

Supplementary information

Extending the π -Conjugated Benzoimidazoisindolone-based n-Type Ladder Polymers: Highly Air-robust Aqueous Sodium-ion Electrochromic Energy Storage and Mechanism

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Table of Contents

Results and discussion.....	1
Figure S1. Film thickness	1
Figure S2. Absorption spectra	2
Figure S3. Raman spectra	3
Figure S4. Electron gain and loss mechanism	4
Figure S5. XPS spectra	5
Figure S6. Plots of the j_p vs. $v^{1/2}$	6
Figure S7. N ₂ adsorption and desorption curves	7
Figure S8. Electrochemical impedance spectroscopy (EIS).....	8
Figure S9. Electrochromic cycling stability of BBL-EC film	9
Figure S10. GCD curves and specific capacity	10
Figure S11. Electrochemical cycling stability under illumination conditions.....	11
Figure S12. Electrochromic performance of BBL-EC at different temperatures	12
Figure S13. Electrochromic performance of BAL-EC at different temperatures.....	13
Figure S14. Electrochromic performance of ECD	14
Figure S15. GCD curves.....	15
Figure S16. Raman spectra during the charge-discharge	16
Figure S17. GCD curves and capacity retention	17
Table S1. Detailed fitting parameters of Nyquist plots for BBL-EC ladder polymer	18
Table S2. Detailed fitting parameters of Nyquist plots for BAL-EC ladder polymer	19
Table S3. Capacitive Performance of Electrochromic Polymers	20
Table S4. Performance Comparison Table of n-Type Aqueous Electrochromic Energy Storage Materials	21

Results and discussion

Figure S1. Film thickness

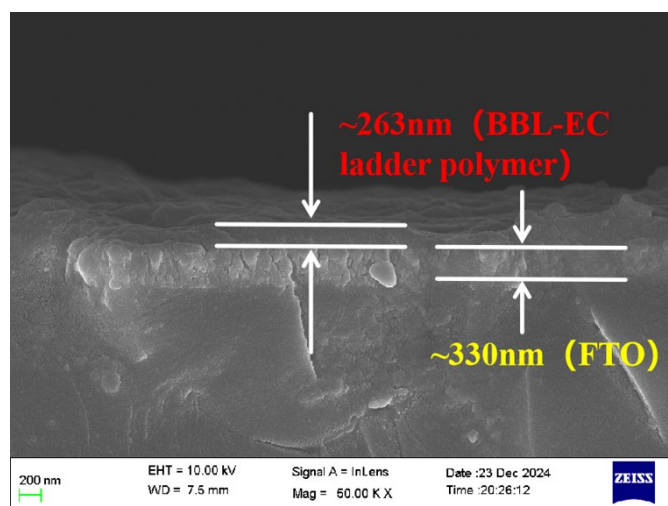


Figure S1. Film thickness of BBL-EC.

Figure S2. Absorption spectra

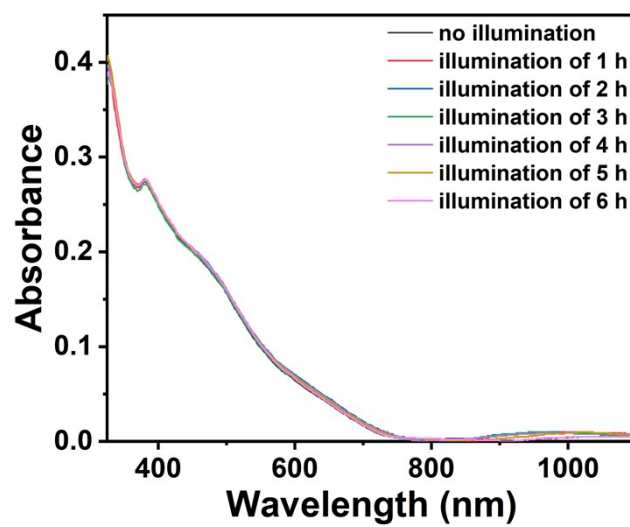


Figure S2. Absorption spectra of BBL-EC at different exposure times and in the air.

Figure S3. Raman spectra

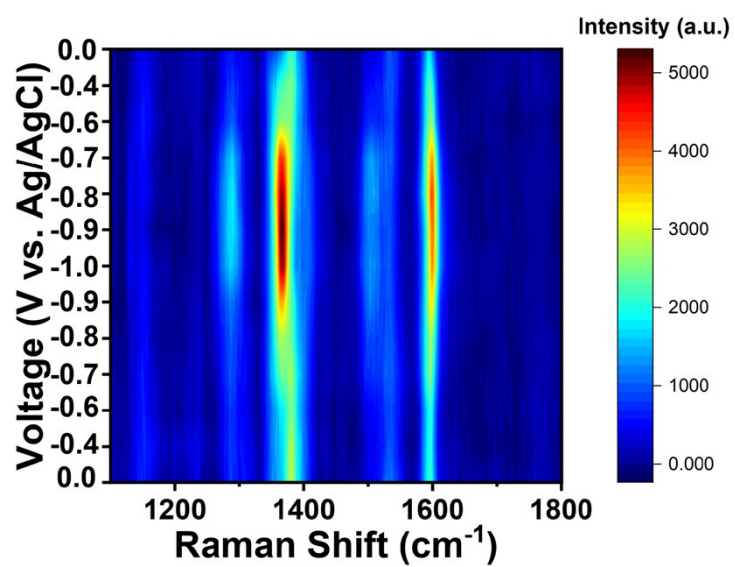


Figure S3. 2D mapping of the Raman spectra of BBL-EC.

Figure S4. Electron gain and loss mechanism

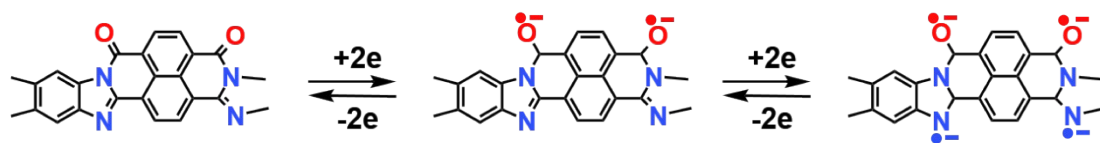


Figure S4. Electron gain and loss mechanism in electrochemical REDOX process of BBL-EC.

Figure S5. XPS spectra

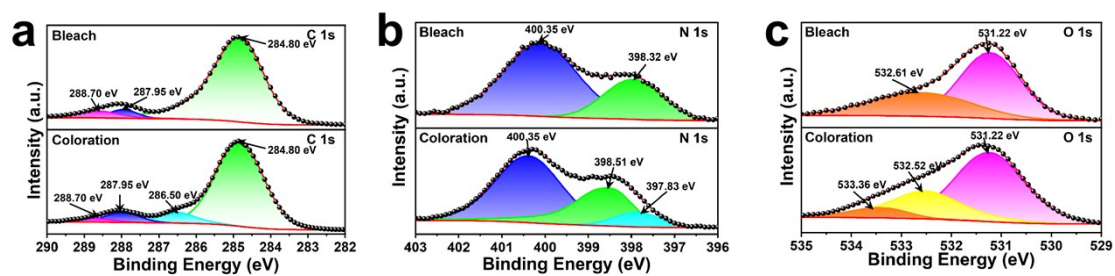


Figure S5. XPS (a) C 1s, (b) N 1s and (c) O 1s spectra of colored/bleached states of BBL-EC.

Figure S6. Plots of the j_p vs. $v^{1/2}$

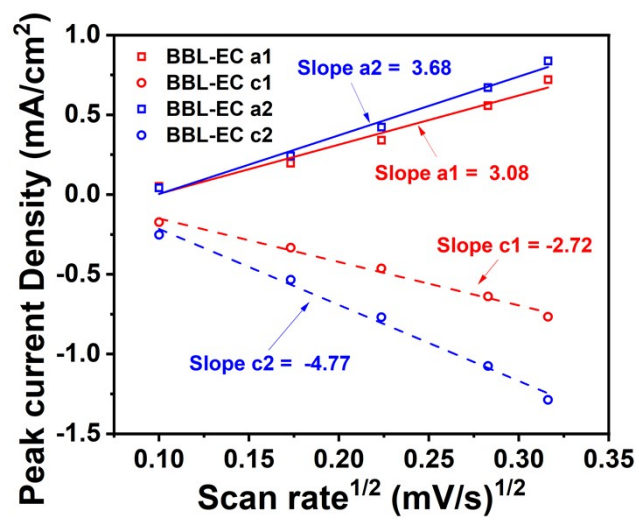


Figure S6. Plots of the j_p vs. $v^{1/2}$ of BBL-EC.

Figure S7. N₂ adsorption and desorption curves

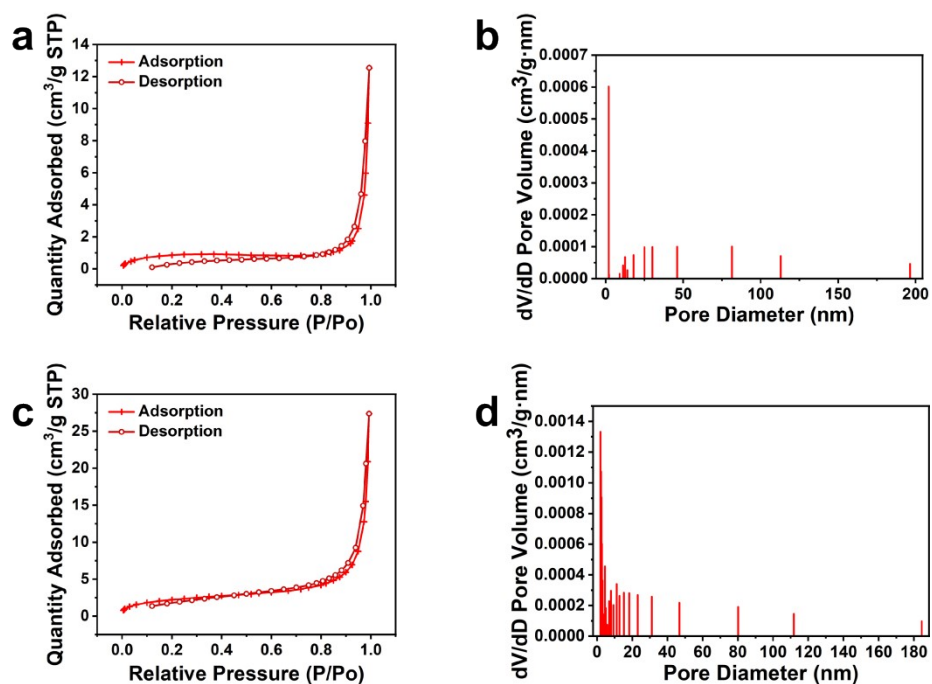


Figure S7. N₂ adsorption and desorption curves at 77K of (a) BBL-EC and (c) BAL-EC; Pore size distribution of (b) BBL-EC and (d) BAL-EC.

Figure S8. Electrochemical impedance spectroscopy (EIS)

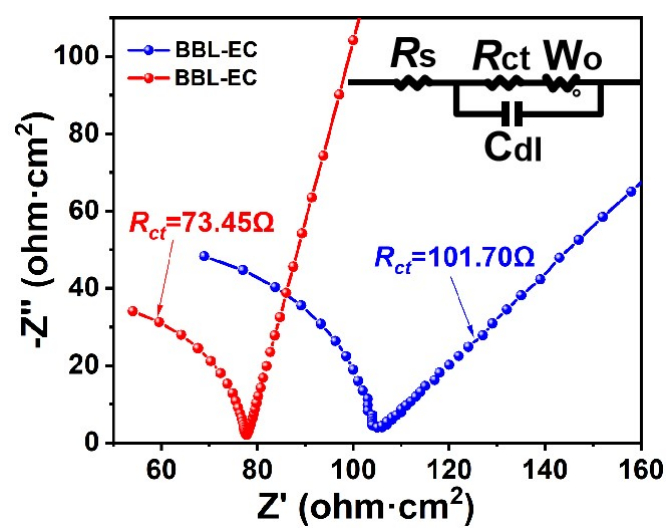


Figure S8. EIS (inset is equivalent circuit diagram) of BBL-EC and BAL-EC.

Figure S9. Electrochromic cycling stability of BBL-EC film

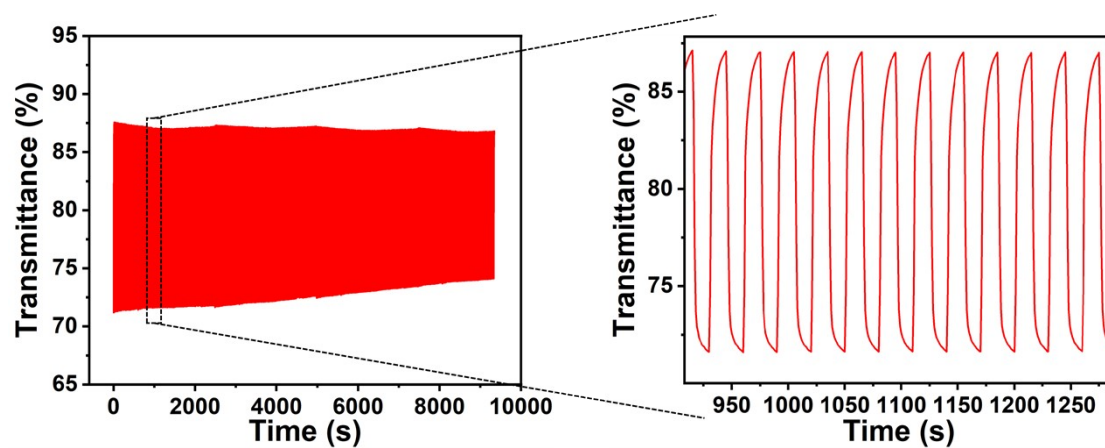


Figure S9. Electrochromic cycling stability of BBL-EC film.

Figure S10. GCD curves and specific capacity

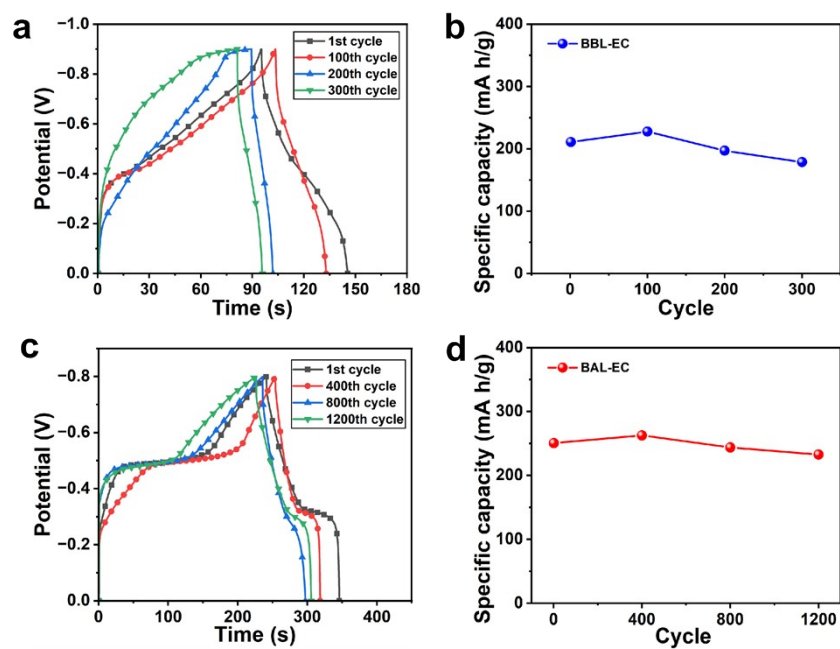


Figure S10. GCD curves and specific capacity of (a, b) BBL-EC and (c, d) BAL-EC at 0.3 mA/cm².

Figure S11. Electrochemical cycling stability under illumination conditions

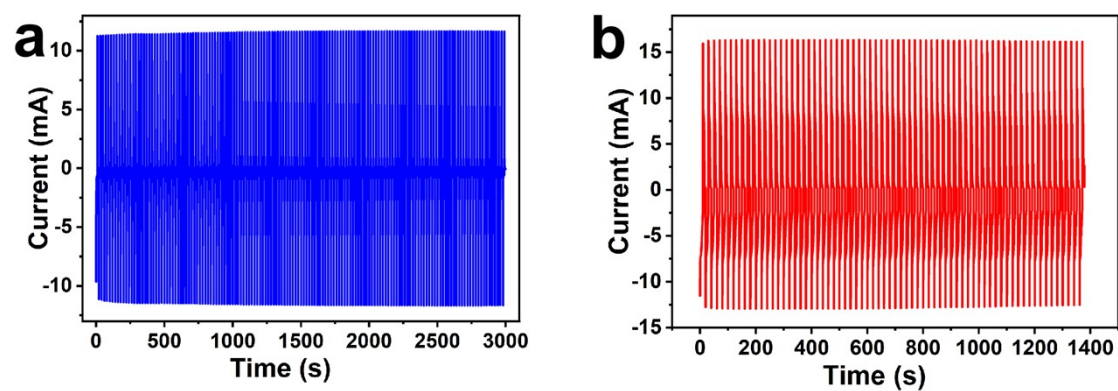


Figure S11. Electrochemical cycling stability of (a) BBL-EC and (b) BAL-EC under illumination conditions.

Figure S12. Electrochromic performance of BBL-EC at different temperatures

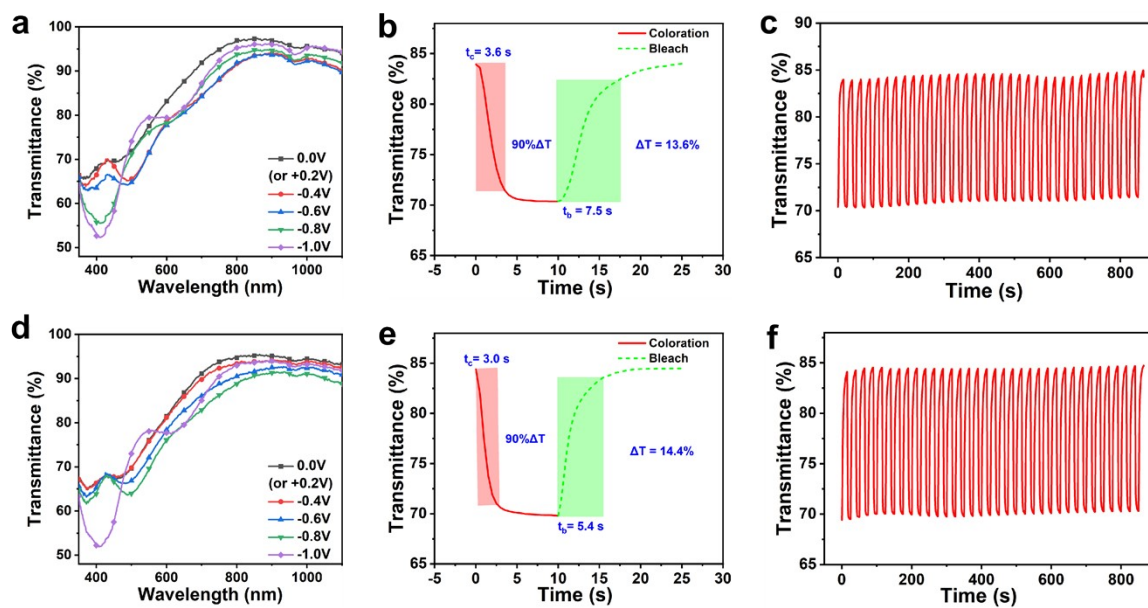


Figure S12. Electrochromic performance of BBL-EC at different temperatures. (a, d) Spectroelectrochemistry ; (b, e) response times; and (c, f) cycling stability tests at 0°C and 40°C.

Figure S13. Electrochromic performance of BAL-EC at different temperatures

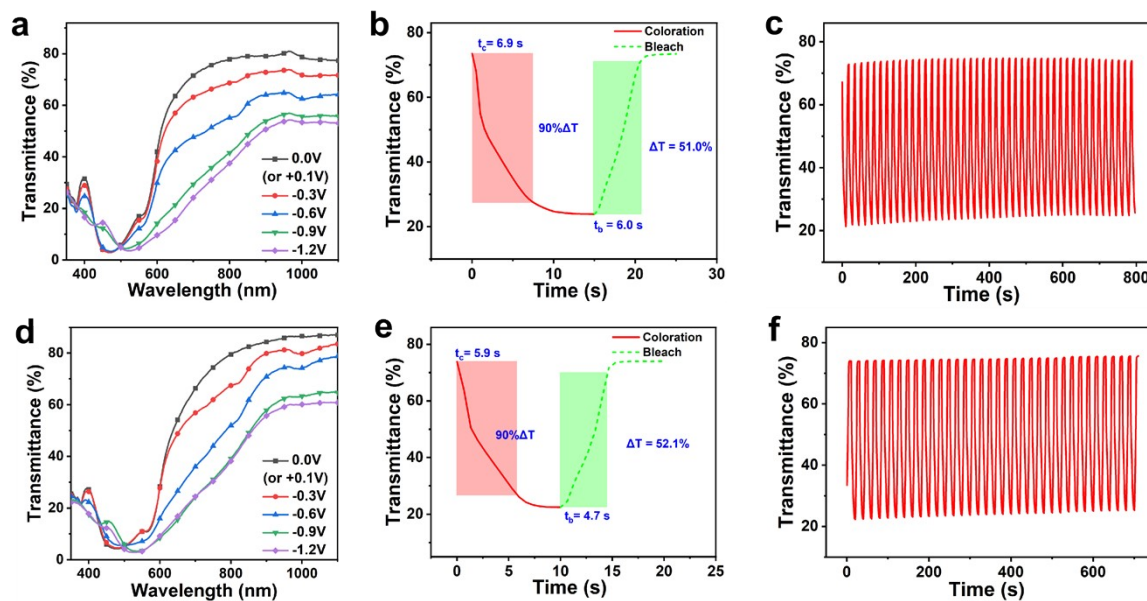


Figure S13. Electrochromic performance of BAL-EC at different temperatures. (a, d) Spectroelectrochemistry ; (b, e) response times; and (c, f) cycling stability tests at 0°C and 40°C.

Figure S14. Electrochromic performance of ECD

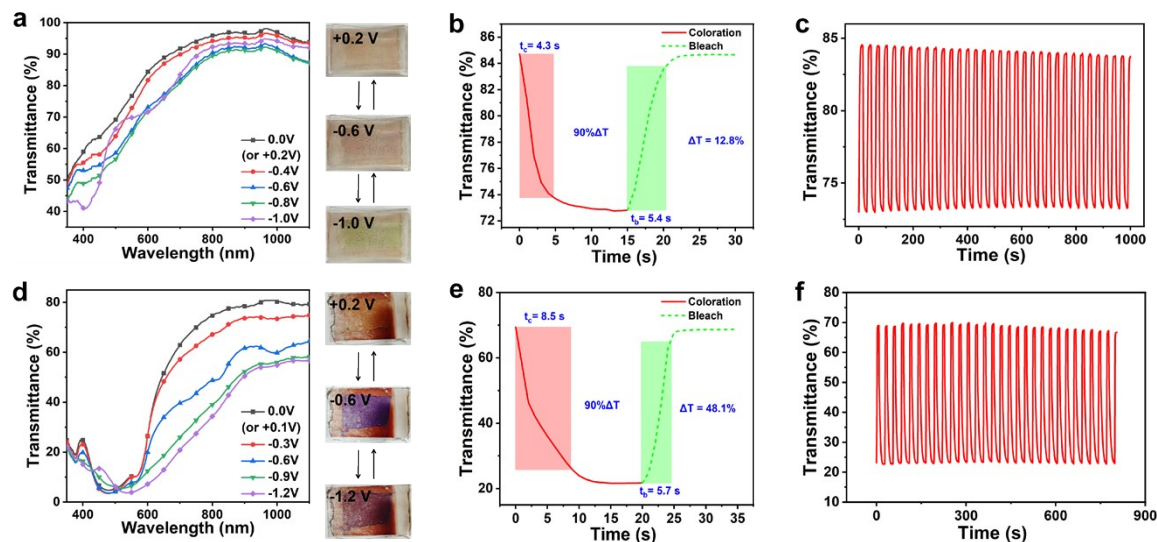


Figure S14. Comparison of the electrochromic performance of ECD-BBL and ECD-BAL. Spectroelectrochemistry and electrochromic digital photos, coloring/bleaching time and optical contrast, long-term cycling life of (a, b, c) ECD-BBL and (d, e, f) ECD-BAL.

Figure S15. GCD curves

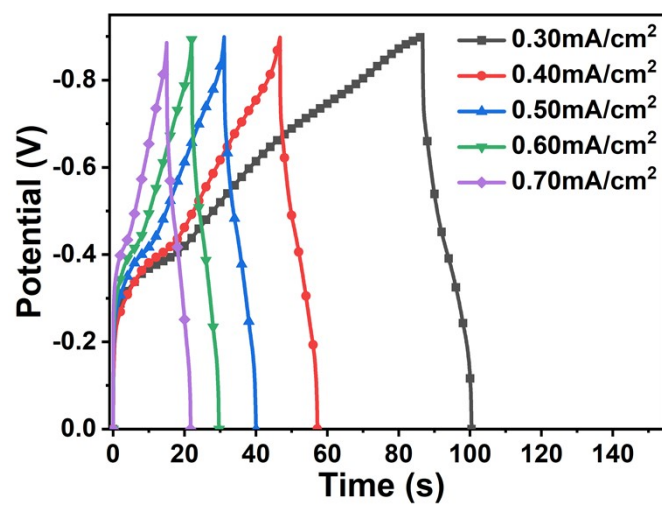


Figure S15. GCD curves of BBL-EC.

Figure S16. Raman spectra during the charge-discharge

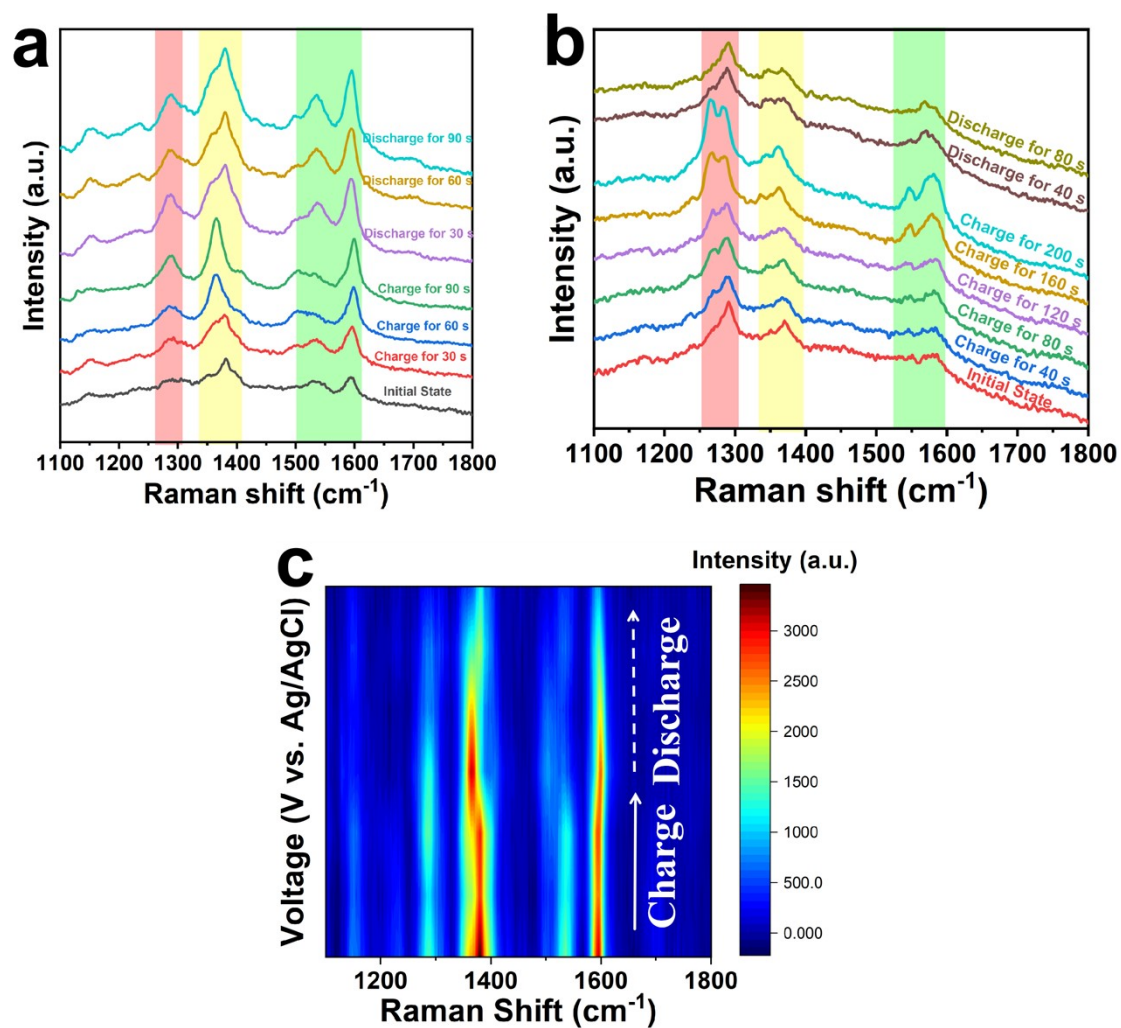


Figure S16. 1D and 2D mapping of the Raman spectra during the charge-discharge process of (a, c) BBL-EC and (b) BAL-EC.

Figure S17. GCD curves and capacity retention

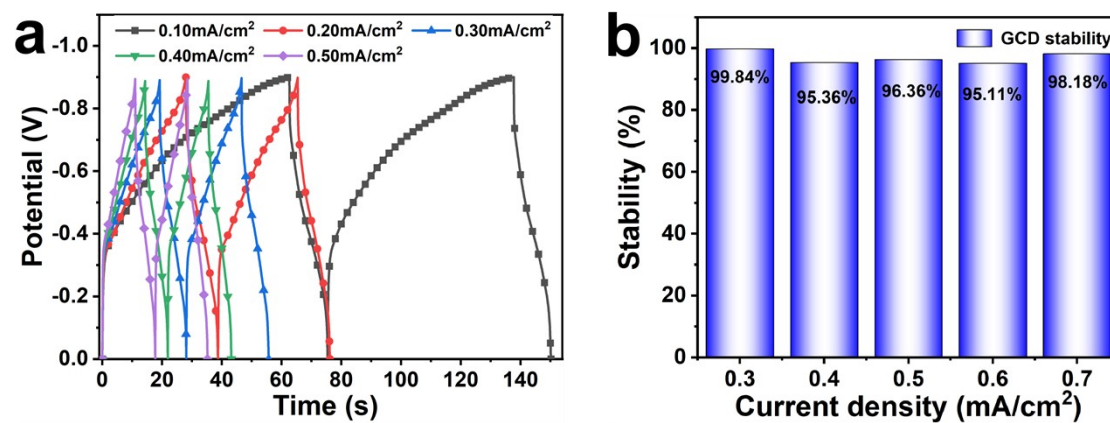


Figure S17. (a) GCD curves and (b) capacity retention of BBL-EC.

Table S1. Detailed fitting parameters of Nyquist plots for BBL-EC ladder polymer

Element	Value	Error	Error %
R_s	2.543	0.15558	6.118
R_{ct}	101.7	0.19682	0.19487
W _o -R	617	20.826	3.3715
W _o -T	0.09786	0.0068706	7.0208
W _o -P	0.56259	0.0029934	0.53207
C _{dl}	1.139×10^{-9}	1.0387×10^{-11}	0.91218

Table S2. Detailed fitting parameters of Nyquist plots for BAL-EC ladder polymer

Element	Value	Error	Error %
R_s	5.341	0.1219	2.2823
R_{ct}	73.45	0.12564	0.17414
W _o -R	0.0071394	2.2257×10^{-5}	0.31175
W _o -T	2.865×10^{-8}	1.0331×10^{-10}	0.36059
W _o -P	0.43228	0.00089009	0.20591
C _{dl}	1.545×10^{-9}	1.6486×10^{-11}	1.067

Table S3. Capacitive Performance of Electrochromic Polymers

EC films	Capacitance	Ref.
TPAB (n-type)	2.38 (mF cm ⁻²)	[1]
PECE (p-type)	3.04 (mF cm ⁻²)	[2]
PECE (p-type)	3.04 (mF cm ⁻²)	
P(TPACz) (p-type)	0.433 (mF cm ⁻²)	[3]
P(TPACz)/WO ₃ (p-type)	1.94 (mF cm ⁻²)	
WO ₃ @PANI (p-type)	147.8 (mF cm ⁻²)	[4]
PM-TPI COF (p-type)	89.8 (mAh g ⁻¹)	[5]
NT-TPI COF (p-type)	141.0 (mAh g ⁻¹)	
TAPA-PMDA PI COF (p-type)	94.07 (mAh g ⁻¹)	[6]
TAPA-NTCDA PI COF (p-type)	86.57 (mAh g ⁻¹)	
PB/PPy-Zn (n-type)	109.9 (mAh g ⁻¹)	[7]
BBL-EC (n-type)	203.49 (mAh g ⁻¹)	This work
BAL-EC (n-type)	257.22 (mAh g ⁻¹)	This work

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Table S4. Performance Comparison Table of n- and p-Type Electrochromic Energy Storage Materials

EC films	ΔT (%)	t_b/t_c ^{a)} (s)	CE (cm ² /C)	Electrolyte	Cyclic Stability	Capacitance	Ref.
pPMQT	32.20 (490 nm)	2.1/1.7	116.52	Zn(TfO) ₂	-	217.13 (mAh g ⁻¹)	[8]
pNTQT	38.49 (474 nm)	2.4/1.8	182.86	Zn(TfO) ₂	-	349.79 (mAh g ⁻¹)	[8]
TPAB	9 (654 nm)	6.49/7.03	41.81	LiClO ₄ /PC	-	2.23 (mF cm ⁻²)	[1]
TPANB	10 (680 nm)	3/4	36.19	LiClO ₄ /PC	-	2.13 (mF cm ⁻²)	[1]
ALV-BF ₄ _X	25	4.03/3.45	56	LiClO ₄ /AC N	Retain 56% (after 100 cycles)	1.713×10 ⁻⁴ (F cm ⁻²)	[9]
ALV-ClO ₄ _X	29	2.84/0.53	82	LiClO ₄ /AC N	Retain 72% (after 100 cycles)	1.538×10 ⁻⁴ (F cm ⁻²)	[9]
ALV-PF ₆ _X	25	3.14/3.02	58	LiClO ₄ /AC N	Retain 83% (after 100 cycles)	2.499×10 ⁻⁴ (F cm ⁻²)	[9]
ALV-TFSI_X	22	3.53/3.76	77	LiClO ₄ /AC N	Retain 80% (after 100 cycles)	8.94×10 ⁻⁴ (F cm ⁻²)	[9]
BBL	30	-/21	-	Drinking Water	Retain 90% (after 1000 cycles)	-	[10]
ProDOT-Am	-	2/>10	-	NH ₄ Cl/H ₂ O	Retain 60% (after 100 cycles)	-	[11]
LPANI	-	0.1/0.39	-	PMMA/PC /LiClO	Retain 98% (after 200 cycles)	-	[12]
BBL@CNT	-	-	-	H ₂ /H ₂ O	Retain 99.7% (after 500 cycles)	268 (mAh g ⁻¹)	[13]
BAL-EC	53.8 (740 nm)	6.4/6.2	20.62	NaCl/H ₂ O	No obvious ΔT % drop (after 1200 cycles)	257.22 (mAh g ⁻¹)	This work

a) t_c : coloring time, t_b : bleaching time.

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