Electronic supplementary information

for

Syntheses and structures of cationic osmium bis(σ-B–H) borane complexes

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General information

X-ray diffraction: Single-crystal X-ray diffraction data were collected on a Bruker D8 Venture diffractometer equipped with a Photon 100 CMOS detector using Mo K α radiation ($\lambda = 0.71073$ Å). All of the data were corrected for absorption effects using the multi-scan technique. Final unit cell parameters were based on all observed reflections from integration of all frame data. The structures were solved with the ShelXT structure solution program using Intrinsic Phasing and refined with the ShelXL refinement package using Least Squares minimization that implanted in Olex2. For all compounds, all non-H atoms were refined anisotropically unless otherwise stated, and hydrogen atoms were introduced at their geometric positions and refined as riding atoms unless otherwise stated. For 2a and 2e the Cy groups and NTf counter anions were found disordered over two positions, and several restraints were used in order to improve refinement stability. For 2a, the distance between Os and terminal H is kept during refinement at the value of 1.576 Å using DFIX restraint. For 2e, the distance between Os and terminal H is kept during refinement at the value of 1.585 Å using DFIX restraint. CCDC 2434259 (2e) and 2434260 (2a) contain the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures/.



Fig. S2. ¹³C{¹H} NMR (101 MHz, 298 K, CDCl₃) spectrum of 1a.



Fig. S3 $^{11}B{}^{1}H$ NMR (128 MHz, 298 K, CDCl₃) spectrum of 1a.



oxidation products of PCy₃.]

NMR spectra of 1b



Scheme S2











Fig. S9 ¹⁹F{¹H} NMR (377 MHz, 298 K, CDCl₃) spectrum of **1b**.

NMR spectra of 1c



Scheme S3



Fig. S11 $^{13}C\{^{1}H\}$ NMR (101 MHz, 298 K, CDCl₃) spectrum of 1c .



Fig. S12 ¹¹B{¹H} NMR (128 MHz, 298 K, CDCl₃) spectrum of **1c**.



oxidation products of PCy₃.]



Fig. S14 ¹⁹F{¹H} NMR (377 MHz, 298 K, CDCl₃) spectrum of 1c.



Fig. S15 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 1d.





Fig. S20 ¹³C{¹H} NMR (101 MHz, 298 K, CDCl₃) spectrum of 1e.



90 85 80 75 70 65 60 55 50 45 40 35 30 25 20 15 10 5 0 -5 -10 -15 -20 -25 -30 -35 -40 -45 -50 -55 -60

Fig. S21 ¹¹B{¹H} NMR (128 MHz, 298 K, CDCl₃) spectrum of **1e**.



during the synthesis of $OsHCl(CO)(PCy_3)_2$, which might be one of the oxidation products of PCy_3 .]

NMR spectra of 1f



Scheme S6



Fig. S23 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 1f.



Fig. S24 ${}^{13}C{}^{1}H$ NMR (101 MHz, 298 K, CDCl₃) spectrum of 1f.



Fig. S25 ${}^{11}B{}^{1}H{}$ NMR (128 MHz, 298 K, CDCl₃) spectrum of 1f.







Fig. S27 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 2a.





Fig. S29 $^{11}B{}^{1}H$ NMR (128 MHz, 298 K, CDCl₃) spectrum of 2a.



Fig. S30 ${}^{31}P{}^{1}H$ NMR (162 MHz, 298 K, CDCl₃) spectrum of 2a.

X-ray crystal structure analysis of complex 2a: formula $C_{46}H_{76}BCl_2F_6NO_5OsP_2S_2$, M = 1235.04 g/mol, dull light orange crystal, $0.18 \times 0.13 \times 0.1 \text{ mm}$, a = 16.850(4), b = 18.960(5), c = 16.990(4) Å, a = 90, $\beta = 94.936(7)$, $\gamma = 90^{\circ}$, V = 5408(2) Å³, $\rho_{calc} = 1.517$ g·cm⁻³, $\mu = 2.657$ mm⁻¹, empirical absorption correction (0.5202 \leq T \leq 0.7463), Z = 4, monoclinic, space group $P2_1/n$, $\lambda = 0.71073$ Å, T = 120 K, ω and φ scans, 98977 reflections collected ($\pm h$, $\pm k$, $\pm l$), 9513 independent ($R_{int} = 0.0900$) and 8174 observed reflections [$I > 2\sigma(I)$], 692 refined parameters, R = 0.0364, $wR^2 = 0.0963$, max. (min.) residual electron density 1.58 (-1.23) e.Å⁻³.



Fig. S31 Molecular structure of 2a (thermal ellipsoids are shown at the 30% probability level).

NMR spectra of 2b



Scheme S8



S18



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Fig. S35 ³¹P{¹H} NMR (162 MHz, 298 K, CDCl₃) spectrum of 2b.



Fig. S36 ¹⁹F{¹H} NMR (377 MHz, 298 K, CDCl₃) spectrum of **2b**.

NMR spectra of 2c



Scheme S9



Fig. S37 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 2c.





Fig. S39 $^{11}B{}^{1}H$ NMR (128 MHz, 298 K, CDCl₃) spectrum of 2c.



Fig. S40 ³¹P{¹H} NMR (162 MHz, 298 K, CDCl₃) spectrum of **2c.**



Fig. S41 ¹⁹F{¹H} NMR (377 MHz, 298 K, CDCl₃) spectrum of 2c.

NMR spectra of 2d



Scheme S10



Fig. S42 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 2d.



Fig. S44 ¹¹B{¹H} NMR (128 MHz, 298 K, CDCl₃) spectrum of 2d.



Fig. S45 ³¹P{¹H} NMR (162 MHz, 298 K, CDCl₃) spectrum of 2d.

NMR spectra and crystallographic data of 2e







Fig. S46 ¹H NMR (400 MHz, 298 K, CDCl₃) spectrum of 2e.





Fig. S48 ¹¹B{¹H} NMR (128 MHz, 298 K, CDCl₃) spectrum of **2e**.



Fig. S49 ³¹P{¹H} NMR (162 MHz, 298 K, CDCl₃) spectrum of **2e**.

X-ray crystal structure analysis of complex 2e: formula $C_{48}H_{80}BF_6NO_5OsP_2S_2$, M = 1192.20 g/mol, clear light orange crystal, 0.26 $\times 0.15 \times 0.1$ mm, a = 16.553(2), b = 16.8037(18), c = 19.306(2) Å, a = 90, $\beta = 90, \gamma = 90^{\circ}, V = 5370.0(11) \text{ Å}^3, \rho_{\text{calc}} = 1.475 \text{ g} \cdot \text{cm}^{-3}, \mu = 2.576 \text{ mm}^{-1},$ empirical absorption correction (0.4492 \leq T \leq 0.7460), Z = 4, orthorhombic, space group $P2_12_12_1, \lambda = 0.71073 \text{ Å}, T = 120 \text{ K}, \omega$ and φ scans, 55997 reflections collected ($\pm h, \pm k, \pm l$), 15354 independent ($R_{\text{int}} = 0.0979$) and 12197 observed reflections [$I > 2\sigma(I)$], 794 refined parameters, R = 0.0443, $wR^2 = 0.0875$, max. (min.) residual electron density 1.60 (-1.18) e.Å^{-3}.



Fig. S50 Molecular structure of **2e** (thermal ellipsoids are shown at the 30% probability level).

IR spectra of 1a-1f and 2a-2e







Fig. S52 IR spectrum of 1b.



Fig. S53 IR spectrum of 1c.



Fig. S54 IR spectrum of 1d.



Fig. S55 IR spectrum of 1e.



Fig. S56 IR spectrum of 1f.



Fig. S57 IR spectrum of 2a.



Fig. S58 IR spectrum of 2b.



Fig. S59 IR spectrum of 2c.



Fig. S60 IR spectrum of 2d.



Fig. S61 IR spectrum of 2e.

Computational Details

Geometry optimizations of **1a** and **2a** were performed at the PBE0¹ /def2-SVP^{2,3} level, in combination with RIJOCSX⁴ and the def2/J⁵ auxiliary basis set planted in the ORCA5.0.1⁶⁻⁸ suite of programs. During the calculations, the D3 dispersion correction suggested by Grimme et al^{9,10} was used. All species were characterized as local minima by analytic frequency calculations. For AIM analysis, single-point calculations on the computationally optimized geometries were performed with in M06-L¹¹ /def2-SVP level, in combination with Split-RI-J¹² and the def2/J auxiliary basis set. The topology of the electron density was conducted through the Multiwfn program package.¹³

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Os	4.420492000	4.740779000	16.310368000
Р	4.519154000	5.365920000	14.004702000
Р	4.186911000	4.619323000	18.686492000
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С	0.966854000	8.360866000	12.488969000
Н	0.457774000	9.318405000	12.298339000
Н	0.273223000	7.568915000	12.154157000
С	9.634624000	1.409361000	14.932897000
Н	10.550083000	0.867917000	14.682872000
Н	4.019723000	6.261625000	16.384753000
Н	4.863729000	2.922593000	16.041197000
Н	6.143592000	4.791472000	16.370328000

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