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Supporting Information

Tris(pentafluoroethyl)difluorophosphorane for Fluoride Abstraction and Ligand Exchange Reactions of *N*-Heterocyclic Carbene and Cyclic Alkyl(amino)carbene Copper(I) Fluorides

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Contents

- 1) Crystallographic Details
- 2) NMR spectra of Compounds

1) Crystallographic Details

Crystal data were collected on a Bruker X8 Apex-2 diffractometer with a CCD area detector and graphite monochromated Mo–K α radiation or a Rigaku XtaLAB Synergy-DW diffractometer with an Hy-Pix-6000HE detector and monochromated Cu–K α radiation equipped with an Oxford Cryo 800 cooling unit. Crystals were immersed in a film of perfluoropolyether oil on a glass fiber MicroMountTM (MiTeGen) and data were collected at 100 K. Images were processed with Bruker or CrySalis software packages and equivalent reflections were merged. Corrections for Lorentz-polarization effects and absorption were performed if necessary and the structures were solved by direct methods. Subsequent difference Fourier syntheses revealed the positions of all other non-hydrogen atoms. Structures were solved by using the ShelXTL software package.¹ All non-hydrogen atoms were refined anisotropically. All hydrogen atoms were assigned to idealized geometric positions and were included in structure factors calculations.

Crystallographic data for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as supplementary publication no.s CCDC 2444352 (1), CCDC 2444336 (3), CCDC 2444351 (5), CCDC 2444348 (6), CCDC 2444349 (7), CCDC 2444343 (10), CCDC 2444350 (11), CCDC 2444347 (12), CCDC 2444346 (16), CCDC 2444340 (17), CCDC 2444345 (19), CCDC 2444342 (20), CCDC 2444338 (21), CCDC 2444341 (23), CCDC 2444344 (24), CCDC 2444339 (26), CCDC 2444337 (27). Copies of the data can be obtained free of charge on application to CCDC.



Figure S1. Molecular structure of [(IDipp)Cu(NH₃)]***FAP**⁻ (**1**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms except the ones directly bound to N1 and co-crystallized solvent molecule are omitted for clarity. Only one of two independent molecules in the asymmetric unit is shown. Selected bond lengths [Å] and angles [°]: Cu–C1 1.882(3), Cu–N1 1.908(3), N2–C1 1.360(4), N3–C1 1.353(43); C1–Cu–N1 177.19(15).

Crystal Data for [(IDipp)Cu(NH₃)]⁺**FAP**⁻ **(1)**: $C_{67}H_{79}Cl_3Cu_2F_{36}N_6P_2$, $M_r = 1947.73$, T = 100.00(2) K, $\lambda = 1.54184$ Å, colorless block, $0.200 \times 0.290 \times 0.400$ mm³, triclinic space group $P^{1;-}$, a = 12.7224(2) Å, b = 16.7168(2) Å, c = 21.0326(2) Å, $\alpha = 98.6100(10)^\circ$, $\beta = 107.3470(10)^\circ$, $\gamma = 95.1630(10)^\circ$, V = 4178.42(10) Å³, Z = 2, $\rho_{calcd} = 1.548$ g·cm⁻³, $\mu = 2.990$ mm⁻¹, F(000) = 1972, 84475 reflections,

 $-16 \le h \le 16$, $-21 \le k \le 19$, $-26 \le l \le 26$, $2.240^{\circ} < \theta < 80.839^{\circ}$, completeness 97.6%, 17912 independent reflections, 15317 reflections observed with [l > 2**23** (I)], 1552 parameters, 1638 restraints, *R* indices (all data) $R_1 = 0.0981$, $wR_2 = 0.2629$, final *R* indices [l > 2σ (I)] $R_1 = 0.0904$, $wR_2 = 0.2520$, largest difference peak and hole 1.380 and $-1.053 e Å^{-3}$, GooF = 1.034.



Figure S2. Molecular structure of [(IDipp)Cu(C₇H₁₀N₂)]***FAP**⁻ (**3**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms and co-crystallized solvent molecule are omitted for clarity. Only one of two independent molecules in the asymmetric unit is shown. Selected bond lengths [Å] and angles [°]: Cu–C1 1.875(2), Cu–N1 1.8785(19), N2–C1 1.357(3), N3–C1 1.355(3); C1–Cu–N1 175.45(9).

Crystal Data for [(IDipp)Cu(C₇H₁₀N₂)]⁺**FAP**⁻ **(3):** $C_{81}H_{93}Cl_3Cu_2F_{36}N_8P_2$, $M_r = 2158.00$, T = 100.00(2) K, $\lambda = 1.54184$ Å, colorless block, $0.117 \times 0.196 \times 0.283$ mm³, triclinic space group P¹; , a = 11.33150(10) Å, b = 19.5059(2) Å, c = 23.4078(3)°, $\alpha = 102.6080(10)°$, $\beta = 99.0610(10)°$, $\gamma = 102.3770(10)°$, V = 4817.16(9) Å³, Z = 2, $\rho_{calcd} = 1.488$ g·cm⁻³, $\mu = 2.660$ mm⁻¹, F(000) = 2196, 70913 reflections, $-14 \le h \le 14$, $-24 \le k \le 23$, $-28 \le I \le 29$, $1.980° < \theta < 74.501°$, completeness 98.5%, 19406 independent reflections, 17127 reflections observed with [I > 2σ (I)], 1655 parameters, 2541 restraints, *R* indices (all data) $R_1 = 0.0584$, $wR_2 = 0.1571$, final *R* indices [I > 2σ (I)] $R_1 = 0.0533$, $wR_2 = 0.1519$, largest difference peak and hole 1.562 and -1.211 e Å⁻³, GooF = 1.063.



Figure S3. Molecular structure of $[(IDipp)Cu(C_{13}H_9N)]^{+}FAP^{-}$ (5) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.8789(18), Cu-N1 1.9038(15), N2-C1 1.359(2), N3-C1 1.358(2); C1-Cu-N1 179.62(7).

Crystal Data for [(IDipp)Cu(C₁₃H₉N)]⁺**FAP**⁻ **(5):** C₄₆H₄₅CuF₁₈N₃P, M_r = 1076.36, T = 100.00(2) K, $\lambda = 0.71073$ Å, colorless block, $0.070 \times 0.160 \times 0.240$ mm³, monoclinic space group $P2_1/c$, a = 16.1433(4) Å, b = 16.7833(3) Å, c = 17.5037(4) Å, $\beta = 103.057(3)^\circ$, V = 4619.80(18) Å³, Z = 4, $\rho_{calcd} = 1.548$ g·cm⁻³, $\mu = 0.618$ mm⁻¹, F(000) = 2192, 53839 reflections, $-22 \le h \le 20$, $-21 \le k \le 24$, $-23 \le I \le 24$, $1.969^\circ < \theta < 31.243^\circ$, completeness 81.8%, 12308 independent reflections, 9039 reflections observed with [I > 2 σ (I)], 630 parameters, 0 restraints, *R* indices (all data) $R_1 = 0.0657$, $wR_2 = 0.1047$, final *R* indices [I > 2 σ (I)] $R_1 = 0.0399$, $wR_2 = 0.0939$, largest difference peak and hole 0.608 and -0.548 e Å⁻³, GooF = 1.016.



Figure S4. Molecular structure of $[(IDipp)Cu(\eta^1-O=C_{13}H_9N)]^*FAP^-$ (6) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms and a co-crystallized solvent molecule are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.863(2), Cu-O 1.8313(16), N1-C1 1.356(3), N2-C1 1.352(3), C1-Cu-O 169.58(9), Cu-O-C2 139.64(16).

Crystal Data for [(IDipp)Cu(\eta^1-O=C₁₃H₉N)]⁺**FAP**⁻ (6): C₄₉H₅₂CuF₁₈N₃OP, M_r = 1135.44, T = 100.00(2) K, $\lambda = 1.54184$ Å, colorless block, 0.150×0.170×0.480 mm³, triclinic space group P^{1} ; , a = 12.70530(10) Å, b = 13.77550(10) Å, c = 16.5290(2) Å, $\alpha = 73.0900(10)^{\circ}$, $\beta = 68.1900(10)^{\circ}$, $\gamma = 86.4690(10)^{\circ}$, V = 2566.21(5) Å³, Z = 2, $\rho_{calcd} = 1.469$ g·cm⁻³, $\mu = 1.837$ mm⁻¹, F(000) = 1162, 53129 reflections, -16 ≤ h ≤ 13, -17 ≤ k ≤ 17, -20 ≤ l ≤ 20, 3.008° < θ < 79.995°, completeness 97.9%, 10954 independent reflections, 9734 reflections observed with [l > 2 σ (l)], 656 parameters, 0 restraints, *R* indices (all data) $R_1 = 0.0572$, $wR_2 = 0.1495$, final *R* indices [l > 2 σ (l)] $R_1 = 0.0522$, $wR_2 = 0.1450$, largest difference peak and hole 1.034 and -0.991 e Å⁻³, GooF = 1.075.



Figure S5. Molecular structure of [(IDipp)Cu(C₄H₁₀S)]***FAP**⁻ (7) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.887(3), Cu–S 2.1705(8), N1–C1 1.351(3), N2–C1 1.353(3); C1–Cu–S 176.49(8).

Crystal Data for [(IDipp)Cu(C₄H₁₀S)]⁺**FAP**⁻ **(7):** C₃₇H₄₆CuF₁₈N₂PS, M_r = 987.33, T = 100.00(2) K, $\lambda = 1.54184 \text{ Å}$, colorless block, 0.099×0.193×0.362 mm³, monoclinic space group P2₁/n, a = 15.3271(2) Å, b = 16.1495(2) Å, c = 17.9666(3) Å, $\beta = 101.9160(10)^{\circ}$, V = 4351.35(11) Å³, Z = 4, $\rho_{calcd} = 1.507 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 2.481 \text{ mm}^{-1}$, F(000) = 2016, 33531 reflections, $-19 \le h \le 18$, $-20 \le k \le 11$, $-22 \le I \le 20$, $3.456^{\circ} < \theta < 74.496^{\circ}$, completeness 98.4%, 8751 independent reflections, 7643 reflections observed with [I > 2 σ (I)], 551 parameters, 0 restraints, R indices (all data) R₁ = 0.0635, wR₂ = 0.1589, final R indices [I > 2 σ (I)] R₁ = 0.0572, wR₂ = 0.1531, largest difference peak and hole 1.802 and $-1.040 \ e \ A^{-3}$, GooF = 1.053.



Figure S6. Molecular structure of $[{IDipp}Cu]_2(C_2N_3H_3)_2]^{2+}$ **2FAP**⁻ (**10**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms except the ones bound to H5 and N6 and a co-crystallized solvent molecule are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.937(6), Cu–N3 2.038(5), Cu–N4 2.035(4), N1–C1 1.350(7), N2–C1 1.354(7), C1–Cu–N3 128.50(19), C1–Cu–N4 132.12(19).

Crystal Data for [{(IDipp)Cu}₂(C₂N₃H₃)₂]²⁺**2FAP**⁻ (10): C₇₃H₈₁Cl₉Cu₂F₃₆N₁₀P₂, M_r = 2290.54, T = 100.00(2) K, $\lambda = 1.54184$ Å, colorless block, 0.130×0.190×0.260 mm³, monoclinic space group P2₁/m, a = 14.6881(2) Å, b = 21.2716(5) Å, c = 14.9560(3) Å, $\beta = 91.888(2)^{\circ}$, V = 4670.31(16) Å³, Z = 2, $\rho_{calcd} = 1.629$ g·cm⁻³, $\mu = 4.331$ mm⁻¹, F(000) = 2308, 48516 reflections, $-18 \le h \le 18$, $-26 \le k \le 26$, $-19 \le l \le 14$, 2.956° < θ < 80.331°, completeness 98.6%, 48516 independent reflections, 40756 reflections observed with [I > 2 σ (I)], 721 parameters, 654 restraints, *R* indices (all data) *R*₁ = 0.1142, *wR*₂ = 0.2842, final *R* indices [I > 2 σ (I)] *R*₁ = 0.1028, *wR*₂ = 0.2751, largest difference peak and hole 2.562 and -2.645 e Å⁻³, GooF = 1.041.



Figure S7. Molecular structure of [{(cAAC^{Me})Cu}₂{C₆H₁₂N₂}]²⁺**2FAP⁻** (**11**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.900(4), Cu-N1 1.927(3), N2-C1 1.306(5); C1-Cu-N1 176.68(15).

Crystal Data for [{(cAAC^{Me})Cu}₂{C₆H₁₂N₂}]²⁺2FAP⁻ (11): $C_{58}H_{74}Cu_2F_{36}N_4P_2$, $M_r = 1700.23$, T = 100.00(2) K, $\lambda = 0.71073$ Å, colorless block, $0.120 \times 0.180 \times 0.330$ mm³, monoclinic space group $P2_1/n$, a = 13.8496(4) Å, b = 17.3310(5) Å, c = 14.6695(4) Å, $\beta = 101.873(2)^\circ$, V = 3445.75(17) Å³, Z = 2, $\rho_{calcd} = 1.639$ g·cm⁻³, $\mu = 0.802$ mm⁻¹, F(000) = 1724, 41410 reflections, $-17 \le h \le 16$, $-25 \le k \le 21$, $-20 \le I \le 18$, 2.185° < θ < 31.270°, completeness 80.2%, 9022 independent reflections, 6872 reflections observed with [I > 2 σ (I)], 931 parameters, 2055 restraints, *R* indices (all data) $R_1 = 0.0566$, $wR_2 = 0.0948$, final *R* indices [I > 2 σ (I)] $R_1 = 0.0367$, $wR_2 = 0.0860$, largest difference peak and hole 0.471 and -0.322 e Å⁻³, GooF = 1.036.



Figure S8. Molecular structure $[{(cAAC^{Me})Cu}_2{C_4H_4N_2}]^2+2FAP^-$ (12) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.890(3), Cu-N1 1.906(2), N2-C1 1.299(4); C1-Cu-N1 170.45(12).

Crystal Data for [{(cAAC^{Me})Cu}₂{C₄H₄N₂}]²⁺2FAP⁻ (12): C₅₆H₆₆Cu₂F₃₆N₄P₂, M_r = 1668.14, T = 100.00(2) K, $\lambda = 1.54184$ Å, yellow block, 0.090×0.160×0.220 mm³, monoclinic space group P2₁/n, a = 13.98110(10) Å, b = 17.2021(2) Å, c = 14.09990(10) Å, $\beta = 102.6220(10)^{\circ}$, V = 3309.13(5) Å³, Z = 2, $\rho_{calcd} = 1.674$ g·cm⁻³, $\mu = 2.562$ mm⁻¹, F(000) = 1684, 34985 reflections, $-11 \le h \le 17$, $-21 \le k \le 21$, $-17 \le I \le 18$, 4.034° < θ < 79.988°, completeness 97.8%, 7062 independent reflections, 6386 reflections observed with [I > 2 σ (I)], 685 parameters, 720 restraints, *R* indices (all data) *R*₁ = 0.0666, *wR*₂ = 0.1852, final *R* indices [I > 2 σ (I)] *R*₁ = 0.0625, *wR*₂ = 0.0625, largest difference peak and hole 1.232 and -0.425 e Å⁻³, GooF = 1.058.



Figure S9. Molecular structure of $[(IDipp)Cu(N\equiv CMe)_2]^*FAP^-$ (16) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.909(2), Cu–N1 1.967(2), Cu–N2 1.938(2), N3–C1 1.360(3), N4–C1 1.358(3), N1–C2 1.133(3), N2–C3 1.123(4); C1–Cu–N1 121.11(9), C1–Cu–N2 131.08(10), N1–Cu–N2 107.71(9).

Crystal Data for [(IDipp)Cu(N=CMe)_2]***FAP**⁻ **(16):** $C_{37}H_{42}CuF_{18}N_4P$, $M_r = 979.25$, T = 100.00(10) K, $\lambda = 1.54184$ Å, colorless block, $0.116 \times 0.146 \times 0.249$ mm³, monoclinic space group $P2_1/c$, a = 12.60010(10) Å, b = 22.0470(2) Å, c = 16.4358(2) Å, $\beta = 106.5990(10)^\circ$, V = 4375.51(8) Å³, Z = 4, $\rho_{calcd} = 1.487$ g·cm⁻³, $\mu = 2.044$ mm⁻¹, F(000) = 1992, 45059 reflections, $-15 \le h \le 15$, $-26 \le k \le 27$, $-19 \le I \le 20$, $3.449^\circ < \theta < 74.500^\circ$, completeness 99.3%, 8901 independent reflections, 7750 reflections observed with $[I > 2\sigma(I)]$, 560, parameters, 0 restraints, *R* indices (all data) $R_1 = 0.0602$, $wR_2 = 0.1113$, final *R* indices $[I > 2\sigma(I)]$ $R_1 = 0.0503$, $wR_2 = 0.1064$, largest difference peak and hole 0.478 and $-0.501 = Å^{-3}$, GooF = 0.928.



Figure S10. Molecular structure of [(IDipp)Cu(N≡CPh)]***FAP**⁻ (**17**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu−C1 1.8848(15), Cu−N1 1.8453(14), N2–C1 1.350(2), N3–C1 1.3503(19), N1–C2 1.141(2); C1–Cu−N1 178.03(7), Cu−N1–C2 176.85(15).

Crystal Data for [(IDipp)Cu(N=CPh)]⁺**FAP**⁻ **(17):** $C_{40}H_{41}CuF_{18}N_3P$, $M_r = 1000.27$, T = 99.99(10) K, $\lambda = 1.54184$ Å, colorless block, $0.178 \times 0.203 \times 0.294$ mm³, monoclinic space group $P2_1/c$, a = 13.9829(2) Å, b = 17.2015(3) Å, c = 18.1556(3) Å, $\beta = 96.7310(10)^\circ$, V = 4336.81(12) Å³, Z = 4, $\rho_{calcd} = 1.532$ g·cm⁻³, $\mu = 2.071$ mm⁻¹, F(000) = 2032, 32423 reflections, $-17 \le h \le 16$, $-21 \le k \le 19$,

 $-21 \le l \le 22, 3.551^{\circ} < \theta < 74.494^{\circ}$, completeness 98.0%, 8700 independent reflections, 7608 reflections observed with $[I > 2\sigma(I)]$, 576 parameters, 0 restraints, *R* indices (all data) $R_1 = 0.0378$, $wR_2 = 0.0865$, final *R* indices $[I > 2\sigma(I)]$ $R_1 = 0.0320$, $wR_2 = 0.0835$, largest difference peak and hole 0.399 and $-0.559 = Å^{-3}$, GooF = 1.072.



Figure S11. Molecular structure of [(SIDipp)Cu(NH₂Ph)]***FAP**⁻ (**19**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms except these of the nitrogen atom of the aniline ligand and a co-crystallized solvent molecule are omitted for clarity. Only one of two independent molecules in the asymmetric unit is shown. Selected bond lengths [Å] and angles [°] (average out of two independent molecules in the asymmetric unit): Cu–C1 1.893(3), Cu–N1 1.935(3), N2–C1 1.331(4), N3–C1 1.335(4), N1–C2 1.452(4); C1–Cu–N1 171.88(13), Cu–N1–C2 110.9(2).

Crystal Data for [(SIDipp)Cu(NH₂Ph)]⁺**FAP**⁻ **(19):** $C_{85}H_{98}Cu_2F_{36}N_6P_2$, $M_r = 2076.71$, T = 99.98(12) K, $\lambda = 1.54184$ Å, colorless block, $0.220 \times 0.111 \times 0.080$ mm³, monoclinic space group $P2_1/c$, a = 10.93360(10) Å, b = 43.2109(5) Å, c = 19.8400(2) Å, $\beta = 98.0870(10)^\circ$, V = 9280.21(17) Å³, Z = 4, $\rho_{calcd} = 1.486$ Mg/m³, $\mu = 1.956$ mm⁻¹, F(000) = 4248, 67731 reflections, $-13 \le h \le 13$, $-54 \le k \le 53$, $-24 \le I \le 15$, $2.471^\circ < \theta < 74.499^\circ$, completeness 97.6%, 18532 independent reflections, 15520 reflections observed with [I > 2 σ (I)], 1714 parameters, 3349 restraints, *R* indices (all data) $R_1 = 0.0650$, $wR_2 = 0.1577$, final *R* indices [I > 2σ (I)] $R_1 = 0.0770$, $wR_2 = 0.1653$, largest difference peak and hole 1.153 and -0.486 e Å⁻³, GooF = 1.051.



Figure S12. Molecular structure of [(cAAC^{Me})Cu(NH₂Ph)]***FAP**⁻ (**20**) in the solid state (ellipsoids set at the 50% probability level; the Dipp substituent is shown as a wire-and-stick model). Hydrogen atoms except these of the nitrogen atom of the aniline ligand are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.882(3), Cu-N1 1.942(3), N1-C3 1.457(4); C1-Cu-N1 175.47(12), Cu-N1-C3 109.99(18).

Crystal Data for [(cAAC^{Me})Cu(NH₂Ph)]⁺**FAP**⁻ **(20):** $C_{32}H_{38}CuF_{18}N_2P$, $M_r = 887.15$, T = 99.98(10) K, $\lambda = 1.54184$ Å, colorless block, $0.092 \times 0.172 \times 0.215$ mm³, monoclinic space group $P2_1/c$, a = 13.91300(10) Å, b = 15.00150(10) Å, c = 17.89340(10) Å, $\beta = 96.1210(10)^\circ$, V = 3713.34(4) Å³, Z = 4, $\rho_{calcd} = 1.587$ g·cm⁻³, $\mu = 2.323$ mm⁻¹, F(000) = 1800, 40061 reflections, $-17 \le h \le 17$, $-18 \le k \le 18$, $-22 \le I \le 16$, $3.833^\circ < \theta < 74.503^\circ$, completeness 100%, 7610 independent reflections, 6946 reflections observed with [I > 2σ (I)], 495 parameters, 0 restraints, *R* indices (all data) $R_1 = 0.0652$, $wR_2 = 0.1750$, final *R* indices [I > 2σ (I)] $R_1 = 0.0614$, $wR_2 = 0.1705$, largest difference peak and hole 2.251 and -0.847 e Å⁻³, GooF = 1.034.



Figure S13. Molecular structure of [(IDipp)Cu(NHPh₂)]***FAP**⁻ (**21**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms except these of the nitrogen atom of the diphenylamine ligand are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.8716(17), Cu–N1 1.9492(15), N2–C1 1.358(2), N3–C1 1.353(2), N1–C2 1.471(2), N1–C2 1.471(2), N1–C3 1.461(2); C1–Cu–N1 175.97(7), Cu–N1–C2 113.95(11), Cu–N1–C3 95.36(10), C2–N1–C3 118.09(4).

Crystal Data for [(IDipp)Cu(NHPh₂)]⁺**FAP**⁻ **(21)**: $C_{45}H_{47}CuF_{18}N_3P$, $M_r = 1066.36$, T = 99.9(5) K, $\lambda = 1.54184$ Å, colorless block, $0.063 \times 0.090 \times 0.189$ mm³, monoclinic space group $P2_1/c$, a = 12.75280(10) Å, b = 21.8800(2) Å, c = 16.76400(10) Å, $\beta = 96.4180(10)^\circ$, V = 4648.36(6) Å³, Z = 4, $\rho_{calcd} = 1.524$ g·cm⁻³, $\mu = 1.971$ mm⁻¹, F(000) = 2176, 69991 reflections, $-14 \le h \le 15$, $-27 \le k \le 26$, $-20 \le I \le 20$, $3.334^\circ < \theta < 74.498^\circ$, completeness 99.8%, 9467 independent reflections, 8483 reflections observed with $[I > 2\sigma(I)]$, 847 parameters, 1182 restraints, R indices (all data) R₁ = 0.0416, wR₂ = 0.0996, final R indices $[I > 2\sigma(I)]$ R₁ = 0.0375, wR₂ = 0.0970, largest difference peak and hole 0.848 and -0.397 e Å⁻³, GooF = 1.049.



Figure S14. Molecular structure of $[(cAAC^{Me})Cu(NC_5H_5)]^*FAP^-$ (23) in the solid state (ellipsoids set at the 50% probability level; the Dipp substituent is shown as a wire-and-stick model). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu-C1 1.8853(17), Cu-N1 1.9038(15), N2-C1 1.300(2); C1-Cu-N1 174.25(6).

Crystal Data for [(cAAC^{Me})Cu(NC₅H₅)]***FAP**⁻ **(23)**: $C_{31}H_{36}CuF_{18}N_2P$, $M_r = 873.13$, T = 99.9(3) K, $\lambda = 1.54184$ Å, colorless block, $0.124 \times 0.198 \times 0.211$ mm³, monoclinic space group $P2_1/c$, a = 14.05780(10) Å, b = 16.1196(2) Å, c = 15.84600(10) Å, $\beta = 94.2150(10)^\circ$, V = 3581.09(6) Å³, Z = 4, $\rho_{calcd} = 1.619$ g·cm⁻³, $\mu = 2.398$ mm⁻¹, F(000) = 1768, 52927 reflections, $-17 \le h \le 17$, $-20 \le k \le 20$, $-19 \le I \le 17$, $3.917^\circ < \theta < 74.504^\circ$, completeness 99.8%, 7311 independent reflections, 6579 reflections observed with $[I > 2\sigma(I)]$, 723 parameters, 1341 restraints, *R* indices (all data) $R_1 = 0.0382$, $wR_2 = 0.0946$, final *R* indices $[I > 2\sigma(I)]$ $R_1 = 0.0345$, $wR_2 = 0.0916$, largest difference peak and hole 0.378 and -0.277 e Å⁻³, GooF = 1.052.



Figure S15. Molecular structure of $[(IDipp)Cu(NC_5H_3F_2)]^*FAP^-$ (24) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.878(2), Cu–N1 1.916(2), N2–C1 1.357(3), N3–C1 1.351(3); C1–Cu–N1 172.2(1).

Crystal Data for [(IDipp)Cu(NC₅H₃F₂)]⁺**FAP**⁻ **(24):** C₃₈H₃₉CuF₂₀N₃P, M_r = 1012.23, T = 99.99(10) K, $\lambda = 1.54184$ Å, colorless block, 0.162×0.234×0.298 mm³, orthorhombic space group *Pbca*, a = 18.63510(10) Å, b = 20.20240(10) Å, c = 23.25420(10) Å, V = 8754.60(7) Å³, Z = 8, $\rho_{calcd} = 1.536$ g·cm⁻³, $\mu = 2.126$ mm⁻¹, F(000) = 4096, 89464 reflections, $-23 \le h \le 23$, $-24 \le k \le 25$, $-25 \le l \le 29$, $3.745^{\circ} < \theta < 74.503^{\circ}$, completeness 100%, 8953 independent reflections, 8252 reflections

observed with $[I > 2\sigma(I)]$, 802 parameters, 1293 restraints, *R* indices (all data) $R_1 = 0.0559$, $wR_2 = 0.1470$, final *R* indices $[I > 2\sigma(I)]$ $R_1 = 0.0529$, $wR_2 = 0.1442$, largest difference peak and hole 0.922 and -0.488 e Å⁻³, GooF = 1.032.



Figure S16. Molecular structure of [{(IDipp)Cu(μ -ONC₅H₅)}₂]²⁺**2FAP**⁻ (**26**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms and two co-crystallized solvent molecules are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu1–C1 1.873(3), Cu2–C2 1.876(3), Cu1–O1 2.054(2), Cu2–O1 2.027(2), Cu1–O2 2.028(2), Cu2–O2 2.070(2), Cu1–Cu2 3.3117(8), O1–N1 1.357(4), O2–N2 1.348(3); C1–Cu1–O1 142.16(11), C1–Cu1–O2 145.85(11), C2–Cu2–O1 144.71(12), C2–Cu2–O2 143.59(11), Cu1–O1–Cu2 108.47(11), Cu1–O2–Cu2 107.84(10).

Crystal Data for [{(IDipp)Cu(\mu-ONC₅H₅)}₂]²⁺2FAP⁻ (26): C₈₈H₉₀Cu₂F₄₀N₆O₂P₂, M_r = 2212.67, T = 99.98(11) K, \lambda = 1.54184 Å, colorless block, 0.143×0.209×0.424 mm³, triclinic space group P^{1;-}, a = 15.15180(10) Å, b = 15.53180(10) Å, c = 22.82480(10) Å, \alpha = 90.2190(10)°, \beta = 103.7860(10)°, \gamma = 112.7070(10)°, V = 4784.79(6) Å³, Z = 2, \rho_{calcd} = 1.536 Mg/m³, \mu = 2.016 mm⁻¹, F(000) = 2248, 197077 reflections, -18 \le h \le 18, -19 \le k \le 19, -28 \le l \le 25, 3.102^{\circ} < \theta < 74.503^{\circ}, completeness 100%, 19494 independent reflections, 17389 reflections observed with [I > 2\sigma(I)], 1520 parameters, 1830 restraints, *R* **indices (all data) R_1 = 0.0829, wR_2 = 0.2530, final** *R* **indices [I > 2\sigma(I)] R_1 = 0.0778, wR_2 = 0.2293, largest difference peak and hole 1.881 and -1.060 e Å⁻³, GooF = 1.069.**



Figure S17. Molecular structure of $[(IDipp)Cu(\eta^{1}-O=CPh_{2})]$ ***FAP**⁻ (**27**) in the solid state (ellipsoids set at the 50% probability level; Dipp substituents are shown as wire-and-stick models). Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Cu–C1 1.866(3), Cu–O 1.8624(19), N1–C1 1.352(3), N2–C1 1.359(2), O–C2 1.246(3); C1–Cu–O 167.61(9), Cu–O–C2 133.48(18), C3–C2–C4 123.0(2), O–C2–C3 119.1(2), O–C2–C4 117.9(2).

Crystal Data for [(IDipp)Cu(η^{1} **-O=CPh**₂**)]**⁺**FAP**⁻ (**27):** C₄₆H₄₆CuF₁₈N₂OP, M_r = 1079.36, T = 100.00(10) K, $\lambda = 1.54184 \text{ Å}$, colorless block, 0.179×0.206×0.225 mm³, monoclinic space group $P2_1/c$, a = 14.08110(10) Å, b = 19.34660(10) Å, c = 17.91920(10) Å, $\beta = 96.8600(10)^\circ$, V = 4846.63(5) Å³, Z = 4, $\rho_{calcd} = 1.479 \text{ g} \cdot \text{cm}^{-3}$, $\mu = 1.909 \text{ mm}^{-1}$, F(000) = 2200, 73441 reflections, $-17 \le h \le 14$, $-24 \le k \le 24$, $-22 \le I \le 21$, 3.375° < θ < 74.499°, completeness 99.7%, 9877 independent reflections, 8944 reflections observed with [I > 2 σ (I)], 688 parameters, 282 restraints, *R* indices (all data) $R_1 = 0.0609$, $wR_2 = 0.1517$, final *R* indices [I > 2 σ (I)] $R_1 = 0.0565$, $wR_2 = 0.1478$, largest difference peak and hole 1.371 and $-1.070 \in Å^{-3}$, GooF = 1.044.

2) NMR spectra of compounds





Figure S19. $^{13}\text{C}\{^{1}\text{H}\}$ NMR spectrum (125.8 MHz) of 1 recorded in CDCl_3.



Figure S20. $^{19}\mathsf{F}$ NMR spectrum (470.5 MHz) of 1 recorded in CDCl_3.



Figure S21. ³¹P NMR spectrum (202.4 MHz) of 1 recorded in CDCl₃.



Figure S22. ¹H NMR spectrum (500.1 MHz) of 2 recorded in CDCl₃.



Figure S23. ¹³C{¹H} NMR spectrum (125.8 MHz) of 2 recorded in CDCl₃.



Figure S24. ¹⁹F NMR spectrum (470.5 MHz) of 2 recorded in CDCl₃.



Figure S25. ³¹P NMR spectrum (202.4 MHz) of 2 recorded in CDCl₃.



Figure S26. 1 H NMR spectrum (500.1 MHz) of 3 recorded in CDCl₃.



Figure S27. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (125.8 MHz) of 3 recorded in CDCl_3.



Figure S28. ¹⁹F NMR spectrum (470.5 MHz) of 3 recorded in CDCl₃.



Figure S29. ³¹P NMR spectrum (202.4 MHz) of 3 recorded in CDCl₃.







Figure S31. ¹³C{¹H} NMR spectrum (125.8 MHz) of 4 recorded in CDCl₃.



Figure S32. ¹⁹F NMR spectrum (470.5 MHz) of 4 recorded in CDCl₃.



Figure S33. ³¹P NMR spectrum (202.4 MHz) of 4 recorded in CDCl₃.



Figure S34. ¹H NMR spectrum (500.1 MHz) of 5 recorded in CDCl₃.



Figure S35. ¹³C{¹H} NMR spectrum (125.8 MHz) of 5 recorded in CDCl₃ (X: signals for acridine ligand).



Figure S36. ¹⁹F NMR spectrum (470.5 MHz) of 5 recorded in CDCl₃.



Figure S37. ³¹P NMR spectrum (202.4 MHz) of 5 recorded in CDCl₃.







Figure S39. $^{13}C{^1H}$ NMR spectrum (125.8 MHz) of 6 recorded in CDCl₃ (X: signals for acridone ligand).



Figure S40. ¹⁹F NMR spectrum (470.5 MHz) of 6 recorded in CDCl₃.



Figure S41. ³¹P NMR spectrum (202.4 MHz) of 6 recorded in CDCl₃.



Figure S42. ¹H NMR spectrum (600.2 MHz) of 7 recorded in CDCl₃.



Figure 43. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (100.6 MHz) of 7 recorded in CDCl_3.



Figure S44. ¹⁹F NMR spectrum (470.5 MHz) of 7 recorded in CDCl₃.



Figure S45. ³¹P NMR spectrum (202.4 MHz) of 7 recorded in CDCl₃.



Figure S46. ¹H NMR spectrum (500.1 MHz) of 8 recorded in CDCl₃.



Figure S47. ¹³C{¹H} NMR spectrum (100.6 MHz) of 8 recorded in CDCl_{3.}



Figure S48. ¹⁹F NMR spectrum (470.5 MHz) of 8 recorded in CDCl₃.



Figure S49. ³¹P NMR spectrum (202.4 MHz) of 8 recorded in CDCl₃.



Figure S50. ¹H NMR spectrum (500.1 MHz) of 9 recorded in CDCl₃.



Figure S51. ${}^{13}C{}^{1}H$ NMR spectrum (100.6 MHz) of 9 recorded in CDCl_{3.}



Figure S52. ¹⁹F NMR spectrum (470.5 MHz) of **9** recorded in CDCl₃.



Figure S53. $^{\rm 31}\text{P}$ NMR spectrum (202.4 MHz) of 9 recorded in CDCl_3.



Figure S54. ¹H NMR spectrum (400.1 MHz) of 10 recorded in CDCl₃.



Figure S55. ¹³C{¹H} NMR spectrum (125.8 MHz) of **10** recorded in CDCl_{3.}



Figure S56. ^{19}F NMR spectrum (470.5 MHz) of 10 recorded in CDCl_3.



Figure 57. ³¹P NMR spectrum (202.4 MHz) of 10 recorded in CDCl₃.



Figure S58. ¹H NMR spectrum (500.1 MHz) of 11 recorded in CD₂Cl₂.



gure S59. $^{13}\text{C}\{^{1}\text{H}\}$ NMR spectrum (125.8 MHz) of 11 recorded in CD_2Cl_2.



Figure S60. ¹⁹F NMR spectrum (470.5 MHz) of **11** recorded in CD₂Cl₂.



Figure S61. $^{\rm 31}P$ NMR spectrum (202.4 MHz) of 11 recorded in CD_2Cl_2.



Figure S62. 1 H NMR spectrum (500.1 MHz) of 12 recorded in CD₂Cl₂.



Figure S63. $^{13}\text{C}\{^{1}\text{H}\}$ NMR spectrum (125.8 MHz) of 12 recorded in CD_2Cl_2.



Figure S64. ^{19}F NMR spectrum (470.5 MHz) of 12 recorded in $\text{CD}_2\text{Cl}_2.$



Figure S65. $^{\rm 31}P$ NMR spectrum (202.4 MHz) of 12 recorded in $CD_2Cl_2.$



Figure S66. ¹H NMR spectrum (500.1 MHz) of 13 recorded in CDCl₃.



Figure S67. ¹³C{¹H} NMR spectrum (125.8 MHz) of 13 recorded in CDCl₃.



Figure S68. ¹⁹F NMR spectrum (470.5 MHz) of **13** recorded in CDCl₃.



Figure S69. ³¹P NMR spectrum (202.4 MHz) of 13 recorded in CDCl₃.



Figure S70. ¹H NMR spectrum (400.1 MHz) of 14 recorded in CDCl₃.



Figure S71. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum (100.6 MHz) of 14 recorded in CDCl_3.



Figure S72. ¹⁹F NMR spectrum (470.5 MHz) of 14 recorded in CDCl₃.



Figure S73. $^{\rm 31}P$ NMR spectrum (202.4 MHz) of 14 recorded in CDCl_3.



Figure S74. ¹H NMR spectrum (500.1 MHz) of 15 recorded in CDCl₃.



Figure S75. ¹³C{¹H} NMR spectrum (125.8 MHz) of 15 recorded in CDCl₃.



Figure S76. ¹⁹F NMR spectrum (470.5 MHz) of **15** recorded in CDCl₃.



Figure S77. ³¹P NMR spectrum (202.4 MHz) of 15 recorded in CDCl₃.



Figure S78. ¹H NMR spectrum (500.1 MHz) of 16 recorded in CDCl₃.



Figure S79. $^{13}C{^{1}H}$ NMR spectrum (125.8 MHz) of 16 recorded in CD₂Cl₂.



Figure S80. ¹⁹F NMR spectrum (470.5 MHz) of 16 recorded in CD₂Cl₂.



Figure S81. ^{31}P NMR spectrum (202.4 MHz) of 16 recorded in CD_2Cl_2.



Figure S82. ¹H NMR spectrum (500.1 MHz) of 17 recorded in CDCl₃.



Figure S83. $^{13}C\{^{1}H\}$ NMR spectrum (125.8 MHz) of 17 recorded in CDCl_3.



Figure S84. ¹⁹F NMR spectrum (470.5 MHz) of **17** recorded in CDCl₃.



Figure S85. ³¹P NMR spectrum (202.4 MHz) of 17 recorded in CDCl₃.



Figure S86. ¹H NMR spectrum (500.1 MHz) of 18 recorded in CD₂Cl₂.



Figure S87. $^{13}C{^{1}H}$ NMR spectrum (125.8 MHz) of 18 recorded in CD₂Cl₂.



Figure S88. ^{19}F NMR spectrum (470.5 MHz) of 18 recorded in $\text{CD}_2\text{Cl}_2.$



Figure S89. ³¹P NMR spectrum (202.4 MHz) of 18 recorded in CD₂Cl₂.



Figure S90. ^1H NMR spectrum (500.1 MHz) of 19 recorded in CD_2Cl_2.



Figure S91. ${}^{13}C{}^{1}H$ NMR spectrum (125.8 MHz) of 19 recorded in CD₂Cl₂.



Figure S92. ¹⁹F NMR spectrum (470.5 MHz) of **19** recorded in CD₂Cl₂.



Figure S93. ³¹P NMR spectrum (202.4 MHz) of **19** recorded in CD₂Cl₂.



Figure S94. ¹H NMR spectrum (500.1 MHz) of 20 recorded in CD₂Cl₂.



Figure S95. $^{13}\text{C}\{^{1}\text{H}\}$ NMR spectrum (125.8 MHz) of 20 recorded in CD_2Cl_2.



Figure S96. ¹⁹F NMR spectrum (470.5 MHz) of **20** recorded in CD₂Cl₂.



Figure S97. ^{31}P NMR spectrum (202.4 MHz) of 20 recorded in CD₂Cl₂.



Figure S98. ¹H NMR spectrum (500.1 MHz) of 21 recorded in CDCl₃.



Figure S99. ¹³C{¹H} NMR spectrum (125.8 MHz) of 21 recorded in CDCl₃.



Figure S100. ¹⁹F NMR spectrum (470.5 MHz) of 21 recorded in CDCl₃.



Figure S101. ³¹P NMR spectrum (202.4 MHz) of 21 recorded in CDCl₃.



Figure S102. ¹H NMR spectrum (500.1 MHz) of 22 recorded in CDCl₃.



Figure S103. $^{13}\text{C}\{^{1}\text{H}\}$ NMR spectrum (125.8 MHz) of 22 recorded in CDCl_3.



Figure S104. ¹⁹F NMR spectrum (470.5 MHz) of 22 recorded in CDCl₃.



Figure S105. ³¹P NMR spectrum (202.4 MHz) of 22 recorded in CDCl₃.



Figure S106. ¹H NMR spectrum (500.1 MHz) of 23 recorded in CDCl₃.



Figure S107. ¹³C{¹H} NMR spectrum (125.8 MHz) of 23 recorded in CDCl₃.



Figure S108. ¹⁹F NMR spectrum (470.5 MHz) of 23 recorded in CDCl₃.



Figure S109. ³¹P NMR spectrum (202.4 MHz) of 23 recorded in CDCl₃.



Figure S110. ¹H NMR spectrum (500.1 MHz) of 24 recorded in CDCl₃.



Figure S111. ¹³C{¹H} NMR spectrum (125.8 MHz) of 24 recorded in CDCl₃.



Figure S112. ¹⁹F NMR spectrum (470.5 MHz) of 24 recorded in CDCl₃.



Figure S113. ³¹P NMR spectrum (202.4 MHz) of 24 recorded in CDCl₃.



Figure S114. ¹H NMR spectrum (500.1 MHz) of 25 recorded in CDCl₃.



Figure S115. ¹³C{¹H} NMR spectrum (125.8 MHz) of 25 recorded in CDCl₃.



Figure S116. ¹⁹F NMR spectrum (470.5 MHz) of 25 recorded in CDCl₃.



Figure S117. ³¹P NMR spectrum (202.4 MHz) of 25 recorded in CDCl₃.



Figure S118. ¹H NMR spectrum (400.1 MHz, 236.5 K) of 26 recorded in CDCl₃.



Figure S119. $^{13}C{^{1}H}$ NMR spectrum (100.6 MHz, 236.5 K) of 26 recorded in CDCl₃.



Figure S120. ^{19}F NMR spectrum (470.5 MHz) of 26 recorded in CD_2Cl_2.



Figure S121. ³¹P NMR spectrum (202.4 MHz) of 26 recorded in CD₂Cl₂.



Figure S122. ¹H NMR spectrum (500.1 MHz) of 27 recorded in CDCl₃.



Figure S123. ¹³C{¹H} NMR spectrum (125.8 MHz) of 27 recorded in CDCl₃.



Figure S124. ¹⁹F NMR spectrum (470.5 MHz) of 27 recorded in CDCl₃.



Figure S125. $^{\rm 31}P$ NMR spectrum (202.4 MHz) of 27 recorded in CDCl_3.



Figure S126. ¹H NMR spectrum (500.1 MHz) of 28 recorded in CDCl₃.



Figure S127. $^{13}C{^{1}H}$ NMR spectrum (125.8 MHz) of 28 recorded in CDCl₃.



Figure S128. ¹⁹F NMR spectrum (470.5 MHz) of 28 recorded in CDCl₃.



Figure S129. ³¹P NMR spectrum (202.4 MHz) of 28 recorded in CDCl₃.

References

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