# Anchoring Ligand Engineering Enables Highly Stable MA-Free Perovskite Solar Cells with a Minimal Voc Deficit of 0.32 V

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### **Experimental Section**

### Materials:

Lead iodide (PbI<sub>2</sub>, 99.99%), and [4-(3,6-Dimethyl-9H-carbazol-9-yl)butyl]phosphonic Acid (Me-4PACz) were purchased from TCI Development Co. Formamidinium iodide (FAI,  $\geq$  99.5%) and 4-Chlorophenethylammonium iodide (ClPEAI) were ordered from Greatcell Solar Ltd. Methylammonium bromide (MABr,  $\geq$  99.5%), Methylammonium chloride (MACl,  $\geq$  99.5%) was purchased from Xi'an polymer Light Technology Corp. Lead bromide (PbBr<sub>2</sub>, 99.999%), Cesium iodide (CsI, 99.9%), Methylphosphonic acid (MPA, 98%), Ethanol absolute (EtOH), dimethyl sulfoxide (DMSO,  $\geq$  99.9%), N, N-dimethylformamide (DMF, 99.8%), ethyl acetate (EA  $\geq$  97%), NiOx nanoparticles was obtained Advanced ElectionTechnology Co., Ltd. Ethanolamin ( $\geq$  99%), Ethanol( $\geq$  99.8%) and Al<sub>2</sub>O<sub>3</sub> dispersion (30 nm, 20 wt% in isopropanol), chlorobenzene (CB, 99.8%), Triphenylphosphine oxide (TPPO, 98%) and Bathocuproine (BCP) were obtained from Sigma-Aldrich. Isopropanol ( $\geq$  99.9%) was purchased from Aladdin. PC<sub>61</sub>BM was purchased from Nano-C. All the chemicals were used as received without further purification.

## Preparation of control and target films and devices

The pre-patterned ITO-coated glass substrates ( $25 \times 25 \text{ mm}^2$  with sheet resistance of 15 ohm/sq) were cleaned by ultrasonication in acetone and isopropanol for 10 min each. Before deposition, the bare ITO substrate was treated with UV-ozone for 10min. First, NiOx nanopartical (10 mg ml<sup>-1</sup> in H<sub>2</sub>O) layer was spin coated on as-cleaned ITO substrates at 4000 rpm for 30 s, and then annealed at 150 °C for 10 min in ambient air, followed by immediately transfer into a nitrogen-

filled glove box. The hole transport layers of Me-4PACz and MPA were sequentially deposited on the ITO substrate at 4000 rpm for 30 s and annealed at 100 °C for 10 min, the concentration of precursor was 0.5 mg/mL of Me-4PACz or MPA in alcohol solution. The above processes were performed in the N2-globlox. After that, Al2O3 nanoparticles were diluted in IPA (Al2O3/IPA= 1:50, v:v), spin-coated onto the SAM layer at 5000 rpm for 30 seconds, and subsequently annealed at 100 °C for 5 minutes. the perovskite layer was deposited by using the one-step spin-coating method in the N<sub>2</sub>-filled glovebox. 1.5 M FA<sub>0.9</sub>Cs<sub>0.1</sub>PbI<sub>3</sub> perovskite precursor was prepared by dissolving 232.2 mg FAI, 38.97 mg CsI, 25 mg MACl, 726 mg PbI<sub>2</sub> in conventional solvent system (DMF/DMSO= 5:1, v:v). The perovskite solution was deposited onto the above substrates at 1000 rpm for 5 s and 5000 rpm for 30 s, 280 µL EA was dripped onto the center of the film at 10 s before the end of program, followed by annealing on a hotplate at 110 °C for 30 min. Next, a mixture of 1 mg ClPEAI/MAI (2:1, wt/wt) was dissolved in 1 mL IPA–DMF (200:1, v/v), deposited onto the upper of perovskite films at 4000 rpm for 30 s and then annealed at 100 °C for 5 min. Subsequently, Then TPPO solution was deposited on the CIPEAI layer at 5000 rpm for 30 s and annealed at 85 °C for 5 min. PCBM (20 mg/mL) solution was spin-coated on the perovskite layer at 2000 rpm for 30 s and then annealed for 5 min at 70 °C. BCP (0.25 mg/mL) solution was deposited on the PCBM layer with 4000 rpm 30 s. Finally, the Ag electrode (100 nm) was evaporated on the BCP layer using the thermal evaporation method and the fixed active area of this electrode was 0.04 cm<sup>2</sup>. Note that all solutions are filtered using a 0.22 µm Polytetrafluoroethylen (PTFE) filter before spin coating. For both FA0.82MA0.13Cs0.05Pb(I0.87Br0.13)3 (bandgap  $\approx$  1.59 eV)- and

 $Cs_{0.22}FA_{0.78}PbI_{1.8}Br_{1.2}$  (bandgap  $\approx 1.797$  eV)-bandgap perovskites, we employed the same fabrication process as that used for the 1.55 eV bandgap perovskite

### Characterizations:

*X-ray diffraction (XRD) measurement*: The crystal structure was characterized by Panalytical X'pert powder diffractometer (XRD) with Cu K $\alpha$  radiation ( $\lambda = 1.54178$  Å) and an X'Celerator solid-state stripe detector operated at 40 kV and 30 mA.

*X-ray photoelectron spectroscopy (XPS) measurement*: XPS were performed on an EA 125X U7 Energy Analyser from Scienta Omicron using Al Kalpha radiation with 1486.7 eV excitation energy, 0.05 eV step with 20 eV pass and 5 s dwell was used for XPS analysis.

*Scanning electron microscopy (SEM) images*: The morphologies of the perovskite films were investigated by scanning electron microscope (JEOL JSM-7610F).

A double-beam spectrophotometer (Lambda 950, PerkinElmer) equipped with an integrated sphere was used for the UV-vis absorption measurement.

*Photoluminescence (PL) spectra*: PL was carried out using a steady-state spectroscope (FluoTime 300, PicoQuant GmbH) excited by a 402 nm laser.

*Impedance spectra*: The Mott-Schottky and Nyquist plots measurements were performed on an electrochemical workstation (ZAHNER, Germany). The impedance spectra and built-in potential  $(V_{bi})$  were correspondingly measured in dark and under illumination with a LED lamp (LSW-2, s/n LS 1858).

*Kelvin force microscope (KPFM) images*: KPFM measurements were done with the NX10 (Park Systems), installed inside of the Ar-filled glovebox for the inert atmosphere measurement. PPP-EFM probe (resonance frequency of 75 kHz, Spring constant of 2.8 N m<sup>-1</sup>, coated with PtIr5 on both sides) was used to obtain the 2 × 2  $\mu$ m<sup>2</sup> area images, with a 512 × 512 pixel resolution. Obtained data was processed with Park SmartAnalysis<sup>™</sup> software.

*Current density-voltage (J–V) measurement*: J–V characteristics of the solar cells were measured using a Keithley 2400 source meter. The illumination was provided by a Newport Sol3A solar simulator with an AM1.5G filter, operating at 100 mW cm<sup>-2</sup>, which was calibrated by a standard silicon solar cell from Newport. Both forward and backward scans were performed, and the scan rate of 20 ms/step and a scan step of 40 mv. The external quantum efficiency (EQE) of the perovskite solar cell device was measured by using a QE-R instrument from Enlitechnology.

*Time-of-flight secondary ion mass spectrometry (TOF-SIMS) measurement:* The SIMS depth profiles were acquired by using a Time-of-Flight SIMS IV (Ion-TOF GmbH, Muenster, Germany) equipped with a Bi<sup>+</sup> primary ion gun working at 25 keV in dual beam mode under the UHV conditions. The sputtering was realized with Cs<sup>+</sup> ion guns at 1 keV energy using of active electron flooding to prevent the surface charging to increase the yield of negatively charged secondary ions. While the sputtering Cs<sup>+</sup> gun creating the crater, the primary ion beam with Bi<sup>+</sup> is progressively analyzing the crater bottom. The usual configuration for depth profiles is to sputter 300 × 300 µm<sup>2</sup> from the sample surface and the analysis is then done in the middle of this area  $84 \times 84 \mu m^2$ . After a careful calibration using to known spectral lines, the data files were analyzed by SurfaceLab software provided by Ion-TOF. For the selected ions like I<sup>-</sup> and Ag<sup>-</sup> a peak list was created and then the depth profile evaluated. All samples were measured under same conditions.

**Quasi-Fermi level spitting (QFLS)** *measurements*: QFLS was calculated according to previous our group reports.<sup>1-3</sup> In this work, we calculate the QFLS according to

 $QFLS = QFLS_{rad} + k_BTln^{(ro)}(PLQY)$ 

 $QFLS_{\rm rad} = kTln^{\rm red} \left( \frac{J_G}{J_{0,\rm rad}} \right)$ 

where QFLS<sub>rad</sub> is the QFLS in the radiative limit,  $J_G$ ,  $J_0$ ,  $_{rad}$ , k and T are generation current density, dark radiative saturation current density, Boltzmann constant, thermal equilibrium temperature, respectively.  $J_G$  here is the integrated product of EQE spectra and  $J_0$ ,  $_{rad}$  can be determined by:

$$J_{0,rad} = q \int E Q E_{PV}(E) \phi_{bb}(E) dE$$

$$\phi_{bb}(E) = \frac{2\pi E^2}{\left(h^3 c^2\right)} exp^{(m)} \left(-\frac{E}{kT}\right)$$

where q is the elementary charge, E the photo energy, h the Planck constant and c the light velocity in vacuum.

*Long term stability measurement*: Devices without encapsulation were loaded into a degradation chamber flowed with N<sub>2</sub>-chamber Light source was provided by 4 white LED (XLamp® CMA3090 LED) without using filter. The light intensity was calibrated with a Si reference cell (91150V) bought from Newport. Temperature is controlled via hotplate and monitored at all times with a temperature detector (PT100). The J-V characteristics were measured by LabView program with different interval time (2-10 minutes in the early stages and 45-90 minutes in the later stages).



Figure S1 XPS spectra of I 3d peaks of perovskite surface control and modified by TPPO (Target).



**Figure S2** Top-view SEM images of the (a) control and (b) control re-annealing films, Sample structure: ITO/NiOx/SAM/Perovskite



**Figure S3** Cross-section SEM image of control and target PSCs, the structure of ITO/NiO@SAM/perovskite/TPPO (w/o.w)/PCBM/BCP/Ag.



Figure S4 XRD image of control, Control re-annealing, Control@IPA re-anneaing, Target samples



**Figure S5**. AFM image of silicon wafer(Si), Control, Control@IPA,and target samples.sample artucture: Si/NiOx/SAM/Perovskite(w, w/o TPPO).Note The only difference between Control and Control@IPA is that Control@IPA was rinsed with IPA.



**Figure S6**. AFM image of control and target samples. sample artucture: Si/NiOx/SAM/Perovskite(w, w/o TPPO)/PCBM



Figure S7. UV-Vis spectra of Control and Target sample.



**Figure S8** Mott–Schottky plots of PSCs based on control, target. Nyquist plots of PSCs devices based on control and perovskite with TPPO, The Nyquist plots of sample are measured at different light intensity



Figure S9  $J_{sc}$  versus light intensity of Control and Target samples



**Figure S10** Eg, estimation obtained from the inflection point of the EQE spectra by locating the maximum point ( $\lambda g$ ) of the Gaussian-like derivate  $\partial EQE/\partial\lambda$  (1240/800=1.55 eV).



**Figure S11** Statistics of key photovoltaic parameters (open-circuit voltage ( $V_{OC}$ ), fill factor (FF), short-circuit current density ( $J_{SC}$ ), PCE) obtained from the *J-V* characteristic of 25 control devices, and devices modified with different concentrations of TPPO (0.0, 0.125, 0.25, 0.35 and 0.5 mg ml<sup>-1</sup>).



Figure S12 The *J*-*V* curve of some individual target PSC with a high  $V_{OC}$  of 1.23V (reverse scan).



**Figure S13** *J*-*V* curve of 1 cm<sup>2</sup>-active layer.



Figure S14 J-V curves of different band gaps with, without, TPPO modification (Reverse scan)



**Figure S15** Long-term operational stability of device's *J*sc, *V*oc, FF under continuous output at MPP conditions (1 sun, LED lamps with ultraviolet filter; 65 °C (a-c) and 85 °C (e-g) in N<sub>2</sub>-filled Chamber.



Figure S16 the PL evolution of control and target sample after 300 h under 1suns and 65 °C



Figure S17 the evolution of XRD under ambient conditions without (a) and with (b) TPPO



Figure S18 Water contact angle of the films processed without (a) and with TPPO-treated (b).

			PeLEDs	
No	Title	EQE (%)	Journal	Summary
1	Distribution control enables	25.6(2.5 times)		TPPO regulates quantum well distribution
	efficient reduced-dimensional			and passivates defects, boosting efficiency
	perovskite LEDs		Nature 599, 594–598 (202, DOI:	and color purity in reduced-dimensional
			10.1038/s41586-021-03997-z	PeLEDs
2	Efficient all-thermally	16.4(2times)		The use of TPPO provides efficient all-
	evaporated perovskite light-			thermal evaporated PeLED for active matrix
	emitting diodes for active-		Nature Photo.	displays
	matrix displays		DOI: 10.1038/s41566-023-01177-1	
3	Phosphine oxide additives for	Review		A comprehensive review summarizing
	perovskite light-emitting diodes			phosphine oxide ligands, including TPPO,
	and solar cells		Chem, 2023, DOI:	with emphasis on their widespread use in
			10.1016/j.chempr.2023.01.002	PeLEDs.
4	Efficient Perovskite Light-	21.01(1.58times)		TPPO was introduced at the buried interface
	Emitting Diodes by Buried			to reduce interfacial defects, achieving an
	Interface Modification with			EQE of 21.01%.
	Triphenylphosphine Oxide		ACS Applied Materials & Interfaces,	
			2023, DOI: 10.1021/acsami.2c19123	
5	Phosphine Oxide Additives for	9.5(2.1times)		Study demonstrating enhanced luminance and
	High-Brightness Inorganic			stability of inorganic PeLEDs using TPPO
	Perovskite Light-Emitting			additives.
	Diodes		Advanced Optical Materials, 2021,	
			DOI: 10.1002/adom.202101602	
6	Bilayer phosphine oxide	4.87(2 times)	Science Bulletin, 2023, DOI:	Use of bilayer phosphine oxide including

# **Table S1** Statistics on the application of TPPO in perovskite devices and LEDs

	modification toward efficient and large-area pure-blue perovskite quantum dot LEDs		10.1016/j.scib.2023.09.014	TPPO for performance enhancement in large- area blue QD PeLEDs.
7	Fluorine-modified passivator for efficient vacuum-deposited pure-red PeLEDs	12.6(640nm,6tim es)	Light: Science & Applications, 2025, DOI: 10.1038/s41377-025-01740-1	Study utilizing fluorinated TPPO analogs in red-emitting vacuum-deposited PeLEDs.
8	Trade-off between efficiency and stability in Mn <sup>2+-</sup> doped perovskite LEDs	14(4.7time)	Device, 2023, DOI: 10.1016/j.device.2023.100017	TPPO and similar phosphine oxide ligands were found effective for efficiency improvement.
9	Effect of Passivating Molecules and Antisolvents on Lifetime of Green Dion–Jacobson Perovskite Light-Emitting Diodes	12.42 (520nm,1.11time s)	ACS Applied Materials & Interfaces, 2023, DOI: 10.1021/acsami.3c02170	Study shows TPPO helps improve lifetime of quasi-2D PeLEDs.
10	Perovskite LEDs with an external quantum efficiency exceeding 25%		Nature Materials, 2022,	Comparative study of fluorinated and non- fluorinated phosphine oxides including TPPO.
11	A roadmap for the commercialization of perovskite light emitters		Nature Reviews Materials, 2022, DOI: 10.1038/s41578-022-00459-4	Review of commercialization strategies including surface treatments like TPPO.
12	The regulatory effect of triphenylphosphine oxide on	Review	Journal of Materials Chemistry C, 2021, DOI: 10.1039/D1TC01320C	TPPO functions as an interfacial modulator that enhances perovskite film crystallinity and

	perovskites for morphological			improves its photoluminescence performance
	and radiative improvement			by effective surface passivation.
13	Exploring the Critical Factors	11.5(477nm,1.82		
	Toward Spectrally Stable	time)		TPPO effectively passivates surface defects
	Mixed-Halide Blue Perovskite			and suppresses halide segregation, which
	LEDs		ChemPhotoChem,2024, DOI:	stabilizes the emission spectrum and
			10.1002/cptc.202400019	improves efficiency
14	Phosphonate/Phosphine Oxide	11(474nm,3.54ti		TPPO acts as a surface passivating agent that
	Dyad Additive for Efficient and	mes)		suppresses halide ion migration and non-
	Spectrally Stable Blue			radiative recombination, thereby enhancing
	Perovskite Light-Emitting		Angewandte Chemie,2022. DOI:	the spectral stability and efficiency of blue
	Diodes		10.1002/ange.202117374	perovskite LEDs.
15	Lewis Base Passivation of	PL intensity (10		TPPO functions as a strong Lewis base that
	Quasi-2D Ruddlesden–Popper	times)		passivates surface defects and suppresses
	Perovskite for Order of			halide ion migration, thereby significantly
	Magnitude Photoluminescence		ACS Applied Electronic	enhancing the efficiency and spectral stability
	Enhancement and Improved		Materials,2021, DOI:	of perovskite LEDs.
	Stability		10.1021/acsami.2c19123	
16.	Edge stabilization in reduced-	14%(517nm,5.6ti		TPPO passivates Pb <sup>2+</sup> defects and suppresses
	dimensional perovskites" was	me)		non-radiative recombination, boosting
	published in Nature		Nature Communications, 2020, 11,	efficiency and spectral stability in perovskite
	Communications,		170, DOI: 10.1021/acsami.2c19123	LEDs
17	A bilateral interfacial	18.7(520nm,		TPPO passivates interfacial defects and
	passivation strategy promoting	2.43times)		suppresses non-radiative recombination,
	efficiency and stability of			thereby enhancing the efficiency, spectral
	perovskite quantum dot light-		Nature Communications,11,2020,	stability, and operational lifetime of
	emitting diodes		DOI: 10.1038/s41467-020-17633-3	perovskite LEDs

18.	All-Thermally Evaporated Blue	2.47(475nm)		TPPO passivates interfacial defects and
	Perovskite Light-Emitting			inhibits ion migration, thereby enhancing
	Diodes for Active Matrix		Small Methods, 2024.,	efficiency and spectral stability in perovskite
	Displays		DOI:10.1002/smtd.202300712	LEDs.
19.	Additives in Halide Perovskite	7.8(480nm,2.3ti		TPPO passivates defects and modulates
	for Blue-Light-Emitting Diodes:	mes)		crystallization, enhancing efficiency and
	Passivating Agents or		ACS Energy Letters, 2021,	spectral stability in perovskite LEDs.
	Crystallization Modulators?		DOI:10.1021/acsenergylett.1c02232	
20	Multi-site anchoring lead-halide	20.6(514nm,		Benzylphosphonic acid (BPA) enhances
	octahedral by benzylphosphonic	2.6times)		perovskite LED performance by anchoring
	acid to regulate phase			lead-halide octahedra to regulate phase
	distribution for efficient			distribution and passivate defects, resulting in
	PeLEDs		Nano Research ,2024,	improved efficiency and stability
			DOI:10.1007/s12274-024-6914-9	
21	All-Thermally Evaporated			These results highlight the potential of all-
	Perovskite LEDs Toward High-			thermally evaporated processes for the
	Resolution Active-Matrix		Matter.2023, DOI:	scalable production of high-performance blue
	Displays		10.1016/j.matt.2023.05.041	PeLEDs suitable for display application

Perovskite Solar cells										
Architecture	Jsc	Voc(V)	FF(%)	PCE(%)	Reference					
	$(mA cm^{-2})$									
n-i-p	23.7	1.131	79	21.2	Advanced Materials, 2020, 32(23): 1907396.					
<b>p-i-n,</b> MAPbI <sub>3</sub>	21.32	0.94	72	14.9	Optical Materials, 2022, 127: 112264					
n-i-p CsPbBr <sub>3</sub>	7.87	1.28	72	7.28	ACS Applied Materials & Interfaces, 2024,					
					16(50): 69410-69417					
n-i-p MAFACs	23.9	1.15	73.5	20.3	Chemical Engineering Journal, 2020, 396:					
					125010					

n-i-p FAMA-	25.25	1.13	79.3	22.63	Applied Physics Letters, 2025, 126(2)
4FTPPO					
p-i-n MA-free	25.57	1.214	83.8	26.01	This Work

Samples	Intensity of suns	$R_{\rm ct}(\rm ohm)$	$R_{\rm rec}({\rm ohm})$	
	1	340	124.4	
Control	0.8	601	152	
	0.4	1178	213.9	
	0.2	2024	417	
	0.1	3199	905	
	1	309.7	110	
	0.8	406.3	150	
Target	0.4	725	183.1	
	0.2	1246	328.4	
	0.1	2317	714.3	

Table S2 Fitting all parameters of the TRPL spectra. Bimolecular  $(k_B)$ , trapping  $(k_T)$ , and

detrapping ( $k_D$ ), accompanied by trap state density ( $N_T$ ) and doping concentration ( $p_0$ ) extracted from 'Bimolecular-Trapping-Detrapping Model' in 'PEARS (https://pears-tool.herokuapp.com/)'.

Samples	$k_{\rm B} \times 10^{-21}$	$k_{\rm T} \times 10^{-15}$	$K_{\rm D} \times 10^{-1}$	$N_{\rm T} \times 10^{13}$	$P_0 \times 10^{15}$
	(cm <sup>3</sup> /ns)	(cm <sup>3</sup> /ns)	19	$(cm^3)$	$(cm^3)$
			(cm <sup>3</sup> /ns)		
Control	7.57	1.49	9.7	112	6.5
Target	2.07	1.97	113	4	1.4

Table S3 Fitting parameters of the TRPL spectra for the control and TPPO-modified films.

Samples	Photoexcited carrier	Trapping (%)	Bimolecular (%)	Detrapping
	concentrations $(10^{16} \text{cm}^{-3})$			(%)
	4.40	69.77	9.20	21.03
Control	8.34	64.51	15.54	19.95
	17.9	58.19	25.61	16.20
	4.40	46.58	12.39	41.03
Target	8.34	41.82	20.83	37.35
	17.9	34.34	35.07	30.59

Table S4 EIS parameters of the PSCs under different intensity of suns

Devices Structure	Perovskite composition	Voc deficit s	V <sub>oc</sub> [V]	J <sub>SC</sub> [mA cm <sup>-2</sup> ]	FF [%]	PCE [%]	Ref.
ITO/ NiO <sub>X</sub> /Me-4PACz /PVK/C <sub>60</sub> /BCP/Bi/Ag	$FA_{0.95}Cs_{0.05}PbI_{3}$	0.338	1.192	26.47	84.11	26.54	4
p-i-n (1.52) FTO/MeO-2PACz/PVK/C <sub>60</sub> /BCP/Cu	FAPbI <sub>3</sub>	0.4	1.12	25.5	82	24.1	5
p-i-n (1.53) FTO/MeO-2PACz/PVK/C <sub>60</sub> /BCP/Ag	(FAPbI <sub>3</sub> ) <sub>0.95</sub> (MAPbBr <sub>3</sub> ) <sub>0.05</sub>	0.35	1.18	25.9	85.4	26	6
p-i-n (1.52) ITO/SAM/PVK/C <sub>60</sub> /BCP/Au	$FA_{0.72}Cs_{0.1}MA_{0.18}PbI_3$	0.35	1.17	25.4	84	25.1	7
p-i-n (1.55) ITO/PTAA/PVK/Fc-derived/C <sub>60</sub> /BCP/Ag	$\frac{Cs_{0.05}(FA_{0.98}MA_{0.02})_{0.95}Pb(I_{0.98}Br_{0.02})}{_3}$	0.356	1.194	25.93	84.24	26.08	8
p-i-n (1.55) ITO/NiOX/MeO- PhpPACz/PVK/PI/C <sub>60</sub> /BCP/Cu	$Cs_{0.05}MA_{0.1}FA_{0.85}PbI_{3}\\$	0361	1.189	25.37	86.79	26.17	9
p-i-n (1.55) ITO/NiOX/PTAA/Al <sub>2</sub> O <sub>3</sub> /PVK/PCBM/BCP/A g	FA <sub>0.95</sub> Cs <sub>0.05</sub> PbI <sub>3</sub>	0.346	1.184	26.11	84.2	26.03	10
p-i-n (1.55) FTO/2PACz/2D/3D-PVK/2D/C <sub>60</sub> /BCP/Cu	$Cs_{0.01}(FA_{0.98}MA_{0.02})_{0.95}PbI_3$	0.372	1.178	25.67	86.47	26.15	11
p-i-n (1.53) ITO/2PACz/2D/PVK/2D/C <sub>60</sub> /SnO <sub>2</sub> /IZO/Cu	$Cs_{0.025}MA_{0.075}FA_{0.90}PbI_{3} \\$	0.34	1.19	24.94	85.9	25.63	12
p-i-n (1.53) ITO/2PACz/PVK/2D/C <sub>60</sub> /SnO <sub>2</sub> /IZO/Cu	$Cs_{0.01}(FA_{0.97}MA_{0.03})_{0.99}Pb(I_{0.97}Br_{0.03})_{3}$	0.345	1.185	25.41	84.6	25.42	13
p-i-n (1.54) FTO/MeO-2PACz/PVK/PCBM/YbOx/Cu	$Rb_{0.05}Cs_{0.05}MA_{0.05}FA_{0.85}Pb(I_{0.95}Br_{0.05})_3$	0.38	1.16	26.1	83	25.2	14
p-i-n (1.55) ITO/NiOx/PTAA/Al <sub>2</sub> O <sub>3</sub> /PVK/PCBM/BCP/Ag	$Cs_{0.05}FA_{0.95}PbI_3$	0.38	1.17	26.2	82.2	25.12	15
p-i-n (1.53) ITO/p-PY/PVK/PCBM/BCP/Ag	$FA_{0.98}Cs_{0.02}PbI_{3}$	0.363	1.167	25.75	81.5	24.5	16
p-i-n (1.544) ITO/DMAcPA/PVK/PCBM/BCP/Ag	$Cs_{0.055}(FA_{0.95}MA_{0.05})_{0.945}Pb(I_{0.9}Br_{0.1})_3$	0.355	1.189	25.69	84.91	25.86 (certi fied 25.39 )	17
p-i-n (1.563) ITO/MeO-4PACz@PVK/C <sub>60</sub> /BCP/Ag	$Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}Pb(I_{0.95}Br_{0.05})_3$	0.362	1.201	24.8	84.5	25.16	.18
p-i-n (1.55) ITO/ DC-PA /PVK/C <sub>60</sub> /BCP/Ag	$Cs_{0.06}MA_{0.14}FA_{0.80}PbI_{3} \\$	0.36	1.19	24.55	84.78	24.8 (certi fied2 4.5)	19
p-i-n (1.53) FTO/NiOx/Me-4PACz/(with or without) DPPP/PVK/PEAI/C <sub>60</sub> /SnO <sub>2</sub> /Ag	FA <sub>0.95</sub> Cs <sub>0.05</sub> PbI <sub>3</sub> +0.6mol% MAPbBr <sub>3</sub>	0.37	1.16	25.73	82.5	24.5	20
p-i-n (1.55) ITO/PTAA/PVK/C <sub>60</sub> /BCP/Ag	$\frac{Cs_{0.05}(FA_{0.98}MA_{0.02})_{0.95}Pb(I_{0.98}Br_{0.02})}{_3}$	0.366	1.184	25.68	82.32	25	21
p-i-n (1.56) ITO/MeO-4PACz@PVK/C <sub>60</sub> /BCP/Ag	$\frac{Cs_{0.05}(FA_{0.98}MA_{0.02})_{0.95}Pb(I_{0.98}Br_{0.02})}{3}$	0.37	1.19	24.78	83.07	24.5	22
p-i-n (1.55) ITO/MeO-2PACz/PVK/PCBM/BCP/Ag	$\frac{Cs_{0.05}(FA_{0.98}MA_{0.02})_{0.95}Pb(I_{0.98}Br_{0.02})}{_3}$	0.373	1.177	24.8	84.3	24.6	23
p-i-n (1.563) ITO/Me-4PACz@PVK/C <sub>60</sub> /BCP/Ag	$Cs_{0.05}(FA_{0.95}MA_{0.05})_{0.95}Pb(I_{0.95}Br_{0.05})_3$	0.355	1.208	25.08	84.37	25.56	24
p-i-n (1.56) ITO/PTAA/PVK/C <sub>60</sub> /BCP/Cu	$Cs_{0.05}(FA_{0.92}MA_{0.08})_{0.95}Pb(I_{0.92}Br_{0.08})_{3.}$	0.39	1.17	24.1	81.6	23	25
p-i-n (1.55) ITO/PTAA-P1/PVK/C <sub>60</sub> /BCP/Ag	$Cs_{0.05}(FA_{0.92}MA_{0.08})_{0.95}Pb(I_{0.92}Br_{0.08})_3$	0.38	1.17	25.50	83.28	24.89	26
p-i-n (1.55) ITO/2PACZ/PVK/C <sub>60</sub> /BCP/Ag	$Cs_{0.03}(FA_{0.90}MA_{0.10})_{0.97}PbI_3$	0.35	1.20	24.7	82	24.3	27

Table S5 Performance parameters of p-i-n solar cells with  $\sim 1.55~eV$  bandgap

p-i-n (1.55) ITO/Me-4PACz@MPA/PVK/PCBM/BCP/Ag	$Cs_{0.1}FA_{0.90}PbI_3$	0.336	1.214	25.57	83.8	26.01	This
p-i-n (1.55) ITO/Me-4PACz@MPA/PVK/PCBM/BCP/Ag	Cs <sub>0.1</sub> FA <sub>0.90</sub> PbI <sub>3</sub> (active layer 1cm <sup>2</sup> )	0.320	1.230	25.55	78	24.51	work

# Table S6. PV Parameters of the Best-Performing Perovskite Solar Cells

Daviasa	Scan		EE (0/)	I ( A?)	PCE	Hysteresis
Devices	direction	$V_{\rm OC}(\mathbf{V})$	FF (%)	$J_{\rm SC} ({\rm mA~cm^2})$	(%)	Index
Control	Forward	1.155	78.9	25.40	23.15	0.027
	Reverse	1.161	80.8	25.30	23.79	
Target	Forward	1.195	83.4	25.61	25.52	0.019
	Reverse	1.214	83.8	25.57	26.01	

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