## Supporting Information

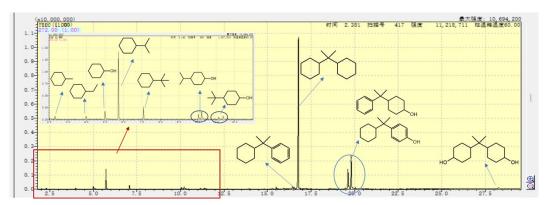
Chemical recycling of polycarbonate waste into advanced aviation fuel candidates via nickel-oxygen vacancy dual sites

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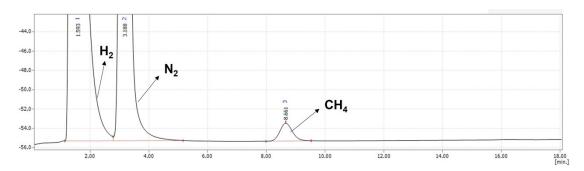
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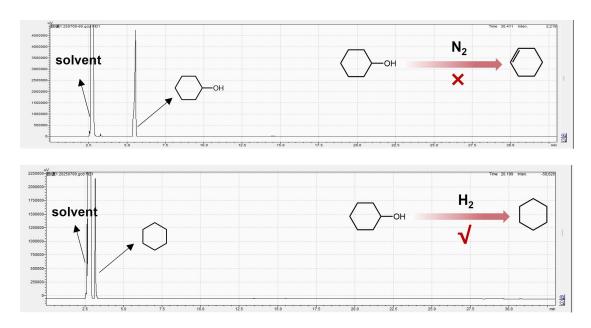
**Figure S1.** Photographs of polycarbonate (PC) powder and the obtained C15 dicycloalkane product.



**Figure S2.** GC-MS spectra of products detected in PC conversion. Reaction conditions: 0.3 g PC, 0.06 g 15Ni/CeO<sub>2</sub>, 30 mL cyclopentane, 4 MPa H<sub>2</sub>, 523 K.



**Figure S3.** GC chromatogram of the gas phase products from the HDO of PC over the 15Ni/CeO<sub>2</sub>. Reaction conditions: 0.3 g PC, 0.06 g 15Ni/CeO<sub>2</sub>, 30 mL cyclopentane, 4 MPa H<sub>2</sub>, 523 K.



**Figure S4.** GC chromatogram of products using cyclohexanol as feedstock. Reaction conditions: 0.3 g cyclohexanol, 0.06 g 15Ni/CeO<sub>2</sub>, 30 mL cyclopentane,  $N_2/H_2$  atmosphere, 453 K.

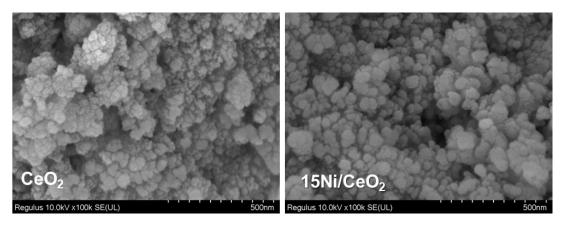


Figure S5. SEM images of CeO<sub>2</sub> and 15Ni/CeO<sub>2</sub>.

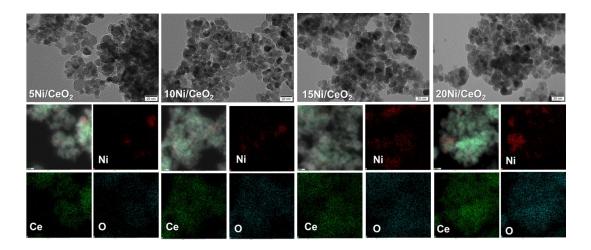
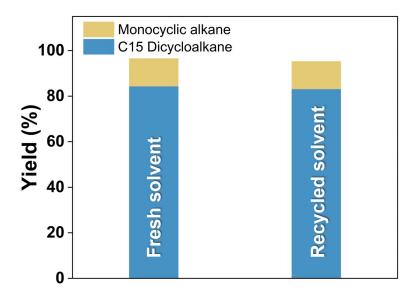


Figure S6. TEM images and EDS spectra of Ni/CeO $_2$  catalysts.



**Figure S7.** The HDO results of PC powder using the recycled solvent. Reaction conditions: 0.3 g PC, 0.06 g 15Ni/CeO<sub>2</sub>, 30 mL cyclopentane/recycled cyclopentane, 4 MPa H<sub>2</sub>, 523 K.

Table S1. Yield of saturated alkanes under different catalytic systems.

	Catalysts	Yield of saturated hydrocarbons (%)			
Entry		Monocyclic saturated alkanes	Bicyclic saturated alkanes	Total	Ref.
1	Rh/C + H-USY	/	86.9	86.9	1
2	Ru-Ni/Hβ	8.2	89.4	97.6	2
3	Ru-ReO <sub>x</sub> /SiO <sub>2</sub> +HZSM-5	5.7	93.4	99.1	3
4	Ni/HZSM-5	18.1	81.2	99.3	4
5	$Pd/C + La(OTf)_3$	4	95	99	5
6	This work	12.1	84.4	96.5	

Table S2. Properties and chemical composition of pristine PC powder.

Ayyana aa maalaaylan yyai aht 8	Chemical composition (%) b			
Average molecular weight <sup>a</sup> -	С	Н	О	
35976	74.58	5.35	17.75	

<sup>&</sup>lt;sup>a</sup> Measured by U.S. Agilent PL-GPC50 & Agilent PL-GPC220 gel permeation chromatography (GPC).

<sup>&</sup>lt;sup>b</sup> Measured by an elemental analyzer (EA, Elementar Unicube, Germany).

**Table S3.** Characteristics of Ni/CeO<sub>2</sub> catalysts.

	Ni	BET Surface area (m <sup>2</sup> /g) <sup>b</sup>			Pore size	$\mathrm{D}_{\mathrm{Ni}}$
Sample	content (wt%) a	Micropore area	External area	Total surface area	(nm) °	(%) d
5Ni/CeO <sub>2</sub>	4.30	2.56	41.87	44.44	11.1261	0.58
10Ni/CeO <sub>2</sub>	10.36	2.24	44.23	46.46	10.3561	0.52
15Ni/CeO <sub>2</sub>	15.76	1.88	42.77	44.65	10.3778	0.38
20Ni/CeO <sub>2</sub>	21.04	1.84	44.78	46.62	8.8266	0.28

<sup>&</sup>lt;sup>a</sup> Ni content of catalyst was determined by ICP-OES.

<sup>&</sup>lt;sup>b</sup> Micropore surface area and external surface area were calculated by t-plot method, specific surface area was calculated by BET equation.

<sup>&</sup>lt;sup>c</sup> Pore size was determined using the BJH.

<sup>&</sup>lt;sup>d</sup> The Ni dispersions was determined by H<sub>2</sub> plus chemisorption.

## References

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