

## Supplemental information for

### Measurement of octanol-air and air-water partition coefficients of isopropyl and isobutyl nitrate

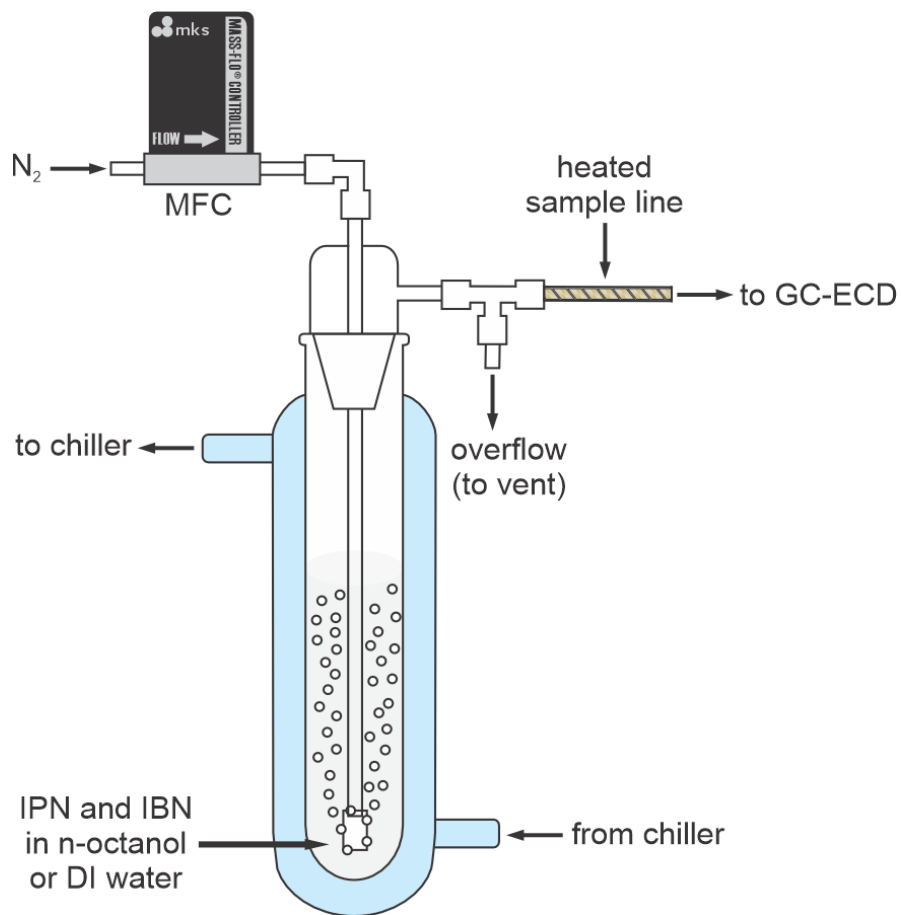
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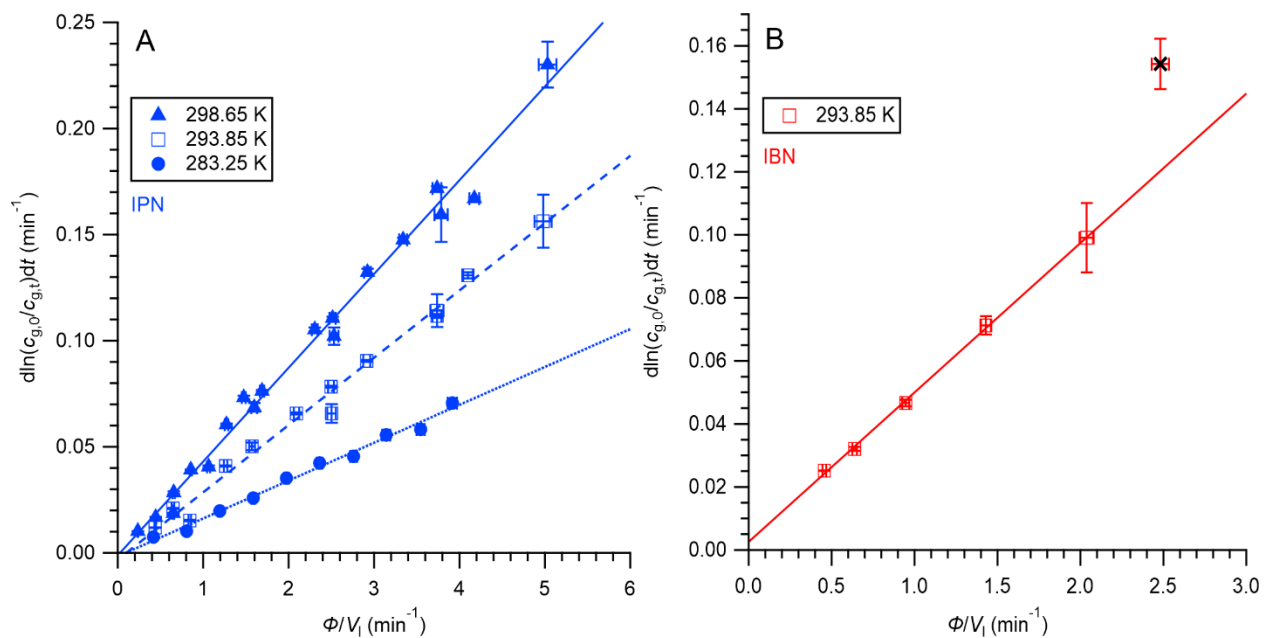
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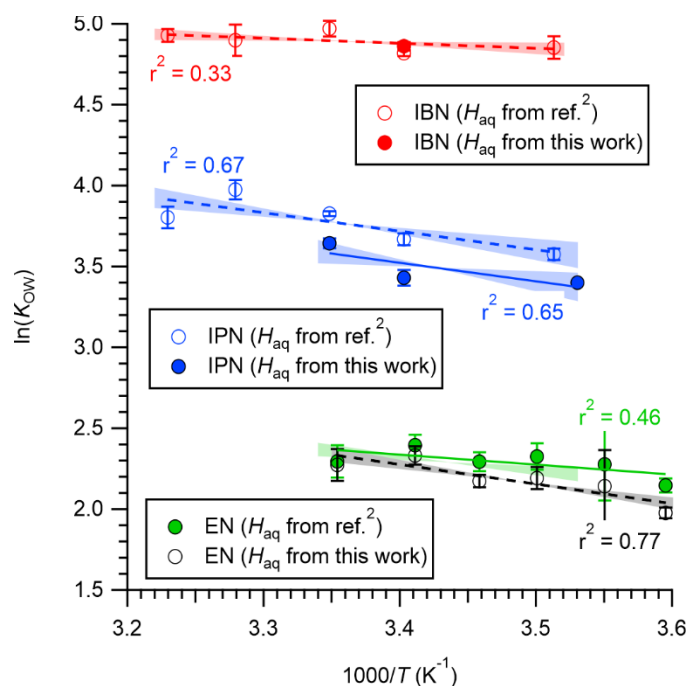
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30 **Figure S1. Schematic of the jacketed bubble column apparatus** (not to scale). MFC = mass flow controller. GC-ECD = gas chromatograph with electron capture detection. DI water = deionized water.



35 **Figure S2. Plots of  $\frac{d\ln(c_{g,0}/c_{g,t})}{dt}$  versus  $\Phi/V_1$  for (A) IPN and (B) IBN in DI water.** The IBN datum suspected to have not achieved equilibrium (crossed out in Table S3) is shown as a bolded cross ( $\times$ ) symbol. The straight lines are linear fits to the data at each temperature. Error bars are  $\pm 1\sigma$  precision.



40 **Figure S3. Natural logarithm of the octanol-water partition coefficients ( $K_{OW}$ ) of IPN, IBN, and EN**  
**as a function of  $1000/T$ .** All  $K_{OW}$  values shown were calculated according to Eq. (2). For IPN (blue colour)  
 and IBN (red colour), the solid symbols ( $\bullet$ ) represent  $K_{OW}$  values calculated using  $H_{oct}$  and  $H_{aq}$  values  
 measured in this work. For EN, the solid symbols ( $\bullet$ , green colour) represent  $K_{OW}$  values calculated using  
 $H_{oct}$  and  $H_{aq}$  measured by Easterbrook et al.<sup>1</sup> The open symbols ( $\circ$ ) represent  $K_{OW}$  values calculated using  
 45  $H_{aq}$  values derived from the temperature dependence parameterizations of IPN, IBN, and EN reported by  
 Kames and Schurath.<sup>2</sup> The lines are linear fits to the data. The shaded regions represent upper and lower  
 error bounds for each linear fit, i.e.,  $\ln(K_{OW}) = (m+\sigma_m)/T + (b-\sigma_b)$  and  $\ln(K_{OW}) = (m-\sigma_m)/T + (b+\sigma_b)$ . Error  
 bars are  $\pm 1\sigma$  precision.

50 **Table S1. Overview of experimental conditions for IPN and IBN in n-octanol.** Uncertainties are at the  $\pm 1\sigma$  level. The  $\ln(c_{g,0}/c_{g,t})/dt$  data marked with a dash (-) indicates that the specified compound was not present.

Internal reference	$T_{\text{setpoint}}$ (°C)	$T_{\text{corrected}}$ (°C)	$V_1$ (mL)	$\Phi$ (mL min <sup>-1</sup> )	$\frac{\Phi}{V_1}$ (min <sup>-1</sup> )	$\frac{d}{dt} \ln\left(\frac{c_{g,0}}{c_{g,t}}\right)$ for IPN (10 <sup>3</sup> min <sup>-1</sup> )	r (%)	$\frac{d}{dt} \ln\left(\frac{c_{g,0}}{c_{g,t}}\right)$ for IBN (10 <sup>3</sup> min <sup>-1</sup> )	r (%)
KE 240705-3	25.00±0.01	25.5±0.5	100.0±0.6	190±1	1.90±0.02	2.391±0.008	99.73	1.094±0.003	99.69
KL 240708-3	25.00±0.01	25.5±0.5	100.0±0.6	159±1	1.59±0.01	-	-	0.752±0.007	99.55
KL 240710-2	25.00±0.01	25.5±0.5	100.0±0.6	505±2	5.05±0.03	0.005.81±0.01	99.99	-	-
KL 240711-1	25.00±0.01	25.5±0.5	100.0±0.6	316±1	3.16±0.02	3.87±0.02	99.93	-	-
KL 240711-2	25.00±0.01	25.5±0.5	100.0±0.6	565±2	5.65±0.04	6.44±0.02	99.96	2.412±0.006	99.97
KL 240712-2	25.00±0.01	25.5±0.5	100.0±0.6	253±1	2.53±0.02	3.109±0.006	99.97	1.140±0.002	99.95
KL 240717	25.00±0.01	25.5±0.5	100.0±0.6	65±1	0.65±0.01	0.713±0.003	99.83	-	-
KL 240718	25.00±0.01	25.5±0.5	100.0±0.6	439±2	4.39±0.03	5.23±0.01	99.98	1.890±0.003	99.99
KL 240719	25.00±0.01	25.5±0.5	150.0±2.0	65±1	0.43±0.01	0.513±0.001	99.96	-	-
KL 240722	25.00±0.01	25.5±0.5	150.0±2.0	347±1	2.31±0.03	2.70±0.01	99.94	0.955±0.003	99.95
KL 240724	25.00±0.01	25.5±0.5	150.0±2.0	147±1	0.98±0.01	1.115±0.001	99.99	0.383±0.003	99.75
KL 240725-1	25.00±0.01	25.5±0.5	150.0±2.0	537±2	3.58±0.05	4.25±0.01	99.98	1.531±0.007	99.95
KL 240725-2	25.00±0.01	25.5±0.5	150.0±2.0	443±2	2.96±0.04	3.489±0.006	99.99	1.328±0.003	99.99
KL 240726	25.00±0.01	25.5±0.5	150.0±2.0	224±1	1.49±0.02	1.793±0.004	99.91	0.6484±0.0009	99.97
KL 240729	25.00±0.01	25.5±0.5	150.0±2.0	570±2	3.80±0.05	4.554±0.008	99.99	1.632±0.002	99.99
AB 241009-1	20.00±0.01	20.7±0.5	100.0±2.0	403±2	4.03±0.08	3.78±0.01	99.97	1.424±0.008	99.83
AB 241010-1	20.00±0.01	20.7±0.5	100.0±2.0	95±1	0.95±0.02	0.86±0.03	98.28	0.31±0.03	91.10
AB 241010-2	20.00±0.01	20.7±0.5	100.0±2.0	187±1	1.87±0.04	1.765±0.006	99.93	0.620±0.002	99.95
AB 241011-1	20.00±0.01	20.7±0.5	100.0±2.0	64±1	0.64±0.01	0.613±0.002	99.95	0.221±0.001	99.68
AB 241015-1	20.00±0.01	20.7±0.5	100.0±2.0	498±2	4.98±0.10	4.68±0.02	99.97	1.79±0.01	99.86
AB 241126-2	37.00±0.01	36.5±0.5	100.0±2.0	68±1	0.68±0.02	1.530±0.010	99.87	0.91±0.03	96.12
AB 241127-1	37.00±0.01	36.5±0.5	100.0±2.0	426±2	4.26±0.09	9.5±0.2	99.76	3.38±0.07	99.51
AB 241127-2	37.00±0.01	36.5±0.5	100.0±2.0	133±1	1.33±0.03	2.82±0.01	99.97	1.26±0.03	98.19
AB 241128-1	37.00±0.01	36.5±0.5	100.0±2.0	589±2	5.89±0.12	13.4±0.2	99.76	4.9±0.1	99.11
AB 241129-1	37.00±0.01	36.5±0.5	100.0±2.0	296±1	2.96±0.06	7.3±0.1	99.29	2.57±0.02	99.84
KE 250106-2	32.00±0.01	31.8±0.5	100.0±2.0	130±1	1.30±0.03	1.91±0.05	97.17	0.555±0.009	98.66
KE 250107-1	32.00±0.01	31.8±0.5	100.0±2.0	576±2	5.76±0.12	8.31±0.06	99.93	3.53±0.08	98.83
KE 250107-2	32.00±0.01	31.8±0.5	100.0±2.0	387±2	3.87±0.08	5.73±0.05	99.85	3.1±0.1	93.88

55 **Table S1 (continued).** Overview of experimental conditions for IPN and IBN in n-octanol. Uncertainties are at the  $\pm 1\sigma$  level. The  $\ln(c_{g,0}/c_{g,t})/dt$  data marked with a dash (-) indicates that the specified compound was not present.

Internal reference	$T_{\text{setpoint}}$ (°C)	$T_{\text{corrected}}$ (°C)	$V_1$ (mL)	$\Phi$ (mL min <sup>-1</sup> )	$\frac{\Phi}{V_1}$ (min <sup>-1</sup> )	$\frac{d}{dt} \ln\left(\frac{c_{g,0}}{c_{g,t}}\right)$ for IPN (10 <sup>3</sup> min <sup>-1</sup> )	r (%)	$\frac{d}{dt} \ln\left(\frac{c_{g,0}}{c_{g,t}}\right)$ for IBN (10 <sup>3</sup> min <sup>-1</sup> )	r (%)
KE 250108-1	32.00±0.01	31.8±0.5	100.0±2.0	481±2	4.81±0.10	7.08±0.06	99.83	2.77±0.04	98.92
KE 250109-1	32.00±0.01	31.8±0.5	100.0±2.0	323±1	3.23±0.07	4.68±0.03	99.91	1.82±0.02	98.97
KE 250110-1	32.00±0.01	31.8±0.5	100.0±2.0	229±1	2.29±0.05	3.55±0.02	99.88	1.323±0.005	99.87
KE 250113-1	32.00±0.01	31.8±0.5	100.0±2.0	419±2	4.19±0.09	6.42±0.05	99.88	2.33±0.03	99.50
KE 250121-1	10.00±0.01	11.5±0.5	100.0±2.0	299±1	2.99±0.06	2.04±0.02	99.63	0.75±0.02	94.97
KE 250122-1	10.00±0.01	11.5±0.5	100.0±2.0	475±2	4.75±0.10	3.17±0.03	99.87	1.04±0.01	99.24
KE 250123-1	10.00±0.01	11.5±0.5	100.0±2.0	359±1	3.59±0.07	2.48±0.05	99.45	0.81±0.02	97.38
KE 250124-1	10.00±0.01	11.5±0.5	100.0±2.0	180±1	1.80±0.04	1.214±0.008	99.63	0.415±0.007	97.77
KE 250125-1	10.00±0.01	11.5±0.5	100.0±2.0	417±2	4.17±0.08	2.74±0.02	99.82	0.93±0.01	98.27
KE 250127-1	10.00±0.01	11.5±0.5	100.0±2.0	240±1	2.40±0.05	1.70±0.01	99.71	0.58±0.01	97.73
KE 250129-1	20.00±0.01	20.7±0.5	100.0±2.0	252±1	2.52±0.05	-	-	0.80±0.02	96.75
KE 250130-1	20.00±0.01	20.7±0.5	100.0±2.0	562±2	5.62±0.11	5.98±0.04	99.95	2.06±0.05	99.24
KE 250131-1	20.00±0.01	20.7±0.5	100.0±2.0	157±1	1.57±0.03	1.64±0.01	99.81	0.550±0.003	99.50
KE 250203-1	20.00±0.01	20.7±0.5	100.0±2.0	339±1	3.39±0.07	3.50±0.04	99.15	1.23±0.01	99.45
KE 250204-1	20.00±0.01	20.7±0.5	100.0±2.0	280±1	2.80±0.06	2.84±0.02	99.87	1.007±0.006	99.75
KE 250205-1	20.00±0.01	20.7±0.5	100.0±2.0	96±1	0.96±0.02	1.010±0.006	99.85	0.37±0.01	95.61
KE 250206-1	20.00±0.01	20.7±0.5	100.0±2.0	465±2	4.65±0.09	5.01±0.04	99.90	1.67±0.01	99.66
KE 250207-1	37.00±0.01	36.5±0.5	100.0±2.0	229±1	2.29±0.05	4.97±0.05	99.86	1.77±0.01	99.73
KE 250210-1	37.00±0.01	36.5±0.5	100.0±2.0	488±2	4.88±0.10	9.51±0.09	99.88	4.06±0.05	99.76
KE 250211-1	37.00±0.01	36.5±0.5	100.0±2.0	362±1	3.62±0.07	7.28±0.06	99.91	3.14±0.03	99.78
KE 250212-1	32.00±0.01	31.8±0.5	100.0±2.0	164±1	1.64±0.03	3.22±0.04	99.71	1.11±0.02	98.28
KE 250213-1	32.00±0.01	31.8±0.5	100.0±2.0	67±1	0.67±0.02	1.131±0.005	99.80	0.443±0.003	98.73
KE 250218-1	32.00±0.01	31.8±0.5	100.0±2.0	360±1	3.60±0.07	6.39±0.09	99.79	2.41±0.05	97.06

**Table S2. Overview of experimental conditions for IPN in DI water.** Uncertainties are at the  $\pm 1\sigma$  level.

Internal reference	$T_{\text{setpoint}}$ (°C)	$T_{\text{corrected}}$ (°C)	$V_1$ (mL)	$\Phi$ (mL min <sup>-1</sup> )	$\frac{\Phi}{V_1}$ (min <sup>-1</sup> )	$\frac{d}{dt} \ln \left( \frac{c_{g,0}}{c_{g,t}} \right)$ (min <sup>-1</sup> )	r (%)
KL 240731	25.00±0.01	25.5±0.5	150.0±2.0	67±1	0.44±0.01	0.0167±0.0002	99.89
KL 240801-1	25.00±0.01	25.5±0.5	150.0±2.0	626±3	4.17±0.06	0.1670±0.0006	100.0
KL 240801-2	25.00±0.01	25.5±0.5	150.0±2.0	253±2	1.69±0.02	0.0763±0.0004	99.98
KL 240801-3	25.00±0.01	25.5±0.5	150.0±2.0	377±2	2.51±0.04	0.1106±0.0008	99.98
KL 240801-4	25.00±0.01	25.5±0.5	150.0±2.0	97±1	0.65±0.01	0.0186±0.0003	99.81
KL 240802-1	25.00±0.01	25.5±0.5	150.0±2.0	128±1	0.85±0.01	0.0392±0.0002	99.99
KL 240802-2	25.00±0.01	25.5±0.5	150.0±2.0	35±1	0.23±0.01	0.01029±0.00003	99.91
KL 240806-1	25.00±0.01	25.5±0.5	150.0±2.0	346±2	2.31±0.03	0.1052±0.0010	99.97
KL 240806-2	25.00±0.01	25.5±0.5	150.0±2.0	191±1	1.27±0.02	0.0605±0.0005	99.99
KL 240806-3	25.00±0.01	25.5±0.5	150.0±2.0	501±2	3.34±0.05	0.1477±0.0010	99.99
KL 240806-4	25.00±0.01	25.5±0.5	150.0±2.0	159±1	1.06±0.02	0.0406±0.0006	99.94
KL 240806-5	25.00±0.01	25.5±0.5	150.0±2.0	439±2	2.92±0.04	0.132±0.002	99.56
KL 240806-6	25.00±0.01	25.5±0.5	150.0±2.0	221±1	1.48±0.02	0.0732±0.0009	99.97
KL 240807-1	25.00±0.01	25.5±0.5	150.0±2.0	560±2	3.74±0.05	0.172±0.001	99.98
KL 240807-2	8.50±0.01	10.1±0.5	150.0±2.0	588±3	3.92±0.05	0.0706±0.0002	99.99
KL 240807-4	8.50±0.01	10.1±0.5	150.0±2.0	471±2	3.14±0.04	0.0557±0.0003	99.99
KL 240807-5	8.50±0.01	10.1±0.5	150.0±2.0	296±2	1.97±0.03	0.03519±0.00006	99.99
KL 240808-1	8.50±0.01	10.1±0.5	150.0±2.0	354±2	2.36±0.03	0.0424±0.0004	99.98
KL 240808-2	8.50±0.01	10.1±0.5	150.0±2.0	179±1	1.19±0.02	0.01977±0.00008	99.98
KL 240808-3	8.50±0.01	10.1±0.5	150.0±2.0	238±2	1.58±0.02	0.0259±0.0001	99.99
KL 240808-4	8.50±0.01	10.1±0.5	150.0±2.0	121±1	0.81±0.01	0.01024±0.00008	99.93
KL 240808-5	8.50±0.01	10.1±0.5	150.0±2.0	414±2	2.76±0.04	0.0455±0.0001	99.98
KL 240809-1	8.50±0.01	10.1±0.5	150.0±2.0	62±1	0.42±0.01	0.00756±0.00007	99.95
KL 240809-2	8.50±0.01	10.1±0.5	150.0±2.0	532±2	3.55±0.05	0.0582±0.0003	99.98
KL 240809-3	20.00±0.01	20.7±0.5	150.0±2.0	615±3	4.10±0.06	0.131±0.001	99.96
KL 240809-4	20.00±0.01	20.7±0.5	150.0±2.0	126±1	0.84±0.01	0.0152±0.0004	99.67
KL 240809-6	20.00±0.01	20.7±0.5	150.0±2.0	65±1	0.44±0.01	0.01186±0.00006	99.98
KL 240812-1	20.00±0.01	20.7±0.5	150.0±2.0	374±2	2.50±0.04	0.0783±0.0005	99.98
KL 240812-2	20.00±0.01	20.7±0.5	150.0±2.0	189±1	1.26±0.02	0.0410±0.0002	99.99

60 **Table S2 (continued). Overview of experimental conditions for IPN in DI water. Uncertainties are at the  $\pm 1\sigma$  level.**

Internal reference	$T_{\text{setpoint}}$ (°C)	$T_{\text{corrected}}$ (°C)	$V_1$ (mL)	$\Phi$ (mL min <sup>-1</sup> )	$\frac{\Phi}{V_1}$ (min <sup>-1</sup> )	$\frac{d}{dt} \ln \left( \frac{c_{g,0}}{c_{g,t}} \right)$ (min <sup>-1</sup> )	r (%)
KL 240812-3	20.00±0.01	20.7±0.5	150.0±2.0	560±2	3.74±0.05	0.112±0.001	99.98
KL 240812-4	20.00±0.01	20.7±0.5	150.0±2.0	314±2	2.09±0.03	0.0657±0.0005	99.98
KL 240812-5	20.00±0.01	20.7±0.5	150.0±2.0	437±2	2.91±0.04	0.0905±0.0004	99.98
KE 250704-1	25.00±0.01	25.5±0.5	100.0±2.0	253±1	2.53±0.05	0.102±0.004	99.76
KE 250704-2	25.00±0.01	25.5±0.5	100.0±2.0	378±2	3.78±0.08	0.16±0.01	99.68
KE 250704-3	25.00±0.01	25.5±0.5	100.0±2.0	65±1	0.65±0.02	0.0284±0.0006	99.78
KE 250707-1	25.00±0.01	25.5±0.5	100.0±2.0	503±2	5.03±0.10	0.23±0.01	99.89
KE 250707-2	25.00±0.01	25.5±0.5	100.0±2.0	160±1	1.60±0.03	0.069±0.002	99.83
KE 250708-1	20.00±0.01	20.7±0.5	100.0±2.0	374±2	3.74±0.08	0.114±0.008	99.55
KE 250708-2	20.00±0.01	20.7±0.5	100.0±2.0	250±1	2.50±0.05	0.066±0.004	99.56
KE 250708-3	20.00±0.01	20.7±0.5	100.0±2.0	65±1	0.65±0.02	0.0209±0.0003	99.83
KE 250709-1	20.00±0.01	20.7±0.5	100.0±2.0	498±2	4.98±0.10	0.16±0.01	99.68
KE 250709-2	20.00±0.01	20.7±0.5	100.0±2.0	157±1	1.57±0.03	0.050±0.002	99.58

**Table S3. Overview of experimental conditions for IBN in DI water. Uncertainties are at the  $\pm 1\sigma$  level.**

Internal reference	$T_{\text{setpoint}}$ (°C)	$T_{\text{corrected}}$ (°C)	$V_1$ (mL)	$\Phi$ (mL min <sup>-1</sup> )	$\frac{\Phi}{V_1}$ (min <sup>-1</sup> )	$\frac{d}{dt} \ln \left( \frac{c_{g,0}}{c_{g,t}} \right)$ (min <sup>-1</sup> )	r (%)
<del>KE 250714-1</del> *	<del>20.00±0.01</del>	<del>20.7±0.5</del>	<del>100.0±2.0</del>	<del>248±1</del>	<del>2.48±0.05</del>	<del>0.154±0.008</del>	<del>99.47</del>
KE 250714-2	20.00±0.01	20.7±0.5	100.0±2.0	95±1	0.95±0.02	0.047±0.001	99.89
KE 250714-5	20.00±0.01	20.7±0.5	100.0±2.0	64±1	0.64±0.01	0.0321±0.0006	99.88
KE 250715-1	20.00±0.01	20.7±0.5	100.0±2.0	204±1	2.04±0.04	0.10±0.01	98.78
KE 250715-2	20.00±0.01	20.7±0.5	100.0±2.0	143±1	1.43±0.03	0.071±0.003	99.83
KE 250715-3	20.00±0.01	20.7±0.5	100.0±2.0	46±1	0.46±0.01	0.02521±0.00007	100.0

65 \* not used in the analysis.

**Table S4. Linear regression analyses of  $d/dt(\ln(A_0/A_t))$  versus  $\Phi/V_1$  for the data shown in Figures 3 and S2. Uncertainties are at the  $\pm 1\sigma$  level.**

Compound	Solvent	$T$ (K)	Slope	Intercept ( $\text{min}^{-1}$ )	$r$ (%)
IPN	n-octanol	284.65	$(0.65 \pm 0.02) \times 10^{-3}$	$(1.0 \pm 0.8) \times 10^{-4}$	99.76
		293.85	$(1.03 \pm 0.04) \times 10^{-3}$	$-(0.6 \pm 1.3) \times 10^{-4}$	99.37
		298.65	$(1.16 \pm 0.02) \times 10^{-3}$	$(0.7 \pm 0.6) \times 10^{-4}$	99.86
		304.95	$(1.42 \pm 0.09) \times 10^{-3}$	$(4.0 \pm 3.0) \times 10^{-4}$	98.57
		309.65	$(2.16 \pm 0.14) \times 10^{-3}$	$(0.4 \pm 5.2) \times 10^{-4}$	98.72
IBN	n-octanol	284.65	$(2.05 \pm 0.14) \times 10^{-4}$	$(0.8 \pm 0.5) \times 10^{-4}$	99.04
		293.85	$(3.67 \pm 0.07) \times 10^{-4}$	$-(0.3 \pm 0.2) \times 10^{-4}$	99.83
		298.65	$(4.14 \pm 0.20) \times 10^{-4}$	$-(0.3 \pm 0.2) \times 10^{-4}$	98.93
		304.95	$(6.25 \pm 0.61) \times 10^{-4}$	$-(0.2 \pm 2.1) \times 10^{-4}$	96.40
		309.65	$(7.76 \pm 0.30) \times 10^{-4}$	$(2.3 \pm 1.1) \times 10^{-4}$	99.54
IPN	DI water	283.25	$(1.78 \pm 0.06) \times 10^{-2}$	$-(1.6 \pm 1.5) \times 10^{-3}$	99.51
		293.85	$(3.17 \pm 0.10) \times 10^{-2}$	$-(3.1 \pm 2.8) \times 10^{-3}$	99.46
		298.25	$(4.42 \pm 0.12) \times 10^{-2}$	$-(1.1 \pm 2.9) \times 10^{-3}$	99.42
IBN	DI water	293.25	$(4.74 \pm 0.08) \times 10^{-2}$	$(2.6 \pm 1.0) \times 10^{-3}$	99.96

**Table S5. Dimensionless Henry's law constants of EN, IPN, and IBN. Uncertainties are at the  $\pm 1\sigma$  level. n/d = not determined.**

Compound	$T$ (K) / Reference	$H_S^{cc}$ (unitless)					
		283.25	284.65	293.85	298.65	304.95	309.65
EN in n-octanol	1	896 $\pm$ 200 (281.7 K)	n/d	535 $\pm$ 30 (293.2 K)	378 $\pm$ 37 (298.2 K)	n/d	n/d
IPN in n-octanol	This work	n/d	1547 $\pm$ 53	972 $\pm$ 36	864 $\pm$ 13	706 $\pm$ 43	463 $\pm$ 30
IBN in n-octanol	This work	n/d	4869 $\pm$ 339	2726 $\pm$ 50	2417 $\pm$ 119	1599 $\pm$ 156	1289 $\pm$ 50
EN in DI water	1	91.9 $\pm$ 1.1 (281.7 K)	n/d	48.8 $\pm$ 1.5 (293.2 K)	38.0 $\pm$ 0.7 (298.2 K)	n/d	n/d
IPN in DI water	This work	56.0 $\pm$ 2.0	n/d	31.5 $\pm$ 1.0	22.6 $\pm$ 0.6	n/d	n/d
IBN in DI water	This work	n/d	n/d	21.1 $\pm$ 0.4	n/d	n/d	n/d

75 **Table S6. Liquid-phase loss rate constants of EN, IPN and IBN. Uncertainties are at the  $\pm 1\sigma$  level. n/d = not determined.**

Compound	$T$ (K) / Reference	$k_1$ ( $10^{-5} \text{ s}^{-1}$ )					
		283.25	284.65	293.85	298.65	304.95	309.65
EN in n-octanol	1	0.8±1.9 (281.7 K)	n/d	-0.9±1.3 (293.2 K)	-4.7±2.3 (298.2 K)	n/d	n/d
IPN in n-octanol	This work	n/d	0.17±0.13	-0.10±0.21	0.11±0.09	0.67±0.50	0.07±0.86
IBN in n-octanol	This work	n/d	0.13±0.08	-0.06±0.04	0.13±0.11	-0.04±0.36	0.39±0.18
EN in DI water	1	0.4±0.6 (281.7 K)	n/d	0.1±3.8 (293.2 K)	5.8±2.2 (298.2 K)	n/d	n/d
IPN in DI water	This work	-2.7±2.6	n/d	-5.2±4.6	-1.8±4.9	n/d	n/d
IBN in DI water	This work	n/d	n/d	4.3±1.7	n/d	n/d	n/d

## References

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