# Supporting Information

# **Tough and Sustainable Solid-Solid Phase Change Materials**

# Achieved via Reversible Crosslinking for Thermal Management

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### **1. Experimental section**

#### **Materials**

PEG (*Mn*=4000), Pyromellitic dianhydride (PMDA), and 2, 4, 6-Tris (dimethylaminomethyl) phenol (DMP-30, 95%) were provided by Shanghai Macklin ESO chemical Co., Ltd. (Shanghai, China). EP (bisphenol, E51) is provided by Nan Ya Epoxy Resin (Kunshan) Co. Ltd. Hydroxy silicone oil (Mn=2000) was purchased from Silicone Oil House (Shenzhen) Chemical Products Centre (Shenzhen, China). Boron nitride (BN, 50  $\mu$ m) was purchased from Anhui Zhong Hang Nano Technology Development Co., Ltd. (Anhui, China).  $\gamma$  - (2, 3-epoxypropoxy) propytrimethosysilane (KH-560) was provided by Kangjin New Material Technology Co. Ltd. (Guangdong, China).

#### **Synthesis**

*Preparation of PEG/PMDA:* The preparation procedure for PEG/PMDA resembles our previous work. As previously reported[1], the dried PEG is positioned in a two-necked flask and heated to  $120 \,^{\circ}$  until it fully melts. Subsequently, the appropriate mass of PMDA, calculated based on a mass ratio of PEG to PMDA of 10:1, is introduced and allowed to react for 2 hours. Afterward, the mixture is poured into a PTFE mold to cool.

*Preparation of* epoxy terminated silicone *EPTS*: In a flask, 70 grams of hydroxyl silicone oil (with a molecular weight of 2000) and 11 grams of KH-560 were mixed and subjected to a reaction at 90  $^{\circ}$ C for a duration of 6 hours. Following this, a reduced-pressure distillation apparatus was attached to eliminate the volatile small molecules,

resulting in the production of EPTS.

*Preparation of SSPCMs:* The prepared EPTS was blended with EP in the appropriate ratio. Next, PEG/PMDA were melted and mixed with the EPTS/EP blend at a ratio where the carboxylic acid groups to epoxy groups was 1:1. A specific amount of the catalyst, 2, 4, 6-Tris (dimethylaminomethyl) phenol (DMP-30, 95%), was then added and stirred thoroughly until a homogeneous mixture was achieved. The ratio of EPTS to EP was displayed in Table S1. The sample was subsequently cured at 150 °C for 5 hours and named PEG/PMDA/EPTS/EP. To create thermally conductive phase change composites, boron nitride (BN) was incorporated, and the final composite was designated as PEG/PMDA/EPTS/EP/BNXX, where "XX" indicates the mass fraction of BN (as detailed in Table S3).

#### **Characterization**

The microstructure of the composites was determined by using Hitachi Regulus 8100 (Japan) microscope operating under vacuum with a 10 kV voltage. DSC curves were registered on a differential scanning calorimeter (TA Q20) at a selected scanning rate. All the samples were heated from  $-30 \,^{\circ}$ C to 120  $^{\circ}$ C at a scanning rate of 5  $^{\circ}$ C min<sup>-1</sup> to eliminate the thermal history. XRD graph lines were surveyed with a Bruker D8 diffractometer using filtered Cu-K $\alpha$  radiation and scanned in the range of 5-60  $^{\circ}$ t a generator voltage of 40 kV. The temperature-dependent XRD graph lines surveyed with a Bruker D8 diffractometer equipped with a variable-temperature system using filtered Cu-K $\alpha$  radiation and scanned in the range of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV. The temperature of 5-60  $^{\circ}$ at a generator voltage of 40 kV.

thermogravimetric analyzer (TA Q50) in an inert atmosphere (nitrogen) with a heating rate of  $10 \,^{\circ}$  min<sup>-1</sup>. Thermal conductivities of the composites were measured by a thermal constant analyzer (Hotdisk TPS2500S) and all samples had dimensions of 30  $\times 30 \times 7$  mm. The surface morphology of the specimens was observed using a stereomicroscope (AOSVI, T2-3M180). The temperature distribution of the specimen was recorded using a MAG32 infrared thermography camera (Magnity Electronics) for the purpose of visualizing the thermal management of the specimen. The stress–strain property of the specimens were measured by using a universal testing machine. The stress-strain curves were tested by using a dumbbell-type specimen with a width of 4 mm and a thickness of 2 mm. The chemical structure of the materials is confirmed by NMR spectrometer (JEOL JNM-ECZ600R/S1, Japan).



**Figure S1** (a) FTIR spectra of PEG, PMDA and PEG/PMDA, (b) XRD patterns of PEG, PMDA and PEG/PMDA, (c) <sup>1</sup>H NMR of PEG/PMDA, (d) DSC curves of PEG,

PEG/PMDA and PEG/PMDA/EPTS/EP.



Figure S2 <sup>1</sup>H NMR of EPTS.

Table S1 The formula for SSPCMs with varying EPTS content

	PEG/PMDA (mass)	EPTS (mass)	EP (mass)	EPTS:EP
PEG/PMDA/EPTS/EP -1	20	1.67	1.67	1:1
PEG/PMDA/EPTS/EP -2	20	0.9	1.8	1:2
PEG/PMDA/EPTS/EP-4	20	0.47	1.9	1:4
PEG/PMDA/EPTS/EP	20	0.24	1.95	1:8

 $\begin{tabular}{ll} Table S2 \ Phase transition enthalpy and temperature of SSPCMs with varying EPTS \end{tabular}$ 

Sample	$T_c$ (°C)	$\Delta H_c$ (J/g)	$T_m$ (°C)	$\Delta H_m$ (J/g)
PEG	36.28	211.6	62.47	234.4
PEG/PMDA	28.42	135.1	49.57	137.0
PEG/PMDA/EPTS/EP-1	27.42	80.38	48.15	77.87
PEG/PMDA/EPTS/EP-2	24.06	83.15	49.02	78.44
PEG/PMDA/EPTS/EP-4	30.06	88.16	46.35	87.25
PEG/PMDA/EPTS/EP	27.33	92.01	45.20	90.03



Figure S3 Stress-strain curve of the product obtained from the reaction between EP



and PEG/PMDA.

Figure S4 the tensile strength of SSPCMs reported in literatures ([2-13])



Figure S5 Thermally induced flexibility and self-healing of (a) PEG/PMDA/EPTS/EP-1,

(b) PEG/PMDA/EPTS/EP-2 and (c) PEG/PMDA/EPTS/EP-4.



Figure S6 Temperature-dependent XRD spectra of PEG/PMDA/EPTS/EP at different

temperatures.



Figure S7 Phase transition of pure PEG and PMDA/PEG upon heating at 80 °C.



Figure S8 (a) TG and (b) DTG curves of PEG, PEG/PMDA, PEG/PMDA/EPTS/EP

## and PEG/PMDA/EPTS/EP/BN.

	PEG/PMDA	EPTS	EP	BN
	(mass)	(mass)	(mass)	(%)
PEG/PMDA/EPTS/EP	20	0.24	1.95	0
PEG/PMDA/EPTS/EP/BN20	20	0.24	1.95	20
PEG/PMDA/EPTS/EP/BN30	20	0.24	1.95	30
PEG/PMDA/EPTS/EP/BN40	20	0.24	1.95	40

## Table S3 PEG/PMDA/EPTS/EP composites filled with BN

Table S4 Phase transition enthalpy and temperature of SSPCMs composites

Sample	$T_c$ (°C)	$\Delta H_c$ (J/g)	$T_m$ (°C)	$\Delta H_m$ (J/g)
PEG/PMDA/EPTS/EP	27.33	92.01	45.20	90.03
PEG/PMDA/EPTS/EP/BN20	28.90	72.90	46.62	71.83
PEG/PMDA/EPTS/EP/BN30	30.06	59.36	45.25	56.36
PEG/PMDA/EPTS/EP/BN40	29.29	44.64	45.20	39.92

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