

Supplementary Information

**Silicalite-1 zeolite encapsulated Cu-ZnO nanoparticles for selective CO₂
hydrogenation to oxygenates**

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Figure Captions

Fig. S1 Pore size distribution of CZ(x)@S-t catalysts and CZ@Si precursor via N₂-sorption analysis.

Fig. S2 NH₃-TPD profiles of Cu@S-48 and CZ(0.4)@S-48 catalyst.

Fig. S3 H₂-TPR profiles of Cu@S-48 and CZ(0.4)@S-48 catalyst.

Fig. S4 Catalytic activity with time on stream (TOS) on CZ@Si for CO₂ hydrogenation to methanol. Catalytic reaction condition: T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction.

Fig. S5 Catalytic activity with time on stream (TOS) on CZ(0.9)@S-t catalysts for CO₂ hydrogenation to form DME and methanol oxygenates at the reaction condition of T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction; (a) CZ(0.9)@S-12, (b) CZ(0.9)@S-24, (c) CZ(0.9)@S-48, (d) CZ(0.9)@S-72 .

Fig. S6 Catalytic activity with time on stream (TOS) on CZ(0.9)@S-t catalysts for CO₂ hydrogenation to form DME and methanol oxygenates at the reaction condition of T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction; (a) Cu@S-48, (b) CZ(0.4)@S-48.

Fig. S7 in situ DRIFTS analysis results of CZ(0.9)@S-12 under the reaction condition of T = 250 °C, P = 2.0 MPa, H₂/CO₂ molar ratio of 3/1.

Table S1 Chemical compositions of CZ(x)@S-t catalysts measured by XRF analysis

Table S2 Previously reported zeolite-encapsulated Cu-based catalysts for CO₂ hydrogenation activity to oxygenates.

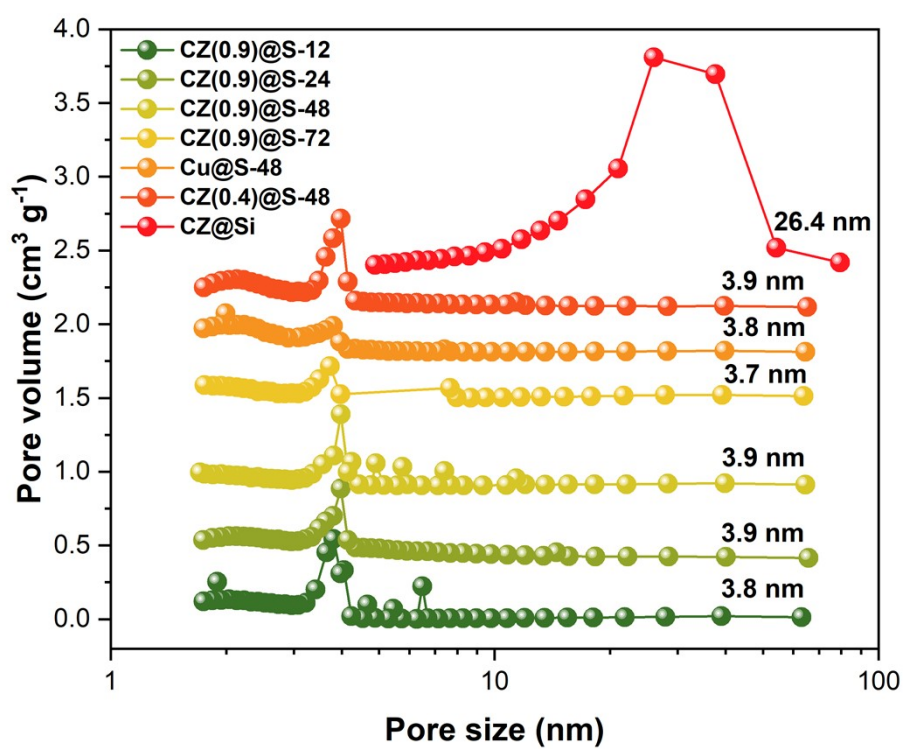


Fig. S1 Pore size distribution of CZ(x)@S-t catalysts and CZ@Si precursor via N₂-sorption analysis.

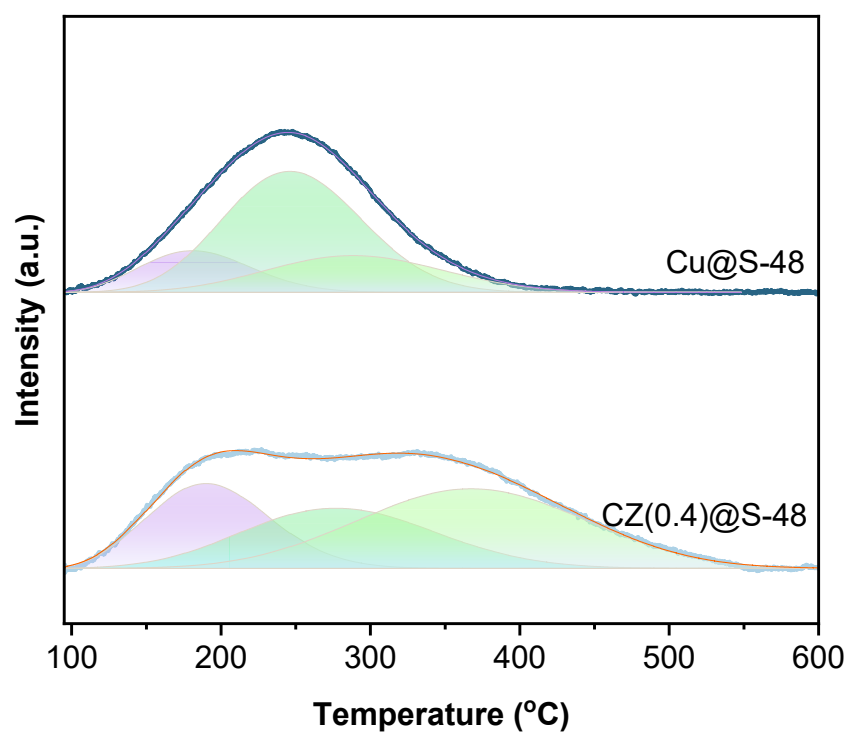


Fig. S2 NH₃-TPD profiles of Cu@S-48 and CZ(0.4)@S-48 catalyst.

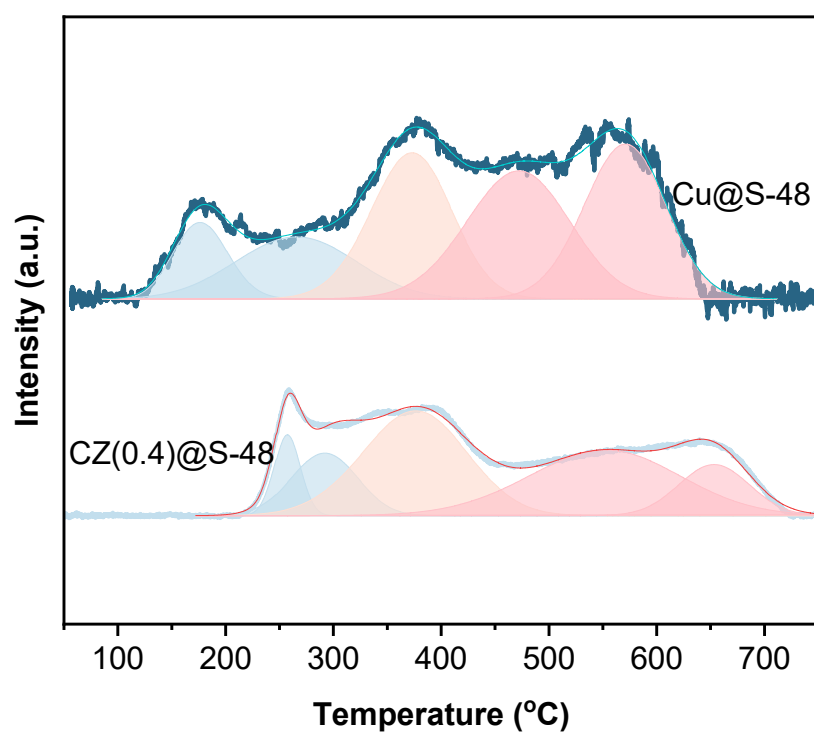


Fig. S3 H₂-TPR profiles of Cu@S-48 and CZ(0.4)@S-48 catalyst.

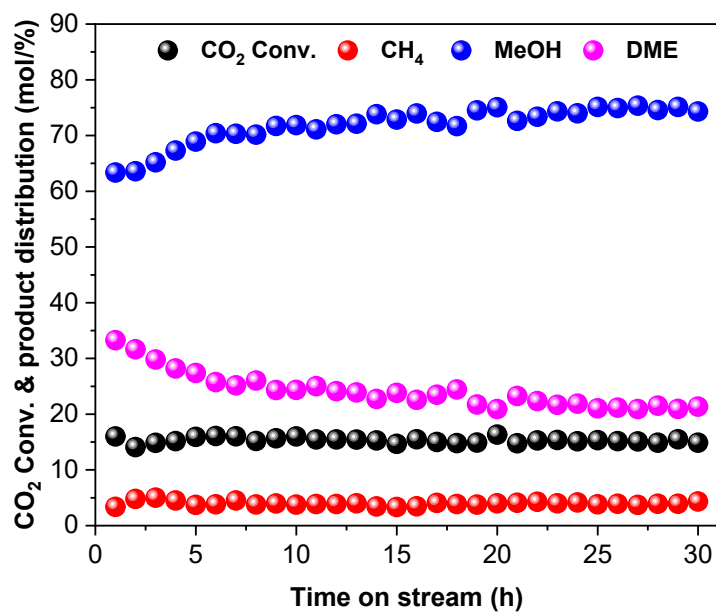


Fig. S4 Catalytic activity with time on stream (TOS) on CZ@Si for CO₂ hydrogenation to methanol. Catalytic reaction condition: T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction.

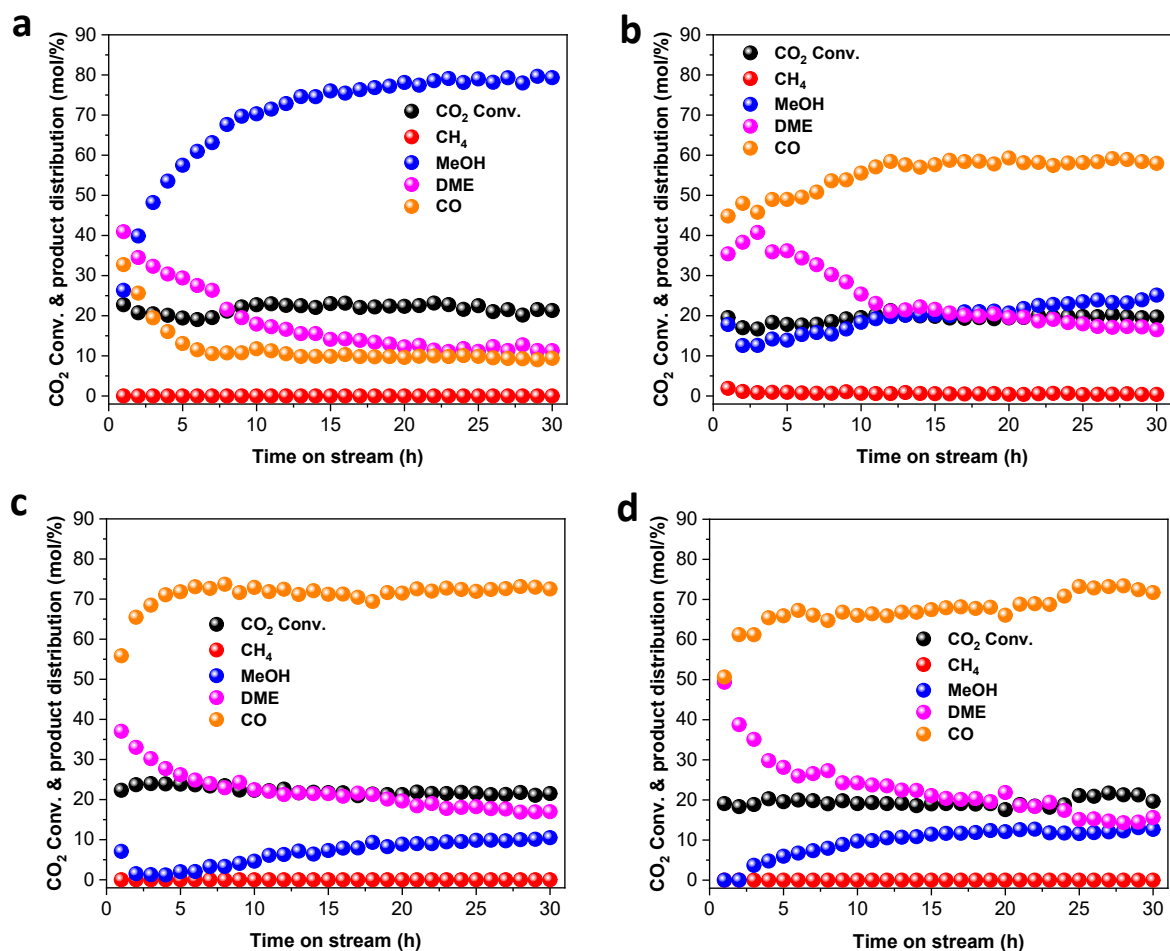


Fig. S5 Catalytic activity with time on stream (TOS) on CZ(0.9)@S-t catalysts for CO₂ hydrogenation to form DME and methanol oxygenates at the reaction condition of T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction; (a) CZ(0.9)@S-12, (b) CZ(0.9)@S-24, (c) CZ(0.9)@S-48, (d) CZ(0.9)@S-72.

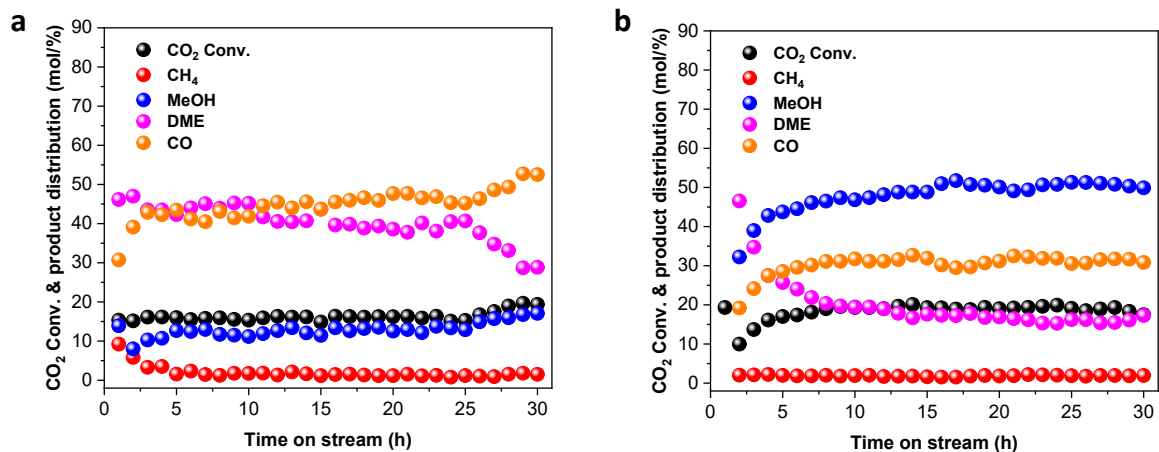


Fig. S6 Catalytic activity with time on stream (TOS) on CZ(0.9)@S-t catalysts for CO₂ hydrogenation to form DME and methanol oxygenates at the reaction condition of T = 250 °C, P = 5.0 MPa, H₂/CO₂ molar ratio = 3 (4% N₂ balance) at space velocity (SV) = 3000 mL g_{cat}⁻¹ h⁻¹ for 30 h reaction; (a) Cu@S-48, (b) CZ(0.4)@S-48.

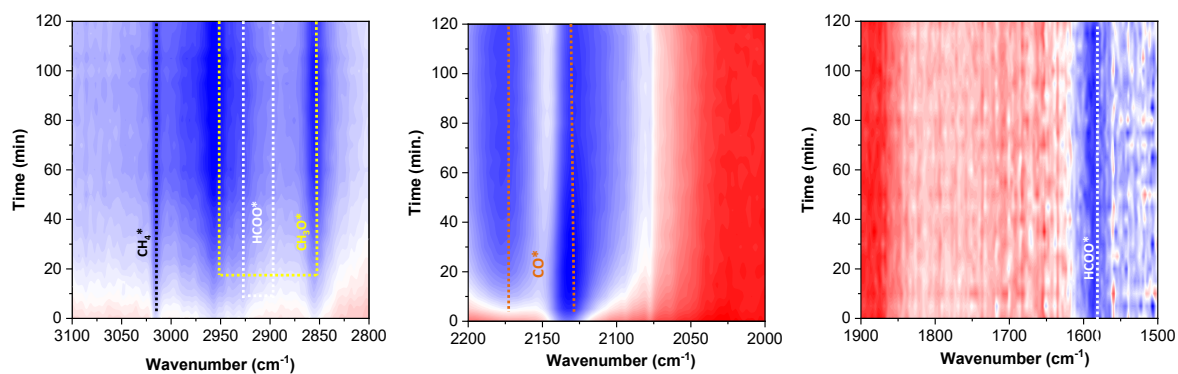


Fig. S7 *in situ* DRIFTS analysis results of CZ(0.9)@S-12 under the reaction condition of T = 250 °C, P = 2.0 MPa, H₂/CO₂ molar ratio of 3/1.

Table S1 Chemical compositions of CZ(x)@S-t catalysts measured by XRF analysis

Catalyst	XRF compound content (wt%)		
	CuO	ZnO	SiO ₂
CZ(0.9)@S-12	3.9	4.2	91.9
CZ(0.9)@S-24	3.7	4.1	92.2
CZ(0.9)@S-48	3.7	4.3	92.0
CZ(0.9)@S-72	3.5	4.0	92.5
Cu@S-48	3.5	-	96.5
CZ(0.4)@S-48	1.6	4.1	94.3

Table S2 Previously reported zeolite-encapsulated Cu-based catalysts for CO₂ hydrogenation activity to oxygenates.

Catalysts	Temperature (°C)	CO ₂ conversion (%)	Methanol selectivity (%)	DME selectivity (%) [*]	Ref.
ZnO/Cu@S-1	250	4.7	55.3	N	1
Cu/ZnO _x @Na-ZSM-5	250	6.3	77.8	N	2
CX-220	290	10.2	91.2	N	3
Cu0.2m-NaY	250	~0.5	~68	N	4
Cu@FAU	240	11.5	89.5	N	5
Cu-ZnOx/13X	290	~10	~83	N	6
CuO-ZnOx@H-S-1	250	~3	85	N	7
CuO-ZnO@Cu-MOR	240	~21	~97	N	8
mhZSM-5-AlCu	280	7.3	~24	20	9
CZ@Z	260	20.8	19.4	62.2	10
CZ(0.9)@S-48	250	21.5	10.5	72.5	This work

^{*}N represents no DME formation reported.

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