

Supporting Information

Computational Investigation of Perfluorocarbonates as Multi-Functional Diluents in Localized High Concentration Electrolytes for Lithium Metal Batteries: Insights from MD and DFT Simulations

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Table S1. Type and quantity of non-solvating solvents, along with their corresponding dielectric constants, for seven types of LHCE systems in the final frame of the MD simulation.

No.	FEC	DFEC	DFEC1	DFEC2	tri-FEC	tetra-FEC	TTE	1,2-DME	Dielectric constant
1	152	0	0	0	0	0	0	14	72.93
2	0	136	0	0	0	0	0	12	26.51
3	0	0	154	0	0	0	0	16	31.24
4	0	0	0	157	0	0	0	4	10.12
5	0	0	0	0	158	0	0	10	8.76
6	0	0	0	0	0	142	0	10	2.62
7	0	0	0	0		0	100	11	6.42

No.1-7 are LiFSI/1,2-DME/FEC, LiFSI/1,2-DME/DFEC, LiFSI/1,2-DME/DFEC1, LiFSI/1,2-DME/DFEC2, LiFSI/1,2-DME/Tri-FEC, LiFSI/1,2-DME/Tetra-FEC, LiFSI/1,2-DME/TTE, respectively.

In each LHCE system, the amount of 1,2-DME not involved in the solvation is relatively small compared to that of each diluent.

Therefore, for the calculation of HOMO/LUMO of Li⁺ solvation structures, free anions, and solvent molecules in the seven types of LHCE systems (**Figure 6c-d**), an implicit solvation model was used, and the dielectric constant of the mixed solvent was determined via a linear combination of the dielectric constants presented in **Figure 1d**.¹

Table S2. Distance of Ni-O bond (In FEC and its derivatives, the O refers to the carbonyl oxygen in FEC and its derivatives).

FEC	DFEC	DFEC1	DFEC2	tri-FEC	tetra-FEC	TTE
2.551 Å	2.730 Å	2.605 Å	2.795 Å	2.859 Å	2.864 Å	3.466 Å

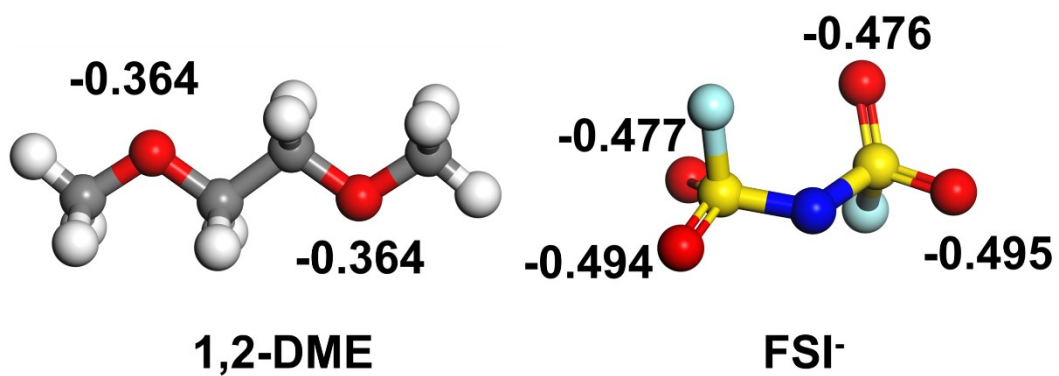


Figure S1 Charge distribution on oxygen atoms of 1,2-DME and FSI⁻ anions without considering the solvation model.

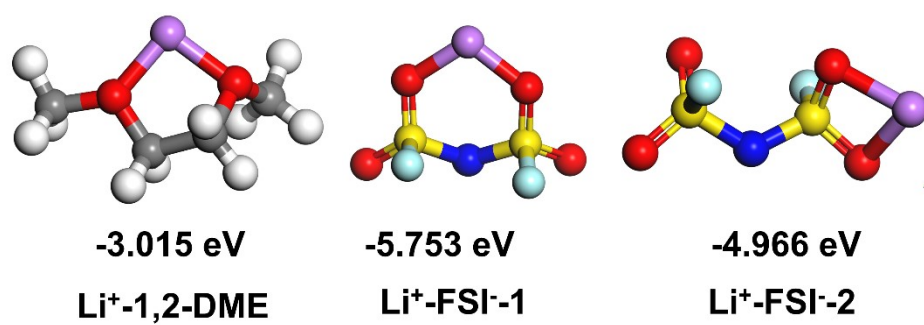


Figure S2 Binding energies between solvent/anion (1,2-DME, FSI⁻) and Li⁺.

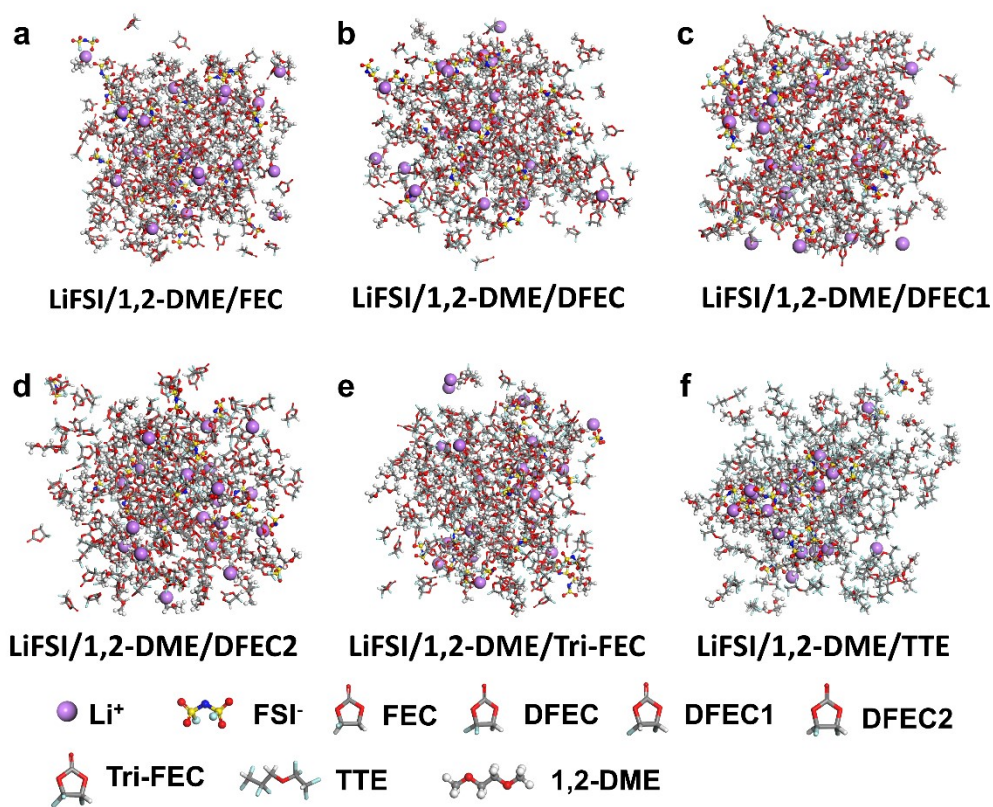


Figure S3 a-b) Snapshots of different LHCE systems of LiFSI/1,2-DME/FEC, LiFSI/1,2-DME/DFEC, LiFSI/1,2-DME/DFEC1, LiFSI/1,2-DME/DFEC2, LiFSI/1,2-DME/Tri-FEC, LiFSI/1,2-DME/Tetra-FEC, LiFSI/1,2-DME/TTE.

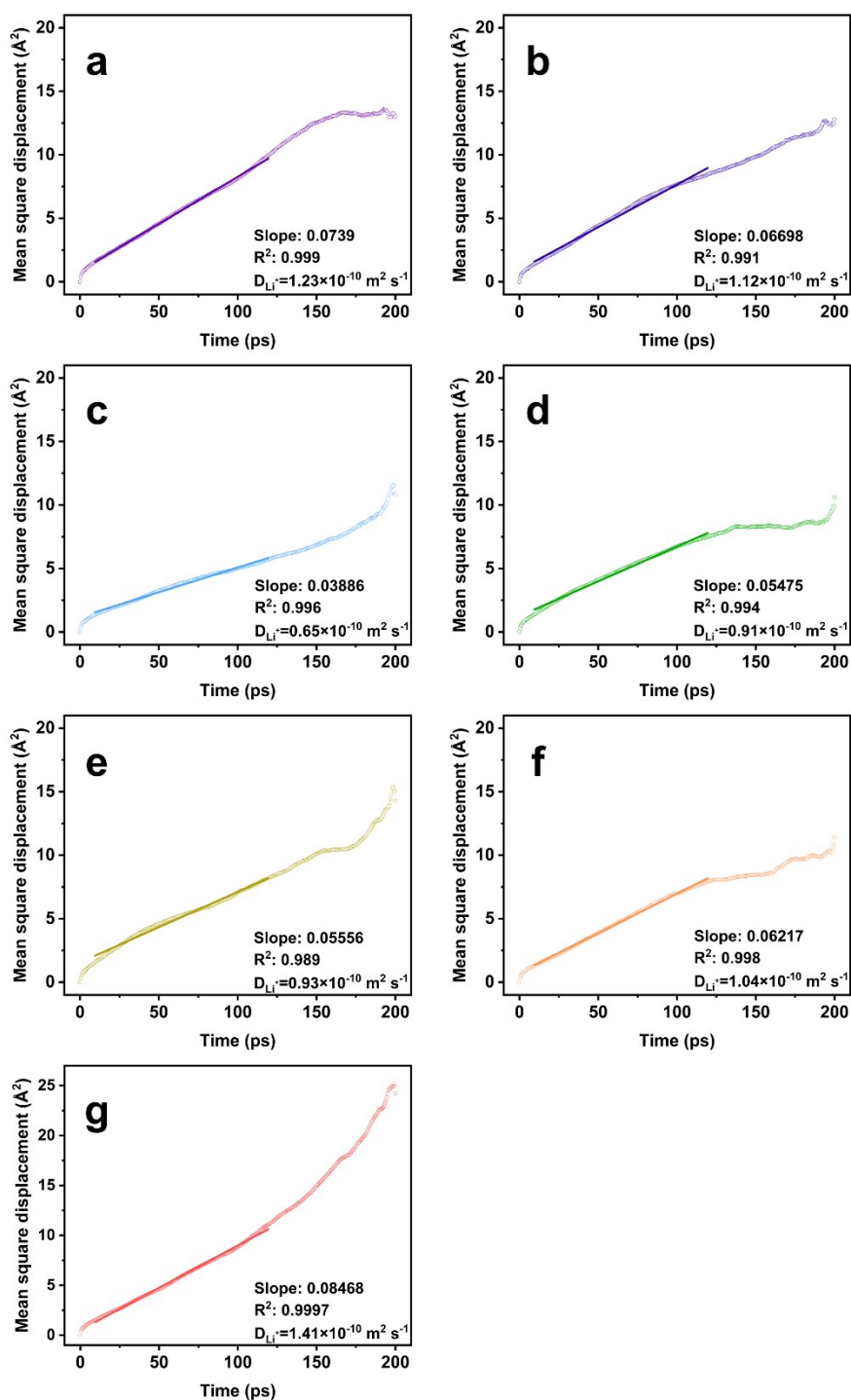


Figure S4 a-b) Li⁺ mean square displacement curves and corresponding fits of LiFSI/1,2-DME/ FEC, LiFSI/1,2-DME/ DEFC, LiFSI/1,2-DME/ DEFC1, LiFSI/1,2-DME/DEFC2, LiFSI/1,2-DME/Tri-FEC, LiFSI/1,2-DME/Tetra-FEC, LiFSI/1,2-DME/TTE at 298 K.

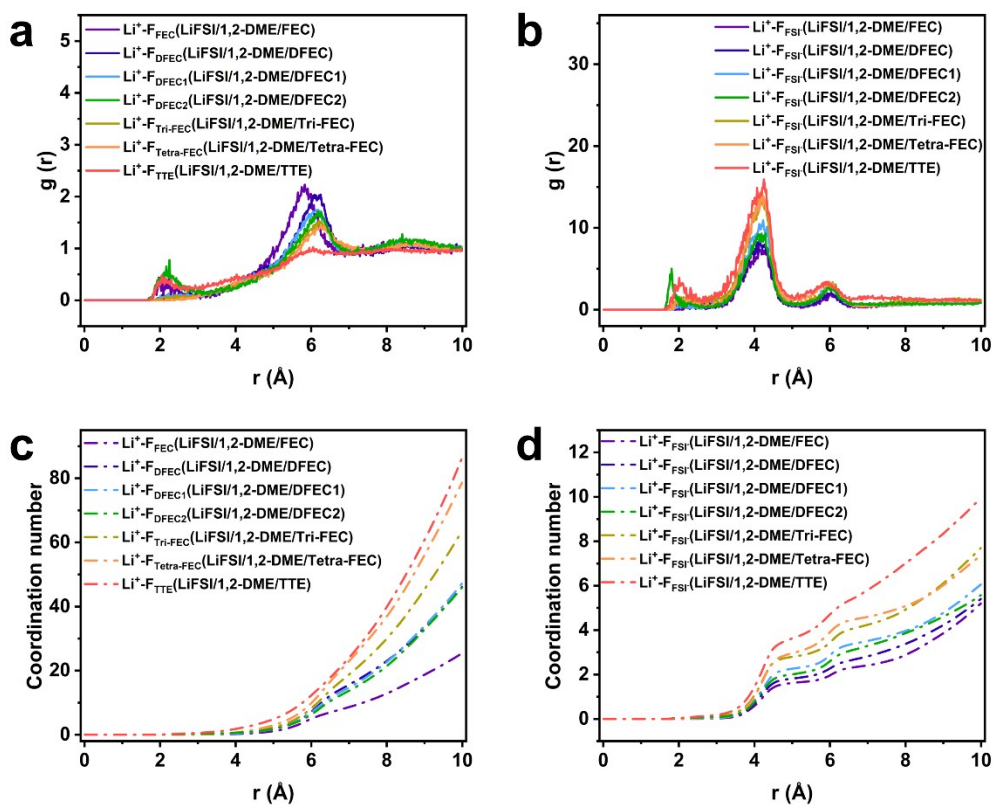


Figure S5 a-b) Radial distribution function of Li^+ - $\text{F}_{\text{diluent}}$, Li^+ - F_{FSI^-} of LiFSI/1,2-DME/FEC, LiFSI/1,2-DME/DFEC, LiFSI/1,2-DME/DFEC1, LiFSI/1,2-DME/DFEC2, LiFSI/1,2-DME/Tri-FEC, LiFSI/1,2-DME/Tetra-FEC, LiFSI/1,2-DME/TTE at 298 K; c-d) corresponding coordination number around Li^+ in all the LHCE systems.

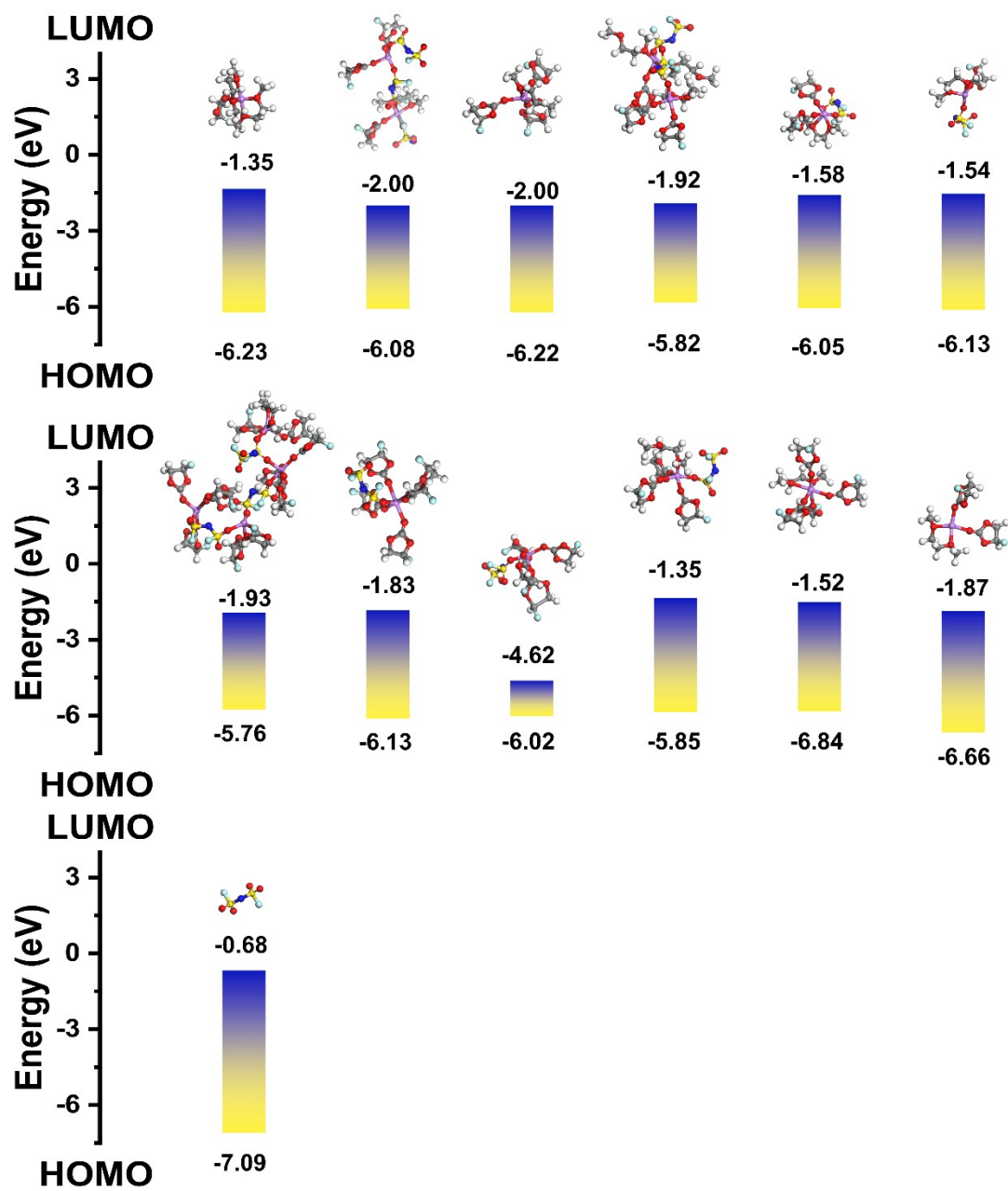


Figure S6 LUMO and HOMO levels of the Li clusters in the last frame in the LiFSI/1,2-DME/ FEC LHCE system.

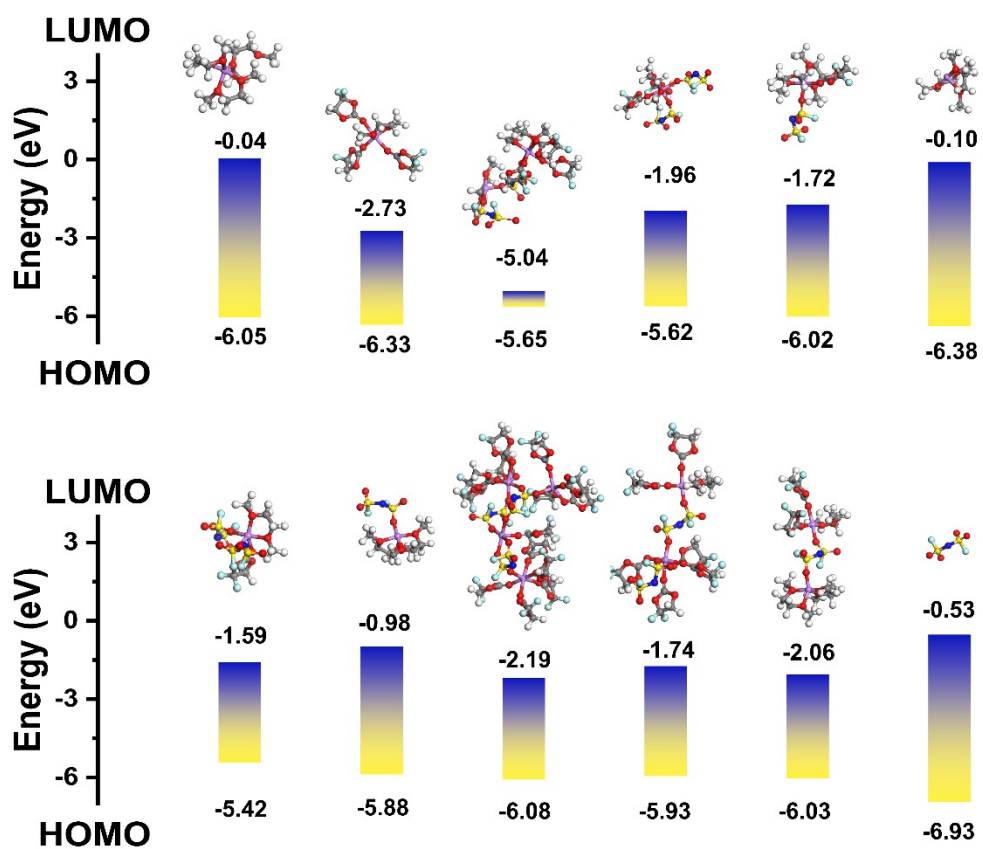


Figure S7 LUMO and HOMO levels of the Li clusters in the last frame in the LiFSI/1,2-DME/ DFEC LHCE system.

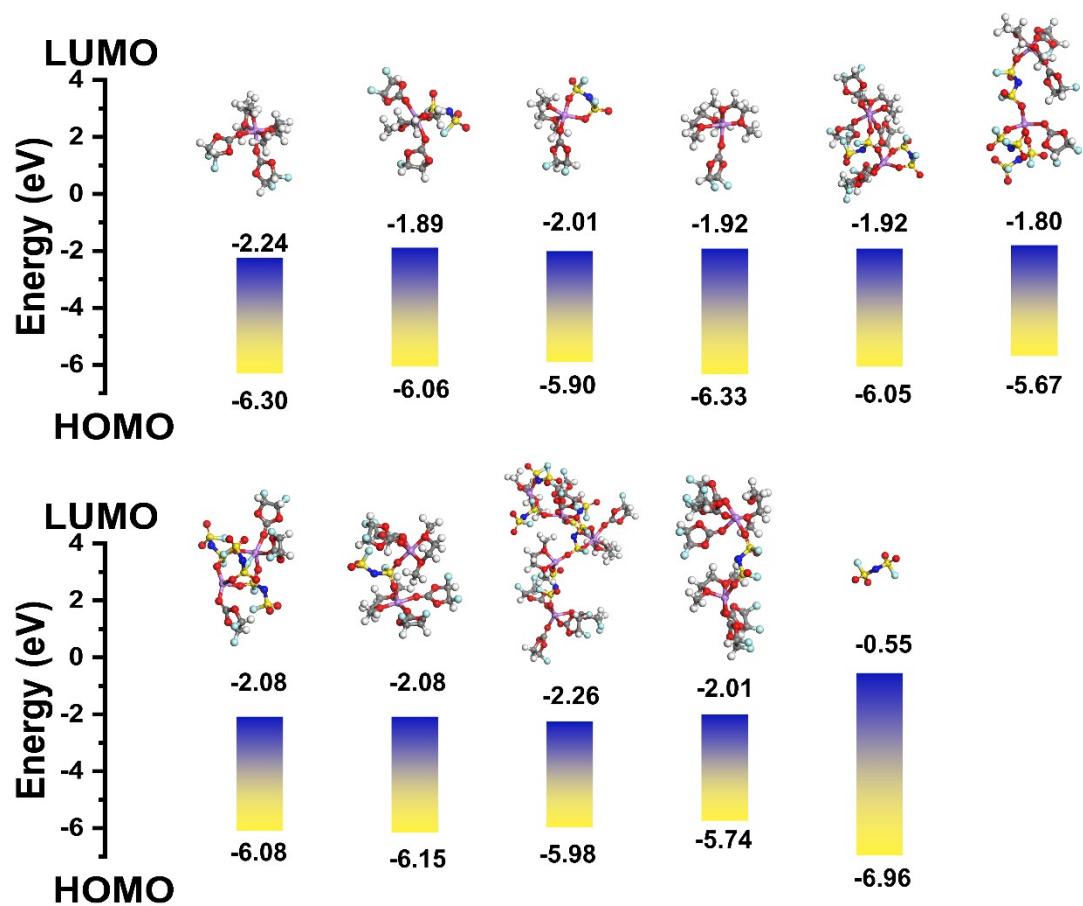


Figure S8 LUMO and HOMO of the Li clusters in the last frame in the LiFSI/1,2-DME/ DFEC1 LHCE system.

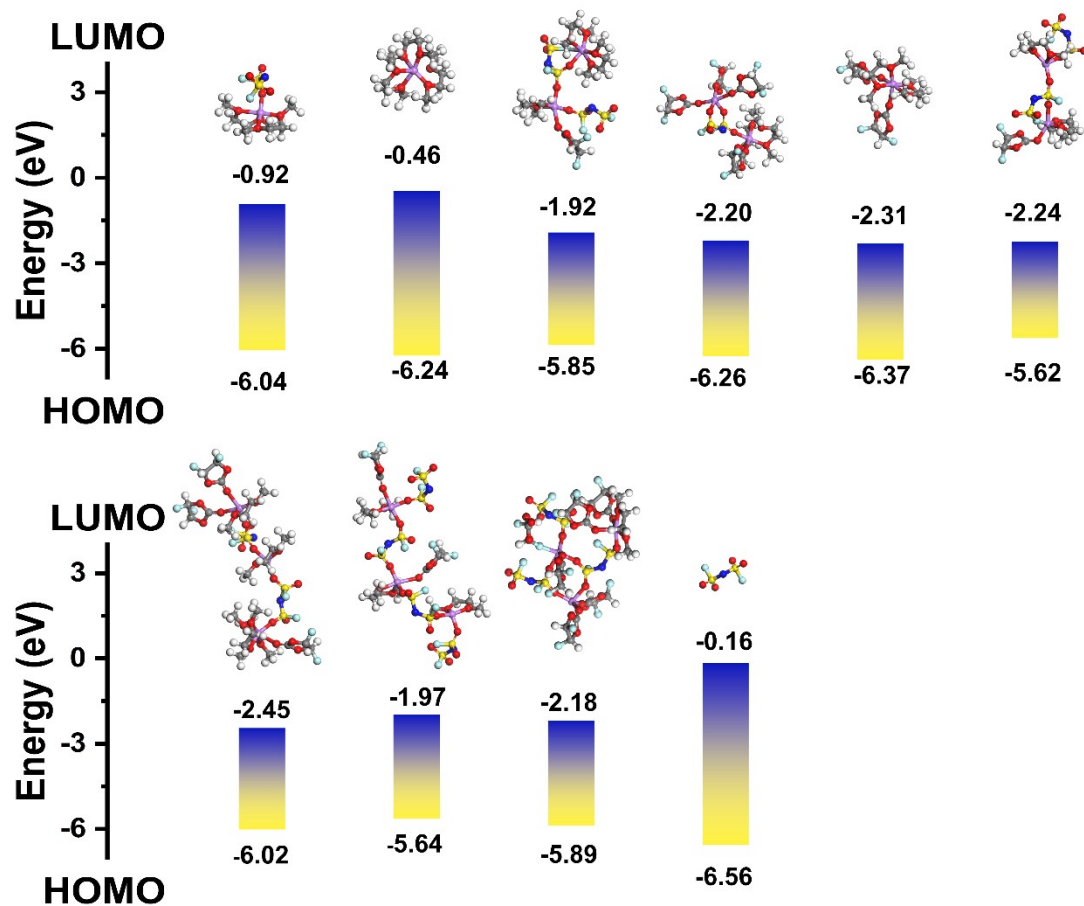


Figure S9 LUMO and HOMO levels of the Li clusters in the last frame in the LiFSI/1,2-DME/ DFEC2 LHCE system.

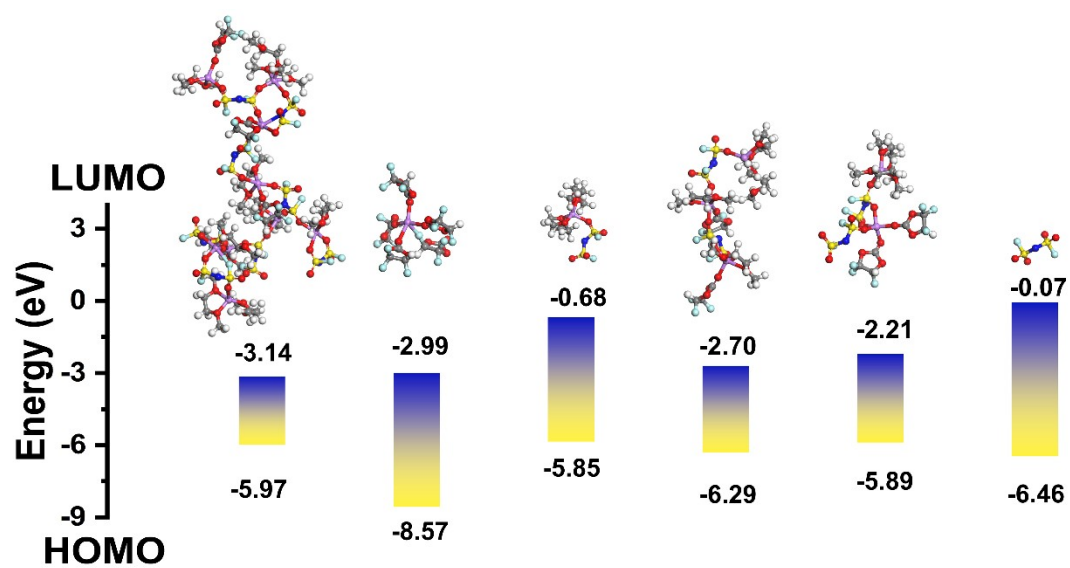


Figure S10 LUMO and HOMO levels of the Li clusters in the last frame in the LiFSI/1,2-DME/ Tri-FEC LHCE system.

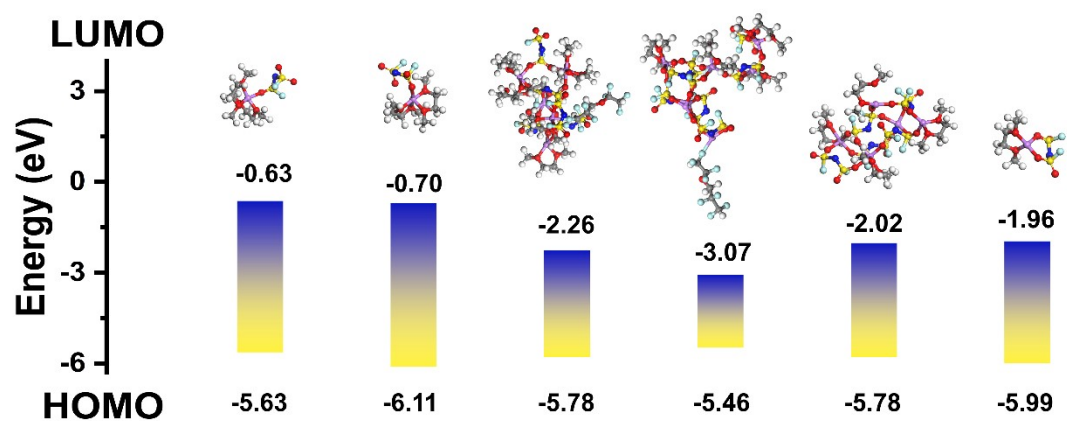


Figure S11 LUMO and HOMO levels of the Li clusters in the last frame in the LiFSI/1,2-DME/ TTE LHCE system.

1. N. Yao, X. Chen, X. Shen, R. Zhang, Z. H. Fu, X. X. Ma, X. Q. Zhang, B. Q. Li and Q. Zhang, *Angew. Chem. Int. Ed.*, 2021, **60**, 21473-21478.