

Supplementary data for:

***Development of Adsorption-Desorption Technique of Rare-Earth
Metal Ions using Phase Transition Behavior of Zwitterionic
Polymer Brushes***

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Equations:

Equation S1:

Polymer mass per gram silica:

$$W = \frac{5.332 \text{ mg polymer (from TG)}}{7.958 \text{ mg silica (from TG)}} = 0.67 \text{ g/g}$$

Equation S2:

Number of chains per gram silica:

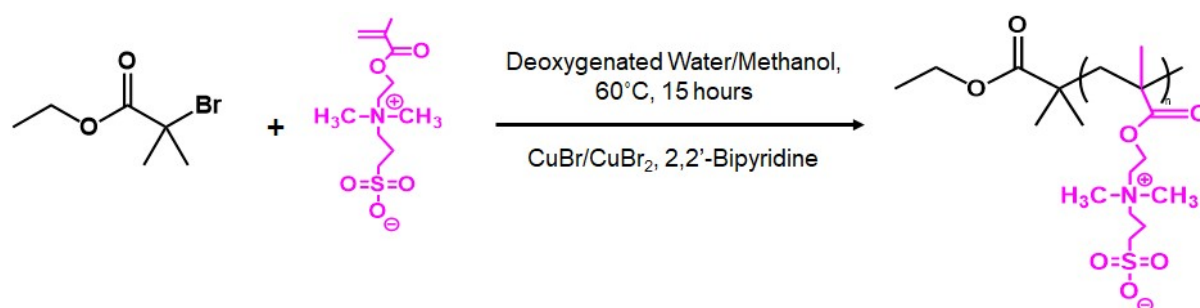
$$N_{\text{chains/g}} = \frac{W}{M_n} \times N_A = \frac{0.67}{13160} \times 6.022 \times 10^{23} = 3.066 \times 10^{19} \text{ chains/g}$$

Equation S3:

Thus, the density would be:

$$\sigma = \frac{\frac{N_{\text{chains}}}{g}}{S} = \frac{3.066 \times \frac{10^{19} \text{ chains}}{g}}{\frac{200 \text{ m}^2}{g}} = 0.1533 \times \frac{10^{18} \text{ chains}}{\text{m}^2} = 0.1533 \text{ chains/nm}^2$$

Scheme:



Scheme S1 Schematic illustrations of synthesis procedure of bulk poly(DMAPS).

Figures:

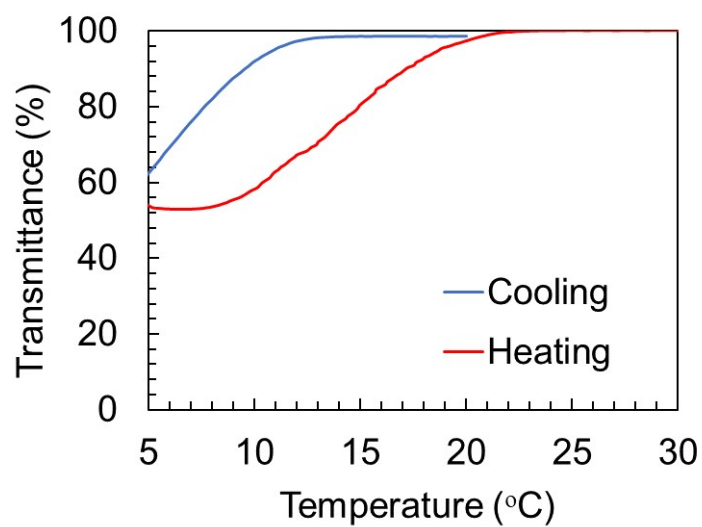


Figure S1. Temperature dependence of optical-transmittance plots for a poly(DMAPS)-containing aqueous solution under cooling and heating processes.

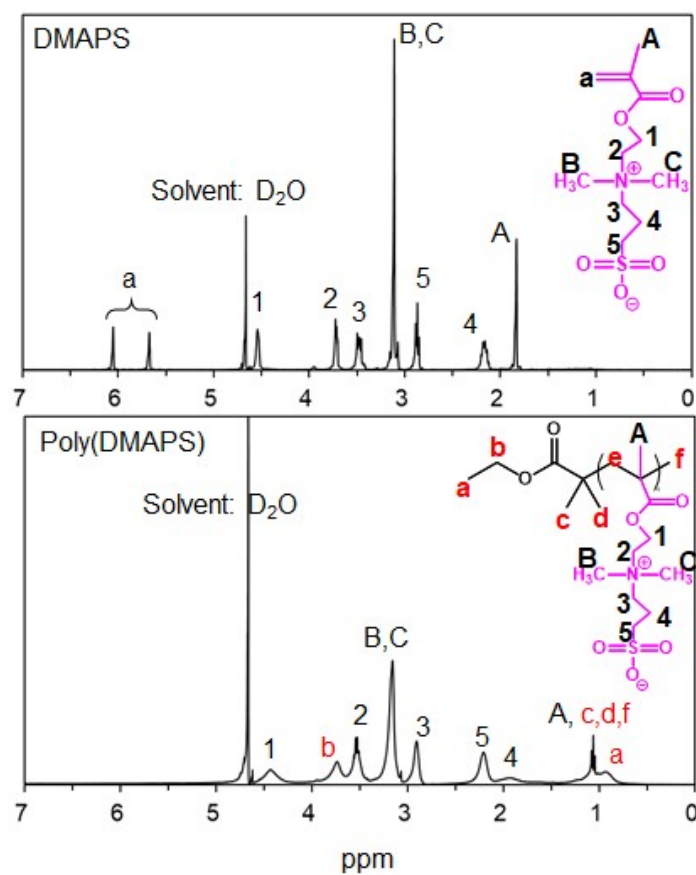


Figure S2. ¹H-NMR spectra of DMAPS monomer and the synthesized bulk poly(DMAPS).

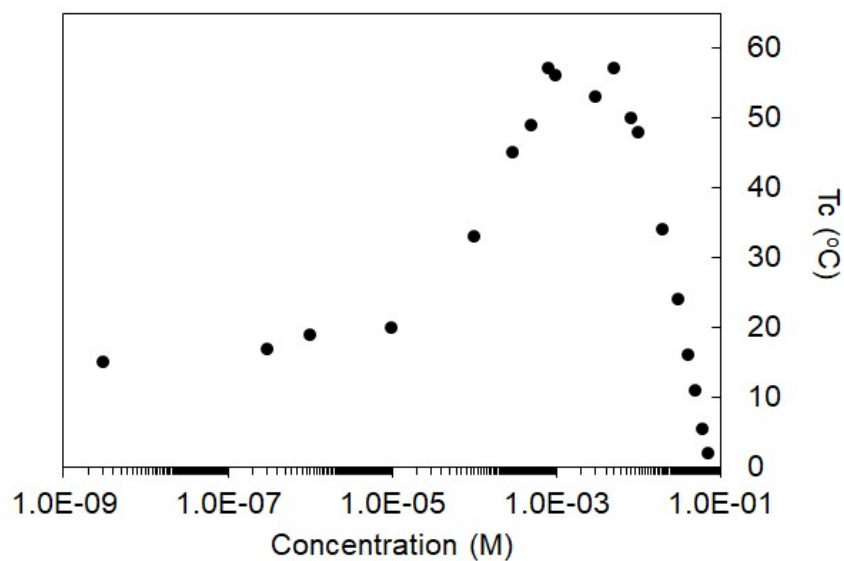


Figure S3. La(III) concentration dependence of T_c of poly(DMAPS) in aqueous solutions containing La ions.

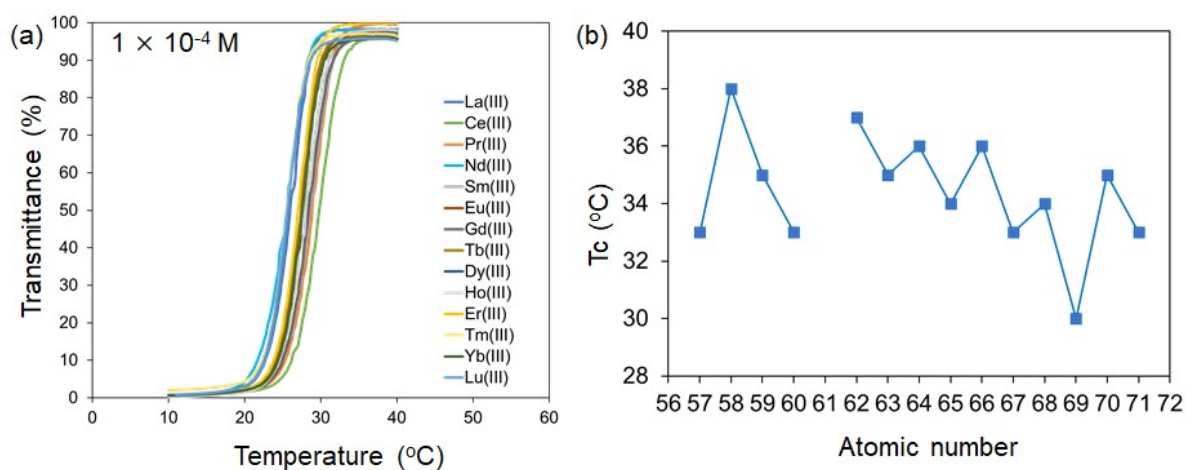


Figure S4. (a) Temperature dependence of optical-transmittance plots for the poly(DMAPS) in all Ln ions (1×10^{-4} M) and (b) the Ln atomic number dependence of the UCST.

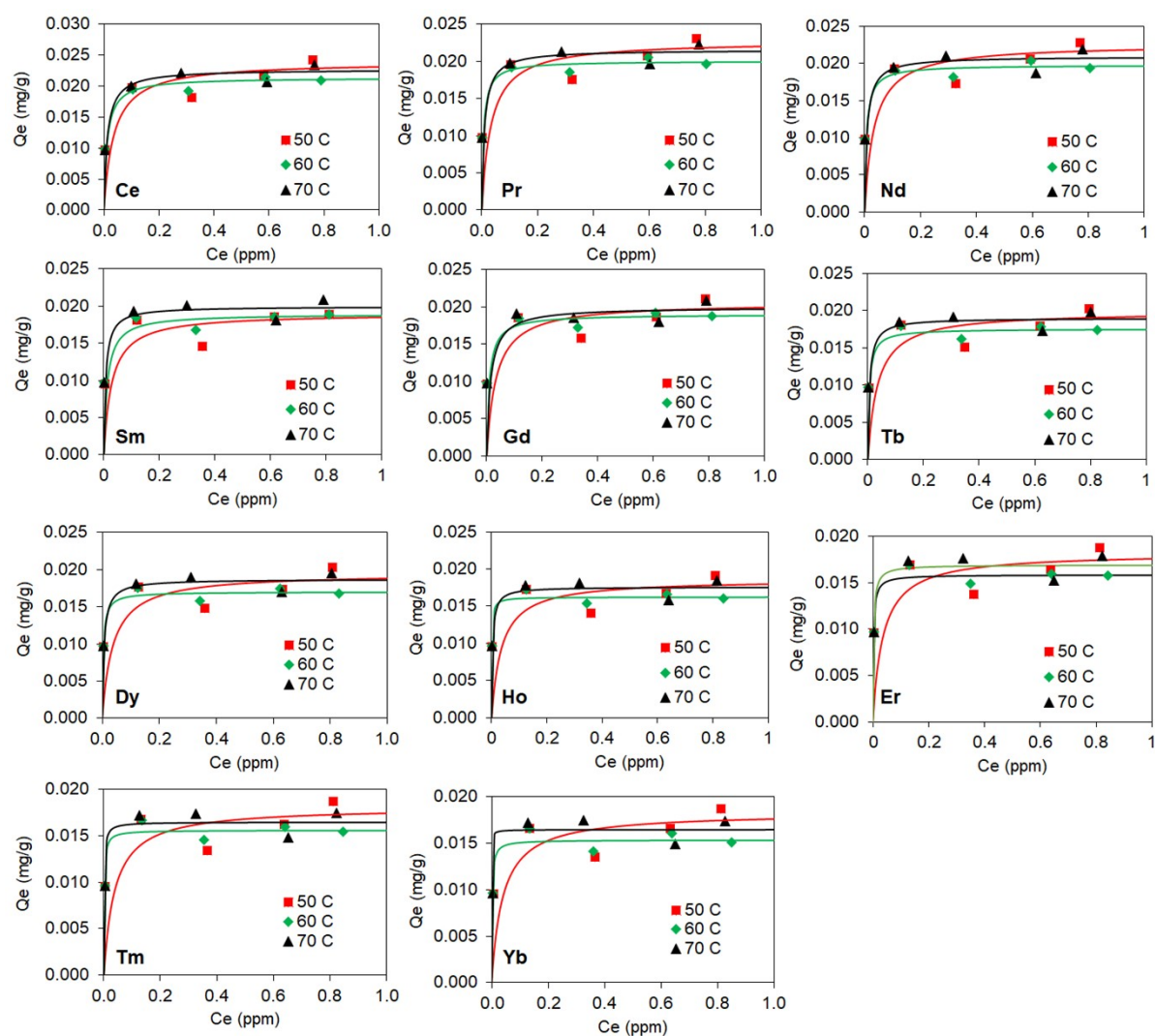


Figure S5. Effects of initial concentrations on all Ln(III) except La, Eu, and Lu at 50 – 70 °C, respectively.

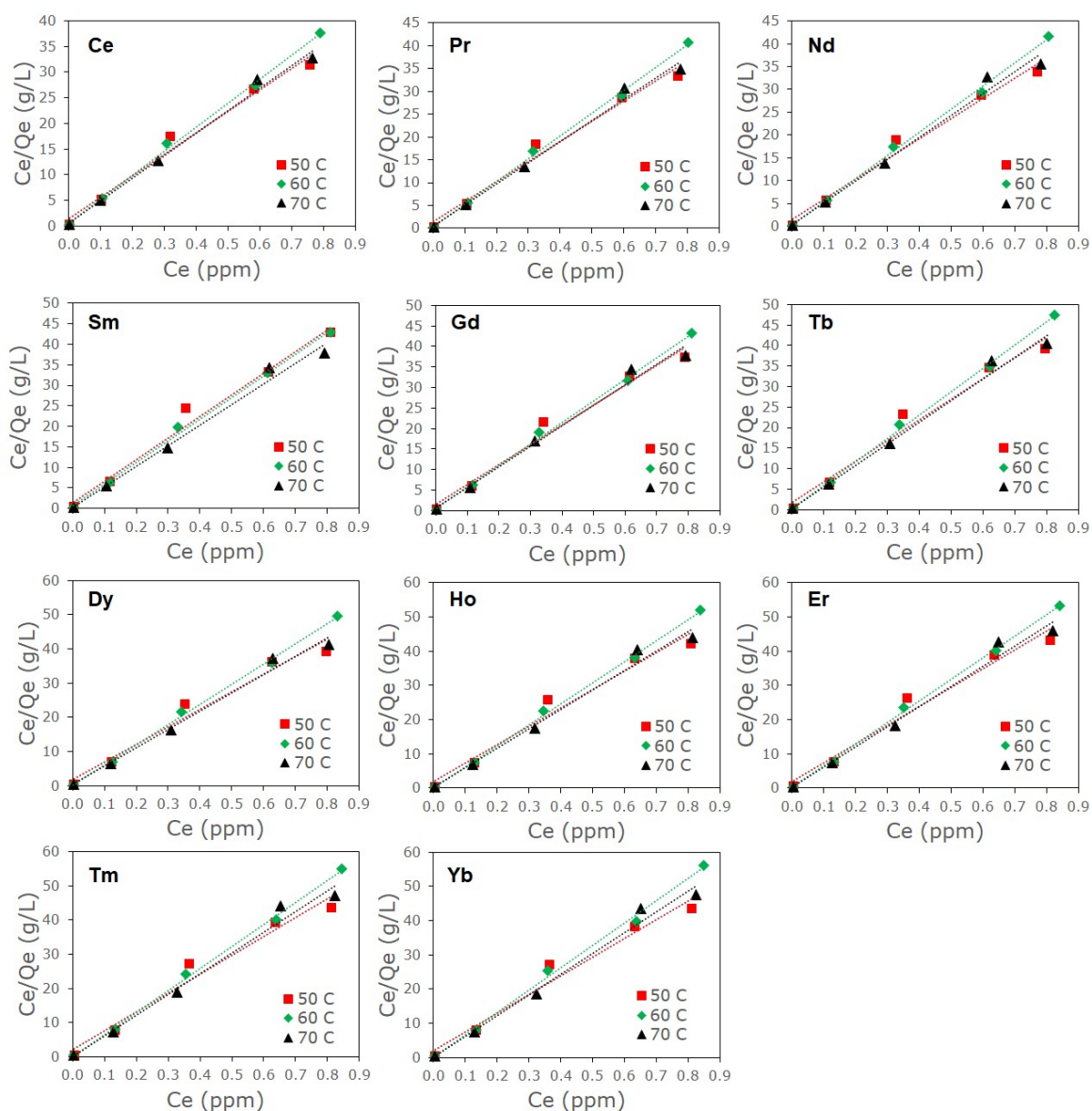


Figure S6. Linear regression analysis based on Langmuir adsorption isotherm model fittings for all Ln(III) except La, Eu, and Lu at 50 – 70 °C.

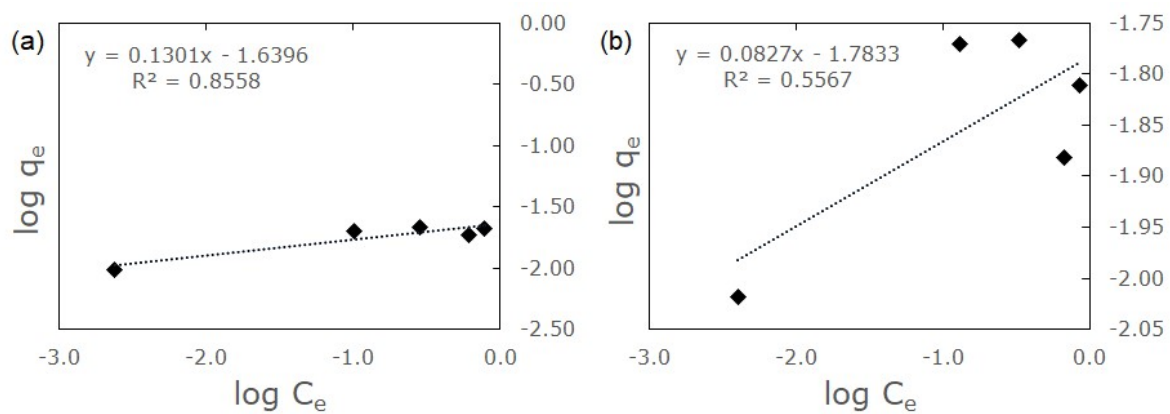


Figure S7. Freundlich isotherm model fittings for (a) La and (b) Lu ions, respectively.

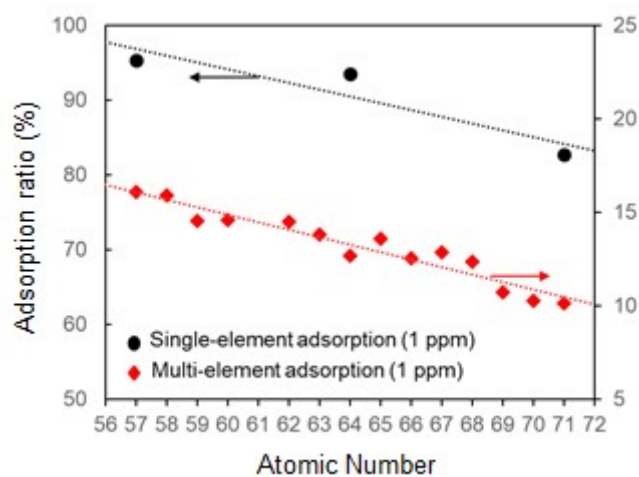


Figure S8. Adsorption ratios of Ln ions in single- and multi-elemental solutions.

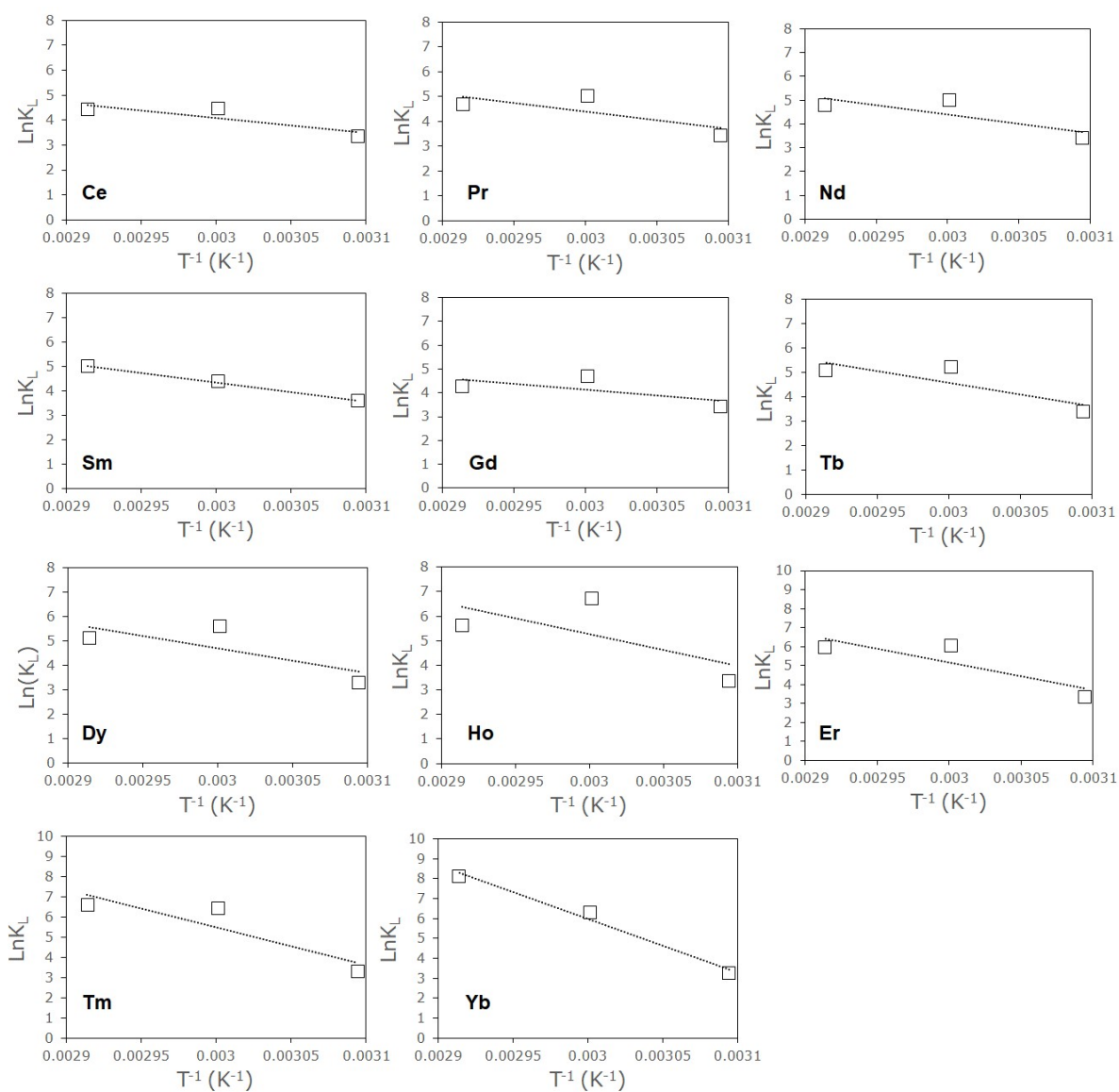


Figure S9. The plots of $\text{Ln}K_L$ vs. $1/T$ (van't Hoff plot) of all Ln(III) except La, Eu, and Lu.

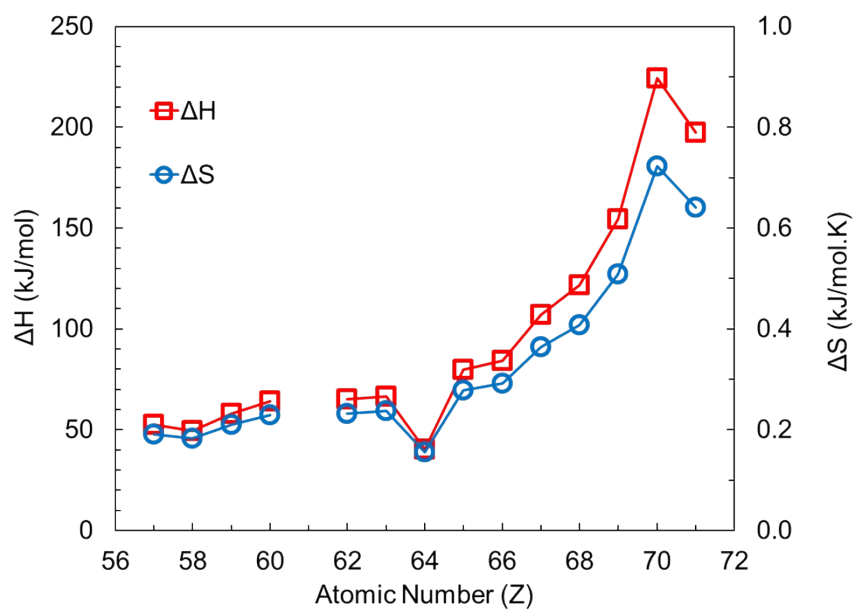


Figure S10. Atomic number dependence of standard enthalpy change (ΔH°) and standard entropy change (ΔS°) for all Ln ions.