

Supporting Information

Reduction and adsorption capacities of soils for Cr(VI) and quantitative contributions of key influencing factors

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Table S1 Summary of existing studies on Cr(VI) sorption in soils

Soil Type	Soil pH	Reaction conditions (time, temperature, soil/water ratio)	Initial Cr(VI) concentration (mg/L)	Analysis Type of sorption capacity	Sorption capacity (mg/kg)	Conclusion
Soils from 35 DoD facilities across the continental U.S. (Jardine et al. 2013)	3.15-9.01	3 replicate of 48 h; 25 ± 1 °C; 1:10	250	Saturated	50-750	Cr(VI) sorption increased with increasing soil TOC and decreasing pH.
Paddy soil (Li et al. 2012)	5.26-5.57	24 h; 25 ± 1 °C; 1:10	0–100	Equilibrium	0-300	Cr(VI) adsorption increased with increasing initial Cr(VI) concentration.
SOM-removed soil samples (Gu et al. 2017a)	3.0 - 10.0	48 h, 25 °C, 1:200	0.538 – 27.2	Equilibrium	0-1414	Cr(VI) retention was strongly dependent on the pH of the reaction solution and Fe oxide content.
Soil from 7 different regions of China (Liu et al. 2021)	6.89-8.21	12 h, /, 1:10	5-400	Equilibrium	/	Fe _D was the most crucial factor influencing Cr(VI) adsorption, while soil texture controlled its transport; OM and Al _{OX} were secondary factors.
Cinnamon soil, fluvo-aquic soil, and loess soil. (Liu et al. 2019)	8.26-8.50	24 h; 25 °C; 1:25	160	Equilibrium	180-220	Cr(VI) adsorption equilibrium time and capacity varied with soil type and soil layer.

References

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