

Supplementary Information

for

Modelling the Binding of Cytotoxic Dinuclear Nickel

Complexes to two Neighboring Phosphate Esters of DNA

Using Dicarboxylate Ligands

Thomas Riediger,^a Maximilian Böhm,^a Michael Kapiza,^a Anja Stammeler,^a Jan Oldengott,^a and Thorsten Glaser^{*,a}

^a *Lehrstuhl für Anorganische Chemie I, Fakultät für Chemie, Universität Bielefeld, Universitätsstr. 25, D-33615 Bielefeld, Germany*

^{*} *To whom correspondence should be addressed. E-mail: thorsten.glaser@uni-bielefeld.de (T.G.)*

Table S1. Crystal data and refinement parameters.

	[(Htom ^{6-Me}){Ni ^{II} (μ -succ)Ni ^{II} }]-(BPh ₄)•2CH ₃ CN	[(Htom ^{6-Me}){Ni ^{II} (μ -glut)Ni ^{II} }]-(BPh ₄)•2CH ₃ CN
Empirical formula	C ₇₂ H ₇₁ BN ₈ Ni ₂ O ₆	C ₇₃ H ₇₃ BN ₈ Ni ₂ O ₆
Formula weight	1272.59	1286.62
Crystal system	monoclinic	monoclinic
Space group	<i>P</i> 2 ₁ / <i>c</i>	<i>P</i> 2 ₁ / <i>c</i>
<i>a</i> [Å]	18.6047(11)	18.6053(8)
<i>b</i> [Å]	24.1873(15)	24.5063(10)
<i>c</i> [Å]	14.9384(9)	14.9770(7)
α [°]	90	90
β [°]	113.012(2)	112.945(2)
γ [°]	90	90
<i>V</i> [Å ³]	6187.3(7)	6288.4(5)
<i>Z</i>	4	4
ρ [g cm ⁻³]	1.366	1.359
μ [mm ⁻¹]	0.670	0.660
<i>F</i> (000)	2672.0	2704.0
Crystal size [mm ³]	0.37 × 0.19 × 0.03	0.42 × 0.28 × 0.23
Radiation	MoK α (λ = 0.71073 Å)	MoK α (λ = 0.71073 Å)
2 θ range [°]	2.91 to 60.39	3.32 to 66.69
	-26 ≤ <i>h</i> ≤ 26	-28 ≤ <i>h</i> ≤ 28
<i>hkl</i> ranges	-34 ≤ <i>k</i> ≤ 34	-37 ≤ <i>k</i> ≤ 37
	-21 ≤ <i>l</i> ≤ 21	-23 ≤ <i>l</i> ≤ 23
Collected refl.	231626	311779
Unique refl., <i>R</i> _{int}	18191, 0.0469	24173, 0.0427
Observed refl. (<i>I</i> > 2 σ (<i>I</i>))	14261	19402
Completeness	0.990	0.992
Data/restraints/param.	18191/0/812	24173/0/821
Goodness-of-fit on <i>F</i> ²	1.008	1.052
<i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2 σ (<i>I</i>))	0.0335, 0.0794	0.0410, 0.0999
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	0.0508, 0.0876	0.0571, 0.1102
Largest peak/hole [e Å ⁻³]	0.48/-0.34	1.97/-0.71
CCDC numbers	2445426	2445427

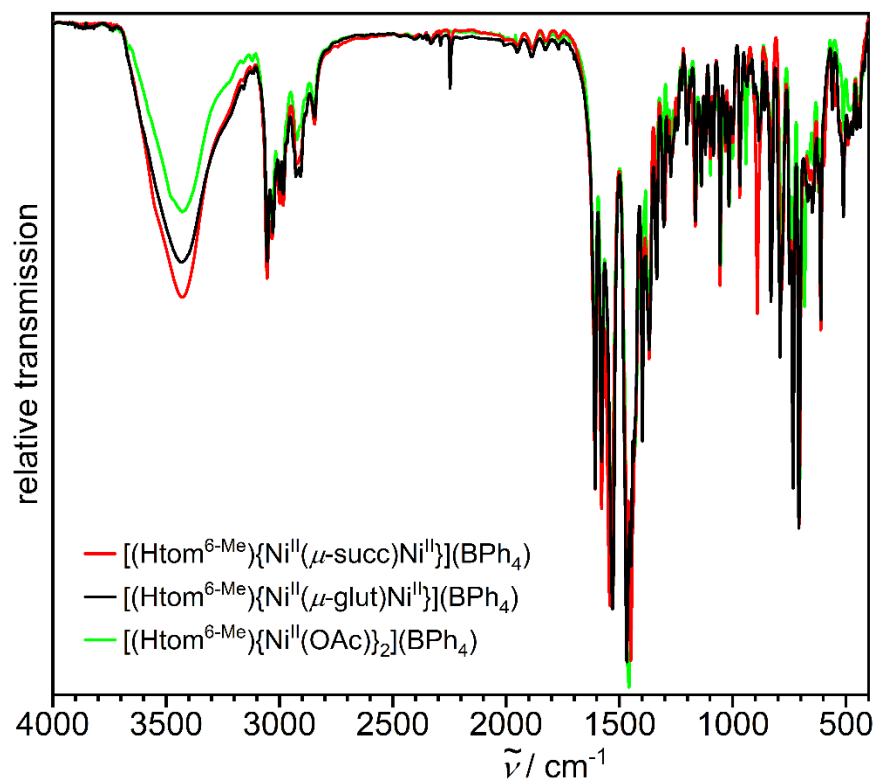


Figure S1. FTIR spectra of $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-succ})\text{Ni}^{\text{II}}\}](\text{BPh}_4)$, $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-glut})\text{Ni}^{\text{II}}\}](\text{BPh}_4)$, and $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\text{OAc})\}_2](\text{BPh}_4)$ for comparison.

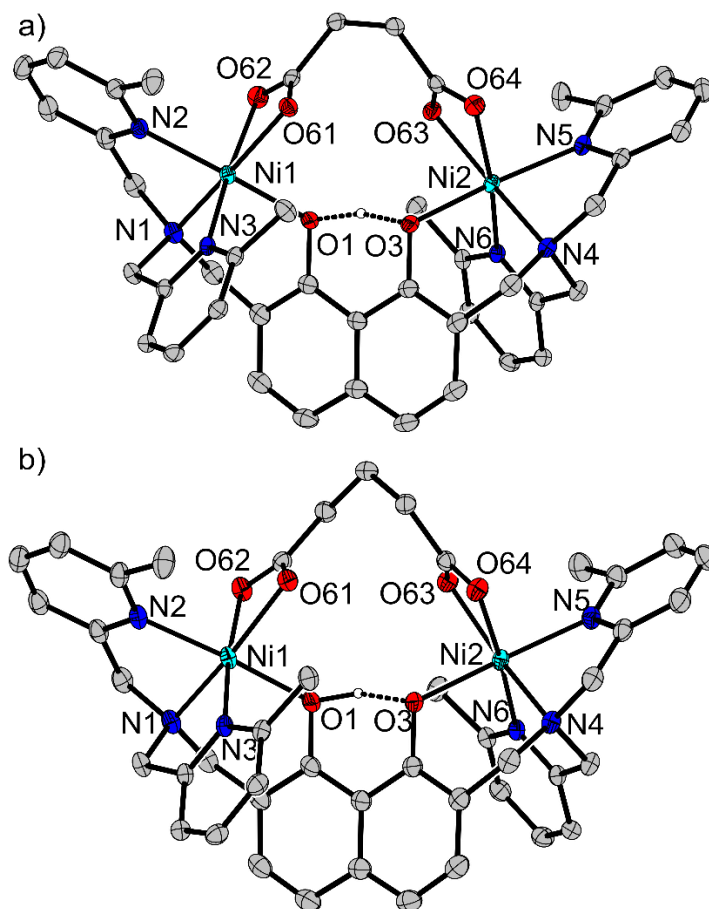


Figure S2. Thermal ellipsoid plots of a) $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-succ})\text{Ni}^{\text{II}}\}]^+$ in single-crystals of $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-succ})\text{Ni}^{\text{II}}\}](\text{BPh}_4)_2 \cdot 2\text{CH}_3\text{CN}$ and b) $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-glut})\text{Ni}^{\text{II}}\}]^+$ in single-crystals of $[(\text{Htom}^{6\text{-Me}})\{\text{Ni}^{\text{II}}(\mu\text{-glut})\text{Ni}^{\text{II}}\}](\text{BPh}_4)_2 \cdot 2\text{CH}_3\text{CN}$ at the 50% probability level. Hydrogen atoms except those coordinated at oxygen atoms omitted for clarity.