

Supplementary Information

Trace-LiCl-assisted Synthesis of High-Loading Ordered Pt₃Co Intermetallic Catalysts for Oxygen Reduction Reaction in Fuel Cells

Yutao Ni^{a,b}, Jiahui Song^{a,b}, Huiying Lan^a, Xue Jing,^{a,b} Wenwen Shi^{a,*}, Ruimin Ding^{a,*},
Xi Yin^{a,*}

^a State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese Academy of Sciences, Taiyuan, Shanxi 030001, China

^b School of Chemical Engineering, University of Chinese Academy of Sciences, Beijing 100049, China

*Corresponding author. E-mail addresses: shiwenwen@sxicc.ac.cn (W. Shi), dingrm@sxicc.ac.cn (R. Ding), xiyin@sxicc.ac.cn (X. Yin).

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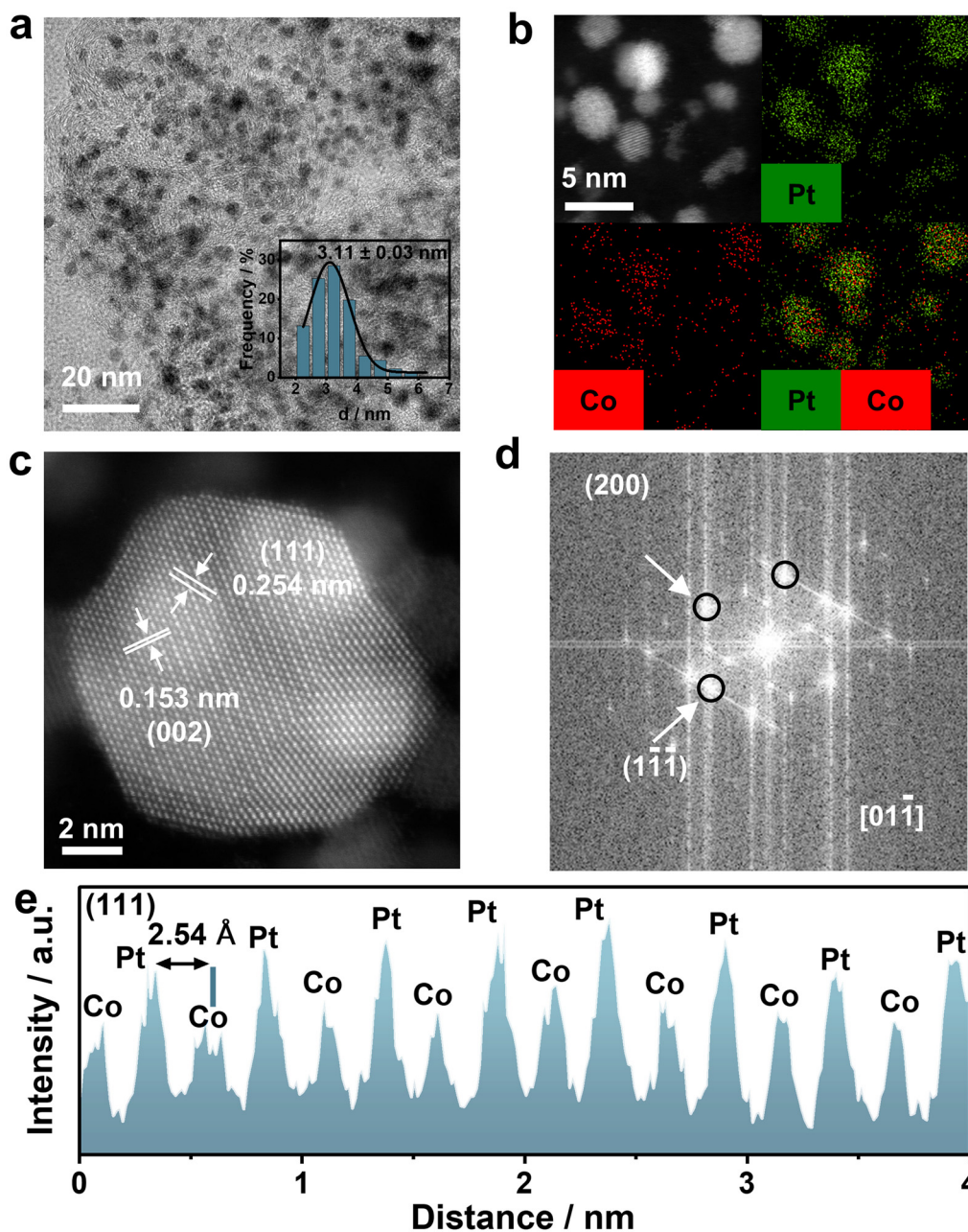


Figure S1. (a) Representative TEM image of Pt₃Co/C(Li_{0.0})-700 °C-2 h. (b) EDS-mapping images of Pt₃Co/C(Li_{0.0})-700 °C-2 h. (c) Atomic-resolution HAADF-STEM image of Pt₃Co/C(Li_{0.0})-700 °C-2 h. (d) FFT patterns of highlighted area from (c). (e) Line profile of Pt₃Co/C(Li_{0.0})-700 °C-2 h from part (c).

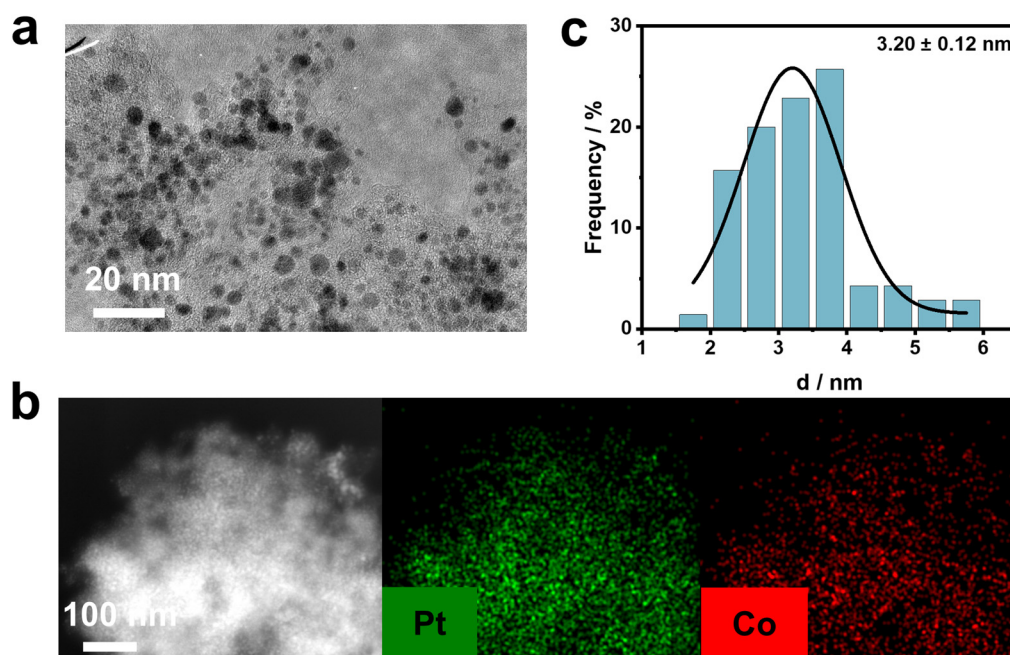


Figure S2. (a) Representative TEM image of Pt₃Co/C(Li_{0.3})-700 °C-2 h. (b) EDS-mapping of Pt₃Co/C(Li_{0.3})-700 °C-2 h. (c) Corresponding particle size distribution histogram derived from (a), with mean diameter indicated.

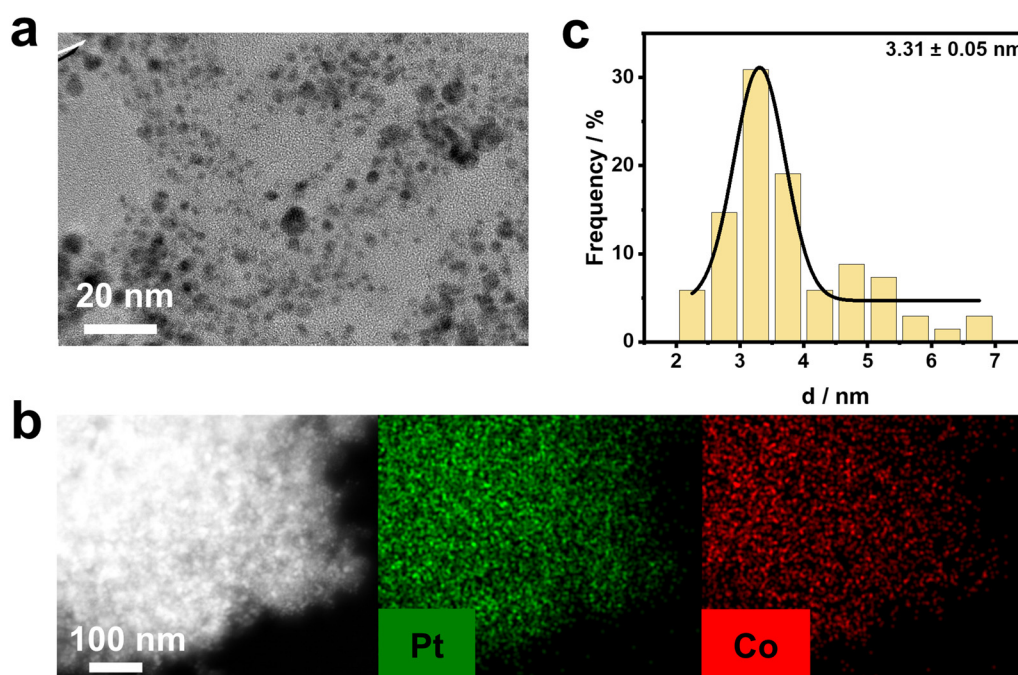


Figure S3. (a) Representative TEM image of Pt₃Co/C(Li_{0.5})-700 °C-2 h. (b) EDS-mapping of Pt₃Co/C(Li_{0.5})-700 °C-2 h. (c) Corresponding particle size distribution histogram derived from (a), with mean diameter indicated.

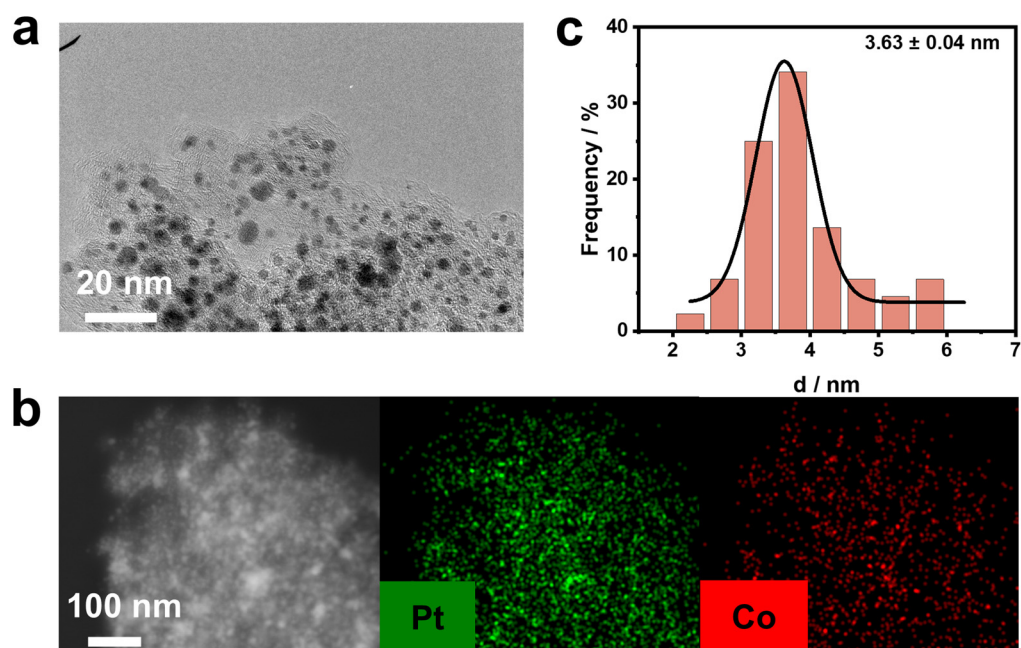


Figure S4. (a) Representative TEM image of Pt₃Co/C(Li_{0.7})-700 °C-2 h. (b) EDS-mapping of Pt₃Co/C(Li_{0.7})-700 °C-2 h. (c) Corresponding particle size distribution histogram derived from (a), with mean diameter indicated.

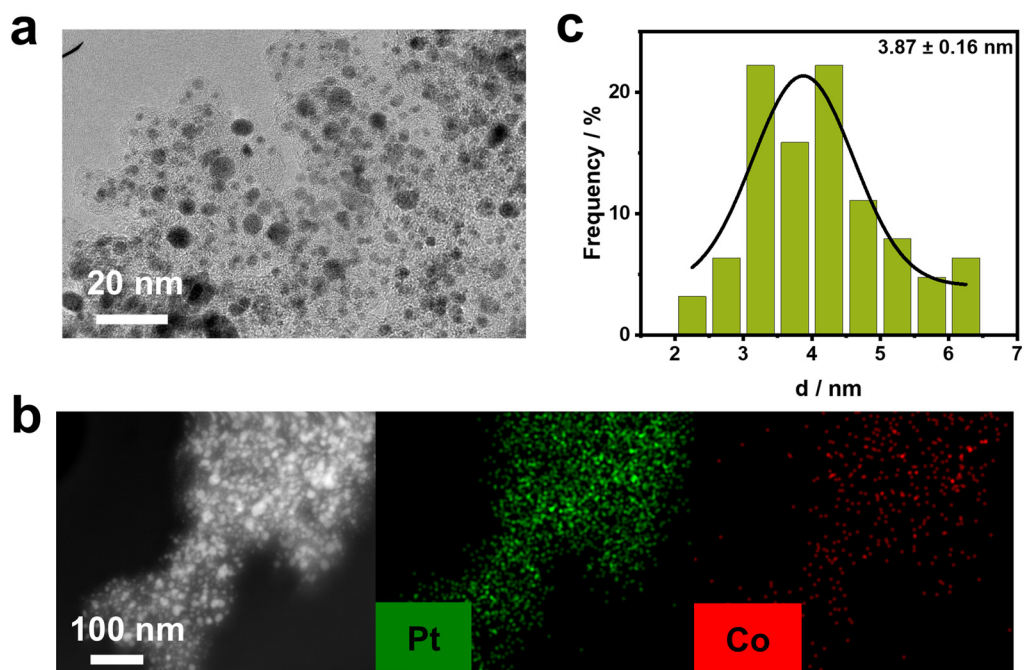


Figure S5. (a) Representative TEM image of Pt₃Co/C(Li_{1.0})-700 °C-2 h. (b) EDS-mapping of Pt₃Co/C(Li_{1.0})-700 °C-2 h. (c) Corresponding particle size distribution histogram derived from (a), with mean diameter indicated.

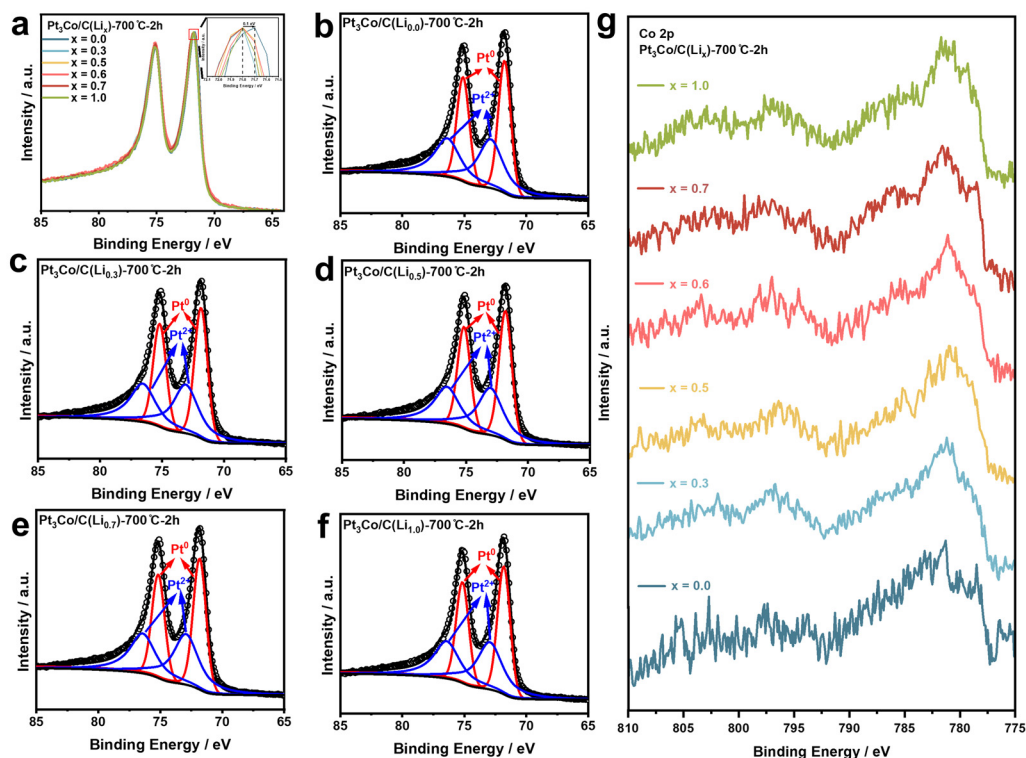


Figure S6. (a) High-resolution Pt 4f XPS of Pt₃Co/C(Li_x)-700 °C-2 h. High-resolution Pt 4f XPS and the corresponding peak deconvolution of (b) Pt₃Co/C(Li_{0.0})-700 °C-2 h (c) Pt₃Co/C(Li_{0.3})-700 °C-2 h (d) Pt₃Co/C(Li_{0.5})-700 °C-2 h (e) Pt₃Co/C(Li_{0.7})-700 °C-2 h (f) Pt₃Co/C(Li_{1.0})-700 °C-2 h. (g) High-resolution Co 2p XPS of Pt₃Co/C(Li_x)-700 °C-2 h (Weak signal not fitted for analysis).

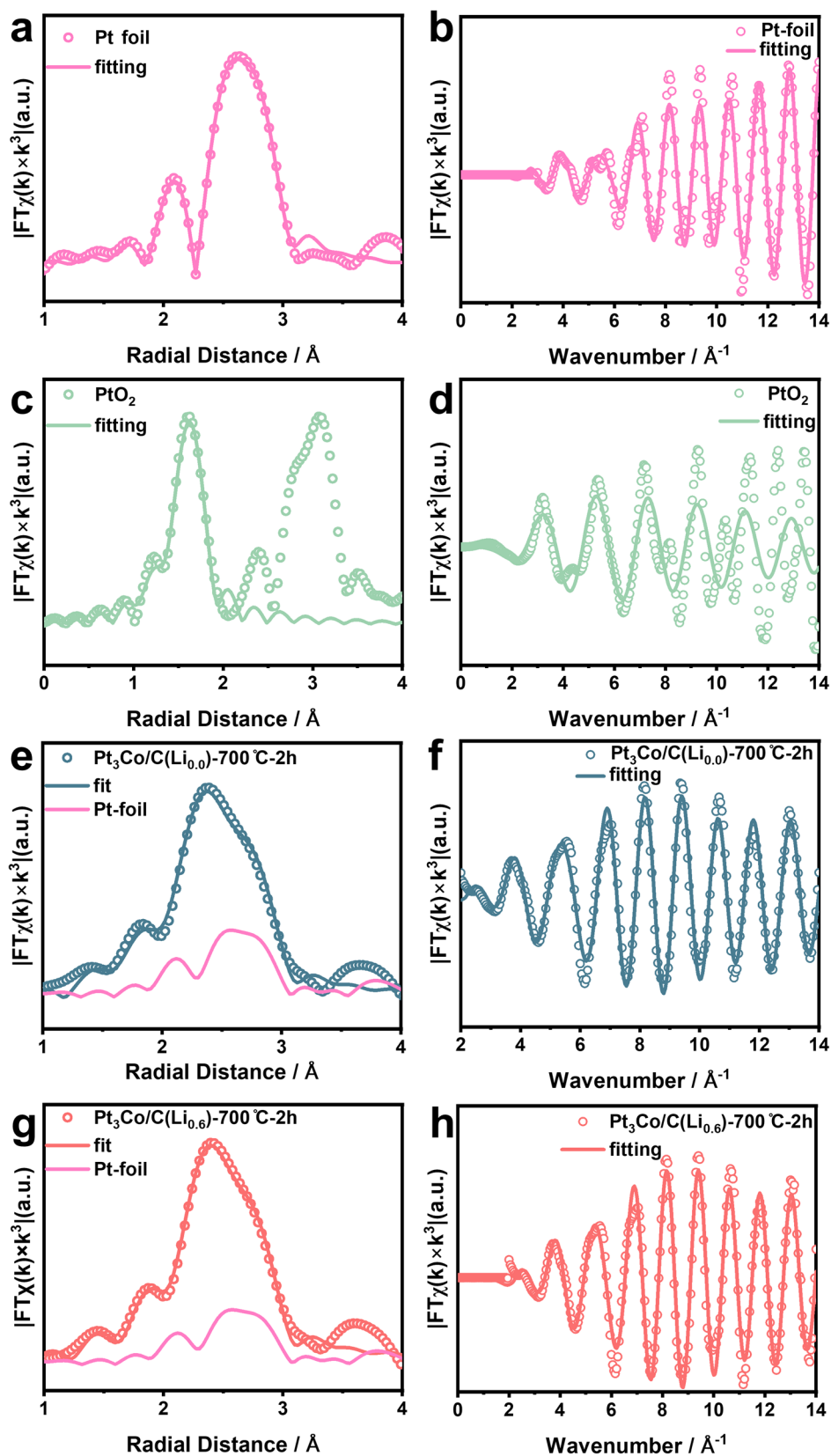
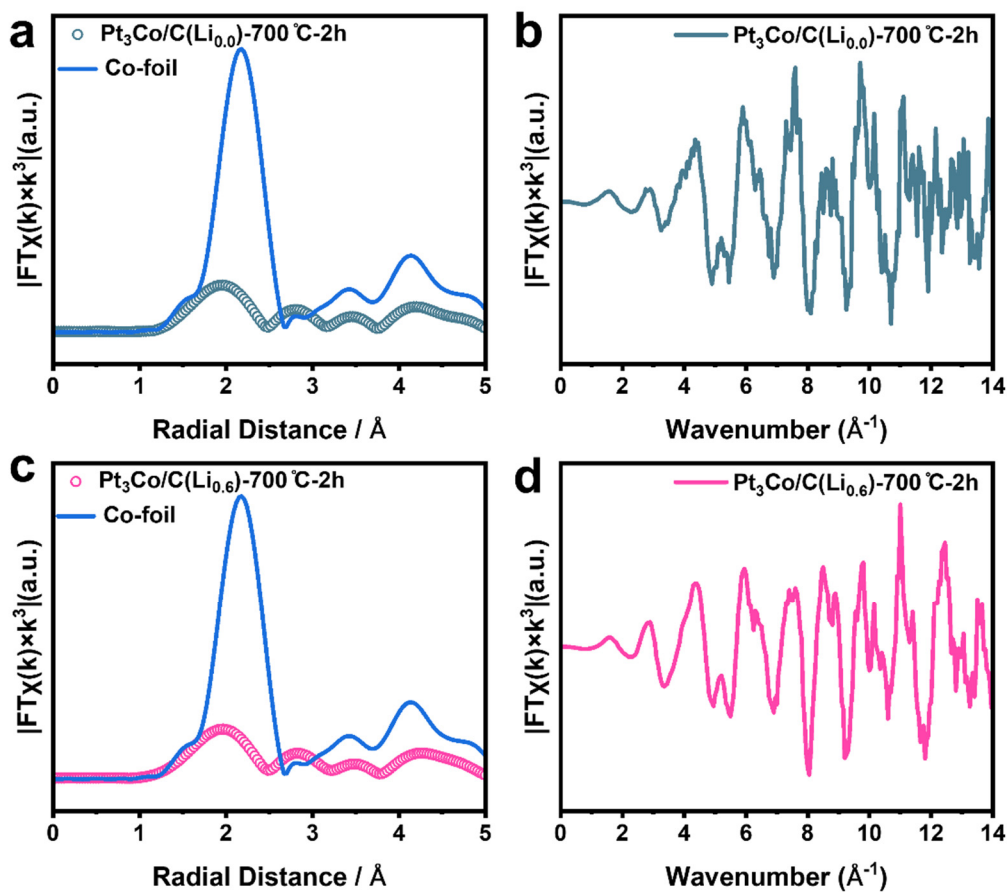


Figure S7. FT-EXAFS curves and EXAFS oscillation function and corresponding fitting curve at Pt L₃-edge of Pt foil ((a) and (b)), PtO₂ ((c) and (d)), Pt₃Co/C(Li_{0.6})-



700 °C-2 h ((e) and (f)), Pt₃Co/C(Li_{0.6})-700 °C-2 h ((g) and (h)).

Figure S8. FT-EXAFS curves and EXAFS oscillation function at Co K-edge of Pt₃Co/C(Li_{0.0})-700 °C-2 h ((a) and (b)), Pt₃Co/C(Li_{0.6})-700 °C-2 h ((c) and (d)) (Weak signal not fitted for analysis).

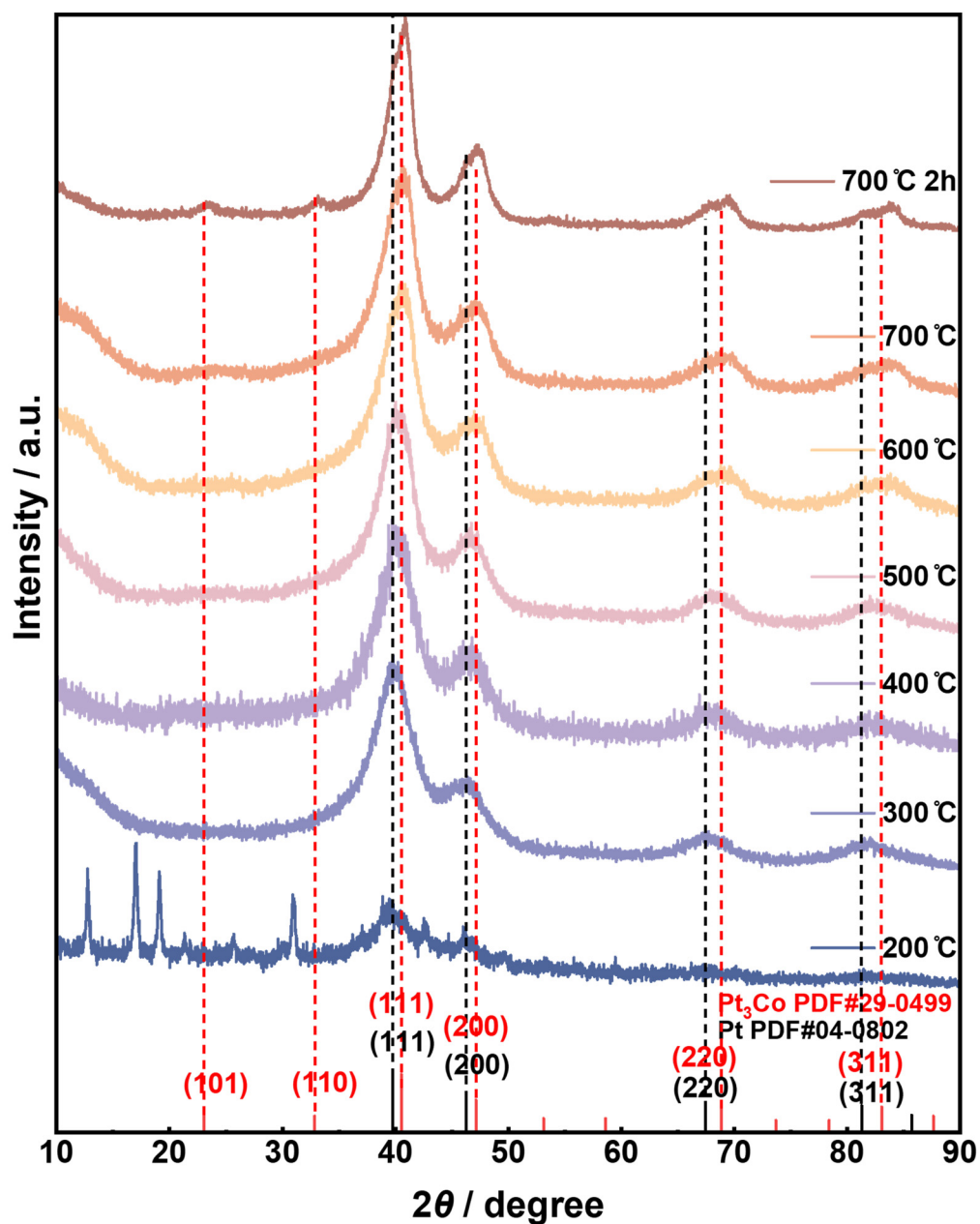


Figure S9. Temperature/time-evolution ex-situ XRD patterns of Pt₃Co/C(Li_{0.0})-700 °C-2 h catalyst during the annealing process.

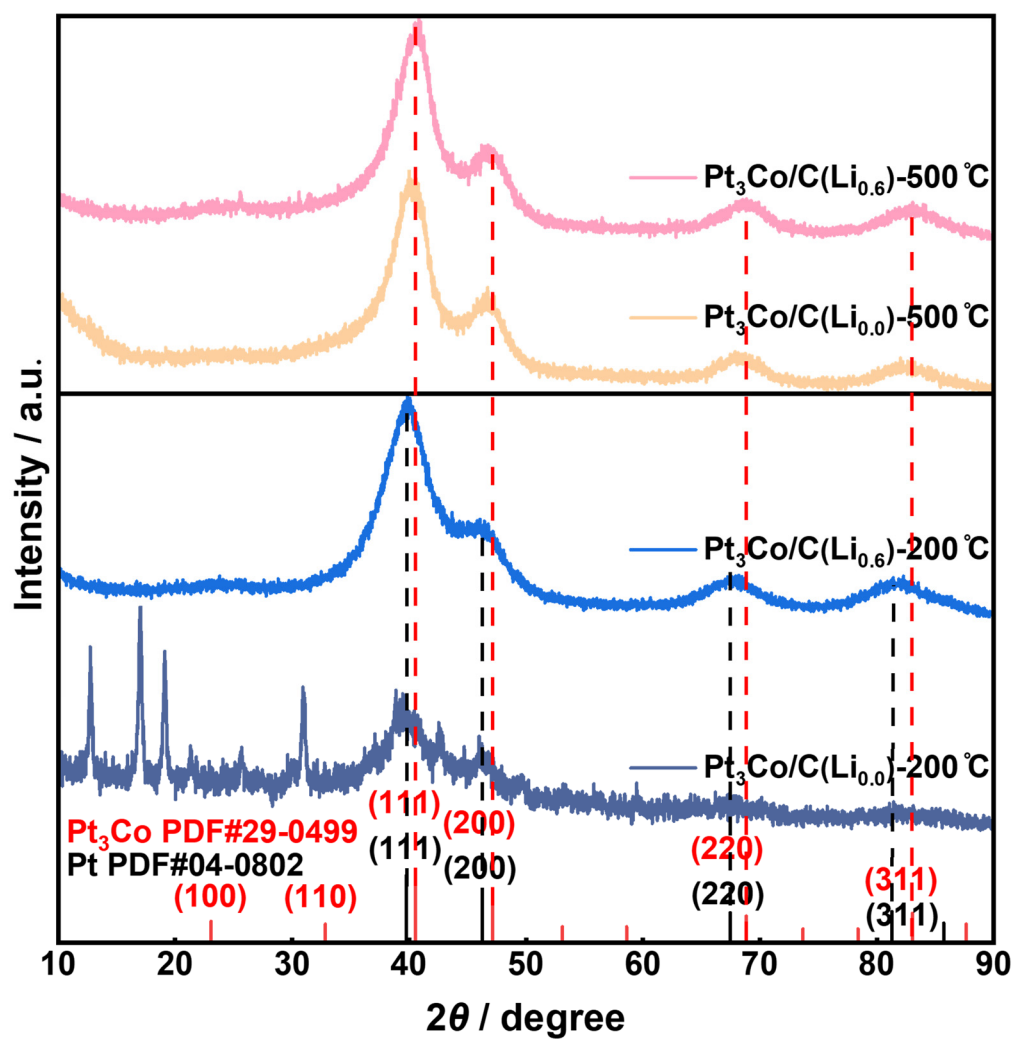


Figure S10. XRD patterns of $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-200^\circ\text{C}$, $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-200^\circ\text{C}$, $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-500^\circ\text{C}$ and $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-500^\circ\text{C}$.

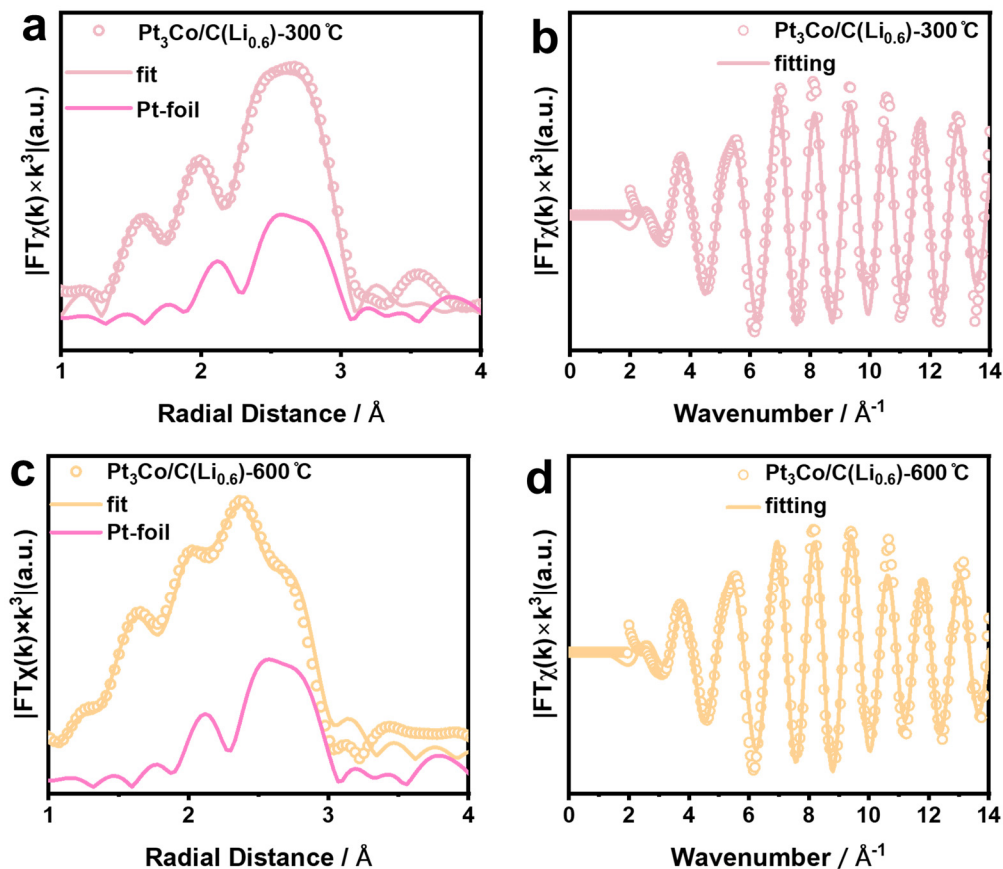


Figure S11. (a) FT-EXAFS curves at Pt L₃-edge of Pt₃Co/C(Li_{0.6})-300°C. (b) EXAFS oscillation function at Pt L₃-edge of Pt₃Co/C(Li_{0.6})-300°C. (c) FT-EXAFS curves at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-600°C. (d) EXAFS oscillation function at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-600°C.

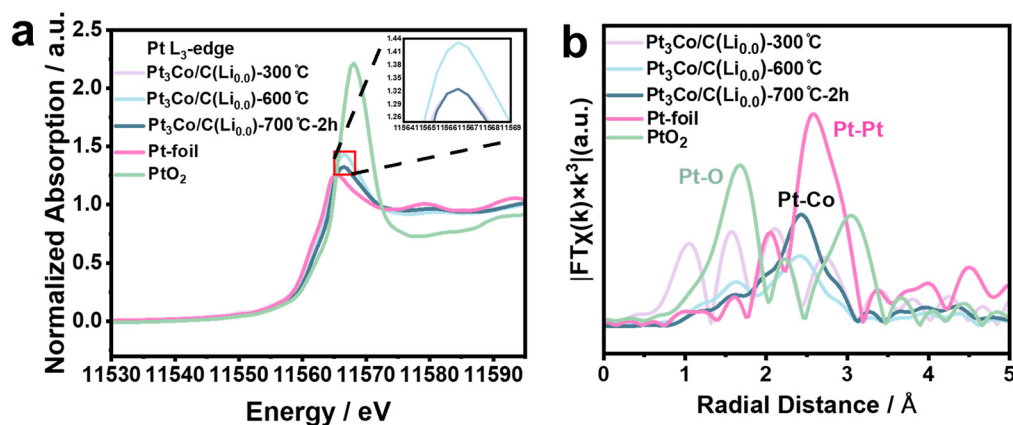


Figure S12. (a) Pt L₃-edge normalized XANES spectra of Pt foil, PtO₂, Pt₃Co/C(Li_{0.0})-300 °C, Pt₃Co/C(Li_{0.6})-600 °C and Pt₃Co/C(Li_{0.6})-700 °C-2 h. (b) Corresponding FT-EXAFS spectra of (a).

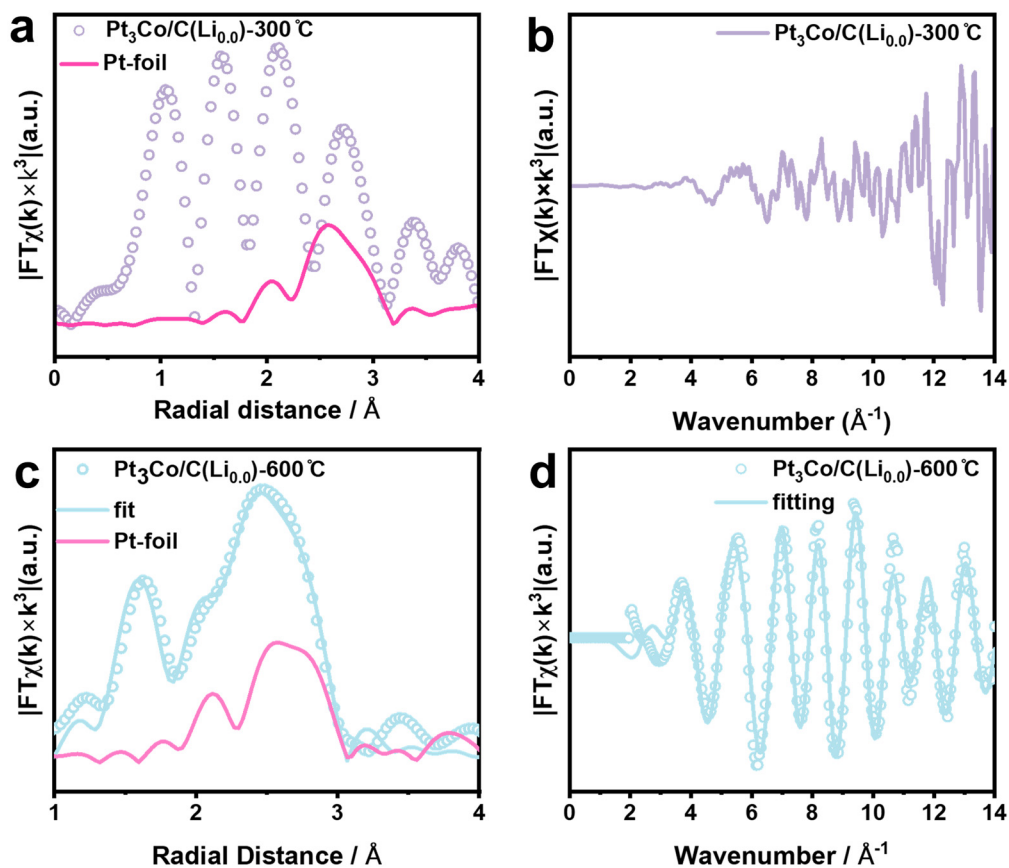


Figure S13. (a) FT-EXAFS curves at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-300°C. (b) EXAFS oscillation function at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-300°C (Weak signal not fitted for analysis). (c) FT-EXAFS curves at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-600°C. (d) EXAFS oscillation function at Pt L₃-edge of Pt₃Co/C(Li_{0.0})-600°C.

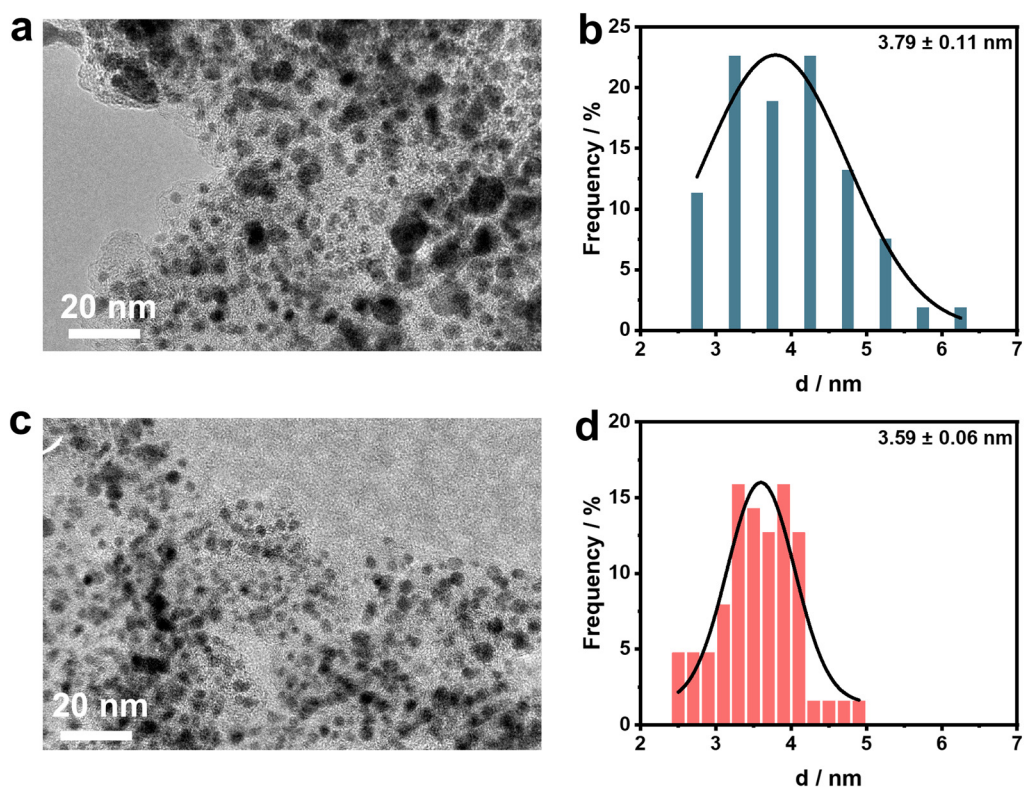


Figure S14. (a)TEM images of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.0})$ -700 °C-2 h after ADT. (b) Size distribution histogram of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.0})$ -700 °C-2 h after ADT. (c) TEM images of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.6})$ -700 °C-2 h after ADT. (d) Size distribution histogram of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.6})$ -700 °C-2 h after ADT.

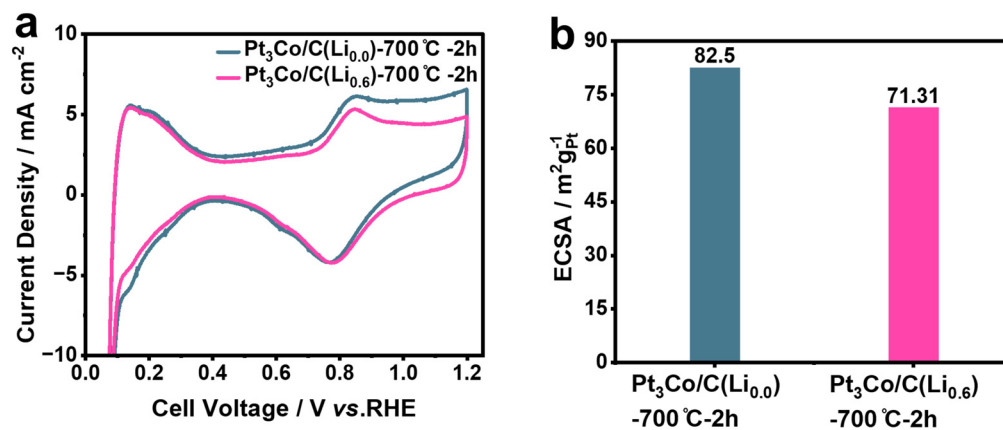


Figure S15. (a) Comparison of CV curves for Pt₃Co/C(Li_{0.0})-700 °C-2 h and Pt₃Co/C(Li_{0.6})-700 °C-2 h. (b) Comparison of ECSA for Pt₃Co/C(Li_{0.0})-700 °C-2 h and Pt₃Co/C(Li_{0.6})-700 °C-2 h.

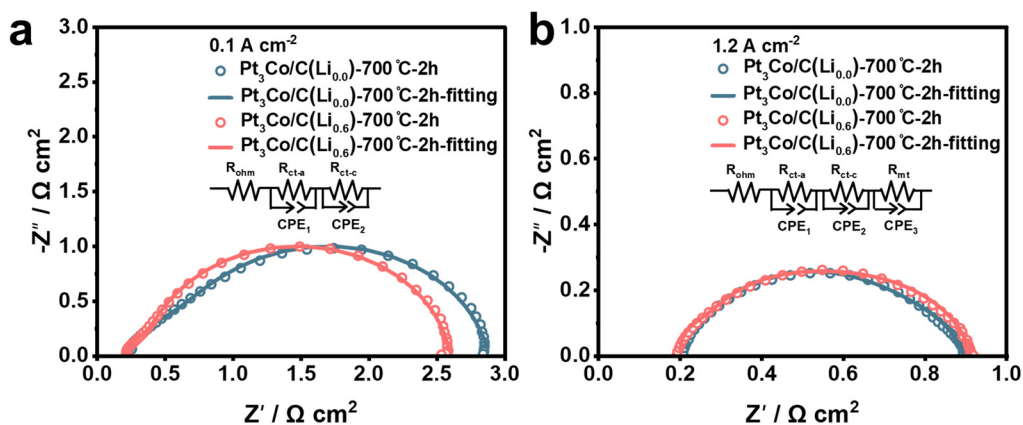


Figure S16. (a) Nyquist plots of EIS spectra recorded at 0.1 A cm^{-2} for $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-700^\circ\text{C-2 h}$ and $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-700^\circ\text{C-2 h}$. (b) Nyquist plots of EIS spectra recorded at 1.2 A cm^{-2} for $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-700^\circ\text{C-2 h}$ and $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-700^\circ\text{C-2 h}$.

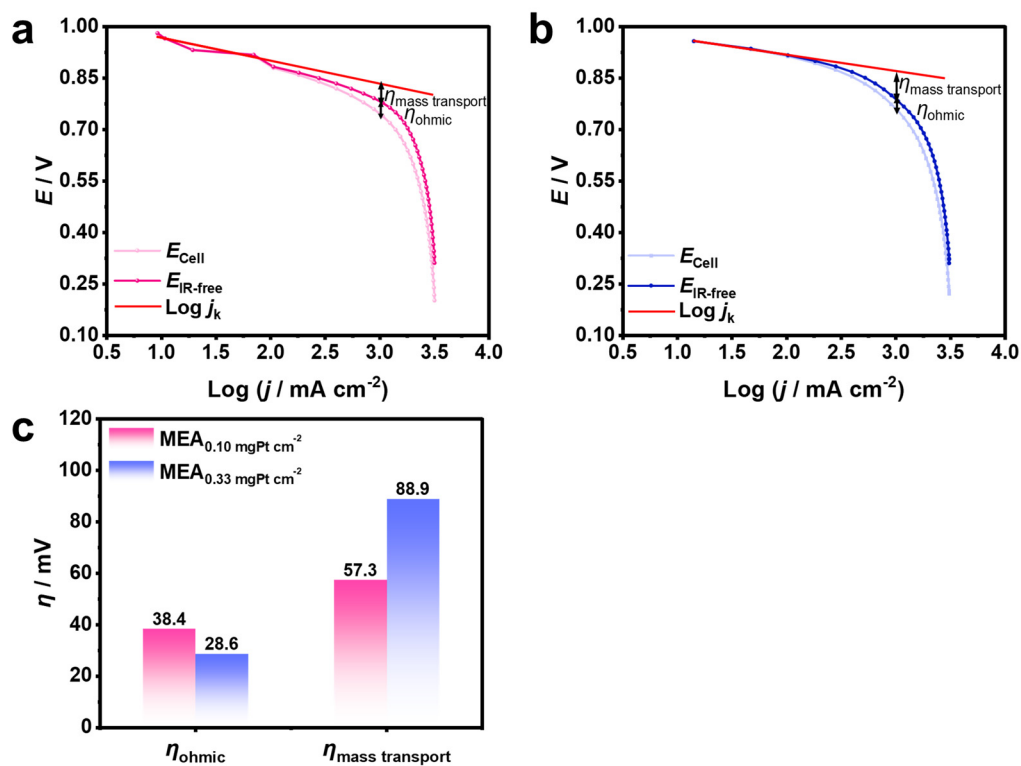


Figure S17. (a) Tafel plots and overpotential analysis for $\text{MEA}_{0.10 \text{ mgPt cm}^{-2}}$ under H_2 -air. (b) Tafel plots and overpotential analysis for $\text{MEA}_{0.33 \text{ mgPt cm}^{-2}}$ under H_2 -air. (c) Comparison of η_{ohmic} and $\eta_{\text{mass transport}}$ for $\text{MEA}_{0.10 \text{ mgPt cm}^{-2}}$ and $\text{MEA}_{0.33 \text{ mgPt cm}^{-2}}$.

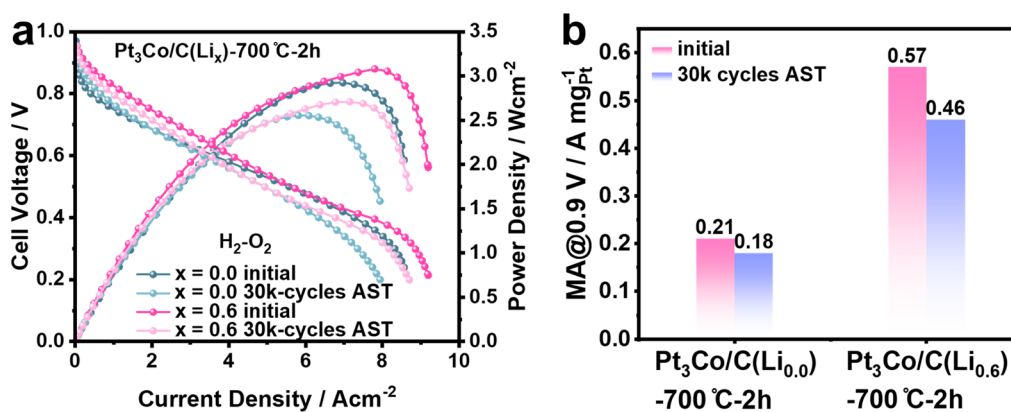


Figure S18. (a) Comparison of steady-state polarization curves of the $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-700\text{ }^\circ\text{C}-2\text{ h}$ and $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-700\text{ }^\circ\text{C}-2\text{ h}$ before and after AST under $\text{H}_2\text{-air}$. (b) Comparison of MA for $\text{Pt}_3\text{Co/C(Li}_{0.0}\text{)}-700\text{ }^\circ\text{C}-2\text{ h}$ and $\text{Pt}_3\text{Co/C(Li}_{0.6}\text{)}-700\text{ }^\circ\text{C}-2\text{ h}$ before and after AST.

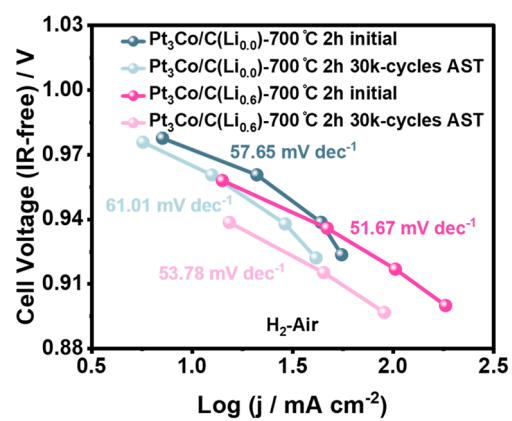


Figure S19. Tafel curves of the Pt₃Co/C(Li_{0.0})-700 °C-2 h and Pt₃Co/C(Li_{0.6})-700 °C-2 h before and after AST under H₂-air.

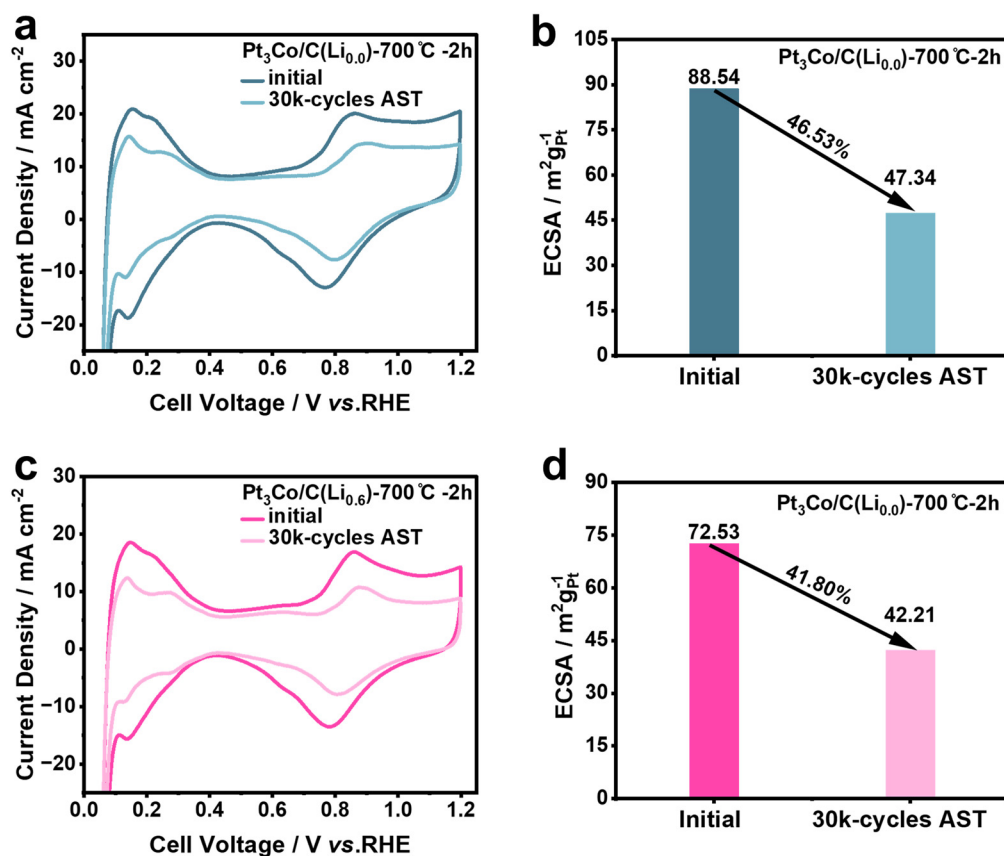


Figure S20. (a) Comparison of CV curves for Pt₃Co/C(Li_{0.0})-700 °C-2 h before and after AST. (b) Comparison of ECSA for Pt₃Co/C(Li_{0.0})-700 °C-2 h before and after AST. (c) Comparison of CV curves for Pt₃Co/C(Li_{0.6})-700 °C-2 h before and after AST. (d) Comparison of ECSA for Pt₃Co/C(Li_{0.6})-700 °C-2 h before and after AST.

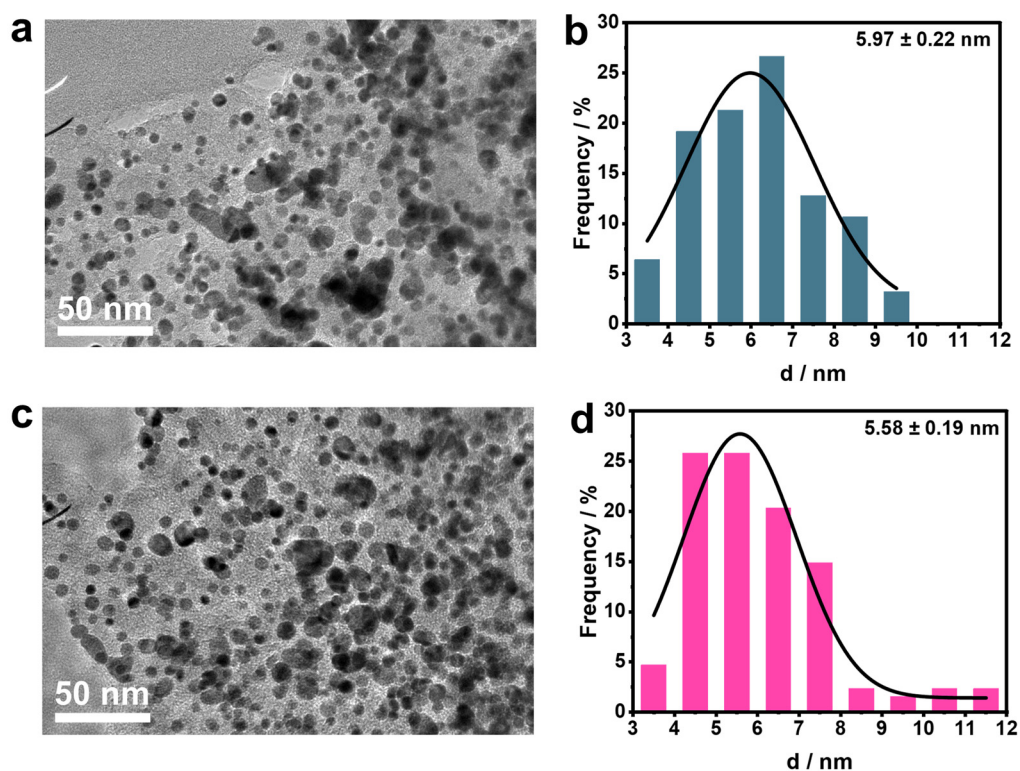


Figure S21. (a)TEM images of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.0})$ -700 °C-2 h after AST. (b) Size distribution histogram of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.0})$ -700 °C-2 h after AST. (c) TEM images of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.6})$ -700 °C-2 h after AST. (d) Size distribution histogram of $\text{Pt}_3\text{Co}/\text{C}(\text{Li}_{0.6})$ -700 °C-2 h after AST.

Table S1. The mass fractions and atomic ratios of each element in different samples
(The mass fraction was determined by ICP-OES).

Sample (Pt₃Co/C(Li_x)-700 °C-2 h)	Pt /wt. %	Co /wt. %	Li /wt. %	Atomic ratio Pt:Co:Li
<i>x</i> = 0.0	49.12	5.15	0.00	1:0.34
<i>x</i> = 0.3	47.15	4.41	0.20	1:0.31:0.12
<i>x</i> = 0.5	50.36	4.54	0.19	1:0.30:0.11
<i>x</i> = 0.6	50.43	5.46	0.21	1:0.36:0.12
<i>x</i> = 0.7	51.46	5.63	0.22	1:0.36:0.12
<i>x</i> = 1.0	48.91	5.32	0.27	1:0.36:0.15
<i>x</i> = 0.6 (after washing)	52.30	4.57	0.013	1:0.29:0.007

Table S2. Comparison of particle size (d) and Pt loading between Pt₃Co/C(Li_{0.6})-700 °C-2 h and previously reported conventional PtM-IMC catalysts.

Catalysts	Pt loading (wt%)	d (nm)	Refs.
Pt ₃ Co/C(Li _{0.6})-700 °C-2 h	50.43	3.58 ± 0.02	This work
PtCoNiGaZn@Pt/C	40.53	3.15	1
PtCo-O/C-6	30.6	3.2 ± 1.0	2
O-Pt ₃ Co/NC1100	12.5	4.6	3
L1 ₀ -PtPdFe/C	22.8	6.12 ± 0.8	4
i-CoPt@Pt/KB	40.05	< 5	5
PtFe/NcPS	4.8	1.48 ± 0.82	6
Pt ₃ Co/Te-C	18.2	4.2 ± 1.03	7
Pt ₁ Co ₁ -IMC@Pt/C	44.7	5.6	8
DC-Pt ₁ Co ₁ /CNT	8	≈ 4	9
L1 ₀ -NPtCo@Pt/C	21.8	≈ 5	10

Table S3. Analysis of Pt 4f XPS peak including peak position, half peak width and integral area.

Sample	Pt ⁰ (eV)		Pt ²⁺ (eV)		Pt ⁰	Pt ²⁺
(Pt ₃ Co/C(Li _x)-700 °C-2 h)	4f _{7/2}	4f _{5/2}	4f _{7/2}	4f _{5/2}	4f (%)	4f (%)
<i>x</i> = 0.0	71.75	75.10	72.87	76.37	52.49	47.51
<i>x</i> = 0.3	71.85	75.19	73.08	76.55	53.14	46.86
<i>x</i> = 0.5	71.82	75.16	73.03	76.53	54.08	45.92
<i>x</i> = 0.6	71.80	75.16	73.03	76.53	55.24	44.76
<i>x</i> = 0.7	71.83	75.17	72.93	76.43	52.60	47.40
<i>x</i> = 1.0	71.81	75.15	72.94	76.43	52.60	47.40

All XPS spectra were fitted using Gaussian-Lorentzian peak shapes. The binding energy scale was calibrated using adventitious carbon (C 1s at 284.80 eV). For Pt 4f spectra, the doublet separation was fixed at 3.35 eV with an area ratio of 4:3 (Pt 4f_{7/2} Pt 4f_{5/2}).

Table S4. The fitting parameters of Pt L₃-edge for various samples.

Sample	Path	CN	$R(\text{\AA})$	$\sigma^2(\text{\AA}^2)$	ΔE_0 (eV)	R factor
Pt foil	Pt-Pt	12.11	2.76	0.0046	5.46	0.0183
PtO ₂	Pt-O	5.48	2.02	0.0025	0.0599	0.0187
	Pt-O	1.15	1.99	0.0026		
Pt ₃ Co/C(Li _{0.0})-600°C	Pt-Co	0.88	2.66	0.0095	0.8590	0.0142
	Pt-Pt	7.09	2.72	0.0095		
	Pt-O	0.31	1.98	0.0000		
Pt ₃ Co/C(Li _{0.0})-700°C-2h	Pt-Co	1.27	2.68	0.0080	0.8853	0.0133
	Pt-Pt	9.12	2.71	0.0080		
	Pt-O	0.33	1.98	0.0000		
Pt ₃ Co/C(Li _{0.6})-300°C	Pt-Cl	0.57	2.26	0.0087	0.3014	0.0040
	Pt-Pt	8.05	2.74	0.0087		
	Pt-O	0.67	1.99	0.0024		
Pt ₃ Co/C(Li _{0.6})-600°C	Pt-Co	0.92	2.67	0.0090	0.2511	0.0092
	Pt-Pt	8.69	2.72	0.0090		
	Pt-O	0.25	1.99	0.0000		
Pt ₃ Co/C(Li _{0.6})-700°C-2h	Pt-Co	1.56	2.68	0.0076	0.5270	0.0089
	Pt-Pt	9.13	2.72	0.0076		

CN , coordination numbers; R , distance between absorber and backscatter atoms; σ^2 , Debye-Waller factor to account for both thermal and structural disorders; ΔE_0 , inner potential correction; R -factor, a measure of the quality of the EXAFS fit.

Table S5. Particle size distribution and diffraction peak position (DPP) changes of (111), (200), and (220) planes at different temperatures for Pt₃Co/C(Li_{0.0}) and Pt₃Co/C(Li_{0.6}).

T	DPP₍₁₁₁₎		DPP₍₂₀₀₎		DPP₍₂₂₀₎	
	Pt₃Co/C(Li_{0.0})	Pt₃Co/C(Li_{0.6})	Pt₃Co/C(Li_{0.0})	Pt₃Co/C(Li_{0.6})	Pt₃Co/C(Li_{0.0})	Pt₃Co/C(Li_{0.6})
200 °C	/	39.73	/	46.25	/	67.53
300 °C	39.78	39.90	46.27	46.44	67.67	68.40
400 °C	40.28	40.39	46.54	46.61	68.24	68.45
500 °C	40.38	40.51	46.78	47.10	68.49	68.78
600 °C	40.52	40.53	47.23	47.18	68.69	68.79
700 °C	40.53	40.57	47.22	47.20	68.95	69.03
700 °C-2 h	40.82	40.58	47.35	47.22	69.07	69.04

Table S6. MA, SA, ECSA, and $E_{1/2}$ of Pt₃Co/C(Li_x)-700 °C-2 h.

Sample (Pt₃Co/C(Li_x)-700 °C-2 h)	MA@0.9 V A mg_{Pt}⁻¹	SA@0.9 V mA cm⁻²	ECSA m² g_{Pt}⁻¹	$E_{1/2}$ V vs. RHE
$x = 0.0$	0.55 ± 0.01	1.03 ± 0.01	52.62 ± 0.39	0.922 ± 0.004
$x = 0.3$	0.60 ± 0.05	1.19 ± 0.07	52.13 ± 0.24	0.929 ± 0.004
$x = 0.5$	0.66 ± 0.02	1.31 ± 0.01	50.32 ± 0.81	0.926 ± 0.002
$x = 0.6$	0.86 ± 0.04	1.75 ± 0.07	49.23 ± 0.61	0.934 ± 0.004
$x = 0.7$	0.62 ± 0.00	1.28 ± 0.01	48.97 ± 0.28	0.928 ± 0.001
$x = 1.0$	0.60 ± 0.04	1.23 ± 0.06	48.41 ± 0.49	0.920 ± 0.002

Table S7. EIS parameters of the MEA at 0.1 and 1.2 A cm⁻².

Catalyst	0.1 A cm⁻²		1.2 A cm⁻²		
	<i>R_{ohm}</i> / Ω cm ²	<i>R_{ct-c}</i> / Ω cm ²	<i>R_{ohm}</i> / Ω cm ²	<i>R_{ct-c}</i> / Ω cm ²	<i>R_{mt}</i> / Ω cm ²
Pt₃Co/C(Li_{0.0})-700°C-2h	0.23	2.27	0.22	0.39	0.23
Pt₃Co/C(Li_{0.6})-700°C-2h	0.22	2.23	0.20	0.25	0.42

Table S8. Comparison of ORR activity and durability between Pt₃Co/C(Li_{0.6})-700°C-2h and previously reported conventional PtM-IMC catalysts.

Catalysts	MA @0.9V(RDE) A mg _{Pt} ⁻¹	PPD / Wcm ⁻²		MA Loss (AST) %	Refs.
		H ₂ -Air	H ₂ -O ₂		
Pt ₃ Co/C(Li _{0.6})-700°C-2h	0.86	1.23	2.92	19.3	This work
PtCoNiGaZn@Pt/C	0.65	0.96	2.44	12.5	1
PtCo-O/C-6	1.28	1.25	2.38	18.5	2
O-Pt ₃ Co/NC1100	1.21	1.27	-	-	3
L1 ₀ -PtPdFe/C	0.89	-	-	32.8	4
i-CoPt@Pt/KB	2.07	1.18	-	35.5	5
PtFe/NcPS	1.35	-	-	33.8	6
Pt ₃ Co/Te-C	0.30	-	2.32	26	7
Pt1Co1-IMC@Pt/C	0.53	1.23	2.30	24.8	8
Pt ₃ Ga	2.18	1.06	-	-	11
Pt ₅₆ Mn ₄₄	0.53	-	1.36	-	12
PtNiMoAu NWs/C	2.42	0.92	1.75	22.6	13
PtCo@Gnp	-	1.01	-	34.2	14
L1 ₀ -CoPt/Pt	-	1.1	-	19.6	15
L1 ₀ -Cr-PtFe/C	2.05	1.15	-	25	16

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