

Supplementary Information

Co-recycling of spent cathode and anode via redox-mediated lithiation

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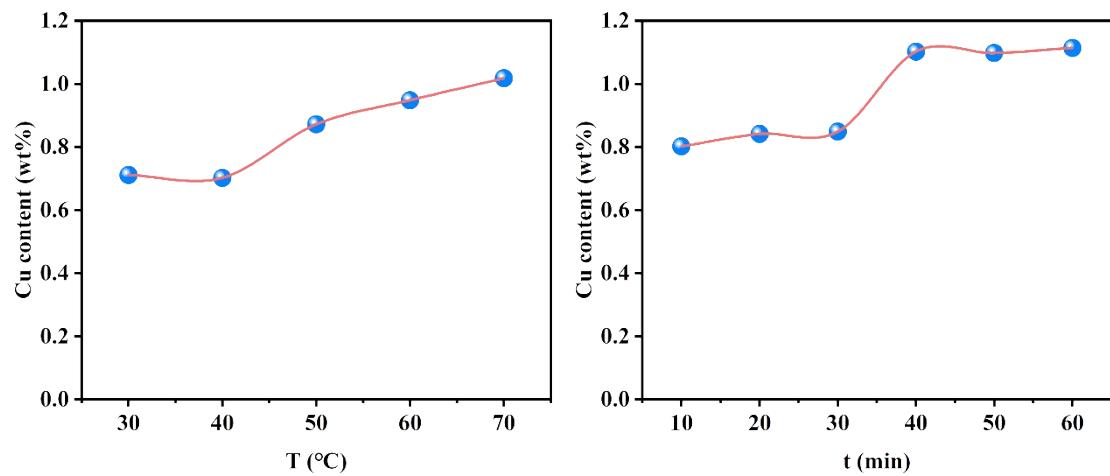
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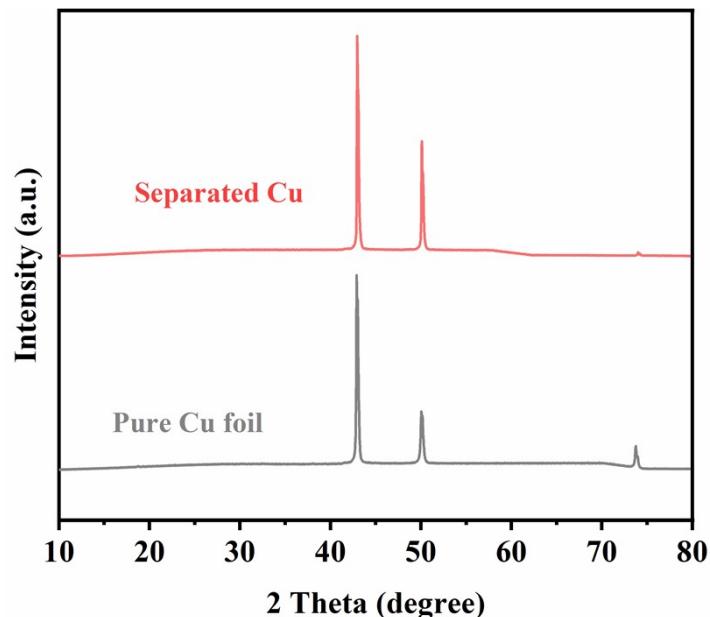
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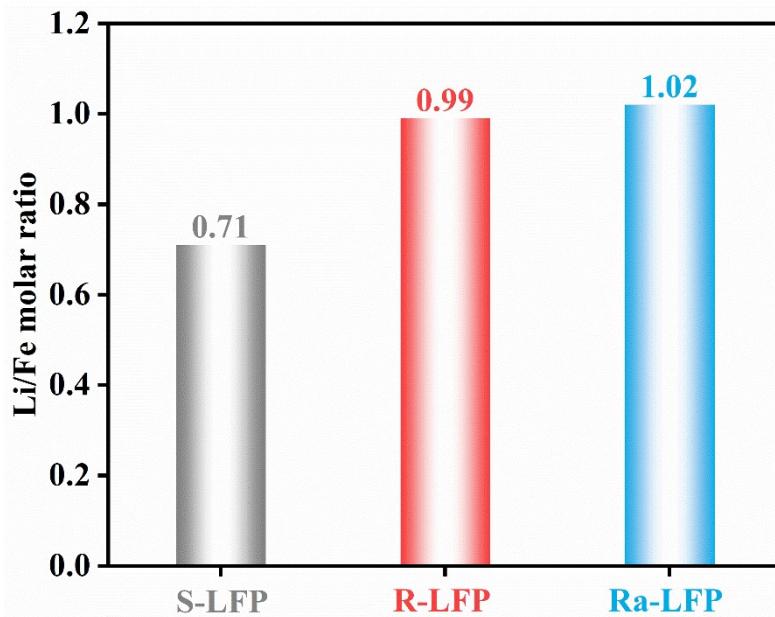
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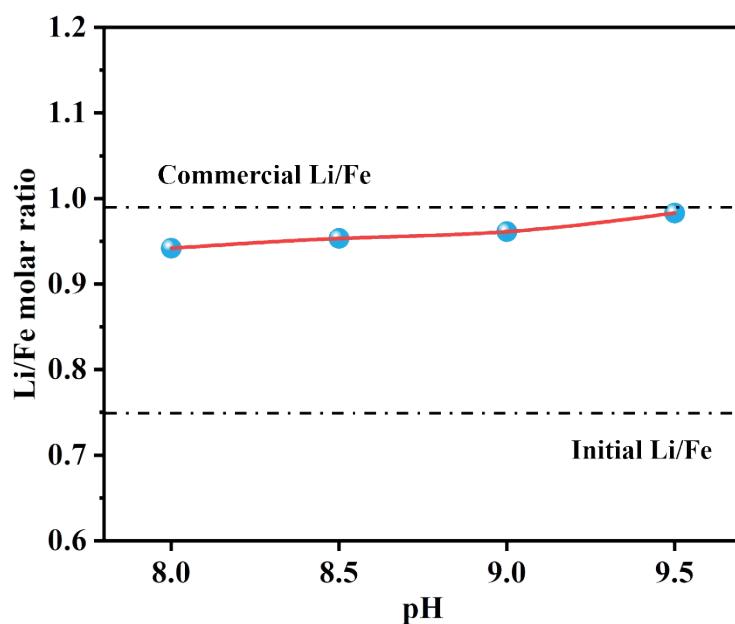
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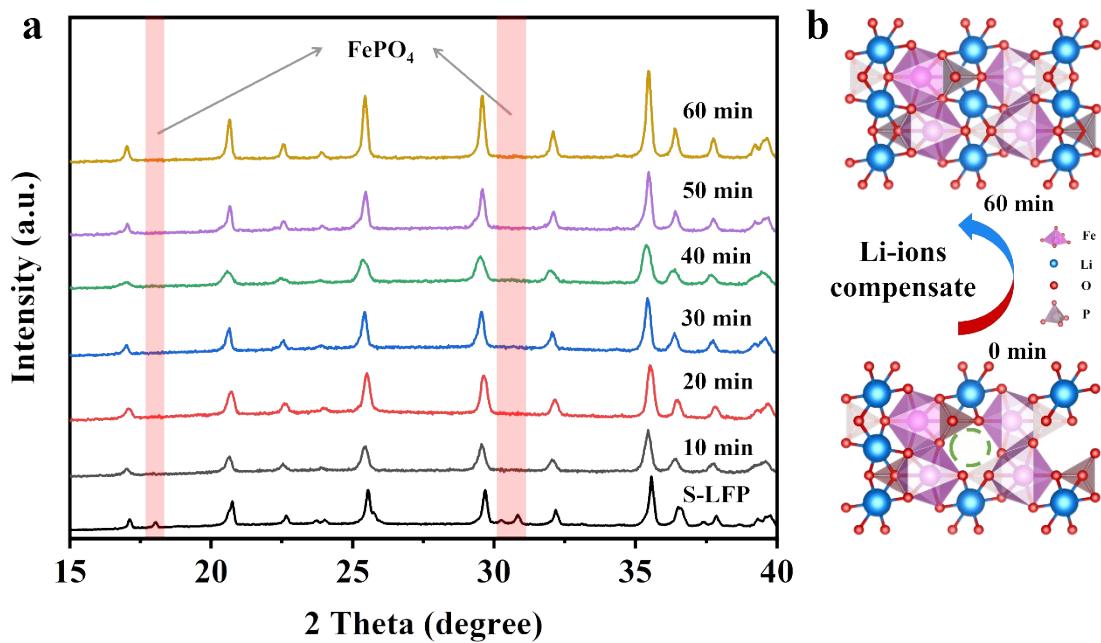
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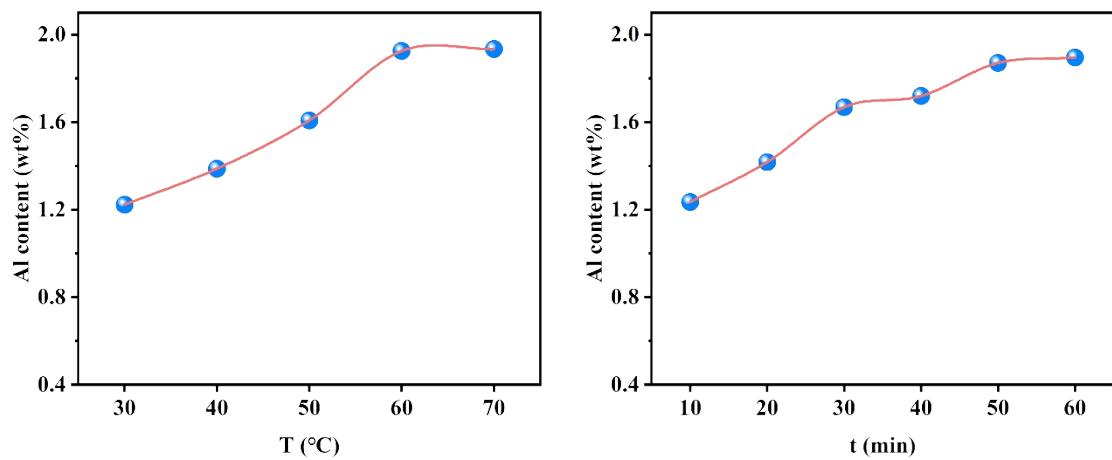
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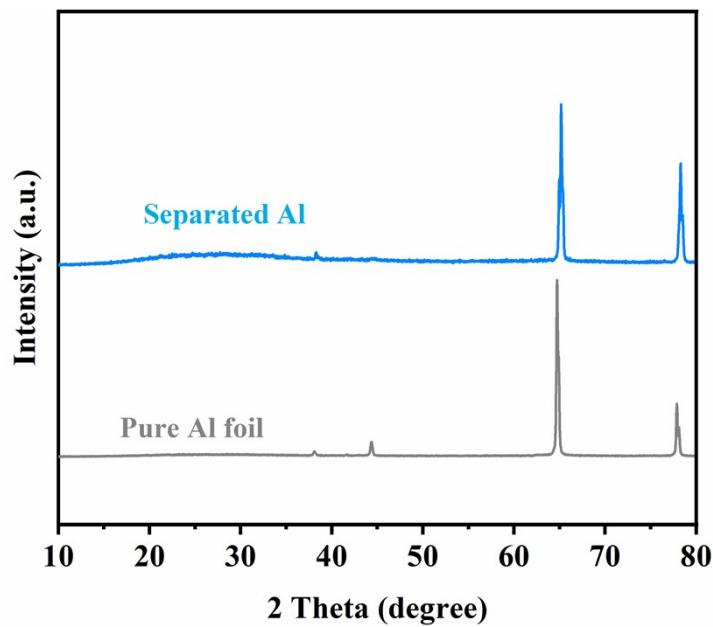
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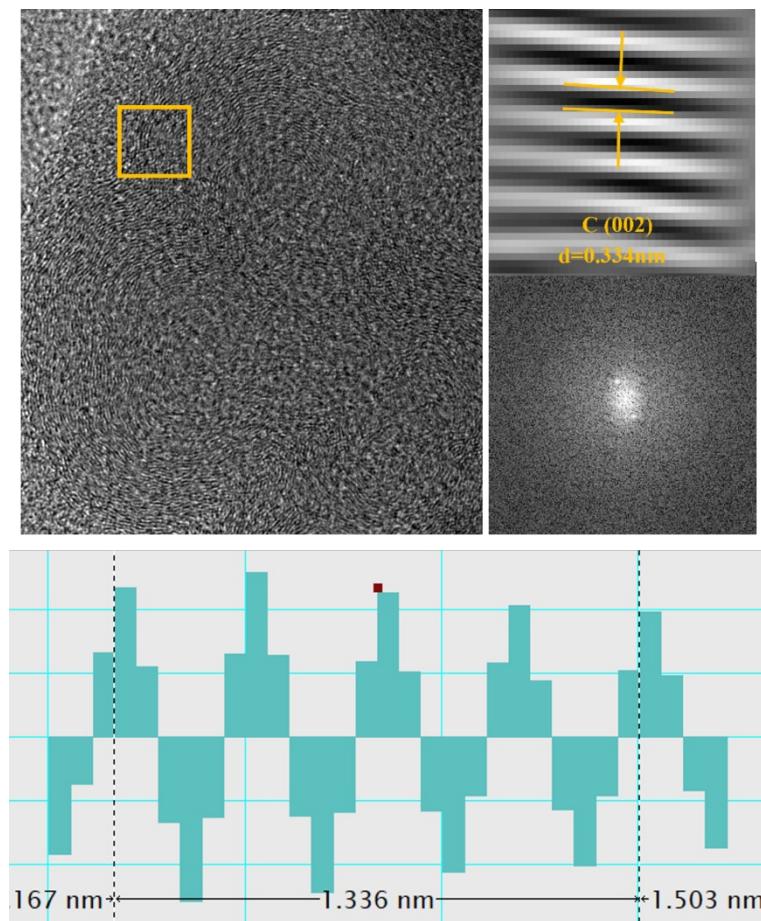
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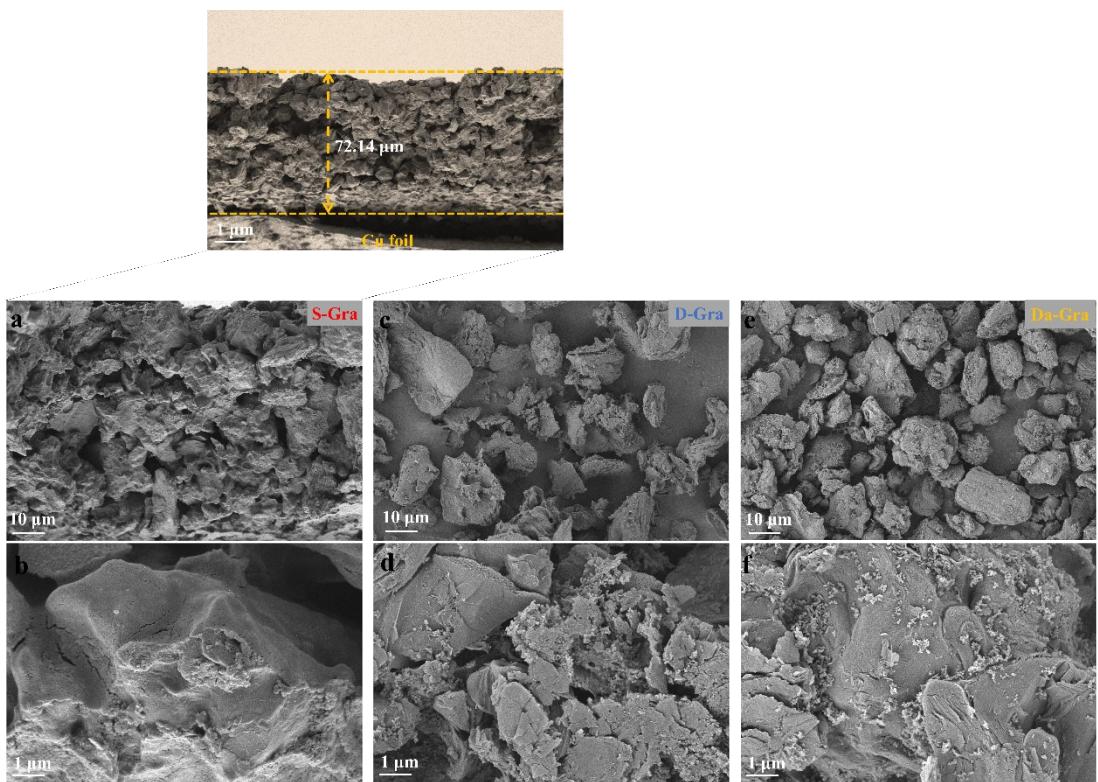
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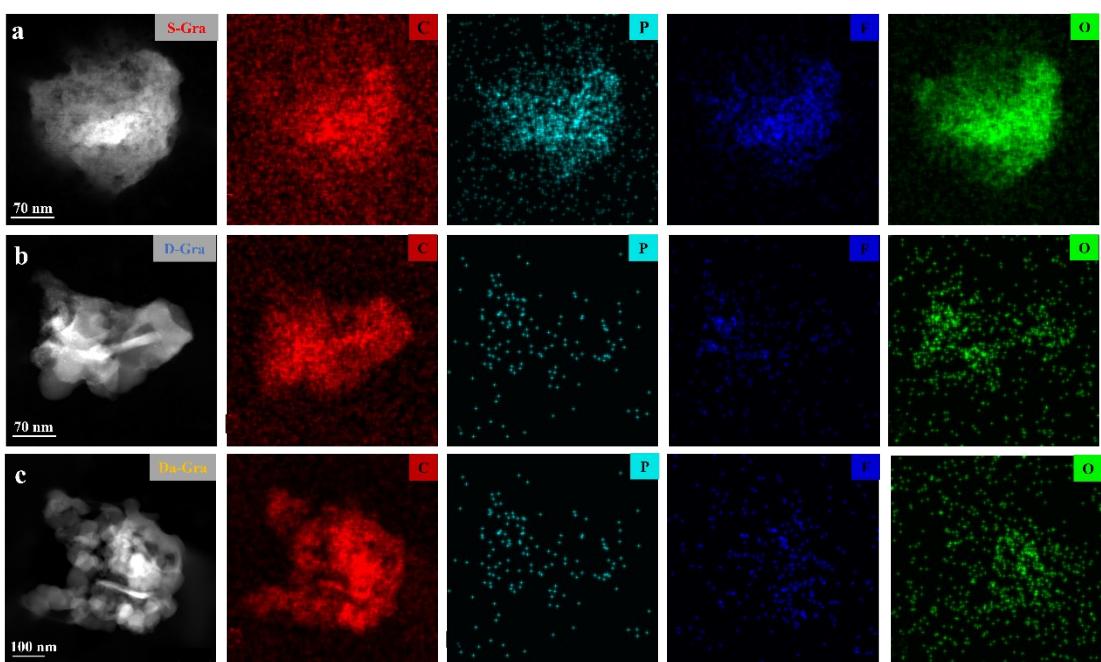
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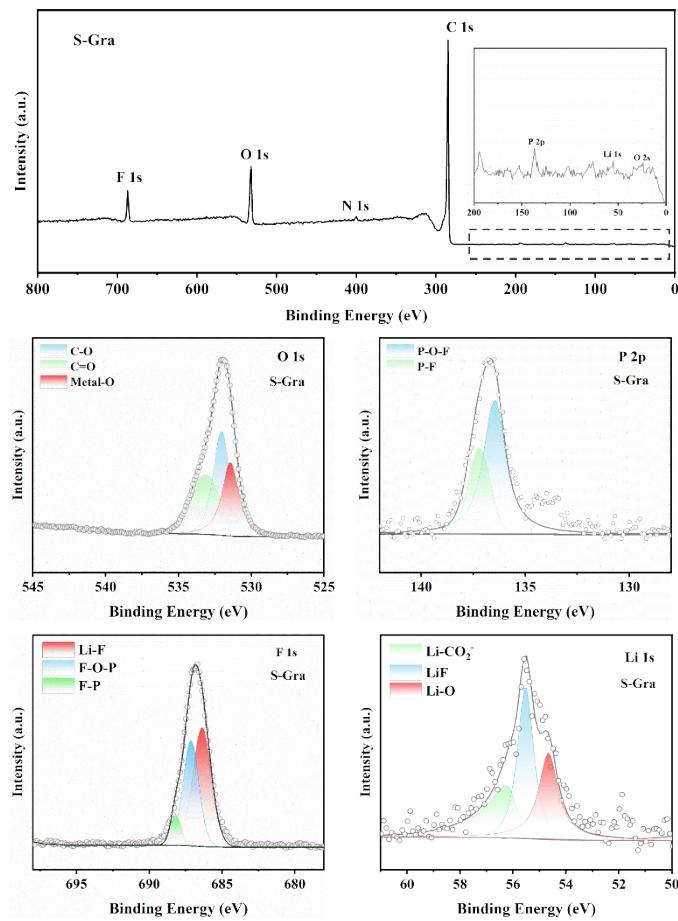
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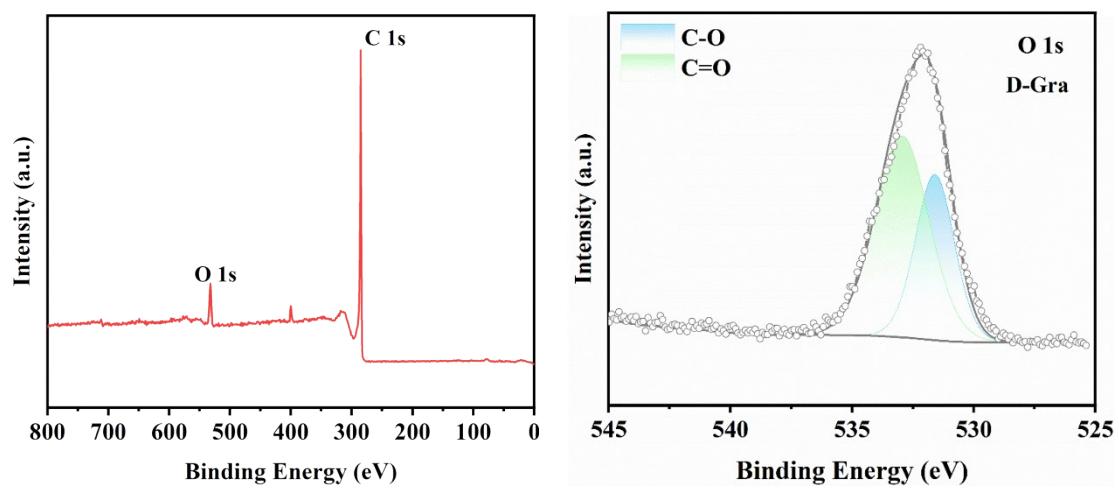
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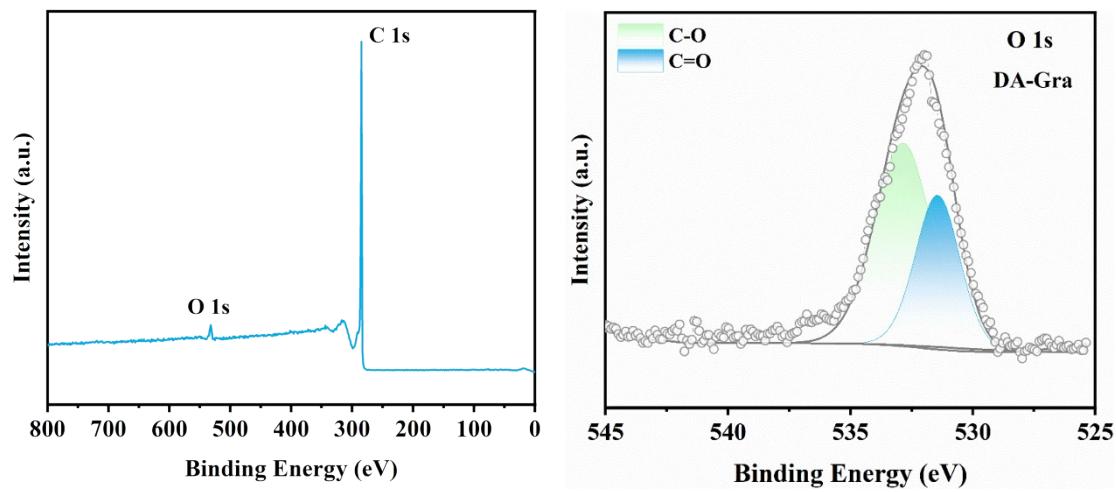
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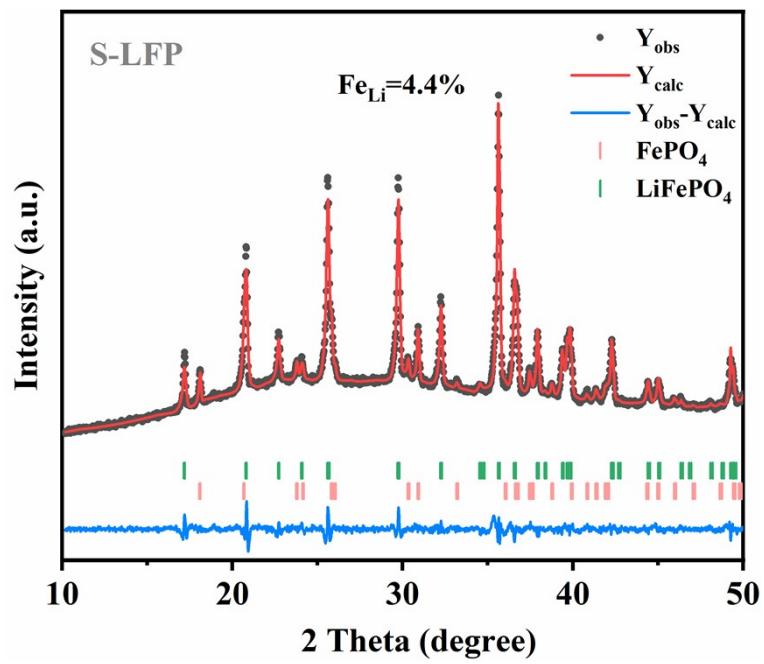
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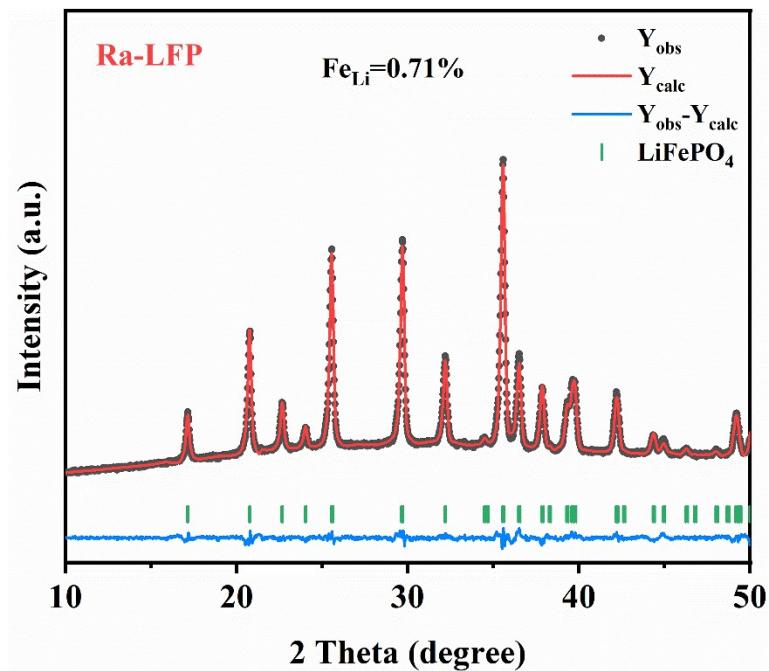
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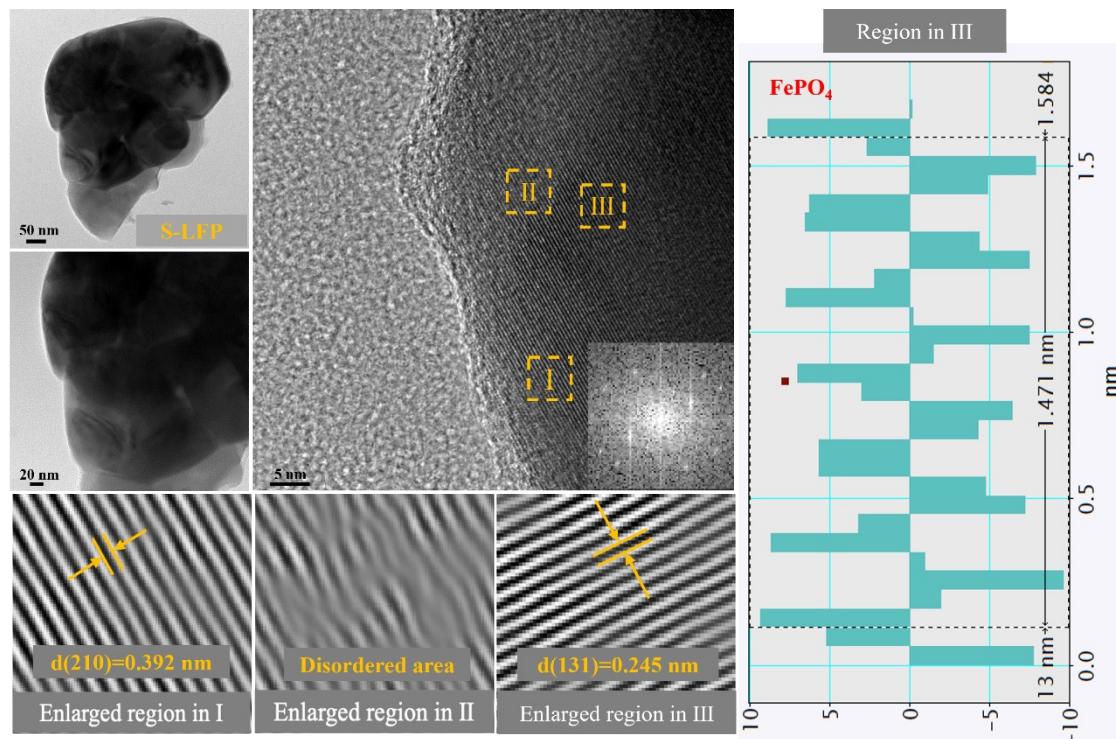
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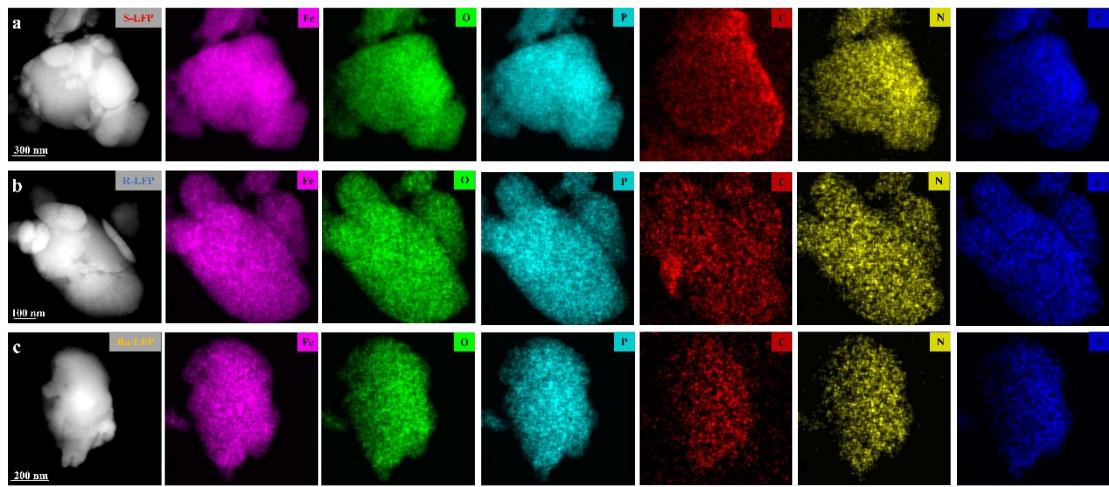
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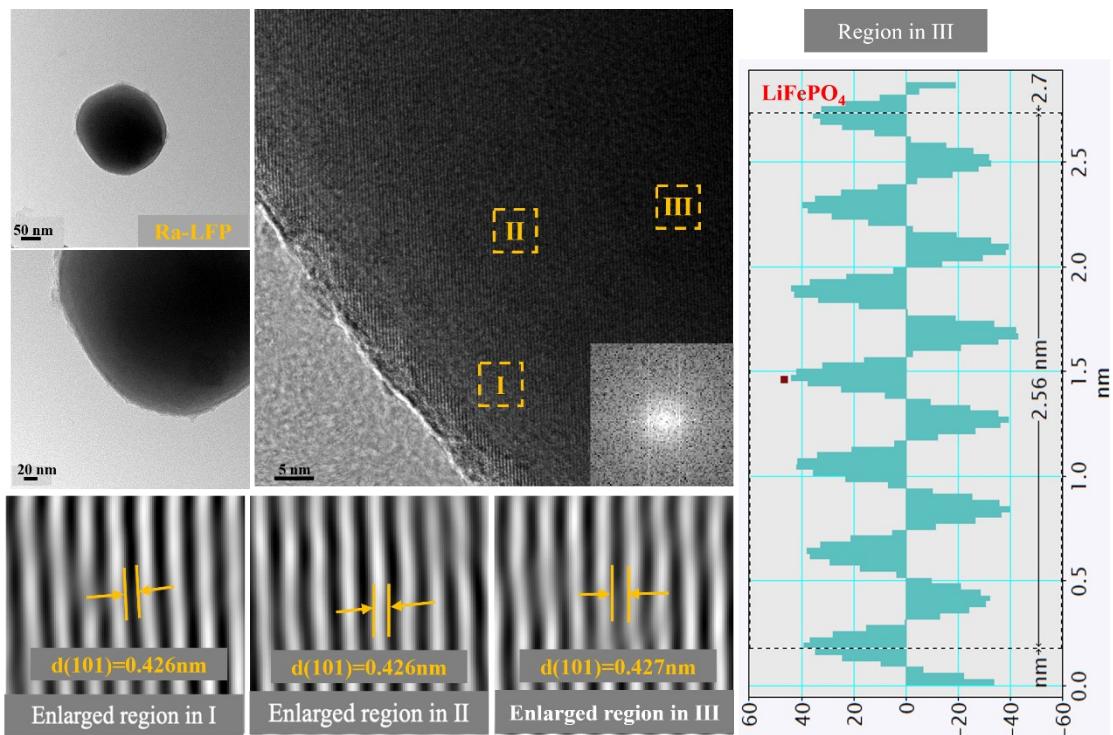
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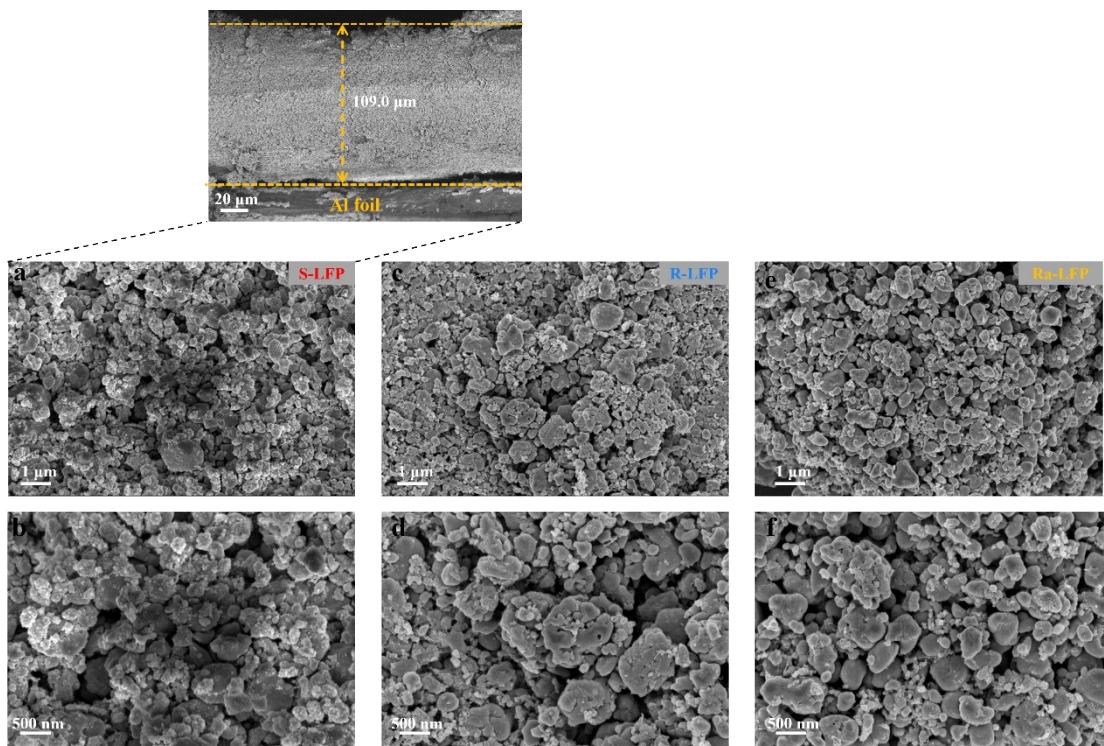
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Supplementary Fig. 17 EDS elemental maps of (a) S-LFP, (b) R-LFP and (c) Ra-LFP.

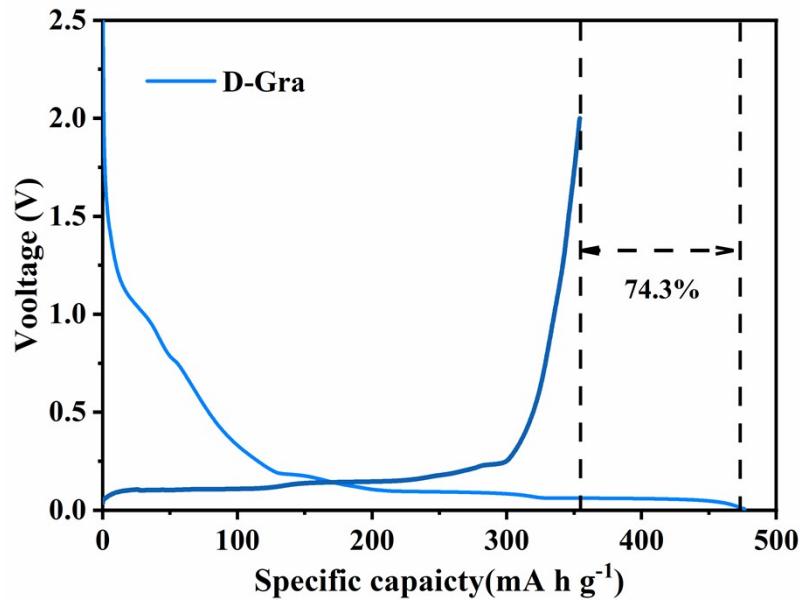


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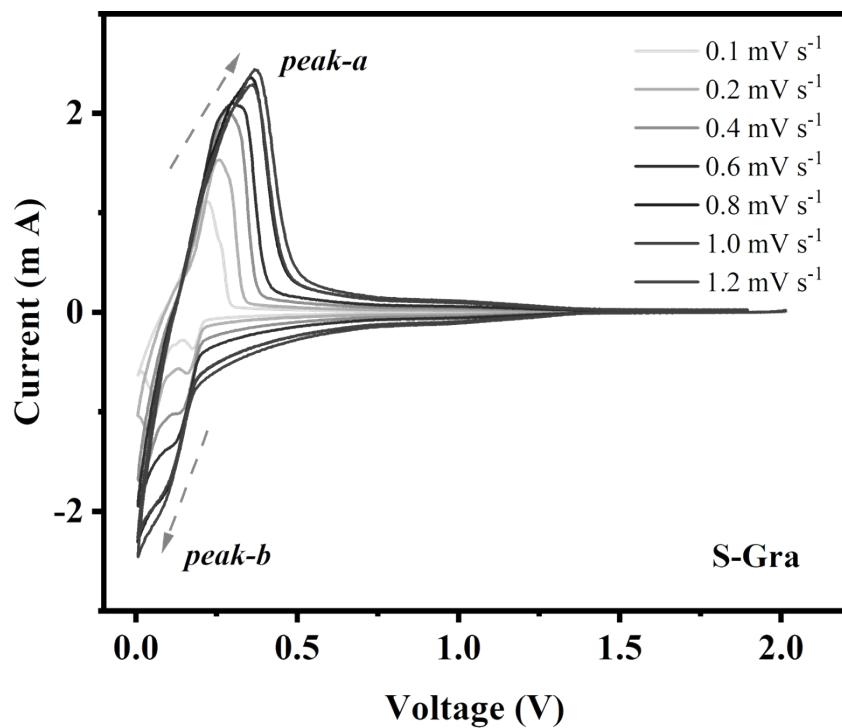


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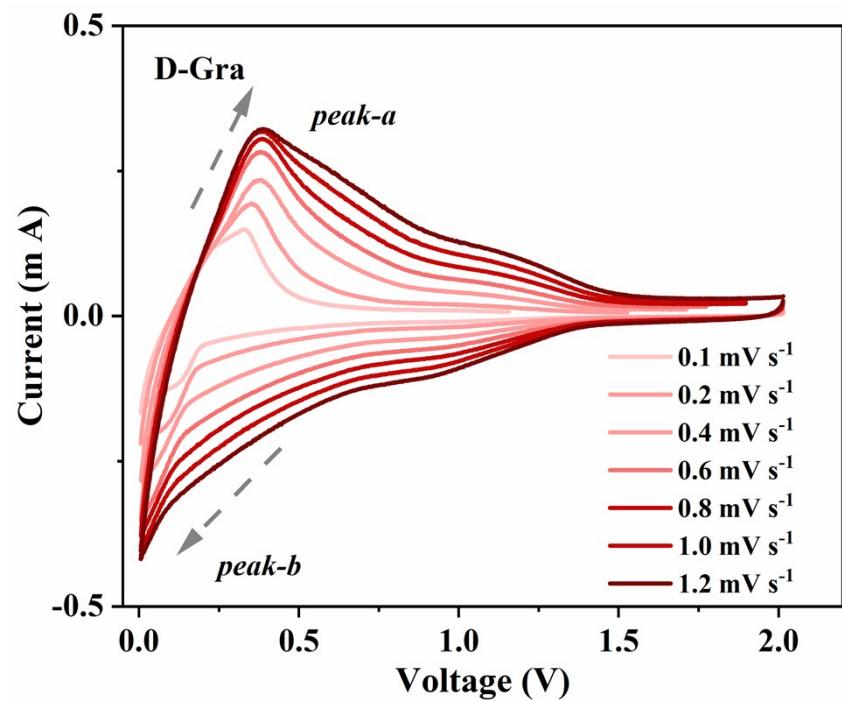
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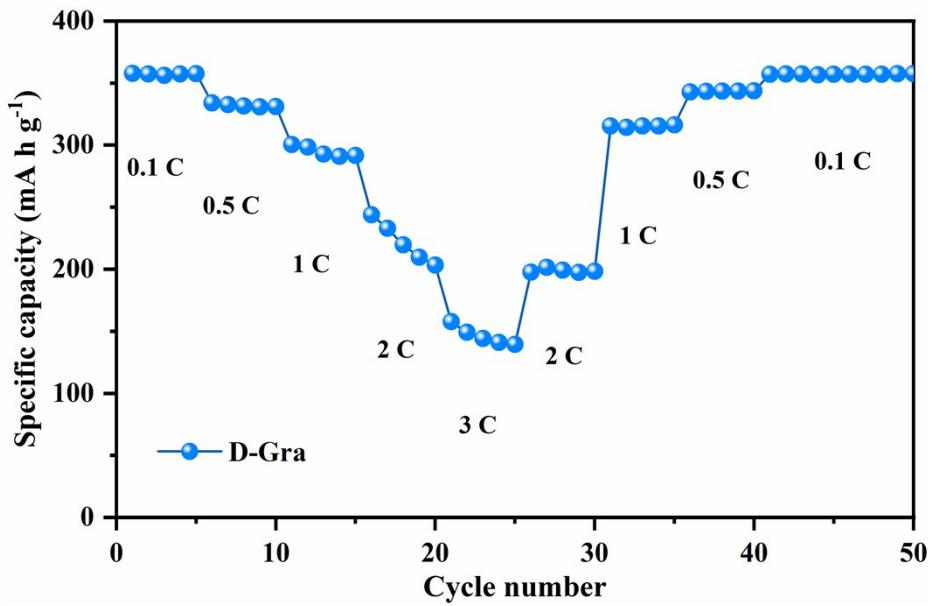
Supplementary Fig. 20 Initial charge and discharge curves of D-Gra at 0.1 C.



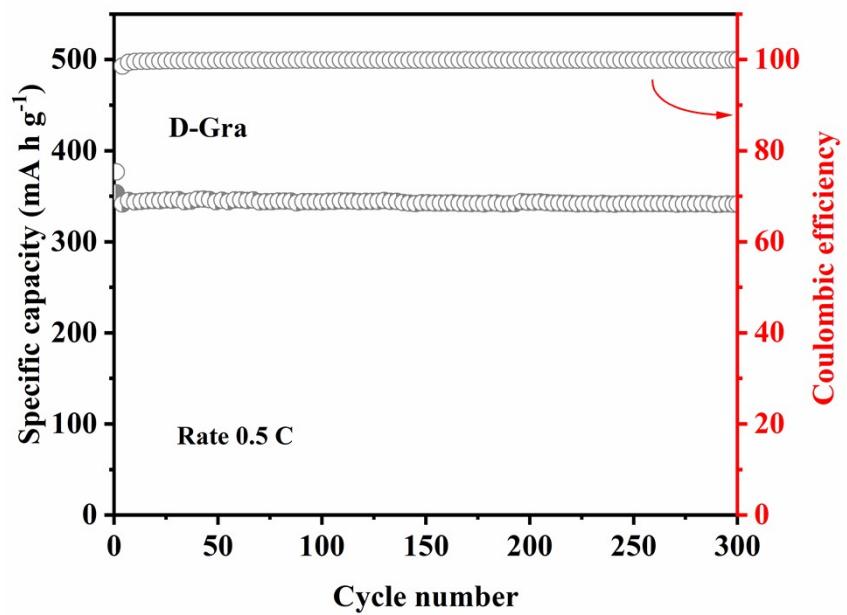
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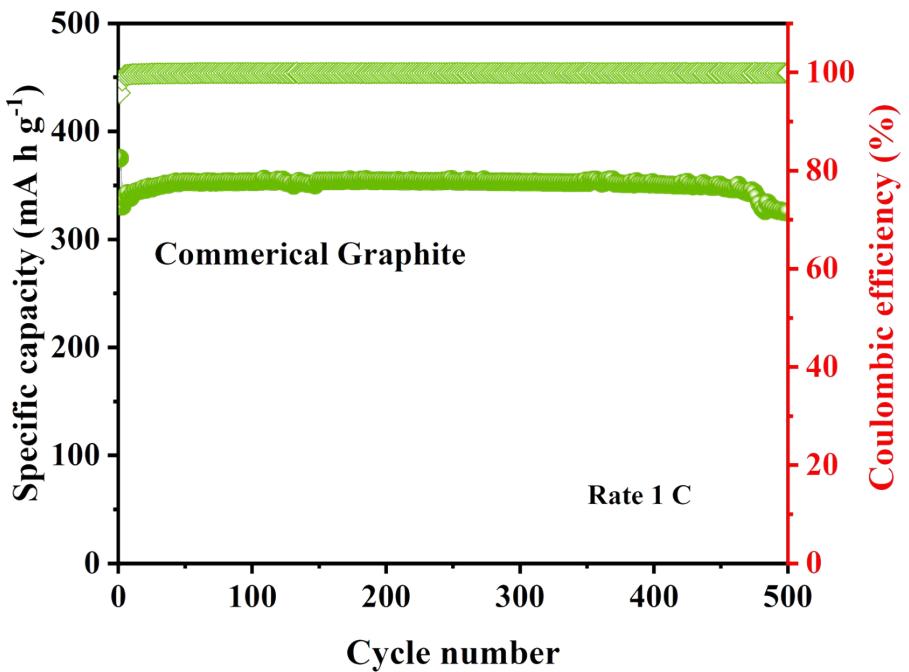
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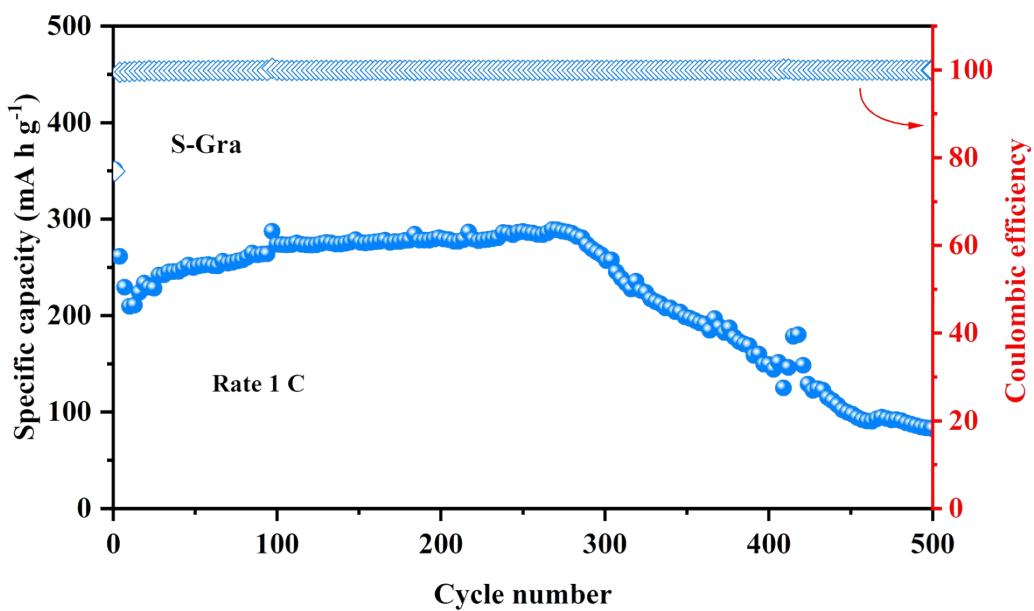
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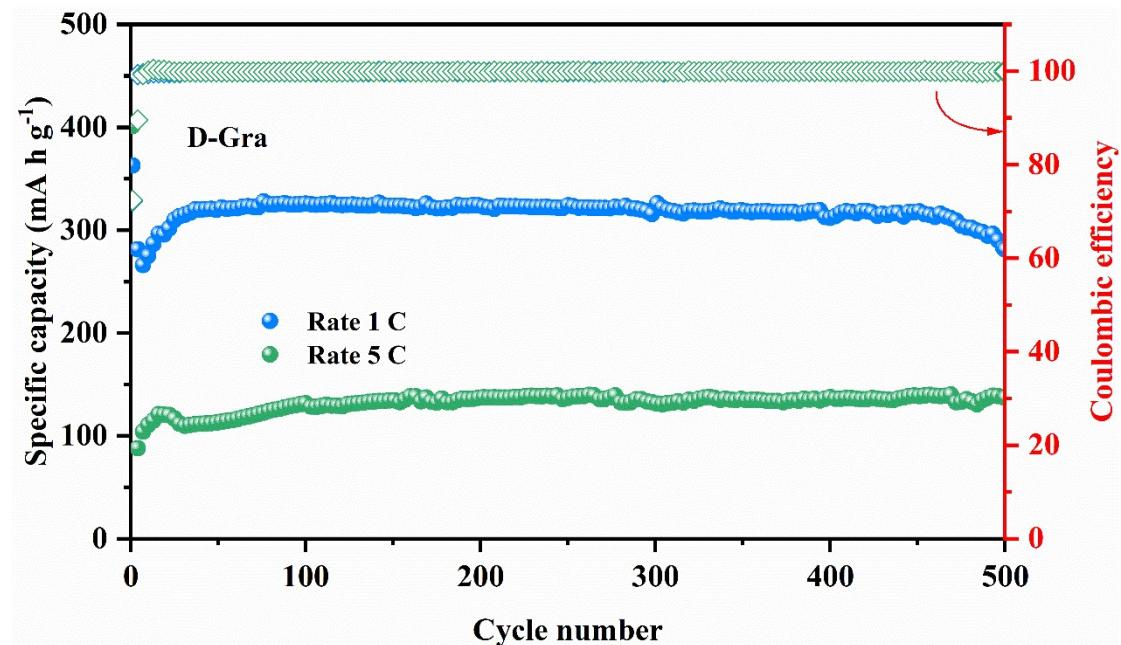
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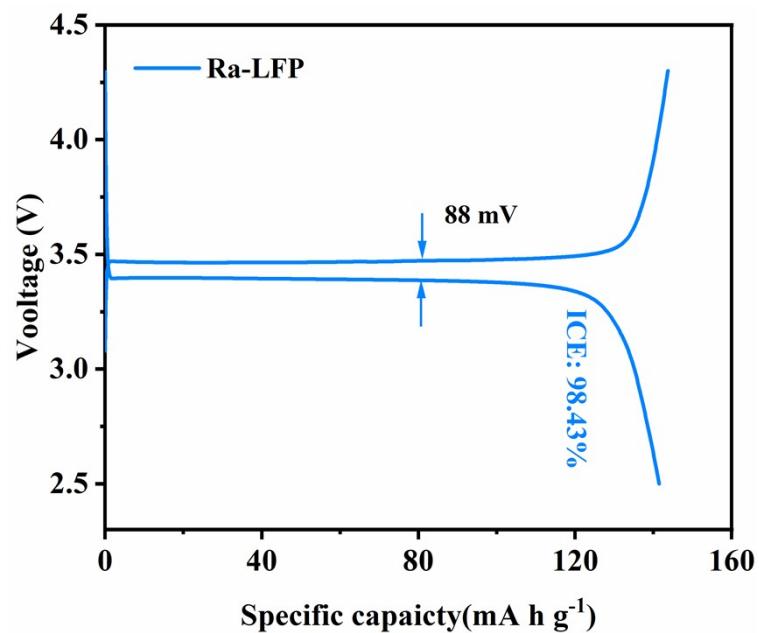
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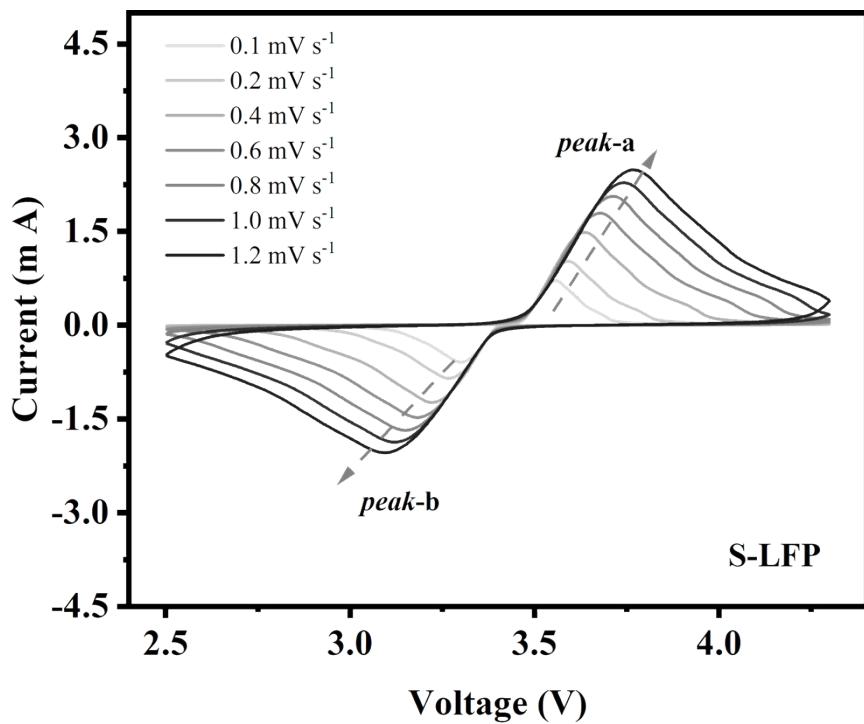
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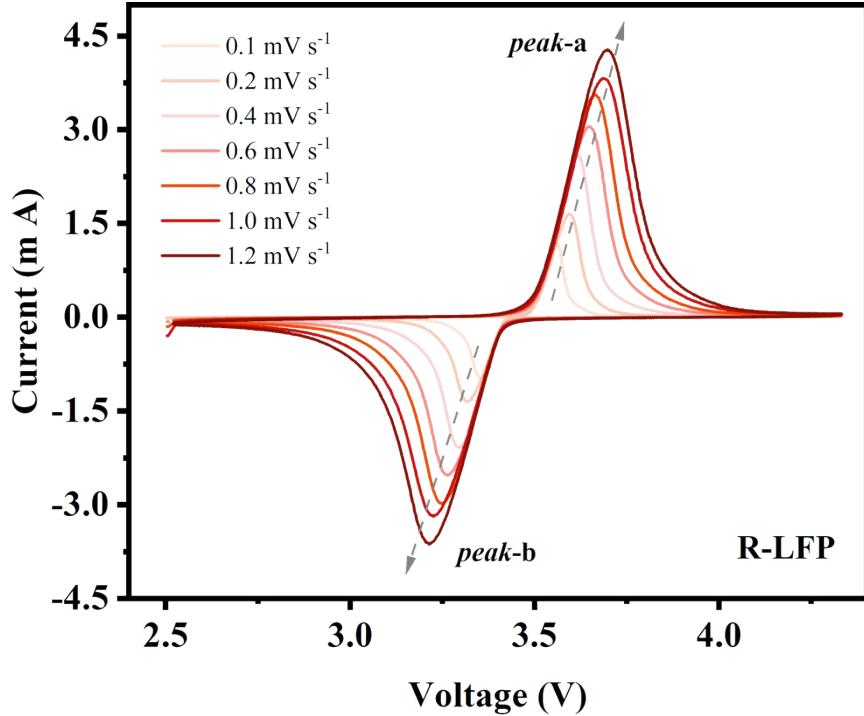
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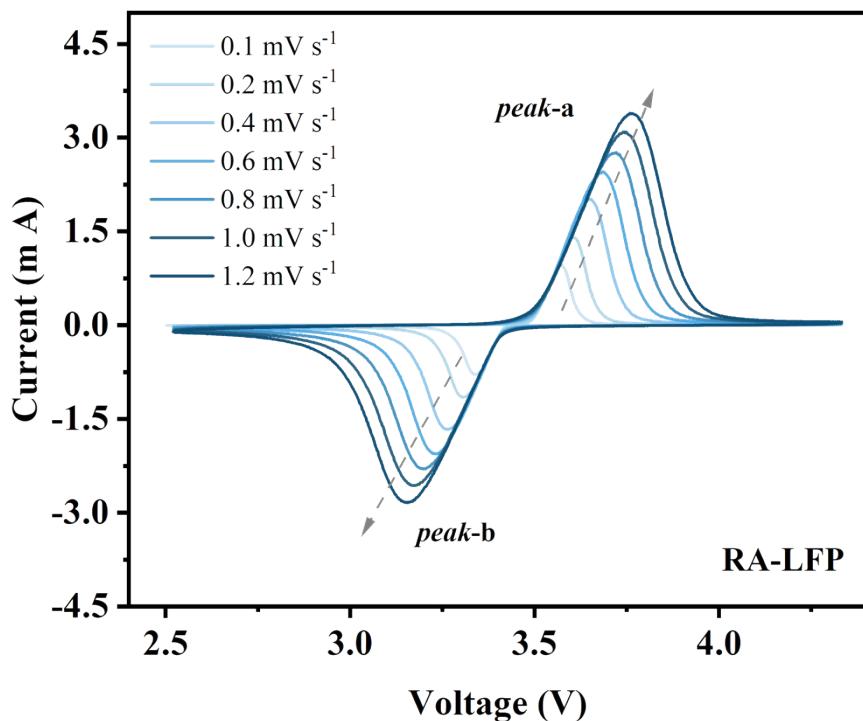
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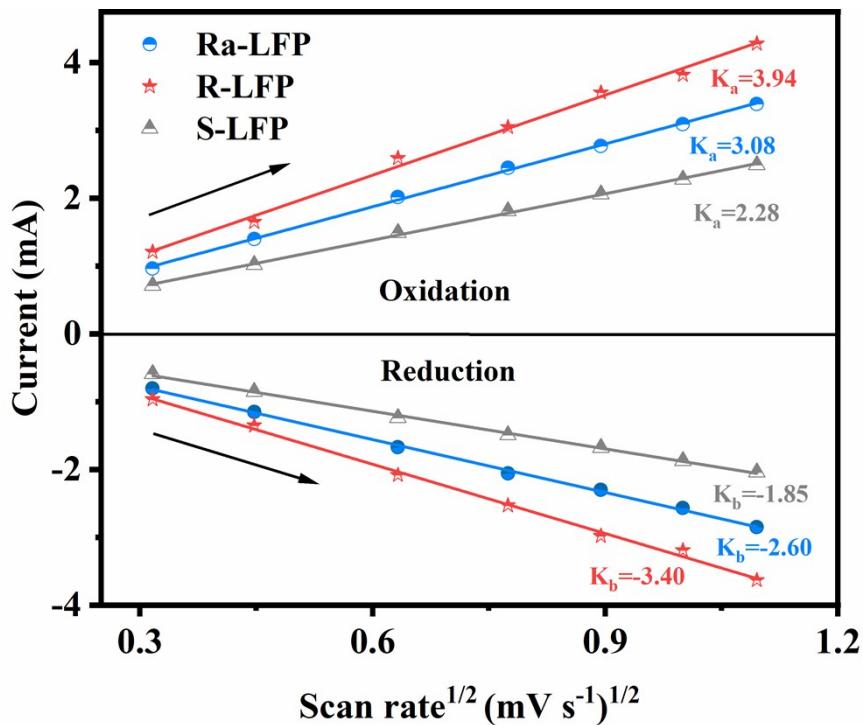
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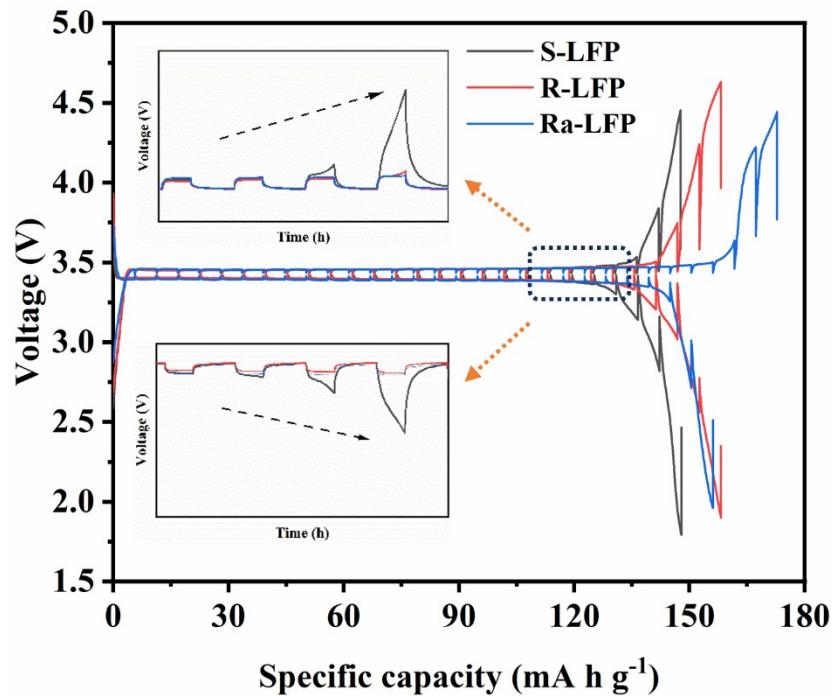
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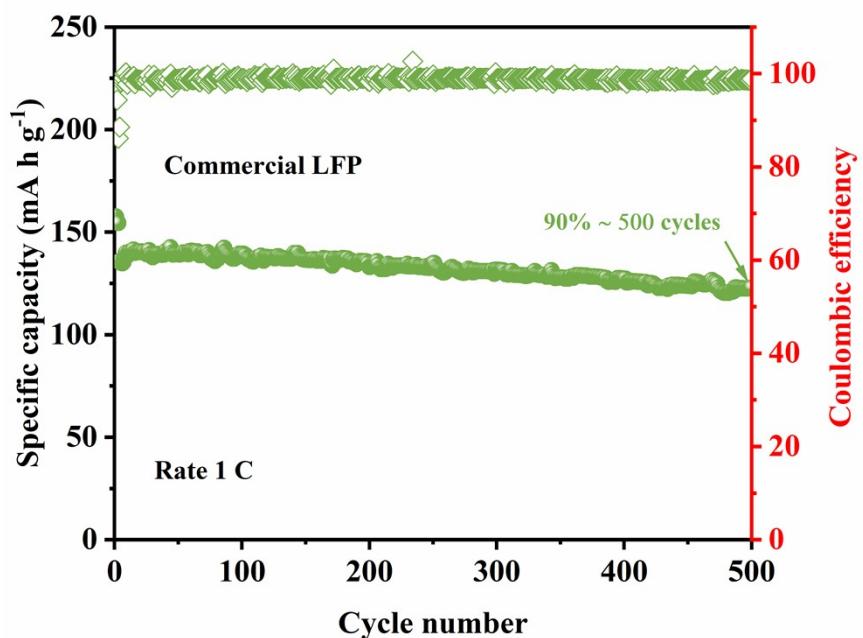
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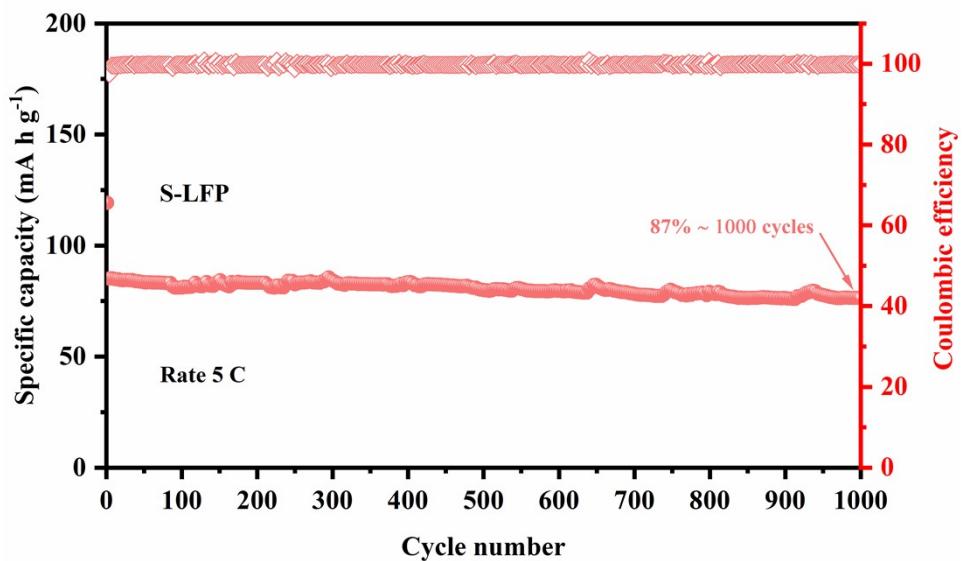
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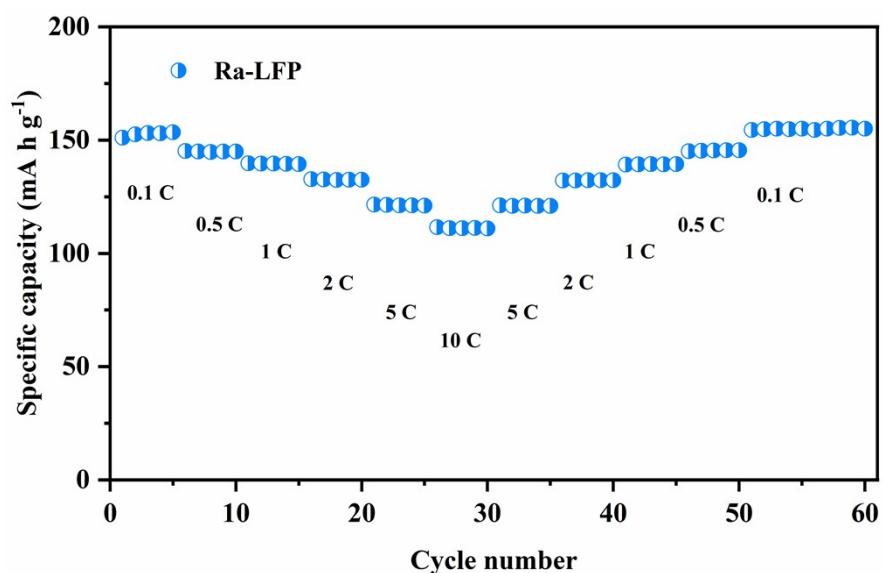
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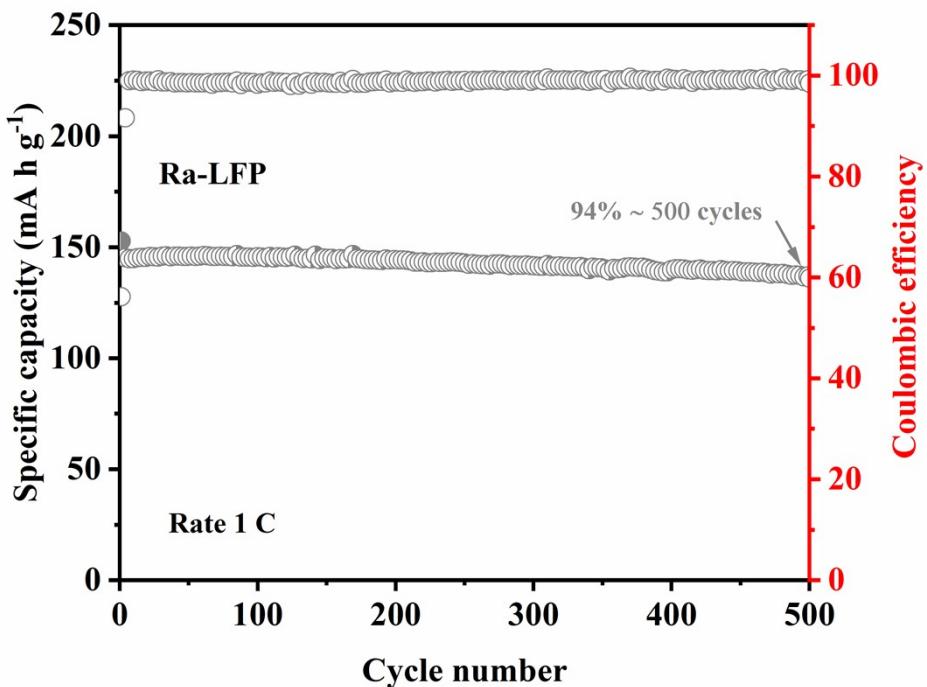
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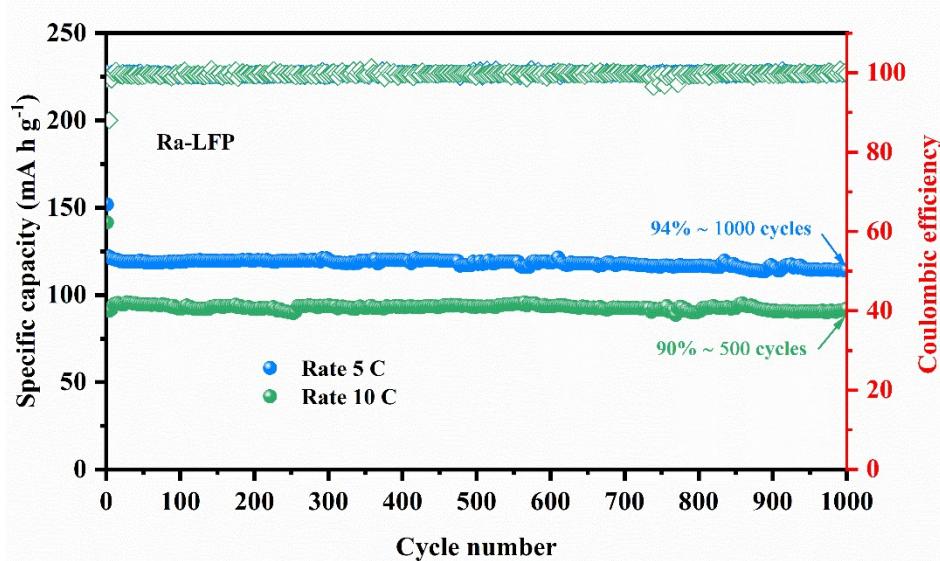
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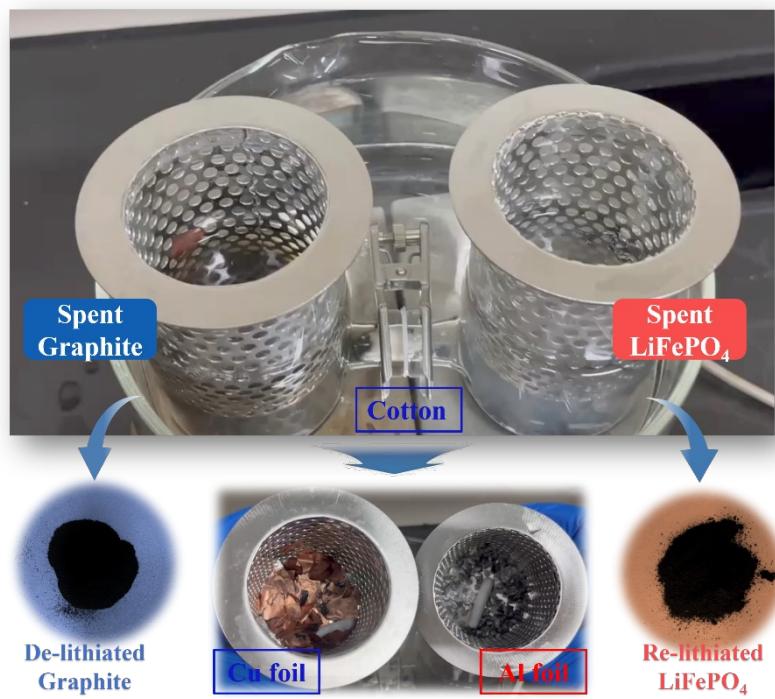
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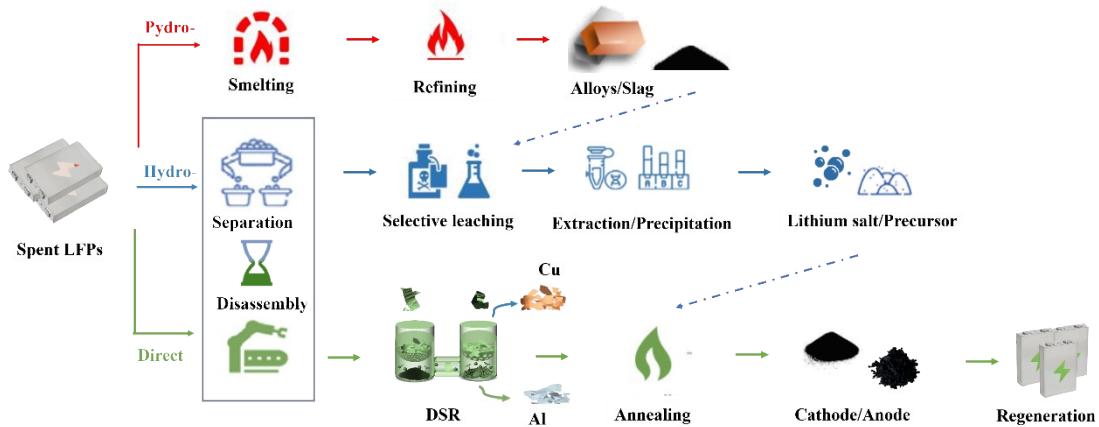
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Supplementary Fig. 38 Cycling performance of Ra-LFP at 5 and 10 C.



Supplementary Fig. 39 A digital photograph of spent cathode and anode after direct separation and regeneration in the scaled-up Hype cell configuration.



Supplementary Fig. 40 The general and typical processes in different LFPs recycling methods, including pyrometallurgy (Pyro-), hydrometallurgy (Hydro-) and direct (DSR).

Supplementary Table 1. ICP results for Cu contents in water solution at different temperatures and time after Cu foil-graphite layer separation operation.

Temperature (°C)	Cu (mg/L)	Time (min)	Cu (mg/L)
30	0.62	10	0.70
40	0.61	20	0.73
50	0.76	30	0.74
60	0.82	40	0.96
70	0.89	50	0.96
/	/	60	0.97

Supplementary Table 2. ICP results for Li/Fe molar ratio of R-LFP with different lithiation temperatures after diluting 1000 times.

Temperature (°C)	Li (mg/L)	Fe (mg/L)	P (mg/L)
30	0.85	8.17	4.42
40	0.92	8.24	4.43
50	0.96	8.2	4.47
60	1.04	8.69	4.66
70	1.09	4.77	8.91

Supplementary Table 3. ICP results for Li/Fe molar ratio of R-LFP with different pH values after diluting 1000 times.

pH	Li (mg/L)	Fe (mg/L)	P (mg/L)
8	1.02	8.79	4.72
8.5	1.05	8.92	4.79
9	1.03	8.69	4.65
9.5	1.09	8.91	4.77

Supplementary Table 4. ICP results for Li/Fe molar ratio of R-LFP with different lithiation time after diluting 1000 times.

time	Li (mg/L)	Fe (mg/L)	P (mg/L)
10	1.01	8.54	4.57
20	1.03	8.73	4.71
30	1.06	8.95	4.71
40	1.08	9.14	4.78
50	1.06	8.81	4.68
60	1.09	8.91	4.77

Supplementary Table 5. ICP results for Al contents in water solution at different temperatures and time after Al foil- LiFePO₄ layer separation operation.

Temperature (°C)	Al (mg/L)	Time (min)	Al (mg/L)
30	0.13	10	0.13
40	0.15	20	0.15
50	0.17	30	0.18
60	0.21	40	0.19
70	0.22	50	0.20
/	/	60	0.21

Supplementary Table 6. Structural parameters obtained from Rietveld refinement of the XRD pattern of S-LFP. **Phase 1 LiFePO₄:** Space group: *Pnma*, $a=10.327364$ Å, $b=6.005226$ Å, $c=4.690605$ Å, $\alpha=\beta=\gamma=90^\circ$. **Phase 2 FePO₄:** Space group: *Pnma*, $a=9.809450$ Å, $b=5.786042$ Å, $c=4.779844$ Å, $\alpha=\beta=\gamma=90^\circ$.

Atoms	Site	Wyckoff positions			Occupancy	Site	Wyckoff positions			Occupancy
Li	4a	0	0	0	0.95598	NA				
Fe	4a	0	0	0	0.04402	4a	0.00000 0.00000	0.00000		0.01200
Fe	4c	0.28136	0.25000	0.97040	0.95598	4c	0.27211 0.94679	0.25000		0.98800
Li	4c	0.28136	0.25000	0.97040	0.04402	NA				
P	4c	0.09424	0.25000	0.41413	1	4c	0.09546 0.37701	0.25000		1
O	4c	0.09725	0.25000	0.73635	1	4c	0.11840 0.71821	0.25000		1
O	4c	0.44984	0.25000	0.19902	1	4c	0.43774 0.16289	0.25000		1
O	8d	0.16375	0.04192	0.28121	1	8d	0.1676 0.24428	0.06044		1

Supplementary Table 7. Structural parameters obtained from Rietveld refinement of the XRD pattern of R-LFP. **Phase LiFePO₄:** Space group: *Pnma*, *a*=10.322839 Å, *b*=6.001675 Å, *c*=4.687203 Å, $\alpha=\beta=\gamma=90^\circ$.

Atoms	Site	Wyckoff positions			Occupancy
Li	4a	0	0	0	0.99154
Fe	4a	0	0	0	0.00846
Fe	4c	0.28163	0.25000	0.97251	0.99154
Li	4c	0.28163	0.25000	0.97251	0.00846
P	4c	0.09558	0.25000	0.42261	1
O	4c	0.09015	0.25000	0.74942	1
O	4c	0.45244	0.25000	0.21473	1
O	8d	0.15987	0.04707	0.28457	1

Supplementary Table 8. Structural parameters obtained from Rietveld refinement of the XRD pattern of Ra-LFP. **Phase LiFePO₄:** Space group: *Pnma*, $a=10.346512\text{ \AA}$, $b=6.016788\text{ \AA}$, $c=4.700098\text{ \AA}$, $\alpha=\beta=\gamma=90^\circ$.

Atoms	Site	Wyckoff positions			Occupancy
Li	4a	0	0	0	0.99286
Fe	4a	0	0	0	0.00714
Fe	4c	0.28267	0.25000	0.97510	0.99286
Li	4c	0.28267	0.25000	0.97510	0.00714
P	4c	0.09278	0.25000	0.42706	1
O	4c	0.08867	0.25000	0.75660	1
		0.44914	0.25000	0.22520	
O	4c				1
O	8d	0.16091	0.05684	0.28491	1

Supplementary Table 9. The discharge capacity resulting from the direct recycling of published articles varies at different rates.

No.	0.1 C	0.2 C	0.5 C	1 C	2 C	5C	10 C	Ref.
S1	151.3	148.7	146.6	141.3	132.2	113.1	92.4	1
S2	/	/	/	140.1	131.6	112.8	93.0	2
S3	/	/	143	140	134	117	/	3
S4	/	150.9	140	125.7	110.1	89.2	71.3	4
S5	/	/	138	/	/	/	/	5
S6	151.1	/	142.8	134.8	123	104.2	87.9	6
S7	/	/	/	139.2	131.2	116	105.5	7
S8	/	/	/	147.9	136.1	113.6	87.2	8
S9	157	/	/	/	127	111	97	9
S10	/	/		139.1	/	/	100.1	10
S11	148.7	/	143.6	135.4	123.3	102.5	/	11
S12	156.6	/	151.8	145.9	/	122.1	/	12

Supplementary Table 10 Basic data of techno-economic analysis by Hydro- and DSR recycling.

No.	Item	Market price	Unit	Update Data	Data Sources
1	Spent LFP (pouch cell)	8000	¥ t ⁻¹	Dce. 27 th 2024	<u>SMM</u>
2	Li ₂ CO ₃	75400	¥ t ⁻¹	Dce. 27 th 2024	<u>SMM</u>
3	LiOH	69798	¥ t ⁻¹	Dce. 27 th 2024	<u>SMM</u>
4	Na ₂ CO ₃	3000	¥ t ⁻¹	Dce. 27 th 2024	<u>100PPI</u>
5	H ₂ SO ₄ (98%)	380	¥ t ⁻¹	Dce. 27 th 2024	<u>100PPI</u>
6	H ₂ O ₂ (27.5%)	1800	¥ t ⁻¹	Dce. 27 th 2024	<u>100PPI</u>
7	NaSO ₃	2800	¥ t ⁻¹	Dce. 27 th 2024	<u>100PPI</u>
8	Ar	2400	¥ t ⁻¹	Dce. 27 th 2024	<u>100PPI</u>
9	Electricity	0.61	¥ kW·h ⁻¹	Dce. 27 th 2024	<u>ZZJSJ</u>
10	Water	2.96	¥ t ⁻¹	Dce. 27 th 2024	<u>ZZJSJ</u>
11	Average labor cost	73000	¥ a ⁻¹		<u>DOI:</u>
12	Sewage Treatment	40	¥ t ⁻¹	Dce. 27 th 2024	<u>10.1038/s41467-023-36197-6</u>
13	LiFePO ₄ cathode material	37510	¥ t ⁻¹	Dce. 27 th 2024	<u>SMM</u>
a) 1\$ = 7.2985 ¥ (Update time: 2024/12/27); b) SMM (https://www.smm.cn/), 100PPI (https://www.100ppi.com/ppi/), ZZJSJ (https://zzjsj.zhengzhou.gov.cn/).					

Supplementary Table 11 Cost analysis of hydro- recycling strategy.

Cost analysis based on hydrometallurgical recovery												
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)					
Raw material	1	Spent LFP battery	8000	1.00	250 kg of spent LFP can be sorted from per 1 ton of spent battery	DOI: 10.1002/adma.202414048	8000					
					Subtotal (¥)			8000				
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)					
Reagent	2	H ₂ SO ₄	380	0.0885	Leaching agent	DOI:10.1021/acssuschemeng.7b01594	33.63					
	3	H ₂ O ₂	1800	0.1116	Oxidating agent		200.88					
	4	Na ₂ CO ₃	3000	0.0541	Precipitating agent		162.3					
					Subtotal (¥)			396.81				
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)					
Average labor	5	Disassembly	73000	0.01		DOI. 10.1016/j.xcrp.2022.100741	730					
	6	Recycling step	73000	0.02			1460					
					Subtotal (¥)			2190				
Main	No.	Item	Unit Price (¥ t ⁻¹ / ¥ k Wh)	Dose (t/k Wh)	Note	Data Sources	Cost (¥)					
Electricity & water	7	Electricity	0.61	500		DOI:10.1038/s41467-023-36197-6	305					
	8	Water	2.96	500			1480					
								1785				
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)					
Equipment depreciation	9	Equipment depreciation	5000	0.25		DOI: 10.3969/j.issn.1009-847X.2018.10.006	1250					
					Subtotal (¥)			1250				
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)					
Sewage treatment	10	Sewage treatment	40	20		DOI: 10.3969/j.issn.1009-847X.2018.10.006	800					
					Subtotal (¥)			800				
Total Cost (¥)							14421.81					
In table S6, the cost per one ton of spent LFP battery was calculated using the hydro-route. The recycling process can be divided into three steps: disassembling the spent battery, separating cathode active materials and subsequently conducting hydro-leaching to extract lithium. It is assumed that the level of deficiency in the spent LFP battery remains consistent (Li _{0.8} FePO ₄) based on hydro- recovery, DSR and hydro- synthesized methods.												

Supplementary Table 12 Cost analysis of hydro- and re-synthesized recovery strategy.

Cost analysis based on hydrometallurgical and re-synthesized recovery strategy													
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Raw material	1	Spent LFP battery	8000	1.00	250 kg of spent LFP can be sorted from per 1 ton of spent battery	DOI: 10.1002/adma.202414048	8000						
	Subtotal (¥)						8000						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Reagent	2	H ₂ SO ₄	380	0.0885	Leaching agent	DOI: 10.1016/j.cej.2023.147201	33.63						
	3	H ₂ O ₂	1800	0.1116	Oxidating agent		200.88						
	4	Na ₂ CO ₃	3000	0.0541	Precipitating agent		162.3						
	5	Li ₂ CO ₃	75400	0.0146	Used as the supplement of lithium during the annealing process	DOI: 10.1039/DITA07757K	1100.84						
	6	Ar	2400	0.03	Used as the protective atmosphere during regeneration process		72						
	Subtotal (¥)						1569.65						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Average labor	7	Disassembly	73000	0.01		DOI: 10.1016/j.xcrp.2022.100741	730						
	8	Recycling & synthesized	73000	0.03			2190						
	Subtotal (¥)						2920						
Main	No.	Item	Unit Price (¥ t ⁻¹ / ¥ k Wh)	Dose (t/k Wh)	Note	Data Sources	Cost (¥)						
Electricity & water	9	Electricity	0.61	800		DOI:10.1038/s41467-023-36197-6	488						
	10	Water	2.96	500			1480						
Subtotal (¥)							1968						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Equipment depreciation	11	Equipment depreciation	5000	0.25		DOI: 10.3969/j.issn.1009-847X.2018.10.006	1250						
	Subtotal (¥)						1250						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Sewage treatment	12	Sewage treatment	40	20		DOI: 10.3969/j.issn.1009-847X.2018.10.006	800						
	Subtotal (¥)						800						
Total Cost (¥)							16507.65						
In table S6, the cost per one ton of spent LFP battery was calculated using the hydro-route. The recycling process can be divided into three steps: disassembling the spent battery, separating cathode active materials and subsequently conducting hydro-leaching to extract lithium. It is assumed that the level of deficiency in the spent LFP battery remains consistent (Li _{0.8} FePO ₄) based on hydro- recovery, DSR and hydro- synthesized methods.													

Supplementary Table 13 Cost analysis of DSR recycling strategy.

Cost analysis based on direct separation and regeneration strategy													
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Raw material	1	Spent LFP battery	8000	1.00	310 kg of spent LFP electrode can be sorted from per 1 ton of spent battery	DOI: 10.1002/adma.202414048	8000						
	Subtotal (¥)						8000						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Reagent	2	NaSO ₃	2800	0.0399	Used as reductant in the re-lithiation process	DOI: 10.1016/j.cej.2023.147201	351.92						
	3	LiOH	69798	0.0061	Used as lithium salt to re-lithiate the spent LFP		425.77						
	4	Li ₂ CO ₃	75400	0.0017	Used as the supplement of volatile lithium during the short annealing process	DOI: 10.1016/j.joule.2020.10.008	128.18						
	5	Ar	2400	0.015	Used as the protective atmosphere during regeneration process		36						
	Subtotal (¥)						941.87						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Average labor	6	Disassembly & Recovery	73000	0.02		DOI: 10.1016/j.xcrp.2022.100741	1460						
	Subtotal (¥)						1460						
Main	No.	Item	Unit Price (¥ t ⁻¹ / k Wh)	Dose (t/k Wh)	Note	Data Sources	Cost (¥)						
Electricity & water	7	Electricity	0.61	200		DOI:10.1038/s41467-023-36197-6	122						
	8	Water	2.96	300			888						
	Subtotal (¥)						1010						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Equipment depreciation	9	Equipment depreciation	5000	0.25		DOI: 10.3969/j.issn.1009-847X.2018.10.006	1250						
	Subtotal (¥)						1250						
Main	No.	Item	Unit Price (¥ t ⁻¹)	Dose (t)	Note	Data Sources	Cost (¥)						
Sewage treatment	10	Sewage treatment	40	10		DOI: 10.3969/j.issn.1009-847X.2018.10.006	400						
	Subtotal (¥)						400						
Total Cost (¥)							13061.87						
In table S7, the cost per one ton of spent LFP battery was calculated using the DSR route. The recycling process can be divided into three steps: disassembling the spent battery, separating electrode and subsequently conducting direct re-lithiation to replenish lithium. It is assumed that the level of deficiency in the spent LFP battery remains consistent ($\text{Li}_{0.8}\text{FePO}_4$) based on hydro- recovery and DSR methods.													

Supplementary Table 14 Hydro- recovery revenue analysis.

Revenue analysis based on hydro- recovery								
No.	Item	Material	Recovery yield (%)	Production (t)	Market price (¥ t ⁻¹)	Revenue (¥)	Update Date	Data Sources
1	Lithium salt	Li ₂ CO ₃	95	0.0446	75400	3362.84	Dce. 27 th 2024	SMM
2	Precursor	FePO ₄	95	0.227	10600	2406.2	Dce. 27 th 2024	SMM
3	Anode	Graphite	96	0.125	20400	2550	Dce. 27 th 2024	SMM
4	Current collector	Al foil	90	0.0540	35200	1900.8	Dce. 27 th 2024	SMM
5	Current collector	Cu foil	90	0.090	92420	8317.8	Dce. 27 th 2024	Mysteel
Subtotal (¥) 18537.64								
Note: a) It is assumed that the level of deficiency in the spent LFP battery remains consistent (Li _{0.8} FePO ₄) based on hydro- recovery and DSR methods. b) The proportion of each component is 25% of cathode material, 13% of graphite, 6% of Al foil, 10% of Cu foil, 3% of separator, 16% of electrolyte, and 20% of shell. c) The value of separator, electrolyte, and shell is hard to assess in the real process, therefore, these components are excluded from the revenue analysis. d) Mysteel (https://www.mysteel.com/).								

Supplementary Table 15 Hydro- and re-synthesized recovery revenue analysis.

Revenue analysis based on hydro- recovery								
No.	Item	Material	Recovery yield (%)	Production (t)	Market price (¥ t ⁻¹)	Revenue (¥)	Update Date	Data Sources
1	Cathode material	LiFePO ₄	95	0.2375	37510	8908.625	Dce. 27 th 2024	SMM
3	Anode	Graphite	96	0.125	20400	2550	Dce. 27 th 2024	SMM
4	Current collector	Al foil	90	0.0540	35200	1900.8	Dce. 27 th 2024	SMM
5	Current collector	Cu foil	90	0.090	92420	8317.8	Dce. 27 th 2024	Mysteel
Subtotal (¥) 21677.23								
Note: a) It is assumed that the level of deficiency in the spent LFP battery remains consistent (Li _{0.8} FePO ₄) based on hydro- recovery and DSR methods. b) The proportion of each component is 25% of cathode material, 13% of graphite, 6% of Al foil, 10% of Cu foil, 3% of separator, 16% of electrolyte, and 20% of shell. c) The value of separator, electrolyte, and shell is hard to assess in the real process, therefore, these components are excluded from the revenue analysis. d) Mysteel (https://www.mysteel.com/).								

Supplementary Table 16 DSR revenue analysis.

Revenue analysis based direct separation and regeneration strategy								
No.	Item	Material	Recovery yield (%)	Production (t)	Market price (¥ t ⁻¹)	Revenue (¥)	Update Date	Data Sources
1	Cathode material	LiFePO ₄	95	0.2375	37510	8908.625	Dce. 27 th 2024	<u>SMM</u>
2	Anode	Graphite	98	0.1274	20400	2598.96	Dce. 27 th 2024	<u>SMM</u>
3	Current collector	Al foil	90	0.0540	35200	1900.8	Dce. 27 th 2024	<u>SMM</u>
4	Current collector	Cu foil	90	0.090	92420	8317.8	Dce. 27 th 2024	<u>Mysteel</u>
Subtotal (¥) 21726.19								
Note: a) It is assumed that the level of deficiency in the spent LFP battery remains consistent (Li _{0.8} FePO ₄) based on hydro- recovery and DSR methods. b) The proportion of each component is 25% of cathode material, 13% of graphite, 6% of Al foil, 10% of Cu foil, 3% of separator, 16% of electrolyte, and 20% of shell. c) The value of separator, electrolyte, and shell is hard to assess in the real process, therefore, these components are excluded from the revenue analysis. d) Mysteel (https://www.mysteel.com/).								

Supplementary Table 17 Profit calculation.

Recovery profit results				
No.	Recycling route	Cost (¥)	Revenue (¥)	Profit (¥)
1	Hydro-	14421.81	18537.64	4115.83
2	DSR	12331.87	21726.19	9583.92
3	Hydro- synthesis	16507.65	21677.23	5169.575

Note: 1\$ = 7.2985 ¥ (Update time: 2024/12/27).

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