

Supplementary Information

Table S1. Summary of previous applications of CH₃F as a cell gas.

Analyte	Interference	Sample Matrix	Cell Reactions	Reference
⁸⁷ Sr/ ⁸⁶ Sr ratio	⁸⁷ Rb	Seven materials with different Rb/Sr ratios: 1. Basalt (0.02) 2. Nephelinite (0.03) 3. Basalt (0.11) 4. Basalt (0.55) 5. Basalt (0.14) 6. Diorite (0.28) 7. Rhyolite (0.67)	$^{86,87,88}\text{Sr}^+ + \text{CH}_3\text{F} \rightarrow ^{86,87,88}\text{SrF}^+ + \text{CH}_3$	1
⁸⁹ Sr	⁸⁷ Rb	Dissolute magmatic rocks	$\text{M}^+ + \text{CH}_3\text{F} \rightarrow \text{MF}^+ + \text{CH}_3$	2
⁷⁵ As, ^{77,78,79} Se	⁷⁵ Se, ⁷⁵ Ge, ⁷⁵ Rb, ⁷⁷ Br, ⁷⁷ Mo, ⁷⁷ Ag, ⁷⁸ Kr, ⁷⁸ Mo, ⁷⁸ Rb, ⁷⁸ Ag, ⁷⁹ Br, ⁷⁹ Kr, ⁷⁹ Rb, ⁷⁹ Mo	Synthetic matrices containing elements giving rise to ions interfering either at the original <i>m/z</i> ratios or those of the reaction products	$^{75}\text{As}^+ + \text{CH}_3\text{F} \rightarrow ^{75}\text{AsCH}_2^+ + \text{HF}$ $^{77,78,79}\text{Se} + \text{CH}_3\text{F} \rightarrow ^{77,78,79}\text{SeCH}_2^+ + \text{HF}$	3
⁷⁵ As and ^{77,78} Se	⁴⁰ Ar ³⁵ Cl ⁺ , ⁵⁹ Co ¹⁶ O ⁺ , ³⁶ Ar ³⁸ ArH ⁺ , ³⁸ Ar ³⁷ Cl ⁺ , ³⁶ Ar ³⁹ K ⁺ , ⁴³ Ca ¹⁶ O ₂ ⁺ , ⁴⁰ Ar ²³ Na ¹² C ⁺ , ¹² C ³¹ P ¹⁶ O ₂ ⁺ , ⁴⁰ Ca ³⁵ Cl ⁺ , ³⁸ Ar ³⁷ Cl ⁺ , ⁴⁰ Ca ³⁷ Cl ⁺ , ³⁶ Ar ⁴⁰ ArH ⁺ , ³⁸ Ar ₂ H ⁺ , ¹² C ¹⁹ F ¹⁴ N ¹⁶ O ₂ ⁺ , ⁷⁸ K ⁺ , ⁴⁰ Ar ⁴⁰ Ca ⁺ , ⁶⁴ Ni ¹⁶ O ⁺ , ⁶⁴ Zn ¹⁶ O ⁺ , ³² S ₂ ¹⁶ O ⁺ , ³² S ¹⁶ O ₃ ⁺ , ⁴⁰ Ar ⁴⁰ Ar ⁺ , and ⁴⁰ Ca ⁴⁰ Ca ⁺	Certified reference materials (pine needles, tomato leaves, rice flour, tuna fish tissue, fish protein, plankton, estuarine sediment, oyster tissue, lobster hepatopancreas, and Seronorm TM)	$\text{M}^+ + \text{CH}_3\text{F} \rightarrow \text{MCH}_2^+ + \text{HF}$ $\text{M}^+ + \text{CH}_3\text{F} \rightarrow \text{MF}^+ + \text{CH}_3$ $\text{M}^+ + \text{CH}_3\text{F} \rightarrow \text{MCHF}^+ + \text{H}_2$ $\text{MF}^+ + \text{CH}_3\text{F} \rightarrow \text{MF}_2^+ + \text{CH}_3$ $\text{MCH}_2^+ + \text{CH}_3\text{F} \rightarrow \text{MCH}_2^+(\text{CH}_3\text{F})$	4
La, Ce, Pr, Nd, Sm, Eu, Gd, Tb, Dy, Ho, Er, Tm, Yb, Lu	Lanthanide tailing and hydrides	Vaporised solution containing lanthanide salt	$\text{Ln}^+ + \text{CH}_3\text{F} \rightarrow \text{LnF}^+ + \text{CH}_3$ $\text{Ln}^+ + \text{CH}_3\text{F} \rightarrow \text{Ln}^+(\text{CH}_3\text{F})$ (only for Ln = Dy, Ho, Er, Tm, Yb, Lu)	5
Al, Co, Cr, Mn, Ni, Ti, V	C, Ca, Cl, Fe, K, Mg, Na, P, and S	Blood serum and urine	$\text{M}^+ + \text{CH}_3\text{F} \rightarrow \text{MF}_a(\text{CH}_3\text{F})_b^+$	6

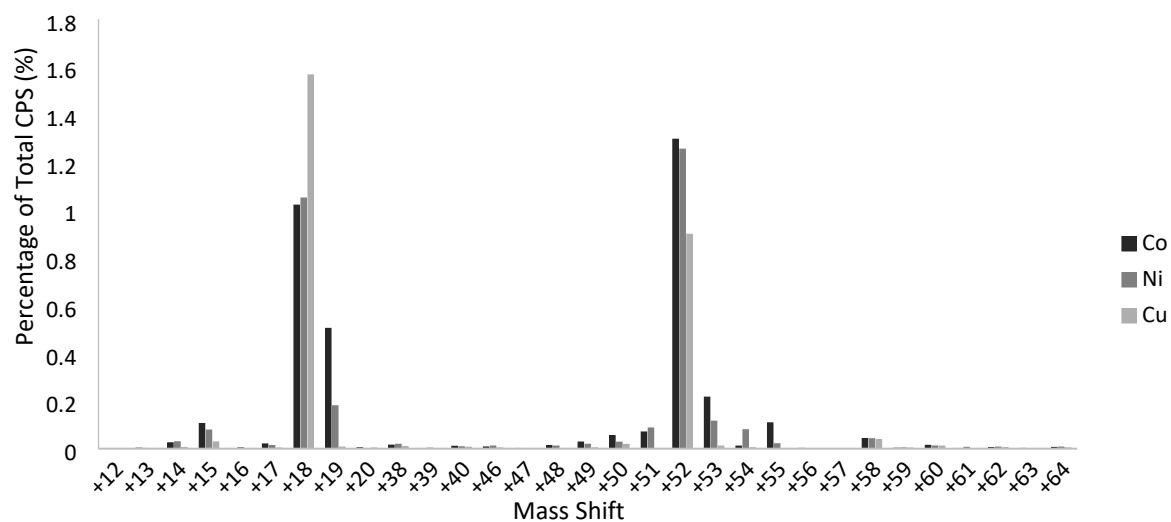


Figure S1. Product ion scan for Co, Ni and Cu using CH_3F with mass shifts from +12 to +64.

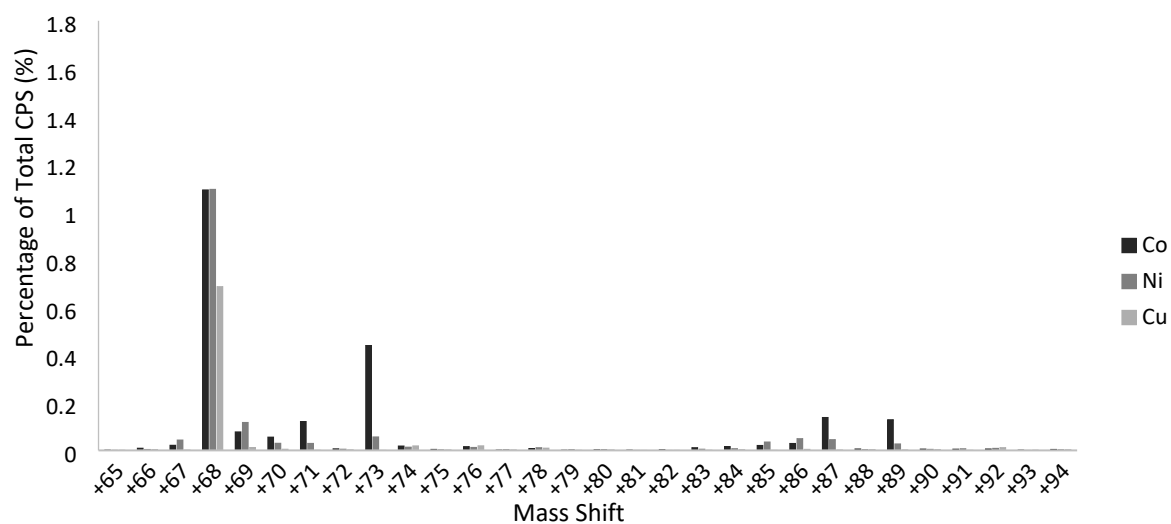


Figure S2. Product ion scan for Co, Ni and Cu using CH_3F with mass shifts from +65 to +94.

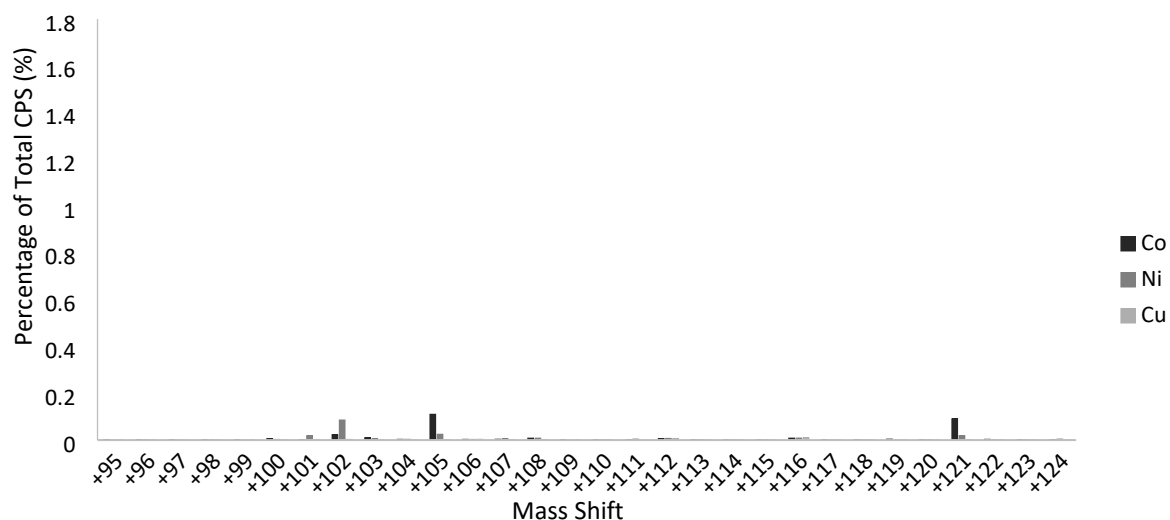


Figure S3. Product ion scan for Co, Ni and Cu using CH₃F with mass shifts from +95 to +124.

Table S2. Summary of the mass shifts measured for each actinide tested.

Actinide	Mass Shifts
²³² Th	On Mass, +19, +34, and +38
²³⁷ Np	On Mass, +1 to +38
²³⁸ U	On Mass, +19, and +34
²³⁹ Pu	On Mass, +1 to +36
²⁴¹ Am	On Mass, +1 to +34

Table S3. Possible separation schemes for actinides, including their optimal CH₃F gas flow rate, cell reaction(s) and mass shift.

Separation Scheme	CH ₃ F Gas Flow Rate (mL/min)	Mass Shift	Cell Reaction
²³⁷ Np / ²³⁹ Pu + ²⁴¹ Am	5	+16	$M^+ + CH_3F \rightarrow MCH_2^+ + HF$ $MCH_2^+ + CH_3F \rightarrow MCH_3 + CH_2F^+$ $MCH_3 + CH_3F \rightarrow MCH_4^+ + CH_2F^-$
²³⁷ Np / ²³⁹ Pu + ²⁴¹ Am	2	+19	$M^+ + CH_3F \rightarrow MF^+ + CH_3$
²³² Th / ²³⁷ Np	6	+38	$M^+ + CH_3F \rightarrow MF^+ + CH_3$ $MF^+ + CH_3F \rightarrow MF_2^+ + CH_3$

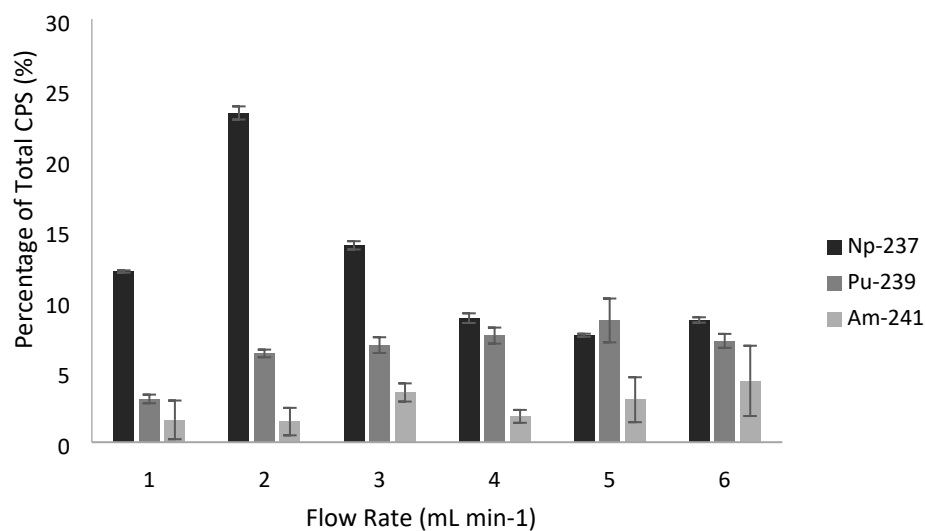


Figure S4. Graph showing the percentage of the total signal for ^{237}Np , ^{239}Pu , and ^{241}Am at mass shift +19 as CH_3F gas flow rate increases.

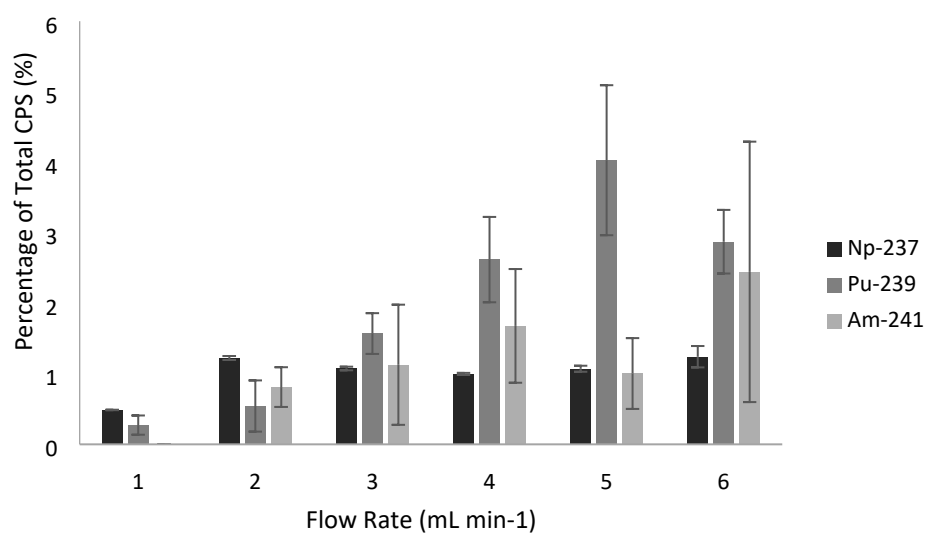


Figure S5. Graph showing the percentage of the total signal for ^{237}Np , ^{239}Pu , and ^{241}Am at mass shift +16 as CH_3F gas flow rate increases.

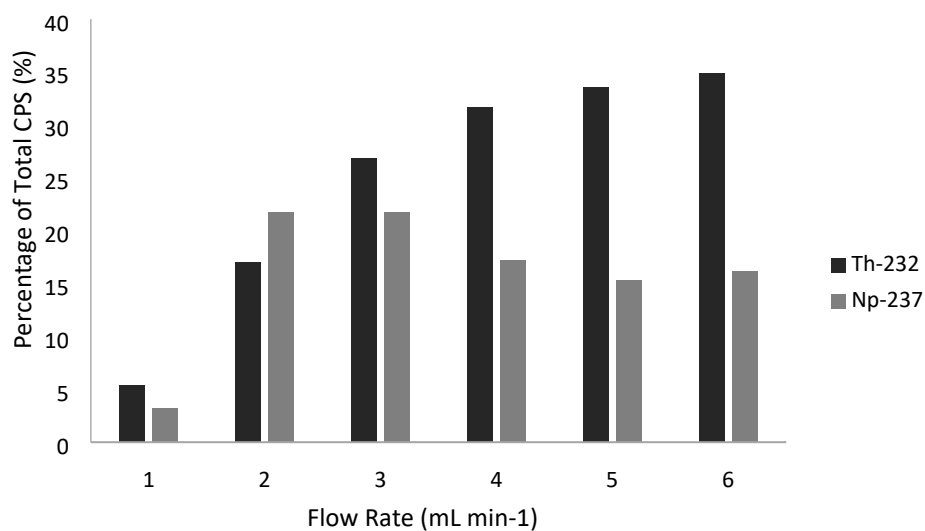


Figure S6. Graph showing the percentage of the total signal for ²³²Th and ²³⁷Np at mass shift +38 as CH₃F gas flow rate increases.

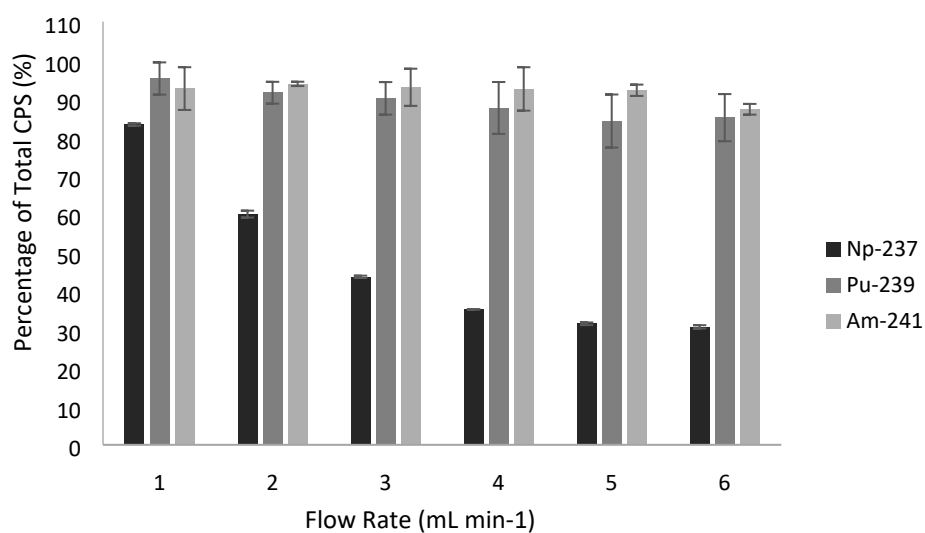


Figure S7. Graph showing the percentage of the total signal for ²³⁷Np, ²³⁹Pu, and ²⁴¹Am remaining on mass as CH₃F gas flow rate increases.

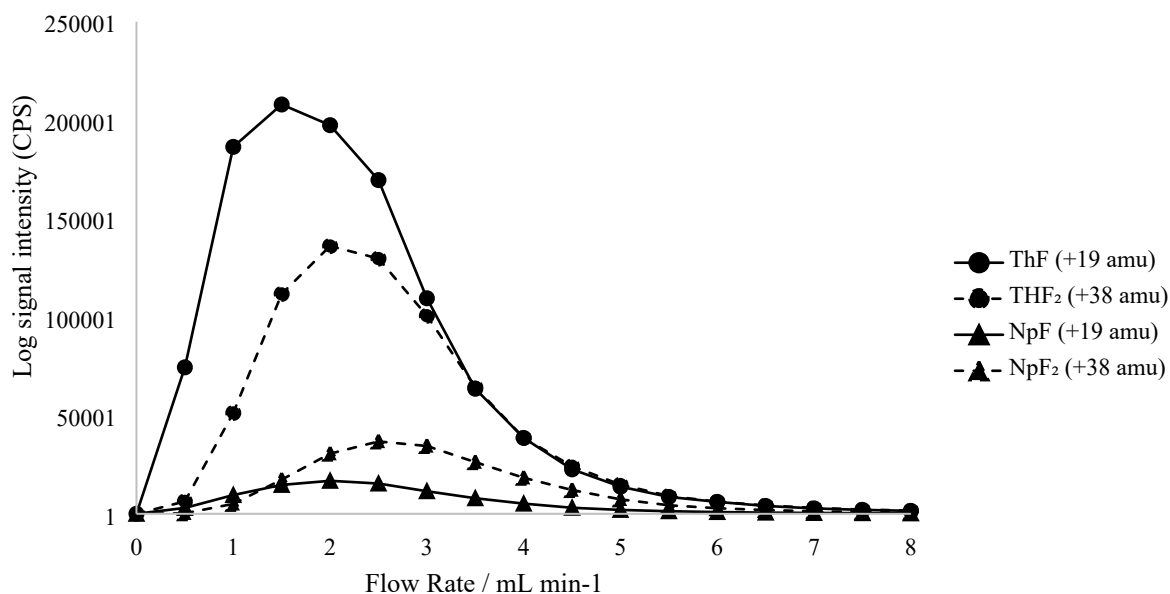


Figure S8. Variation of sensitivity for the formation of the MF (+19 amu) and MF₂ (+38 amu) product ions for ²³²Th and ²³⁷Np with flow rate of CH₃F cell gas.

References

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