

Supplementary Information

Harnessing Chirality in Nanomaterials: Advancing Photocatalysis for Hydrogen Production and Beyond

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Table S1 Representative chiral photo(electro)catalytic systems benchmarked across diverse materials and reactions.

Type	Photocatalyst	Chirality source	Light source	Reaction	Performance (metric)	gain	Ref.
Chiral plasmonic	SiO ₂ @Au@TiO ₂ nanoribbons	Tartrate-templated silica helices	CPL	RhB degradation	Matched CPL yields reaction slope -1.97 vs. -0.69 under opposite CPL ($\sim 2.85\times$ enhancement).		[1]
	Gold-gap-silver nanostructures	L-/D-cysteine	CPL/LP/dark	4-nitrophenol reduction	Matched CPL gives $\sim 70\times$ and $17\times$ higher vs. Au and Au@Ag core-shell NPs.		[2]
	Au NP multilayer films	L-/D-phenylalanine	CPL/LP	Photo-response	Photocurrent enhanced by 2.41 under matched CPL; achiral films show little to no change in current.		[3]
	Pd-on-cAu nanorods	L-/D-cysteine	CPL/LP/dark	TMB oxidation	Reaction kinetics maximized under matched CPL relative to opposite CPL, LP and dark conditions.		[4]
	Branched AgAuPt nanoparticles	L-/D-cysteine	CPL/LP/dark	Water splitting	Matched CPL shows reduced overpotentials than LP.		[5]
Chiral carbon nitride	Helical rod-like g-C ₃ N ₄	Chiral mesoporous silica	Visible	HER	7x higher H ₂ evolution rate than bulk g-C ₃ N ₄ .		[6]
	Bioinspired mesoporous chiral g-C ₃ N ₄	Cellulose nanocrystal	Visible	HER	55x higher H ₂ evolution rate than bulk g-C ₃ N ₄ ; achiral samples show no CPL response.		[7]
Chiral inorganic/hybrid semiconductor	TiO _{2-x} /Cu _x O/Pt	Cellulose nanocrystal	Visible	HER	$\sim 1.6\times$ higher H ₂ production than non-reduced analogue and $\sim 46\times$ vs. pristine TiO ₂ .		[8]
	COF-Cu	L-/D-cysteine	Visible	HER	$\sim 4\times$ higher H ₂ evolution rate than achiral COF control.		[9]
	TiO ₂ -SiO ₂	N-acylamino acid C ₁₄ -L-AlaA	Visible	RhB degradation, HER	$\sim 97.9\%$ removal vs. negligible activity for achiral TiO ₂ -SiO ₂ . \sim Substantial HER activity vs. negligible HER activity for achiral TiO ₂ -SiO ₂ .		[10]

	Au/TiO ₂	Cellulose nanocrystal/D-glucose	UV-vis	HER	4x higher H ₂ evolution rate than achiral Au/TiO ₂ and 31x higher than TiO ₂ .	[11]
	Cd-MOFs	N-(4-pyridylmethyl)-L-leucine	AM 1.5G	HER	Highest H ₂ evolution with chloride ligation than formate and acetate Cd-MOF.	[12]
	Pt-CdSe@CdS nanorods	L-/D-cysteine, histidine	LP	HER	Higher H ₂ evolution than racemic and Pt-free controls.	[13]
CISS-based photo-electrode	CdSe/TiO ₂	Organic linkers	Xe lamp	Water splitting	H ₂ evolution occurs at lower overpotential only with chiral linkers; achiral linkers require higher bias.	[14]
	TiO ₂	Zn-porphyrins molecules	Xe lamp	OER	Substantial OER current and H ₂ O ₂ suppressed below detection; achiral analogues produce 43 ± 5 mmol L ⁻¹ peroxide.	[15]
	BiVO ₄ /CoO _x	L-/D-tartaric acid	AM 1.5G	OER	O ₂ FE increased from ~60% to ~80% and H ₂ O ₂ FE reduced from >30% to <10%.	[16]
	ZnO	L-/D-methionine	Xe lamp	OER	H ₂ O ₂ suppressed by ~5.5x and carrier lifetime extended by up to 2.4x vs. DL-ZnO.	[17]
	α-Fe ₂ O ₃	L-cysteine, glutathione	Visible	Photo-response	Photocurrent increased by up to 4x and free carrier density by ~61%.	[18]
	TiS ₂	L-tartaric acid	Single sunlight	OER	Enhanced photocurrent and suppressed peroxide formation vs. non-chiral TiS ₂ .	[19]

CPL: Circularly polarized light

LP: Linearly polarized light

TMB: Tetramethylbenzidine

Table S2 Representative chiral photo(electro)catalytic systems across diverse materials for CO₂RR.

Photocatalyst	Chirality source	Light source	Reaction	Performance gain (metric)	Ref.
Chiral polyoxometalate–Co hybrid	L-/D-1,10-phenanthroline	Visible	CO ₂ RR	CO production rate ~3× higher than achiral analogue with higher AQY.	[20]
CsPbBr ₃ perovskite nanoplates (NPLs)	R-/S-MBA:Br	AM 1.5G	CO ₂ RR	5x CO ₂ RR efficiencies compared pristine CsPbBr ₃ perovskite NPLs.	[21]
D-BiOBr/Au	D-sorbitol	Xe lamp	CO ₂ RR	2.43x higher CO formation rate vs. achiral BiOBr.	[22]
Cu-In ₂ S ₃	L-/D-cysteine	Xe lamp	CO ₂ RR	Spin polarization promotes C–C coupling, yielding ethanol with ~94% selectivity while suppressing C ₁ products to ~6%, compared with ~36–38% for achiral and racemic analogues.	[23]
ZnS	L-/D-cysteine	Xe lamp	CO ₂ RR with NH ₃	Spin polarization promotes C–C coupling while suppressing competing reaction pathways relative to non-chiral analogues, resulting in average yield enhancements of 3.5% for CO and 13.6% for C ₂₊ products.	[24]

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