

## SUPPORTING INFORMATION

### **Iodine oxidation-driven enhancement in preconcentration and ICP-MS/MS detection of $^{129}\text{I}$ in environmental water**

Weixian Wang <sup>a</sup>, Hui Xu <sup>a</sup>, Shan Xing <sup>b</sup>, Yu Wang <sup>a\*</sup>, Bingyang Meng <sup>a</sup>, Xiang Zhou <sup>a</sup>, Jia Zhan <sup>a</sup>, He Zhao <sup>a</sup>, Haisong Tang <sup>a</sup>, Nana Chai <sup>a</sup>

<sup>a</sup> Northwest Institute of Nuclear Technology, Xi'an, 710024, China

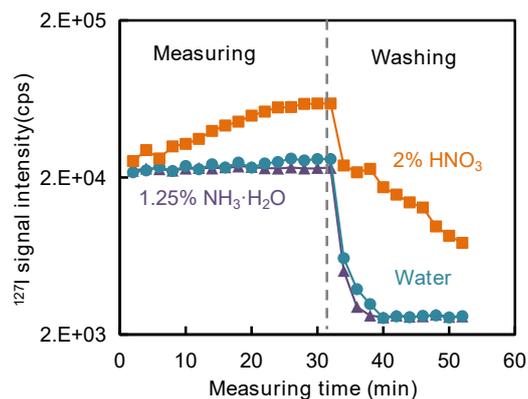
<sup>b</sup> School of Nuclear Science and Technology, Lanzhou University, Lanzhou 730000, China

**Supporting information includes 6 figures and 1 table.**

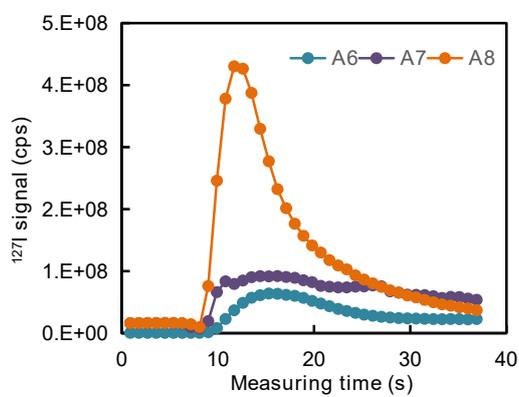
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\* Corresponding author: Northwest Institute of Nuclear Technology, Xi'an, 710024, China.

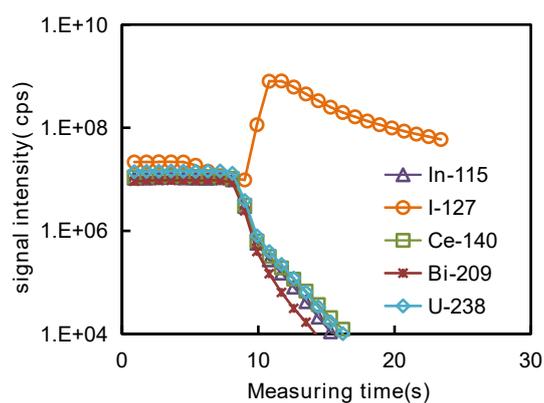
E-mail address: yuwang\_20012011@sina.com (Y. Wang).



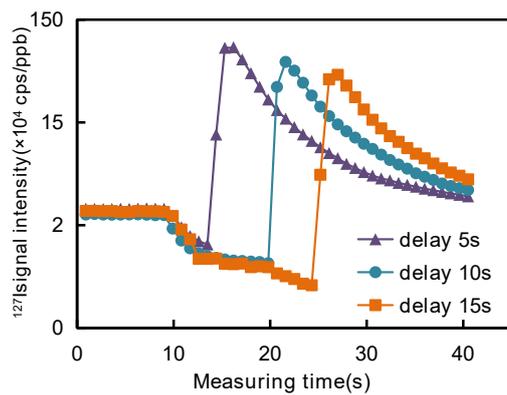
**Fig. S1.** Measuring of  $1 \text{ ng mL}^{-1} {}^{127}\text{I}$  solution and washing with 2%  $\text{HNO}_3$ , water, 1.25%  $\text{NH}_3 \cdot \text{H}_2\text{O}$ .



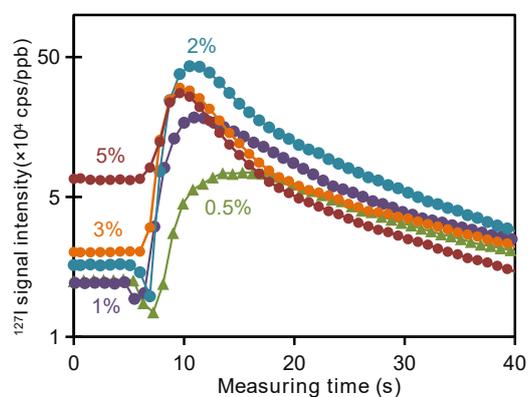
**Fig. S2.** A6: rising  $\text{I}^-$  at water/5% $\text{NH}_3 \cdot \text{H}_2\text{O}$ ; washing with 0.5%  $\text{NaNO}_2$ ; A7: rising  $\text{I}^-$  at 2%  $\text{HNO}_3$ ; washing with 0.5%  $\text{NaNO}_2$ +5%  $\text{NH}_3 \cdot \text{H}_2\text{O}$ ; A8: rising  $\text{I}^-$  at 2% $\text{HNO}_3$ ; washing with 0.5%  $\text{NaNO}_2$ .



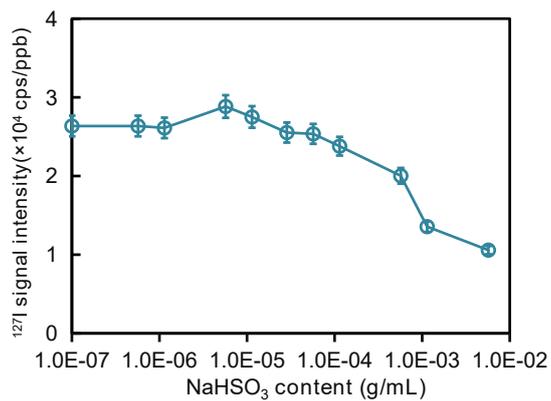
**Fig. S3.** Signals of  $100 \text{ ng mL}^{-1}$  In-115, I-127, Ce-140, Bi-209, U-238 standard solution, then washing with  $5 \times 10^{-4} \text{ g mL}^{-1} \text{NaNO}_2$ .



**Fig. S4.** Effect of delay entry into the oxidant solution on signal jumping.



**Fig. S5.** Effect of different concentration of  $\text{HNO}_3$  in rising solution,  $[\text{}^{127}\text{I}] = 1.0 \mu\text{g mL}^{-1}$ ,  $V_{\text{initial}} = 0.1 \text{ mL}$ , washing with  $5 \times 10^{-4} \text{ g mL}^{-1} \text{ NaNO}_2$ .



**Fig. S6.** The affection of  $\text{NaHSO}_3$  contents on the  $^{127}\text{I}$  signal intensity,  $[\text{}^{127}\text{I}] = 10 \text{ ng mL}^{-1}$ .

1 **Table S1** Stability of the CVG measurement at different batch for  $^{127}\text{I}$  signal intensity (cps (ng  
2  $\text{mL}^{-1})^{-1}$ )

Batches	Direct measure	CVG measure	Jump times
1	$1.67 \times 10^4$	$4.72 \times 10^5$	28.4
2	$1.44 \times 10^4$	$3.93 \times 10^5$	31.2
3	$1.57 \times 10^4$	$3.98 \times 10^5$	25.3
4	$1.63 \times 10^4$	$4.70 \times 10^5$	28.9
5	$1.50 \times 10^4$	$4.36 \times 10^5$	30.3
6	$1.62 \times 10^4$	$4.30 \times 10^5$	26.6

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