# **Supporting Information**

Dual Enabling Photomediated Knoevenagel Condensation and Alkene Perfluoroalkylation Reactions by a Photoresponsive Cadmium-Organic Framework

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Fig. S1 Thermogravimetric curve of Cd-MOF



Fig. S2 Infrared spectra of Cd-MOF



Fig. S4 fluorescence quenching experiments (a) 5mmol benzaldehyde; (b) 6mmol Malonitrile; (c) 0.4mmol  $C_4F_9I$ 

Compound	Cd-MOF
Formula	$C_{21}H_{18}N_2O_7Cd$
Formula weight	522.77
Temperature(K)	150.00
Crystal system	Triclinic
Space group	Pī
<i>a</i> (Å)	10.1022(5)
<i>b</i> (Å)	10.2035(4)
<i>c</i> (Å)	11.3950(6)
α (°)	99.416(2)
β (°)	99.647(2)
γ (°)	119.1130(10)
Volume (Å <sup>3</sup> )	969.83(8)
$ ho_{calc}g/cm^3$	1.790
Z	2
F (000)	524.0
Goodness-of-fit on $F^2$	1.063
$P_{\rm M} P_{\rm M} (I > 2 {\rm sigma}(I))$	$R_1 = 0.0279,$
$K_1, WK_2(1 \ge 2 \operatorname{Sigman}(1))$	$wR_2 = 0.0798$
$P_{\rm e} = W P_{\rm e} (all data)$	$R_1 = 0.0300,$
$\kappa_1, w\kappa_2(a)$ (and $a)$	$wR_2 = 0.0812$

Table S1: Crystallographic data of Cd-MOF

Table S2 : Bond lengths and bond angles for Cd-MOF

Cd-MOF				
Cd1-O6A	2.394(2)	O3-Cd1-O5A	136.98(7)	
Cd1-O6B	2.581(2)	O3-Cd1-N2	133.51(8)	
Cd1-O3	2.1824(19)	O2-Cd1-O6B	171.69(6)	
Cd1-O2	2.326(2)	O2-Cd1-O6A	108.57(7)	

Cd1-O5A	2.399(2)	O3-Cd1-O5A	80.19(7)
Cd1-N2	2.216(2)	O5A-Cd1-O6B	108.11(7)
O6A-Cd1-O6B	76.86(7)	N2-Cd1-O6A	128.45(8)
O6A-Cd1-O5A	54.55(7)	N2-Cd1-O6B	82.78(8)
O3-Cd1-O6A	88.56(7)	N2-Cd1-O2	98.03(8)
O3-Cd1-O6B	79.69(7)	N2-Cd1-O5A	89.40(8)
O3-Cd1-O2	93.97(8)		

#### 1. Calculation of green indicators[1, 2]:

(1). Environmental Factor (E-Factor): The amount of waste generated in the chemical process. The higher the value, the greater the amount of waste.

 $E = \frac{Waste\ quality}{Product\ quality} = \frac{Total\ quality\ of\ raw\ materials\ -\ Product\ quality}{Product\ quality} \times 100\%$ 

(2). Atom Economy (AE): Atomic economy, the number of atoms of the starting material contained in the reaction product.

 $AE = \frac{Molecular \ weight \ of \ product}{The \ sum \ of \ the \ molecular \ weights \ of \ the \ stoichiometric \ reactants} \times 100\%$ 

(3). Carbon Efficiency (CE): The carbon generated in the product accounts for the percentage of carbon in the reactant.

 $CE = rac{Carbon \ content \ in \ products}{Total \ carbon \ content \ in \ raw \ materials}$ 

(4). Reaction mass efficiency (RME): Reaction quality efficiency, evaluation of

reaction cleanliness. The higher the RME value, the cleaner the reaction, and the more

conducive to green chemistry.

 $RME = \frac{quality \ of \ product}{Total \ mass \ of \ chemically \ measured \ reactants}$ 

(5). Mass intensity (MI): Mass Intensity, lower MI values indicate lower costs and more sustainable chemical processes (excluding solvents).

 $MI = \frac{Total \ mass \ in \ the \ chemical \ process}{Quality \ of \ products}$ 

(6). Process mass intensity (PMI): Process quality strength.

 $PMI = \frac{Total \ mass \ in \ the \ chemical \ process \ (including \ solvent)}{Total \ mass \ of \ product}$ 

#### 2. Theoretical calculations

Table S3: Calculated thermodynamic data for pathway II					
ZPE	E+ZPE	G	HF	total free	
				energy	
Bn-O-	0 107572	245 212705	245 244994	245 (412525	245 5647605
radical-	0.10/5/2 -345.313/95	-343.344004 -343.0412323		-343.304/093	

anion					
BOH-	0 101005	245 964095	245 20(544	246 1104595	246 0200225
radical	0.121085	-343.804983	-343.896344	-346.1194383	-346.0299325
Malonitril	0.000116	224 20 521 4	224 222574	004 5154560	224 5125002
e-anion	0.032116	-224.305214	-224.3325/4	-224.5174562	-224.5127002
Malonitril	~ ~				
e	0.045441	-224.845386	-224.873072	-224.9861538	-224.9683988

	read	ctant	pro	duct	ΔGr
	Bn-O-		Malanimila		
II	radical-	Malonitrile	Malonitrile-	BOH-radical	-5.939000912
	anion		anion		

**3. NMR data for the compounds:** 



#### 3a: Benzalmalononitrile (PE: EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-d) δ 7.91 (d, J = 7.5 Hz, 2H), 7.79 (s, 1H), 7.64 (t, J = 7.5 Hz, 1H), 7.54 (t, J = 7.7 Hz, 2H), 7.26 (s, 1H).

<sup>13</sup>C NMR (101 MHz, CHLOROFORM-D) δ 160.11, 134.79, 131.04, 130.88, 129.77, 113.84, 112.68, 82.97。



## 3b: 2-(4-hydroxybenzylidene) malononitrile (PE: EA=10:1)

<sup>1</sup>H NMR (400 MHz, DMSO-D6/ TMS): δ 11.07 (s, 1H), 8.30 (s, 1H), 7.89 (d, J=8Hz, 2H), 6.99-6.95 (m, 2H).

<sup>13</sup>C NMR (100 MHz, CDCl3): δ 163.91, 160.52, 133.89, 122.81, 116.63, 115.11, 114.20, 75.12.[3]



### 3c: 2-(4-methoxybenzylidene) malononitrile (PE: EA=80:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.91 (d, *J* = 6.7 Hz, 2H), 7.65 (s, 1H), 7.01 (d, *J* = 8.9 Hz, 2H), 3.91 (s, 3H).

<sup>13</sup>C NMR (101 MHz, CHLOROFORM-*D*) δ 164.60, 158.72, 133.26, 123.77, 114.91, 114.24, 113.16, 78.81, 55.60.



3d: 2-(4-(trifluoromethyl) benzylidene) malononitrile (PE: EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.02 (d, *J* = 8.2 Hz, 2H), 7.85 (s, 1H), 7.81 (d, *J* = 8.3 Hz, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.20, 135.64, 133.80, 130.88, 126.79, 126.75, 126.71, 126.67, 121.84, 113.10, 112.02, 86.15.



## 3e: 2-(4-nitrobenzylidene)malononitrile (PE:EA=20:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.39 (d, *J* = 6.9 Hz, 2H), 8.08 (d, *J* = 6.9 Hz, 2H), 7.89 (s, 1H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 157.33, 150.76, 136.23, 131.79, 125.12, 113.08, 112.05, 87.98



### 3f: 2-(4-chlorobenzylidene)malononitrile (PE:EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.86 (d, *J* = 8.8 Hz, 2H), 7.74 (s, 1H), 7.52 (d, *J* = 8.7 Hz, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.14, 140.98, 131.67, 129.90, 129.06, 113.26, 112.16, 83.11



## 3g: 2-(3-chlorobenzylidene) malononitrile (PE: EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  7.84 (d, *J* = 6.8 Hz, 2H), 7.73 (s, 1H), 7.60 (d, *J* = 8.6 Hz, 1H), 7.50 (t, *J* = 8.3 Hz, 1H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.05, 135.60, 134.19, 132.09, 130.71, 130.26, 128.15, 113.01, 111.84, 84.44



3h: 2-(2-chlorobenzylidene) malononitrile (PE: EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.26 (s, 1H), 8.18 (d, *J* = 7.6, 3.9 Hz, 1H), 7.55 (qd, *J* = 4.4, 3.6, 2.1 Hz, 2H), 7.48 – 7.40 (m, 1H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 155.87, 136.09, 134.83, 130.46, 129.25, 128.78, 127.56, 112.99, 111.68, 85.47.



## 3j: 2-(naphthalen-1-ylmethylene) malononitrile (PE: EA=100:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*)  $\delta$  8.67 (s, 1H), 8.28 (d, *J* = 7.4 Hz, 1H), 8.12 (d, *J* = 8.3 Hz, 1H), 7.96 (d, *J* = 8.5 Hz, 2H), 7.69 (t, *J* = 6.9 Hz, 1H), 7.67 – 7.58 (m, 2H). <sup>13</sup>C NMR (101 MHz, Chloroform-*d*)  $\delta$  157.58, 134.77, 133.34, 130.88, 129.26, 128.40, 128.36, 127.31, 127.12, 125.22, 122.11, 113.54, 112.32, 84.97



3k: 2-(furan-2-ylmethylene) malononitrile (PE: EA=80:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.80 (d, *J* = 1.7 Hz, 1H), 7.52 (s, 1H), 7.35 (d, *J* = 3.8 Hz, 1H), 6.74 – 6.69 (m, 1H).

<sup>13</sup>C NMR (101 MHz, CHLOROFORM-*D*) δ 149.74, 148.15, 143.22, 123.77, 114.58, 113.95, 112.74, 77.50



31: 2-(thiophen-2-ylmethylene) malononitrile (PE: EA=60:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.92 – 7.84 (m, 2H), 7.79 (d, *J* = 4.1 Hz, 1H), 7.29 – 7.22 (m, 1H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 150.99, 138.11, 136.79, 135.16, 128.83, 113.60, 112.76, 77.97



### 3m: 2-((1-methyl-1H-pyrazol-4-yl) methylene) malononitrile (PE: EA=5:1)

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.21 (s, 1H), 7.97 (s, 1H), 7.68 (s, 1H), 3.99 (s, 3H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 151.03, 142.63, 133.43, 116.15, 113.96, 113.83, 77.84, 39.91.



3n: 2-(quinolin-2-ylmethylene) malononitrile (PE: EA=20:1)

<sup>1</sup>H NMR (400 MHz, CDCl3) δ 8.34 (d, J = 8.2 Hz, 1H), 8.24 (d, J = 7.9 Hz, 1H), 7.99 (s, 1H), 7.90 (d, J = 8.3 Hz, 1H), 7.85 (s, 1H), 7.77 (d, J = 8.4 Hz, 1H), 7.72 (d, J = 6.9 Hz, 1H);

<sup>13</sup>C NMR (150 MHz, CDCl3) δ 157.67, 148.52, 148.34, 137.72, 131.33, 130.67, 129.91, 129.29, 127.80, 122.33, 113.85, 112.31, 87.93;[4]



**6a: 1-methoxy-4-(4,4,5,5,6,6,7,7,7-nonafluoro-2-iodoheptyl) benzene** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.15 – 7.08 (m, 2H), 6.91 – 6.81 (m, 2H), 4.42 (m,1H), 3.81 (s, 3H), 3.18 (m, 2H), 2.88 (m,2H).

<sup>13</sup>C NMR (101 MHz, CHLOROFORM-*D*) δ 158.92, 130.78, 130.18, 114.10, 77.48, 77.16, 76.84, 55.37, 46.32, 40.58, 20.35.

<sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -80.90 (m, 3F), -111.68 – -114.40 (m, 2F), -124.44(m, 2F), -125.83 (m, 2F).

**6b: 1-methoxy-4-(4,5,5,5-tetrafluoro-2-iodo-4-(trifluoromethyl)pentyl)benzene** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.18 – 7.10 (m, 2H), 6.92 – 6.85 (m, 2H), 4.42 (m, 1H), 3.81 (s, 3H), 3.19 (m, 2H), 2.97 – 2.77 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.61, 130.46, 129.86, 117.29-108.45(m,-C<sub>3</sub>F<sub>7</sub>), 113.77, 55.03, 45.98, 40.30, 40.09, 39.88, 20.05.

<sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -80.24(m, 3F), -112.39 – -116.82 (m,2F), -127.79(m, 2F).

6c: 1-(4,4,5,5,6,6,6-heptafluoro-2-iodohexyl)-4-methoxybenzene <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.13 – 7.08 (m, 2H), 6.89 – 6.85 (m, 2H), 4.41 (m, 1H), 3.81 (s, 3H), 3.21 (m, 1H), 3.09 (m, 1H), 2.98 – 2.86 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.61, 130.57, 129.82, 122.13-118.75(m,1C, -CF), 113.74, 93.11-90.41(m, 2C, -2CF<sub>3</sub>), 55.03, 46.43, 46.40, 38.58, 38.40, 22.50
<sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -76.08(s,3F), -77.05(s,3F), -185.15(m,3F)



**6d: 1-methoxy-4-(4,4,5,5,6,6,7,7,8,8,8-undecafluoro-2-iodooctyl) benzene** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.15 – 7.09 (m, 2H), 6.90 – 6.86 (m, 2H), 4.43 (m, 1H),

3.81 (s, 3H), 3.19 (m, 2H), 2.88 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.94, 130.80, 130.19, 120.52-110.70(m,5C, -C<sub>5</sub>F<sub>11</sub>), 114.10, 55.34, 46.33, 40.89, 40.68, 40.47, 20.40.

19F NMR (376 MHz, Chloroform-*d*) δ -80.73(s,3F), -111.39 – -114.25 (m,2F), -122.51(m,2F), -123.74(m,2F), -126.20(m,2F).

**6e: 1-methoxy-4-(4,4,5,5,6,6,7,7,8,8,9,9,9-tridecafluoro-2-iodononyl) benzene** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.16 – 7.10 (m, 2H), 6.92 – 6.85 (m, 2H), 4.43 (m, 1H), 3.81 (s, 3H), 3.26 – 3.10 (m, 2H), 2.98 – 2.77 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.63, 130.49, 129.87, 120.25-110.46(m,6C, -C<sub>6</sub>F<sub>13</sub>), 113.78, 55.02, 46.01, 40.60, 40.39, 40.18, 20.11.

<sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -80.76, -111.41 – -114.23 (m,2F), -121.74, -122.81, -123.56, -126.09.

6f:1-(4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,11-heptadecafluoro-2-iodoundecyl)-4methoxybenzene <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.18 – 7.05 (m, 2H), 6.92 – 6.84 (m, 2H), 4.42 (m, 1H), 3.81 (s, 3H), 3.18 (m, 2H), 2.88 (m, 2H). <sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.96, 130.82, 130.19, 120.58-107.78(m, 8C, -C<sub>8</sub>F<sub>17</sub>), 114.10, 55.34, 46.34, 40.94, 40.73, 40.52, 20.44 <sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -80.76 – -81.27(m,3F), -111.41 – -114.44 (m,2F), -121.53 (m,2F), -121.87 (m,4F), -122.69 (m,2F), -123.51 (m,2F), -126.10 (m,2F).



### 6g:1-(4,4,5,5,6,6,7,7,8,8,9,9,10,10,11,11,12,12,13,13,13-henicosafluoro-2-

iodotridecyl)-4-methoxybenzene <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.14 – 7.11 (m, 2H), 6.89 – 6.86 (m, 2H), 4.46-4.39 (m, 1H), 3.81 (s, 3H), 3.25 – 3.10 (m, 2H), 2.96 – 2.79 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 158.96, 130.84, 130.19, 118.68-110.90(m,10C,

 $-C_{10}F_{21}$ , 114.10, 55.33, 46.32, 40.96, 40.75, 40.54, 20.48

19F NMR (376 MHz, Chloroform-*d*) δ -80.83(s,3F) -111.36 - -114.21 (m,2F), -121.56 - -121.96(m,10F), -122.73, -123.56, -126.15.



**6h: ethyl 2-fluoro-4-iodo-5-(4-methoxyphenyl) pentanoate** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.15 – 7.08 (m, 2H), 6.89 – 6.84 (m, 2H), 4.35 – 4.30 (m, 2H), 3.80 (s, 2H), 3.22 – 3.11 (m, 2H), 2.97 – 2.70 (m, 2H), 1.36 (t, *J* = 7.2 Hz, 3H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 163.56, 163.24, 162.92, 158.48, 130.70, 129.86, 117.52-112.50 (t,-CF<sub>2</sub>), 113.71, 63.08, 55.02, 46.16, 44.12, 43.89, 43.65, 22.59, 13.68.
<sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -100.33 - -102.60 (m,1F), -104.87 - -107.72 (m,1F).



**6i:** (4,4,5,5,6,6,7,7,7-nonafluoro-2-iodoheptyl) benzene <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.37 (m, 1H), 7.36 – 7.29 (m, 2H), 7.24 – 7.19 (m, 2H), 4.48 (m, 1H), 3.32 (m, 1H), 3.21 (m, 1H), 3.01 – 2.79 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 138.40, 128.79, 128.48, 127.19, 120.10-115.69(m,4C, -C<sub>4</sub>F<sub>9</sub>), 46.83, 40.72, 40.52, 40.31, 19.15.

19F NMR (376 MHz, Chloroform-*d*) δ -80.45 – -80.49(m,3F), -110.93 – -114.53 (m,2F), -124.48 – -124.70 (m,2F), -124.73 – -127.11 (m,2F).



**6k: 1,2,3,4,5-pentafluoro-6-(4,4,5,5,6,6,7,7,7-nonafluoro-2-iodoheptyl) benzene** <sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 4.55 – 4.47 (m, 1H), 3.45 – 3.26 (m, 2H), 3.14 – 2.79 (m, 2H).

<sup>13</sup>C NMR (101 MHz, Chloroform-*d*) δ 146.21, 143.72, 139.25, 138.75, 136.23, 118.87-112.66(m, 4C, -C<sub>4</sub>F<sub>9</sub>), 41.83, 41.62, 41.41, 33.58, 14.35. <sup>19</sup>F NMR (376 MHz, Chloroform-*d*) δ -80.73 - -81.49(m,3F), -110.94 - -115.06 (m,2F), -124.16 - -124.40(m,2F), -125.60 - -125.88(m,2F), -142.08 - -142.49 (m,2F), -154.31 - -154.86 (m,1F), -161.44 - -161.86 (m,2F).















Fig.S 8 <sup>13</sup>C NMR spectra of compound 3c









Fig.S 14 <sup>13</sup>C NMR spectra of compound 3f







Fig.S 16<sup>13</sup>C NMR spectra of compound 3g







Fig.S 18  $^{13}\mathrm{C}$  NMR spectra of compound 3h







Fig.S 20 <sup>13</sup>C NMR spectra of compound 3j



Fig.S 21 <sup>1</sup>H NMR spectra of compound 3k



Fig.S 22 <sup>13</sup>C NMR spectra of compound 3k



Fig.S 23 <sup>1</sup>H NMR spectra of compound 31



Fig.S 24 <sup>13</sup>C NMR spectra of compound 31







Fig.S 26 <sup>13</sup>C NMR spectra of compound 3m







Fig.S 28  $^{13}\mathrm{C}$  NMR spectra of compound 6a







Fig.S 30 <sup>1</sup>H NMR spectra of compound 6b







Fig.S 32 <sup>19</sup>F NMR spectra of compound 6b



Fig.S 34 <sup>13</sup>C NMR spectra of compound 6c



Fig.S 35 <sup>19</sup>F NMR spectra of compound 6c



Fig.S 36 <sup>1</sup>H NMR spectra of compound 6d



Fig.S 37 <sup>13</sup>C NMR spectra of compound 6d



Fig.S 38 <sup>19</sup>F NMR spectra of compound 6d







Fig.S 40 <sup>13</sup>C NMR spectra of compound 6e



Fig.S 41 <sup>19</sup>F NMR spectra of compound 6e



Fig.S 42 <sup>1</sup>H NMR spectra of compound 6f



Fig.S 43 <sup>13</sup>C NMR spectra of compound 6f



Fig.S 44 <sup>19</sup>F NMR spectra of compound 6f



Fig.S 45 <sup>1</sup>H NMR spectra of compound 6g



Fig.S 46 <sup>13</sup>C NMR spectra of compound 6g







Fig.S 48 <sup>1</sup>H NMR spectra of compound 6h



Fig.S 49 <sup>13</sup>C NMR spectra of compound 6h



Fig.S 50 <sup>19</sup>F NMR spectra of compound 6h



Fig.S 51 <sup>1</sup>H NMR spectra of compound 6i



Fig.S 52 <sup>13</sup>C NMR spectra of compound 6i



Fig.S 53 <sup>19</sup>F NMR spectra of compound 6i



Fig.S 54 <sup>1</sup>H NMR spectra of compound 6k



Fig.S 55 <sup>13</sup>C NMR spectra of compound 6k



Fig. S 56 <sup>19</sup>F NMR spectra of compound 6k







Fig. S 58 <sup>13</sup>C NMR spectra of H<sub>4</sub>DDB







Fig. S 60 <sup>13</sup>C NMR spectra of 4,4'-bimp

[1] K. Kumari, P. Choudhary, D. Sharma and V. Krishnan, Ind Eng Chem Res 2022,62 158-168.

[2] S. Gajurel, R. Sarkar, F.K. Sarkar, L. Kyndiah and A.K. Pal, ACS omega 2022 7 48087-48099.

[3] B.D. Dond, R.P. Kale and S.N. Thore, Chem Pap 2023 77 2717-2724.

[4] Y. Zhu, H. Li, G. Yan, B. Shi, Y. Zhang, Q. Lin, H. Yao and T. Wei, RSC Adv 2015 5 49953-49957.