

Supporting Information

Supporting Information for *In Situ* SERS and *In Situ* Raman: Interfacial Phenomena and Processes Deciphering

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Supporting Information

To facilitate the reference of researchers, we have compiled an appendix on the Raman peaks and their corresponding behaviors under *in situ* systems. Within the *in situ* detection system, by correlating the Raman shift of the spectral peak to the ranges in our SI, researchers can identify vibrations relevant to their study system. Alternatively, if the species are known, researchers can search for the species name in the SI to obtain reference Raman shifts. However, it's important to note that the sequence of peak significance in the longitudinal order does not directly correlate with the transverse order found in the references. And this table should be used as a general guide rather than a definitive resource.

CONTENTS FOR SUPPORTING INFORMATION

Raman Shift (cm ⁻¹)	Page	Raman Shift (cm ⁻¹)	Page
50 - 149	2	675 - 765	11
150 - 243	3	770 - 851	12
249 - 334	4	860 - 979	13
336 - 418	5	980 - 1050	14
420 - 468	6	1051 - 1188	15
469 - 501	7	1200 - 1370	16
502 - 549	8	1375 - 1581	17
550 - 604	9	1581.2 - 1668	18
605 - 673	10	1669 - 3000	19
		3004 - 3600	19

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	50.0 cm ⁻¹ : ¹
1	Raman active vibrational bands of iodine species (begin)
	60.0 cm ⁻¹ : ²
1	Orthorhombic BiF ₃ (o-BiF ₃)
	62.0 cm ⁻¹ : ²
1	Metallic Bi
	72.0 cm ⁻¹ : ³
1	NiTe ₂
	76.0 cm ⁻¹ : ²
1	o-BiF ₃
	88.0 cm ⁻¹ : ²
1	Metallic Bi
	99.0 cm ⁻¹ : ⁴
1	E ₂ (low) of ZnO ₁₀₀
	110.0 cm ⁻¹ : ^{1,5-7}
1	I ₃ ⁻
2	Raman active vibrational bands of I ₃ ⁻ (begin)
3	The symmetric stretching mode of I ₃ ⁻
4	Vibration mode of I ₃ ⁻
	115.0 cm ⁻¹ : ²
1	o-BiF ₃
	120.0 cm ⁻¹ : ¹
1	Raman active vibrational bands of I ₃ ⁻ (end)
	123.0 cm ⁻¹ : ⁸
1	AlSeCl
	133.0 cm ⁻¹ : ⁸
1	Al ₂ Se ₃
	135.0 cm ⁻¹ : ^{9,10}
1	NiSe ₂
2	The interaction between S ²⁻ and nickel
	142.0 cm ⁻¹ : ⁸
1	AlSeCl
	143.0 cm ⁻¹ : ⁸
1	E ₁ mode of trigonal-phase Se
	145.0 cm ⁻¹ : ^{11,12}
1	The E _g mode of anatase TiO ₂
	146.0 cm ⁻¹ : ¹³
1	Surface Cu ₂ O (Cu ₂ O _{surf})
	148.0 cm ⁻¹ : ¹⁰
1	Elemental Sulfur (S ₈)
	149.0 cm ⁻¹ : ¹⁴
1	S ₈ ²⁻ peak of graphene/S

Continued on next page

	150.0 cm ⁻¹ : ^{2,3,15-17}
1	cubic BiF ₃ (c-BiF ₃)
2	Bending or torsional vibrational modes of S-S bonds in elemental sulfur (S ₈) (begin)
3	Raman peak of Li ² S ₈
4	α-S ₈
5	Vibration mode of S ₈ ²⁻
	151.0 cm ⁻¹ : ¹⁸
1	Symmetric bond stretching of S ₈
	152.0 cm ⁻¹ : ¹¹
1	The E _g mode of anatase TiO ₂ (More defects)
	155.0 cm ⁻¹ : ^{8,19,20}
1	Symmetric bond stretching mode of S ₈ ²⁻
2	
3	TA (transverse acoustic) of graphite electrodes coated with Si layers of different thicknesses
	156.0 cm ⁻¹ : ⁸
1	Al ₂ Se ₃
	160.0 cm ⁻¹ : ⁵
1	I ₅ ⁻
	161.0 cm ⁻¹ : ²¹
1	AlF ₃ crystal
2	The interaction between S ²⁻ and nickel
	163.0 cm ⁻¹ : ⁶
1	Symmetric stretching mode of I ₅ ⁻
	165.0 cm ⁻¹ : ⁷
1	Vibration mode of I ₅ ⁻
	174.0 cm ⁻¹ : ²²
1	Reactive chloride ion species formed by electrochemical CER on Ru/RuO ₂ electrode in Cl electrolyte
	185.0 cm ⁻¹ : ⁸
1	AlSeCl
	187.0 cm ⁻¹ : ²³
1	A _g mode of monoclinic ZrO ₂
	192.0 cm ⁻¹ : ²⁴
1	F1 2g mode of spinel Co ₃ O ₄
	194.0 cm ⁻¹ : ¹⁰
1	The vibration of Ni-S in rhombohedral Ni ₃ S ₂
	199.0 cm ⁻¹ : ¹⁶
1	S ₄ ²⁻
	200.0 cm ⁻¹ : ^{1,15,18,25,26}
1	Bending or torsional vibrational modes of S-S bonds in elemental sulfur (S ₈) (end)
2	Vibration of Ni-S Bond in Ni ₃ S ₂
3	Vibration mode of Li ₂ S ₂
4	Raman active vibrational bands of I ₃ ⁻
5	F _{2g} modes of AO ₄ units in AB ₂ O ₄ spinel structure

Continued on next page

200.8 cm ⁻¹ : ¹⁶	
1	S ₆ ²⁻ /S ₄ ²⁻ Species
202.0 cm ⁻¹ : ¹⁴	
1	Bending mode of S ₄ ²⁻
206.0 cm ⁻¹ : ⁸	
1	Al ₂ Se ₃
210.0 cm ⁻¹ : ²⁷	
1	Solvated Mg-Cl cation species
212.0 cm ⁻¹ : ²⁸	
1	A _g vibrational modes of Se-Se bonds in selenides
215.0 cm ⁻¹ : ¹⁶	
1	α-S ₈
216.0 cm ⁻¹ : ^{10,29}	
1	A _g mode of Se-Se stretching in Ru-NiSe ₂
2	Vibration of Ni-S in rhombohedral Ni ₃ S ₂
217.0 cm ⁻¹ : ²⁵	
1	Vibration of Ni-S Bond in Ni ₃ S ₂
218.0 cm ⁻¹ : ^{3,10,15}	
2	Elemental sulfur (S ₈)
3	Raman peaks of Li ₂ S ₈
219.0 cm ⁻¹ : ^{14,17-19,30}	
1	Bending of symmetric bonds in S ₈
2	Cu ₂ O
3	Symmetric bond stretching mode of S ₈ ²⁻
4	S ₈ ²⁻ peak of graphene/S
5	Vibration mode of S ₈ ²⁻
223.0 cm ⁻¹ : ¹⁵	
1	Hematite, Fe ₂ O ₃
225.0 cm ⁻¹ : ^{8,12}	
1	Solid electrolyte interface (SEI) layer of aluminum chlorochalcogenide
2	The E _g vibration mode of anatase TiO ₂
235.0 cm ⁻¹ : ⁸	
1	A ₁ mode of trigonal-phase Se
236.6 cm ⁻¹ : ³¹	
1	The E _g phonon mode of Ni
240.0 cm ⁻¹ : ²⁸	
1	T _g vibrational modes of Se-Se bonds in selenides
242.0 cm ⁻¹ : ²⁵	
1	Fe-O bond of α-FeOOH
243.0 cm ⁻¹ : ^{32,33}	
1	NO ₂ species
2	Longitudinal mode of surface oxygen
3	S-Ag stretching vibration between H ₂ SO ₄ component and Ag

Continued on next page

249.0 cm ⁻¹ : ³⁴	
1	Lepidocrocite (γ-FeOOH)
250.0 cm ⁻¹ : ^{9,21,35}	
1	NiSe ₂
2	The characteristic peak of Co(OH) ₂
3	Al-F bonds in Na ₅ Al ₃ F ₁₄ crystal (vs)
4	Terminal SS bond (begin)
251.0 cm ⁻¹ : ³⁶	
1	The second-order effect (SOE) of anatase TiO ₂
256.0 cm ⁻¹ : ⁸	
1	Amorphous Se
258.0 cm ⁻¹ : ³⁷	
1	Second order transverse acoustic mode of CeO ₂ lattice (2TA)
259.0 cm ⁻¹ : ³⁸	
1	Bending vibration of O-W-O
260.0 cm ⁻¹ : ³⁹	
1	E _g mode for Co(OH) ₂
262.0 cm ⁻¹ : ⁴⁰	
1	The second-order transverse acoustic mode of FCC fluorite phase of CeO ₂
268.0 cm ⁻¹ : ⁴¹	
1	The E _g mode of Co-O bonds
269.0 cm ⁻¹ : ²³	
1	The E _g mode of Co-O bonds
273.0 cm ⁻¹ : ⁴²	
1	CuO
281.0 cm ⁻¹ : ⁴³	
1	A _g translation mode
2	Symmetric stretching mode of FeS (begin)
282.0 cm ⁻¹ : ¹⁵	
1	Amorphous Mackinawite/Mackinawite
285.0 cm ⁻¹ : ⁴⁴	
1	A _g mode of CuO
288.94 cm ⁻¹ : ⁴⁵	
1	The E _g mode of metal sulfur bonds in Co(Zn)S ₂ /CC
289.0 cm ⁻¹ : ^{15,46}	
1	CuO
2	Hematite
290.0 cm ⁻¹ : ⁴⁷	
1	E-type vibrations of h-Co _{0.34} Fe _{0.33} Ni _{0.33} -LDH
292.0 cm ⁻¹ : ⁴⁸	
1	Frustrated rotation of Cu-CO
296.0 cm ⁻¹ : ^{13,49}	
1	
2	Crystalline CuO nanoparticles

Continued on next page

	298.0 cm ⁻¹ ; 15,16,44
1	Ferrites
2	S ₆ ²⁻
3	S ₄ ²⁻
4	Restricted rotation of adsorbed CO
5	Symmetric stretching mode of FeS (end)
	300.0 cm ⁻¹ ; 10,17
1	Vibration of Ni-S in rhombohedral Ni ₃ S ₂
2	The vibration mode of mid-length LiPS
	303.0 cm ⁻¹ ; 13,30
1	CuC stretching and rotation mode of adsorbed CO (end)
	304.0 cm ⁻¹ ; 25
1	Vibration of Ni-S Bond
	305.0 cm ⁻¹ ; 23
1	δInO ₆ structural units (δInO ₆)
	306.0 cm ⁻¹ ; 38
1	δ(W[=O] ₂) of crystalline Na ₂ WO ₄ phase
	310.0 cm ⁻¹ ; 20,50
1	Adsorbed phosphate ions
2	LA (longitudinal acoustics) of graphite electrodes coated with Si layers of different thicknesses
	312.66 cm ⁻¹ ; 51
1	A _{1g} peak of SnS ₂
	319.0 cm ⁻¹ ; 38
1	Bending vibration of O-W-O
	320.0 cm ⁻¹ ; 50
1	Adsorbed phosphate ions
	321.0 cm ⁻¹ ; 21,22,52
1	Reactive chloride ion species formed by electrochemical CER on Ru/RuO ₂ electrode in Cl ⁻ electrolyte
2	Ni-LDH
3	Symmetric stretching vibration B _{2g} mode of Al-F bonds in Na ₅ Al ₃ F ₁₄ crystal (ν _s)
	324.0 cm ⁻¹ ; 19,53
1	Asymmetric deformation modes of VO ₄ ³⁻
2	Li ₂ S ₆
	325.0 cm ⁻¹ ; 10
1	The interaction between S ²⁻ and nickel
	326.0 cm ⁻¹ ; 18
1	Vibration mode of Li ₂ S ₆
	330.0 cm ⁻¹ ; 4
1	E ₂ (High) - E ₂ (low) of ZnO ₁₀₀
	331.0 cm ⁻¹ ; 30
1	CuO
	334.0 cm ⁻¹ ; 23
1	Monoclinic ZrO ₂

Continued on next page

	336.0 cm ⁻¹ ; 43
1	E _{2g} mode of La(OH) ₃
	339.0 cm ⁻¹ ; 21
1	B _g mode of Al-F bonds in Na ₃ AlF ₆ (Calculation)
	342.0 cm ⁻¹ ; 10
1	Vibrations of Ni-S in rhombohedral Ni ₃ S ₂
	343.0 cm ⁻¹ ; 42
1	Ni oxide NiO
	345.0 cm ⁻¹ ; 23
1	Stretching vibrations of the InOIn structures
	346.0 cm ⁻¹ ; 22
1	Reactive chloride ion species formed by electrochemical CER on Ru/RuO ₂ electrode in Cl ⁻ electrolyte
	351.0 cm ⁻¹ ; 25,54
1	Vibration of Ni-S Bond in Ni ₃ S ₂
2	Mn-OH is deuterated to Mn-OD
	352.0 cm ⁻¹ ; 44
1	Restricted rotation of Cu-CO stretching
	355.0 cm ⁻¹ ; 13
1	CuC stretching and rotation mode of adsorbed CO
	359.0 cm ⁻¹ ; 21
1	Symmetric stretching vibration E _g mode of Al-F bonds in Na ₅ Al ₃ F ₁₄ crystal (ν _s)
	360.0 cm ⁻¹ ; 54
1	MnOH is deuterated to MnOD
	360.51 cm ⁻¹ ; 51
1	A _g ¹ peak of black phosphorus (BP)
	361.0 cm ⁻¹ ; 21
1	Ag mode of Al-F bonds in Na ₃ AlF ₆ (Calculation)
	363.0 cm ⁻¹ ; 35
1	The characteristic peaks of CoOOH
	365.0 cm ⁻¹ ; 55
1	Cu-CO stretching vibration
	367.0 cm ⁻¹ ; 48
1	Stretch of Cu-CO
	374.0 cm ⁻¹ ; 56,57
1	Peak of MoO _{3-x} (0 < x < 1)
2	Formation of Na ₂ O(UO ₃ ·H ₂ O) x precipitation
	375.0 cm ⁻¹ ; 58
1	Active CoOOH intermediate species
	376.0 cm ⁻¹ ; 59
1	Terminal Mn-O (H) tunnel or surface expansion
	378.0 cm ⁻¹ ; 34
1	Lepidocrocite (γ-FeOOH)

Continued on next page

	380.0 cm ⁻¹ : ²³
1	Monoclinic ZrO ₂
	383.0 cm ⁻¹ : ⁶⁰
1	SrTiO ₃ unit cell via La and Co doping (STLC)
	384.0 cm ⁻¹ : ⁴⁹
1	Anatase vibration of TiO ₂
	386.0 cm ⁻¹ : ^{21,25}
1	Fe-O bond of α -FeOOH
2	AlF ₃ crystal
	390.0 cm ⁻¹ : ⁴⁵
1	A _g mode of metal-S in Co(Zn)S ₂ /CC
	398.0 cm ⁻¹ : ^{3,14}
1	Long chain LiPSs (Li ₂ S ₆)
2	Characteristic Raman peaks of long-chain Li ₂ S ₆
	399.0 cm ⁻¹ : ^{12,61}
1	Ag-O vibration
2	B _{1g} vibration mode of anatase TiO ₂
	399.5 cm ⁻¹ : ¹⁶
1	S ₆ ²⁻ /S ₄ ²⁻ Species
	400.0 cm ⁻¹ : ^{15,17,20,62,63}
1	corundum
2	H ₂ SO ₄ electrolyte
3	LO (longitudinal optics) of graphite electrodes coated with Si layers of different thicknesses
4	Vibration mode of short chain LiPS
5	Bending vibration of C-S bonds
6	Stretching vibration of S ₈ (begin)
	401.9 cm ⁻¹ : ¹⁸
1	Vibration mode of Li ₂ S ₆
	404.0 cm ⁻¹ : ¹⁵
1	Hematite
	405.0 cm ⁻¹ : ⁶⁴
1	Vibration of Au-O bonds
	409.0 cm ⁻¹ : ⁶⁵
1	NiO
	410.0 cm ⁻¹ : ⁶⁶
1	Characteristic Raman peaks of α -FeOOH phase
	412.0 cm ⁻¹ : ¹³
1	Cu ₂ O (Cu ₂ O _{surf})
	415.0 cm ⁻¹ : ^{20,67,68}
1	The vibration mode of Cu ₂ O
2	S _n ²⁻
3	Raman peaks of amorphous silicon (a-Si)
	416.0 cm ⁻¹ : ^{19,21}
2	Li ₂ S ₄
	418.0 cm ⁻¹ : ⁶⁹
1	Lattice vibrations of Mo-O bonds

Continued on next page

	420.0 cm ⁻¹ : ^{58,70-72}
1	Bending of symmetric oxygen cages
2	F _{2g} symmetric vibration of CeO ₂
3	Metal adsorbate stretching vibration of linear CO bonds (AuNi COL)
4	Network bending mode of SiO ₂
	422.0 cm ⁻¹ : ³⁶
1	The E _g vibration mode of anatase TiO ₂
	426.0 cm ⁻¹ : ⁷³
1	The vibration mode of Co-O
	431.39 cm ⁻¹ : ⁵¹
1	B _{2g} peak of BP
	432.0 cm ⁻¹ : ⁷⁴
1	The vibration mode of RuO ₂
	437.0 cm ⁻¹ : ⁷⁵
1	E ₂ high mode of ZnO
	439.0 cm ⁻¹ : ⁴
1	E ₂ (High) of ZnO ₁₀₀
	440.0 cm ⁻¹ : ⁶¹
1	Bulk oxygen
	441.0 cm ⁻¹ : ⁷⁶
1	Au-Bi ₂ O ₃ -3
	443.0 cm ⁻¹ : ⁶⁸
1	S _n ²⁻
2	S ₂ O ₃ ²⁻
	445.0 cm ⁻¹ : ^{14,76-78}
1	Au-Bi ₂ O ₃ -3
2	β -Ni(OH) ₂
3	Na ₃ Al ₃ CF ₈
4	Bending mode of S ₄ ²⁻
	446.0 cm ⁻¹ : ^{12,18,79}
1	The E _g bending vibration mode of MO bonds in MOH species
2	Vibration mode of Li ₂ S ₄
3	B _g vibration mode of anatase TiO ₂
	450.0 cm ⁻¹ : ^{47,76,80-83}
1	h-Co _{0.34} Fe _{0.33} Ni _{0.33} -LDH
2	Ni (II) - OH
3	Bi ₂ O ₃
4	Bonding between Mo atom and top S atom (v (Mo ₃ -S))
5	CoII-OH vibration of Co(OH) ₂
6	MII-OH of Co and Fe
7	ClO ₄ ⁻ ions in polycarbonate solution
	451.0 cm ⁻¹ : ^{35,76,84}
1	The Ni-O vibration peak of NiFeLDH
2	Bi ₂ O ₃
3	Asymmetric breathing mode of oxygen atoms surrounding cerium ions in CeO ₂
	452.0 cm ⁻¹ : ^{18,25,85}
1	F _{2g} symmetric vibration (Ce-O-Ce stretching)
2	Ni-O bond stretching vibration
3	Vibration mode of Li ₂ S ₂

Continued on next page

	453.0 cm ⁻¹ : ^{52,85}
1	Ce-O bond
2	Ni-LDH
	455.9 cm ⁻¹ : ¹⁶
1	S ₆ ²⁻ /S ₄₂ -Species
	456.0 cm ⁻¹ : ⁷⁹
1	The stretching vibration of M-O bonds
	457.0 cm ⁻¹ : ^{86,87}
1	NiFe-LDH@NF Ni-O vibration in Ni(OH) ₂ phase on the surface
2	Bending and stretching vibration of Ni-O in NiOOH
	457.97 cm ⁻¹ : ⁵¹
1	A _g ² peak of BP
	458.0 cm ⁻¹ : ⁸⁵
1	Symmetric tensile vibration of Ce-O
	460.0 cm ⁻¹ : ^{41,77,88-91}
1	E _g -CO ²⁺ -OH
2	Triple Degeneracy Mode of F _{2g} Mode in CeO ₂
3	O-Co-O bending mode
4	O-Co-O
5	α-Ni(OH) ₂
6	Perchlorate electrolyte
7	The E _g vibrational modes of Co-O species
	460.9 cm ⁻¹ : ⁹²
1	CoO ₂
	461.0 cm ⁻¹ : ³
1	Long chain LiPSs (Li ₂ S ₅ + Li ₂ S ₄)
	462.0 cm ⁻¹ : ³⁷
1	Vibration model of octahedral local symmetry from the CeO ₂ lattice (F _{2g})
	463.0 cm ⁻¹ : ^{35,93}
1	The characteristic peaks of CoOOH
2	Ni-O vibration of defective NiCoLDH nanocrystals
	464.0 cm ⁻¹ : ⁴⁹
1	Vibration of CeO ₂ nanoparticles
	465.0 cm ⁻¹ : ^{40,80}
1	Ni-O bond
2	The first-order F _{2g} symmetric vibration of CeO ₂
	465.5 cm ⁻¹ : ³¹
1	A _g phonon modes of Ni
	467.0 cm ⁻¹ : ⁹⁴
1	The peak of β-Ni(OH) ₂ phase
	467.6 cm ⁻¹ : ⁴⁶
1	Cu ₂ O
	468.0 cm ⁻¹ : ⁹⁵
1	Ni-O vibration in γ-NiOOH

Continued on next page

	469.0 cm ⁻¹ : ^{28,93}
1	High valence nickel iron hydroxide (NiFe)OOH
2	Characteristic peaks of NiCoOOH
	470.0 cm ⁻¹ : ^{19,20}
1	Li ₂ S ₆ /Li ₂ S ₄ mixture
2	Vibration modes of TO (transverse optic) in graphite electrodes coated with Si layers of different thicknesses
	472.0 cm ⁻¹ : ^{3,10,96}
1	Elemental sulfur (S ₈)
2	The E _g band of LiNiO ₂ (LNO)
3	Raman peaks of Li ₂ S ₈
	473.0 cm ⁻¹ : ^{25,62,80,84,97}
1	C-C _{ring} vibration of 4-AP and 4-NP(4-nitrophenol)
2	Active NiOOH species (γ-NiOOH)
3	Metal (M) O Vibration in Metal OOH
4	γ-NiOOH
5	E _g Ni-O bending vibration mode in NiOOH
	474.0 cm ⁻¹ : ^{18,29,98}
1	NiOOH
2	F _{2g} vibration mode of NiCo ₂ O ₄
3	Anti-symmetric bond bending of S ₈
	475.0 cm ⁻¹ : ^{9,10,16,25,43,70}
1	Depolarized E _g mode of γ-NiOOH phase (bending vibration of oxygen atoms on a plane) of γ-NiOOH phase
2	NiOOH
3	Ni-O vibration in NiOOH
4	The E _g , Ni-O bending vibration modes of γ-NiOOH
5	(Co/Fe) O (OH) structure
	476.0 cm ⁻¹ : ^{35,86,99}
1	Ni-OOH vibration
2	Ni-O vibration of NiOOH
3	The characteristic peak of CoO ₂
	477.0 cm ⁻¹ : ^{14,89,100}
1	Ni-O bending vibration of γ-NiOOH phase
2	The E _g band of NiOOH
3	S ₈ ²⁻ peak of graphene/S
	478.0 cm ⁻¹ : ^{17,19,101}
1	The E _g bending mode of γ-NiOOH
2	Symmetric bond stretching mode of S ₈ ²⁻
3	Vibration mode of S ₈ ²⁻
	479.0 cm ⁻¹ : ⁶⁵
1	E _g bending vibration in NiOOH
	480.0 cm ⁻¹ : ^{35,63,66,88,91,102-104}
1	Formation of CoOOH
2	E _g Ni ²⁺ -O bending
3	Ni(III)-O bending (δ) in NiOOH
4	Polarized A _{1g} mode (stretching) of Ni ^{III} -O in NiOOH
5	Characteristic Raman peaks of γ-NiOOH phase
6	The characteristic peak of CoO ₂
7	The stretching vibration of S-S bonds
8	The E _g vibrational modes of Co-O species

Continued on next page

	481.0 cm ⁻¹ : ¹⁸
1	Depolarization of γ -NiOOH in E _g mode (Bending)
	482.0 cm ⁻¹ : ¹⁰⁵
1	Typical bending δ (NiIII-O)
	483.0 cm ⁻¹ : ⁸⁷
1	Vibration of Co(OH) x
	485.0 cm ⁻¹ : ^{24,30,96}
1	Co(OH) ₂ -phase
2	The E _g mode of spinel Co ₃ O ₄
3	The E _g band of LNO
4	NiOOH species
	487.0 cm ⁻¹ : ^{30,106}
1	Bending of Ni ³⁺ -O
2	NiOOH species
3	Tri- and tetra-cyclohexane rings of SiO ₂
	488.0 cm ⁻¹ : ^{58,72}
1	F _{2g} symmetric vibration mode of CeO ₂
2	D1 and D2 defect modes of tri- and tetra-cyclotrisiloxane in SiO ₂
	489.0 cm ⁻¹ : ^{35,57}
1	The characteristic peaks of CoOOH
2	Adsorbed uranyl
	494.0 cm ⁻¹ : ¹⁰⁷
1	Characteristic of vibrational modes directly involving Fe species in MOFs catalyst
	494.84 cm ⁻¹ : ⁴⁵
1	The stretching vibration of Co-O bonds
	495.0 cm ⁻¹ : ^{45,100}
1	Disordered Ni-OH
2	Co(Zn)OOH
	497.0 cm ⁻¹ : ¹⁰⁸
1	V-O-Ag bonds with different coordination structures in Ag _{1.2} V ₃ O ₈
	497.8 cm ⁻¹ : ⁹²
1	CoOOH species
	498.0 cm ⁻¹ : ³⁰
1	Cu(OH) ₂
	500.0 cm ⁻¹ : ^{22,24,54,78,80,83,91,109,110}
1	CoOOH
2	Terminal SS bond (end)
3	Defective NiOOH or Ni(OH) ₂
4	Interplanar vibration of adjacent MnO ₂ nanosheets
5	The E _g mode of nanocrystalline RuO ₂
6	The bending mode of Co-O-Co
7	Raman spectra of molten NaF-AlF ₃ -Al ₄ C ₃ salts
8	Uncoordinated Si atoms near the surface of poly-c Si thin films
9	The E _g vibrational modes of Co-OOH species
	501.0 cm ⁻¹ : ³⁵
1	The characteristic peaks of CoOOH

Continued on next page

	502.0 cm ⁻¹ : ²³
1	ν (InO ₆) of InO ₆ structural units
	506.0 cm ⁻¹ : ³⁵
1	The characteristic peak of Co(OH) ₂
	507.0 cm ⁻¹ : ⁵⁹
1	Out of plane symmetric stretching vibration of Mn-O in MnO ₆
	507.7 cm ⁻¹ : ¹⁸
1	Vibration mode of Li ₂ S ₃
	509.0 cm ⁻¹ : ^{14,42,58}
1	Active CoOOH intermediate species
2	NiO
3	Characteristic Raman peaks of long-chain Li ₂ S ₆
	510.0 cm ⁻¹ : ⁸¹
1	Terminal [S ₂] ²⁻ (ν (S-S) _{te})
	512.0 cm ⁻¹ : ^{12,49}
1	The vibration mode of anatase
2	A _{1g} vibration mode of anatase TiO ₂
	515.0 cm ⁻¹ : ^{35,44}
1	Characteristic peaks of Co ₃ O ₄
2	OH group
	517.0 cm ⁻¹ : ³⁰
1	The vibration of Ni(OH) ₂
	518.0 cm ⁻¹ : ^{65,74,77}
1	Fe-free and the Fe-containing electrolyte
2	β -Ni(OH) ₂
3	Ru-O characteristic peaks belonging to RuO ₂
	519.0 cm ⁻¹ : ⁷⁹
1	A _{1g} stretching vibration mode of M-O bonds in MOH species
	520.0 cm ⁻¹ : ^{20,83}
1	Crystalline peak of Si
2	Bulk c-Si
	521.0 cm ⁻¹ : ^{93,111}
1	Ni-O vibrations of defective or disordered NiCo LDH nanocrystals
2	Li Si alloy
	522.0 cm ⁻¹ : ⁸⁷
1	Co-O vibration in NCMO
	523.0 cm ⁻¹ : ^{35,112}
1	The characteristic peak of Co(OH) ₂
2	Nafion molecule
	524.0 cm ⁻¹ : ³⁵
1	The characteristic peak of Co(OH) ₂
	525.0 cm ⁻¹ : ^{13,24,31,34,67}
1	CuO _x /(OH) _y species
2	Lepidocrocite (γ -FeOOH)
3	The characteristic peaks of M(metal)-OOH
4	F _{2g} mode of spinel Co ₃ O ₄
5	The vibration mode of Cu ₂ O (begin)

Continued on next page

526.0 cm ⁻¹ : 41,47,55,88,100,101,113	
1	The vibration mode of Cu ₂ O
2	Adsorption of primary intermediates (such as CO _{2 ad}) on Cu
3	M-O vibration of h-Co _{0.34} Fe _{0.33} Ni _{0.33} -LDH
4	A _{1g} Ni ²⁺ -O stretching
5	Formation of FeOOH intermediates
6	Co-O symmetric stretching mode of Co(OH) ₂
7	Co-O
8	FeOOH
527.0 cm ⁻¹ : ⁹³	
1	Characteristic peaks of NiCoOOH
528.0 cm ⁻¹ : 13,65,95	
1	Surface Cu ₂ O (Cu ₂ O _{surf})
2	Thin, disordered NiFe-LDH
3	α-Ni(OH) ₂
530.0 cm ⁻¹ : 26,66,80,91	
1	Defective NiOOH or Ni(OH) ₂
2	Ni(II)-O
3	Typical Raman features of Ni-Fe-LDH
4	T _{2g} mode of CoO/rGO
5	A _g vibrational modes of CoOOH species (begin)
531.0 cm ⁻¹ : ²⁵	
1	Ni-O stretching vibration
532.0 cm ⁻¹ : ²¹	
1	A _g mode of Al-F bond in Na ₃ AlF ₆ (Calculation)
534.0 cm ⁻¹ : ⁶⁷	
1	The vibration mode of Cu ₂ O (end)
536.0 cm ⁻¹ : ²¹	
1	A _{1g} mode of Al-F bonds in Na ₅ Al ₃ F ₁₄ crystal(v _s)
538.0 cm ⁻¹ : ⁸⁶	
1	Ni-O vibrations in surface Ni(OH) ₂ phase of NiFe-LDH@NF
541.0 cm ⁻¹ : ⁸⁴	
1	The Ni-O vibration peak of NiFe LDH
542.0 cm ⁻¹ : ⁸¹	
1	Bridging type [S ₂] ²⁻ (ν(S-S) _{br})
543.0 cm ⁻¹ : ⁵⁸	
1	Vibration peak of Ni-O bond in NiOOH
545.0 cm ⁻¹ : ^{89,95}	
1	Characteristic peaks of Ni ₃ N
2	Ni-O vibration in γ-NiOOH
548.0 cm ⁻¹ : ⁸⁷	
1	Bending and stretching vibration of Co-O in CoOOH
549.0 cm ⁻¹ : ^{35,96}	
1	The characteristic peaks of CoOOH
2	A _{1g} band of LNO

Continued on next page

550.0 cm ⁻¹ : 10,20,24,33,70,78,80,82,91	
1	Ni-O vibration in NiOOH
2	(Co/Fe) O (OH) structure
3	Fe and Ni species, such as α-FeOOH, α-Fe ₂ O ₃ , NiO (begin)
4	Oxygen defect related band
5	A _{1g} vibration mode of Mn-O nanosheets
6	CoII-O vibration
7	Boron (B) (begin)
8	The Raman spectra of molten NaF-AlF ₃ -Al ₄ C ₃ salts
9	A _g vibrational modes of CoOOH species (end)
552.4 cm ⁻¹ : ⁶²	
1	Metal-O Vibration in Metal-OOH
553.0 cm ⁻¹ : ^{9,80,96}	
2	A _{1g} Ni-O stretching vibration mode in NiOOH
3	A _{1g} band of LNO
554.0 cm ⁻¹ : 29,98,101,114	
1	NiOOH
2	Formation of Ni(III)OOH on the surface of NiCo ₂ O ₄
3	Polarized A _{1g} mode (stretching) of Ni ^{III} -O
4	A _{1g} stretching mode of γ-NiOOH
555.0 cm ⁻¹ : 28,65,99	
1	Ni-O vibration of NiOOH
2	Polarized A _{1g} stretching mode of Ni-O(H) in NiOOH
3	High valence Nickel iron hydroxide (NiFe)OOH
556.0 cm ⁻¹ : 25,84,86	
1	Active NiOOH species (β-NiOOH)
2	Ni-OOH vibration
3	A _{1g} , Ni-O stretching vibration mode of γ-NiOOH
4	γ-NiOOH
558.0 cm ⁻¹ : 43,100	
1	Ni-O of γ-NiOOH phase
2	Polarized A _{1g} mode of γ-NiOOH (stretching vibration of oxygen atoms perpendicular to the plane)
560.0 cm ⁻¹ : 62,66,103,104	
1	Ni(III)-O stretching vibration (ν) modes in NiOOH
2	Depolarized E _g mode (bending) of NiIII-O in NiOOH
3	H ₂ SO ₄ electrolyte
4	Characteristic Raman peaks of γ-NiOOH phase
561.0 cm ⁻¹ : ³⁰	
1	NiOOH species
564.0 cm ⁻¹ : ¹⁰⁵	
1	Typical extension of NiOOH (NiIII-O)
565.0 cm ⁻¹ : ³⁰	
1	Tensile Vibration of Ni ³⁺ -O
2	NiOOH species
566.0 cm ⁻¹ : ^{59,101}	
1	In-plane stretching vibration of octahedral layers in δ - MnO ₂
2	Peak value of NiFeLDH system

Continued on next page

	568.0 cm ⁻¹ : ⁴
1	Peaks of oxygen vacancies in Pd/ZnO ₁₀₀ samples
	569.0 cm ⁻¹ : ³⁵
1	The characteristic peaks of CoOOH
	570.0 cm ⁻¹ : ²⁰
1	Formation of Li ₂ O
	571.0 cm ⁻¹ : ¹¹⁵
1	Formation of Co-OH on Co ₂ V ₂ O ₇ and Co ₃ O ₄
	573.0 cm ⁻¹ : ^{41,58}
1	Tensile vibration of A _g -Co-OH
2	Active CoOOH intermediate species
	574.0 cm ⁻¹ : ⁹⁰
1	MnO lattice vibration of the basal plane of the MnO ₂ sheets
	575.0 cm ⁻¹ : ⁵⁴
1	In-plane MnO stretching
	577.0 cm ⁻¹ : ^{90,107}
1	FeOOH
2	MnO lattice vibration of the basal plane of the MnO ₂ sheets
	579.0 cm ⁻¹ : ⁹²
1	The peak of MnO ₂
	580.0 cm ⁻¹ : ^{35,61,85}
1	Oxygen vacancies creation because of the entry of CeO ₂ into the lattice of CuO (end)
2	v(O-Ag-O) in surface and subsurface oxygen motifs
3	The characteristic peak of CoO ₂
	581.0 cm ⁻¹ : ⁵⁶
1	Peak of MoO _{3-x} (0 < x < 1)
	589.0 cm ⁻¹ : ^{4,73}
1	E _{1L} of ZnO ₁₀₀
2	The vibration mode of Ni-O in Ni-OH
	596.0 cm ⁻¹ : ⁷⁹
1	The stretching vibration of the M-O bond, suggesting the presence of a layered metal hydroxide phase
	598.0 cm ⁻¹ : ^{37,40}
1	
2	(D) mode caused by defects in CeO ₂
	600.0 cm ⁻¹ : ^{15,24,45,80,85,102,109}
1	Amorphous cobalt oxide
2	Formation of CoOOH
3	Oxygen vacancy of Fe ₂ O ₃
4	corundum
5	Fe and Ni species, such as α-FeOOH, α-Fe ₂ O ₃ , NiO (end)
6	The E _g mode of Co(OH) ₂
7	Co(Zn)OOH
	601.0 cm ⁻¹ : ³⁰
1	Cu ^{III}
	604.0 cm ⁻¹ : ³⁶
1	A _{1g} vibration mode of anatase TiO ₂

Continued on next page

	605.0 cm ⁻¹ : ^{72,106}
1	Symmetric stretching of the bridging SiOSi in SiO ₂
2	SiOSi symmetrical stretching of SiO ₂
	606.0 cm ⁻¹ : ³⁵
1	The characteristic peaks of CoOOH
	608.0 cm ⁻¹ : ²⁴
1	Co(OH) ₂ -phase
	610.0 cm ⁻¹ : ^{85,101}
1	Oxygen vacancies creation because of the entry of CeO ₂ into the lattice of CuO (begin)
2	Formation of β-FeOOH
	612.0 cm ⁻¹ : ^{35,69,97}
1	C-H bond of 4-NP and 4-AP
2	Lattice vibrations of Mo-O bonds
3	The characteristic peaks of CoOOH
4	Ce-O stretching mode
	613.0 cm ⁻¹ : ^{12,116}
1	Vibration modes of specific bonds in R6G on cavity-based particle-in-quasicavity(PIQC) ZnO
2	A _{1g} vibration mode of anatase TiO ₂
	614.0 cm ⁻¹ : ³⁵
1	The characteristic peaks of CoOOH
	614.8 cm ⁻¹ : ⁵⁴
1	MnO stretching of the MnOOH intermediate
	615.0 cm ⁻¹ : ⁷⁸
1	The Raman spectra of molten NaF-AlF ₃ -Al ₄ C ₃ salts
	618.0 cm ⁻¹ : ³²
1	SO ₃ species
2	Sag stretching
	619.0 cm ⁻¹ : ^{13,22}
1	Surface Cu ₂ O (Cu ₂ O _{surf})
2	A _{1g} mode of nanocrystalline RuO ₂
	620.0 cm ⁻¹ : ^{67,70,106}
1	The vibration mode of Cu ₂ O
2	Stretching of symmetric oxygen cages of A _{2g}
3	Surface MnO _x oligomers
4	Grafting MnOSi bonds in MnO _x /SiO ₂
	620.3 cm ⁻¹ : ⁴⁴
1	B _g mode of CuO
	621.0 cm ⁻¹ : ^{88,89}
1	F _{2g} -Co ²⁺ -OH
2	Double degeneration into longitudinal optical mode of oxide, suggesting that abundant oxygen vacancies in CeO ₂
	623.0 cm ⁻¹ : ^{49,87}
1	Co-O vibration in NCMO
2	The vibration mode of anatase
	624.0 cm ⁻¹ : ⁹²
1	The peak of MnO ₂

Continued on next page

	628.0 cm ⁻¹ : ¹¹³
1	The vibration mode of Cu ₂ O
	630.0 cm ⁻¹ : ^{59,90,114}
1	Out of plane symmetric stretching vibration of Mn-O in MnO ₆
2	Presence of oxygen vacancies on NiGe film (end)
3	Symmetric stretching vibration of MnO band in MnO ₆ octahedra
	635.0 cm ⁻¹ : ^{30,58}
1	A _{1g} vibration of NiCo ₂ O ₄
2	CuO
	636.0 cm ⁻¹ : ¹¹⁰
1	The E _g vibrational mode of Pd-O bonds
	638.0 cm ⁻¹ : ¹²
1	The E _g vibration mode of anatase TiO ₂
	643.0 cm ⁻¹ : ^{34,60}
1	Lepidocrocite (γ-FeOOH)
2	VCo/TiO stretching frequencies (along c axis) of STLC
	645.0 cm ⁻¹ : ^{20,54,58,117}
1	The out-of-plane MnO stretching
2	A _{1g} vibration of NiCo ₂ O ₄
3	second order of the LA (longitudinal acoustic) mode and the overtone of TA (transverse acoustic) and TO optical modes of a-Si
4	The vibrational modes of Pd-O bonds on Pd-based catalyst
	650.0 cm ⁻¹ : ⁹⁰
1	Mn-O bonds of mixed valence manganese
2	Mn ³⁺ /Mn ⁴⁺
	660.0 cm ⁻¹ : ^{107,114}
1	Characteristic of vibrational modes directly involving Co species in MOFs catalyst
2	Presence of oxygen vacancies on NiGe film (begin)
	663.0 cm ⁻¹ : ⁶⁸
1	S ₂ O ₃ ²⁻
	665.0 cm ⁻¹ : ⁴⁹
1	Crystalline Cu ₂ O nanoparticles
	667.0 cm ⁻¹ : ³⁵
1	Characteristic peaks of Co ₃ O ₄
	668.0 cm ⁻¹ : ¹¹⁸
1	the bending out-of-plane mode of -CH, OH, and OC=O in MOF-AgNC(nanocube)
	670.0 cm ⁻¹ : ^{82,98,119,120}
1	A _{1g} vibration mode of NiCo ₂ O ₄
2	Octahedral-coordinated CoIIIO stretch
3	CoII-O stretching vibration of Ag/iron oxide-doped NiO-CoO hollow spheres (FNCO-HSs)
4	Symmetric stretching modes of DMSO molecules
5	Guanine ring stretching vibration of single-stranded DNA
	673.0 cm ⁻¹ : ⁹⁷
1	NO ₂ bending of 4-NP

Continued on next page

	675.0 cm ⁻¹ : ⁷⁰
1	Internal motion of oxygen within the Co(Fe)O ₆ octahedra in the BSCF (Ba _{0.5} Sr _{0.5} Co _{0.8} Fe _{0.2} O _{3-δ}) lattice
	676.0 cm ⁻¹ : ^{106,110}
1	B _{1g} vibrational mode of Pd-O bond
2	Surface MnO _x oligomers
	679.0 cm ⁻¹ : ¹²⁰
1	Guanine ring stretching vibration of duplex DNA
	680.0 cm ⁻¹ : ^{20,22,26,35,61,66,115}
2	Typical Raman features of Ni-Fe-LDH
3	B _{2g} mode of nanocrystalline RuO ₂
4	Vibration of Co ₃ O ₄
5	Characteristic peaks of Co ₃ O ₄
6	Boron (B) (end)
7	A _{1g} mode of CoO/rGO
	684.0 cm ⁻¹ : ⁴²
1	NiO
	685.0 cm ⁻¹ : ^{24,106}
1	A _{1g} mode of spinel Co ₃ O ₄
2	Mn-O-Mn vibration of MnO _x -containing catalysts
	686.0 cm ⁻¹ : ³⁵
1	Characteristic peaks of Co ₃ O ₄
	690.0 cm ⁻¹ : ^{85,91,109}
1	Co (II) - O vibration mode
2	T-vibration of Fe ₂ O ₃
3	A _{1g} vibration mode of Co-O
	694.0 cm ⁻¹ : ²⁴
1	A _{1g} vibrational modes of Co-O in octahedra
	696.0 cm ⁻¹ : ⁸⁸
1	A _{1g} -Co ³⁺ -O vibration
	697.0 cm ⁻¹ : ¹²¹
1	Symmetric stretching mode of O-W-O in WO ₆ octahedra
	698.0 cm ⁻¹ : ¹⁰⁶
1	Surface MnO _x oligomers
	700.0 cm ⁻¹ : ^{50,55,64,94,101,119,120}
1	AuOH species at the surface
2	Surface hydroxyl species on Cu (CuOH)
3	Copper (hydrated) oxide peak (below 700)
4	Tensile vibration of Ni in nickel hydrate (begin)
5	Peak value of NiFeLDH system
6	Anti symmetric stretching mode of DMSO molecules
7	Silica framework vibrational modes (begin)
	704.0 cm ⁻¹ : ³⁸
1	The stretching vibration of O-W-O
	706.0 cm ⁻¹ : ⁶⁷
1	Cu-OH _{ad}

Continued on next page

	710.0 cm ⁻¹ ; 25,72,122
1	The Fe-O bond of β -FeOOH
2	*O-O tensile vibration of OOH bridging adsorption structure
3	the WO ₃ nanoparticles on the SiO ₂ substrate
	715.0 cm ⁻¹ ; 78
1	The characteristic peaks of Na ₃ Al ₃ CF ₈
	725.0 cm ⁻¹ ; 72,101
1	Small peak of β -FeOOH
2	ν_{as} W-O-W bridging mode
	727.0 cm ⁻¹ ; 120
1	Adenine ring stretching vibration of single-stranded DNA
	730.0 cm ⁻¹ ; 71,120
1	In plane bending vibrations (δ CO ₂ ⁻)
2	Adenine ring stretching vibration of single-stranded DNA
	737.0 cm ⁻¹ ; 123
1	The interaction between Na _{1+x} V ₃ O ₈ and silicon oxide
	740.0 cm ⁻¹ ; 72
1	ν_{as} W-O-W bridging mode
	742.0 cm ⁻¹ ; 124
1	In-plane vibration mode of *CO ₂ ⁻
	743.0 cm ⁻¹ ; 125
1	Ni (IV) - Oxo species
	745.0 cm ⁻¹ ; 120
1	Thymine ring stretching vibration of single-stranded DNA
	749.0 cm ⁻¹ ; 120
1	Thymine ring stretching vibration of single-stranded DNA
	750.0 cm ⁻¹ ; 126
1	V-O-V (begin)
	751.0 cm ⁻¹ ; 56
1	Peak of MoO _{3-x} (0 < x < 1)
	752.0 cm ⁻¹ ; 60
1	Co/TiO ₆ octahedral rotation of STL
	753.0 cm ⁻¹ ; 72
1	W-O-W group
	760.0 cm ⁻¹ ; 59,123
1	The stretching vibration mode of MnIV=O species
2	Mapping of Na _{1+x} V ₃ O ₈ /SiO ₂ phase
	761.0 cm ⁻¹ ; 118
1	Vibration of O-C=O
	762.0 cm ⁻¹ ; 90
1	Carboxylate bending mode
	765.0 cm ⁻¹ ; 127
1	ν (OOH*)

Continued on next page

	770.0 cm ⁻¹ ; 2
1	Tetraglyme in the electrolyte
	774.0 cm ⁻¹ ; 116
1	Vibration modes of specific bonds in R6G
	779.0 cm ⁻¹ ; 121
1	Symmetric stretching mode of O-W-O in WO ₆ octahedron
	780.0 cm ⁻¹ ; 126
1	W-O-W
	783.0 cm ⁻¹ ; 112
1	*O-O stretching vibration of OOH adsorption structure
	784.0 cm ⁻¹ ; 106,108
1	V-O-Ag bonds in edge shared triangular bipyramids (V is five coordinated)
2	V-O-Ag bonds with different coordination structures in Ag _{1.2} V ₃ O ₈
3	W-O vibration of WO ₆ unit in Mn-WO ₃
	785.0 cm ⁻¹ ; 119
1	O-O stretching mode of Li ₂ O ₂
	788.0 cm ⁻¹ ; 72
1	Symmetric stretching of SiO ₂ bridging Si-O-Si
	790.0 cm ⁻¹ ; 20
1	LiPF ₆
	796.0 cm ⁻¹ ; 72
1	The combined vibration of ν_{as} (W-O-W) and ν_{as} (W-O-Pt)
	800.0 cm ⁻¹ ; 50,72,128
1	The vibrational mode of adsorbed phosphate ions (begin)
2	Bridge W-O-W
3	Lateral Acoustic (TA) Branch of Graphite
	802.0 cm ⁻¹ ; 38
1	The stretching vibration of O-W-O
	804.0 cm ⁻¹ ; 123,129
1	ν Cu-H
2	V-O-V bonds bridge VO ₅ and VO ₆ units through co angular oxygen atoms
	805.0 cm ⁻¹ ; 49
1	Crystalline WO ₃
	807.0 cm ⁻¹ ; 130
1	Bending vibration of outer ring C-H
	808.0 cm ⁻¹ ; 72
1	the WO ₃ nanoparticles on the SiO ₂ substrate
	810.0 cm ⁻¹ ; 57,106
1	Si-O-Si symmetric stretching of SiO ₂
2	U(V) signal
	811.0 cm ⁻¹ ; 9,131
1	SeO ₃ ²⁻
2	Crystalline Na ₂ WO ₄ phase

Continued on next page

	812.0 cm ⁻¹ : ¹³²
1	Surface bound peroxymonosulfate (PMS)
	813.0 cm ⁻¹ : ¹¹²
1	*O-O stretching vibration of OOH adsorption structure
	815.0 cm ⁻¹ : ^{61,72,123}
1	v(O-O) in surface atomic-molecular hybrid structure: Ag ₂ -O-O-Ag ₂ , v(O-O) in surface Ag ₄ -O-O structure, v(O-O) in molecular oxygen complex Ag _x -O-O stabilized by subsurface atomic oxygen
2	Mapping of Na _{1+x} V ₃ O ₈ /SiO ₂ Phase
3	WOW
	816.0 cm ⁻¹ : ¹³³
1	CH ₃ shear vibration
2	dimethyl phthalate (DMP)
	823.0 cm ⁻¹ : ¹³⁴
1	the vibrational mode of δC-H rocking of metronidazole (MNZ)
	825.0 cm ⁻¹ : ¹⁰³
1	Mo-O stretching mode from MoO ₄ ²⁻
	826.0 cm ⁻¹ : ^{53,133}
1	The stretching mode of the V-O
2	Characteristic peaks of interfering substance diisodecyl o-phthalate (DIDP) (+)
	830.0 cm ⁻¹ : ^{41,126}
1	V-O-V (end)
2	Short symmetric V-O vibration (A _{1g} mode)
	831.0 cm ⁻¹ : ⁴⁰
1	Stretching vibration of OO bonds of peroxo species (O ₂ ²⁻)
	832.0 cm ⁻¹ : ^{122,135}
1	S ₂ O ₈ ²⁻
2	The O-O stretching vibration frequency of peroxide species
	834.0 cm ⁻¹ : ^{9,136}
1	SeO ₄ ²⁻
2	Ni-PMS*
	836.0 cm ⁻¹ : ¹²⁰
1	PO ₂ ⁻¹ stretch
	840.0 cm ⁻¹ : ⁷⁴
1	Stretching vibrational mode of the V-O
	848.0 cm ⁻¹ : ⁴⁷
1	4,4-biphenyldicarboxylic ion (Bpdc ²⁻) ligand dissociated from MOFs
	850.0 cm ⁻¹ : ^{94,114,115}
1	Tensile vibration of Ni in nickel hydrates (end)
2	Vibration of Co ₂ V ₂ O ₇ caused by surface oxidation of VN
3	v(O-O) of an active oxygen species (NiOO [·]) in oxyhydroxide structure
	851.0 cm ⁻¹ : ¹³⁷
1	The formation of O _{ad} *

Continued on next page

	860.0 cm ⁻¹ : ⁴⁹
1	CeVO ₄ nanoparticles
	874.0 cm ⁻¹ : ¹⁰⁶
1	W-O vibration of WO ₆ unit in MnWO ₄
	880.0 cm ⁻¹ : ^{61,125}
1	Reactive oxygen species
2	Stable molecular oxygen complex structure
3	Metastable PMS adsorbed species formed by PMS molecules (*HSO ₅ ⁻ , NiN ₄ -PMS*)
	882.0 cm ⁻¹ : ¹³²
1	SO ₅ ²⁻
	884.0 cm ⁻¹ : ¹⁰⁸
1	V-O-Ag linkages in AgVO ₃ (V is five-coordination)
	885.0 cm ⁻¹ : ¹⁰⁶
1	Surface MnO _x oligomers
	886.0 cm ⁻¹ : ⁸⁷
1	O-Mo-O stretching vibration in NiCoMo oxides (NCMO) phase
	890.0 cm ⁻¹ : ^{2,49}
1	Tetraethyl ether
2	Cu _x V _y O _z nanoparticles
	900.0 cm ⁻¹ : ^{20,65}
1	NiO
2	EC and DEC solvents
	908.0 cm ⁻¹ : ¹³³
1	interference diallyl phthalate (DAP) (-)
	910.0 cm ⁻¹ : ⁴⁹
1	CuWO ₄ nanoparticles
	912.0 cm ⁻¹ : ²⁷
1	Asymmetric stretching of C _α -C _β
	913.0 cm ⁻¹ : ¹⁰³
1	Mo-O stretching vibration mode of MoO ₄ ²⁻
	915.0 cm ⁻¹ : ^{97,130}
1	Bending vibration of NO ₂
2	Ring skeletal vibration of radial orientation
	918.0 cm ⁻¹ : ¹⁰⁷
1	-CN
	922.0 cm ⁻¹ : ¹⁰⁸
1	VOAg linkage in Ag _{1.2} V ₃ O ₈
2	VAgO _x /α-Al ₂ O ₃
	923.0 cm ⁻¹ : ³⁹
1	Co(OH) ₂
	926.0 cm ⁻¹ : ³⁸
1	Crystalline Na ₂ WO ₄ phase

Continued on next page

	927.0 cm ⁻¹ : 138
1	Raman peaks of crystalline Na ₂ WO ₄
	928.0 cm ⁻¹ : 32
1	SO ₃ species
	930.0 cm ⁻¹ : 49,119,139,140
1	The vibrational modes of ν _{O-Cr-O} bonds
2	Ce ₂ (WO ₄) ₃ nanoparticles
3	Asymmetric modes of VO ₂ ⁺
4	Cl-O scaling mode of LiClO ₄
	933.0 cm ⁻¹ : 122,141
1	ν _s (ClO ₄ ⁻)
2	Symmetric stretching vibration mode of ClO ₄ ⁻
	935.0 cm ⁻¹ : 140,142
1	Perchlorate ion
2	Additional modes of VO ₂ HSO ₄
	936.0 cm ⁻¹ : 112
1	Symmetric stretching mode of perchlorate ion (ClO ₄ ⁻)
	937.0 cm ⁻¹ : 123
1	β-NaVO ₃ species (begin)
	942.0 cm ⁻¹ : 106,123
1	β-NaVO ₃ species (end)
2	W=O stretching vibration of isolated Na-WO ₄ surface sites
	945.0 cm ⁻¹ : 123
1	VO ₂ symmetric stretching of crystal state β-NaVO ₃
	950.0 cm ⁻¹ : 131,138
1	W=O bond in WO ₄ (F) unit
2	Raman peaks of Na-WO ₄
	958.0 cm ⁻¹ : 32,72
1	SO ₄ species
2	ν _s (W=O) vibration of bridging W
	960.0 cm ⁻¹ : 72
1	ν _s (W=O) vibration of bridging W
	970.0 cm ⁻¹ : 38,106
1	The peak of W=O bond
2	SiOH stretching mode of the surface hydroxyls
	973.0 cm ⁻¹ : 72
1	ν _s (W=O) vibration of terminal W
	975.0 cm ⁻¹ : 72,143
1	ReO ₄ ⁻
2	Terminal SiOH vibration
3	Terminal W=O bands
	977.0 cm ⁻¹ : 72
1	W1/SiO ₂
	979.0 cm ⁻¹ : 59,132
1	SO ₄ ²⁻

Continued on next page

	980.0 cm ⁻¹ : 15,72,144
1	Vibration of SO ₄ ²⁻ ion
2	Na ₂ SO ₄
3	W4/SiO ₂
4	ν _s (W=O) vibration of terminal W
	981.0 cm ⁻¹ : 138,145
1	Sulfate radical species
2	Mn-WO ₄
	982.0 cm ⁻¹ : 125,136
1	SO ₄ ²⁻
	985.0 cm ⁻¹ : 138
1	WO ₄ species
	986.0 cm ⁻¹ : 88
1	NiOOH peak
	988.0 cm ⁻¹ : 59
1	Symmetric stretching vibrational mode of PO ₃ in HPO ₄ ²⁻
	990.0 cm ⁻¹ : 140,146
1	Reduced form of PPy (CC ring deformation)
2	V=O stretching vibration
	992.0 cm ⁻¹ : 123
1	V=O stretching vibration in VO ₅ polyhedra
	993.0 cm ⁻¹ : 68
1	S ₂ O ₃ ²⁻
	995.0 cm ⁻¹ : 143
1	Re-O-Ti vibration of isolated four coordinated rhenium oxide monomers
	996.0 cm ⁻¹ : 27
1	The ring-breathing mode of the phenyl groups in the [AlPh ₄] ⁻ anions
	997.0 cm ⁻¹ : 49
1	Crystalline V ₂ O ₅
	998.0 cm ⁻¹ : 147
1	In plane bending of benzene ring in TP molecule (δ-CCC)
2	Benzol Toroidal Bend (δ-CCC) in PMBA Molecules
	1000.0 cm ⁻¹ : 120,143
1	Re-O-Ti of isolated four coordinated rhenium oxide monomers
2	Silica framework vibrational modes (end)
	1003.0 cm ⁻¹ : 120
1	Phenyl ring breathing
	1007.0 cm ⁻¹ : 135
1	SO ₄ ²⁻
	1010.0 cm ⁻¹ : 73,128
1	Superoxide (O-O) species adsorbed on the surface of high-entropy alloy (HEA)
2	Symmetric C-O stretching vibration of HCO ₃ ⁻

Continued on next page

	1012.0 cm ⁻¹ ; 49
1	Raman peaks of surface WO _x on TiO ₂ support
	1014.0 cm ⁻¹ ; 48
1	Bicarbonate
	1015.0 cm ⁻¹ ; 137,138,148
1	CO ₂ ²⁻ adsorbed on electrode surface in electrolyte
2	Formation of O ₂ [*]
3	W=O bond in WO _x mono-oxo species
	1022.0 cm ⁻¹ ; 147
1	CH in-plane bending (δ_{CH})
	1025.0 cm ⁻¹ ; 143
1	Surface sulfate
	1026.0 cm ⁻¹ ; 105,149
1	C-O bonds of methylol groups
2	MeOH
	1027.0 cm ⁻¹ ; 142
1	Stretching of linear dimers and/or high polymers
	1028.0 cm ⁻¹ ; 27,58
1	C $_{\alpha}$ - C $_{\beta}$ symmetric stretching
2	The A _{1g} mode of Ni-O in NiCo ₂ O ₄
	1029.0 cm ⁻¹ ; 110
1	α -NiH vibration mode
	1030.0 cm ⁻¹ ; 49,79,119
1	Vibration of pyridine ring
2	Raman peaks of surface VO _x on TiO ₂ support
3	CH ₃ rocking mode
	1035.0 cm ⁻¹ ; 140,142
1	ν (S=O)
2	HSO ₄ ⁻ ..H ₃ O ⁺ sulfur oxygen bond stretching vibration
	1039.0 cm ⁻¹ ; 123
1	Isolation vanadium oxo (V=O) in [VO ₄]/SiO ₂ (500-air)
	1040.0 cm ⁻¹ ; 142
1	In-phase S=O stretching mode
	1041.0 cm ⁻¹ ; 127
1	ν (OH [*])
	1044.0 cm ⁻¹ ; 32
1	SO ₄ species
	1046.0 cm ⁻¹ ; 32
1	NO ₃ species
	1047.0 cm ⁻¹ ; 88,140,144
1	The formation of O [*] on the Co sites (C-O [*])
2	Stretching of NO ₃ ⁻
3	HSO ₄ ⁻
	1050.0 cm ⁻¹ ; 62,146,150
1	H ₂ SO ₄ electrolyte
2	C-H in-plane deformation in neutral PPy units
3	NO ₃ ⁻ ion

Continued on next page

	1051.0 cm ⁻¹ ; 42,142
1	Out-of-phase S=O stretching mode
2	Ni oxide NiO
	1054.0 cm ⁻¹ ; 34
1	Lepidocrocite (γ -FeOOH)
	1057.0 cm ⁻¹ ; 73
1	Symmetric stretching vibration mode of SO ₄ ²⁻
	1059.0 cm ⁻¹ ; 132
1	SO ₅ ²⁻
	1060.0 cm ⁻¹ ; 136
1	Vibration mode of HSO ₅ ⁻
	1062.0 cm ⁻¹ ; 110,125
1	2P _{LO} of NiO vibrational modes
2	HSO ₅ ⁻
	1064.0 cm ⁻¹ ; 88,101,151
1	Active oxygen species (O [*]) at the Ni sites
2	C-O stretching vibration of CO ₃ ²⁻
3	Carbonate ions inserted into D-NiFeALDH
	1065.0 cm ⁻¹ ; 55
1	Adsorbed carbonate
	1069.0 cm ⁻¹ ; 48
1	Carbonate
	1071.0 cm ⁻¹ ; 147,152
1	In plane bending and C-S stretching of benzene ring in p-mercaptobenzoic acid (PMBA) molecules ($\delta_{CCC} + \gamma_{CS}$)
2	C-S vibration (δ_{CS})
3	Aromatic ring breathing, symmetric C-H plane bending and C-S extension
	1072.0 cm ⁻¹ ; 105
1	[O-O] Vibration
	1073.0 cm ⁻¹ ; 122
1	*OH species in Pt ₃ Co@Pt-SAC
	1074.0 cm ⁻¹ ; 135
1	S ₂ O ₈ ²⁻
	1077.0 cm ⁻¹ ; 130
1	Peak value of 4-aminothiophenol (4-ATP)
	1080.0 cm ⁻¹ ; 146,153
1	4-ATP
2	C-H in-plane deformation in polarons
	1083.0 cm ⁻¹ ; 154
1	C-S vibration in p-nitrothiophenol (PNTTP) and DMAB
	1084.0 cm ⁻¹ ; 130
1	C-S telescopic vibration
	1087.0 cm ⁻¹ ; 79
1	Surface OOH [*] intermediate

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	1090.0 cm ⁻¹ ; ⁶⁵	
1	NiO	
	1094.0 cm ⁻¹ ; ¹⁴³	
1	ν_{asym} (S-O) of surface sulfates	
	1100.0 cm ⁻¹ ; ^{20,50,120}	
1	The vibrational mode of adsorbed phosphate ions (end)	
2	Li ₂ CO ₃	
3	C-H mode	
	1120.0 cm ⁻¹ ; ⁹⁰	
1	Vibration of $\nu_{\text{C-O}}$ and $\delta_{\text{C-O-H}}$	
	1124.0 cm ⁻¹ ; ¹²⁴	
1	Symmetric stretching mode of *CO ₂ ⁻	
	1130.0 cm ⁻¹ ; ⁴⁴	
1	Multi phonon (MP) transition mode of CuO	
	1139.0 cm ⁻¹ ; ^{150,155}	
1	Head-to-Head Coupling Product p,p-	
	dimercaptoazobenzene (DMAB)	
2	-NH ₂	
	1144.0 cm ⁻¹ ; ¹³⁰	
1	C-N symmetric stretching	
	1150.0 cm ⁻¹ ; ¹⁴³	
1	Surface sulfate on the catalyst	
	1152.0 cm ⁻¹ ; ¹⁵⁶	
1	-NH ₂	
	1158.0 cm ⁻¹ ; ¹¹⁸	
1	CH wagging of HCOO ⁻	
	1160.0 cm ⁻¹ ; ¹⁴⁸	
1	Symmetric stretching vibration of *CO ₂ ⁻ (ν_{s} CO ₂ ⁻)	
	1161.0 cm ⁻¹ ; ¹⁵⁷	
1	In-plane C-H bending	
	1170.0 cm ⁻¹ ; ³³	
1	2LO overtone	
	1174.0 cm ⁻¹ ; ³⁷	
1	Second order longitudinal mode of CeO ₂ lattice (2LO)	
	1178.0 cm ⁻¹ ; ¹³⁰	
1	In-plane ring CH bending	
	1180.0 cm ⁻¹ ; ⁴⁰	
1	The second-order longitudinal optical mode of the fcc fluorite phase of CeO ₂	
	1181.0 cm ⁻¹ ; ¹⁵⁸	
1	Bending vibration mode of C-H stretching vibration	
	1188.0 cm ⁻¹ ; ¹³⁴	
1	The in-plane bending mode of metronidazole (MNZ) molecules	

Continued on next page

	1200.0 cm ⁻¹ ; ^{94,107,114}	
1	-OH (begin)	
2	Tensile vibration of nickel in nickel hydrates	
3	Vibration of reactive oxygen species (NiOO ⁻) in the structure of oxygen hydrides	
	1220.0 cm ⁻¹ ; ¹⁵⁹	
1	Disordered graphite lattice with A _{1g} symmetry	
	1221.0 cm ⁻¹ ; ¹⁵⁷	
1	C-N stretching	
	1225.0 cm ⁻¹ ; ¹⁵⁵	
1	Head-to-tail coupling product 4-mercapto-N-phenylquinone diimine (NPQD)	
	1227.0 cm ⁻¹ ; ¹¹⁸	
1	The stretching vibrational mode of CO and rocking mode of -CH	
	1240.0 cm ⁻¹ ; ¹²⁰	
1	Thymine pattern	
	1250.0 cm ⁻¹ ; ³³	
1	The presence of the salt form of flexible-chain PAMPSA	
	1255.0 cm ⁻¹ ; ¹⁴⁶	
1	The presence of the salt form of flexible-chain PAMPSA	
	1257.0 cm ⁻¹ ; ¹²⁰	
1	Amide III stretching mode	
	1270.0 cm ⁻¹ ; ¹¹⁸	
1	Fermi resonance of CO ₂ and -CH rocking vibration of MOF-801	
	1280.0 cm ⁻¹ ; ⁴⁷	
1	Bpdc ²⁻ ligand dissociated from MOFs	
	1283.0 cm ⁻¹ ; ^{58,134}	
1	Twisting of MNZ molecule	
2	Symmetric tensile vibration of C-O	
	1288.0 cm ⁻¹ ; ¹²⁴	
1	OH- deformation in *COOH	
	1290.0 cm ⁻¹ ; ¹⁶⁰	
1	-NO ₂ peak	
	1300.0 cm ⁻¹ ; ^{34,107,130}	
1	Ring C-C stretching vibration	
2	Lepidocrocite (γ -FeOOH)	
3	-OH (end)	
	1308.0 cm ⁻¹ ; ¹²⁹	
1	Stretching vibration (ν_{s} CO ₂ ⁻)	
	1310.0 cm ⁻¹ ; ¹⁵	
1	Hematite	
	1317.0 cm ⁻¹ ; ⁷³	
1	D-bands in carbon nanofibers (CNFs)	

Continued on next page

	1320.0 cm ⁻¹ ; 75,159
1	Disordered graphite lattice with A _{1g} symmetry
2	Amorphous layer of C@ZnO sample
	1330.0 cm ⁻¹ ; 160
1	Adsorption on Au-2
	1333.0 cm ⁻¹ ; 130
1	O-N-O stretching vibration
	1335.0 cm ⁻¹ ; 3,119,144
1	Graphitic carbon (D band)
2	Stretching vibration of N=O in NO ₂ ⁻
3	D-bands related to lattice structure defects in carbon materials
	1336.0 cm ⁻¹ ; 97
1	Bending vibration of NO ₂
	1337.0 cm ⁻¹ ; 12
1	sp ³ disordered carbon in TiO ₂ -CNT (carbon nanotube) film
	1340.0 cm ⁻¹ ; 19,127,149,160
1	C-H bonds of methylol groups
2	v(NO ₂)
3	Vibration modes of disordered carbon (D)
4	Structural defects in graphene (D-band)
	1342.0 cm ⁻¹ ; 54
1	O-O stretching and contraction of Mn-OOD
	1344.2 cm ⁻¹ ; 161
1	D peak of C-C (related to defects)
	1346.0 cm ⁻¹ ; 1
1	D-band of graphite
	1347.0 cm ⁻¹ ; 58
1	Water bending
	1350.0 cm ⁻¹ ; 20,54,90,162
1	Breathing patterns of K-point A _{1g} phonons
2	Structural defects of graphite electrodes
3	Breathing modes of sp ² hexagonal ring in multilayer graphene foam
4	Carboxyl group
	1352.0 cm ⁻¹ ; 162
1	Defects in carbon fiber structure
	1354.0 cm ⁻¹ ; 54
1	O-O stretching vibration of Mn-OOH intermediate
	1360.0 cm ⁻¹ ; 63
1	D-band of organic matrix
2	G-band of carbon nanotubes
	1364.0 cm ⁻¹ ; 116
1	Vibration modes of specific bonds in R6G
	1370.0 cm ⁻¹ ; 160
1	-NO ₂ peak

Continued on next page

	1375.0 cm ⁻¹ ; 144
1	-NH ₂ bending vibration
	1389.0 cm ⁻¹ ; 163
1	C=N bonds in organic ligands of MOFs
	1390.0 cm ⁻¹ ; 160
1	Azo group of DMAB
	1391.0 cm ⁻¹ ; 130
1	N-N stretching
	1395.0 cm ⁻¹ ; 158
1	Vibration of C=C _{wing}
	1411.0 cm ⁻¹ ; 144
1	The NO ₂ stretching of NO ₃ ⁻
	1416.0 cm ⁻¹ ; 79
1	In- and out-of-phase stretching modes of the carboxylate group
	1420.0 cm ⁻¹ ; 107
1	COOH functional group
	1422.0 cm ⁻¹ ; 124
1	Symmetric extension of *CO ₂ ⁻
	1426.0 cm ⁻¹ ; 142
1	δ(C-H) mode
	1435.0 cm ⁻¹ ; 130
1	C-H in-plane bending mode
	1438.0 cm ⁻¹ ; 118
1	Vibration of O-C=O
	1440.0 cm ⁻¹ ; 160
1	azo group of DMAB
	1441.0 cm ⁻¹ ; 154
1	N=N vibration in DMAB
	1442.0 cm ⁻¹ ; 155
1	Head-to-head coupling product DMAB
	1445.0 cm ⁻¹ ; 34,133
1	Methylene deformation mode (δCH ₂)
2	CH ₃ shear vibration
3	di(2-ethylhexyl) phthalate (DEHP)
	1450.0 cm ⁻¹ ; 20,118
1	CH ₂ scissoring of polystyrene (PS)
2	Amorphous carbon
	1452.0 cm ⁻¹ ; 150
1	Asymmetric bending vibration of NH ₄ ⁺
	1482.0 cm ⁻¹ ; 157
1	C=N stretching of quinone (Q)
	1490.0 cm ⁻¹ ; 155
1	Head-to-tail coupling product NPQD

Continued on next page

	1500.0 cm ⁻¹ ; 120,159
1	The presence of amorphous carbon
2	C-C mode
	1510.0 cm ⁻¹ ; 35
1	C=C _{ring} in DFF
	1515.0 cm ⁻¹ ; 35
1	C=C _{ring} in HMFA
	1517.0 cm ⁻¹ ; 30
1	The disappearance of C=C _{ring} vibration mode in 2,5-furandicarboxylic acid (FDCA)
	1518.0 cm ⁻¹ ; 35
1	C=C _{ring} in FDCA
	1523.0 cm ⁻¹ ; 30,35
1	C=C ring in 5-hydroxymethylfurfural (HMF) molecules
	1525.0 cm ⁻¹ ; 145
1	Vibration of -NH bonds
	1526.0 cm ⁻¹ ; 156
1	-NH
	1528.0 cm ⁻¹ ; 35,150
1	C=C _{ring} in FDCA
2	-NH
3	Adsorbed H ₂ NOH
	1538.0 cm ⁻¹ ; 120,130
1	Ring C-C stretching vibration
2	Amide II stretching mode
	1540.0 cm ⁻¹ ; 129,148
1	Asymmetric stretching vibration of *CO ₂ ⁻ (ν _{as} CO ₂ ⁻)
	1547.0 cm ⁻¹ ; 129
1	ν _{as} CO ₂ ⁻
	1550.0 cm ⁻¹ ; 118
1	The O=C=O asymmetric vibration of CO ₂ and stretching vibration of polystyrene (PS)
	1570.0 cm ⁻¹ ; 130
1	Phenyl ring mode
	1577.0 cm ⁻¹ ; 147,164
1	C-C stretching mode (γ _{cc})
2	The stretching of the in-plane CC bonds
	1579.0 cm ⁻¹ ; 165
1	In-plane vibration of sp ² carbon atoms in graphite
	1580.0 cm ⁻¹ ; 3,166
1	Graphitic carbon (G band)
2	The G-band of N-doped porous carbon matrix (NDPCM)
	1581.0 cm ⁻¹ ; 127,152
1	Aromatic C-C extension, asymmetric C-H in-plane bending
2	Graphene carbon (G)

Continued on next page

	1581.2 cm ⁻¹ ; 19
1	Ordered sp ² bonded carbon (G-band) of graphene
	1581.9 cm ⁻¹ ; 161
1	G peak of C-C (related to graphitic carbon)
	1582.0 cm ⁻¹ ; 147
1	C-C stretching mode (γ _{cc})
	1585.0 cm ⁻¹ ; 12
1	Sp ² bonded carbon atoms in TiO ₂ -CNT films
	1587.0 cm ⁻¹ ; 130
1	Ring C-C stretching vibration
	1590.0 cm ⁻¹ ; 19,20,71,79,90,159,160
1	The in- and out-of-phase stretching modes of the carboxylate group
2	Ring vibration of defect free graphite lattice (E _{2g} symmetry)
3	The stretching vibration of -NH ₂ groups
4	Ordered sp ² bonded carbon (G-band) of graphene
5	G band related to sp ² -linked carbon atoms in uncoated graphite electrodes
6	High-frequency bond stretching of sp ² carbon pairs in rings and chains
7	Adsorbed carboxyl species (*CO ₂ ⁻)
	1592.0 cm ⁻¹ ; 167
1	Characteristic peaks of 4-mercapto-benzonitrile (4-MBN) molecules
	1593.0 cm ⁻¹ ; 130,164
1	Peak value of 4-ATP
2	The stretching of the in-plane CC bonds
	1596.0 cm ⁻¹ ; 97,162
1	G-bands of CFP, CCFC, and FCFC
2	Crystallinity in Carbon Fiber Structure
	1597.0 cm ⁻¹ ; 146
1	C=C stretching in the oxidized pyrrole ring
	1598.0 cm ⁻¹ ; 116
1	Vibration modes of specific bonds in 3,3',4,4-tetrachlorobiphenyl (one congener of polychlorinated biphenyls, PCB-77)
	1600.0 cm ⁻¹ ; 47,166,168
1	Bpdc ²⁻ ligand dissociated from MOFs
2	C=C vibration
3	Peak of H ₂ O
	1601.0 cm ⁻¹ ; 58
1	The bending vibration of water
	1605.0 cm ⁻¹ ; 54
1	The G band of the E _{2g} phonon of sp ² carbons
	1606.0 cm ⁻¹ ; 169
1	HOH bending of water

Continued on next page

	1607.0 cm ⁻¹ ; ¹²⁰
1	C=C stretching modes
	1608.0 cm ⁻¹ ; ⁷³
1	G-band in CNFs
	1609.0 cm ⁻¹ ; ¹⁵⁸
1	Vibration mode of C=C _{ring}
	1610.0 cm ⁻¹ ; ¹⁶²
1	benzene ring
	1612.0 cm ⁻¹ ; ¹²⁰
1	C=C stretching peak
	1617.0 cm ⁻¹ ; ¹⁶³
1	C=C Bond in Organic Ligands of MOFs
	1620.0 cm ⁻¹ ; ^{130,159,170}
1	Ring C-C stretching vibration
2	ν_{IP} (Aromatic C-C)
3	the presence of a poorly ordered graphite structure
	1632.0 cm ⁻¹ ; ¹⁵⁰
1	Adsorbed H ₂ NOH
	1637.0 cm ⁻¹ ; ⁷⁷
1	α -Ni(OH) ₂
	1638.0 cm ⁻¹ ; ¹²⁹
1	δ H ₂ O
	1639.0 cm ⁻¹ ; ¹⁵⁵
1	Head-to-tail coupling product NPQD
	1639.4 cm ⁻¹ ; ¹⁷¹
1	Bending vibration of H-OH bonds
	1640.0 cm ⁻¹ ; ^{94,120}
1	Vibration of water molecules
2	The bending vibration mode of water
	1642.0 cm ⁻¹ ; ^{118,133}
1	the C=O vibration/OH rocking of HOCO*
2	Characteristic peaks of interfering substance diallyl phthalate (DAP)
	1643.1 cm ⁻¹ ; ¹⁷¹
1	Bending vibration of H-OH bonds
	1650.0 cm ⁻¹ ; ³⁴
1	The bending mode of NH groups
	1661.0 cm ⁻¹ ; ^{30,118}
1	Vibration of C=C
2	Aldehyde groups of HMF molecules
	1664.0 cm ⁻¹ ; ¹⁴⁹
1	C=O bond of aldehyde group
	1668.0 cm ⁻¹ ; ¹²⁰
1	The carbonyl stretching vibration of thymine, guanine, and cytosine

Continued on next page

	1669.0 cm ⁻¹ ; ³⁰
1	The peak associated with aldehyde oxidation
	1727.0 cm ⁻¹ ; ⁹⁰
1	$\nu_{C=O}$
	1750.0 cm ⁻¹ ; ¹⁶⁶
1	Peak of H ₃ O ⁺
	1780.0 cm ⁻¹ ; ¹²⁰
1	The stretching vibration peak of thiosuccinimide
	1800.0 cm ⁻¹ ; ^{48,67}
1	Stretching of C $\equiv$ O
2	*CO on hollow sites (begin)
	1860.0 cm ⁻¹ ; ¹⁷²
1	C $\equiv$ O tensile vibration of CO (CO _{bridge})
	1900.0 cm ⁻¹ ; ⁶⁷
1	*CO on bridge sites (begin)
2	*CO on hollow sites (end)
	1939.0 cm ⁻¹ ; ⁴⁴
1	Adsorbed *CO intermediates on the CuO@C-600 electrode
	1980.0 cm ⁻¹ ; ¹¹³
1	Interactions between CO* intermediates
	1998.0 cm ⁻¹ ; ¹⁷³
1	CO _{bridge} on Cu ₂ O-TiO ₂ interface
	2000.0 cm ⁻¹ ; ^{67,174}
1	C $\equiv$ O stretching of *CO (begin)
2	*CO on top sites (linearly *CO) (begin)
3	*CO on bridge sites (end)
	2010.0 cm ⁻¹ ; ¹²⁹
1	ν_s CO
	2056.0 cm ⁻¹ ; ¹²⁸
1	Expansion and contraction vibration of C=O
	2070.0 cm ⁻¹ ; ¹¹³
1	C-O stretching vibrational mode of adsorbed *CO on a Cu top site (*CO _{atop})
	2085.0 cm ⁻¹ ; ¹²⁹
1	$\nu_{C\equiv O}$
	2100.0 cm ⁻¹ ; ⁴⁸
1	Stretching of C $\equiv$ O
	2120.0 cm ⁻¹ ; ^{67,173}
1	*CO on top sites (linearly *CO) (end)
2	CO _{atop} on Cu ₂ O-TiO ₂ Interface
	2180.0 cm ⁻¹ ; ¹⁷⁵
1	The characteristic peak of cyanide bridge (C $\equiv$ N)
	2200.0 cm ⁻¹ ; ¹⁷⁴
1	C $\equiv$ O stretching of *CO (end)

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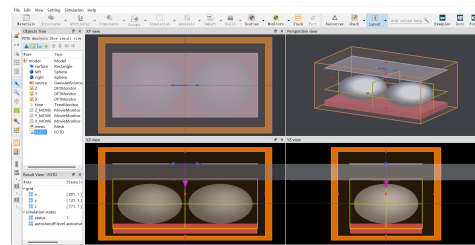
2208.0 cm ⁻¹ : ¹⁵⁸	
1	The vibration mode of C≡N
2230.0 cm ⁻¹ : ¹⁶⁷	
1	Characteristic peaks of 4-MBN molecules
2530.0 cm ⁻¹ : ¹²⁷	
1	ν (S-H)
2532.0 cm ⁻¹ : ¹⁴¹	
1	Stretching vibration of S-H, ν (S-H)
2541.0 cm ⁻¹ : ¹⁰⁷	
1	-CH ₃
2550.0 cm ⁻¹ : ¹⁷⁶	
1	O-D vibration
2580.0 cm ⁻¹ : ¹²⁰	
1	The distinctive thiol SH stretching mode
2670.0 cm ⁻¹ : ¹²⁷	
1	Vibration modes of 2D layers
2677.0 cm ⁻¹ : ¹	
1	2D bands of graphite
2700.0 cm ⁻¹ : ⁹⁰	
1	D peak related to the number of graphene layers and orientation
2800.0 cm ⁻¹ : ³⁴	
1	Antisymmetric (ν_s CH ₂) stretching mode of methylene groups
2848.0 cm ⁻¹ : ⁶⁷	
1	Intermediate products -CH _x in CO ₂ RR
2874.0 cm ⁻¹ : ⁶⁷	
1	Intermediate products -CH _x in CO ₂ RR
2900.0 cm ⁻¹ : ¹⁷⁷	
1	metal-OH band of OH*
2904.0 cm ⁻¹ : ⁶⁷	
1	Intermediate products -CH _x in CO ₂ RR
2934.0 cm ⁻¹ : ¹⁶⁸	
1	C-H vibration
2936.0 cm ⁻¹ : ⁶⁷	
1	Intermediate products -CH _x in CO ₂ RR
2961.0 cm ⁻¹ : ⁶⁷	
1	Intermediate products -CH _x in CO ₂ RR
3000.0 cm ⁻¹ : ^{34,90}	
1	Antisymmetric (ν_a CH ₂) stretching mode of methylene groups
2	O-H stretching of hydroxyl groups

Continued on next page

3004.0 cm ⁻¹ : ³⁴	
1	The ethylenic elongation (ν CH(C=C))
3200.0 cm ⁻¹ : ^{174,176,177}	
1	Tetrahedrally coordinated H-bonded water
2	Raman peaks of interfacial water
3	O-H vibration
3400.0 cm ⁻¹ : ^{174,176}	
1	Trihedrally coordinated H-bonded water
2	O-H vibration
3458.0 cm ⁻¹ : ¹⁶⁹	
1	OH stretching mode of water
3500.0 cm ⁻¹ : ^{90,177}	
1	Raman peaks of interfacial water
2	O-H stretching of COOH groups
3600.0 cm ⁻¹ : ¹⁷⁴	
1	The H-bonding-free water with the dangling OH bonds

Scheme of oFDTD

- Simulation Software:** The FDTD simulation is conducted using ANSYS Lumerical FDTD 2020 R2.4.
- Simulation Objects:** The objects under simulation include a silicon wafer and two metallic particles made of noble metal (Fe, Co, Ni, Cu, Rh, Pd, Ag, Os, Ir, Pt, Au).



The diatomic FDTD simulation model

Table 1 Noble Metals Used in Diatomic FDTD Simulation Model

Metal	Optical Constants Source
Fe	Fe (Iron) - Palik
Co	n: 2.5769, k: 4.6527 ^{178,179}
Ni	Ni (Nickel) - Palik
Cu	Cu (Copper) - Palik
Rh	Rh (Rhodium) - Palik
Pd	Pd (Palladium) - Palik
Ag	Ag (Silver) - Johnson and Christy
Os	n: 4.2664, k: 3.25 ^{178,180}
Ir	n: 2.4852, k: 4.9865 ^{178,181}
Pt	Pt (Platinum) - Palik
Au	Au (Gold) - Johnson and Christy

- Silicon Wafer Specifications:** The silicon wafer measures 90nm in length, 45nm in width, and 10nm in height.

4. **Metallic Particles Specifications:** Each metallic particle has a radius of 20nm.
5. **Light Source:** A Gaussian light source with an amplitude of 1 is employed.
6. **Wavelength:** The simulation operates at a central wavelength of 532nm and 785nm with a wavelength span of 300nm.
7. **Monitor Setup:** The simulation monitors the X, Y, and Z cross-section along the axis defined by the line connecting the centers of the two metallic particles, and exports the data for further analysis.
8. **Mesh Size:** The maximum mesh step sizes, dx, dy, and dz, are set to 0.5nm to ensure accurate simulation results.
9. **FDTD Region:** The FDTD computational region encompasses all objects within a size of 100nm by 60nm by 85nm.

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