

Supporting Information

Stratified Sulfatometallates: Unlocking a New Dimension in Sulfatomolybdates

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Crystal Data and Structure Refinements

Table S1. *R*-factors of the structural refinements of Na₂[MoO₂(SO₄)₂] distinguished between satellites and main reflections determined using single-crystal diffraction at *T* = 293 K

	Na ₂ [MoO ₂ (SO ₄) ₂] Main reflections	Na ₂ [MoO ₂ (SO ₄) ₂] Satellites
<i>R</i> _{int}	0.0508	0.1355
<i>R</i> ₁ (all data)	0.0710	0.2413
<i>wR</i> ₂	0.0997	0.2124

Table S2. Atomic coordinates, Wyckoff symbols and isotropic displacement parameters *U*_{eq} / Å² in Na₂[MoO₂(SO₄)₂] at *T* = 293 K. Standard deviations are given in parentheses. For sodium cations Na6, Na7, Na8, Na10, Na11, Na12 and Na13, the isotropic displacement factor is given

Atom	Wyckoff symbol	S.O.F.	<i>x</i>	<i>y</i>	<i>z</i>	<i>U</i> _{eq}
Mo1	4 <i>e</i>		0.62710	0.09990	0.07070	0.02598(10)
Mo2	4 <i>e</i>		0.37090	0.40190	0.07390	0.02589(10)
S1	4 <i>e</i>		0.8706(1)	0.1494(1)	0.1428(1)	0.0259(3)
S2	4 <i>e</i>		0.1271(1)	0.3510(1)	0.1380(1)	0.0259(3)
S3	4 <i>e</i>		0.6231(1)	0.4135(1)	0.1043(1)	0.0279(3)
S4	4 <i>e</i>		0.3751(1)	0.0851(1)	0.1019(1)	0.0279(3)
Na1	4 <i>e</i>	0.557(11)	0.0022(6)	0.7505(5)	0.1291(8)	0.068(4)
Na2	4 <i>e</i>	0.570(9)	0.2411(6)	0.7177(3)	0.1754(8)	0.052(3)
Na3	4 <i>e</i>	0.729(9)	0.7570(2)	0.7842(2)	0.1800(3)	0.0421(11)
O41	4 <i>e</i>		0.4734(2)	0.1006(2)	0.9967(4)	0.0312(9)
O11	4 <i>e</i>		0.8578(3)	0.2274(2)	0.0668(4)	0.0448(11)
O21	4 <i>e</i>		0.1419(3)	0.2740(2)	0.0583(4)	0.0416(11)
O31	4 <i>e</i>		0.4759(2)	0.5997(2)	0.0025(4)	0.0333(9)
Na4	4 <i>e</i>	0.421(10)	0.0589(10)	0.0161(9)	0.9932(18)	0.018(3)
O42	4 <i>e</i>		0.3595(2)	0.9979(2)	0.1096(4)	0.0334(9)
Om21	4 <i>e</i>		0.3885(3)	0.4670(2)	0.2355(4)	0.0391(10)
Om11	4 <i>e</i>		0.6101(2)	0.0355(2)	0.2347(4)	0.0386(10)
O22	4 <i>e</i>		0.2111(2)	0.4087(2)	0.0719(3)	0.0276(8)
O32	4 <i>e</i>		0.6428(2)	0.4999(2)	0.1102(4)	0.0362(10)
Om12	4 <i>e</i>		0.6241(2)	0.1902(2)	0.1677(4)	0.0366(10)
Om22	4 <i>e</i>		0.3739(2)	0.3128(2)	0.1743(4)	0.0358(10)
O33	4 <i>e</i>		0.7082(3)	0.3721(3)	0.0227(5)	0.0596(14)
O12	4 <i>e</i>		0.7884(2)	0.0930(2)	0.0695(4)	0.0300(9)
O13	4 <i>e</i>		0.6528(2)	0.6571(2)	0.1711(3)	0.0329(9)
O23	4 <i>e</i>		0.3527(2)	0.8400(2)	0.1750(3)	0.0332(9)
O14	4 <i>e</i>		0.9704(2)	0.1114(2)	0.1110(4)	0.0413(11)
O43	4 <i>e</i>		0.2883(3)	0.1211(2)	0.0146(5)	0.0540(13)
O24	4 <i>e</i>		0.0280(2)	0.3886(2)	0.1048(4)	0.0465(12)
O34	4 <i>e</i>		0.6045(3)	0.3849(2)	0.2755(4)	0.0490(13)
Na5	4 <i>e</i>	0.428(11)	0.9278(12)	0.4814(8)	0.9909(14)	0.027(4)
O44	4 <i>e</i>		0.1076(3)	0.6147(2)	0.2259(5)	0.0539(13)

Na6	4e	0.402	0.3561(9)	0.6495(8)	0.3063(19)	0.0433(14)
Na7	4e	0.155	0.6353(19)	0.8436(16)	0.321(5)	0.079(7)
Na8	4e	0.237	0.8737(6)	0.4548(5)	1.0159(10)	0.0149(17)
Na9	4e	0.135	0.1168(16)	0.0334(13)	0.980(3)	-0.001(5)
Na10	2c	0.076	1	1/2	1	0.020(8)
Na11	4e	0.095	0.635(4)	0.794(3)	0.211(5)	0.052(7)
Na12	4e	0.111	0.1263(17)	0.080(2)	1.032(3)	0.007(4)
Na13	4e	0.134	0.6295(16)	0.8792(16)	0.370(4)	0.031(4)

Table S3. Anisotropic displacement parameters $U_{ij} / \text{\AA}^2$ in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$ at $T = 293$ K. Standard deviations are given in parentheses

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Mo1	0.02613(18)	0.0324(2)	0.01940(16)	-0.00387(13)	0.00189(12)	-0.00070(13)
Mo2	0.02584(18)	0.0323(2)	0.01953(17)	-0.00411(13)	-0.00167(12)	0.00031(12)
S1	0.0234(4)	0.0329(5)	0.0215(4)	-0.0006(4)	0.0007(3)	0.0033(4)
S2	0.0248(5)	0.0320(5)	0.0208(4)	-0.0014(4)	0.0005(3)	-0.0033(4)
S3	0.0206(4)	0.0310(5)	0.0320(5)	0.0027(4)	-0.0013(4)	0.0039(4)
S4	0.0217(4)	0.0306(5)	0.0314(5)	0.0031(4)	0.0007(4)	-0.0021(4)
Na1	0.075(6)	0.097(8)	0.032(4)	-0.057(5)	0.002(3)	-0.003(4)
Na2	0.083(6)	0.028(3)	0.046(4)	-0.013(3)	0.018(3)	-0.005(2)
Na3	0.057(2)	0.0297(16)	0.0392(18)	-0.0107(12)	-0.0054(13)	0.0018(11)
O41	0.0222(13)	0.0427(18)	0.0289(15)	-0.0065(12)	-0.0005(11)	0.0014(12)
O11	0.060(2)	0.0405(19)	0.0342(17)	-0.0107(16)	-0.0016(15)	0.0172(14)
O21	0.052(2)	0.0350(17)	0.0381(17)	-0.0099(15)	0.0042(15)	-0.0147(14)
O31	0.0273(15)	0.0460(19)	0.0266(14)	-0.0061(12)	0.0017(11)	-0.0022(12)
Na4	0.007(5)	0.019(4)	0.029(4)	-0.008(3)	-0.002(4)	0.001(3)
O42	0.0436(18)	0.0279(15)	0.0288(15)	-0.0033(13)	0.0041(12)	0.0064(12)
Om21	0.0414(17)	0.051(2)	0.0248(15)	-0.0106(15)	0.0011(12)	-0.0067(14)
Om11	0.0390(17)	0.052(2)	0.0251(15)	-0.0072(14)	-0.0013(12)	0.0055(14)
O22	0.0241(13)	0.0331(15)	0.0257(14)	0.0013(11)	-0.0016(11)	0.0011(11)
O32	0.0477(19)	0.0280(16)	0.0328(16)	-0.0124(13)	-0.0071(13)	-0.0006(12)
Om12	0.0323(16)	0.0440(18)	0.0334(16)	-0.0017(13)	0.0032(12)	-0.0079(13)
Om22	0.0298(15)	0.0433(19)	0.0345(16)	-0.0023(13)	-0.0002(12)	0.0095(13)
O33	0.039(2)	0.064(3)	0.076(3)	0.0200(18)	0.0010(18)	-0.020(2)
O12	0.0233(13)	0.0364(16)	0.0303(15)	0.0016(11)	0.0000(11)	-0.0073(12)
O13	0.0381(16)	0.0352(16)	0.0254(14)	-0.0035(13)	-0.0002(12)	-0.0001(12)
O23	0.0427(17)	0.0318(16)	0.0250(14)	-0.0026(13)	0.0000(12)	0.0030(12)
O14	0.0222(14)	0.060(2)	0.0417(18)	0.0058(14)	0.0031(13)	-0.0016(15)
O43	0.0332(18)	0.064(3)	0.064(2)	0.0058(17)	-0.0032(16)	0.020(2)
O24	0.0265(16)	0.081(3)	0.0326(17)	0.0084(16)	-0.0049(13)	-0.0009(16)
O34	0.048(2)	0.058(2)	0.041(2)	-0.0070(18)	-0.0090(16)	0.0167(17)
Na5	0.018(11)	0.034(5)	0.028(4)	-0.003(6)	-0.014(5)	-0.004(3)
O44	0.053(2)	0.074(3)	0.0350(18)	0.0039(19)	0.0102(16)	0.0157(17)
Na9	-0.009(7)	-0.026(9)	0.031(12)	-0.031(6)	-0.012(7)	0.010(8)

Table S4. Atomic coordinates, Wyckoff symbols and isotropic displacement parameters $U_{eq} / \text{\AA}^2$ in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$ at $T = 373$ K. Standard deviations are given in parentheses. For sodium cations Na3, Na4, Na7, Na8 and Na9, the isotropic displacement factor is given

Atom	Wyckoff symbol	S.O.F.	x	y	z	U_{eq}
Na1	8d	0.648(9)	0.7571(4)	0.3241(5)	0.7161(2)	0.648(9)
Na2	8d	0.294(7)	0.010(3)	0.3736(9)	0.7651(13)	0.294(7)
Na3	8d	0.171(12)	-0.0211(17)	0.4935(19)	0.5039(10)	0.171(12)
Na4	8d	0.063(13)	-0.069(7)	0.481(4)	0.519(2)	0.063(13)
Na5	8d	0.411(15)	-0.1216(8)	0.5041(9)	0.5405(5)	0.411(15)
Na6	8d	0.16(4)	0.649(2)	0.245(6)	0.671(4)	0.16(4)
Na7	8d	0.10(4)	0.6426(19)	0.222(5)	0.641(4)	0.10(4)
Na8	8d	0.073(16)	0.642(2)	0.145(8)	0.629(2)	0.073(16)
Na9	4a	0.01(2)	4	4	-4	0.01(2)
Mo1	8d		0.37180(3)	0.57341(4)	0.59877(2)	0.03086(16)
O1	8d		0.3892(3)	0.7345(4)	0.5339(2)	0.0476(9)
O2	8d		0.3757(2)	0.6724(5)	0.6875(2)	0.0456(8)
S1	8d		0.12908(8)	0.64180(13)	0.64971(7)	0.0334(3)
O11	8d		0.1494(2)	0.8276(4)	0.6586(2)	0.0383(7)
O12	8d		0.2121(2)	0.5714(4)	0.59247(19)	0.0360(7)
O13	8d		0.1432(3)	0.5643(5)	0.7268(2)	0.0524(10)
O14	8d		0.0292(3)	0.6074(5)	0.6129(3)	0.0595(11)
S2	8d		0.62351(8)	0.60358(16)	0.58538(7)	0.0366(3)
O21	8d		0.5255(2)	0.4977(4)	0.5999(2)	0.0403(7)
O22	8d		0.6420(3)	0.6090(4)	0.4991(2)	0.0423(8)
O23	8d		0.6071(3)	0.7742(6)	0.6149(3)	0.0712(13)
O24	8d		0.7089(3)	0.5197(7)	0.6246(3)	0.0757(14)

Table S5. Anisotropic displacement parameters $U_{ij} / \text{\AA}^2$ in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$ at $T = 373$ K. Standard deviations are given in parentheses

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
Na1	0.104(3)	0.051(2)	0.043(2)	0.019(2)	0.0092(19)	0.0006(15)
Na2	0.093(12)	0.038(4)	0.083(18)	0.022(7)	0.067(13)	0.011(5)
Na5	0.040(5)	0.055(4)	0.048(4)	-0.017(3)	-0.015(3)	0.015(3)
Na6	0.085(14)	0.102(17)	0.04(2)	0.002(11)	0.011(10)	0.011(16)
Mo1	0.0288(2)	0.0221(2)	0.0417(3)	-0.00194(13)	0.00471(14)	-0.00079(14)
O1	0.0471(19)	0.0320(17)	0.064(2)	-0.0009(14)	0.0125(16)	0.0087(16)
O2	0.0391(18)	0.0406(19)	0.057(2)	-0.0020(14)	0.0020(15)	-0.0120(16)
S1	0.0262(5)	0.0261(5)	0.0478(6)	-0.0010(4)	0.0025(4)	0.0043(4)
O11	0.0427(17)	0.0251(15)	0.0471(19)	0.0012(13)	-0.0040(14)	0.0043(14)
O12	0.0313(15)	0.0324(17)	0.0444(18)	-0.0004(12)	-0.0012(12)	-0.0048(13)
O13	0.065(2)	0.043(2)	0.049(2)	0.0003(17)	0.0140(18)	0.0179(16)
O14	0.0269(16)	0.046(2)	0.106(3)	-0.0063(15)	-0.0116(19)	0.003(2)
S2	0.0251(5)	0.0413(6)	0.0435(6)	-0.0016(4)	-0.0021(4)	-0.0077(5)
O21	0.0315(15)	0.0321(16)	0.057(2)	0.0014(13)	0.0080(14)	0.0016(14)
O22	0.053(2)	0.0331(17)	0.0406(18)	-0.0096(14)	0.0053(14)	0.0015(14)
O23	0.065(3)	0.053(3)	0.095(3)	-0.015(2)	0.015(2)	-0.035(2)

024	0.047(2)	0.104(4)	0.076(3)	0.004(2)	-0.024(2)	0.019(3)
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Table S6. Atomic coordinates, Wyckoff symbols and isotropic displacement parameters $U_{eq} / \text{\AA}^2$ in $\beta\text{-K}_2[\text{MoO}_2(\text{SO}_4)_2]$. Standard deviations are given in parentheses

Atom	Wyckoff symbol	x	y	z	U_{eq}
K1	4e	1.06036(7)	0.14163(5)	0.38368(8)	0.03123(13)
K2	4e	0.36525(7)	0.85228(6)	0.57500(8)	0.03653(14)
Mo1	4e	0.73860(2)	0.55604(2)	0.50925(2)	0.01827(6)
O1	4e	0.5926(2)	0.52038(18)	0.3763(3)	0.0329(4)
O2	4e	0.7600(3)	0.44778(16)	0.6435(3)	0.0378(5)
S1	4e	0.64663(6)	0.69180(5)	0.82676(7)	0.01721(11))
O11	4e	0.75672(18)	0.78185(14)	0.8699(2)	0.0215(3)
O12	4e	0.50561(19)	0.73298(17)	0.8321(2)	0.0297(4)
O13	4e	0.6843(2)	0.59260(17)	0.9255(2)	0.0323(4)
O14	4e	0.65244(19)	0.66503(15)	0.6466(2)	0.0233(3)
S2	4e	0.91252(6)	0.39121(5)	0.29178(7)	0.01854(11))
O21	4e	1.0611(2)	0.36618(16)	0.3635(3)	0.0321(4)
O22	4e	0.87750(19)	0.50452(14)	0.3632(2)	0.0227(3)
O23	4e	0.8211(2)	0.30468(17)	0.3347(3)	0.0437(6)
O24	4e	0.9015(3)	0.40821(18)	0.1171(2)	0.0398(5)

Table S7. Anisotropic displacement parameters $U_{ij} / \text{\AA}^2$ in $\beta\text{-K}_2[\text{MoO}_2(\text{SO}_4)_2]$. Standard deviations are given in parentheses

Atom	U_{11}	U_{22}	U_{33}	U_{12}	U_{13}	U_{23}
K1	0.0391(3)	0.0195(2)	0.0349(3)	0.0025(2)	0.0056(2)	-0.0007(2)
K2	0.0354(3)	0.0378(3)	0.0386(3)	0.0058(3)	0.0125(3)	0.0065(3)
Mo1	0.02039(10)	0.01622(9)	0.01895(10)	-0.00174(7)	0.00540(7)	-0.00156(7)
O1	0.0254(9)	0.0365(10)	0.0372(11)	-0.0096(8)	0.0060(8)	-0.0134(9)
O2	0.0557(14)	0.0254(10)	0.0366(11)	0.0045(9)	0.0196(10)	0.0074(8)
S1	0.0193(2)	0.0205(3)	0.0153(2)	-0.00275(19)	0.00410(19)	-0.00082(19)
O11	0.0201(8)	0.0215(8)	0.0238(8)	-0.0031(6)	0.0060(6)	-0.0047(6)
O12	0.0192(8)	0.0426(11)	0.0283(9)	-0.0023(8)	0.0075(7)	-0.0076(8)
O13	0.0427(11)	0.0274(9)	0.0244(9)	-0.0069(8)	-0.0014(8)	0.0078(7)
O14	0.0299(9)	0.0247(8)	0.0156(7)	0.0023(7)	0.0051(6)	-0.0025(6)
S2	0.0187(2)	0.0162(2)	0.0199(2)	0.00061(19)	0.00083(19)	-0.00247(19)
O21	0.0224(9)	0.0246(9)	0.0446(11)	0.0058(7)	-0.0084(8)	-0.0102(8)
O22	0.0257(8)	0.0173(8)	0.0270(9)	0.0028(6)	0.0098(7)	-0.0029(6)
O23	0.0376(12)	0.0228(10)	0.0754(17)	-0.0092(8)	0.0234(11)	-0.0055(10)
O24	0.0669(16)	0.032(1)	0.0179(9)	0.0141(10)	-0.0005(9)	-0.0054(8)

Satellite Reflections in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$ in the Reciprocal Space

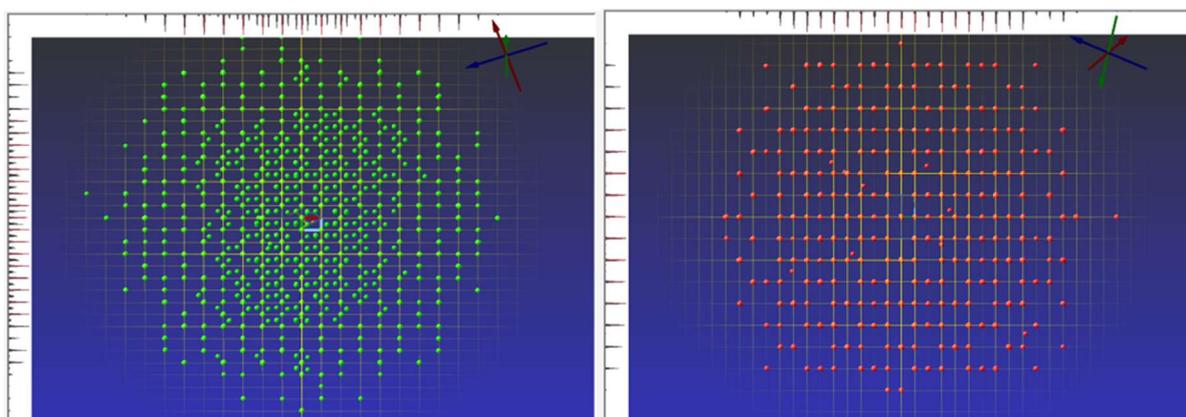


Figure S1. Reciprocal Space at 293 K (left, green) and 373 K (right, red) depicted in the software package APEX.^[1]

De Wolff Section

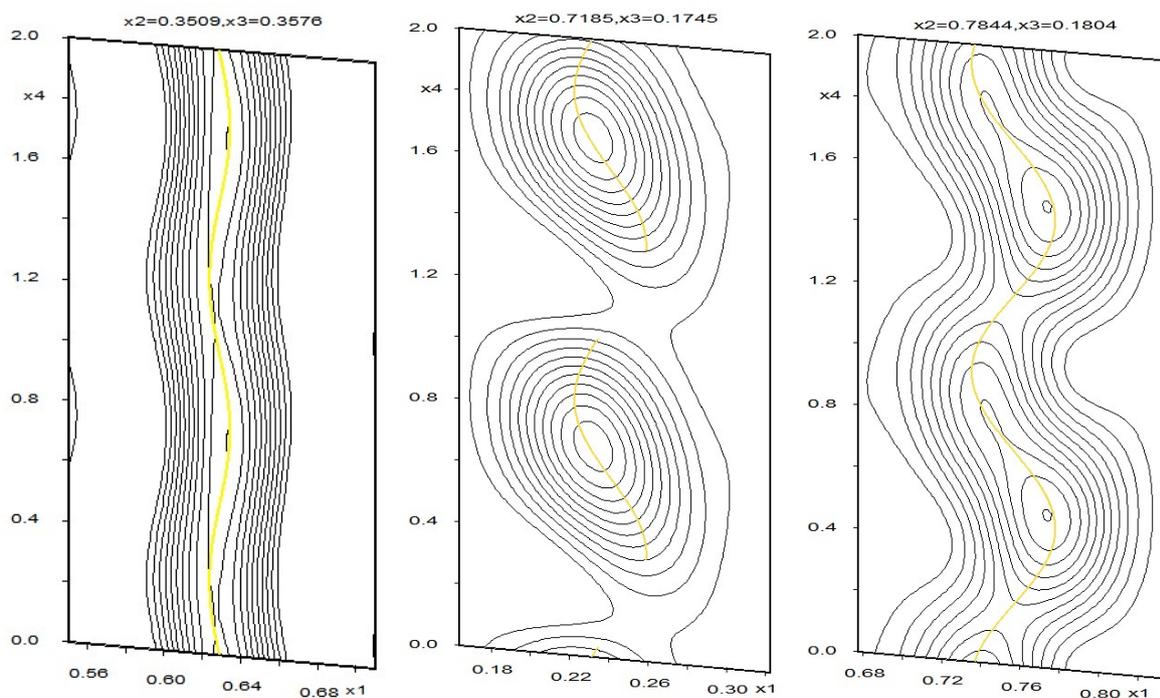


Figure S2. Representative de Wolff plots of S, Na2-x1-x4, Na3-x1-x4.

Centroid Shift of Molybdenum

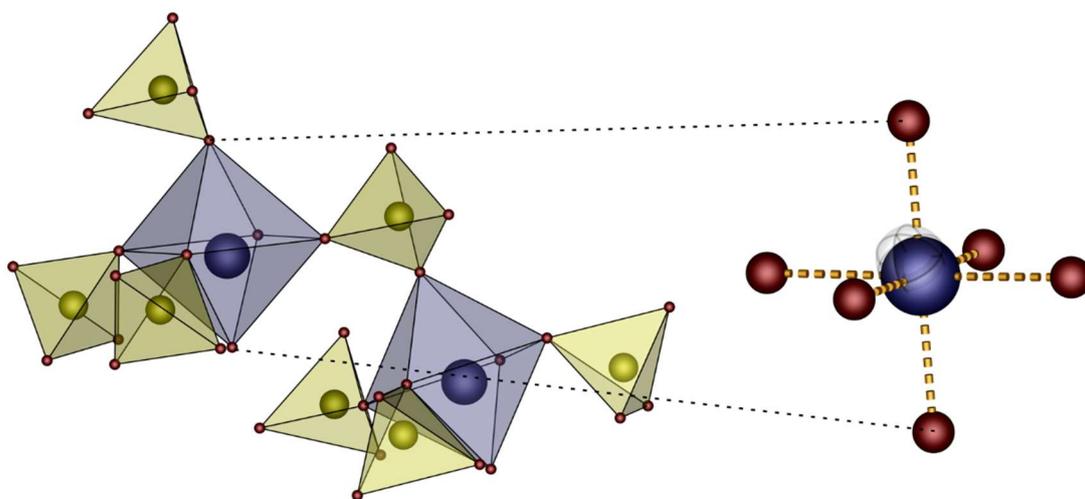


Figure S3. Centroid shift in the molybdate octahedra due to the number of coordinating sulfate tetrahedra in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$.

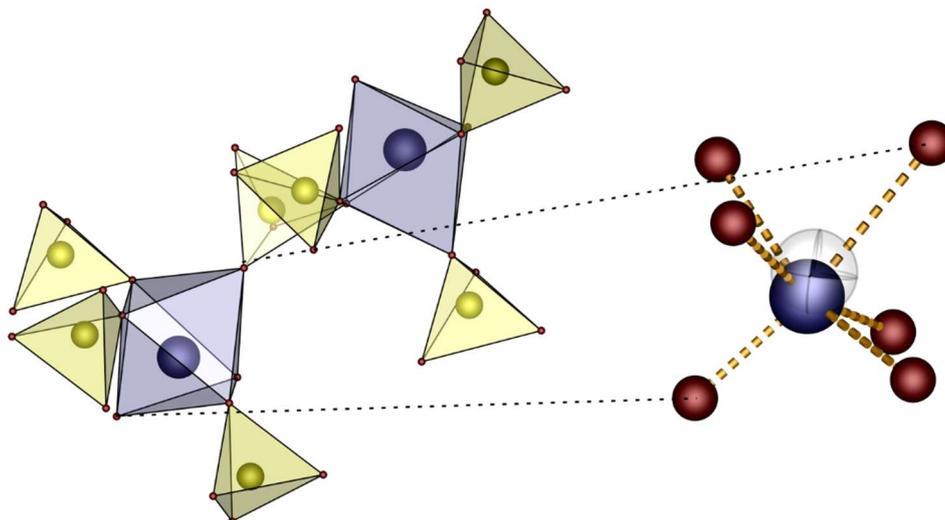


Figure S4. Centroid shift in the molybdate due to the number of coordinating sulfate tetrahedra in $\text{K}_2[\text{MoO}_2(\text{SO}_4)_2]$.

Coordination of Sodium Cations

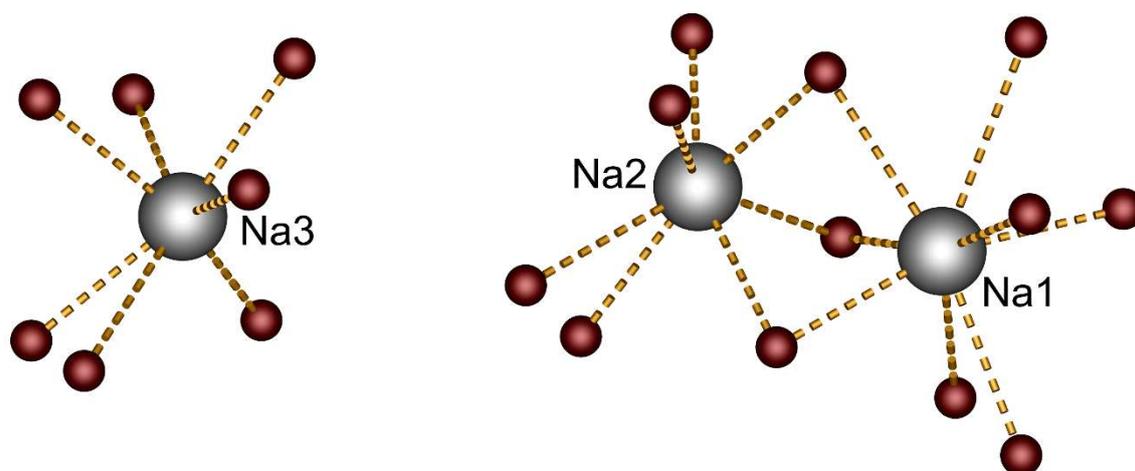


Figure S5. Coordination environment of chosen sodium cations in $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$.

Selected Ionic Radii after Shannon

Table S8. Ionic radii / pm after Shannon^[2] with respect to their coordination number

Ion	Charge	Coordination	Ionic radius
K	1	IX	155
Na	1	VII	112
Na	1	VIII	124
Mo	6	VI	59
S	6	IV	12
O	-2	III	136
O	-2	IV	138

FT-IR Spectroscopy

Table S9. Assignments of vibrations in the FT-IR spectra

Wavenumber / cm^{-1}	Vibration
1400—950	$\nu_{\text{sym}}(\text{S—O})^{[3-6]}$
	$\nu_{\text{asym}}(\text{S—O})^{[3-6]}$
650—400	$\delta(\text{S—O})^{[3-6]}$
990—600	$\nu_{\text{sym}}(\text{Mo—O})^{[7]}$
	$\nu_{\text{asym}}(\text{Mo—O})^{[7]}$
< 400 (not detectable)	$\delta(\text{Mo—O})^{[7]}$
3700—3500	$\nu(\text{O—H})^{[8]}$
1630—1600	$\delta(\text{O—H})^{[8]}$

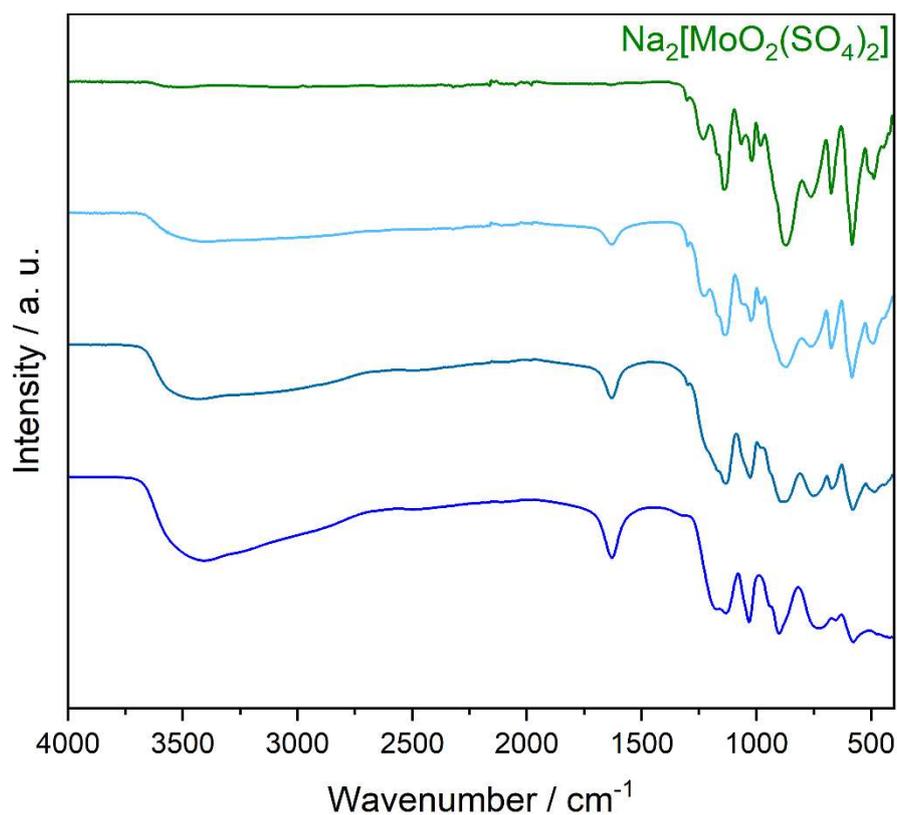


Figure S6. FT-IR spectra of $\text{Na}_2[\text{MoO}_2(\text{SO}_4)_2]$ after contact to ambient air. Dark green directly after opening the ampoule; light blue after exposing to air, middle blue after 30 min; dark blue after one night.

MAPLE Calculations

Table S10. Calculated MAPLE values of the binary and tertiary compounds

Compound	MAPLE Value in kJ mol ⁻¹
K ₂ SO ₄ ^[9]	33098
Na ₂ O ^[10]	2907
MoO ₃ ^[11]	25353
SO ₃ ^[12]	29930

References

- [1] APEX VI, Bruker AXS Inc., Madison, Wisconsin, USA **2025**.
- [2] R. D. Shannon, *Acta Cryst. A* **1976**, 32, 751.
- [3] M. V. Barashkov, A. I. Komyak, S. N. Shashkov, *J. Appl. Spectrosc.* **1999**, 1, 100.
- [4] V. Ramakrishnan, V. U. Nayar, G. Aruldas, *Infrared Phys.* **1985**, 25, 607.
- [5] A. Periasamy, S. Muruganand, M. Palaniswamy, *Rasayan J. Chem.* **2009**, 4, 981.
- [6] R. M. Atkins, K. A. Gingerich, *Chemical Physics Letters* **1978**, 53, 347.
- [7] G.-A. Nazri, C. Julien, *Solid State Ion.* **1992**, 53, 376.
- [8] K. Nakamoto, *Infrared and Raman Spectra of Inorganic and Coordination Compounds*, Wiley **2008**.
- [9] N. V. Zubkova, I. V. Pekov, D. A. Ksenofontov, V. O. Yapaskurt, D. Y. Pushcharovsky, E. G. Sidorov, *Dokl. Earth Sc.* **2018**, 479, 339.
- [10] E. Zintl, A. Harder, B. Dauth, *Z. Elektrochem, Angew. Phys. Chem.* **1934**, 40, 588.
- [11] J. B. Parise, E. M. McCarron, A. W. Sleight, E. Prince, *MSF* **1988**, 27-28, 85.
- [12] R. Pascard, C. Pascard-Billy, *Acta Cryst.* **1965**, 18, 830.