

Production of Assorted Nanoparticle Structures using Silver-Cerium Redox Chemistry for Inactivation of Coronavirus

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| | | Peak Center | FWHM | Average Crystallite Diameter(nm) |
|-----------------------|----|-------------|--------|----------------------------------|
| 1:1.2 (60-min mixing) | Ce | 28.55 | 1.277 | 6.418 |
| | Ag | 38.22 | 0.5625 | 14.95 |
| 1:1 (60-min mixing) | Ce | 28.29 | 1.202 | 6.813 |
| | Ag | 37.95 | 0.4499 | 18.67 |
| 1:0.5 (60-min mixing) | Ce | 28.60 | 0.8798 | 9.319 |
| | Ag | 38.17 | 0.3606 | 23.32 |
| 1:0.2 (60-min mixing) | Ce | 28.57 | 0.8423 | 9.732 |
| | Ag | 38.14 | 0.3659 | 22.97 |
| 1:1.2 (2-min mixing) | Ce | 28.54 | 1.000 | 8.190 |
| | Ag | 38.16 | 0.3874 | 21.70 |
| 1:0.5 (2-min mixing) | Ce | 28.51 | 1.005 | 8.157 |
| | Ag | 38.10 | 0.3820 | 22.00 |
| 1:0.2 (2-min mixing) | Ce | 28.67 | 0.9573 | 8.565 |
| | Ag | 38.23 | 0.4521 | 18. |

Table S1 Average Crystallite Diameter determined via the Scherrer equation. Lower angle peaks for high density atomic planes were baseline-corrected and fit with Gaussian functions in *Origin* graphing software. Full-width half maxima and analytically determined peak centers were then used to calculate average crystallite diameters using the Scherrer equation as

$$D_{hkl} = k * \lambda / (B_{hkl} * \cos(\theta)) \quad (1)$$

where D_{hkl} is the calculated average crystal diameter, k is a shape factor approximated as 0.9, λ is the x-ray source wavelength (1.540598 Å), and θ is the Bragg angle of the diffraction peak (hkl).

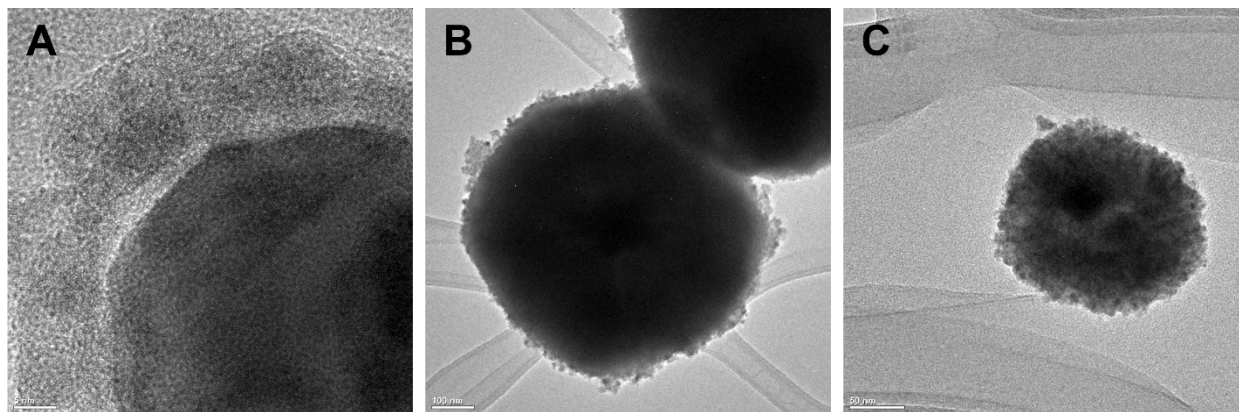


Figure S1 Additional High-resolution Transmission Electron Microscopy images. Formulations 1.2, 0.5, and 0.2 images, A to C, respectively.

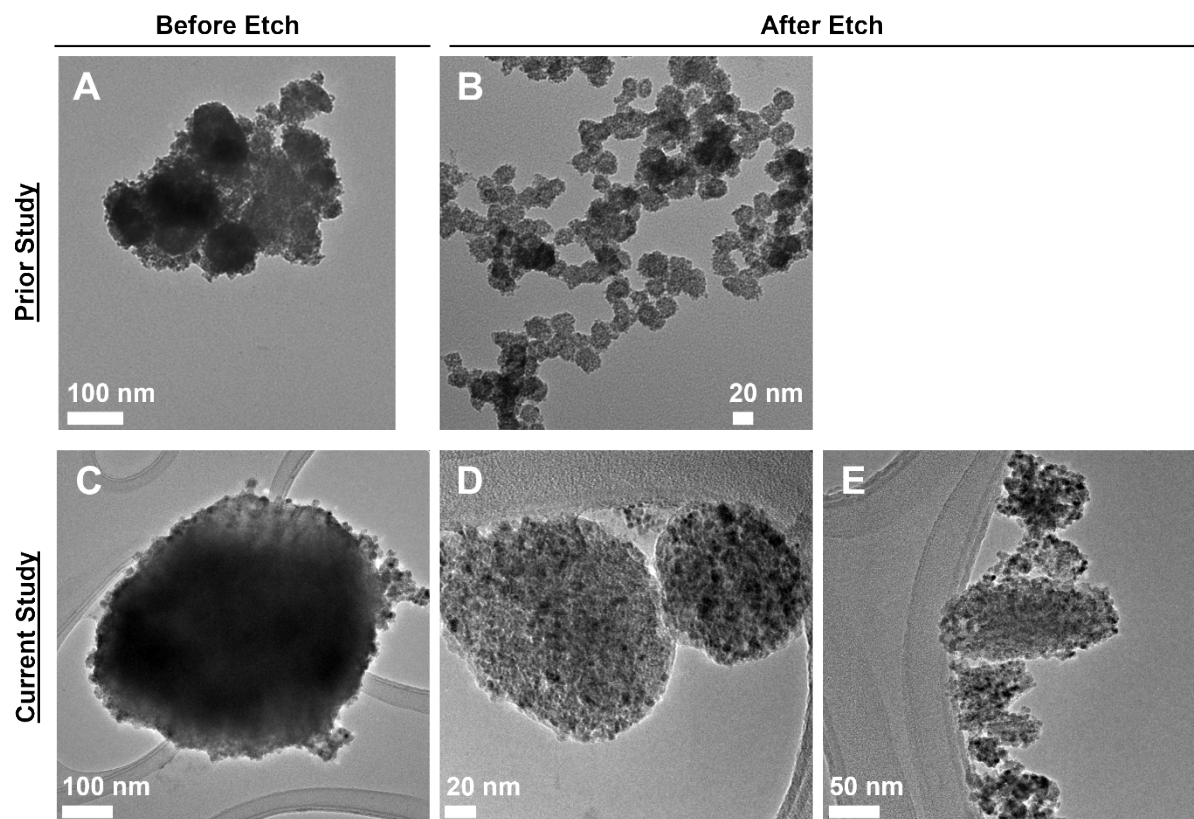


Figure S2 Comparison between Nanoparticles from Prior and Current Synthesis Methods before and after Etching. TEM images of AgCNP1 (**A,B**) and *Formulation 0.2* (**C-E**) before (**A,C**) and after (**B,D,E**) etching in concentrated ammonium hydroxide. Following etching, the surface of particles appear to be decorated with small silver nanophases. Dissolution of larger silver phases may occur in *Formulation 0.2* (**C**), as seen for AgCNP1 (**A**), and potentially lead to re-deposition of silver at the particle surface from diamine complexes.

A

| Sample Name | SOD Activity (%) | | |
|-----------------|--------------------|--------------------|-------------------|
| | 0.08 μM | 0.83 μM | 8.3 μM |
| Formulation 0.2 | 5.55+/-9.63 | 22.2+/-9.62 | 98.6+/-0.25 |
| Formulation 0.5 | 0.00+/-0.00 | 33.3+/-0.01 | 99.6+/-0.25 |
| Formulation 1.2 | 0.00+/-0.00 | 72.2+/-9.62 | 99.3+/-0.19 |

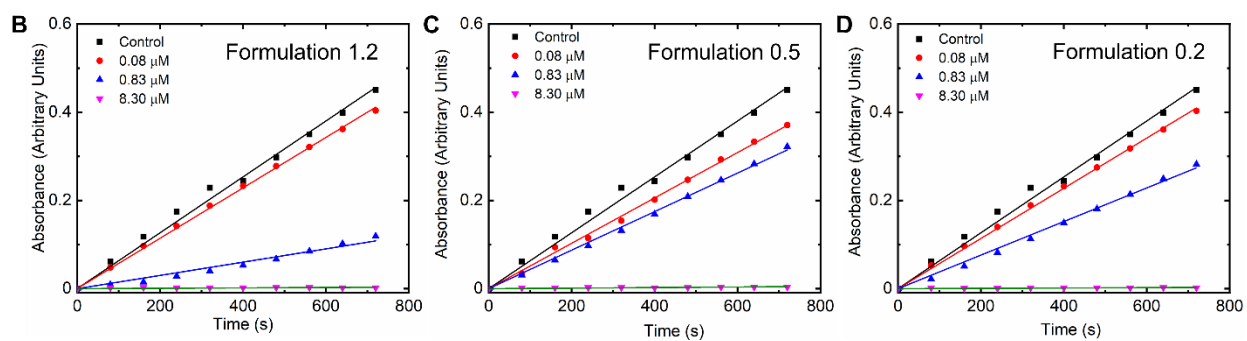


Figure S3 Superoxide Dismutase-mimetic assay. Determined percent SOD activity data shows a positive correlation with nanoparticle concentration and silver content.