

## Synthesis and Study of Donor-Acceptor Conjugated Polymers based on Isoindigo units *via* Metal Free Aldol Polymerization Strategy

Prerak R. Patel,<sup>‡</sup> Mayur J. Patel, <sup>#</sup> Parameswar K. Iyer,<sup>†#\*</sup> Sanjio S. Zade<sup>‡\*</sup> and Arun L. Patel<sup>‡\*</sup>

<sup>‡</sup>*Department of Chemistry, Faculty of Science, The Maharaja Sayajirao University of Baroda, Vadodra-390002, Gujarat, India. E-mail: [arunpatel\\_5376@yahoo.co.in](mailto:arunpatel_5376@yahoo.co.in)*

<sup>†</sup>*Centre of Nanotechnology, Indian Institute of Technology Guwahati, Guwahati, Assam-781039, India.*

<sup>#</sup>*Department of Chemistry, Indian Institute of Technology Guwahati, Guwahati, Assam-781039, India. E-mail: [pki@iitg.ac.in](mailto:pki@iitg.ac.in)*

<sup>‡</sup>*Department of Chemical Sciences, Indian Institute of Science Education and Research (IISER) Kolkata, Mohanpur-741246, India. E-mail: [sanjiozade@iiserkol.ac.in](mailto:sanjiozade@iiserkol.ac.in)*

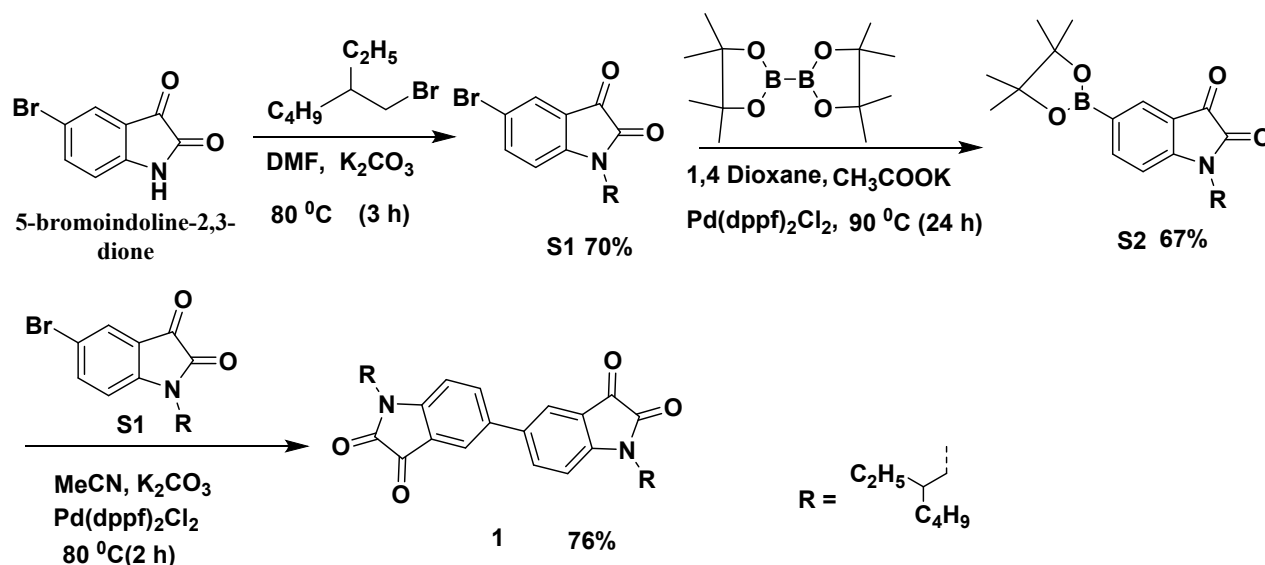
## CONTENTS

Sr No	Contents	Page No
1	Materials and Instrumentation	S2
2	Synthesis of 1,1'-bis(2-ethylhexyl)-[5,5'-biindoline]-2,2',3,3'-tetraone (1)	S3-S4
3	Synthesis of 5,5'-([2,2'-bithiophene]-5,5'-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (3)	S5
4	Synthesis of 2,7-dibromo-9-(2-ethylhexyl)-9H-carbazole (S5)	S6
5	Synthesis of 5,5'-(9-(2-ethylhexyl)-9H-carbazole-2,7-diyl) bis(1-(2-ethylhexyl) indoline-2,3-dione) (5)	S7-S8
6	Synthesis of (E)-5,5'-Dibromo-1,1'-bis(2-hexyldecyl)-[3,3'-biindolinylidene]-2,2'-dione (9)	S8-S9
7	Spectral data of intermediate compounds and monomers	S10-S27
8	Spectral data of polymers	S28-S30
9	References	S31

## Materials and Instrumentation

All the chemical were reagent grade and used as it was purchased. Moisture-sensitive reaction were carried out using dry solvents in an inert environment of dry nitrogen. Reaction progress was monitored by thin-layer chromatography (TLC) using Merck 60 F254 aluminum-coated plates. Synthesized compounds were purified by column chromatography using Silica gel (60-120 mesh and 100-200 mesh). NMR spectra were recorded on a Bruker Avance-III 400 spectrometer, NMR in  $\text{CDCl}_3$  and  $\text{DMSO-D}_6$ . High resolution mass spectra (HRMS) were recorded on Xevo G2-XS QTOF Mass Spectrometer.

## Synthesis of 1,1'-Bis(2-ethylhexyl)-[5,5'-biindoline]-2,2',3,3'-tetraone (**1**)



**Scheme S1** Synthesis of 1,1'-bis(2-ethylhexyl)-[5,5'-biindoline]-2,2',3,3'-tetraone (**1**) from 5-bromoindoline-2,3-dione.

Compound **1** was synthesized according to the modified literature procedure.<sup>1,2</sup>

**5-Bromo-1-(2-ethylhexyl)indoline-2,3-dione (S1)**: Under nitrogen atmosphere, 5-bromoisatin (2.0g, 8.84 mmol) and anhydrous potassium carbonate (5.492g, 39.8 mmol) were dissolved in dry DMF (18 mL) and heated for 10 minutes at 50 °C. After adding ethylhexyl bromide (2.73 mL, 15.4 mmol), the reaction mixture was stirred at 80 °C for 4 hours. The reaction mixture was poured into 180 mL of H<sub>2</sub>O and then extracted with ethyl acetate. The organic fraction was washed with brine and water and dried over Na<sub>2</sub>SO<sub>4</sub>. A crude product was obtained as a dark crimson oil, which was further purified by column chromatography using 5% ethyl acetate in pet ether.

**5-Bromo-1-(2-ethylhexyl)indoline-2,3-dione (S1)**: Orange red solid (2.45g, 70%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.69-7.71 (dd,  $J_1 = 7.2$  Hz,  $J_2 = 2$  Hz, 2H), 6.80-6.82 (dd, 1H,  $J_1 = 8$  Hz,  $J_2 = 1.6$  Hz), 3.56-3.62 (t, 2H), 1.72-1.81 (s, 1H), 1.28-1.43 (m, 8H), 0.90-0.96 (m, 6H). IR (KBr, cm<sup>-1</sup>): 2959, 2927, 2864, 1732, 1605, 1471, 1467, 1438, 1326, 1182, 825, 711, 468.

**1-(2-Ethylhexyl)-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)indoline-2,3-dione (S2)**: In dry two-necked round bottom flask, a mixture of 1,4 dioxane (20 mL), compound **S1** (1.0g, 2.9 mmol), bis(pinacolato)diborane (1.12g, 4.4 mmol), and potassium acetate (0.580g, 5.9 mmol) was taken. After degassing for 30 minutes with sparging nitrogen, catalyst, Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (0.103g, 0.14mmol) was added and the reaction mixture was heated at 100 °C for 24 hours under a nitrogen atmosphere. The reaction mixture was then poured into 200 mL of H<sub>2</sub>O and

the organic layer was extracted with ethyl acetate. The organic layer was washed with H<sub>2</sub>O and brine, dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, and concentrated to obtain red oil as the crude product. The crude product was further purified by column chromatography using 20% ethyl acetate in pet ether to get red oil.

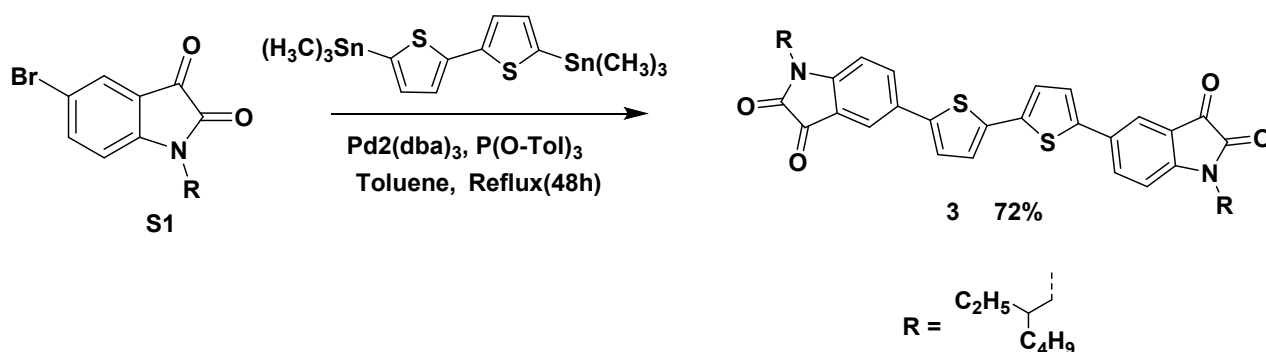
1-(2-Ethylhexyl)-5-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)indoline-2,3-dione (**S2**): Orange red liquid (0.81g, 67%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.06-8.06 (d, J = 0.8 Hz, 1H), 8.01-8.03 (dd, J<sub>1</sub> = 8 Hz, J<sub>2</sub> = 1.2 Hz, 1H), 6.87-6.89 (d, J = 8 Hz, 1H), 3.61-3.64 (m, 2H), 1.78 (s, 1H), 1.35 (s, 12H), 1.27-1.33 (m, 8H) 0.87-0.95 (m, 6H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 183.34, 158.71, 153.53, 145.05, 131.88, 117.10, 109.73, 84.27, 44.40, 37.33, 30.52, 28.56, 25.02, 23.89, 22.99, 14.06, 10.55.

1,1'-Bis(2-ethylhexyl)-[5,5'-biindoline]-2,2',3,3'-tetraone (**1**)

Compound **S2** (0.860g, 2.5 mmol) and compound **S1** (1.10g, 2.88mmol) were dissolved in 45 mL acetonitrile and the reaction mixture was stirred for 5 minutes. 2 M K<sub>2</sub>CO<sub>3</sub> (0.84g, 6.1 mmol) solution was added to the reaction mixture, followed by 30 minutes of nitrogen purging. The catalyst, [Pd(dppf)<sub>2</sub>Cl<sub>2</sub>] (0.10g, 0.147 mmol) was added and the reaction mixture was heated at reflux temperature for 6 hours. After the completion of the reaction, 200 mL of water was added and extracted with ethyl acetate. The organic layers are washed with saturated brine solution, then with water, and finally dried by adding anhydrous Na<sub>2</sub>SO<sub>4</sub>. After solvent evaporation, the crude product is further purified by column chromatography over silica gel using 30% ethyl acetate in petroleum ether as eluent.

1,1'-Bis(2-ethylhexyl)-[5,5'-biindoline]-2,2',3,3'-tetraone (**1**): Brown red solid (1.0g, 76%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.75-7.78 (m, 4H), 6.99-7.01 (d, J<sub>1</sub> = 8 Hz, 2H), 3.65-3.68 (m, 4H), 1.84-1.87 (m, 2H), 1.26-1.46 (m, 16H), 0.92-0.99 (m, 12H). <sup>13</sup>C NMR (100MHz, CDCl<sub>3</sub>): δ 183.43, 158.43, 150.89, 136.10, 134.89, 123.17, 118.16, 111.05, 44.59, 37.44, 30.61, 28.81, 28.66, 24.83, 23.87, 23.00, 14.02, 10.55. HRMS(ES<sup>+</sup>): C<sub>32</sub>H<sub>41</sub>N<sub>2</sub>O<sub>4</sub> requires 517.3066, found 517.3075.

### Synthesis of 5,5'-([2,2'-bithiophene]-5,5'-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**3**)



**Scheme S2** Synthesis of 5,5'-([2,2'-bithiophene]-5,5'-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**3**) from compound **S1**

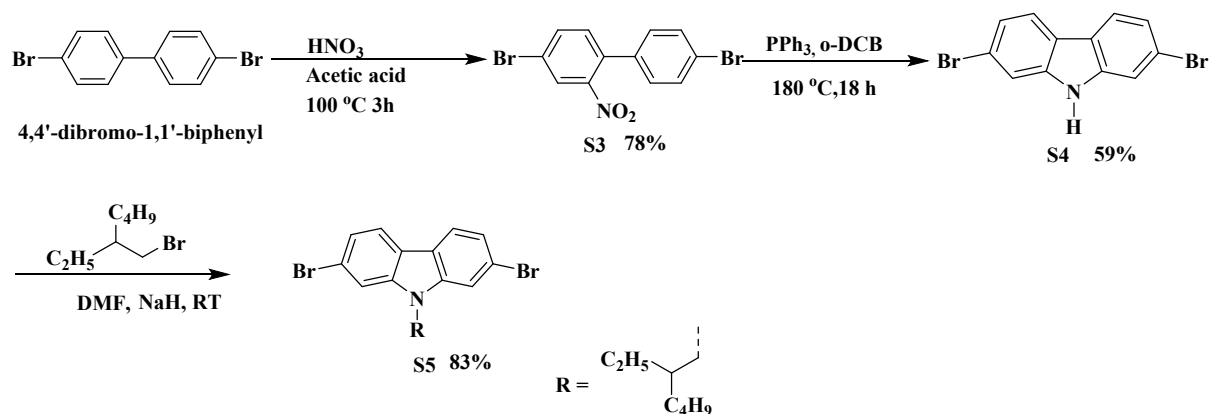
Compound **3** was synthesized according to the modified literature procedure.<sup>3</sup>

#### 5,5'-([2,2'-bithiophene]-5,5'-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**3**)

Compound **S1** (0.604g, 1.79 mmol) was taken in two neck round bottom flask and dissolved in 30 mL toluene. 5,5'-Bis(trimethylstannyl)-2,2'-bithiophene (0.400g, 0.814 mmol), Pd<sub>2</sub>(dba)<sub>3</sub> (0.074g, 0.081 mmol), and P(o-tol)<sub>3</sub> (0.318g, 0.651 mmol) were added into the reaction mixture under nitrogen atmosphere. The reaction mixture was stirred for 24 hours at 110 °C, cooled down to room temperature, water and dichloromethane were added. The organic extract was washed with water and then dried over with anhydrous Na<sub>2</sub>SO<sub>4</sub>. After removal of the solvent under reduced pressure, the crude product was purified by silica gel chromatography with eluent 40% DCM in petroleum ether as an eluent to give compound **3** as a black solid (0.600g, 72.2 %).

5,5'-([2,2'-Bithiophene]-5,5'-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**3**): Brown solid (0.600g, 72.2%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.80-7.84 (m, 4H), 7.18-7.22 (dd, J<sub>1</sub>= 12.8 Hz, J<sub>2</sub>= 3.6 Hz, 4H), 6.93-6.95 (d, J<sub>1</sub>= 8 Hz, 2H), 3.64-3.68 (m, 4H), 1.83-1.85 (m, 2H), 1.27-1.45 (m, 16H), 0.91-0.99 (m, 12H). <sup>13</sup>C NMR (100MHz, CDCl<sub>3</sub>): 183.45, 158.39, 150.34, 141.06, 136.70, 134.90, 130.04, 124.92, 124.13, 122.17, 118.05, 110.90, 44.58, 37.49, 30.64, 28.64, 24.00, 23.01, 14.04, 10.57. HRMS(ES<sup>+</sup>): C<sub>40</sub>H<sub>45</sub>N<sub>2</sub>O<sub>4</sub>S<sub>2</sub> requires 681.2821, found 681.2816. IR (KBr, cm<sup>-1</sup>): 3438, 2957, 2928, 2862, 1730, 1617, 1588, 1529, 1482, 1436, 1353, 1331, 1280, 1187, 1116, 1033, 832, 789, 715, 465.

### Synthesis of 2,7-dibromo-9-(2-ethylhexyl)-9*H*-carbazole (S5)



**Scheme S3** Synthesis of compound **S5** from 4,4'-dibromo-1,1'-biphenyl.

Compound **S5** was synthesized according to the modified literature procedure.<sup>4-6</sup>

#### Synthesis of 4,4'-dibromo-2-nitro-1,1'-biphenyl (**S3**)

In a 250 mL two necked round bottom flask, 4,4'-dibromo- biphenyl (4g, 12.82 mmol) in glacial AcOH (60 mL) was taken and heated at 100 °C. To this solution fuming HNO<sub>3</sub> (100%, 18.5 mL) and H<sub>2</sub>O (1.5 mL) was added. The reaction mixture was stirred at 100 °C for 30 minutes and then allowed to cool to room temperature. The resultant reaction mixture was poured into water and subsequently extracted with DCM. The organic layer was then washed successively with water and dried over Na<sub>2</sub>SO<sub>4</sub>. The compound **S3** was obtained by evaporation of solvent under reduced pressure.

**4,4'-Dibromo-2-nitro-1,1'-biphenyl (**S3**):** Yellow solid (3.60g, 78%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.09 (s, 1H), 7.88-7.90 (d, J = 8.4, 2H), 7.58–7.59 (d, J = 1.6 Hz, 2H), 7.37-7.39 (dd, J<sub>1</sub> = 8 Hz, J<sub>2</sub> = 1.6 Hz, 2H).

#### Synthesis of 2,7-dibromocarbazole (**S4**)

4,4'-Dibromo-2-nitro-1,1'-biphenyl (**S3**) (6g, 16.8 mmol) and PPh<sub>3</sub> (11g, 42.0 mmol) were dissolved in *o*-dichlorobenzene (35 mL) under nitrogen atmosphere and refluxed for 24 hours. After the completion of reaction, the solvent was evaporated and the crude product was purified by column chromatography over silica gel by using 20% ethyl acetate in petroleum ether.

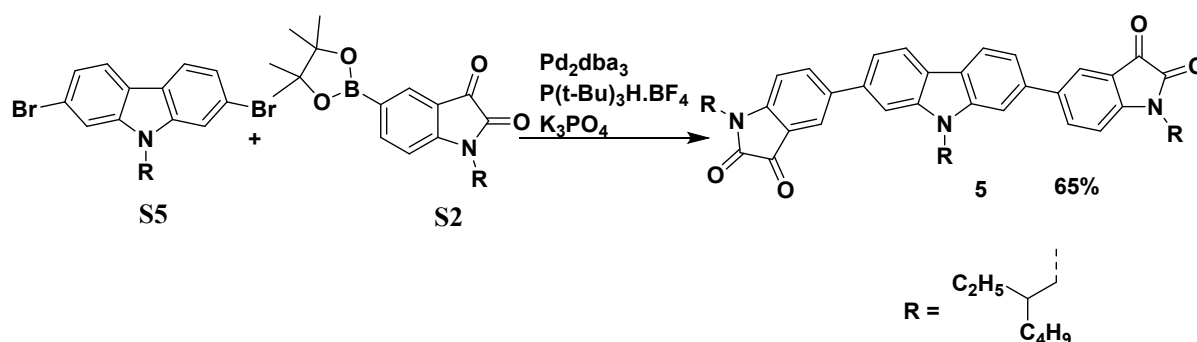
**2,7-Dibromocarbazole (**S4**):** White solid (3.25g, 59%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 8.05-8.06 (d, J = 2 Hz, 1H), 7.77- 7.80 (dd, J = 8.4 Hz, J = 2 Hz, 1H), 7.60 (s, 1H), 7.57-7.58 (d, J = 2 Hz, 1H), 7.31-7.33 (d, J = 8.4 Hz, 1H), 7.17-7.19 (dd, J = 1.6 Hz, J = 6.4 Hz, 2H).

### Synthesis of 2,7-dibromo-9-(2-ethylhexyl)-9*H*-carbazole (**S5**)

In a 100 mL two necked round bottom flask, compound **S4** (5g, 15.4 mmol) in DMF (50 mL) was taken. To this reaction mixture, NaH (60% w/w suspension in mineral oil) (865mg, 21.6 mmol) was added. After 1 hour, 2-ethylhexyl bromide (3.83g, 19.84 mmol) was added and the reaction mixture was stirred at room temperature for more 20 hours under nitrogen atmosphere. After the completion of the reaction, the reaction mixture was poured into water and the product was extracted in ethyl acetate. Organic fraction was washed with brine and dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>, followed by solvent distillation under reduced pressure. The resultant crude product was subjected to the column chromatography over silica gel by using 5% ethyl acetate in petroleum ether to get desired pure product.

2,7-Dibromo-9-(2-ethylhexyl)-9*H*-carbazole (**S5**): White solid (7.0g, 83%). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 7.90-7.92 (d, J = 8.4 Hz, 2H), 7.52 (s, 2H), 7.34-7.37 (dd, J<sub>1</sub> = 8.2 Hz, J<sub>2</sub> = 1.2 Hz, 2H), 4.06-4.08 (d, J = 6 Hz, 2H), 1.99-2.09 (m, 1H), 1.24–1.41 (m, 24H), 0.86–0.91 (m, 6H).

### Synthesis of 5,5'-(9-(2-ethylhexyl)-9*H*-carbazole-2,7-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**5**)



**Scheme S4** Synthesis of 5,5'-(9-(2-ethylhexyl)-9*H*-carbazole-2,7-diyl) bis(1-(2-ethylhexyl) indoline-2,3-dione) (**5**) from compound **S5**.

### 5,5'-(9-(2-Ethylhexyl)-9*H*-carbazole-2,7-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) (**5**)

Compound **S5** (0.500g, 1.14 mmol), compound **S2** (0.951g, 2.47 mmol) and Pd<sub>2</sub>(dba)<sub>3</sub> (47mg, 0.052 mmol), P(t-Bu)<sub>3</sub>.BF<sub>4</sub> (62mg, 0.216 mmol) were added into clean and dry two neck round bottom flask under nitrogen atmosphere. Then THF (40 ml) and potassium phosphate (1.7g, 2M) were added by injection in sequence. The reaction mixture was stirred at 80 °C for 24

hours. After cooling down, the mixture was poured into water, organic layer was extracted with dichloromethane, dried over anhydrous  $\text{Na}_2\text{SO}_4$ , and purified by silica gel column chromatography using 50% Ethyl acetate in petroleum ether.

5,5'-(9-(2-ethylhexyl)-9*H*-carbazole-2,7-diyl)bis(1-(2-ethylhexyl)indoline-2,3-dione) **(5)**: Dark red solid (0.60g, 65%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  8.16-8.18 (d,  $J$  = 8 Hz, 2H), 7.94-7.97 (m, 4H), 7.52 (s, 2H), 7.42-7.44 (dd,  $J$  = 8 Hz,  $J$  = 1.2 Hz, 2H), 7.01-7.03 (d,  $J$  = 8.0 Hz, 2H), 4.25-4.29 (m, 2H), 3.67-3.70 (m, 4H), 2.09 (s, 1H), 1.88– 1.89 (d, 2H), 1.27-1.46 (m, 32H), 0.88-1.01 (m, 22H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ ):  $\delta$  183.83, 158.64, 150.36, 142.05, 138.08, 137.08, 137.04, 124.11, 122.15, 120.96, 118.19, 188.11, 110.79, 106.99, 47.57, 44.53, 39.38, 37.49, 30.96, 30.62, 28.66, 24.86, 24.02, 23.02, 14.05, 11.03, 10.60. HRMS ( $\text{ES}^+$ ):  $\text{C}_{52}\text{H}_{64}\text{N}_3\text{O}_4$  requires 794.4897, found 794.4905. IR (KBr,  $\text{cm}^{-1}$ ): 3455, 3055, 2958, 2927, 2865, 1736, 1617, 1501, 1459, 1332, 1332, 1254, 1186, 1118.

Compound **9** was synthesized using the modified reported literature process.<sup>7,8</sup>

#### 5-Bromoindolin-2-one (**7**)

5-Bromoisatin (2.50 g, 11.0 mmol) was dissolved in methanol (20 mL). Following this, hydrazine monohydrate (1.5 mL, 31 mmol) was introduced, and the resulting mixture was stirred at reflux for 1 hour. Subsequently, the reaction mixture was cooled to 0 °C, subjected to suction filtration to get a bright yellow solid. This yellow solid was combined with a solution of sodium ethoxide in ethanol. Sodium metal (0.97 g, 42 mmol) had been dissolved in absolute ethanol (25 mL) to form the sodium ethoxide solution. The reaction mixture was then stirred at reflux temperature in air for 0.5 hours. The resultant mixture was poured over ice and acidified to pH 1 with dilute hydrochloric acid. The resulting solid was collected *via* suction filtration and dried under high vacuum, yielding compound **7** as a beige solid.

5-bromoindolin-2-one (**7**): Beige solid (1.84 g, 78.4% yield).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ ):  $\delta$  10.51 (s, 1H), 7.38-7.35 (d,  $J$  = 11.2 Hz, 2H), 6.76-6.77 (d,  $J$  = 5.2 Hz, 1H), 3.51 (s, 2H).

#### (*E*)-5,5'-Dibromo-[3,3'-biindolinylidene]-2,2'-dione (**8**)

5-Bromoisatin (1.23 g, 5.44 mmol) and compound **7** (1.16 g, 5.47 mmol) were suspended in glacial acetic acid (35 mL). Following this, concentrated hydrochloric acid (0.25 mL) was added and the reaction mixture was stirred at reflux temperature for a duration of 18 hours. The mixture was then cooled to room temperature and subjected to suction filtration. The resulting solid was washed with water, ethanol and ethyl acetate before being dried under high vacuum.



(*E*)-5,5'-Dibromo-[3,3'-biindolinylidene]-2,2'-dione (**8**): Blackish red solid (1.98 g, 86.7% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 11.13 (s, 2H), 7.30-7.31 (d, *J* = 1.6 Hz, 2H), 7.53-7.58 (dd, *J*<sub>1</sub> = 2 Hz, *J*<sub>2</sub> = 8.4 Hz, 2H), 6.82-6.84 (d, *J* = 8 Hz, 2H).

(*E*)-5,5'-Dibromo-1,1'-bis(2-hexyldecyl)-[3,3'-biindolinylidene]-2,2'-dione (**9**)

Compound **8** (2.10 g, 5.00 mmol) and K<sub>2</sub>CO<sub>3</sub> (6.91 g, 52.2 mmol) were suspended in dry DMF (100 mL). 2-Hexyldecylbromide (3.34 g, 10.9 mmol) was added into the reaction mixture and the reaction mixture was stirred at 100 °C for 20 hours. After completion of the reaction, the reaction mixture was poured over ice and subjected to extraction with Et<sub>2</sub>O. The combined organic layers were washed successively with H<sub>2</sub>O, brine and then dried over anhydrous Na<sub>2</sub>SO<sub>4</sub>. Subsequently, the solvent was concentrated under reduced pressure, yielding the crude product. Purification of the crude product was accomplished through silica gel column chromatography using 60% dichloromethane in petroleum ether. The solvent was evaporated under reduced pressure to get the desired product as a red solid (1.37 g, 31.3% yield).

(*E*)-5,5'-Dibromo-1,1'-bis(2-hexyldecyl)-[3,3'-biindolinylidene]-2,2'-dione (**9**): Red solid (1.37g, 31.3% yield). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>): δ 9.41-9.41 (d, *J* = 1.6 Hz, 2H), 7.48-7.50 (dd, *J*<sub>1</sub> = 2 Hz, *J*<sub>2</sub> = 8.4 Hz, 2H), 6.70-6.80 (d, *J* = 8.4 Hz, 2H), 3.74-3.78 (t, *J* = 7.2 Hz, 4H), 2.19 (m, 4H), 1.62-1.68 (m, 35H), 0.86-0.93 (m, 12H).

## Spectral data of intermediate compounds and monomers

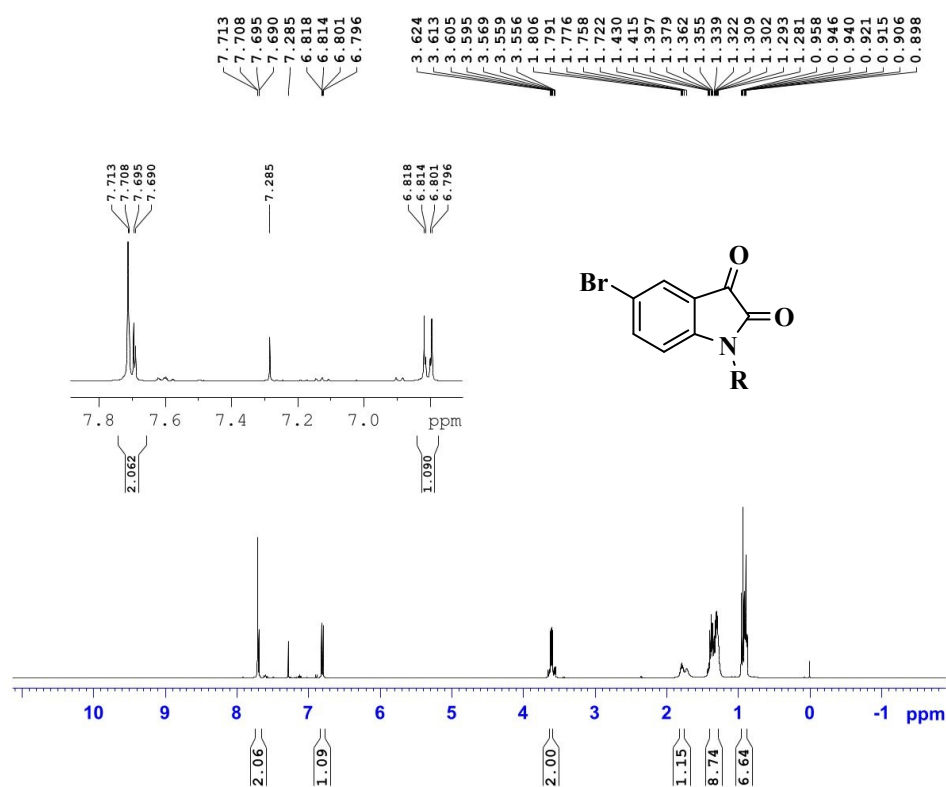


Figure S1: <sup>1</sup>H NMR spectra of compound S1

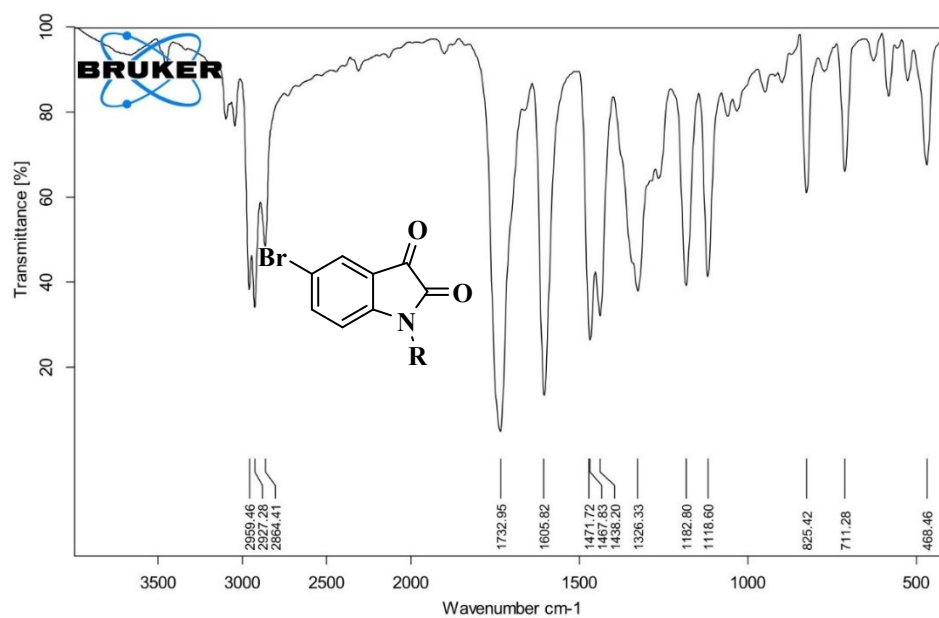


Figure S2: IR spectrum (KBr pellet) of compound S1

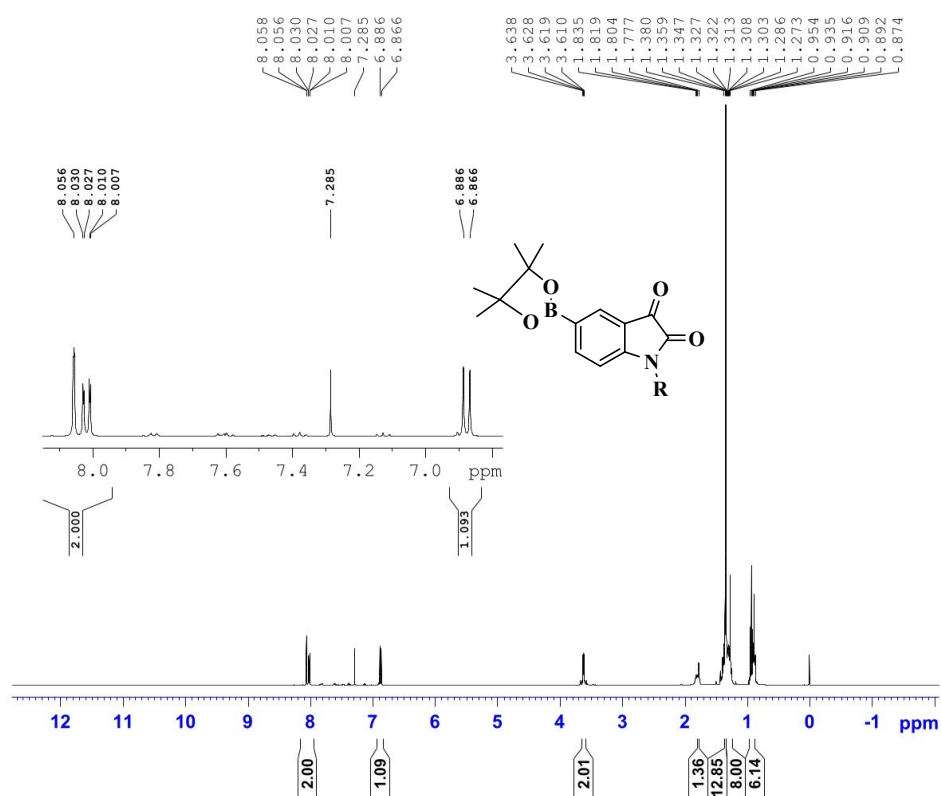


Figure S3: <sup>1</sup>H NMR spectra of compound S2

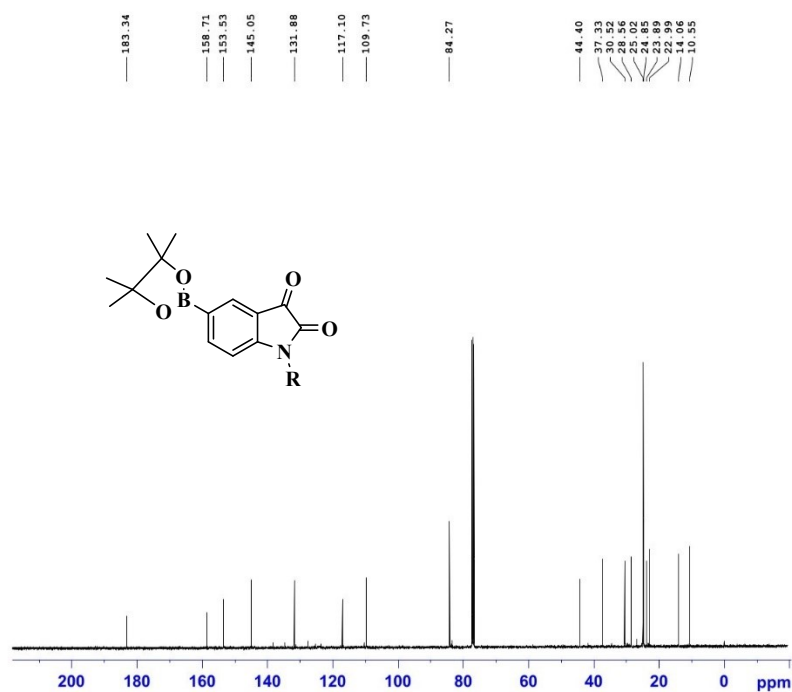


Figure S4: <sup>13</sup>C NMR spectra of compound S2

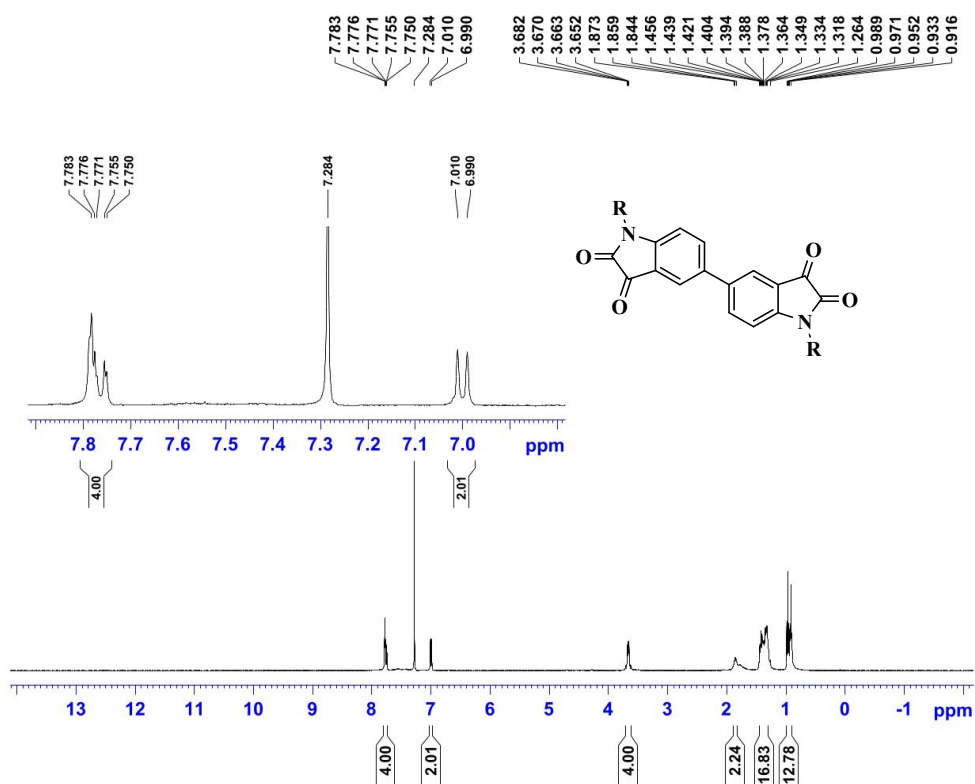


Figure S5: <sup>1</sup>H NMR spectra of compound **1**

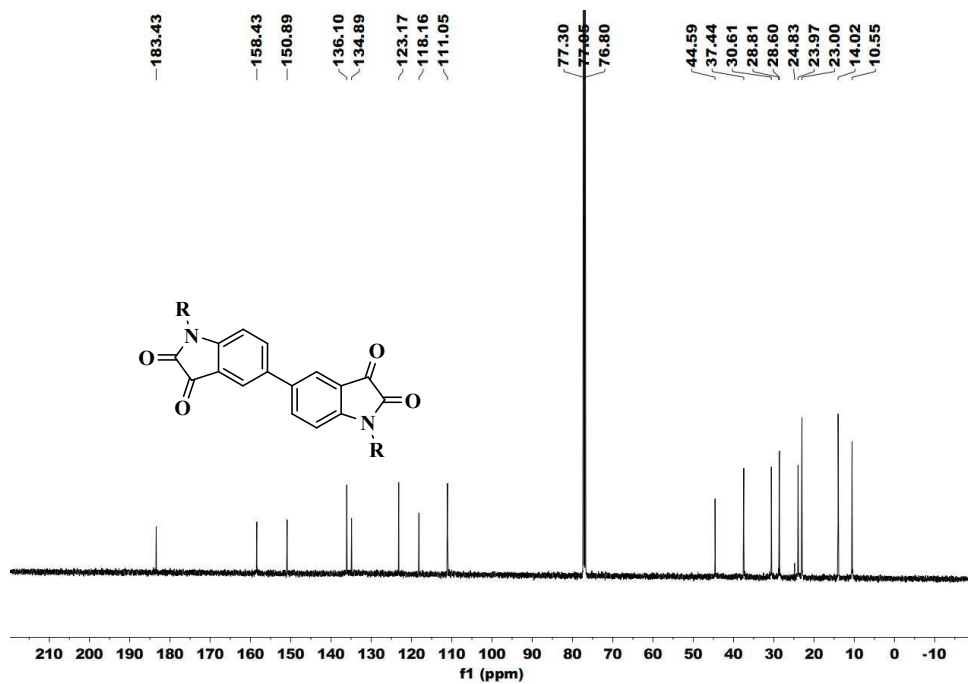


Figure S6: <sup>13</sup>C NMR spectra of compound **1**

## Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

19 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

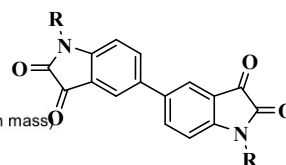
Elements Used:

C: 0-32 H: 0-100 N: 0-2 O: 0-4

090223\_PPBI 22 (0.240)

Test Name :

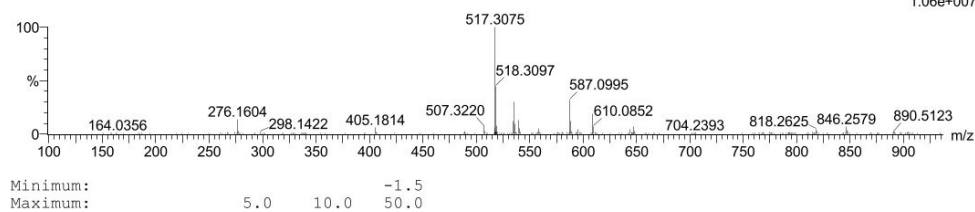
1: TOF MS ES+



IITRPR

XEVO G2-XS QTOF  
090223\_PPBI

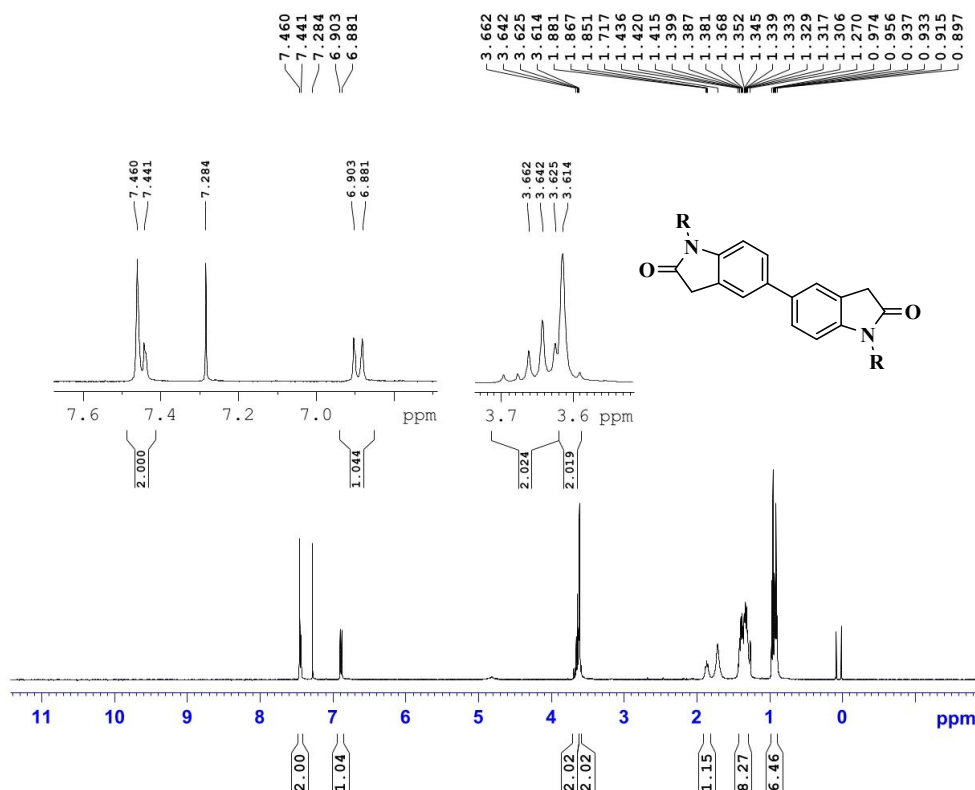
1.06e+007



Minimum: -1.5  
Maximum: 5.0 10.0 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
517.3075	517.3066	0.9	1.7	13.5	799.0	n/a	n/a	C32 H41 N2 O4

Figure S7: HR-MS data of compound 1

Figure S8:  $^1\text{H}$  NMR spectra of compound 2

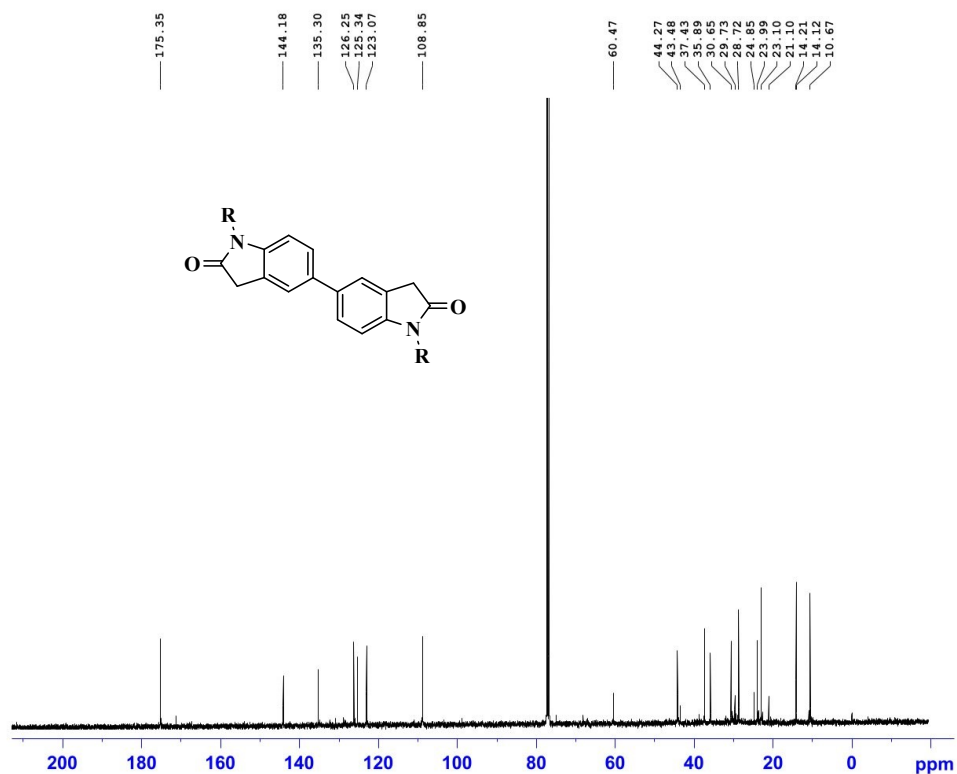


Figure S9: <sup>13</sup>C NMR spectra of compound 2

#### Elemental Composition Report

Page 1

##### Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

34 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

Elements Used:

C: 0-32 H: 0-100 N: 0-2 O: 0-4

090223\_PPRBI 72 (0.737)

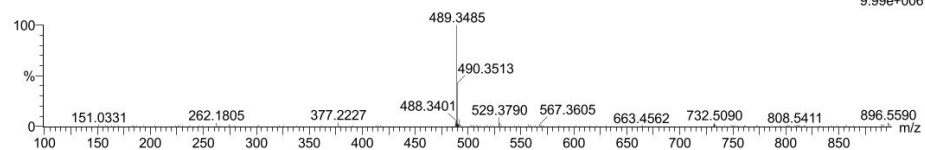
Test Name :

1: TOF MS ES+

IITRPR

XEVO G2-XS QTOF  
090223\_PPRBI

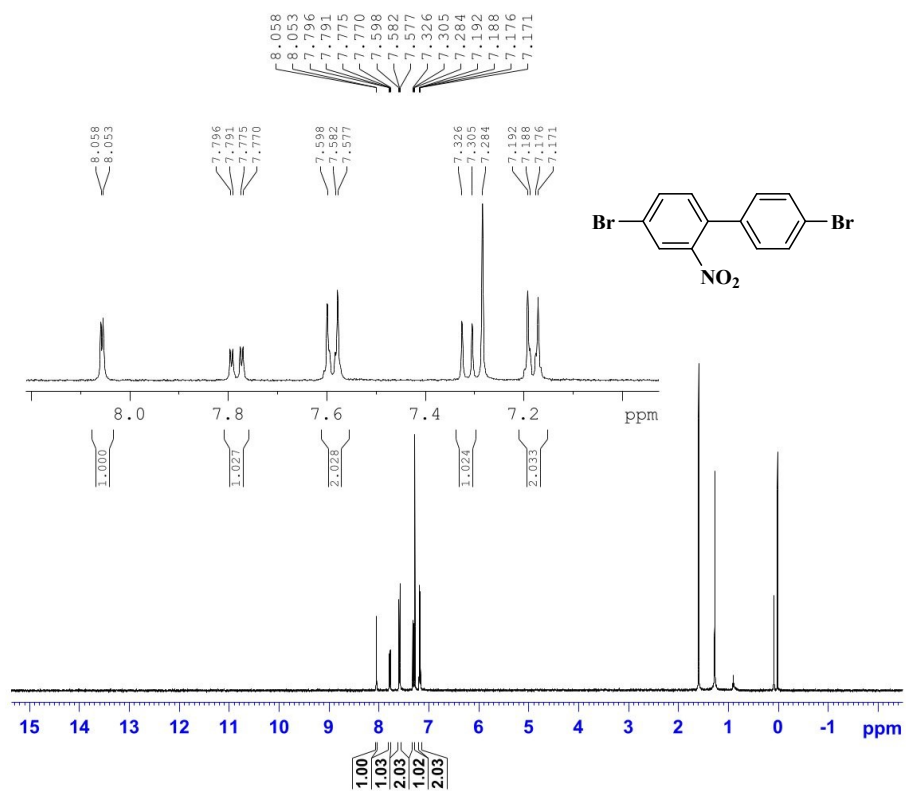
9.99e+006



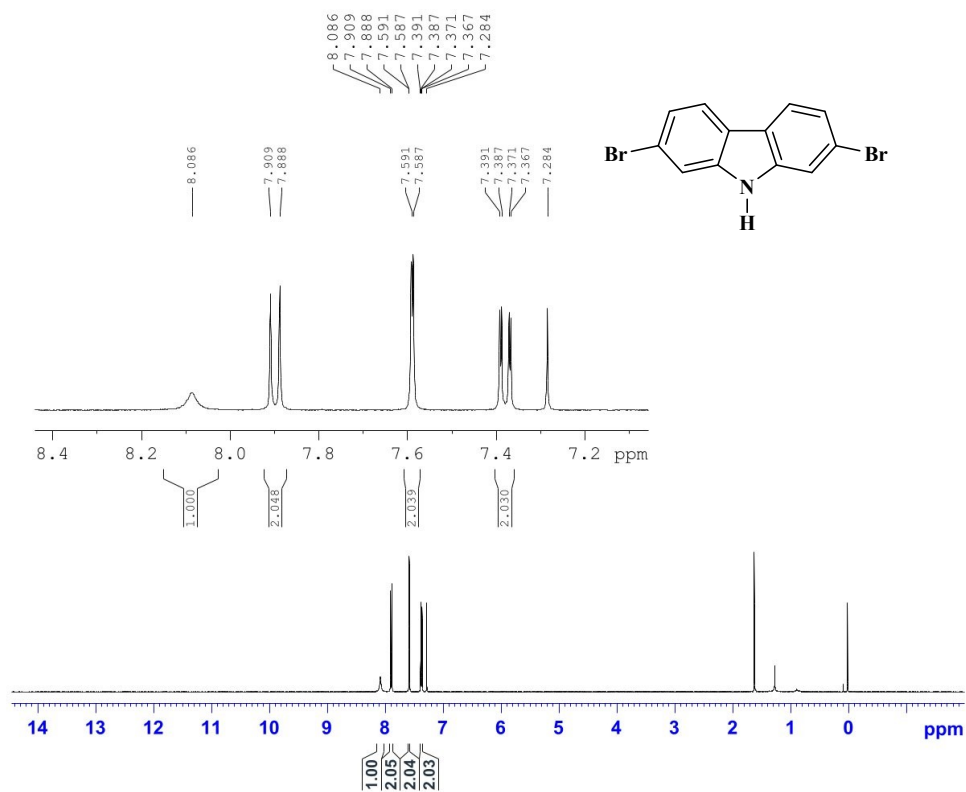
Minimum: -1.5  
Maximum: 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
489.3485	489.3481	0.4	0.8	11.5	848.8	n/a	n/a	C32 H45 N2 O2

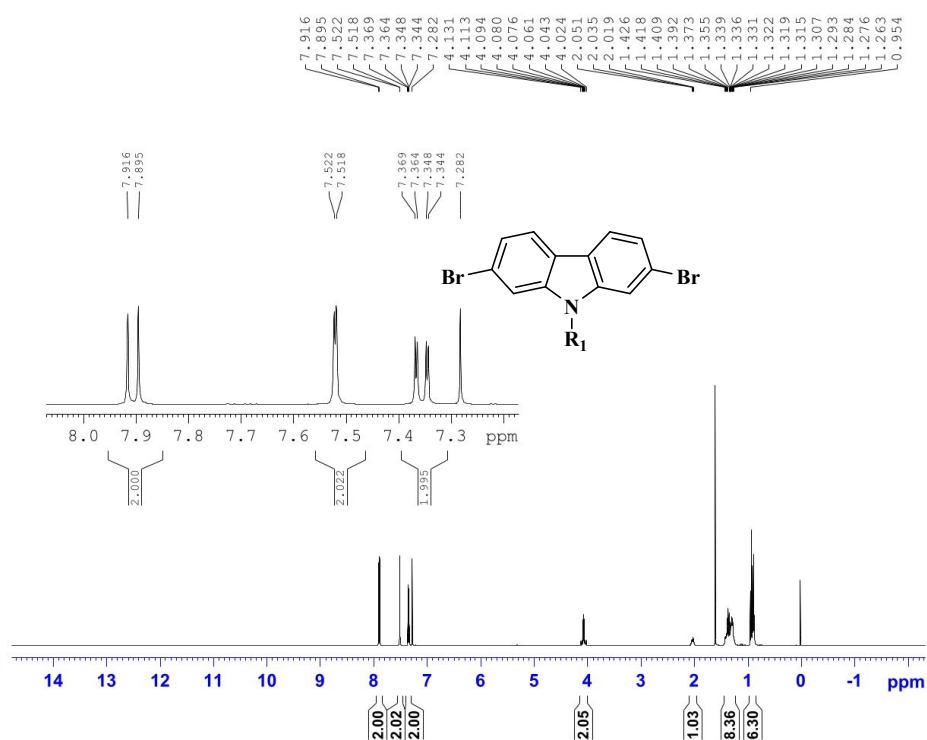
Figure S10: HR-MS data of compound 2



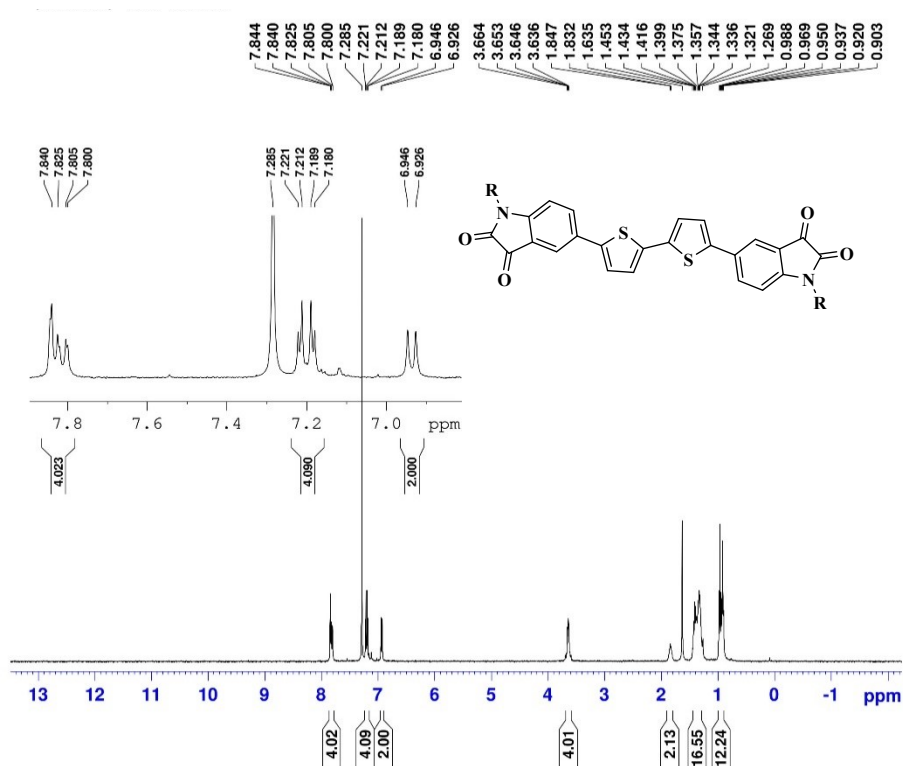
**Figure S11:** <sup>1</sup>H NMR spectra of compound S3



**Figure S12:** <sup>1</sup>H NMR spectra of compound S4



**Figure S13:**  $^1H$  NMR spectra of compound **S5**



**Figure S14:**  $^1H$  NMR spectra of compound **3**



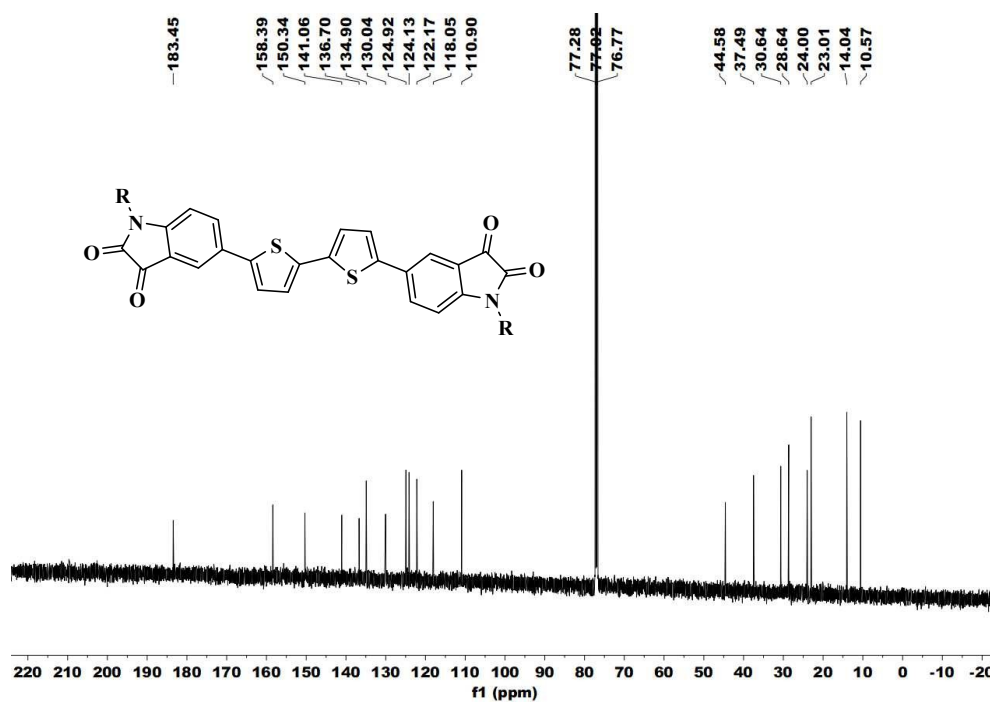


Figure S15:  $^{13}\text{C}$  NMR spectra of compound **3**

## Elemental Composition Report

Page 1

### Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

3 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

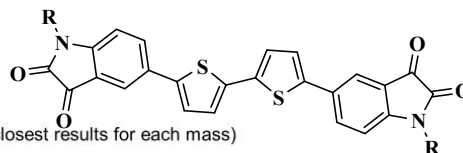
Elements Used:

C: 40-40 H: 1-100 N: 2-2 O: 2-4 S: 2-2

220823\_IBT 48 (0.497) Cm (47:54)

Test Name :

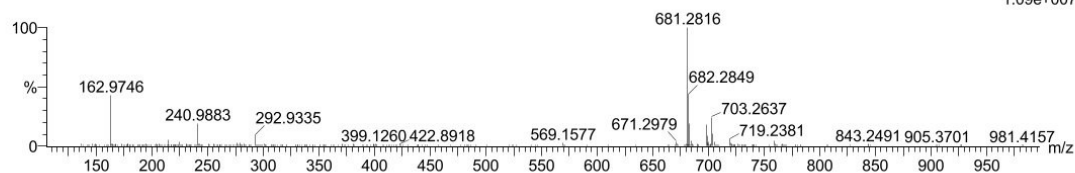
1: TOF MS ES+



IITRPR

XEVO G2-XS QTOF  
220823\_IBT

1.09e+007



Minimum: -1.5  
Maximum: 5.0 20.0 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
681.2816	681.2821	-0.5	-0.7	19.5	785.1	n/a	n/a	C40 H45 N2 O4 S2

Figure S16: HR-MS data of compound **3**

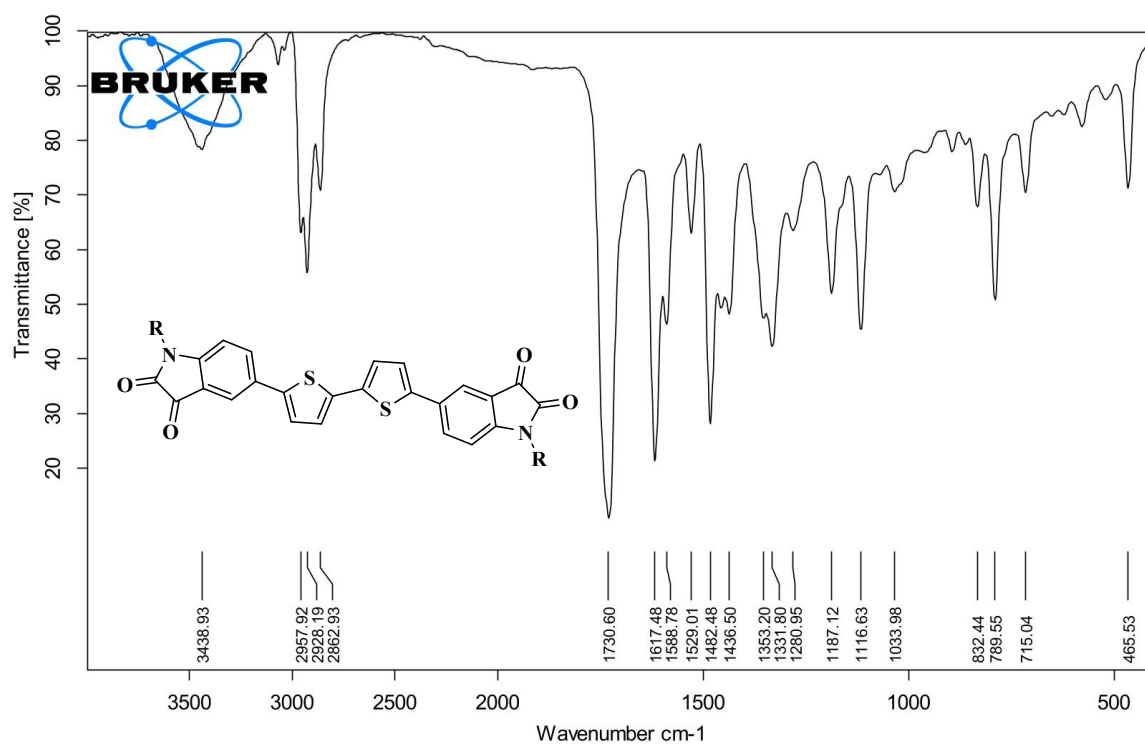


Figure S17: IR spectrum (KBr pellet) of compound 3

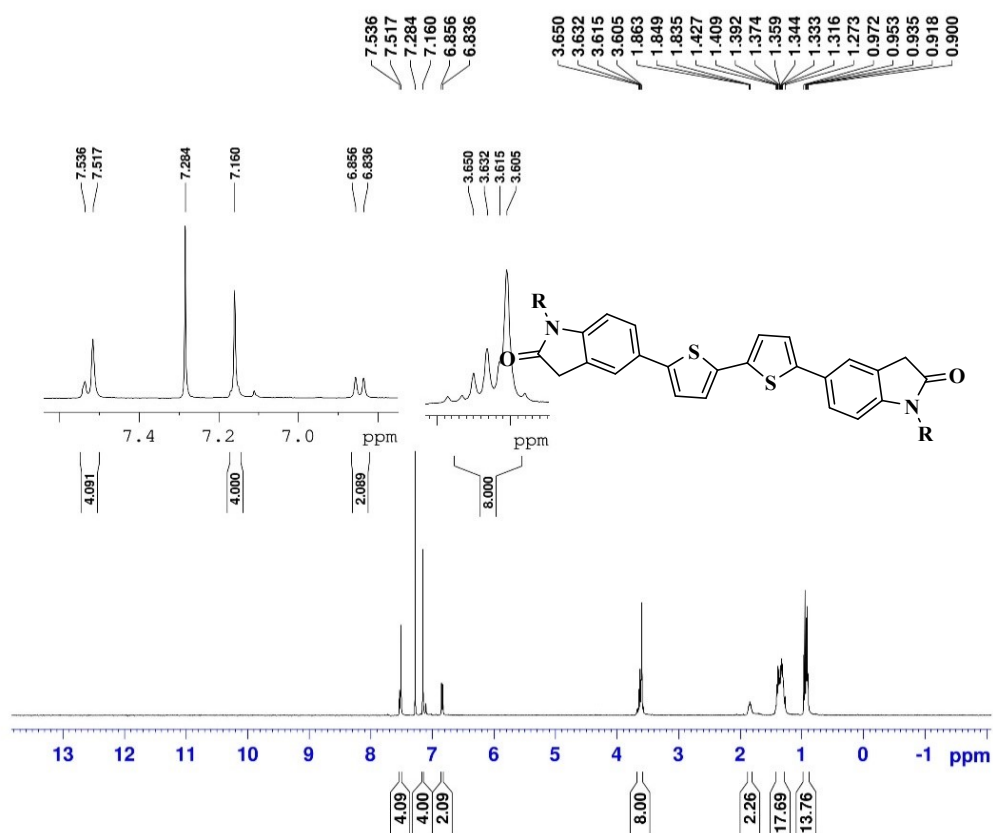


Figure S18: <sup>1</sup>H NMR spectra of compound 4

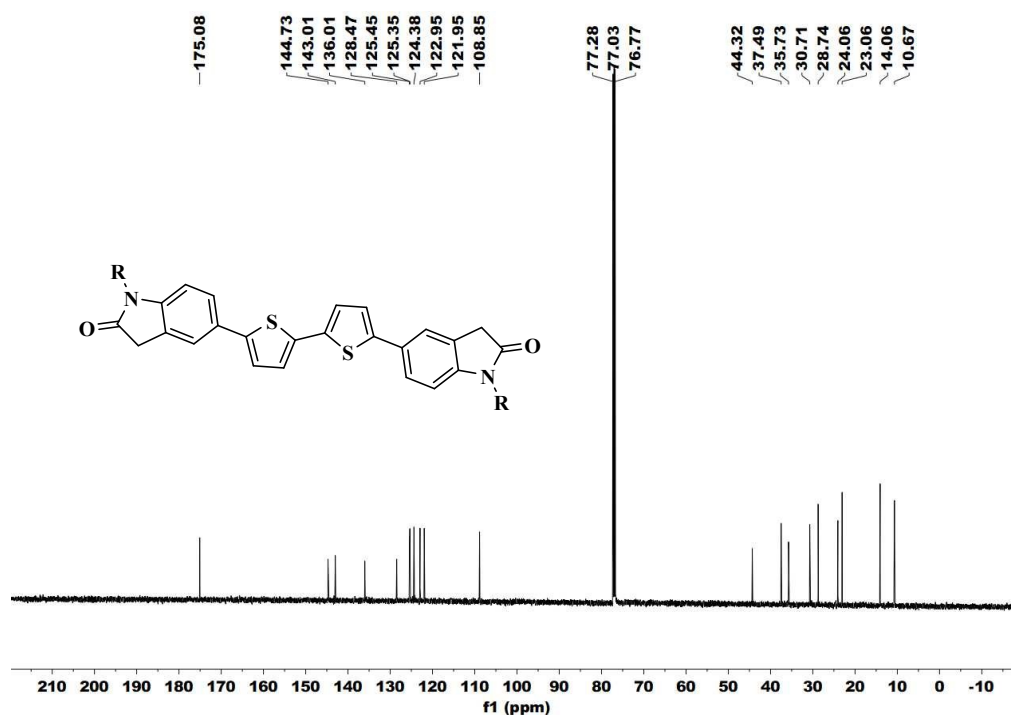


Figure S19: <sup>13</sup>C NMR spectra of compound 4

#### Elemental Composition Report

Page 1

##### Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

3 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

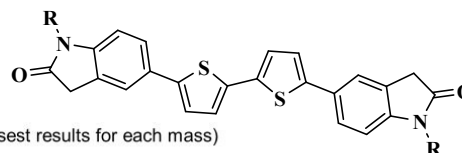
Elements Used:

C: 40-40 H: 1-100 N: 2-2 O: 2-4 S: 2-2

220823\_IBTRED 27 (0.294) Cm (27:41)

Test Name :

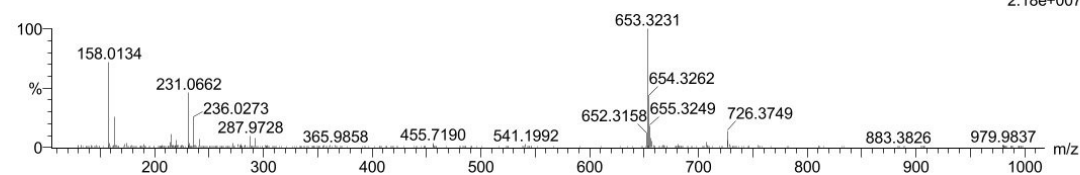
1: TOF MS ES+



IITRPR

XEVO G2-XS QTOF  
220823\_IBTRED

2.18e+007



Minimum: -1.5  
Maximum: 5.0 20.0 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
653.3231	653.3235	-0.4	-0.6	17.5	928.8	n/a	n/a	C40 H49 N2 O2 S2

Figure S20: HR-MS data of compound 4

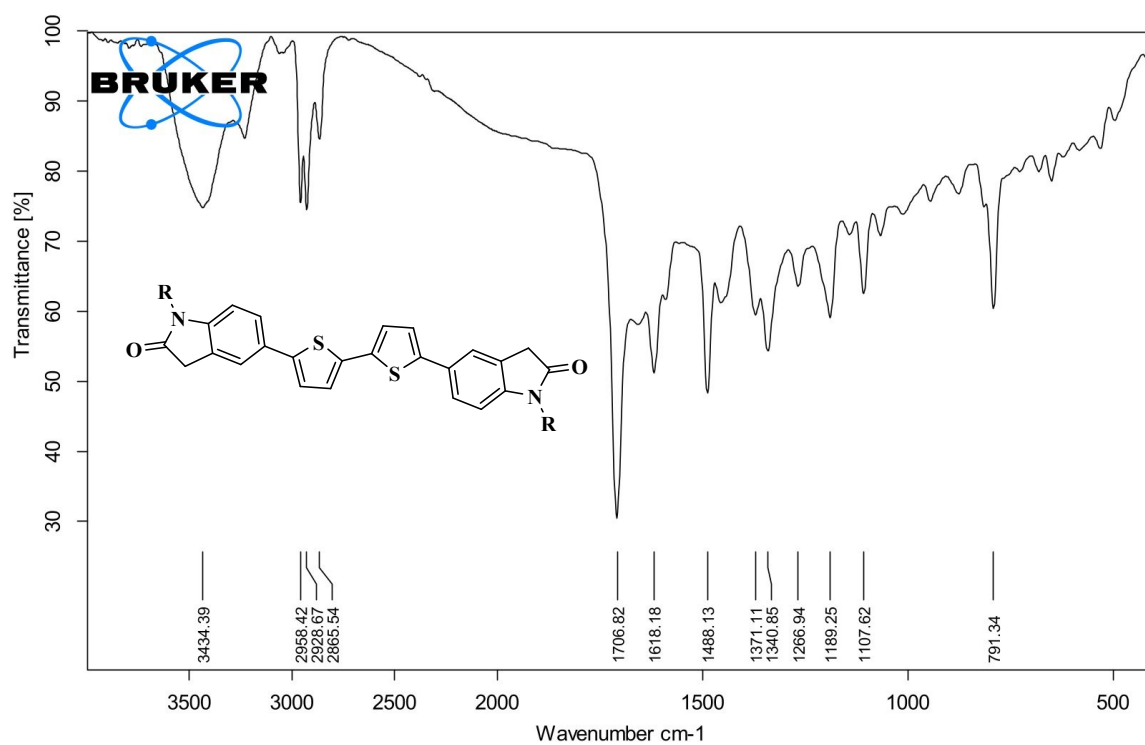


Figure S21: IR spectrum (KBr pellet) of compound 4

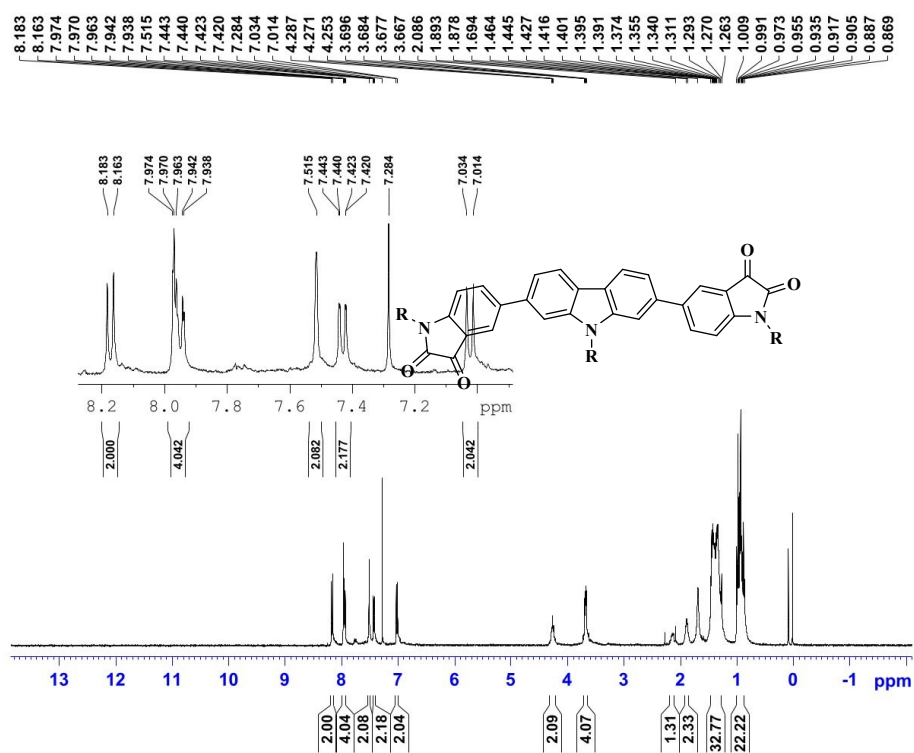


Figure S22: <sup>1</sup>H NMR spectra of compound 5

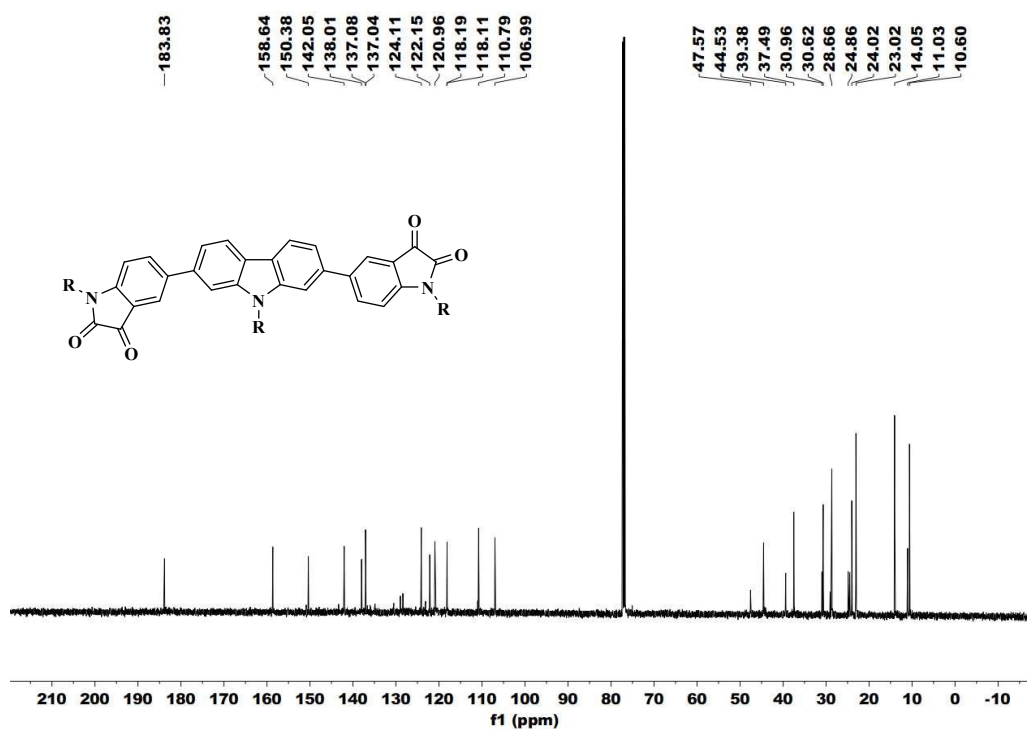


Figure S23:  $^{13}\text{C}$  NMR spectra of compound 5

## Elemental Composition Report

Page 1

### Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Even Electron Ions

14 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

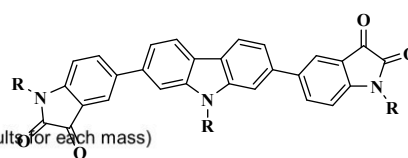
Elements Used:

C: 0-52 H: 0-80 N: 0-3 O: 0-4

090223\_PPCIS 31 (0.328)

Test Name :

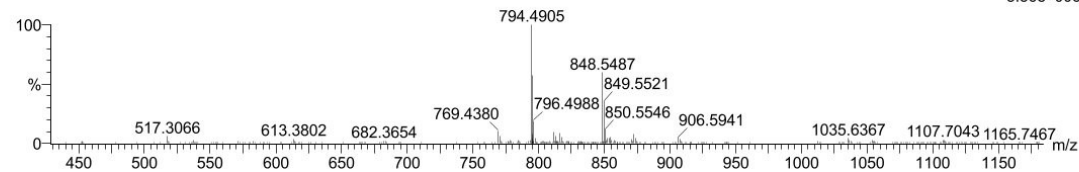
1: TOF MS ES+



IITRPR

XEVO G2-XS QTOF  
090223\_PPCIS

5.36e+006



Minimum: -1.5  
Maximum: 5.0 10.0 50.0

Mass	Calc. Mass	mDa	PPM	DBE	i-FIT	Norm	Conf(%)	Formula
794.4905	794.4897	0.8	1.0	22.5	592.7	n/a	n/a	C52 H64 N3 O4

Figure S24:  $^{13}\text{C}$  NMR spectra of compound 5

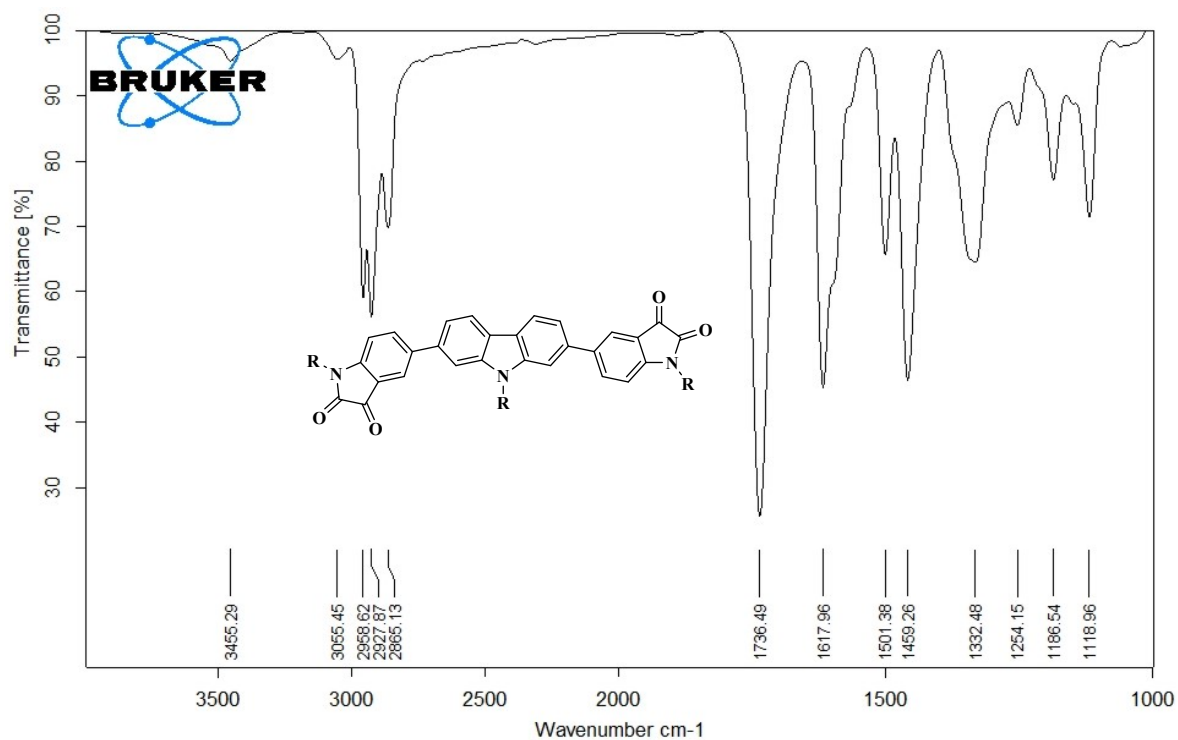


Figure S25: IR spectrum (KBr pellet) of compound **5**

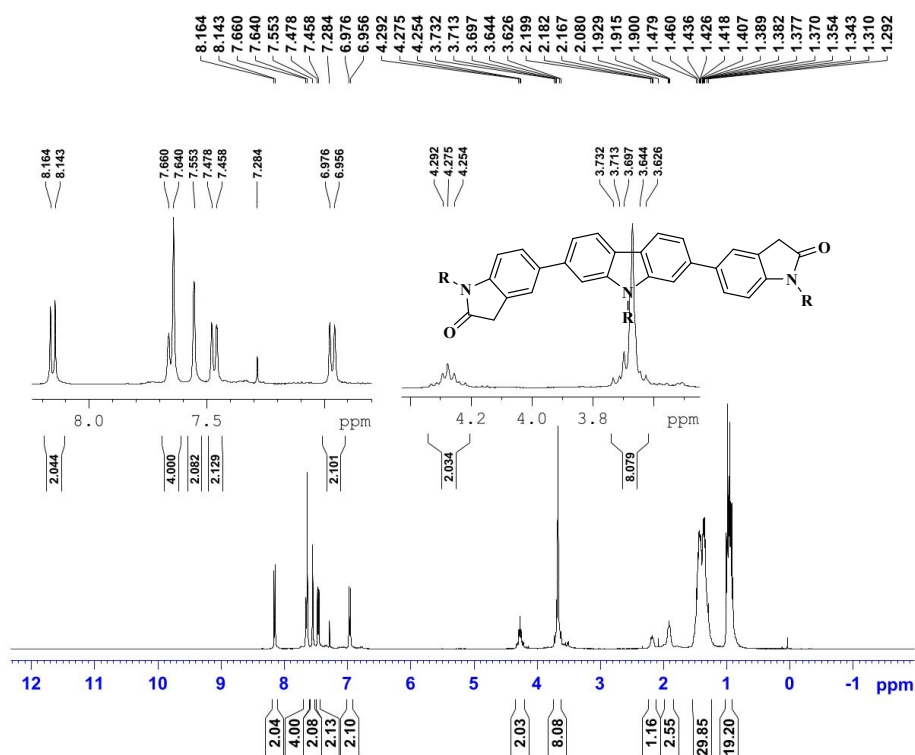


Figure S26:  $^1\text{H}$  NMR spectra of compound **6**

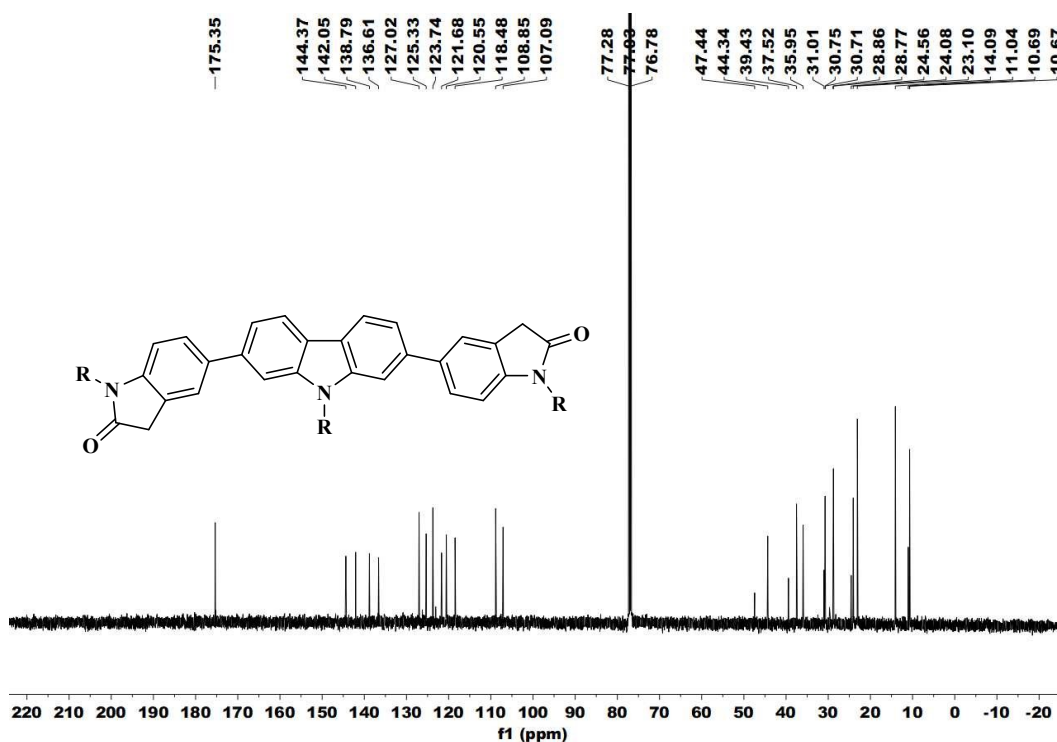


Figure S27: <sup>13</sup>C NMR spectra of compound 6

## Elemental Composition Report

Page 1

### Single Mass Analysis

Tolerance = 5.0 mDa / DBE: min = -1.5, max = 50.0

Element prediction: Off

Number of isotope peaks used for i-FIT = 3

Monoisotopic Mass, Odd and Even Electron Ions

38 formula(e) evaluated with 1 results within limits (up to 50 closest results for each mass)

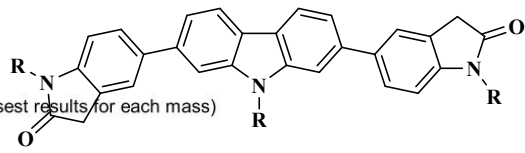
Elements Used:

C: 0-52 H: 0-100 N: 0-3 O: 0-4

090223\_PPCISRED 25 (0.277)

Test Name :

1: TOF MS ES+



XEVO G2-XS QTOF  
090223\_PPCISRED

2.68e+007

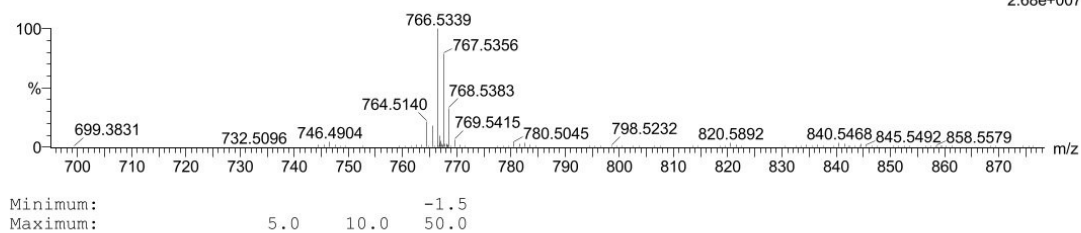
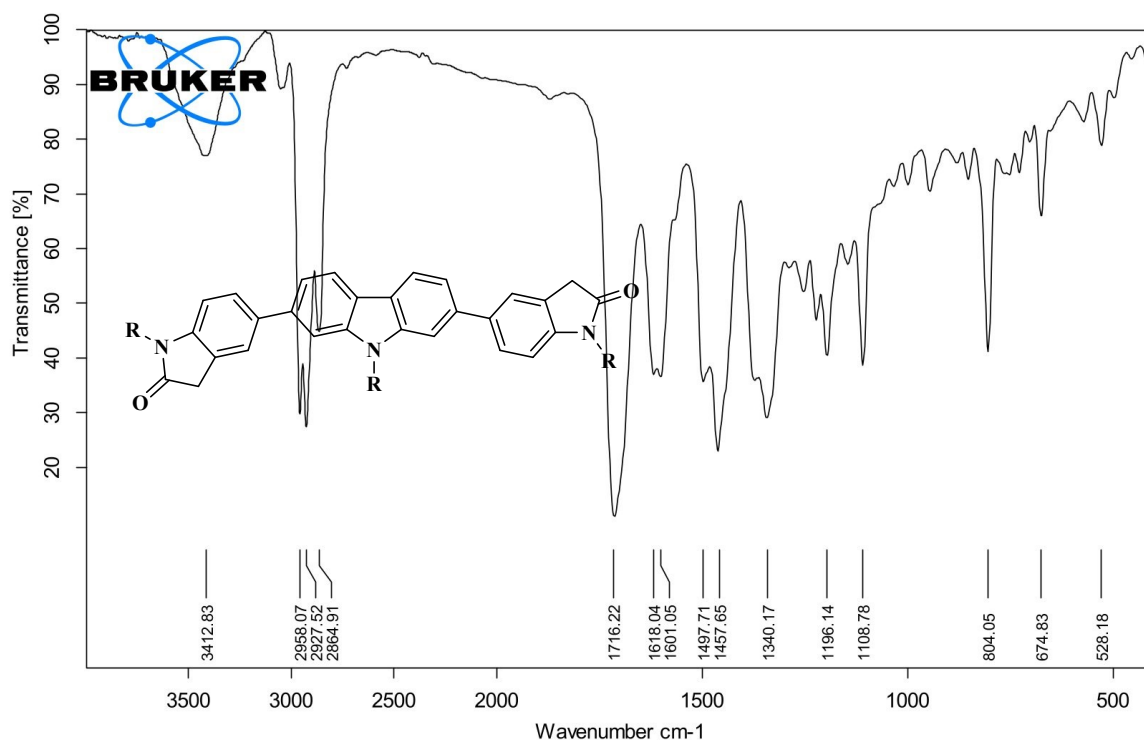
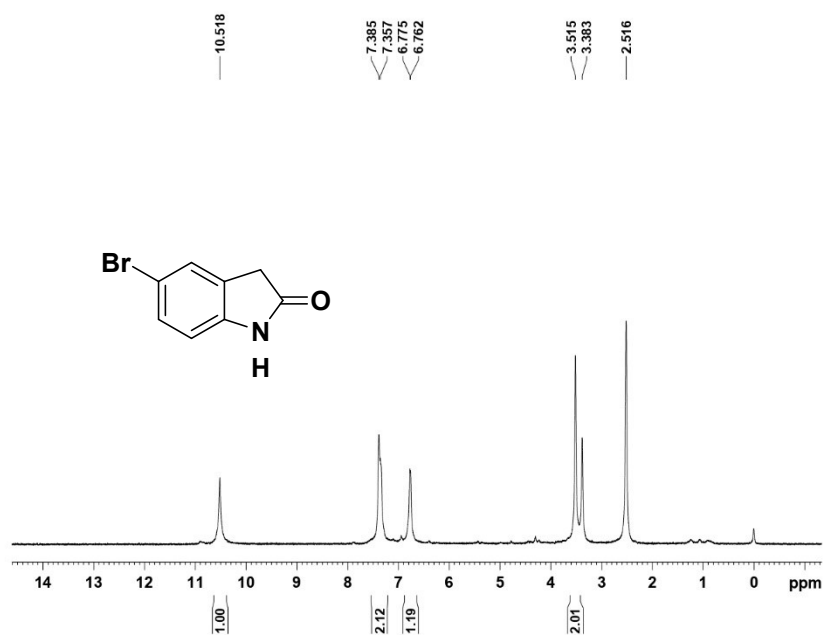


Figure S28: HR-MS data of compound 6



**Figure S29:** IR spectrum (KBr pellet) of compound **6**



**Figure S30:**  $^1\text{H}$  NMR spectra of compound **7**



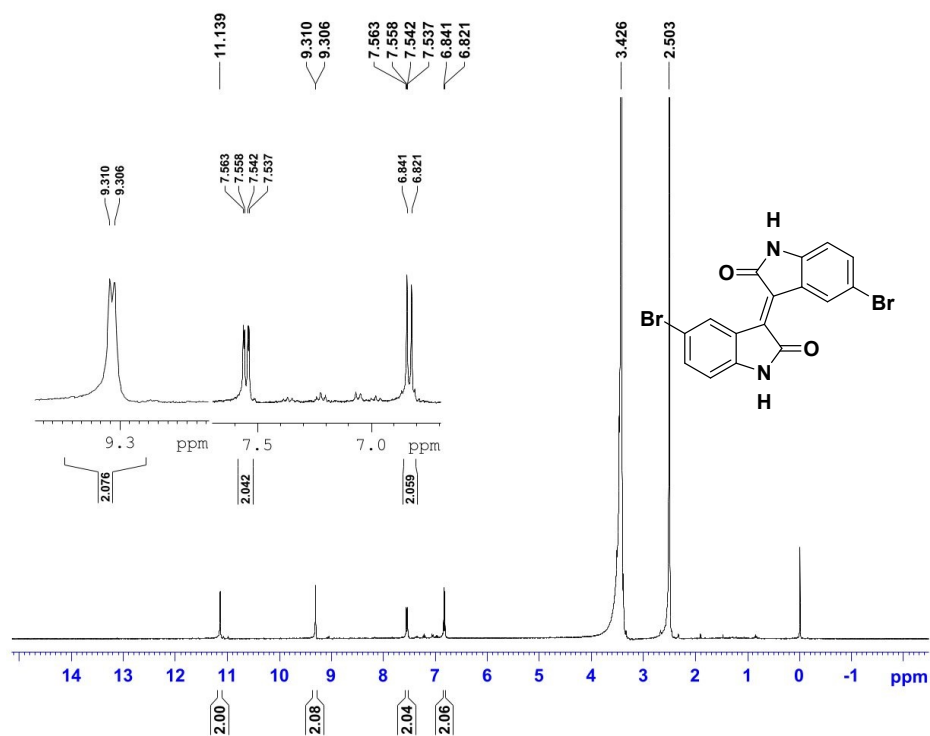


Figure S31: <sup>1</sup>H NMR spectra of compound 8

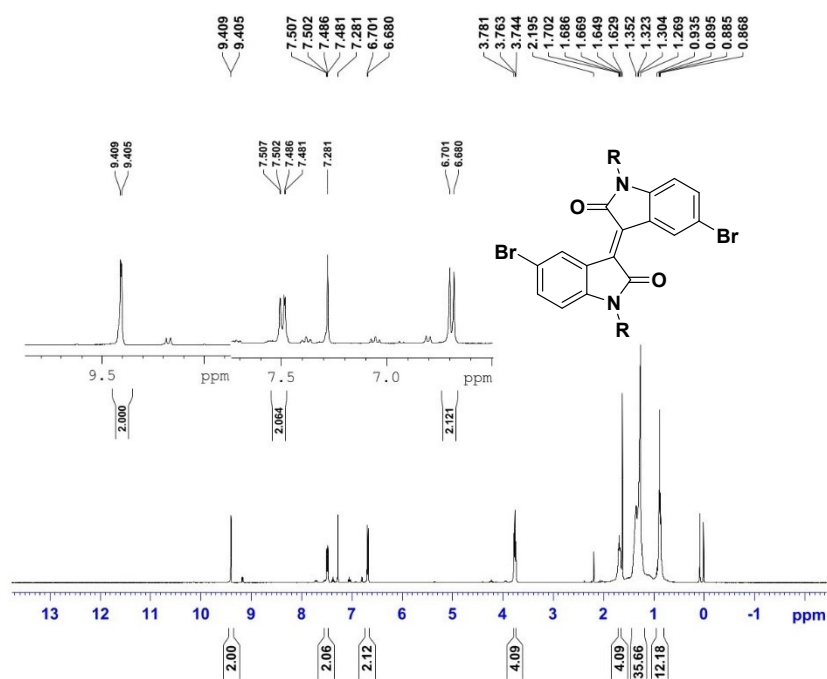


Figure S32: <sup>13</sup>C NMR spectra of compound 9

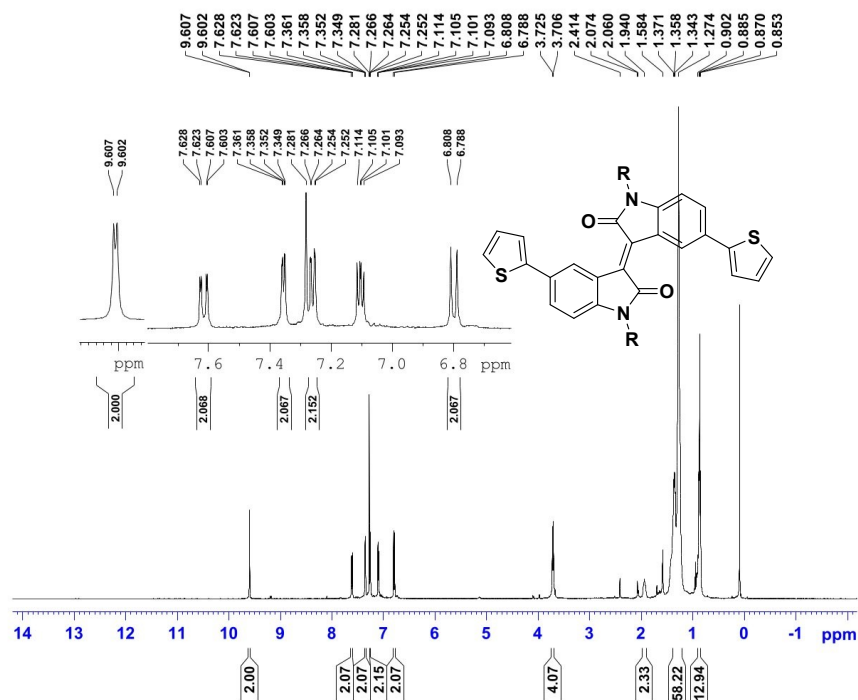


Figure S33: <sup>1</sup>H NMR spectra of compound 10

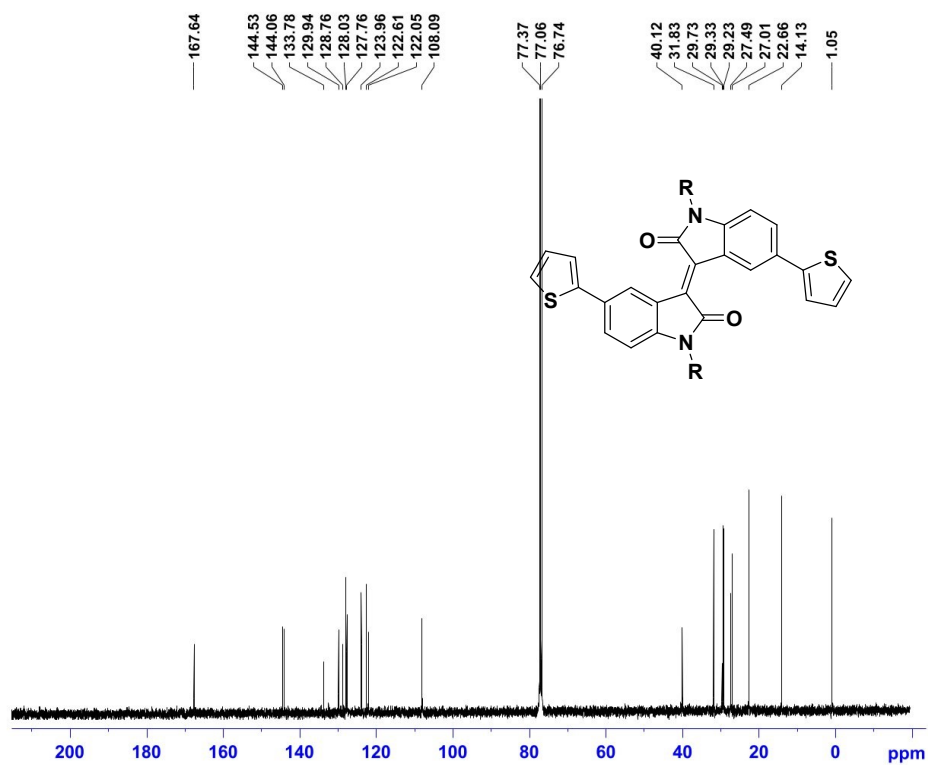


Figure S34: <sup>13</sup>C NMR spectra of compound 10

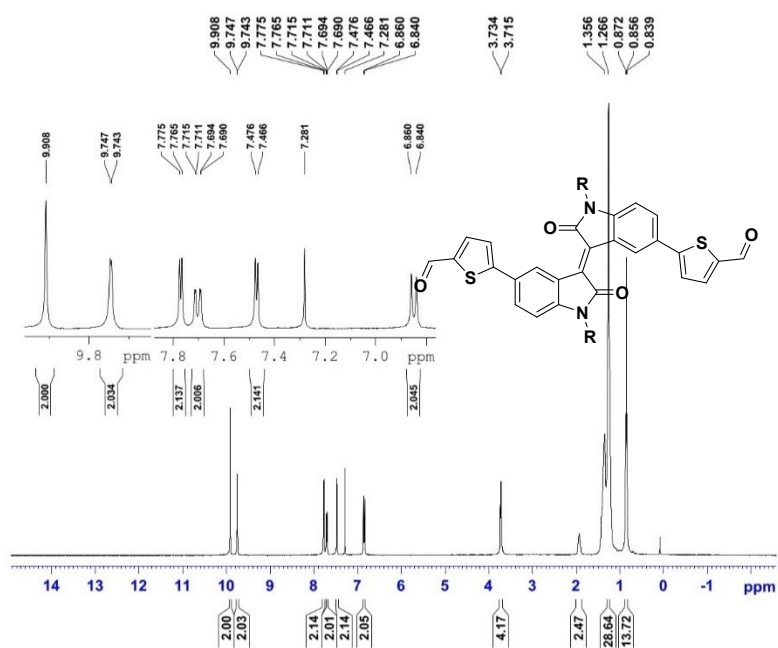


Figure S35: <sup>1</sup>H NMR spectra of compound 11

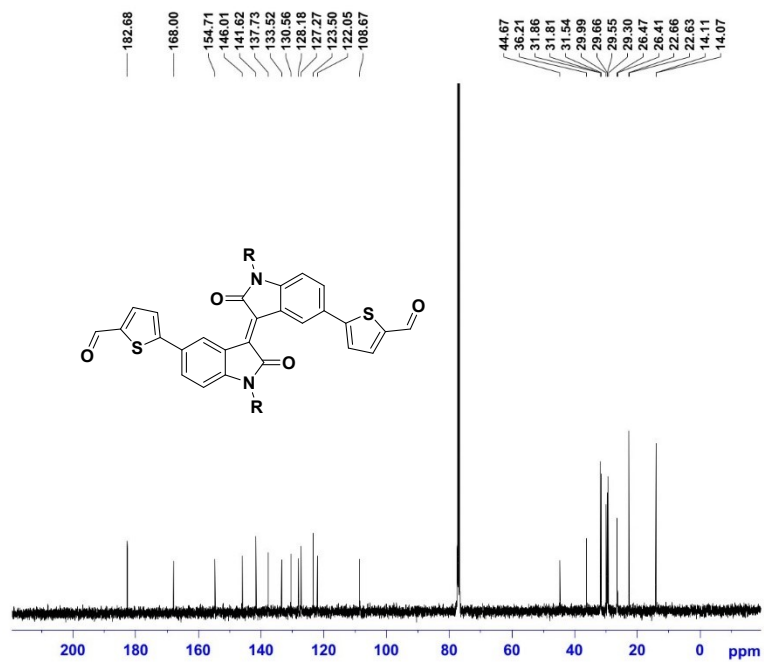


Figure S36: <sup>13</sup>C NMR spectra of compound 11

## Spectral data of polymers

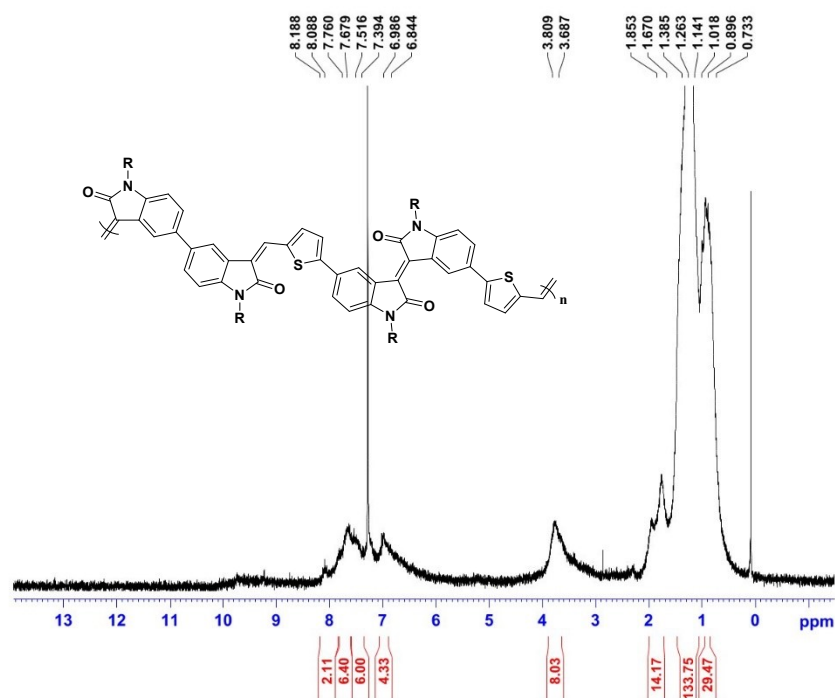


Figure S37: <sup>1</sup>H NMR spectra of polymer AIRI

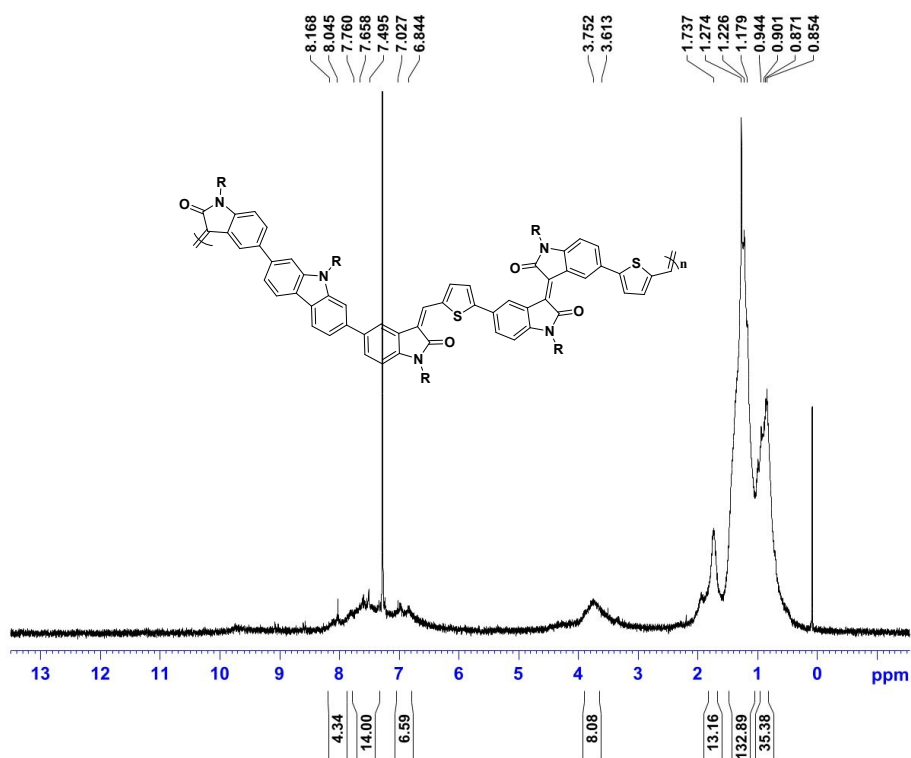


Figure S38: <sup>1</sup>H NMR spectra of polymer AICRI

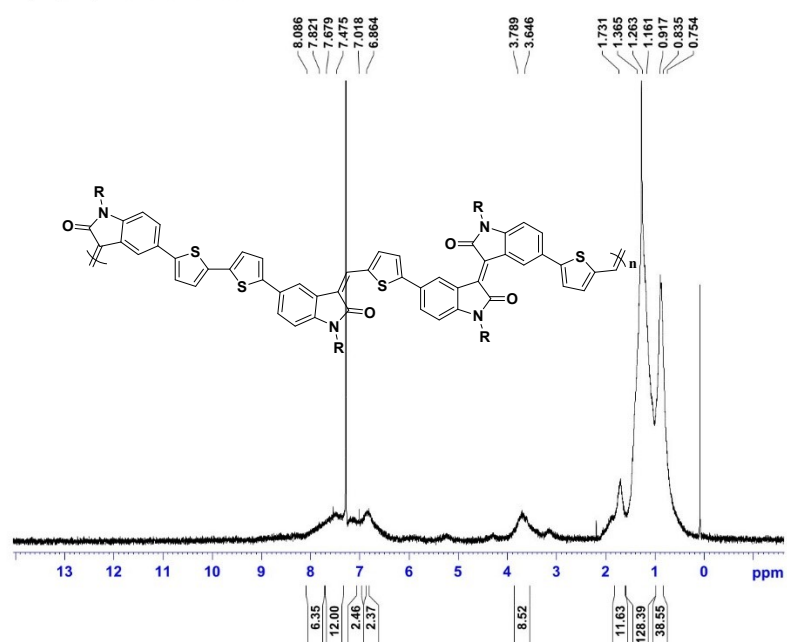


Figure S39:  $^1\text{H}$  NMR spectra of polymer AITRI

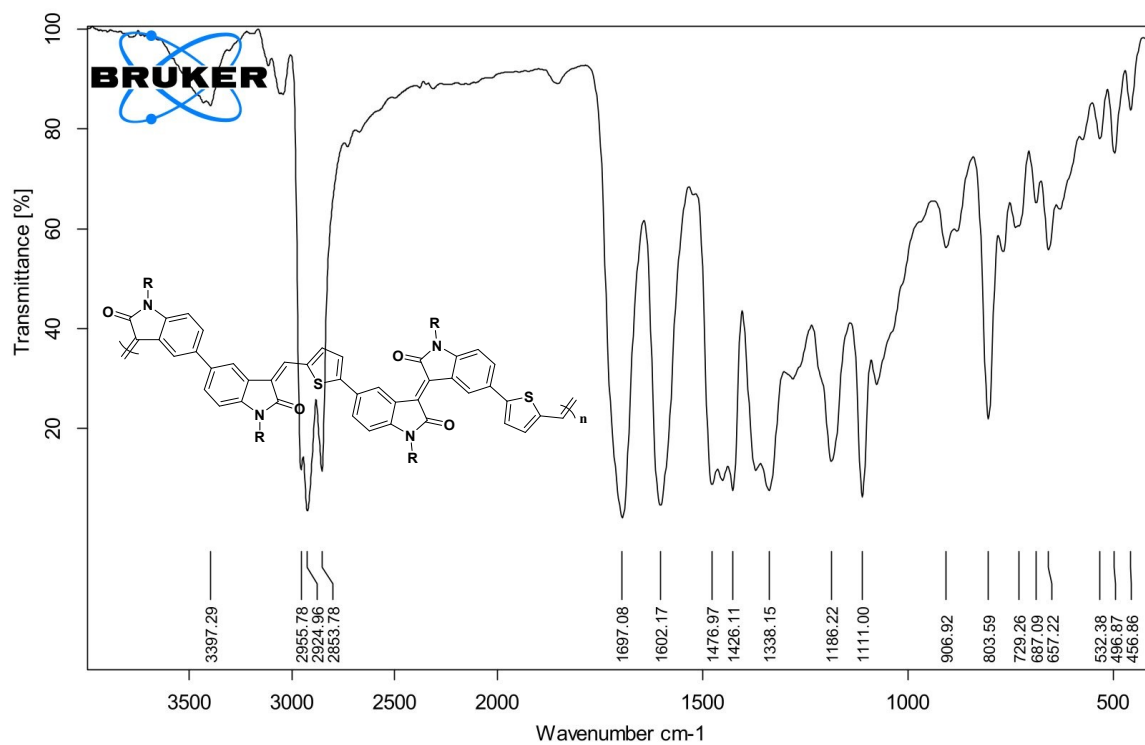


Figure S40: IR spectrum (KBr pellet) of polymer AITRI

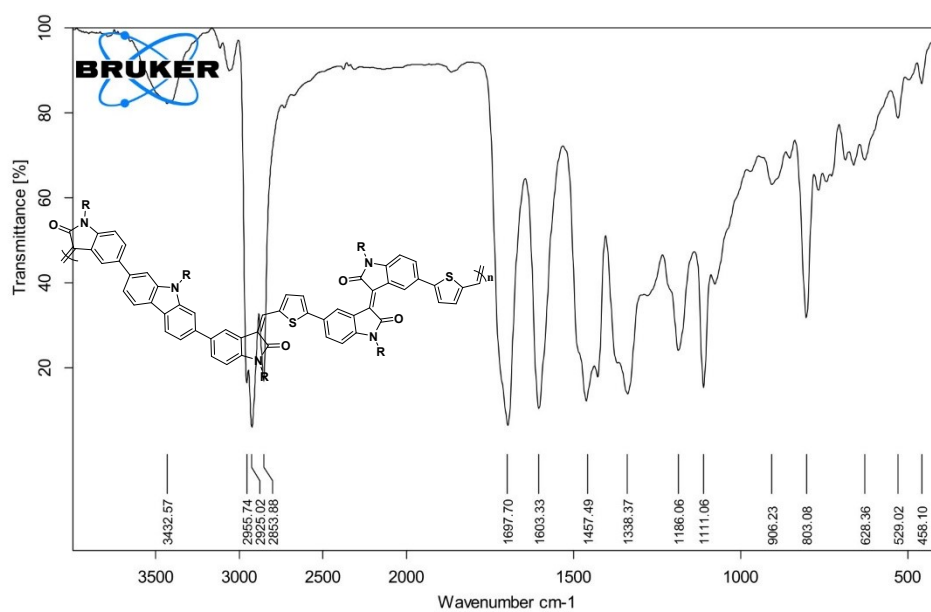


Figure S41: IR spectrum (KBr pellet) of polymer AICRI

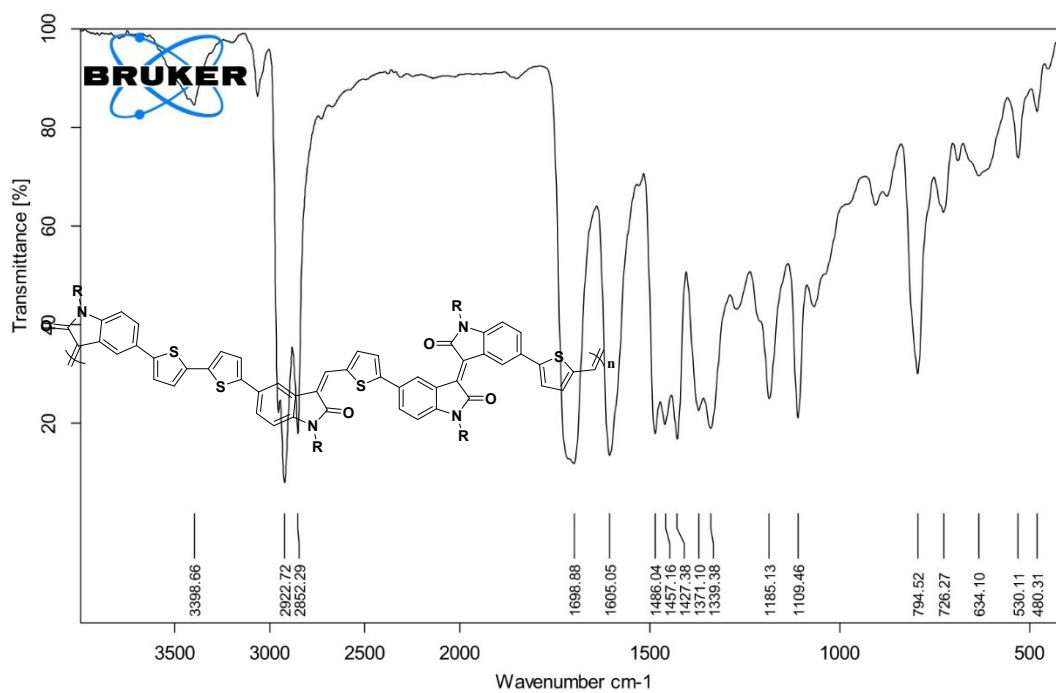


Figure S42: IR spectrum (KBr pellet) of polymer AITRI

## References

- 1 P. R. Patel, Y. G. Kapdi, S. S. Soni, S. S. Zade and A. L. Patel, *New J. Chem.*, 2024, 48, 7007-7019
- 2 Y. W. Huang, Y. C. Lin, J. S. Li, W. C. Chen and C. C. Chueh, *J. Mater. Chem. C*, 2021, **9**, 9473–9483.
- 3 Y. Jiang, J. Chen, Y. Sun, Q. Li, Z. Cai, J. Li, Y. Guo, W. Hu and Y. Liu, *Adv. Mater.*, 2019, **31**, 1900342.
- 4 F. Dierschke, A. C. Grimsdale and K. Müllen, *Synthesis (Stuttg.)*, 2003, 2470–2472.
- 5 H. Wang, P. Cheng, Y. Liu, J. Chen, X. Zhan, W. Hu, Z. Shuai, Y. Li and D. Zhu, *J. Mater. Chem.*, 2012, **22**, 3432–3439.
- 6 H. Li, S. Sun, T. Salim, S. Bomma, A. C. Grimsdale and Y. M. Lam, *J. Polym. Sci. Part A Polym. Chem.*, 2012, **50**, 250–260.
- 7 V. S. Kadam, P. A. Bhatt, H. K. Machhi, S. S. Soni, S. S. Zade and A. L. Patel, *Nano Sel.*, 2020, **1**, 491–498.
- 8 T. Lei, Y. Lei, Y. Cao, C.-J. Fan, S.-C. Liu, J. Yuan and Pei, *J. Am. Chem. Soc.*, 2011, **133**, 6099–6101.