

Supporting information

Electrochemical selective recovery of thorium from rare earth by amidoxime modified graphite felt electrode

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Table S1 Chemicals and materials used in the experiments

Compound	Supplier	CAS	Purity, wt%
Polyacrylonitrile	Damas-beta	25014-41-9	99%
Hydroxylamine Hydrochloride	Damas-beta	5470-11-1	99%
Carbon felt	-	-	99%
Thorium nitrate	-	13823-29-5	99%
NaOH	Greagent	1310-73-2	≥ 98%
Potassium bromide	Damas-beta	7758-02-3	≥ 99.99%
N,N-dimethylformamide	Shanghai Laiyashi Chemical Co., Ltd.	68-12-6	99%
Rare earth oxides (REO)	Fujian Changting Golden Dragon Rare Earth Co., Ltd.	-	≥ 99%
Rare earth nitrates (RE(NO ₃) ₃)	Prepared in our lab		> 99%
Deionized water	Prepared by Shanghai Hetai Smart-S30UV ultrapure water system	-	The resistivity is 18.2 MΩ*cm

Table S2 Comparison with electrochemical and functionalized adsorbents

Method	Material	Q mg/g	Performance	Ref.
Physicochemical adsorption	KMS-5	90.91	pH = 1.0, t = 0.5 h	[1]
	MOF	350	pH = 3.0, t = 6 h	[2]
	COF-DL229	513	pH = 4.0, t = 1.0 h	[3]
	[NH ₄] ⁺ [COF-SO ₃ ⁻]	395	pH = 2.8, t = 2 h	[4]
	Nanoribbons	457.9	pH = 3.0, t = 4 h	[5]
	Ti ₂ CTxMXene	213.2	pH = 3.0, t = 12 h	[6]
	GO-PDA	718	pH = 4.0, t = 6 h	[7]
	Graphene	515	pH = 3.0, t = 24 h	[8]
	Graphene oxide	214.6	pH = 2.6, t = 4 h	[9]
Electrosorption	Activated carbon	8.4	pH = 3.0, t = 3 h, Voltage 0.8 V	[10]
	g-C ₃ N ₄ /AX	275.74	pH = 4.0, t = 5h, Voltage 1.0 V	[11]
	g-C ₃ N ₄ /P	341.93	pH = 3.0, t = 5h, Voltage 0.2 V	[12]
	PANI/AC	238.3	pH = 3.0, t = 5h, Voltage 2.4 V	[13]
	CF-AO	3158	pH = 3.0, t = 12 h, Voltage 3.0V	This work
	CF-AO	1142	pH = 3.0 t = 1 h, Voltage 3.0 V	This work

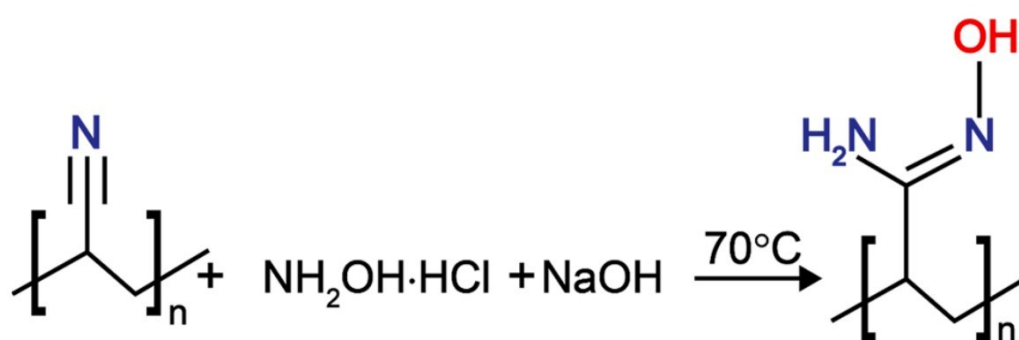


Fig. S1. Reaction chemistry equation

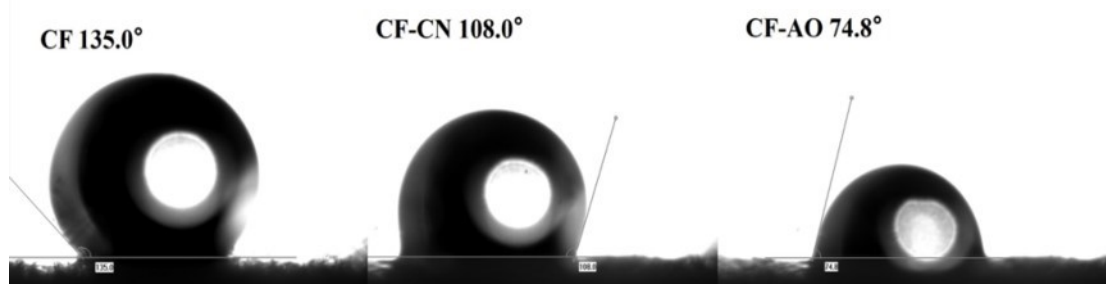


Fig. S2. Contact Angle between the material and water

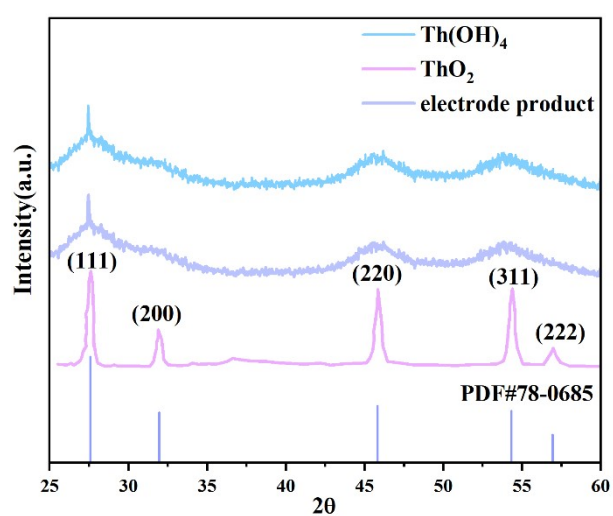


Fig. S3. PXRD of electrode products

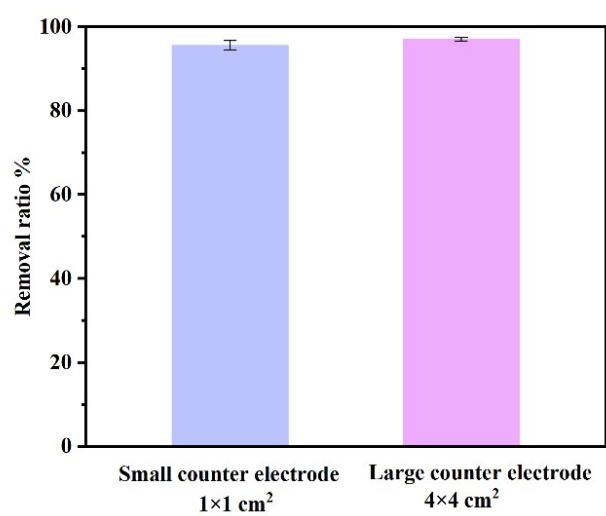


Fig. S4. Effect of counter electrode size

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