	1
C2	0.2490(3)	0.3270(3)	0.5	0.0320	0.0410	0.0300	−0.0046	0.0000	0.0000	1
C3	0.1936(4)	0.2480(4)	0.5	0.0420	0.0400	0.0510	−0.0060	0.0000	0.0000	1
C4	0.0973(4)	0.2580(4)	0.5	0.0420	0.0570	0.0770	−0.0150	0.0000	0.0000	1
C5	0.0569(4)	0.3468(4)	0.5	0.0300	0.0650	0.0820	−0.0090	0.0000	0.0000	1
C6	0.1116(3)	0.4269(4)	0.5	0.0310	0.0490	0.0590	−0.0010	0.0000	0.0000	1
C7	0.2077(3)	0.4162(3)	0.5	0.0330	0.0390	0.0300	−0.0063	0.0000	0.0000	1
C8	0.2863(3)	0.4836(3)	0.5	0.0320	0.0370	0.0300	−0.0035	0.0000	0.0000	1
C9	0.2750(3)	0.5825(3)	0.5	0.0290	0.0350	0.0340	0.0027	0.0000	0.0000	1
H3	0.220715	0.188613	0.5	0.0530 ^{a)}						1
H4	0.059126	0.204981	0.5	0.0700 ^{a)}						1
H5	−0.008150	0.352429	0.5	0.0710 ^{a)}						1
H6	0.084320	0.486260	0.5	0.0550 ^{a)}						1

a) thermal parameters for hydrogen atoms are U_{eq} .

Table S2 Crystal data and structure refinement results for reproduced x-Li(Pc) by chemical reduction of x-Li(Pc)Br_x.

Chemical formula	C ₃₂ H ₁₆ LiN ₈
Formula weight	519.47
Crystal description	Black needle
<i>T</i> (K)	293
Crystal system	Tetragonal
Space group	<i>P4/mcc</i>
<i>a</i> (Å)	13.8420 (5)
<i>c</i> (Å)	6.4865 (4)
<i>V</i> (Å ³)	1242.82 (12)
<i>Z</i>	2
<i>d</i> _{cal} (g cm ⁻³)	1.388
Radiation	MoKα
Wavelength (Å)	0.71073
<i>μ</i> (mm ⁻¹)	0.086
No. of measured reflections	4383
No. of independent reflection	802
No. of observed reflections	551 (<i>I</i> > 2σ (<i>I</i>))
<i>R</i> (<i>I</i> > 2σ (<i>I</i>))	<i>R</i> ₁ = 0.0427
	<i>wR</i> ₂ = 0.1464
Goodness-of-fit	0.980
Parameters	63

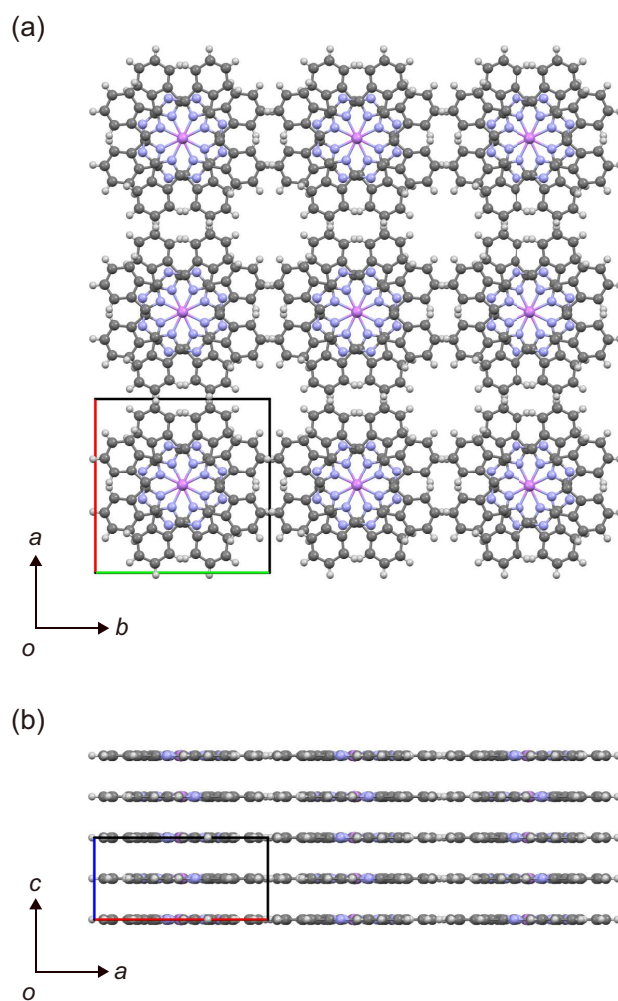


Fig. S2 Crystal structure of reproduced $x\text{-Li(Pc)}$ by chemical reduction of $x\text{-Li(Pc)Br}_x$ viewed along the c -axis (a) and the b -axis (b).

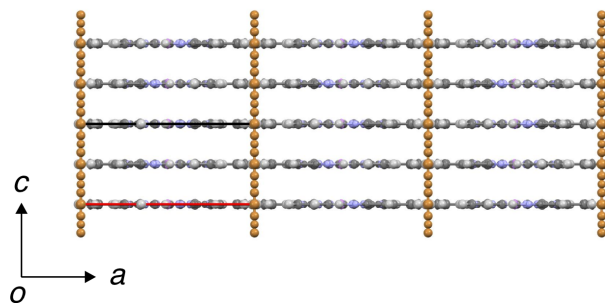


Fig. S3 Crystal structure of $x\text{-Li(Pc)Br}_x$ employing the split-atom model for Br atoms.